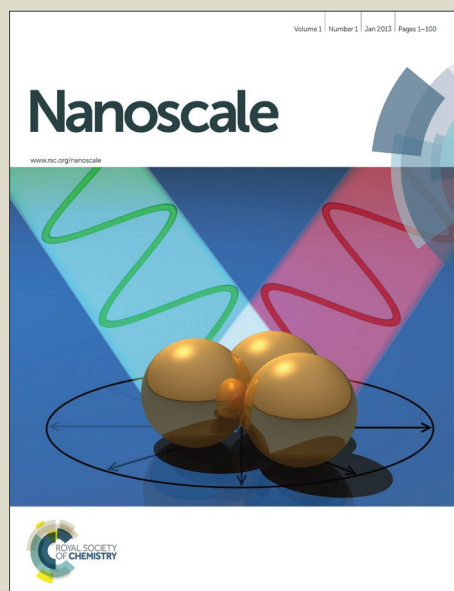


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1 **SERS-active Sensor Based on Heterogeneous Gold Nanostar Core-**
2 **Silver Nanoparticle Satellite Assemblies for Ultrasensitive Detection**
3 **of AflatoxinB1**

4 Aike Li^{1*}, Lijuan Tang², Dan Song¹, Shanshan Song², Wei Ma², Liguang Xu², Hua
5 Kuang², Xiaoling Wu², Liqiang Liu², Xin Chen³, Chuanlai Xu^{2*}

6
7 A Surface-enhanced Raman scattering (SERS) sensor based on gold nanostar (Au NS)
8 core-silver nanoparticle (Ag NP) satellites was fabricated for the first time to detect
9 aflatoxinB1 (AFB1). We constructed the SERS sensor using AFB1 aptamer
10 (DNA1)-modified Ag satellites and a complementary sequence (DNA2)-modified Au
11 NS core. The Raman label (ATP) was modified onto the surface of Ag satellites. The
12 SERS signal was enhanced when the satellite NP was attached to the Au core NS. The
13 AFB1 aptamer on the surface of Ag satellites would bind to targets when AFB1 was
14 present in the system, Ag satellites were then removed and the SERS signal decreased.
15 This SERS sensor showed superior specificity for AFB1 and the detection linear range
16 was from 1 to 1,000 pg/mL with the limit of detection (LOD) of 0.48 pg/mL. The
17 excellent recovery experiment using peanut milk demonstrated that the sensor could
18 be applied in food and environmental detection.

19

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1. Introduction

Aflatoxins (AF), which are secondary fungal metabolites, are a group of mycotoxins produced by *Aspergillus flavus* and *A. parasiticus*.¹ Aflatoxins are toxic and generally recognized as potent carcinogens, mutagens and teratogens.² The aflatoxins mainly include B1, B2, G1, and G2.³ Of these, AFB1 has received considerable attention and is listed as a group I carcinogen by the International Agency for Research on Cancer (IARC) due to its abundance and high toxicity.⁴ AFB1 is a widely-found residue in raw food materials, such as grains, corn, feedstuffs and peanuts. Exposure to a low level of AFB1 can cause immunosuppression and immunotoxicity, cancer, and even death. Thus, taking into account the risks of AFB1, regulatory limits for the level of this toxin have been established in many countries. In the United States and China, the current maximum allowed levels of AFB1 in foodstuffs is 20 ng/g.⁵ Therefore, there is an urgent need to develop more robust methods for monitoring AFB1 levels in food to ensure food safety.

Conventional methods for AFB1 quantification in foodstuff samples have been developed and include liquid chromatography–tandem mass spectrometry (LC–MS/MS),⁶ high-performance liquid chromatography–tandem mass spectrometry (HPLC–MS/MS)⁷ and enzyme-linked immunosorbent assay (ELISA).⁸ However, these methods are mainly based on instruments, are time-consuming, expensive, and require professional expertise. These drawbacks have limited their wide application. To solve these problems, various novel biosensors have also been developed.^{9–12} For example, a colorimetric immunoassay, ultrasensitive aptasensor, photoelectrochemical immunosensor, and fluorescence sensor. Although these biosensor techniques, to some extent, have solved these problems, low sensitivity and stringent sample pretreatments need to be improved.

1 In recent years, Surface-enhanced Raman scattering (SERS) encoded noble metal
2 nanomaterials have increased in popularity as important tools in biosensors.^{13,14}
3 However, the SERS signal of an encoded single nanomaterial was too weak for wide
4 application in ultrasensitive detection. Although aggregations can largely improve the
5 SERS signal, uncontrollable aggregation results in signal heterogeneity with low
6 reproducibility of the SERS assay.¹⁵ Therefore, controllable assemblies are critical in
7 the fabrication of SERS-active sensors.¹⁶ Many controllable assemblies have been
8 fabricated, such as dimers,¹⁷ trimers,¹⁸ pyramids,¹⁹ and chains.²⁰ Recently, core–
9 satellite assemblies have gained increasing attention.^{14,21} Our group has also
10 fabricated a variety of core-satellite assemblies used in ultrasensitive detection. For
11 example, Au-Ag NP core–satellite assemblies,²² Au-UC NP core–satellite
12 assemblies,²³ and Au NR-Au NP core–satellite assemblies.²⁴ However, to date, there
13 are no reports on a SERS sensor using Au NS core–Ag NP satellite assemblies. Gold
14 nanostars (Au NS) are an outstanding SERS substrate due to their anisotropic shape
15 with the core and many tips.^{25,26} The structure couples the plasmons focused on the
16 multi-sharp tips which produce lots of “hotspots” that enhance the electromagnetic
17 field around the particle.²⁷ Importantly, silver nanoparticles (Ag NPs) are also a good
18 SERS candidate with significant SERS enhancement.²⁸ Therefore, in this study, we
19 fabricated a new SERS sensor of Au NS core–Ag NP satellite assemblies using AFB1
20 aptamer (DNA1)-modified Ag satellites and a complementary sequence
21 (DNA2)-modified Au NS core for AFB1 detection.³

