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**Research Papers** 

Running header: preferential uptake of brick shaped nanoparticles in endothelium.

# Differential Internalization of Brick Shaped Iron Oxide Nanoparticles by Endothelial Cells

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### Abstract

Nanoparticles targeting endothelial cells to treat diseases such as cancer, oxidative stress, and inflammation have traditionally relied on ligand-receptor based delivery. The present studies examined the influence of nanoparticle shape in regulating preferential uptake of nanoparticles in endothelial cells. Spherical and brick shaped iron oxide nanoparticles (IONPs) were synthesized with identical negatively charged surface coating. The nanobricks showed a significantly greater uptake profile in endothelial cells compared to nanospheres. Application of an external magnetic field significantly enhanced the uptake of nanobricks but not nanospheres. Transmission electron microscopy revealed differential internalization of nanobricks in endothelial cells compared to epithelial cells. Given the reduced uptake of nanobricks in endothelial cells compared to epithelial cells, and the ability of IONP nanobricks to interfere with caveolae-mediated endocytosis process, a caveolae-mediated pathway is proposed as the mechanism for differential internalization of nanobricks in endothelial cells.

Keywords: shape, iron oxide nanoparticles, drug delivery, nanobrick, endothelial cells, endocytosis, caveolae

### 1 Background

2 There is a growing interest in developing iron oxide nanoparticles (IONPs) as platforms for drug delivery applications.<sup>1-3</sup> In this regard, IONPs provide several advantages: 1) The ability to target to areas of 3 4 interest using externally applied magnetic field, thereby increasing local therapeutic concentrations of 5 IONPs and decreasing potential toxicity related to systemic circulation. 2) Monitoring capabilities for 6 IONPs using MRI. 3) Favorable biocompatibility profile. 4) Flexibility of surface modification to create 7 multifunctional complexes for advanced drug delivery applications involving intracellular or plasma membrane targets. The interaction between IONPs and the cell membrane is largely determined by their 8 physiochemical properties such as surface coating and shape.<sup>4, 5</sup> Our group has previously examined the 9 effect of surface charge on the cellular uptake of IONPs.<sup>6</sup> We found that positively charged IONPs have a 10 11 significantly higher uptake profile compared to negatively charged ones, likely due to electrostatic 12 interactions between positively charged IONPs and the negatively charged plasma membrane of the cell. 13 As a result, negatively charged nanoparticles appeared to be better candidates to advance in our drug 14 delivery platform due to the potential for longer circulation times and reduced clearance. However, the 15 charge related effects on internalization were non-specific as they were present in a variety of different 16 cell types.<sup>6</sup>

17

Various pathological conditions such as cancer, cardiovascular disease, inflammation, and oxidative stress would benefit from the preferential delivery of nanoparticles to the vascular endothelium.<sup>7-9</sup> To achieve the cell specific delivery, targeting ligands are often grafted onto the NPs to increase the delivery efficiency. For instance, intracellular adhesion molecule (ICAM), vascular cell adhesion molecule (VCAM), and platelet-endothelial cell adhesion molecule (PECAM-1) have been used to target endothelial cells.<sup>10-12</sup> However, these approaches are often associated with variability in outcome due to different receptor expression levels between patients or heterogeneity of endothelial cells within different tissue.<sup>13</sup> Therefore, a generalized approach that preferentially target endothelial cells without ligand
 receptor interaction would be advantageous.

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28 In addition to surface charge, nanoparticle shape may also play a role in cell interactions. To date, there 29 are few reports concerning non-spherical nanoparticles. Recent work with theoretical modeling revealed the role of nanoparticle shape and membrane rigidity on cellular uptake.<sup>14</sup> However, only a handful of 30 31 studies provide side-by-side comparison of spherical and non-spherical nanoparticle interactions with biological environments.<sup>15</sup> Recent advances in synthesis techniques have enabled creation of brick 32 shaped IONPs.<sup>16</sup> We hypothesize that changing the IONP shape will influence both the cellular uptake in 33 34 endothelial cells and the ability to augment cell uptake with application of an external magnetic field. 35 Toward that end, the uptake profiles of iron oxide nanospheres and nanobricks of similar size in various 36 cell types were examined to address the impact of IONP shape. In addition, the mechanism of preferential 37 uptake of the nanobrick IONPs in endothelial cells was determined with evidence suggesting a caveolin-38 dependent process. By understanding the relationship between IONP shape and cell surface domains, our 39 work provides insight into the development of IONPs for specifically targeting endothelial cells. 40

41

### 43 Methods

### 44 *Materials*

All chemical reagents were purchased from Sigma Aldrich (St. Louis, MO) and cell culture reagents from
Invitrogen Canada Inc. (Burlington, ON) unless otherwise specified.

### 47 Nanoparticle synthesis and characterization

48 Sphere shaped iron oxide nanoparticles were prepared under mild conditions at room temperature as

49 previously described.<sup>17</sup> They were prepared by adding N-

50 (trimethoxysilylpropyl)ethylenediaminetriacetate trisodium salt (EDT, 3 mmol, from a solution

51 concentration of 45% in water) (Gelest, Morrisville, PA) directly to a reaction vessel containing IONPs .

