

Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

PAPER

Accelerated Fmoc Solid-phase Synthesis of Peptides with Aggregation-disrupting Backbones

Yi-Chao Huang,^{a,b} Chao-Jian Guan,^a Xiang-Long Tan,^b Chen-Chen Chen,^{a,c} Qing-Xiang Guo^c and Yi-Ming Li^{*a}

Received (in XXX, XXX) Xth XXXXXXXXX 20XX, Accepted Xth XXXXXXXXX 20XX
DOI: 10.1039/b000000x

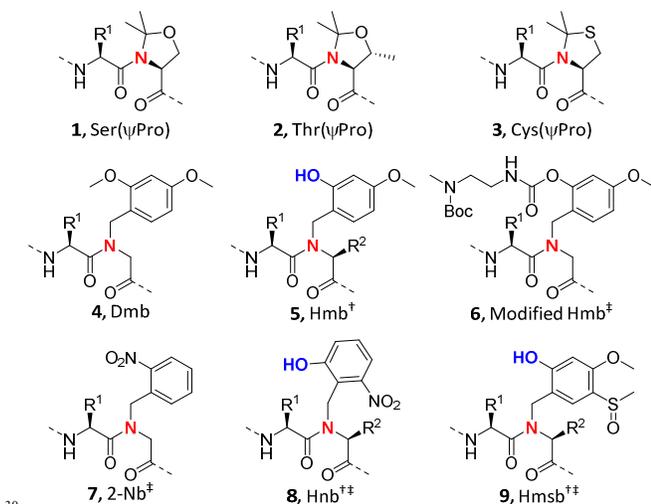
In this work, we describe an accelerated solid-phase synthetic protocol of ordinary or difficult peptides involving air-bath heating and amide protection. For the Hmsb-based backbone amide protection, an optimized acyl shift condition using 1,4-dioxane was discovered. The efficiency and robustness of the protocol was validated in the course of preparation of classical difficult peptides and ubiquitin protein segments.

Introduction

Total synthesis of peptides and proteins provides valuable molecular tools for chemical biology and medicinal chemistry.¹ Despite the introduction of routine solid phase peptide synthesis (SPPS) by Merrifield and the development of powerful native chemical ligation (NCL) by Kent,^{2,3} the protein assembly process is often plagued by the presence of difficult peptide sequences. For Boc SPPS, repetitive TFA treatment of resins combined with *in situ* neutralization coupling can moderate the aggregation effect.⁴ However, the toxicity of hydrogen fluoride as the cleavage reagent makes Boc SPPS less favorable to peptide chemists. While in the case of Fmoc SPPS where on-resin aggregation can easily occur through inter- or intramolecular chain interactions, difficult sequences pose a more severe bottleneck toward preparation of high-quality peptides.⁵ To address the issue of difficult sequences, many disaggregating strategies have been devised which generally fall into two categories.

Figure 1 Typical backbone protecting groups used during Fmoc SPPS. [†]These protecting groups are less sensitive to the steric hinderance of residue R¹ and R². [‡]These protecting groups are stable under conventional TFA cleavage conditions.

The first type of strategies (physical means) combat difficult sequences through adjusting the external environment during SPPS, which includes: 1) switching typical polystyrene resins to low-substitution high-swelling hydrophilic resins such as PEG-based resins,⁶ 2) using alternative solvents or solvent mixtures (DMF, NMP, DMSO) for better swelling effects,⁷ 3) adding chaotropic salts (LiCl, KSCN, NaClO₄) into the coupling solution,⁸ 4) employing more aggressive coupling reagents such as HATU/DIEA or COMU/DIEA single/multiple couplings for a prolonged time,^{9,10} and 5) applying sophisticated microwave or conventional oil-bath heating apparatus.¹¹ The second group of strategies (chemical means) take advantage of backbone amide protection to suppress the amide N-H involved on-resin β -sheet formation (Figure 1).¹² Kiso and Mutter group designed an elegant aggregation-breaking method by the incorporation of “click peptides” or “switch peptides”.¹³ This method mostly relies on the synthesis of depsipeptides (O-acyl isopeptides) followed by *in-solution* O \rightarrow N intramolecular acyl transfer to restore native Xaa-Ser/Thr peptide bonds. Besides this depsipeptide approach, commercially available Fmoc-protected pseudoproline dipeptides have gained wide popularity as aggregation-breaking building blocks 1–3.¹⁴ An important feature is that pseudoproline structures can be converted to native amides during acidic cleavage. Unfortunately, their utility is limited to Xaa-Ser, Xaa-Thr or less frequently Xaa-Cys containing sequences. Another popular method exploits the acid lability of electron-rich benzyl protected amide. Both 2,4-dimethoxybenzyl (Dmb) 4 and 2-hydroxy-4-methoxybenzyl (Hmb) 5 groups were found to have considerable aggregation-disrupting effects.^{15,16} These groups can also be cleaved under TFA to recover native amide bonds. Importantly, Fmoc-Asp(OtBu)-(Dmb)Gly-OH and Fmoc-Asp(OtBu)-(Hmb)Gly-OH have proven to be reliable building blocks to completely avoid the notorious aspartimide formation.¹⁷



Recently an Hmb derivative **6** was developed to allow the incorporation of an oligo-arginine tag which is conducive to the synthesis of transmembrane peptides,¹⁸ almost exclusively difficult sequences. The major drawback of these benzyl type protecting groups is the huge steric hindrance around the tertiary amide. The choice of R² is usually constricted to Gly and Ala. This is also the case for a photo-labile amide protecting group 2-nitrobenzyl (2-Nb) **7**.¹⁹