22

23 **2. Material and methods**

24 **2.1. Materials**

25 Thiolated DNA aptamer was purchased from Shanghai Sangon Biological

1 Engineering Technology & Services Co. Ltd, China. The aptamer was purified by
2 HPLC and suspended in deionized water from a Milli-Q device (18.2 M Ω , Millipore,
3 Molsheim, France). Unless stated otherwise, all chemicals used in this study were
4 purchased from Sigma–Aldrich.

5 The detailed sequences of the oligonucleotides are as follows:

6 AFB1 aptamer (DNA1):

7 HS-5'-GTTGGGCACGTGTTGTCTCTCTGTGTCTCGTGCCCTTCGCTAGGCCAC -3'

8 Complementary sequence (DNA2):

9 HS-5'-CAGAGAGACAACACGTGCCCAAC -3'

10

11 **2.2. Instrumentation**

12 UV-Vis spectra were acquired using a UNICO 2100 PC UV-Vis spectrophotometer
13 and processed with Origin Lab software. Transmission electron microscopy (TEM)
14 images were obtained using a JEOL JEM-2100 operating at an acceleration voltage of
15 200 kV. Raman spectra were measured using a LabRam-HR800 Micro-Raman
16 spectrometer with Lab-spec 5.0 software attached to a liquid cell. The slit and pinhole
17 were set at 100 and 400 μ m, respectively, in the confocal configuration, with a
18 holographic grating (600 g/mm) and an air-cooled He-Ne laser giving 632.8 nm
19 excitation with a power of \sim 8 mW. The spectrum resolution of Raman was \leq 0.65
20 cm^{-1} and the diameter of the laser spot was 7720.16 μ m.

21

22 **2.3. Synthesis of gold nanostars**

23 Gold nanostars (Au NSs) were prepared from a seed-mediated growth method.²⁹
24 Initially, citrate solution (15 mL, 1%) was added into a boiling solution of HAuCl₄
25 (100 mL, 1 mM) under vigorous stirring at 380°C. After reacting for 10 min, the seeds

were cooled to room temperature and then stored at 4°C. After the seed preparation, the Au NSs were fabricated. Silver nitrate (AgNO_3) solution (200 μL , 3 mM), ascorbic acid solution (100 μL , 0.1 M), and 200 μL seeds were added to HAuCl_4 solution (20 mL, 0.25 mM) and quickly stirred for 1 min. The reaction was then terminated by centrifuging the colloidal solution at 3,000 rpm for 5 min. Finally, the precipitate was concentrated 10 times and then 0.01 M phosphate buffer (PBS) at a final concentration of 1 nM was added.

2.4. Synthesis of silver nanoparticles

Silver nanoparticles (Ag NPs) were prepared using a routine method with slight modifications. Initially, poly (N-vinyl-2-pyrrolidone) (PVP) (5 mL, 1%) and freshly prepared sodium borohydride (NaBH_4) (0.45 mL, 0.1 M) were mixed with 20 mL of distilled water. Next, AgNO_3 (5 mL, 10 mM) and PVP (5 mL, 1% by weight) were simultaneously injected into the mixture using a constant-flow pump at a rate of 30 mL/h under high-speed stirring. The reaction solution was kept at 80°C for 2 h to remove the unreacted NaBH_4 and then stored at 4°C.³⁰

2.5. Fabrication of DNA-modified Au NSs and Ag NPs

Ag NPs were modified with AFB1 aptamer (DNA1) at a molar ratio of 4:1. Ag NPs were resuspended in 0.01 M PBS buffer at a final concentration of 10 nM by centrifugation. DNA1 (4 μL , 1 μM) was added to the Ag NPs solution (100 μL , 10 nM) under vigorous stirring. The Raman Label (4-ATP) solution was then added at a final concentration of 10 μM . After incubation at 37°C for 8 h, the particles were centrifuged at 13,000 rpm for 15 min and washed three times with PBS buffer (100 μL , 0.01 M). The deposit was collected and resuspended in 100 μL PBS buffer.

Au NSs were modified with AFB1 complementary (DNA2) at a molar ratio of 200:1. Briefly, DNA2 (2 μ L, 10 μ M) was added into the as-prepared Au NSs solution (100 μ L, 1 nM). Then 5 M of NaCl solution was added at a final concentration of 500 mM under vigorous stirring. After reacting for 8 h, excess DNA was removed by centrifugation (three times) at 3,000 rpm for 5 min. The deposit was resuspended in 100 μ L PBS buffer.