52 The mixture was allowed to react overnight with stirring and the final product was purified by dialysis

53 (MWCO 30000) against deionized (DI) water over 48 hours and was freeze dried and resuspended in

54 sterile PBS prior to experiments. Brick shaped IONPs with EDT surface coating was synthesized and

55 prepared as recently described.<sup>16</sup>

56

Nanoparticle crystallographic properties of both the nanospheres and nanobricks were measured with
powder x-ray diffraction experiments using a Brüker diffractometer (D8 Discover with Davinci;
Karlsruhe, Germany). Both nanoparticle systems were identified as iron oxide through Reitveld
refinement incorporating the effects of the nanocrystalline nature of the samples (e.g. Scherrer broadening
effects).

62

The IONP size distribution in DI water was determined initially through photon correlation spectroscopy (PCS) at a fixed scattering angle (90°) using a Horiba Nano-Partica SZ-100 series instrument (Horiba Instruments Inc., Irvine, CA). The same instrument allowed for the assessment of particle surface charge (zeta potential) by the measurement of IONP electrophoretic mobilities using phase analysis light 67 scattering. The magnetization of dry nanoparticle powder samples were recorded at room temperature as a 68 function of applied magnetic field (0 - 4 T) using a Quantum Design MPMS XL SQUID magnetometer 69 (San Diego, CA).

### 70 Cell culture

A mouse brain derived microvessel endothelial cell line, bEnd.3 (American type tissue culture collection, Manassas, VA), was used as a cell culture model of the blood-brain barrier (BBB). The bEnd.3 cells (passage number 15-30) were cultured in DMEM (Hyclone, Logan, UT) supplemented with 10% heatinactivated FBS (Hyclone, Logan, UT), 50 U/mL penicillin and streptomycin (MP Biomedicals, Solon, OH) at 37°C and 5% CO<sub>2</sub>. Cells were expanded in T-75 tissue culture flasks, and seeded at 2x10<sup>4</sup> cells per cm<sup>2</sup> on 6 or 12 well plates for uptake and cytotoxicity studies, respectively. Culture medium was changed every 2 days. All experiments were performed on confluent monolayers (typically 4-5 days post seeding)

78 seeding).

### 79 Cellular Uptake of IONP compositions

80 Confluent monolayers of bEnd.3 cells grown on 6-well culture plates (Costar, Lowell, MA) were treated 81 with culture media containing either nanosphere or nanobrick compositions  $(2.5 \mu g/mL - 100 \mu g/mL of$ 82 Fe). After treatment with IONPs, cells were placed in a humidified  $CO_2$  incubator maintained at 37°C. 83 After 4 hours, the IONP solutions were removed and the cell monolayers were washed 3X with ice cold 84 phosphate buffered saline (PBS) to remove unbound nanoparticles. Cells were lysed by the addition of 85 500 µl of 0.2 M NaOH and IONP content determined based on the ferrozine assay described below. 86 Cellular accumulation was examined in both the presence and absence of a static magnetic field created 87 by placing the cells over a platform containing cylindrical rare earth magnets (19mm diameter, 3mm 88 height) (Lee Valley, Ottawa, ON, Canada). Cells remained in the magnetic field for the duration of the 89 experiment.

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For mechanistic studies of IONP uptake experiments were performed at both 4<sup>o</sup> and 37<sup>o</sup> C and in the 90 91 presence of various endocytotic inhibitors. Cells were pretreated with chlorpromazine (7 µg/mL), 92 methyl-beta-cyclodextrin (10mM), genistein (200µM), monensin (25µM), or cytochalasin D (5 µg/mL) 93 for 30 min at 37°C. Cells were exposed to the nanobricks for 1 h at 37°C in the presence of the various 94 endocytotic inhibitors. Cell association of nanobricks was determined as described below. 95 Additional studies using known markers of caveolae mediated endocytosis, alexa fluor 488-labeled 96 cholera toxin subunit B (CTB) and tetramethylrhodamine conjugated bovine serum albumin (BSA) were 97 examined for cellular uptake. For these studies, cells were exposed to CTB (3.5 µg/mL), BSA (10 98 µg/mL) for 2 h either alone or following 15-min pretreatment with various concentrations of the iron-99 oxide nanobricks. Cells were washed and lysed and fluorescence determined using a Synergy HT plate 100 reader (BioTek, Winooski, VT).