Interestingly, the phenol group in Hmb **5** permits fast phenol ester formation and then tertiary amide generation through six-membered acyl migration.¹⁶ This special mechanism slightly enhanced the residue (R¹ and R²) tolerance of Hmb **5** compared with Dmb **4**. To speed up the rate-limiting acyl shift step, several electron-withdrawing modifications (**8** and **9**) on 2-hydroxybenzyl were reported having better acyl shift kinetics.^{20,21} An additional advantage of these backbone protecting groups is that they are stable to standard acidic cleavage cocktail. This feature is instrumental because peptides containing tertiary amide backbones usually bear better HPLC elution behavior (shorter retention time and sharper peak) and greater solubility in the LC elution buffer (0.1% TFA, acetonitrile/water = 1/1, pH 1) or the NCL buffer (6 M Gn-Cl, pH 4-8).^{18,22,23} After post-cleavage purifications and segment ligations of backbone-protected peptides, 2-hydroxy-6-nitrobenzyl (Hnb) **8** and 2-hydroxy-4-methoxy-5-methylsulfinylbenzyl (Hmsb) **9** were found to be cleavable by using UV illumination and sulfoxide-reducing TFA (Figure S1, Scheme S1) respectively to generate free peptides. A similar safety-catch structure 2-methoxy-4-methylsulfinylbenzyl (Mmsb) was also reported by Albercio group very recently.²⁴

Previous work from Offer's group indicated that Hmsb **9** could improve the purity and solubility of on-resin aggregation-prone peptide sequences for both manual and automated SPPS.²¹ However the tolerance of **9** to β -branched amino acids was not investigated. In addition, to test the scope and feasibility of this method, we thought it necessary to prepare longer peptides (>25 aa) with multiple Hmsb-protected amides. Herein, we discovered a new solvent incubation protocol which exhibited better acyl shift kinetics to incorporate **9**. Our results showed that the new acyl transfer condition could tolerate couplings of sterically hindered β -branched amino acids (Ile, Val, Thr). Moreover, we successfully prepared several difficult sequences (70%-95% HPLC purities of crude products) using air-bath heating batch-based SPPS in conjunction with Hmsb-based backbone amide protection. The timescale of a single deprotection-washing-coupling SPPS cycle can be shortened by as much as 90 min using the new air-bath heating apparatus.

Results and Discussion

Discovery of the optimized acyl shift condition in dioxane

On our route to extend the applicability of Fmoc SPPS with Hmsb-based backbone protection, we noticed Alewood group's report that O-acylation proceeds fast and quantitatively while the following acyl transfer step is sluggish. According to previous literature,^{20a,21,25} treatment of resins with nonpolar solvents could facilitate on-resin intramolecular acyl transfer. Therefore, we

began our acyl shift optimization by screening various solvents or solvent mixtures.

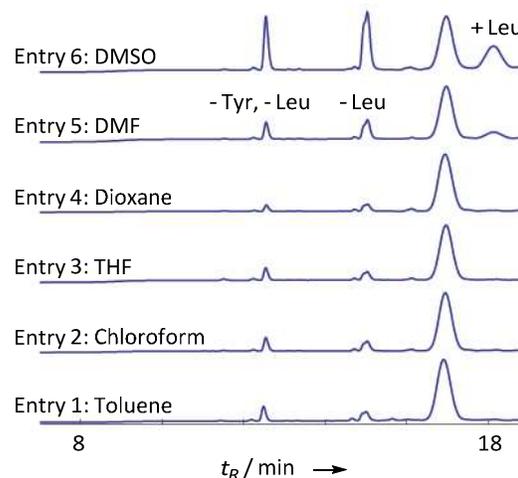
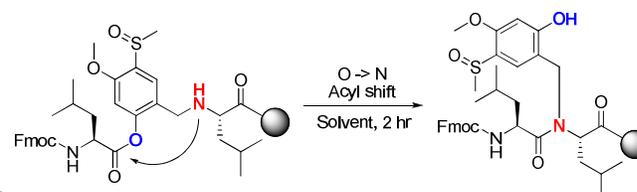


Figure 2 HPLC traces of H-Tyr-Leu-(Hmsb)Leu-Ser-Lys-NH₂ synthesized under different acyl shift conditions. All acyl shift reactions were carried out in 5% (v/v) DIEA/solvent at 50 °C for 2 hr.

Table 1 Optimization of the on-resin acyl shift reaction during Fmoc SPPS to prepare H-Tyr-Leu-(Hmsb)Leu-Ser-Lys-NH₂.



Entry	Solvent (mixture)	DIEA [v/v %]	Temperature [°C]	HPLC purity [%]
1	toluene	5	50	82
2	chloroform	5	50	83
3	THF	5	50	84
4	1,4-dioxane	5	50	90
5	DMF	5	50	74
6	DMSO	5	50	40
7	THF/MeOH = 4/1	5	50	67
8	THF/H ₂ O = 4/1	5	50	66
9	1,4-dioxane	0	50	91
10	1,4-dioxane	0	25	77
11	DMF	0	25	66
12	1,4-dioxane	0	75	97
13	(Acyl shift step skipped)			52