2.6. Fabrication of the Au NS core–Ag NP satellite sensor

Initially, 100 μ L of prepared Ag NPs-DNA1 and 20 μ L of prepared Au NSs-DNA2 were mixed under gentle shaking. Next, eight AFB1 standards were added at final concentrations of 0, 1, 2, 10, 50, 100, 500, and 1000 pg/mL, respectively. The solutions were then incubated at 37°C for 2 h. The Ag NPs were removed by centrifugation (three times) at 2,000 rpm for 3 min. Finally, TEM, UV–Vis and SERS spectroscopy were used to monitor this process.

2.7. Analysis of specificity and selectivity

Seven biotoxins (AFG1 (aflatoxin G1), AFG2 (aflatoxin G2), AFB2 (aflatoxin B2), M1 (aflatoxin M1), DON (deoxynivalenol), OTA (ochratoxin A), and FB1 (fumonisin B1)) were used to determine the specificity of the as-fabricated sensor. The concentrations of other analytes (100 ng/mL) were 100 times greater than that of AFB1 (1 ng/mL). All other detection procedures were identical to those used for AFB1.

2.8. Recovery of AFB1 spiked in peanut milk samples

The peanut milk was purchased from a local market and was confirmed to be negative

1 for AFB1 by the China Entry-Exit Inspection and Quarantine Bureau (CIQ) in Jiangsu.
2 The peanut milk was centrifuged at 7,000 rpm for 10 min to remove fat from the
3 supernatant. The milk was then diluted fivefold in PBS buffer, and both Ag
4 NPs-DNA1 and Au NSs-DNA2 were added under vigorous stirring. The AFB1
5 standard sample at three concentrations (0.6, 3, and 6 pg/mL) were immediately
6 added.

7

8 **3. Results and discussion**

9 **3.1. Establishment of the heterogeneous core-satellites SERS-aptasensor for** 10 **AFB1 detection**

11 As illustrated in **Scheme 1**, firstly, DNA1-modified Ag NPs were mixed with
12 DNA2-modified Au NSs. Following incubation for 2 h at 37°C, in the absence of
13 AFB1, the Ag NP satellite assemblies with Au NS cores led to the formation of
14 core-satellites architecture with an intense SERS signal. However, in the presence of
15 AFB1 targets, the AFB1 aptamer (DNA1) prefers to switch its configuration to
16 combine with the AFB1 target, which led to the release of complementary
17 oligonucleotide (DNA2). Correspondingly, the core-satellites were disaggregated into
18 dispersed nanoparticles with low SERS signals.

19 The Ag NPs, with a diameter of 8 nm, were synthesized by a routine method. The
20 Au NSs, with a diameter of 45 nm, were prepared by a seed-mediated growth method.
21 As shown in **Fig. 1**, the synthesized Ag NPs and Au NSs showed unique morphology
22 and good dispersity. They also exhibited a localized surface plasmon resonance at 400
23 nm and 852 nm, respectively (**Fig. 2**). In order to fabricate the Au NS core-Ag NP
24 satellite heterogeneous structure, the Ag NP satellites and Au NS cores were modified
25 with DNA1 and DNA2 at a molar ratio of 4:1 and 200:1, respectively. According to

the fluorescence methods³¹, the number of DNA1 and DNA2 coupled on Au NPs and Au NSs was calculated to be 1.1 ± 0.4 and 23.6 ± 0.9 , respectively. Ag NP-DNA1 and Au NS-DNA2 were then mixed for 2 h. As shown in **Fig. 2**, the UV-Vis spectra of the core-satellites architecture compared with the spectra for Ag NPs and Au NSs showed no obvious change, which demonstrated that there were no large aggregations in this system. Furthermore, the assembled core satellites nanostructures were present in aqueous solution for 1 month in the absence of DNA cleaving enzymes. Therefore, the architecture can be applied as an aptasensor.

For AFB1 detection, eight AFB1 standards were added to the system containing Ag NP-DNA1-4-ATP and Au NS-DNA2 at final concentrations of 0, 1, 2, 10, 50, 100, 500, and 1,000 pg/mL, respectively. Following incubation for 2 h, the dissociated Ag NP-DNA1-4-ATP was removed by centrifugation (three times) at 2,000 rpm for 3 min. The representative TEM images of AFB1 at different concentrations are shown in **Fig. 3**. The number of Ag NPs surrounding Au NS in the presence of different concentration was statistically analyzed from more than 50 assemblies, respectively. As shown in **Fig. S1**, the number of Ag NP gradually was decreased with the concentration of target increased. In the sample without AFB1, the satellite Ag NPs completely surrounded the core Au NS, and had the highest assembly efficiency. However, there were no satellites around the core at the high concentration of AFB1 (1,000 pg/mL). The complexity and proportion of the assemblies in relation to the concentration of AFB1 were clearly seen. The UV-vis spectra at various concentrations of AFB1 are shown in **Fig. 4**. The absorbance spectra of these samples at 852 nm were not obviously different to each other. However, the absorbance spectra at 400 nm decreased when the concentration of AFB1 increased. This was because more Ag NPs could not assemble around the Au NSs when the concentration