### 101 Analytical assay for measuring IONPs

102 Quantitative determination of IONP content in cell and media samples was performed using the Ferrozine 103 assay. As the Ferrozine assay is an absorbance-based assay for determining soluble iron concentrations. 104 IONPs in the cell lysate and media samples were first solubilized by adding 500 µL of concentrated HCl 105  $(\sim 12M)$  to 500 µL of cell lysate or media samples. This mixture was incubated for 1 h at room 106 temperature with gentle shaking and then neutralized with 500 µL of 12M NaOH. Once the samples were 107 neutralized,  $120 \,\mu\text{L}$  of hydroxylamine hydrochloride (2.8 M) in 4M HCl was added and the samples 108 incubated for 60 min at room temperature with gentle shaking. Following this incubation, 50 uL of 10 M 109 ammonium acetate solution (pH 9.5) and 300 uL of 10mM ferrozine in 0.1M ammonium acetate solution 110 were added to each sample. Absorbance was measured at 562 nm using a Synergy HT plate reader 111 (BioTek, Winooski, VT). Quantitative assessment of IONP concentration was based on a standard curve 112 prepared by serial dilutions of 1000 ppm iron atomic absorption standard (Fisher Scientific, Ottawa, ON). 113 Samples from the cell lysates were normalized for protein content using BCA protein assay kit (Pierce, 114 Rockford, IL).

### 115 Electron Microscopy

116 The cellular localization of IONPs compositions was examined using transmission electron microscopy. 117 For these studies, cells were incubated with IONPs at 50µg/mL concentration in media for 2 hours. After 118 incubation, cells were washed 3X with PBS and collected using 0.25% trypsin EDTA (Hyclone, Logan, 119 UT). After centrifugation, the cell pellets where fixed in 3% glutaraldehyde in 0.1M phosphate buffer (pH 120 7.3), followed by post-fixation in 1% osmium tetroxide in 0.1M phosphate buffer (pH 7.3). Cells were then dehydrated and embedded in Epon 812 using standard techniques.<sup>18</sup> Thin sections were stained with 121 122 uranyl acetate and lead citrate, viewed and photographed in a Philips CM 10 electron microscope (FEI, 123 Hillsboro, OR, USA). In order to eliminate observer bias, sections were examined without foreknowledge 124 of their source.

### 125 Statistical analysis

126 All data were expressed as mean  $\pm$  SEM. All values were obtained from at least three independent

127 experiments. Statistical significance was evaluated using one-way ANOVA followed by post-hoc

128 comparison of the means using the Tukey's test.

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### 130 **Results**

### 131 *Physico-chemical characterization of IONPs*

Physico-chemical parameters of the nanobrick and nanosphere compositions are provided in Figure 1. 132 133 Both nanobricks and nanospheres were silanized and had free carboxylic acid functional groups on their 134 surfaces resulting in zeta potentials of approximately -40 mV. The TEM images confirming the different shapes of IONPs have previously been published.<sup>16, 17</sup> The dimensions of IONP core for the nanobricks 135 136 were approximately 15 nm x 10 nm x 5nm while the nanosphere was around 8 nm in diameter. The 137 saturation magnetization, determined by fitting the high field magnetization to a straight line after background subtraction (diamagnetic signal from the sample holder), was  $50 \pm 5$  A m<sup>2</sup> kg<sup>-1</sup> and  $10 \pm 2$  A 138  $m^{2} kg^{-1}$  for the nanobricks and nanospheres, respectively. The saturation magnetization is the largest 139 140 magnetization that a material can exhibit in an applied magnetic field. Samples with larger saturation 141 magnetizations have greater magnetic response and thus are likely more useful for targeted delivery using 142 an externally applied magnetic field. A more detailed description of the nanoparticle's characterization is 143 provided in the Supplementary Information.

### 144 Preferential internalization of nanobrick in endothelial cells

Quantitative uptake analysis was performed in the bEnd.3 mouse brain endothelial cell line. (Figure 2a) In 145 146 absence of magnetic field, there was a significantly greater uptake of nanobrick compared to nanosphere 147 compositions at all concentrations above  $5\mu g/mL$ . In the presence of external magnetic field, cell 148 association of nanobrick was substantially increased compared to nanosphere. At the highest 149 concentration examined (100µg/mL), there was a 30-fold and 10-fold increase in uptake of nanobricks 150 compared to nanospheres with and without a magnetic field, respectively. This surprising finding suggests 151 that despite the negative surface charge, brick shaped IONPs are taken up by brain endothelial cells to a 152 greater extent than spherical counterparts. Furthermore, the shape of IONPs affected their magnetization 153 value and ability to interact with cells in the presence of an external magnetic field gradient. Potential

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toxicity of nanobricks related to bEnd.3 cells was investigated (Figure S1). Nanobricks appear to be nontoxic even at 100 ug/mL concentration.