Initially, we set two fundamental requirements for appropriate solvents, good resin-swelling capability and high boiling point. The first criterion is crucial for fast on-resin kinetics and the second is beneficial in case of high-temperature SPPS.²⁶ Based on this, we tested the effects of six solvents towards the on-resin acyl shift reaction on a model pentapeptide, H-Tyr-Leu-(Hmsb)Leu-Ser-Lys-NH₂ (Figure 2). All deprotection and couplings were conducted under standard room temperature Fmoc SPPS conditions except the acyl shift step. Hmsb was synthesized using a slightly modified procedure compared with earlier paper (Scheme S2). For good comparison, all reactions were conducted at 50°C for 2 hr in the presence of 5% (v/v %) DIEA (Table 2, entries 1-8). Modified cocktail K (TFA/H₂O/thioanisole/EDT = 87/5/5/3) was used as the cleavage reagent for sulfoxide-containing peptides.^{27,28}

As expected,²⁵ all of the nonpolar solvents afforded the crude Hmsb-protected peptide with good HPLC purity (Table 1, entries 1-4). Interestingly, 1,4-dioxane worked better than other less polar solvents (90% HPLC purity), possibly due to its excellent ability in resin-swelling and chain-solvation. Inferior results were obtained when aprotic polar solvents DMF and DMSO were used (Table 1, entries 5-6). In the case of DMSO, three byproducts were detected through HPLC and ESI-MS (Figure 2), which corresponded to a Tyr¹ and Leu² deletion product, a Leu² deletion product, and a surprising Leu² addition product. The deletion product indicated incomplete acyl shift reaction of Leu³. The appearance of the addition byproduct might be caused by premature Fmoc deprotection by the N-terminal secondary amine of peptide chain in an aprotic polar solvent in the presence of base at higher temperature. Addition of water or methanol as a co-solvent during acyl shift failed to produce the product with a higher yield (Table 1, entries 7-8). Further experiments showed that additional base was unnecessary and the acyl shift happened more efficiently at higher temperature without DIEA (Table 1, entries 9-12). Therefore, neat dioxane (b.p. 101°C) was chosen as the ideal solvent in the following acyl-shift experiments. Encouragingly, compared with Offer's previous protocol using DMF as the sole solvent during coupling and acyl shift, our new 1,4-dioxane involved acyl shift protocol gave a noticeable improvement (Table 1, entry 11-13).

Scope of Hmsb protection of backbone amides

To speed up Hmsb incorporation, we decided to perform both O-acylation and acyl shift reactions at elevated temperature. Based on recent reports about microwave or oil-bath involved SPPS,^{11c,11d} we believed that manual SPPS at 50-90°C could be attempted which might afford comparable results to microwave-based automated SPPS.²⁹ Due to concerns of racemization and other side-reactions,³⁰ we carried out base-free DIC/Oxyma couplings at 75°C (air-bath heating) for all the Fmoc amino acids except Arg, Cys and His.³¹ Cysteine was coupled at 50°C while Arg and His were coupled at room temperature. Next, we synthesized Hmsb protected sequences containing racemization-prone Phe, Ser, Cys residues (Figure 3 and Figure S2). Note that both Ac-Ser[L]- (Hmsb)Ala-Lys-NH₂ and Ac-Ser[D]- (Hmsb)Ala-Lys-NH₂ showed double peaks on HPLC (peak volume ~ 1:1) with identical molecular weights (Figure 3B). This was not caused by on-resin racemization of the building block but the fact

that Hmsb was synthesized as a racemic sulfoxide compound. Interestingly, racemic Ac-Phe[L]- (Hmsb)Ala-Lys-NH₂ or Ac-Phe[D]- (Hmsb)Ala-Lys-NH₂ seemed inseparable using the same HPLC gradient. All of the LC co-injection traces confirmed that racemization ratio was quite low (< 2%) during the manual high-temperature synthesis of Hmsb-protected peptides (Figure 3).

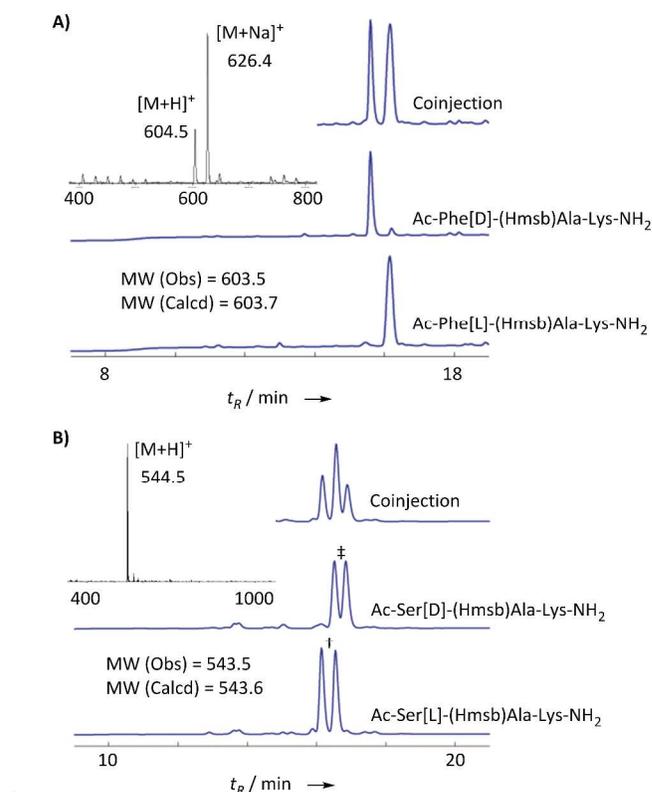
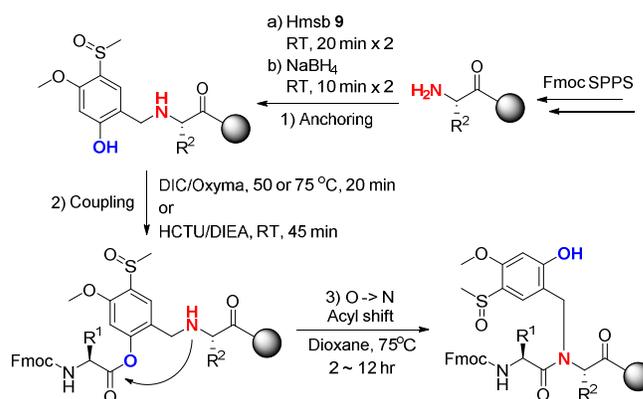


