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Pulsatile Release from pH Triggered Imidazoline Switchable Surfactant Liposomes

 Dylan Y. Hegh,^a Sean M. MacKay^a, and Eng Wui Tan*^a

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The incorporation of an imidazoline (IDZ) based switchable surfactant into the lipid membrane of a liposome produces a system that can be triggered to release its solute upon pH change. However, unlike traditional pH-triggered controlled release systems, IDZ-liposomes are capable of undergoing multiple triggering events, resulting in solute release in a pulsatile manner. Furthermore near-total release can be achieved incrementally with temporal control. A mechanism for this reversible behaviour is also proposed, and dependence on the physical properties of the surfactant is discussed.

1. Introduction

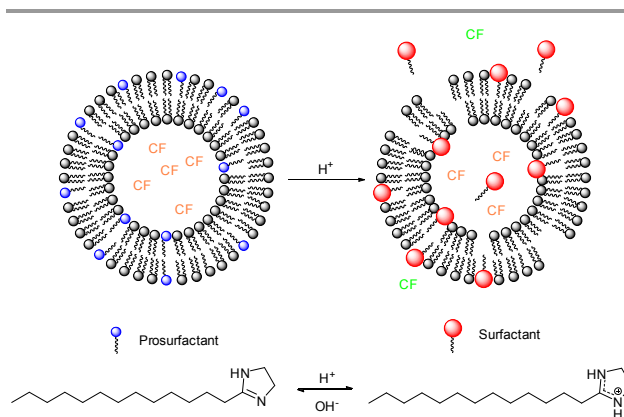
Liposomes have been widely investigated for the applications in systemic¹ and topical drug² delivery due to their ability to impart control over the release profile of a therapeutic agent. Tremendous progress has been made in the development of liposome-based systems since their discovery, for their ability to encapsulate a therapeutic agent and target it to specific tissue types; however content release is generally achieved via passive means.³ Thus there is a growing demand for the design of liposome-based systems which incorporate mechanisms to facilitate on-demand release of the encapsulated contents.

Stimuli-responsive liposomes are one potential avenue in meeting this challenge.⁴ These systems possess a switchable moiety, for example, an additional constituent or structural modification to the lipid comprising the membrane, which can be activated via the application of a trigger. Numerous stimuli have been investigated to activate liposome systems, generally utilizing a change in conditions, such as pH,⁴ temperature,⁵ or CO₂ levels,⁶ or an external input of energy, for example light,⁷ oscillating magnetic fields,⁸ or ultrasound^{9,10}. These systems generally release their contents entirely, either all at once or in a slow continuous fashion. More sophisticated stimuli-responsive liposomes are those that have been adapted to exhibit a pulsatile release profile, through repeated triggering of a reversible switch, providing enhanced temporal control over drug release. Examples include the addition of near infrared absorbing gold nanoparticles to the liposome surface^{11,12} and the addition of a UV activated azobenzene moiety to cholesterol¹³.

One method to increase the permeability and, thereby induce the release of an encapsulated solute, is through the

use of surfactants.¹⁴⁻¹⁶ A recent extension of this strategy for achieving surfactant mediated controlled release from liposomes is through the incorporation of a switchable surfactant.⁶ A switchable surfactant (SS) can be reversibly converted, typically via a chemical or photochemical trigger, into a “prosurfactant” (PS), whose amphiphilicity is greatly diminished, with an associated increase in lipophilicity. This enhanced lipophilic nature allows the prosurfactant to reside benignly within the liposome membrane, until it is “switched” to the surfactant form, via the application of a trigger, inducing solute release, for example the self-quenching fluorescent dye carboxyfluorescein (CF) as shown in Figure 1.

A number of systems have previously been described incorporating SSs with weakly basic,^{6,17} acidic,¹⁸ or cleavable head groups,¹⁹ where activation of these systems is primarily achieved through a change in pH (Figure 1). Typically, liposomes incorporating SS with ‘classical’ architecture of alkyl tail and charged head group release a fraction of the total encapsulated solute upon application of the trigger. Incorporation of classical PS in greater quantities to induce additional release is unfeasible as PS incorporation above 20 mol% generally inhibits



^a Department of Chemistry, University of Otago, Dunedin. Tel: +64 3 4797926, E-mail: ewtan@chemistry.otago.ac.nz.

Figure 1. Schematic of PS (blue) to surfactant (red) induced CF release from DPPC (grey) liposomes triggered via the external addition of H⁺. pH triggered interconversion of the SS C₁₃IDZ between the PS and surfactant forms.

stable liposome formation.^{20,21} Alternatively, it has been suggested by Asokan (2005)¹⁷ that an increase in the head group polarity of SS results in greater membrane permeabilisation. However, despite the use of the very hydrophilic N,N-dimethylacetamide (DMA) head group by Hegh et al. (2014)⁶ which resulted in greater solute release, a single triggering event was found to release up to 40%, after which the system was found to be irreversible.