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157 Uptake studies with the nanobrick was expanded to include primary human lung and brain endothelial 158 cells as well as Madin-Darby canine kidney (MDCK) epithelial cell line with two fold purpose: 1) To 159 investigate whether there was any selectivity to endothelial cells versus epithelial cells; and 2) To 160 examine whether enhanced uptake of the nanobricks was specific to brain endothelial cells compared to 161 other endothelial beds. External magnetic field significantly enhanced cellular uptake of nanobricks in 162 both the lung and brain microvessel endothelial cells but not in epithelial cells (MDCK) (Figure 2b). 163 While accumulation of the nanobrick IONPs in the presence of an external magnetic field was 164 significantly greater in the endothelial cells compared to the epithelial cell line, there was no apparent 165 differences between endothelial cells from different vascular beds (Figure 2b). Transmission electron 166 microscopy (TEM) of the various cell preparations confirmed that nanospheres were loosely bound on the 167 cell surface and not internalized by bEnd.3 cells (Figure 3a) or human hepatocellular liver carcinoma cell 168 line HepG2 (Figure 3b). By contrast, large amounts of nanobricks were found inside the bEnd.3 169 endothelial cells (Figure 3c) but few were found inside HepG2 (Figure 3d) or MDCK epithelial cell lines 170 (Figure 3e), confirming the finding that the nanobricks were selectively internalized in endothelial cells.

### 171 Internalization of nanobrick in bEnd.3 cells via caveolae mediated endocytosis

To understand the selectivity of nanobricks to endothelial cells, we examined the potential mechanism of internalization. The observation that nanobrick accumulation in bEnd.3 cells was temperature dependent with significantly less uptake at 40 compared to 370 C suggested an energy dependent endocytic process (Figure S2). To determine which type of process was responsible for uptake of the nanobricks, confluent bEnd.3 cell monolayers were pretreated with inhibitors for clathrin mediated endocytosis

177 (chlorpromazine), caveolae mediated endocytosis (MβCD, genistein), macropinocytosis (cytochalasin D),

178 and endosome maturation (monensin) for 30 min, and uptake of nanobricks at 100 ug/mL was determined

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179 (Figure 4a). There was a significant inhibition of nanobrick uptake in the M $\beta$ CD and genistein treatment 180 groups (Figure 4a). These findings were confirmed in TEM studies showing diminished IONP association 181 in bEnd3 in the presence of genistein compared to controls receiving the nanobricks alone (Figure 4b,c). 182 The inhibition observed with genistein and MbCD was not attributable to toxicity based inhibition of 183 uptake as none of the inhibitors examined showed cytotoxicity at the concentrations examined in bEnd.3 184 cells (Figure S3). None of the other treatment groups examined significantly impacted on nanobrick 185 accumulation in bEnd3 cells (Figure 4a), suggesting that nanobrick internalization in bEnd.3 cells was 186 mediated via a caveolae dependent endocytosis pathway.

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188 To ascertain whether elevated caveolae mediated endocytosis in endothelial cells contributes to the 189 selective internalization of nanobricks observed in endothelial cells, additional studies were performed 190 with known markers of caveolae-mediated endocytosis. The uptake of CTB and BSA is 6-fold and 12-191 fold greater in bEnd.3 cells than MDCK cells, respectively. (Figure 5a) The increase in uptake of CTB 192 and BSA in the bEnd3 was correlated with an increase in the expression of caveolin-1 compared to 193 epithelial MDCK cell line. Expression of caveolin-1 in another endothelial cell line hCMEC/D3 was also 194 elevated. (Data not shown) Additional evidence of potential interaction of nanobricks in caveolae-195 mediated endocytosis is the ability of the nanobricks to inhibit the uptake of fluorescently-labeled BSA in 196 a concentration dependent manner. (Figure 5b)

### 198 Discussion

Previous studies by our laboratory and others<sup>19-22</sup> have demonstrated the importance of surface charge of 199 200 IONPs for cellular uptake. In the present study, negatively charged IONPs of different shape were utilized 201 to examine the influence of shape on cellular uptake. While there are some publications regarding the synthesis of different shaped IONPs,<sup>23, 24</sup> these methods are typically thermal decomposition based 202 203 generating nanoparticles that are not directly dispersible in water and therefore not readily amenable to 204 cell based interactions. With regards to the possible impact of IONP shape on cell uptake, to date the 205 shape-dependent impact on the cell accumulation have been limited to macrophages, fibroblast, and cancer cells.<sup>25-27</sup> The current studies are the first to demonstrate a shape related effect on IONP 206 207 accumulation in endothelial cells. Our results demonstrated that brick shaped IONPs were preferentially 208 taken up by endothelial cells compared to sphere shaped IONP with identical surface coatings. In 209 addition, when studies were performed in the presence of a magnetic field, the endothelial sensitivity for 210 nanobrick accumulation was even more apparent, being substantially greater than epithelial cell 211 preparations. The selective uptake of the nanobricks by endothelial cells appears to be due to caveolae-212 mediated endocytosis, which is more prevalent in endothelial cells compared to epithelial cells examined. 213 214 As the nanobricks are slightly larger than the nanospheres (15 x 10 x 5nm for nanobricks vs 8nm diameter 215 nanospheres), there is a possibility that differences in size may also contribute to the increased 216 accumulation of IONP nanobricks in the endothelial cells. Previous studies demonstrated a size dependent effect on IONP accumulation in the Caucasian colon adenocarcinoma cell line (Caco2).<sup>28</sup> However, it 217 218 should be noted that those IONP had a positive surface charge and were considerably larger (30 - 100 nm 219 core diameter) than the IONP used in the present study. Given the EDT surface coating used in the 220 present study, the studies of Saito et al reporting no size dependent effect on the accumulation of 221 negatively charged IONP in cells may be more relevant. In this study, the cellular uptake of alkali-treated 222 dextran coated IONPs (-15mV zeta potential) with particle sizes of 28 and 74 nm, were compared to that