Figure 3 Racemization tests of A) Ac-Phe[L/D]- (Hmsb)Ala-Lys-NH₂ and B) Ac-Ser[L/D]- (Hmsb)Ala-Lys-NH₂. †Ac-Ser[L]- (Hmsb)Ala-Lys-NH₂ containing racemic sulfoxides (~ 1:1) are separable by HPLC. ‡Ac-Ser[D]- (Hmsb)Ala-Lys-NH₂ containing racemic sulfoxides are separable by HPLC.



Scheme 1 Fully optimized incorporation of Hmsb 9 consists of three steps, 1) anchoring of salicylaldehyde onto free amino resin through reductive amination; 2) coupling of Fmoc-AA-OH to form a phenolate under room temperature or high temperature conditions; 3) on-resin acyl shift in 1,4-dioxane to make a protected tertiary amide bond.

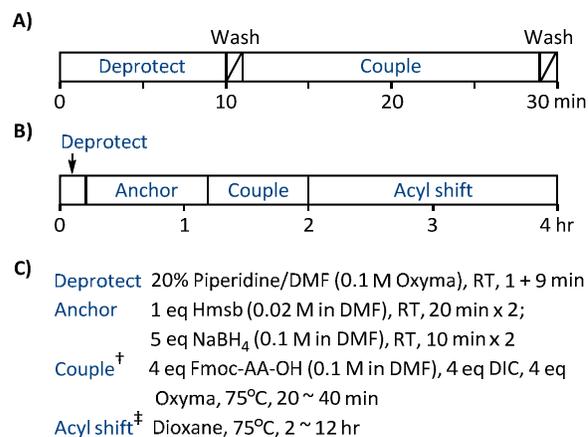
Table 2 Scope of the on-resin acyl shift reaction during Fmoc SPPS using a model sequence H-Met(O)-Xaa-(Hmsb)Yaa-Ser-Lys-NH₂.

Entry	Xaa	Yaa	Acyl shift time [hr]	HPLC purity [%]
1	Leu	Leu	2	95
2	Asn	Asn	2	96
3	Phe	Phe	2	78
4 [†]	Arg	Arg	2	80
5	Pro	Gln	2	86
6	Trp	Tyr	2	70
7	Ile	Gly	12	90
8	Ile	Ala	12	86
9	Val	Ala	12	85
10	Thr	Ala	12	82
11	Val	Leu	24	75
12	Leu	Val	24	< 5

[†]The acyl shift of Fmoc-Arg(Pbf)-OH was performed at room temperature rather than 75°C.

To study the reaction scope of the on-resin acyl shift reaction, we prepared another model pentapeptide H-Met(O)-Xaa-(Hmsb)Yaa-Ser-Lys-NH₂. The reaction efficiency was evaluated based on the HPLC purities of crude products using various Xaa/Yaa combinations (Figure S3 and Table 2). We found that most β-linear amino acids are compatible with the on-resin acyl shift (Table 2, entries 1-6). The crude peptides displayed good to excellent HPLC purity. More importantly, Xaa seems to tolerate all the sterically hindered β-branched amino acids when Yaa = Gly or Ala (Table 2, entries 7-10). However, Val-(Hmsb)Leu and Leu-(Hmsb)Val gave worse results indicating that the acyl shift kinetics is more sensitive to the steric bulkiness of Yaa than Xaa (Table 2, entries 11-12).^{19,20a} We believed that these experimental data would be equally helpful when amino acid coupling and acyl shift were encountered after Hmb **5** and Hnb **8** incorporation.

With the scope of Hmsb protection of backbone amides investigated, we updated the procedure of incorporation of Hmsb, 1) anchoring via on-resin two-step reductive amination; 2) coupling via O-acylation; 3) acyl shift via incubation in dioxane (Scheme 1). Furthermore, we depicted a timeline for Fmoc SPPS which is based on air-bath heating (50°C and 75°C). Traditional Fmoc SPPS requires 45 to 120 min to introduce each amino acid building block. In comparison, our new coupling cycle (deprotection, washing and coupling) takes only 30 min to complete with an easily accessible air-bath shaker as the only additional piece of equipment for manual SPPS (Scheme 2A and Figure S4). The actual coupling time at high temperature is around 15 min (excluding 3-5 min for heat exchange). Piperidine deprotection was carried out at room temperature to suppress any base-induced side reactions.



Scheme 2 A) Standard manual SPPS timeline. B) Manual SPPS timeline with Hmsb incorporation. C) The SPPS protocol includes four steps, deprotection, anchoring, coupling and acyl shift. [†]HCTU/DIEA (RT, 45 min) or DIC/Oxyma (RT, 90 min) coupling is chosen for Fmoc-Arg(Pbf)-OH to prevent δ-lactam formation of arginine. DIC/Oxyma (RT, 45 min) coupling is chosen for Fmoc-His(Trt)-OH to suppress racemization. DIC/Oxyma (50°C, 20 min) coupling is chosen for Fmoc-Cys(Trt)-OH to suppress racemization. DIC/Oxyma (75°C, 20 min) coupling is used for the rest 17 canonical amino acids. [‡]The acyl shift step can be carried out for 2 ~ 12 hr, depending on the steric bulkiness of both the acceptor and donor amino acids of the acyl transfer. For Fmoc-Arg(Pbf)-OH and Fmoc-His(Trt)-OH, the temperature should be set at RT to minimize any side reaction.