1 of AFB1 increased, which led to dissociated Ag NPs being removed by centrifugation.
2 This was further evidence that the AFB1 target can successfully combine with its
3 aptamer to decrease the number of Ag NP satellites surrounding the core Au NSs. The
4 change in UV-Vis spectra was not sensitive enough for quantitative determination.
5 Therefore, the Raman spectrum was used to estimate the sensitivity of this sensor. The
6 SERS intensity of 4-ATP significantly declined with increased concentration of AFB1,
7 which corresponded to the decreased assembly of core satellites (**Fig. 5a**). Calibration
8 curves were then plotted with the Raman signal at $1,136\text{ cm}^{-1}$ as a logarithmic
9 function of AFB1 concentration. A standard curve was obtained with an excellent
10 correlation R^2 of 0.995 and a wide linear range from 1 to 1,000 pg/mL (**Fig. 5b**). The
11 limit of detection was determined by three times the standard deviation of the blank
12 solution and was calculated to be 0.48 pg/mL, which was lower than previously
13 reported sensors.^{2,3,5}

14

15 **3.2. Specificity and selectivity of the sensor for AFB1 detection**

16 To evaluate the selectivity and specificity of the as-fabricated sensor, seven biotoxins
17 (G1, G2, AFB2, M1, DON, OTA, FB) were used. As illustrated in **Fig. 6**, the SERS
18 signal was significantly decreased only when AFB1 was added. Although the
19 concentrations of the other targets (100 ng/mL) were 100 times higher than AFB1 (1
20 ng/mL), no change in the SERS signal was observed. These results demonstrated that
21 this method showed high selectivity for AFB1 detection.

22

23 **3.3. Recovery of AFB1 spiked in peanut milk samples**

24 To confirm the feasibility of this method, peanut milk samples were used as a
25 substrate to perform AFB1 recovery experiments. The peanut milk samples were

1 purchased from a local market and were confirmed to be negative for AFB1 by
2 Jiangsu Entry-Exit Inspection and Quarantine Bureau (CIQ) in. AFB1 standard
3 solutions were added to the samples at a final concentration of 0.6, 3, and 6 pg/mL,
4 and each spiked level was carried out for five parallel tests. As shown in **Table 1**, the
5 recoveries of AFB1 were 88.33%–103.66%. And precision, expressed as the standard
6 deviation, was assessed for five replicates with the same spiked samples
7 (10.83%–18.33%). The results indicated that this assay would be useful for detect
8 AFB1 residues in real samples.

9

10 **4. Conclusion**

11 In summary, we developed a simple and novel SERS-aptasensor for the detection of
12 AFB1 based on heterogeneous Au NS core–Ag NP satellite nanostructures. This novel
13 SERS sensor for the detection of AFB1 was fabricated for the first time. The sensor
14 quantified the concentration of AFB1 with a limit of detection of 0.48 pg/mL and
15 confirmed the feasibility and selectivity of this sensor. Therefore, we believe that this
16 SERS sensor may have potential applications in the future.

17

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22 201310128, 201310135).

23

24 **References**

25 1. Y. Lin, Q. Zhou, Y. Lin, D. Tang, G. Chen, D. Tang, *Biosens. Bioelectron.* 2015, 74, 680.