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of carboxymethyl dextran IONPs (-24mV zeta potential) of similar size in a macrophage cell line,
RAW264. While there was a clear surface charge dependency in cell accumulation, with the alkali-treated
dextran coated IONPs having greater accumulation than the carboxymethyl dextran coated IONPs, no
significant difference was found in the cellular accumulation of the large (74 nm diameter) and small (28
nm diameter) IONPs of the same coating.<sup>29</sup> Taken together, these studies would suggest that for the
negatively charged particles with low membrane association, size is not the predominant factor for
determining cellular accumulation.

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231 We found that brick shaped IONPs could enhance the affinity between surface coating and cell membrane 232 compositions. An increased contact area with the cell surface provides potentially more sites for 233 interaction and has been previously identified as an important contributor to enhance nanoparticle 234 targeting effects.<sup>30</sup> Our finding is in line with recent publications of shape related effect on polystyrene 235 NPs. Barua et al reported that rod shaped polystyrene NPs have enhanced antibody binding specificity to three breast cancer cell lines compared to spherical and disk shaped NPs.<sup>31</sup> Using *in silico* and *in vivo* 236 237 approaches, Kohlar and colleagues demonstrated rod shaped polystyrene NPs with antibody against 238 intracellular adhesion molecule (ICAM) or transferrin receptor exhibited higher internalization in brain and lung endothelial cells than spherical counterparts under flow conditions.<sup>32</sup> Hence, it is speculated that 239 240 by changing the IONPs from sphere to brick, the negatively charged surface coating interacts with 241 multiple discrete sites on the cell membrane that contributes to the selective binding of the nanobricks to 242 endothelial cells. This may provide advantages especially when second-generation nanobrick 243 compositions are created that have additional endothelial ligand targeting capabilities. We further 244 hypothesize that a low affinity ligand grafted on nanobrick surface would exhibit a stronger interaction to 245 its receptor than grafted on nanospheres. Such studies are currently ongoing. 246

Generally speaking, physiochemical properties of IONPs such as shape and surface coating would be
expected to have an impact on the internalization pathway. Studies by Hsu et al demonstrated that

249 chitosan coated IONPs and hyaluronan-modified chitosan coated IONPs may activate different 250 endocytosis pathways. In these studies, chitosan coated IONPs favored uptake by clathrin mediated 251 endocytosis, while the hyaluronan modified chitosan favored more caveolae mediated endocytosis 252 routes.<sup>33</sup> Our previously published studies using positively charged amino silane coated and negatively 253 charged amino silane with EDT functionalized end groups demonstrated that the negatively charged EDT 254 coated nanospheres had a much lower cellular accumulation than the positive charged IONP. This 255 observation, that negatively charged IONP had lower uptake than positively charged IONP of similar size 256 and shape, held up across a variety of cells including brain endothelial cells, as well as primary cultured neurons and astrocytes.<sup>6</sup> This is due to the fact that negatively charged surface reduces nonspecific 257 258 electrostatic interactions between the NPs and cell surface. The results of the present study, that the EDT 259 coated nanobricks with identical surface coating and similar size as the nanospheres showed dramatic 260 increases in uptake in endothelial cells, suggest that while the coating of the nanoparticle is important, so 261 too is the shape. Furthermore at least for the EDT coated IONPs, shape appears to be a bigger determinant 262 of caveolae-mediated vesicular transport.

263

264 Of the various vesicular internalization processes, caveolae mediated endocytosis is predominantly found in endothelial cells.<sup>34</sup> Therefore, targeting to endothelial cells may be achieved by interacting with 265 266 caveolae localized in lipid rafts within the plasma membrane. The current study certainly points to a 267 caveolae-mediated mechanism for the endothelial selective uptake of the nanobrick IONP. The evidence 268 in support of this is the increased expression of caveolin in endothelial cells compared to the epithelial 269 cells and the ability of inhibitors of caveolae-mediated uptake to significantly reduce nanobrick IONP 270 accumulation in endothelial cells. In addition, the nanobrick IONPs were able to prevent the cellular 271 uptake of two macromolecules, CTB and BSA, which are known to enter into endothelial cells through 272 caveolae-mediated endocytosis in a concentration dependent manner consistent with competitive 273 inhibition of caveolae biding sites. Previous studies grafting anionic polyelectrolytes of varied 274 hydrophobicity to nanospheres reported endothelial cell targeting of NPs via a caveolae-mediated

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endocytic process.<sup>35</sup> These findings together suggest that non-spherical nanoparticles with negative
surface charges are likely to have the greatest affinity for caveolae-based uptake.