Further tests of Fmoc SPPS assisted by air-bath heating

To evaluate the robustness of our new fast Fmoc SPPS protocol, we selected four difficult sequences from the literature which was tested via air-bath heating Fmoc SPPS alone and another two sequences through high-temperature SPPS plus Hmsb incorporation. Four difficult peptides, ACP(65-74), ABC 20mer, PnIA(A10L) and HIV-1 PR(81-99) were synthesized on a 0.1 mmol scale using 30 min coupling cycle (Scheme 2A). As shown in Figure 4, crude peptide products were obtained at good to excellent HPLC purity (70%-95% based on peak area ratios, see Table 3). For all of the four peptides, the whole main peak on analytical HPLC was collected and characterized by direct injection ESI-MS.

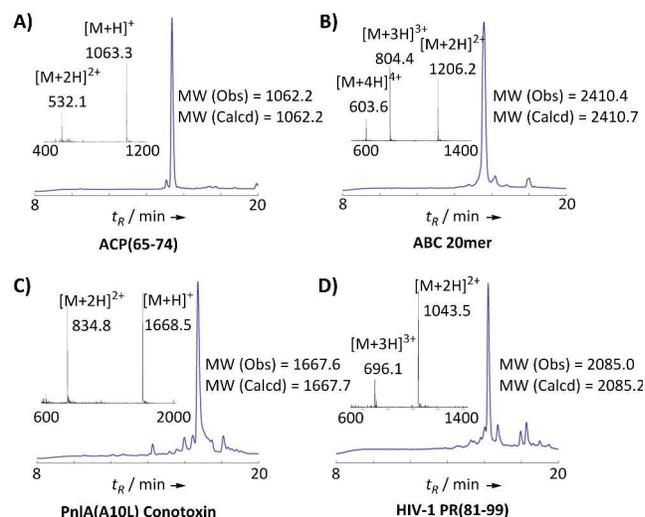


Figure 4 Fmoc SPPS of difficult peptides using air-bath heating. A) ACP(65-74): H-VQAAIDYING-NH₂; B) ABC 20mer: H-VYWTSPFM(O)KLIHEQCNRADG-NH₂; C) PnIA(A10L) Conotoxin: H-GCCSLPPCALNNPDYC-NH₂; D) HIV-1 PR(81-99): H-PVNIIGRNLTLQIGCTLNF-NH₂.

Two additional sequences PolyL 10mer and toxin protein Mambalgin-1(19-40) were further synthesized using both air-bath heating SPPS and Hmsb backbone protection. PolyL10 was found insoluble in common HPLC elution buffer H₂O/ACN = 1/1 when prepared through high-temperature SPPS alone. However, when Hmsb was incorporated between Leu⁵ and Leu⁶ as a backbone kink, we obtained the fully soluble and chemically homogeneous product (Figure 5A and Table 3). In the same manner, we targeted another difficult sequence Mambalgin-1(19-40). In a control experiment, air-bath SPPS gives the crude peptide with average purity. In contrast, both one (between Leu³⁴ and Gln³⁵) or two (between Leu³⁰ and Lys³¹; Gln³⁵ and Gly³⁶) amide-protected Mambalgin-1(19-40) crude peptides showed better HPLC purity than the former (Figure 5B and Table 3).

Finally, we applied our new protocol to two protein segments, ubiquitin(1-45) and ubiquitin(46-76), to verify its robust performance. Both sequences were prepared with or without Hmsb incorporation using air-bath heating SPPS. As shown in Figure 6 and Table 3, backbone amide protection significantly improved the qualities of crude products. Two Hmsb deletion peaks could be identified from the mass spectrum of ubiquitin(1-45) which probably resulted from partial decomposition of tertiary amide in the ionization room of the mass spectrometer. No significant deletion or truncation side-products could be found on the mass spectra of the main peaks. After semi-preparative HPLC purification, ubiquitin H-M(O)¹-F⁴⁵-NHNH₂ [G¹⁰-(Hmsb)K¹¹, L¹⁵-(Hmsb)E¹⁶] and ubiquitin H-C⁴⁶-G⁷⁶-NH₂ [D⁵²-(Hmsb)G⁵³, L⁶⁷-(Hmsb)H⁶⁸] were deprotected in a modified TFA cocktail B (2.5% H₂O, 2.5% TIPS, 2.5% EDT, 1% Bu₄NI) to produce the target peptides H-M¹-F⁴⁵-NHNH₂ and H-C⁴⁶-G⁷⁶-NH₂ (Figure 6 and Scheme S1) in > 80% HPLC yield. Compared with previously reported additives such as NH₄I/Me₂S,³² NH₄I/TIPS and TMSBr or Bu₄NBr/thioanisole/EDT,^{33,34} in our hands, Bu₄NI assisted sulfoxide reduction gave essentially equal results. However, the Hmsb removal condition would yet need to be optimized to give cleaner deprotection results.