- 1 2. S. Srivastava, V. Kumar, M.A. Ali, P.R. Solanki, A. Srivastava, G. Sumana, P.S. Saxena, A.G.
2 Joshi, B.D. Malhotra, *Nanoscale*. 2013, 5(7), 3043.
- 3 3. W.B. Shim, M.J. Kim, H. Mun, M.G. Kim, *Biosens. Bioelectron.* 2014, 62, 288.
- 4 4. X. Xu, X. Liu, Y. Li, Y. Ying, *Biosens. Bioelectron.* 2013, 47, 361.
- 5 5. W. Xu, Y. Xiong, W. Lai, Y. Xu, C. Li, M. Xie, *Biosens. Bioelectron.* 2014, 56, 144.
- 6 6. A. Bacaloni, C. Cavaliere, F. Cucci, P. Foglia, R. Samperi, A. Laganà, *J.Chromatogr. A*. 2008,
7 1179(2), 182.
- 8 7. W.S. Khayoon, B. Saad, T.P. Lee, B. Salleh, *Food. Chem.* 2012, 133(2), 489.
- 9 8. F.Y. Yu, A.V. Gribas, M.M. Vdovenko, I.Y. Sakharov, *Talanta*. 2013, 107, 25.
- 10 9. X. Guo, F. Wen, N. Zheng, Q. Luo, H. Wang, H. Wang, S. Li, J. Wang, *Biosens. Bioelectron.*
11 2014, 56, 340.
- 12 10. T. Li, J.Y. Byun, B.B. Kim, Y.B. Shin, M.G. Kim, *Biosens. Bioelectron.* 2013, 42, 403.
- 13 11. X. Wang, R. Niessner, D. Knopp, *Sensors*. 2014, 14(11), 21535.
- 14 12. S. Wu, N. Duan, C. Zhu, X. Ma, M. Wang, Z. Wang, *Biosens. Bioelectron.* 2011, 30(1), 35.
- 15 13. Y. Wang, B. Yan, L. Chen, *Chem. Rev.* 2013, 113(3), 1391.
- 16 14. Y. Zheng, T. Thai, P. Reineck, L. Qiu, Y. Guo, U. Bach, *Adv. Func. Mater.* 2013, 23(12),
17 1519.
- 18 15. S. Schlücker, *Angew. Chem. Int. Ed.* 2014, 53(19), 4756.
- 19 16. K. Liu, N. Zhao, E. Kumacheva, *Chem. Soci. Rev.* 2011, 40(2), 656.
- 20 17. K.D. Osberg, M. Rycenga, N. Harris, A.L. Schmucker, M.R. Langille, G.C. Schatz, C.A.
21 Mirkin, *Nano. Lett.* 2012, 12(7), 3828.
- 22 18. S. Li, L. Xu, W. Ma, H. Kuang, L. Wang, C. Xu, *Small*. 2015, 11(28), 3435.
- 23 19. L. Xu, W. Yan, W. Ma, H. Kuang, X. Wu, L. Liu, Y. Zhao, L. Wang, C. Xu, *Adv. Mater.* 2015,
24 27(10), 1706.
- 25 20. W. Ma, H. Kuang, L. Xu, L. Ding, C. Xu, L. Wang, N.A. Kotov, *Nat. Commun.* 2013a, 4,
26 2689.
- 27 21. A. Shiohara, S.M. Novikov, D.M. Solís, J.M. Taboada, F. Obelleiro, L.M. Liz-Marzán, *The J.*
28 *Phys. Chem. C*. 2015, 119(20), 10836.
- 29 22. X. Zhao, X. Wu, L. Xu, W. Ma, H. Kuang, L. Wang, C. Xu, *Biosens. Bioelectron.* 2015a, 66,
30 554.
- 31 23. X. Zhao, S. Li, L. Xu, W. Ma, X. Wu, H. Kuang, L. Wang, C. Xu, *Biosens. Bioelectron.*
32 2015b, 70, 372.
- 33 24. L. Xu, H. Kuang, C. Xu, W. Ma, L. Wang, N.A. Kotov, *J. Am. Chem. Soc.* 2012, 134(3),
34 1699.
- 35 25. S. Barbosa, A. Agrawal, L. Rodríguez-Lorenzo, I. Pastoriza-Santos, R.n.A. Alvarez-Puebla,
36 A. Kornowski, H. Weller, L.M. Liz-Marzán, *Langmuir*. 2010, 26(18), 14943.
- 37 26. W. Ma, M. Sun, L. Xu, L. Wang, H. Kuang, C. Xu, *Chem. Commun.* 2013b, 49(44), 4989.
- 38 27. C. Hrelescu, T.K. Sau, A.L. Rogach, F. Jäkel, G. Laurent, L. Douillard, F. Charra, *Nano. Lett.*

- 1 2011, 11(2), 402.
- 2 28. J. Zhao, A.O. Pinchuk, J.M. McMahon, S. Li, L.K. Ausman, A.L. Atkinson, G.C. Schatz, *Acc.*
3 *Chem. Res.* 2008, 41(12), 1710.
- 4 29. H. Yuan, C.G. Khoury, H. Hwang, C.M. Wilson, G.A. Grant, T. Vo-Dinh, *Nanotechnology*.
5 2012, 23(7), 075102.
- 6 30. X. Wu, L. Xu, L. Liu, W. Ma, H. Yin, H. Kuang, L. Wang, C. Xu, N.A. Kotov, *J.Am. Chem.*
7 *Soc.* 2013, 135(49), 18629.
- 8 31. Hurst, S.J., Lytton-Jean, A.K.R., Mirkin, C.A., *Anal. Chem.* 2006, 78 (24), 8313–8318.
9
- 10

1 **Captions:**

2 **Scheme 1** Scheme of AFB1 detection based on SERS method using Au NS core-Ag
3 NP satellites assemblies.

4 **Fig. 1** TEM images of Ag NPs (a) and Au NSs (b).

5 **Fig. 2** The change of UV-Vis spectra of Ag NPs and Au NSs before and after
6 assembly (A) and the changes of hydrodynamic size between Ag NPs and Ag
7 NP-DNA1(B).

8 **Fig. 3** Representative TEM images of AuNS core-AgNP satellites assemblies in the
9 presence of different concentrations of AFB1: (a) 0 pg/mL; (b) 1 pg/mL; (c)10 pg/mL;
10 (d) 50 pg/mL; (e) 100 pg/mL; (f) 1000 pg/mL. Scale bar in the inset corresponds to 50
11 nm.

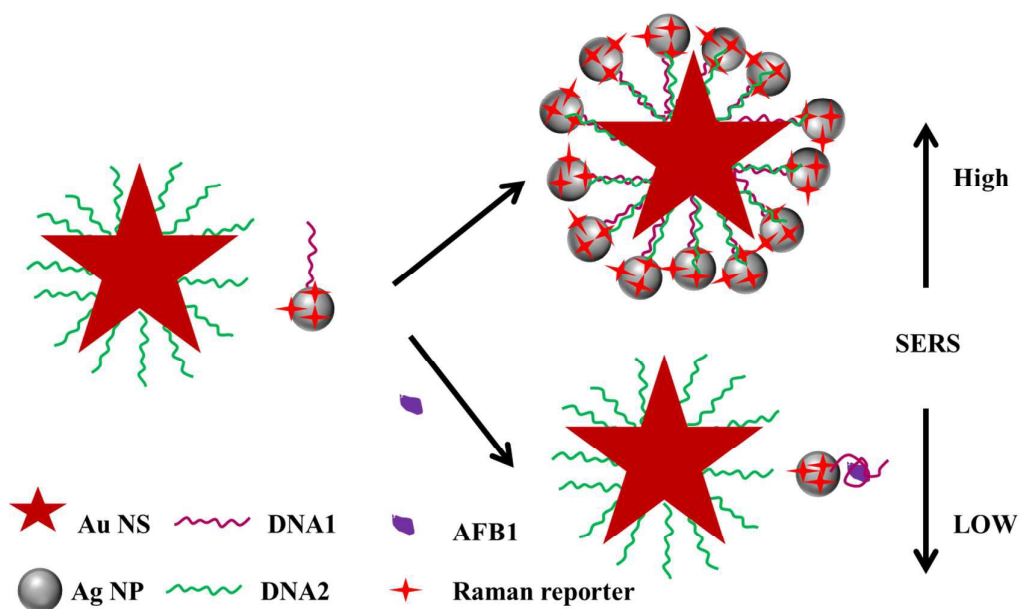
12 **Fig. 4** UV-Vis spectra of the probe in the presence of AFB1 ranging 0 from 1000
13 pg/mL.