To this end, it is reported herein the development, characterization, and behaviour of a stimuli-responsive liposome system incorporating a SS which contains IDZ as the head group (Figure 1). The system can be triggered repeatedly to release CF in a pulsatile manner providing increased release with greater temporal control, which may be useful for pH-triggered topical delivery of bioactives to the skin.

2. Results and discussion

2.1 Characterisation of the system

Surfactant properties

The pK_a and critical micelle temperature (CMT) of C₁₃IDZ.HCl was determined to be 8.3 and 3.2 ± 0.8 °C, respectively (Supplementary 2.5). To ensure that the IDZ was primarily in the surfactant form, CMC determination of C₁₃IDZ.HCl was undertaken at room temperature at a pH of 6.5. The critical micelle concentration (CMC) was measured to be 5.5 mmol L⁻¹ which is expected for similar cationic surfactants of this tail length (Supplementary 2.1).²² The partition coefficients of both PS and SS form of C₁₃IDZ were calculated to be approximately 5 and 2, respectively (Supplementary 2.2). The former value implies that greater than 99% of C₁₃IDZ should reside in the non-polar phase, and thus can be stably incorporated into the lipid membrane of liposome, and the latter value that majority of C₁₃IDZ.HCl should also remain in the membrane and available to be repeatedly activated.

Characterisation and permeability of PS incorporated liposomes

The inclusion of 0, 5, 10 and 20 mol% C₁₃IDZ (relative to 1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine (DPPC)) into the liposome formulation at pH 12.4 resulted in the formation of stable monodisperse liposomes between 100 and 200 nm that retained CF at 25 °C (Supplementary 2.4). However, the incorporation of C₁₃IDZ in amounts greater than 20 mol% showed substantial passive leakage of CF, which is consistent with previous observations when incorporating PSs with morpholine²¹, imidazolyl propionate²³ or DMA⁶ head groups, or with single tailed lipids such as lysolecithin²⁴. This suggests that while the PSs do not retain the surface activity of their surfactant forms, in adequate quantities they nonetheless compromise the bilayer integrity of DPPC liposomes.

To ascertain the effect of PS incorporation on bilayer permeability CF encapsulated 0-20 mol% C₁₃IDZ-liposomes were heated at pH 12.4 and the CF leakage was estimated by

measuring the change in fluorescence. All liposome formulations exhibited approximately the same characteristic temperature until a temperature was reached at which all the

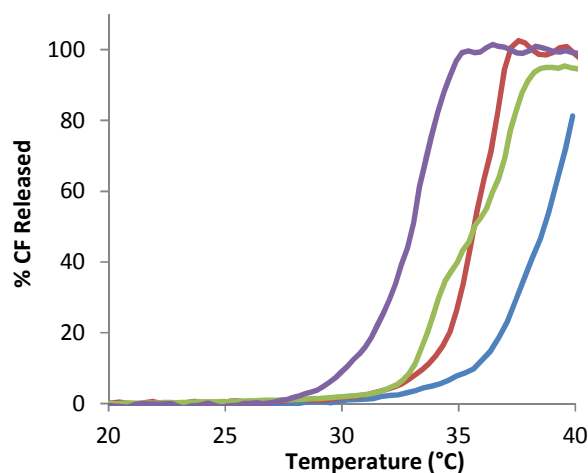


Figure 2. Percentage CF release from 0% (blue), 5% (red), 10% (green) and 20% (purple) C₁₃IDZ incorporated liposomes with increasing temperature (0.7 °Cmin⁻¹).

profile, with leakage slowly increasing with increasing remaining encapsulated CF rapidly leaked (Figure 2). Where these profiles departed from one another was in the temperature at which the leakage occurred, with increasing C₁₃IDZ inducing in a translocation of the leakage profiles to increasingly lower temperatures resulting in significant CF release (> 10%) occurring earlier (Figure 2). This suggests that the incorporation of increasing amounts of C₁₃IDZ results in an increase in the permeability of the bilayer and accounts for the inability to form stable liposomes above 20 mol% C₁₃IDZ. However, despite the increased permeability of liposomes containing ≤ 20 mol% C₁₃IDZ, leakage at 25 °C was negligible enabling additional experimentation to proceed at this temperature (Figure 2).

The apparent pK_a (pK_{a(app)}) of 20 mol% C₁₃IDZ-liposomes was determined to be 6.7, which is significantly less than the pK_a of 8.3 measured for the free surfactant in solution (Supplementary 2.5). This is consistent with the observation that lipophilic forms of pH sensitive amphiphiles are stabilised within a bilayer with a subsequent reduction in pK_a on the order of between 1 to 1.5 units.²⁵ The pK_{a(app)} also gives an estimation of parameters required for incorporation of C₁₃DMA (pH >8.7) and activation which is restricted by CF fluorescence to no lower than pH 5.5.