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278 Caveolae are formed by a group of caveolin protein binding to cholesterol in the lipid raft region of the cell membrane.<sup>36</sup> Although surface chemistry and functional groups can influence IONP cell interaction, 279 280 it has been reported that negatively charged IONPs can interact with cationic lipid domains in the lipid raft.<sup>37</sup> Caveolae are enriched in endothelial cells and present in muscle, fibroblast, and adipocytes.<sup>38</sup> 281 282 Following the pinch off of caveolae from the lipid raft, the fate of caveolae is dependent on the cell type 283 in which endocytosis occurs. In non-endothelial cells, caveolae are subjected to the endosomal-lysosomal 284 system. In endothelial cells, caveolae may bypass the lysosome and transport cargo through vesicular processes across the endothelial cell layer.<sup>39, 40</sup> For this reason, the nanobrick IONPs may potentially be 285 286 exploited for drug and gene delivery applications to tissues underlying endothelial cells such as the brain. 287 These studies are currently ongoing.

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289 Compared to nanospheres, the nanobricks have an increased responsiveness in an applied magnetic field 290 gradient. Based on the modeling and simulation data, (the nanobricks have a preferred direction of 291 magnetization along their largest dimension (see Figure S4) As such, an externally applied magnetic field 292 will act to more preferentially to align the smaller dimensions of the nanobricks along the cell surface, 293 decreasing the area of interaction and thus limiting the effect of the steric repulsion between the cell 294 surface and nanobrick coating. The proposed behavior of the nanobricks in the externally applied 295 magnetic field may help explain the significant increase in uptake of the nanobricks compared to the 296 nanosphere observed in the presence of a magnetic gradient in the present study. In addition to the 297 potential for tissue targeting using external magnetic fields, the magnetic properties of the nanobricks 298 made them ideal candidates for magnet resonance imaging agents. Nanobricks show large and constant 299 transverse relaxivity  $(r_2)$  for medium and high-field MRI compared to gadolinium based contrast agents 300 that peaks at 20 MHz and decreases quickly with high magnetic fields.<sup>16</sup>

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302 The preferential uptake of nanobrick IONPs within vascular endothelial cells combined with the enhanced 303 targeting through application of external magnetic fields has several potential therapeutic applications. 304 The ability to target to the endothelial cells within tumor microvasculature is a prime application for this 305 technology platform. It is generally accepted that angiogenesis is crucial for tumor growth, evasion and metastasis.<sup>41</sup> The creation of new blood vessels to supply oxygen and nutrients to tumor cells is a 306 307 necessary requirement for solid organ tumor growth. Thus, anti-angiogenesis therapy has emerged as a 308 viable treatment strategy to control tumor growth. Recent studies demonstrated the potential of PEG-PLGA nanoparticles for tumor neo-vasculature and tumor cells dual-targeting drug deliverv.<sup>42</sup> The ability 309 310 to focus an external magnetic field within the tumor stroma will not only increase the local concentration 311 of IONPs but also facilitate improved internalization of nanobrick IONPs in endothelial cells. An 312 anticipated result of such focused targeting of the IONPs would be enhanced delivery and potential 313 destruction of the tumor neovasculature. While current anti-angiogenic therapies have been limited in the clinic due to the development of resistance, <sup>43</sup> the targeting of nanobrick IONPs to endothelial cells using 314 315 shape and magnetic fields would make resistance to these delivery vehicles less probable. 316

## 318 Conclusion

Nanoparticle shape plays an important role in the cellular internalization process. Targeting nanoparticles
to endothelial cells can be achieved by modification of shape from a sphere to a brick. Nanobricks
exhibited an improved cellar uptake profile compared to nanospheres despite a negative surface charge.
The larger overall magnetization of the nanobricks resulted in an enhanced uptake in the presence of an
external magnetic field. The preferential uptake of nanobricks in endothelial cells was mediated via
caveolae dependent endocytosis. Our results demonstrate that shape modification offers a general
approach to achieve targeted delivery.

### 328 Table 1: Physico-chemical properties of nanospher and nanobrick IONPs.

	Nanospheres	Nanobricks
Surface Coating	-0-Si 0 -0-S 0 -0-S 0 -0-Si	
TEM Size *	8 nm	15 x 10 x 5 nm
Zeta Potential **	-39 ± 3 mV	-45 ± 3 mV
Saturation Magnetization	10 ± 2 A m <sup>2</sup> kg <sup>-1</sup>	50 ± 5 A m <sup>2</sup> kg <sup>-1</sup>

331 \*\* surface charges of IONPs were measured in triplicate samples using a Nano-partica SZ-100 series

instrument from Horiba. Values represent the mean  $\pm$  SEM (n=3).

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### 334 Figures



337

336 Fig. 1





### 345 Fig. 2



346

347 **Fig 2** Representative TEM images of nanospheres (a, b) and nanobricks (c, d, e) in bEnd.3 (a, c), HepG2

348 (b, d), and MDCK cells (e). The boxed region in each image is magnified 3 times.