14 **Fig. 5** (a) The SERS spectrum of 4-ATP with the different concentration of AFB1;
15 (b) The standard curve based on 1136 cm⁻¹ frequency of 4-ATP.

16 **Fig. 6** Evaluation of the selectivity of SERS sensor at different target analytes. The
17 concentration of AFB1 is 1 ng/mL, other analytes' are 100 ng/mL.

18 **Table 1** Recovery of AFB1 spiked in peanut milk samples

19
20



Scheme 1 Scheme of AFB1 detection based on SERS method using Au NS core-Ag NP satellites assemblies.

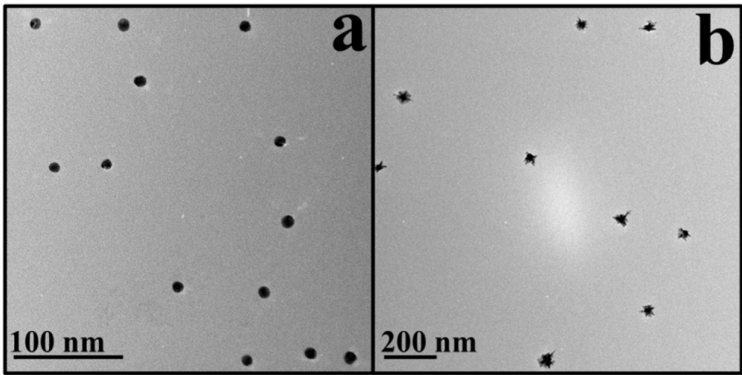


Fig. 1 TEM images of Ag NPs (a) and Au NSs (b).

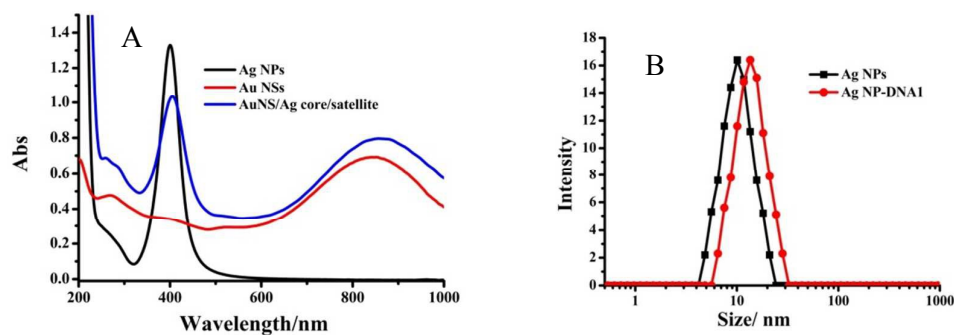
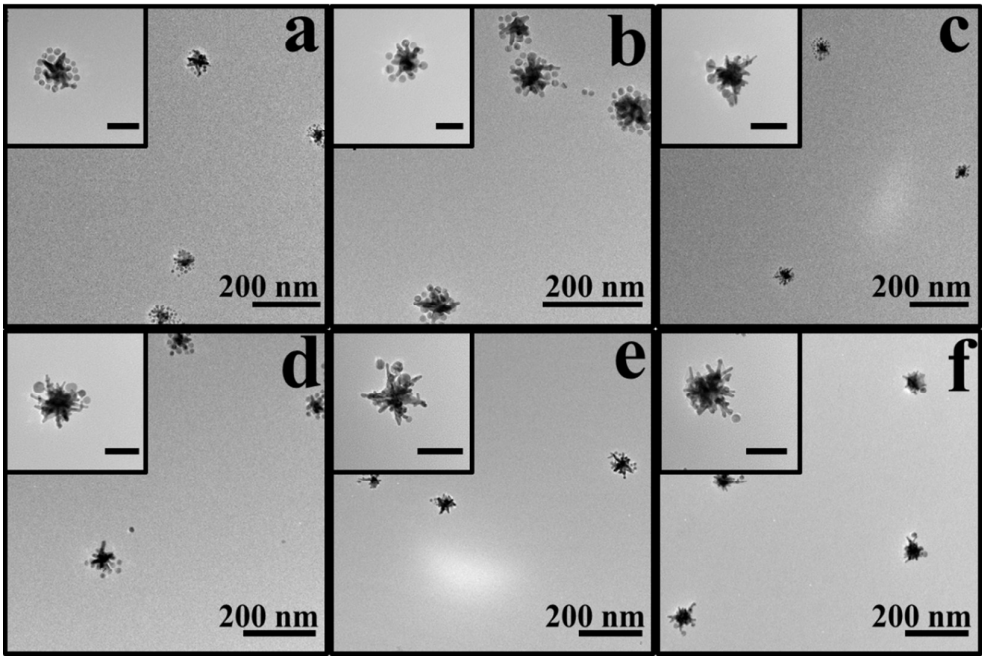


Fig. 2 The change of UV-Vis spectra of Ag NPs and Au NSs before and after assembly (A) and the changes of hydrodynamic size between Ag NPs and Ag NP-DNA1(B).