The concentration of surfactant required to saturate a 1 mmol L⁻¹ liposome suspension was found from OD measurements to be 0.31 mmol L⁻¹ (Supplementary 2.3). At this saturation concentration the membrane water partition coefficient ($P_{mem/w}^{sat}$) of C₁₃IDZ.HCl was calculated to be 1.6. This suggests that provided the surfactant concentration remains below saturation (~ 30% of the DPPC concentration) the majority of the surfactant should remain in the bilayer after PS activation.¹⁶

The effect of triggering, by decreasing the pH to 6.0, on the stability of 20% C₁₃IDZ-liposomes was investigated using DLS and passive leakage of CF with increasing temperature. By DLS, no significant change in the size of C₁₃IDZ-liposomes was observed after activation (Supplementary 2.4). Furthermore, the PDI remained below 0.20 suggesting that the liposomes remained monodisperse and that surfactant induced fusion, which has been observed in other surfactant-liposome suspensions²⁶⁻²⁸, was not significant in this system.

Finally, these liposome suspensions appear unremarkable optically and by DLS with an absence of smaller micelles (<10 nm). The lack of any significant changes in the suspension is thought to be because at a maximum of 20 μmol L⁻¹ the C₁₃IDZ.HCl concentration was not high enough to induce liposome solubilisation, which is typically observed in the mmol L⁻¹ concentration range for surfactants.^{27, 29-32}

2.2 Triggered CF release

The effect on CF release from liposomes incorporating varying amounts of C₁₃IDZ (0, 5, 10 and 20 mol%) upon a decrease of pH, from pH 12.4 to 6.0 (significantly below the pK_a) was investigated and compared to that of a previously published SS C₁₄DMA.⁶ The rate of release was monitored over 250 seconds as no appreciable increase in the percentage release was observed after this interval (Supplementary 3.3). For those liposomes incorporating C₁₃IDZ, the majority of release was observed to have occurred over the first 75 seconds, after which the rate of release decays significantly until a constant rate is attained (Figure 3(A)). In contrast, no significant CF release was observed from the control liposomes (Figure 3(A)).

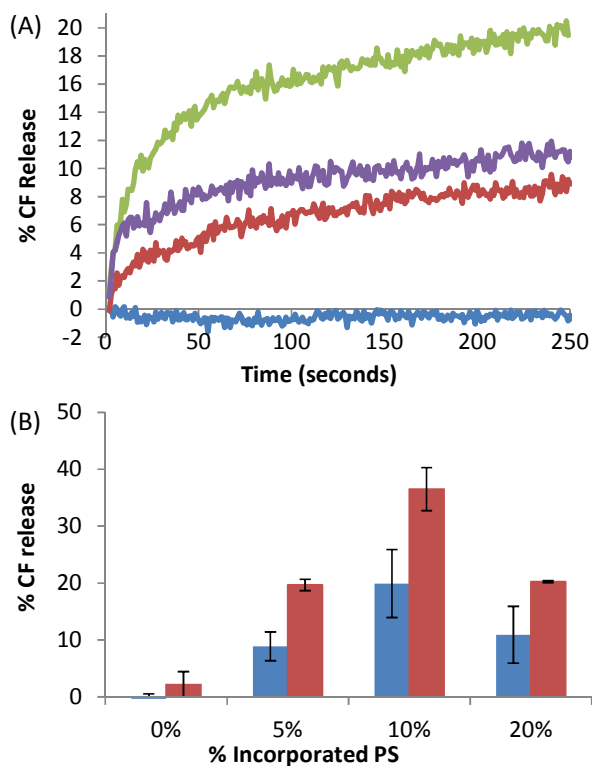


Figure 3. (A) Percentage CF release over 250 seconds from liposomes incorporated with 0% (blue), 5% (red), 10% (green) and 20% (purple) C₁₃IDZ induced by the addition of HCl to change the pH of the suspension from 12.4 to 6.0. (B) Total % CF release after 250 seconds from 0, 5, 10 and 20% C₁₃IDZ- (blue) liposomes induced by switching the pH from 12.4 to 6.0, which is compared to previously investigated PS C₁₄DMA⁶ (red) which was switched from pH 12.4 to 7.4. Error Bars = ± SEM (n = 3).

The total amount of CF released was proportional up to 10 mol% C₁₃IDZ, which exhibited a maximum release of approximately 20%, with an increase of C₁₃IDZ to 20 mol% CF release declined to 11% (Figure 3(B)). The CF release trend with amount of C₁₃IDZ incorporated is consistent with that observed previously for C₁₄DMA-liposomes, although the amount of CF release at comparable PS concentrations is typically 10% lower (Figure 3(B)).⁶

In order to optimise the magnitude of CF release the final pH of the system was investigated. From an initial pH of 12.4 the target pH was adjusted to give varying final pH values and the total percentage of CF released from 10% C₁₃IDZ-liposomes was measured after 250 seconds. Significant CF release was observed to occur at a final pH as high as 8.5, which increased with decreasing final pH to reach a maximum of 24% at pH 7.0 (Figure 4). Continued decreases in final pH, to as low as 5.5, did not liberate additional CF (Figure 4).