**Fig 3** Effect of various endocytosis inhibitors on cellular uptake of nanobricks in bEnd.3 cells. The internalization of nanobricks was significantly decreased by treatment with M $\beta$ CD and Genistein, inhibitors of caveolae mediated endocytosis. This is confirmed by representative TEM that shows substantially greater internalization of nanobricks under control conditions (b) compared to cells treated with genistein (c). The arrows point to nanoparticles. Values represent the mean ± SEM for three cell monolayers per treatment group; \* p<0.05 compared to control.



**Fig 4** Probing caveolae mediated endocytosis pathway in bEnd.3 and MDCK cells using fluorescently labeled BSA and CTB (markers for caveolae-mediated uptake). Caveolae-mediated pathway is prominent in bEnd.3 cells and significantly lower in MDCK cells (a). Western blot analysis shows higher level of caveolin-1 expression on bEnd.3 cells (b). The ability of nanobricks to inhibit the uptake of fluorescently labeled BSA and CTB in bEnd.3 cells suggests a competitive binding of the nanobricks to the caveolae (c). Values are expressed as the mean  $\pm$  SEM for three cell monolayers per treatment group. \*\*\* indicate p<0.001, \*\*\*\* indicate p<0.0001.

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# 378 **Reference**

379	1.	A. K. Silva, N. Luciani, F. Gazeau, K. Aubertin, S. Bonneau, C. Chauvierre, D. Letourneur and
380		C. Wilhelm, Nanomed., 2015, 11, 645-655.
381	2.	M. E. Mertens, S. Koch, P. Schuster, J. Wehner, Z. Wu, F. Gremse, V. Schulz, L. Rongen, F.
382		Wolf, J. Frese, V. N. Gesche, M. van Zandvoort, P. Mela, S. Jockenhoevel, F. Kiessling and T.
383		Lammers, <i>Biomaterials</i> , 2015, <b>39</b> , 155-163.
384	3.	J. P. Chen, P. C. Yang, Y. H. Ma and Y. J. Lu, J. Nanosci. Nanotechnol., 2011, 11, 11089-11094.
385	4.	S. Laurent, A. A. Saei, S. Behzadi, A. Panahifar and M. Mahmoudi, Expert Opin. Drug Deliv.,
386		2014, <b>11</b> , 1449-1470.
387	5.	K. Kaaki, K. Herve-Aubert, M. Chiper, A. Shkilnyy, M. Souce, R. Benoit, A. Paillard, P. Dubois,
388		M. L. Saboungi and I. Chourpa, <i>Langmuir</i> , 2012, <b>28</b> , 1496-1505.
389	6.	Z. Sun, V. Yathindranath, M. Worden, J. A. Thliveris, S. Chu, F. E. Parkinson, T. Hegmann and
390		D. W. Miller, Int. J. Nanomedicine, 2013, 8, 961-970.
391	7.	J. Mai, Y. Huang, C. Mu, G. Zhang, R. Xu, X. Guo, X. Xia, D. E. Volk, G. L. Lokesh, V.
392		Thiviyanathan, D. G. Gorenstein, X. Liu, M. Ferrari and H. Shen, J. Control. Release, 2014, 187,
393		22-29.
394	8.	E. D. Hood, M. Chorny, C. F. Greineder, S. A. I, R. J. Levy and V. R. Muzykantov, <i>Biomaterials</i> ,
395		2014, <b>35</b> , 3708-3715.
396	9.	T. Novokhatska, S. Tishkin, V. Dosenko, A. Boldyriev, I. Ivanova, I. Strielkov and A. Soloviev,
397		<i>Eur. J. Pharmacol.</i> , 2013, <b>718</b> , 401-407.
398	10.	J. Hsu, T. Bhowmick, S. R. Burks, J. P. Kao and S. Muro, J. Biomed. Nanotechnol., 2014, 10,
399		345-354.
400	11.	M. Dan, D. B. Cochran, R. A. Yokel and T. D. Dziubla, <i>PLoS ONE</i> , 2013, 8, e81051.
401	12.	H. Yang, F. Zhao, Y. Li, M. Xu, L. Li, C. Wu, H. Miyoshi and Y. Liu, Int. J. Nanomedicine,
402		2013, 8, 1897-1906.
403	13.	K. Hida, N. Ohga, K. Akiyama, N. Maishi and Y. Hida, <i>Cancer Sci.</i> , 2013, <b>104</b> , 1391-1395.
404	14.	S. Dasgupta, T. Auth and G. Gompper, Nano Lett., 2014, 14, 687-693.
405	15.	N. Hao, L. Li and F. Tang, J. Biomed. Nanotechnol., 2014, 10, 2508-2538.
406	16.	M. Worden, M. A. Bruckman, MH. Kim, N. F. Steinmetz, J. M. Kikkawa, C. LaSpina and T.
407		Hegmann, Journal of Materials Chemistry B, 2015, 3, 6877-6884.
408	17.	Z. Sun, M. Worden, Y. Wroczynskyj, V. Yathindranath, J. van Lierop, T. Hegmann and D. W.
409		Miller, Int. J. Nanomedicine, 2014, 9, 3013-3026.
410	18.	J. H. Luft, The Journal of biophysical and biochemical cytology, 1961, 9, 409-414.
411	19.	Z. Sun, V. Yathindranath, M. Worden, J. A. Thliveris, S. Chu, F. E. Parkinson, T. Hegmann and
412		D. W. Miller, Int. J. Nanomedicine, 2013, 8, 961-970.
413	20.	X. M. Zhu, Y. X. Wang, K. C. Leung, S. F. Lee, F. Zhao, D. W. Wang, J. M. Lai, C. Wan, C. H.
414		Cheng and A. T. Ahuja, Int. J. Nanomedicine, 2012, 7, 953-964.
415	21.	Y. Ge, Y. Zhang, J. Xia, M. Ma, S. He, F. Nie and N. Gu, Colloids Surf. B Biointerfaces, 2009,
416		<b>73</b> , 294-301.
417	22.	A. Villanueva, M. Canete, A. G. Roca, M. Calero, S. Veintemillas-Verdaguer, C. J. Serna, P.
418		Morales Mdel and R. Miranda, <i>Nanotechnology</i> , 2009, <b>20</b> , 115103.
419	23.	B. Karagoz, J. Yeow, L. Esser, S. M. Prakash, R. P. Kuchel, T. P. Davis and C. Boyer, <i>Langmuir</i> ,
420		2014, <b>30</b> , 10493-10502.
421	24.	C. de Montferrand, L. Hu, I. Milosevic, V. Russier, D. Bonnin, L. Motte, A. Brioude and Y.
422		Lalatonne, Acta Biomater., 2013, 9, 6150-6157.
423	25.	W. Xia, H. M. Song, Q. Wei and A. Wei, <i>Nanoscale</i> , 2012, <b>4</b> , 7143-7148.
424	26.	S. Gil, C. R. Correia and J. F. Mano, Advanced healthcare materials, 2015, 4, 883-891.
425	27.	D. Cheng, X. Li, G. Zhang and H. Shi, Nanoscale research letters, 2014, 9, 195.