1
2 **Fig. 3** Representative TEM images of AuNS core-AgNP satellites assemblies in the
3 presence of different concentrations of AFB1: (a) 0 pg/mL; (b) 1 pg/mL; (c) 10 pg/mL;
4 (d) 50 pg/mL; (e) 100 pg/mL; (f) 1000 pg/mL. Scale bar in the inset corresponds to 50
5 nm.

6

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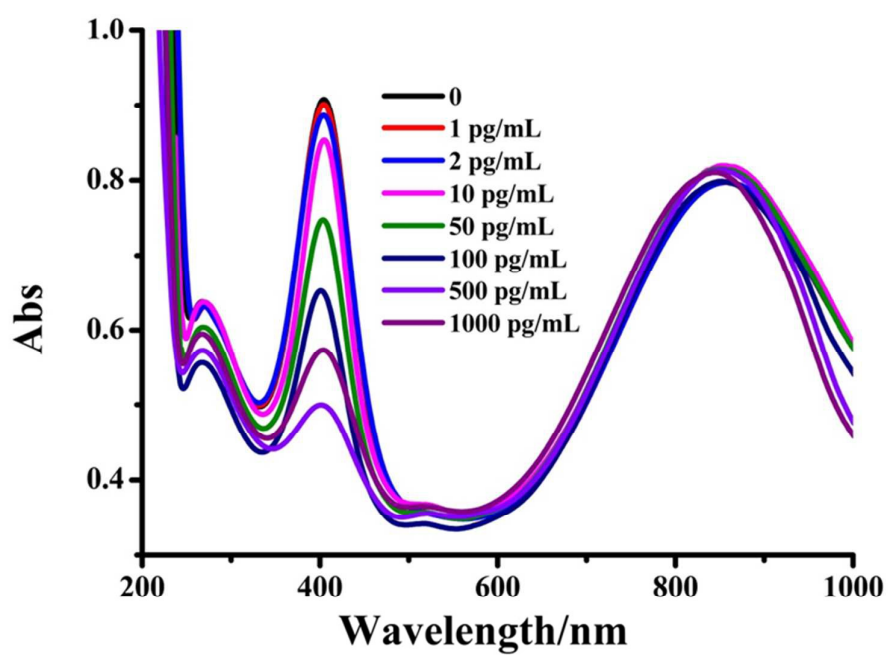
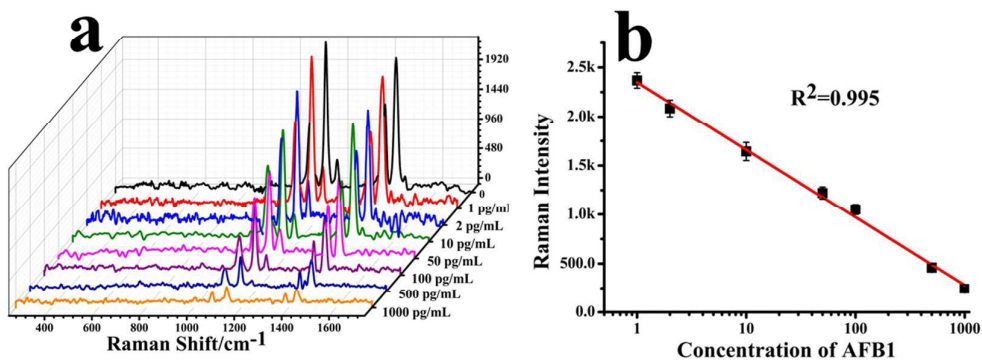


Fig. 4 UV-Vis spectra of the probe in the presence of AFB1 ranging 0 from 1000 pg/mL.



1
2 **Fig. 5** (a)The SERS spectrum of 4-ATP with the different concentration of AFB1; (b)
3 The standard curve based on 1136 cm⁻¹ frequency of 4-ATP.