The concentration of C₁₃IDZ (in the μmol L⁻¹ range) is well below the CMC of C₁₃IDZ.HCl (5.5 mmol L⁻¹) and the critical concentration at which liposome solubilisation is observed (Supplementary 2.1), which is confirmed by the absence of micelles and presence of liposomes in DLS measurements of the system after switching. The mechanism of triggered solute release is therefore believed to occur not by solubilisation, but rather due to a rearrangement of the surfactant in the

membrane after activation of C_{13} IDZ, where the surfactant either remains incorporated within the lipid bilayer, ejected monomerically into the aqueous phase, or a combination of these two effects leading to CF release.

2.3 Multiple triggering

CF release

Further investigations into the mechanism of CF triggered release were undertaken by determining whether C_{13} IDZ-liposomes could exhibit pulsatile release, where repeated triggering induces multiple CF release events by repeatedly 'switching' the suspension back and forth between pH 9.0 and

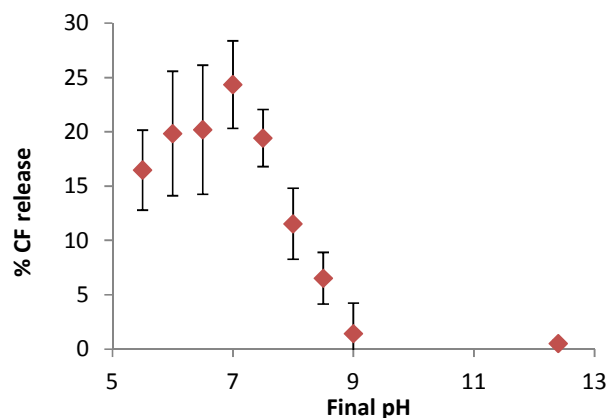


Figure 4. CF release from 10 mol% C_{13} IDZ incorporated liposomes with variation in target pH from an initial pH of 12.4. Error Bars = \pm SEM ($n = 3$). (Note: The results represent an average from three independently prepared batches of liposomes. The large uncertainty below pH 7 is associated with significant pH-dependent decreases in fluorescence intensity.)

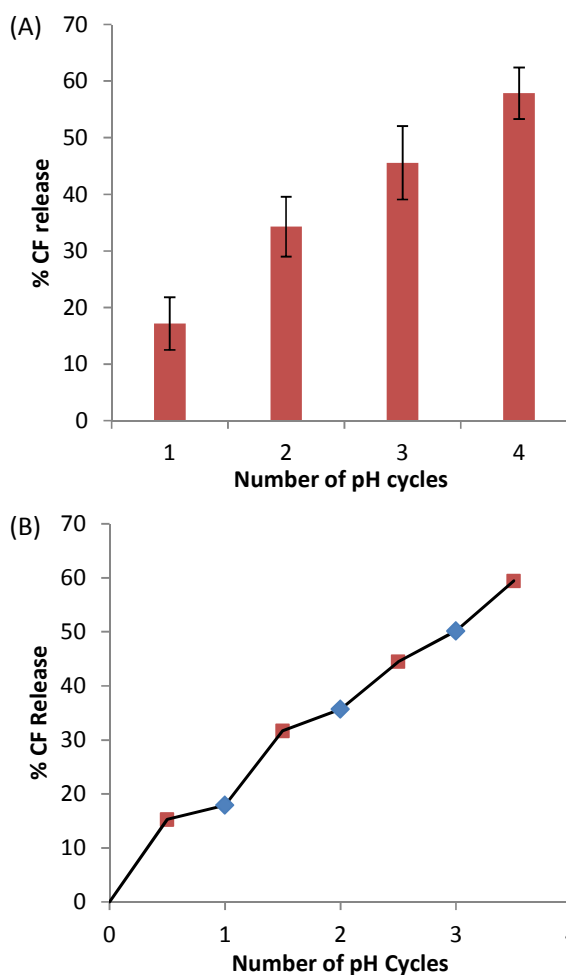


Figure 5. (A) The reversible *in-situ* switching of 10% C_{13} IDZ liposomes from pH 9.0 to 6.5 (B) Reversible switching of a single 10% C_{13} IDZ liposome system from pH 9.0 to 6.5 with CF release measured during both acidification (red) and basification (blue). Error Bars = \pm SEM ($n = 3$).

6.5. These values were chosen as according to results presented in Figure 4 this encompassed the entirety of the CF release, and subsequently was found to release an equivalent amount of CF compared to when the pH was decreased from 12.4 (Figure 5(A)). Over multiple switching cycles, similar amounts of CF release were observed, demonstrating that the system can undergo pulsatile release (Figure 5), which is in contrast to liposomes incorporating the structurally similar C_{14} DMA, which were previously demonstrated by the authors⁶ to not exhibit multiple switching behaviour. Additionally, significant quantities of CF was also observed to release after both acidification and basification (Figure 5(B)).