28.	W. Zhang, M. Kalive, D. G. Capco and Y. Chen, Nanotechnology, 2010, 21, 355103.
29.	S. Saito, M. Tsugeno, D. Koto, Y. Mori, Y. Yoshioka, S. Nohara and K. Murase, Int. J.
	Nanomedicine, 2012, 7, 5415-5421.
30.	E. M. Munoz, J. Correa, R. Riguera and E. Fernandez-Megia, J. Am. Chem. Soc., 2013, 135,
	5966-5969.
31.	S. Barua, J. W. Yoo, P. Kolhar, A. Wakankar, Y. R. Gokarn and S. Mitragotri, Proc. Natl. Acad.
	Sci. U. S. A., 2013, <b>110</b> , 3270-3275.
32.	P. Kolhar, A. C. Anselmo, V. Gupta, K. Pant, B. Prabhakarpandian, E. Ruoslahti and S.
	Mitragotri, Proc. Natl. Acad. Sci. U. S. A., 2013, 110, 10753-10758.
33.	S. H. Hsu, T. T. Ho and T. C. Tseng, Biomaterials, 2012, 33, 3639-3650.
34.	R. G. Parton and K. Simons, Nat. Rev. Mol. Cell Biol., 2007, 8, 185-194.
35.	J. Voigt, J. Christensen and V. P. Shastri, Proc. Natl. Acad. Sci. U. S. A., 2014, 111, 2942-2947.
36.	M. Murata, J. Peranen, R. Schreiner, F. Wieland, T. V. Kurzchalia and K. Simons, Proceedings of
	the National Academy of Sciences of the United States of America, 1995, 92, 10339-10343.
37.	I. M. Adjei, B. Sharma and V. Labhasetwar, Advances in experimental medicine and biology,
	2014, 811, 73-91.
38.	E. J. Smart, G. A. Graf, M. A. McNiven, W. C. Sessa, J. A. Engelman, P. E. Scherer, T. Okamoto
	and M. P. Lisanti, Molecular and cellular biology, 1999, 19, 7289-7304.
39.	A. El-Sayed and H. Harashima, Mol. Ther., 2013, 21, 1118-1130.
40.	M. Simionescu, D. Popov and A. Sima, Cell Tissue Res., 2009, 335, 27-40.
41.	S. M. Weis and D. A. Cheresh, <i>Nat. Med.</i> , 2011, <b>17</b> , 1359-1370.
42.	G. Gu, Q. Hu, X. Feng, X. Gao, J. Menglin, T. Kang, D. Jiang, Q. Song, H. Chen and J. Chen,
	<i>Biomaterials</i> , 2014, <b>35</b> , 8215-8226.
43.	J. M. Ebos and R. S. Kerbel, Nat. Rev. Clin. Oncol., 2011, 8, 210-221.



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