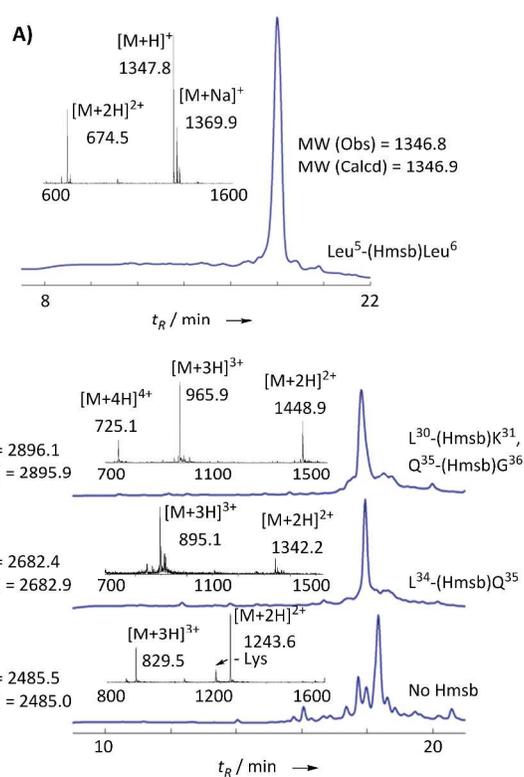


Figure 5 Fmoc SPPS of difficult peptides using air-bath heating along with Hmsb backbone protection. A) PolyL10: H-LLLLL(Hmsb)LLLLL-NH₂; B) Mambalgin-1(19-40): H-CYHNTGMPFRNLKLILQGCSSS-NH₂.

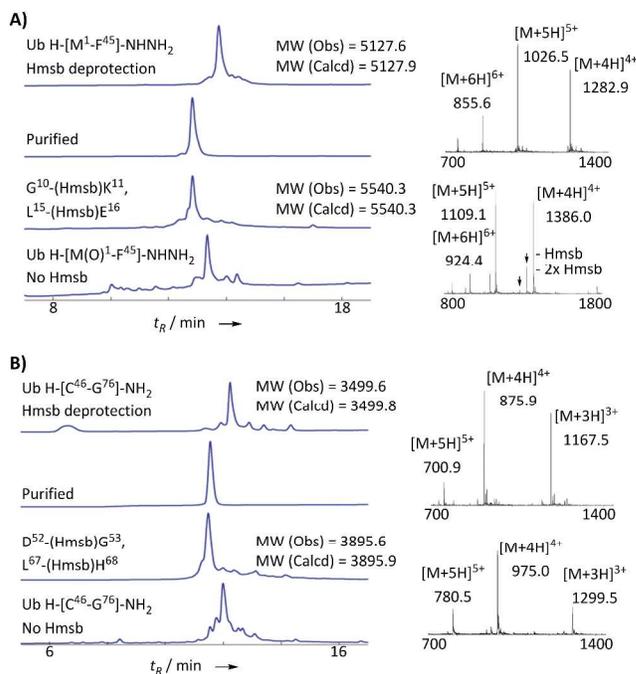


Figure 6 Fmoc SPPS of ubiquitin fragments using air-bath heating along with Hmsb backbone protection. A) Ubiquitin H-M(O)¹-F⁴⁵-NHNH₂; B) Ubiquitin H-C⁴⁶-G⁷⁶-NH₂.

Table 3 Synthesis Results.

Entry	Peptide	Condition ^a	HPLC purity [%]	Crude yield ^b [%]
1	ACP(65-74)	A	95	86
2	ABC20mer	A	92	87
3	PnIA(A10L) Conotoxin	A	76	82
4 [†]	HIV-1 PR(81-99)	A	70	91
5	PolyL10	A	89	85
6	PolyL10	B	N. D.	26
7	Mambalgin-1 (19-40)	A	50	67
8	Mambalgin-1 (19-40)	B	38	62
9	Mambalgin-1 (19-40)	C	72	83
10	Mambalgin-1 (19-40)	D	80	88
11	Ubiquitin(1-45)	A	42	73
12	Ubiquitin(1-45)	D	65	80
13	Ubiquitin(46-76)	A	31	55
14	Ubiquitin(46-76)	D	72	88

^aA = air-bath heating SPPS; B = standard room temperature SPPS; C = air-bath heating SPPS combined with single Hmsb amide protection; D = air-bath heating SPPS combined with double Hmsb amide protection; ^bCrude yields were determined based on the weight of lyophilized crude cleavage product after ether precipitation.

Conclusions

In summary, we successfully improved the efficiency of Hmsb-based backbone amide protection using the key 1,4-dioxane involved acyl shift in an effort to tackle synthesis of difficult peptides. To speed up the whole synthetic scheme further, we devised a fast Fmoc SPPS protocol using a simple air-bath heating apparatus. Manual synthesis of an ordinary or difficult sequence containing 20-40 amino acid residues can be finished within 1-2 days (30 min for each coupling cycle). We considered that the heat and backbone amide protection worked synergistically to improve the peptide quality. The operability of this protocol was validated in the preparation of two protein segments ubiquitin(1-45) C-terminal α -hydrazide and ubiquitin(46-76) C-terminal α -amide which are useful segment building blocks for assembling versatile ubiquitin derivatives through native chemical ligation.^{35,36} Future projects would be focused on the combinatorial use of accelerated Fmoc SPPS and iterative peptide ligations.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (21102083, 21372058 to Y. M. Li; 21272223 to Q. X. Guo).

Notes and references

^aSchool of Medical Engineering, Hefei University of Technology, Hefei 230009, China, Email: yml@hfut.edu.cn; lym2007@mail.ustc.edu.cn.
^bDepartment of Chemistry, Tsinghua University, Beijing100084, China.
^cDepartment of Chemistry, University of Science and Technology of China, Hefei 230026, China.