Zeta-potential measurements of SS incorporated liposomes

The zeta potential of control, 20% C_{13} IDZ- and C_{14} DMA-liposomes at various pH values were determined in an attempt to elucidate the mechanism behind reversible PS mediated solute release. The difference in zeta potential of PS-liposomes can be attributed to positively charged surfactant residing in

the bilayer arising from protonation of the PS. At pH 12.5, the initial zeta potential of C₁₄DMA- and C₁₃IDZ-liposomes were +3 mV and -7 mV, respectively compared to -9 mV for control liposomes (Figure 6). With decreasing pH, the zeta potential of both C₁₄DMA and C₁₃IDZ systems increased from +3 to +21 mV (pH 12.5-7.5) and -7 to +26 mV (pH 12.5-5.5) respectively, compared with -8 to +1.5 mV (pH 12.5-5.5) for the control system (Figure 6). Upon basification back to pH 12.5 the zeta potential of both systems decreased, however C₁₃IDZ returned to approximately its starting value of -8 mV, whereas C₁₄DMA reached a value of -9 mV, much lower than the initial value of +3 mV (Figure 6). This suggests that the C₁₃IDZ in the system returns to its initial state whereas the C₁₄DMA system does not.

Unlike the control liposomes, the PS systems also exhibited biphasic behaviour with a period of constant substantial change

intervals of 0.5 pH units from pH 12.5 down to pH 7.5 or 5.5 (blue) then, with the same suspension, back to pH 12.5 (red). The pK_{a(app)} of the PSs are also indicated.

in zeta potential at higher pH, followed by a period of reduced zeta potential change at lower pH (Figure 6). The biphasic behaviour occurs in both directions for C₁₃IDZ whereas it is only observed upon acidification for C₁₄DMA. Furthermore, the intercept occurs between pH 10.0 and 10.5 in both systems and is therefore not obviously correlated to the pK_{a(app)} of either system.

Despite their structural similarity, C₁₃IDZ.HCl is less hydrophilic ($P_{mem/w}^{sat}$ 1.0 vs 1.6⁶) than C₁₄DMA.HCl. It is thought that the reversible nature of the C₁₃IDZ trigger arises from the reduced hydrophilicity resulting in the majority of the activated SS remaining in the bilayer. This is supported by the observation that the zeta potential measurements of C₁₃IDZ- and C₁₄DMA-liposomes where the former exhibits a biphasic profile at similar zeta values during both basification and acidification whereas the latter exhibits a monophasic profile at significantly lower zeta values during basification. Lower zeta potential values during basification suggests ejection of charged C₁₄DMA.HCl from the bilayer with the system not returning to equilibrium which is consistent with the previous observation that C₁₄DMA-liposomes are limited to a single release event with repeated switching⁶ (Figure 6).

The loss of biphasic character of the C₁₄DMA zeta profile during basification suggests that it is a SS mediated process. However, a biphasic profile is thought not an indication of the initiation SS protonation or deprotonation, as the inflection points of both C₁₃IDZ and C₁₄DMA zeta profiles occur at a similar pH (10.5-9.5) and are not correlated with their respective pK_as (Figure 6). Instead, the similarity in the inflection points is thought to be the result of a sudden change in lipid packing with pH³³ which, in this case, relies on the presence of a single tailed amphiphile in the membrane as the control liposomes do not exhibit biphasic zeta potential behaviour (Figure 6A).

The location of an inflection point in the zeta potential profile also occurs at a similar pH (9.5-10.5) to the pH at which CF release is first detected (9.0), which is observed for both C₁₃IDZ- and C₁₄DMA-liposomes despite their dissimilar pK_a values (Figure 6). Furthermore, increasing zeta potential values at higher pH values (> 10.5) strongly implies PS protonation which occurs in the absence of CF leakage (Figure 4 and