4
5

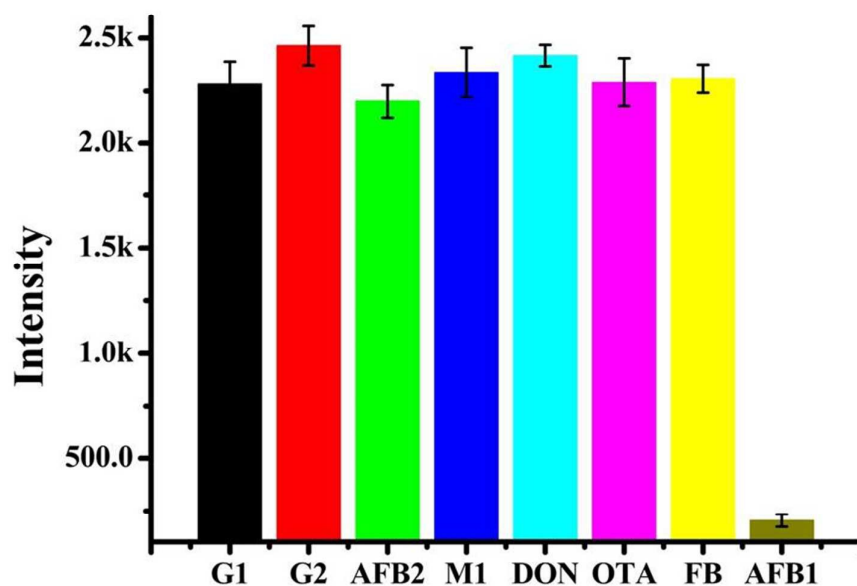


Fig. 6 Evaluation of the selectivity of SERS sensor at different target analytes. The concentration of AFB1 is 1 ng/mL, other analytes' are 100 ng/mL.

1 **Table 1** Recovery of AFB1 spiked in peanut milk samples

Sample	Spiked Concentration pg mL ⁻¹)	Detected Concentration (Mean±SD, ngmL ⁻¹ , n = 5)	Recovery (%) (Mean±SD, n = 5)
Peanut milk	0.6	0.53±0.11	88.33±18.33
	3	2.83±0.34	94.33±11.34
	6	6.22±0.65	103.66±10.83

2

Supporting information

Caption:

Fig. S1 Representative TEM images of Au NS core-Ag NP satellites assemblies in the presence of different concentrations of AFB1 (Top), and statistical analysis of the number of Ag NPs surrounding Au NS in the presence of different concentrations of AFB1 (Bottom).

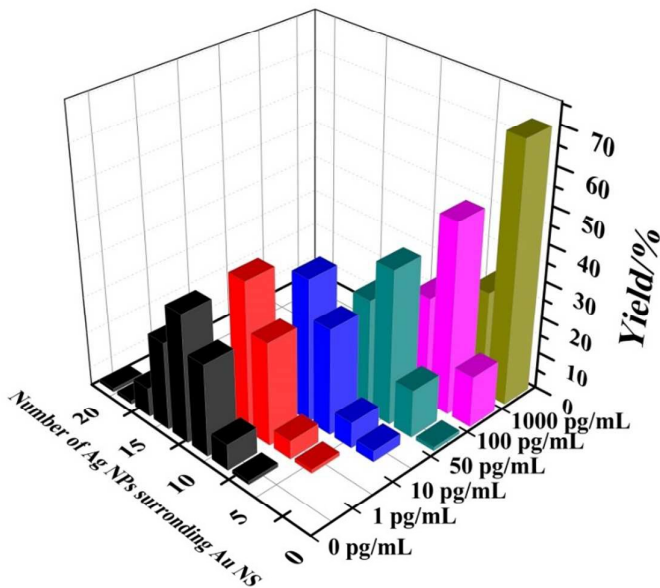
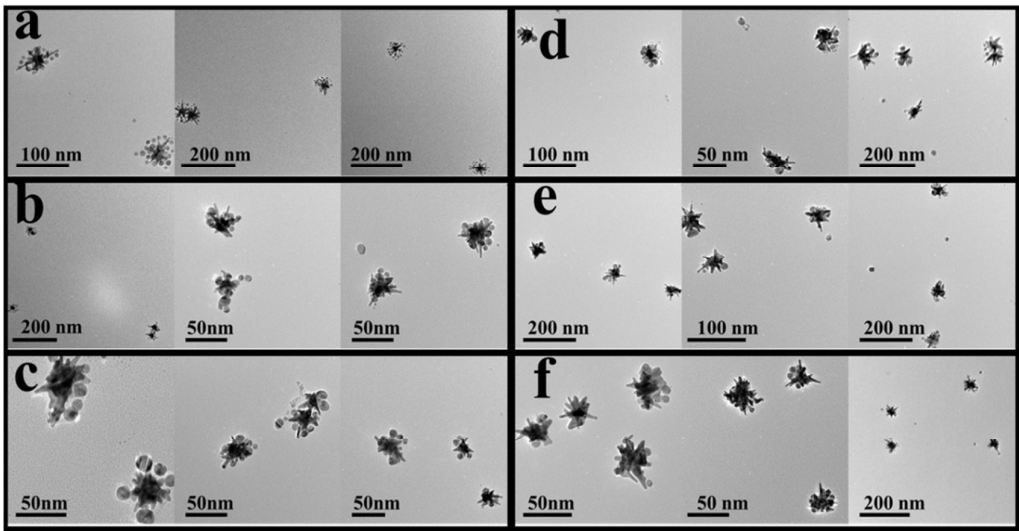


Fig. S1 Representative TEM images of Au NS core-Ag NP satellites assemblies in the presence of different concentrations of AFB1: a) 0 pg/mL; b) 1 pg/mL; c) 10 pg/mL; d) 50 pg/mL; e) 100 pg/mL; f) 1000 pg/mL (Top), and statistical analysis of the number of Ag NPs surrounding Au NS in the presence of different concentrations of AFB1 (Bottom).