[†] Electronic Supplementary Information (ESI) available: general information and details of synthetic procedures, ¹H NMR, ¹³C NMR, HPLC, ESI-MS. See DOI: 10.1039/b000000x/

- (a) S. B. H. Kent, *Chem. Soc. Rev.*, 2009, **38**, 338-351; (b) S. Kent, Y. Sohma, S. Liu, D. Bang, B. Pentelute, K. Mandal, *J. Pept. Sci.*, 2012, **18**, 428-436.
- R. B. Merrifield, *J. Am. Chem. Soc.*, 1963, **85**, 2149-2154.
- P. E. Dawson, T. W. Muir, I. Clarklewis, S. B. H. Kent, *Science*, 1994, **266**, 776-779.
- (a) M. Schnolzer, P. Alewood, A. Jones, D. Alewood, S. B. H. Kent, *Int. J. Pept. Protein Res.*, 1992, **40**, 180-193; (b) L. P. Miranda, P. F. Alewood, *Proc. Natl. Acad. Sci. USA*, 1999, **96**, 1181-1186; (c) M. Schnolzer, P. Alewood, A. Jones, D. Alewood, S. B. H. Kent, *Int. J. Pept. Res. Ther.*, 2007, **13**, 31-44.
- (a) S. B. H. Kent, *Annu. Rev. Biochem.*, 1988, **57**, 957-989; (b) J. P. Tam, Y.-A. Lu, *J. Am. Chem. Soc.*, 1995, **117**, 12058-12063; (c) I. Coin, M. Beyermann, M. Bienert, *Nat. Protoc.*, 2007, **2**, 3247-3256; (d) A. K. Tickler, J. D. Wade, *Curr. Protoc. Protein Sci.*, 18.8.1-18.8.6, November 2007.
- E. Oliveira, A. Miranda, F. Albericio, D. Andreu, A. C. M. Paiva, C. R. Nakaie, M. Tominaga, *J. Pept. Res.*, 1997, **49**, 300-307.
- Y. Garcia-Ramos, M. Paradis-Bas, J. Tulla-Puche, F. Albericio, *J. Pept. Sci.*, 2010, **16**, 675-678.
- R. Steinauer, F. M. F. Chen, N. L. Benoiton, *Helv. Chim. Acta*, 1991, **74**, 628-643.
- L. A. Carpino, *J. Am. Chem. Soc.*, 1993, **115**, 4397-4398.
- A. El-Faham, R. S. Funosas, R. Prohens, F. Albericio, *Chem. Eur. J.*, 2009, **15**, 9404-9416.
- (a) H. M. Yu, S. T. Chen, K. T. Wang, *J. Org. Chem.*, 1992, **57**, 4781-4784; (b) S. L. Pedersen, A. P. Tofteng, L. Malik, K. J. Jensen, *Chem. Soc. Rev.*, 2012, **41**, 1826-1844; (c) J. M. Collins, K. A. Porter, S. K. Singh, G. S. Vanier, *Org. Lett.*, 2014, **16**, 940-943; (d) M. D. Simon, P. L. Heider, A. Adamo, A. A. Vinogradov, S. K. Mong, X. Li, T. Berger, R. L. Policarpo, C. Zhang, Y. Zou, X. Liao, A. M. Spokoynny, K. F. Jensen, B. L. Pentelute, *ChemBioChem* 2014, **15**, 713-720.
- (a) T. Johnson, M. Quibell, R. C. Sheppard, *J. Pept. Sci.*, 1995, **1**, 11-25; (b) J. Offer, M. Quibell, T. Johnson, *J. Chem. Soc., Perkin Trans. 1*, 1996, 175-182.
- (a) Y. Sohma, M. Sasaki, Y. Hayashi, T. Kimura, Y. Kiso, *Chem. Commun.* 2004, 124-125; (b) M. Mutter, A. Chandravarkar, C. Boyat, J. Lopez, S. D. Santos, B. Mandal, R. Mimna, K. Murat, L. Patiny, L. Saucedo, G. Tuchscherer, *Angew. Chem. Int. Ed.*, 2004, **43**, 4172-4178; (c) A. Taniguchi, Y. Sohma, M. Kimura, T. Okada, K. Ikeda, Y. Hayashi, T. Kimura, S. Hirota, K. Matsuzaki, Y. Kiso, *J. Am. Chem. Soc.*, 2006, **128**, 696-697.
- (a) T. Wöhr, F. Wahl, A. Nefzi, B. Rohwedder, T. Sato, X. C. Sun, M. Mutter, *J. Am. Chem. Soc.*, 1996, **118**, 9218-9227; (b) P. Dumy, M. Keller, D. E. Ryan, B. Rohwedder, T. Wöhr, M. Mutter, *J. Am. Chem. Soc.*, 1997, **119**, 918-925.