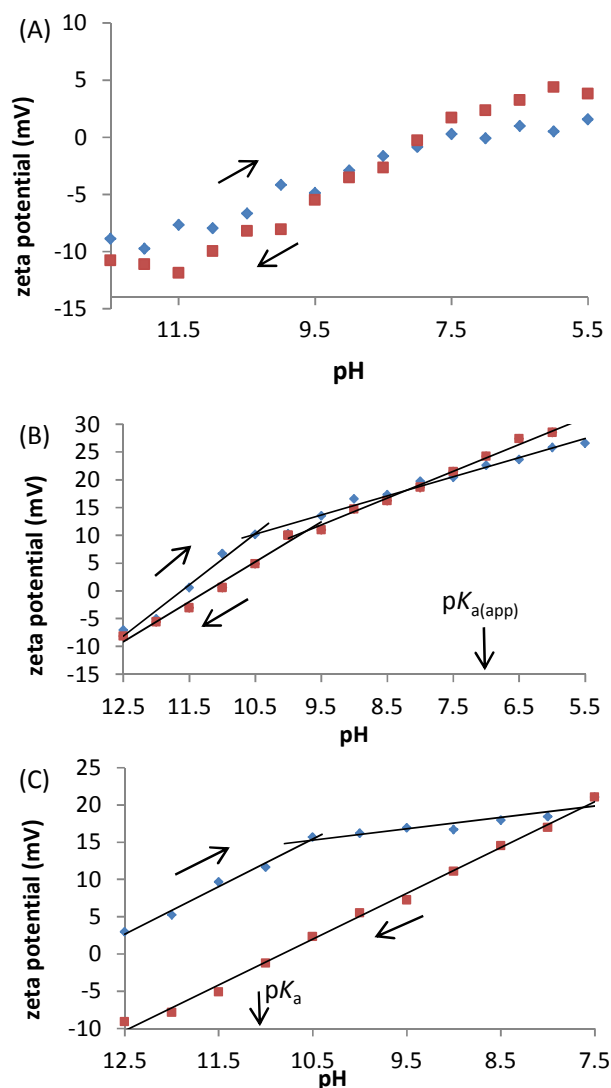


Figure 6. The zeta potential measurements of control (A), 10% C₁₃IDZ- (B) and C₁₄DMA- (C) liposome suspensions (1 mmol L⁻¹). Measurements were recorded at

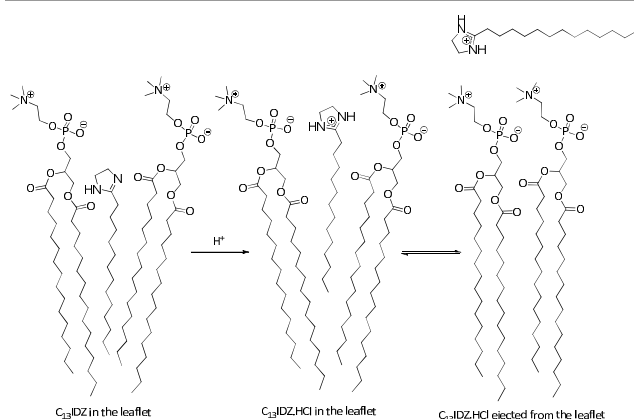


Figure 7. Protonation of C₁₃IDZ PS resulting in shuttling and ejection of the surfactant from the bilayer.

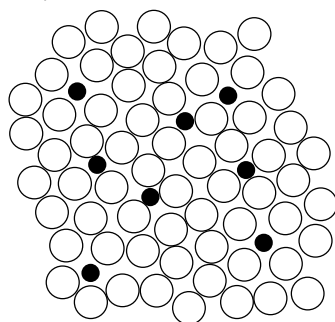


Figure 8. Top down schematic of a DPPC (open circles) membrane incorporating 10% PS (closed circles) showing the relative distribution of the molecules.

Figure 6). This suggests that the bilayer arrangement at lower pH values is required for SS triggered CF release.

The C₁₃IDZ-liposomes were observed to release a reduced amount of CF compared to the C₁₄DMA liposomes, however the relationship between the amount of CF released and the concentration of PS incorporated shows a similar profile (Figure 3). The decline in CF release from 10 to 20% PS in both systems suggests that the SS has a stabilising effect on the bilayer similar to that of cholesterol, whose incorporation in specific quantities is found to reduce bilayer permeability by inducing conformational ordering of lipid chains preserving lateral diffusion, increasing bilayer thickness and mechanical strength, resulting in reduced solute leakage.^{34, 35}

From 10% - 20% PS the additional C₁₃IDZ would possibly have a reduced ability to increase bilayer permeability. It is thought that on a molecular level, PS activation should induce rearrangement in the neighbouring lipid molecules, via a combination of an increase in head group area, shuttling along and ejection from the bilayer (Figure 7). Given ideal mixing, at 10 mol% C₁₃IDZ each PS is on average surrounded by as few as 10 DPPC molecules and thus only 1.5 head group radii (Figure 8). Due to the close proximity of the PS molecules to one another in the bilayer, at higher concentrations, the influence of its activation on the surrounding lipid would be expected to diminish as the lipid molecules have already been perturbed or rearranged to the greatest extent by an adjacent surfactant.

3. Conclusion

Liposomes incorporating the PS C₁₃IDZ were formulated which can be activated by pH change to the surfactant form and, in the process, disrupt the bilayer, triggering release of encapsulated CF. Unlike SS-liposome systems previously investigated, these liposomes exhibited pulsatile release behaviour of the contents, providing temporal control with the added benefit of greater release. The behaviour of which was attributed to the hydrophobicity of C₁₃IDZ.HCl, preventing ejection from the bilayer and resulting in a reversible system.

4. Experimental section

Materials

All chemical reagents for synthesis were of analytical grade, obtained Sigma-Aldrich (St. Louis, MO, USA) and used without further purification. 1,2-Dipalmitoyl-*sn*-glycero-3-phosphocholine (DPPC) was purchased from Avanti Polar Lipids (Alabaster, AL, USA). 5,6-Carboxyfluorescein was obtained from Molekula (Dorset, UK). C₁₄DMA and C₁₄DMA.HCl was available from a previous investigation.⁶

SS synthesis

C₁₃IDZ was synthesised by a method adapted from Mannheimer (1968)³⁶ and Shi and Gu (1997)³⁷ where tetradecanoic acid is refluxed with a slight excess ethylene diamine in toluene until no water was produced (24-36 h) (Supplementary 1.1). To form C₁₃IDZ.HCl, HCl(g) was bubbled through a hot ethereal solution of C₁₃IDZ until a precipitate formed.

Surfactant Characterisation

The critical micelle concentration (CMC) of C₁₃IDZ.HCl was determined using conductivity as described previously.⁶ The critical micelle temperature (CMT) of C₁₃IDZ.HCl was determined using the visual observation method described by Weers (1991)³⁸.

The pK_a of C₁₃IDZ.HCl was experimentally determined via pH titration using a Mettler Toledo MP220 pH Meter (referenced to pH 4.0 and 9.2). A stock solution (4 mL) of C₁₃IDZ.HCl was prepared at a concentration of 1 mmol L⁻¹ (below the CMC) using Milli-Q™ water and thermostated at 25 °C. The initial pH of the resulting solution was recorded and sequential additions of 10 μL of a 0.01 mol L⁻¹ NaOH solution were made, recording the change in pH upon each addition. The pK_a was determined from the derivative of the titration curve (graph of dpH against volume of NaOH) at half the equivalence point.

Liposome preparation

Liposomes consisting of 6 mmol L⁻¹ DPPC, 5 mol% cholesterol, and incorporating varying amounts (0, 5, 10 and 20 mol%) of

C₁₃IDZ, were prepared. Lipids, in chloroform, were mixed in a round bottom flask and the solvent was removed *in vacuo* over one hour to produce a dry thin lipid film. The lipid film was rehydrated with 1 mL phosphate buffer (20 mmol L⁻¹, pH 12.4) containing 100 mmol L⁻¹ CF. To ensure complete rehydration, whilst avoiding saponification, solutions were heated for short periods (<1 min) to between 45 and 50 °C, above the gel to liquid-crystal phase transition of DPPC (41 °C)³⁹ and vortexed until absence of the lipid film on the glass was observed. Unilamellar liposomes of high homogeneity were formed by extruding the solution 11 times through an Avanti® Mini-Extruder (Avanti Polar Lipids Alabaster, AL, USA), heated to 50 °C, fitted with a 1000 nm polycarbonate membrane, and a further 11 times through a 200 nm membrane.

The untrapped CF was separated from the liposomes using a Sephadex® G-100 column and phosphate buffer (20 mmol L⁻¹, pH 12.4) as the eluant. The liposome fraction was collected and the DPPC concentration was determined using the Stewart Assay⁴⁰ whereupon the concentration was adjusted to 1 mmol L⁻¹ with phosphate buffer pH 12.4. To ensure consistent results the final suspension was then subjected to any further measurements within four hours of being collected.

Partitioning of surfactant into liposomes

The membrane water partition coefficient ($P_{mem/w}$) of C₁₃IDZ.HCl was calculated by measuring the optical density (OD) of a DPPC liposome suspension during surfactant mediated solubilisation using the method outlined by de la Maza and Parra (1995)⁴¹ (Supplementary 2.3). OD measurements were taken at 350 nm in duplicate on an Asus UVM3HO plate reader in Immuno 96 Microwell™ Plates (Nunc) over a range of DPPC concentrations (0.5, 1.0, 1.5, 2.0 mmol L⁻¹) to which increasing amounts of C₁₃IDZ.HCl were added, to give a total volume of 200 μL (Supplementary 2.3). The concentration of surfactant was increased until a decrease in OD drop followed by stabilisation was observed. Before measurement, the mixtures were equilibrated for 1 hr at room temperature.

CF release assay

For each measurement, 100 μL of the CF loaded liposome suspension (pH 12.4) was added to a quartz fluorescence cuvette filled with 2.9 mL of phosphate buffer (20 mmol L⁻¹) at the desired pH (9.0, 8.5, 8.0, 7.5, 7.0, 6.5, 6.0, 5.5). The resultant suspension was thoroughly mixed for 1 second by aspirating and expelling with a pipette. The sample temperature within the fluorimeter was maintained at the desired temperature using a Tecron Bio-100 adjustable thermostat controlled heating unit. The fluorescence intensity of the CF (Ex. 465 nm, Em. 520 nm) was then recorded in a on a LS50B luminescence fluorimeter (Perkin-Elmer) at a scattering angle of 90°. To determine the initial fluorescence, the fluorescence intensity over the 10 seconds directly after addition and mixing of the liposome suspension from this data the fluorescence at $T = 0$ was determined. After 250 seconds

100 μL of a solution of 10% T-X 100 was added to the liposome suspension to lyse the liposomes, releasing all remaining encapsulated CF. Fluorescence intensity was again measured after lysis, and corrected for dilution (Supplementary 3.2). CF fluorescence intensity is both pH dependent and not linear with concentration, thus the raw fluorescence values were converted to CF concentration using the pH appropriate standard curve (Supplementary 2.3). The % CF release was calculated by the following equation;

$$\% CF \text{ release} = \frac{(CF_t - CF_0)}{(CF_{TX-100} - CF_0)} \times 100 \quad \text{Eq. 1}$$

where, CF_0 and CF_t are the CF concentrations initially and at time t respectively, whereas CF_{TX-100} is the average fluorescence over 20 seconds after the addition of TX-100. All samples were measured in triplicate and the standard error was calculated at final release which was recorded as an average of the last five measurements in order to increase the signal to noise ratio of the data.

To measure CF release arising from consecutive switching, the initial switch from pH 12.4 to 6.5 was undertaken as described as above. To switch from pH 6.5 back to 12.4, a volume (~75 μL) of 1 M NaOH was added until the desired pH was reached as measured by a MP220 pH meter (Mettler Toledo) and the resultant fluorescence was measured over 250 seconds. To switch back to pH 7.4, a volume of 1 M HCl was added (~50 μL) and fluorescence was measured over 250 seconds. After switching, TX 100 was added to quantify maximum fluorescence. The % CF released was determined as described above, with dilution being taken into account.

SS-Liposome Characterisation measurement

The size and polydispersity of the C₁₃IDZ-liposomes at pH 12.4 and 6.0 were determined using dynamic light scattering (DLS) as described previously.⁶

The apparent pKa ($pK_{a(app)}$) of 20 mol% C₁₃IDZ-liposome suspensions were determined using the pH titration method outlined above with the following modifications. A 5 mmol L⁻¹ liposome suspension was prepared to give a PS concentration of 1 mmol L⁻¹ in a 4 mL volume of unbuffered Milli-Q™ H₂O (pH 12.4). Sequential additions of 10 μL of 0.01 mol L⁻¹ HCl were made, with the subsequent pH recorded upon each addition. The integrity of the liposome bilayer was determined by measuring the fluorescence arising from passive leakage of encapsulated CF. The procedure was similar to that described for the CF release assay (Section 2.7) with the following departures. The fluorescence of CF encapsulated C₁₃IDZ-liposome suspensions (33 μmol L⁻¹, 3 mL, pH 12.4) was measured over temperatures ranging from 20 °C to 40 °C, at a rate of 0.7 °C min⁻¹. From the fluorescence measurements % CF released was calculated using Eq. 1.

Surface charge measurements

The zeta potential of control (0 mol% C₁₃IDZ) and C₁₃IDZ-liposomes were measured on a Malvern Zetasizer Nano ZS

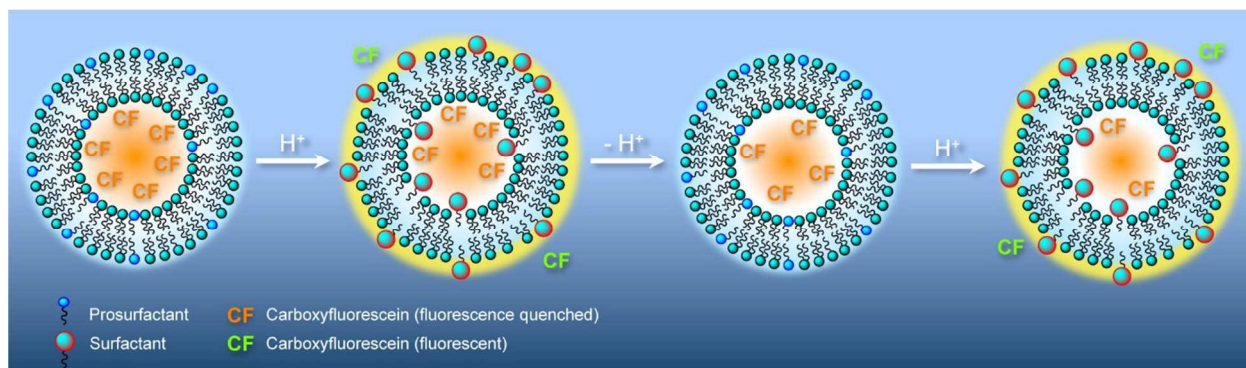
(Malvern Instruments, Worcestershire, U.K.). Measurements were performed in triplicate in a disposable capillary cell (DTS1070, Malvern Instruments, Worcestershire, U.K.) at 25 °C. Control and 20% C₁₃IDZ-liposome suspensions were prepared as previously described, without encapsulated CF and standardised to a DPPC concentration of 1 mmol L⁻¹ before measurement.

Zeta potential measurements were performed. A 30 mL standard solution of 1 mmol L⁻¹ of the required liposome solution at pH 12.5 was made. Using a small volume of concentrated NaOH (1 mol L⁻¹) the pH of this suspension was decreased in increments of 0.5 pH units down to 5.5 then using small volumes of concentrated HCl (1 mol L⁻¹) back to pH 12.5. At each 0.5 pH increment a 1 mL sample was taken and the zeta potential was measured.

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