15. (a) M. Narita, K. Ishikawa, H. Nakano, S. Isokawa, *Int. J. Pept. Protein Res.*, 1984, **24**, 14-24; (b) M. Narita, N. Ohkawa, S. Nagasawa, S. Isokawa, *Int. J. Pept. Protein Res.*, 1984, **24**, 129-134; (c) M. Narita, K. Ishikawa, J. Y. Chen, Y. Kim, *Int. J. Pept. Protein Res.*, 1984, **24**, 580-587.
16. T. Johnson, M. Quibell, D. Owen, R. C. Sheppard, *J. Chem. Soc., Chem. Commun.*, 1993, 369-372.
17. (a) M. Quibell, D. Owen, L. C. Packman, T. Johnson, *J. Chem. Soc., Chem. Commun.*, 1994, **20**, 2343-2344; (b) L. C. Packman, *Tetrahedron Lett.* 1995, **36**, 7523-7526.
18. (a) K. Wahlstrom, O. Planstedt, A. Unden, *Tetrahedron Lett.*, 2008, **49**, 3921-3924; (b) J.-S. Zheng, M. Yu, Y.-K. Qi, S. Tang, F. Shen, Z.-P. Wang, L. Xiao, L. Zhang, C.-L. Tian, L. Liu, *J. Am. Chem. Soc.*, 2014, **136**, 3695-3704.
19. E. C. B. Johnson, S. B. H. Kent, *Chem. Commun.*, 2006, 1557-1559.
20. (a) L. P. Miranda, W. D. F. Meutermans, M. L. Smythe, P. F. Alewood, *J. Org. Chem.*, 2000, **65**, 5460-5468; (b) D. A. Horton, G. T. Bourne, J. Coughlan, S. M. Kaiser, C. M. Jacobs, A. Jones, A. Ruhmann, J. Y. Turner, M. L. Smythe, *Org. Biomol. Chem.*, 2008, **6**, 1386-1395.
21. A.-B. M. Abdel-Aal, G. Papageorgiou, M. Quibell, J. Offer, *Chem. Commun.*, 2014, **50**, 8316-8319.
22. E. C. B. Johnson, S. B. H. Kent, *J. Am. Chem. Soc.*, 2006, **128**, 7140-7141.
23. X. Wu, P. K. Park, S. J. Danishefsky, *J. Am. Chem. Soc.*, 2011, **133**, 7700-7703.
24. M. Paradis-Bas, J. Tulla-Puche, F. Albericio, *Chem. Eur. J.*, 2014, **20**, 15031-15039.
25. D. S. Kemp, D. J. Kerkman, S. L. Leung, G. Hanson, *J. Org. Chem.*, 1981, **46**, 490-498.
26. R. Santini, M. C. Griffith, M. Qi, *Tetrahedron Lett.*, 1998, **39**, 8951-8954.
27. Our original tests showed that phenol is generally unnecessary in both cocktail B (TFA/phenol/H₂O/TIPS = 87/5/5/3) and cocktail K (TFA/phenol/H₂O/thioanisole/EDT = 82/5/5/5/3). Sulfoxide can be slowly reduced to sulfide in TFA in the presence of TIPS, but not thioanisole or EDT. However, thioanisole and EDT can indeed protect sulfide from air oxidation.
28. S. Thennarasu, C.-F. Liu, *Tetrahedron Lett.*, 2010, **51**, 3218-3220.
29. B. Bacsá, K. Horvati, S. Bosze, F. Andreae, C. O. Kappe, *J. Org. Chem.*, 2008, **73**, 7532-7542.
30. S. A. Palasek, Z. J. Cox, J. M. Collins, *J. Pept. Sci.*, 2007, **13**, 143-148.
31. R. Subiros-Funosas, R. Prohens, R. Barbas, A. El-Faham, F. Albericio, *Chem. Eur. J.*, 2009, **15**, 9394-9403.
32. (a) E. Nicolas, M. Vilaseca, E. Giralt, *Tetrahedron*, 1995, **51**, 5701-5710; (b) M. Vilaseca, E. Nicolas, F. Capdevila, E. Giralt, *Tetrahedron*, 1998, **54**, 15273-15286.
33. Z. Dekan, M. Mobli, M. W. Pennington, E. Fung, E. Nemeth, P. F. Alewood, *Angew. Chem. Int. Ed.*, 2014, **53**, 2931-2934.
34. L. Taboada, E. Nicolas, E. Giralt, *Tetrahedron Lett.*, 2001, **42**, 1891-1893.
35. (a) G.-M. Fang, Y.-M. Li, F. Shen, Y.-C. Huang, J.-B. Li, Y. Lin, H.-K. Cui, L. Liu, *Angew. Chem. Int. Ed.*, 2011, **50**, 7645-7649; (b) Y.-C. Huang, C.-C. Chen, S.-J. Li, S. Gao, J. Shi, Y.-M. Li, *Tetrahedron*, 2014, **70**, 2951-2955; (c) J. B. Li, Y. Y. Li, Q. Q. He, Y. M. Li, H. T. Li and L. Liu, *Org. Biomol. Chem.*, 2014, **12**, 5435; (d) Y. M. Li, M. Y. Yang, Y. C. Huang, Y. T. Li, P. R. Chen and L. Liu, *ACS Chem. Biol.* 2012, **7**, 1015-1022; (e) G. M. Fang, J. X. Wang and L. Liu, *Angew. Chem. Int. Ed.* 2012, **51**, 10347-10350; (f) Y. M. Li, Y. T. Li, M. Pan, X. Q. Kong, Y. C. Huang, Z. Y. Hong and L. Liu, *Angew. Chem. Int. Ed.* 2014, **53**, 2198-2202; (g) M. T. Weinstock, M. T. Jacobsen and M. S. Kay, *Proc. Natl. Acad. Sci. U.S.A.* 2014, **111**, 11679-11684; (h) M. Pan, Y. He, M. Wen, F. M. Wu, D. M. Sun, S. J. Li, L. H. Zhang, Y. M. Li and C. L. Tian, *Chem. Commun.* 2014, **50**, 5837-5839; (i) Y. Q. Chen, C. C. Chen, Y. He, M. Yu, L. Xu, C. L. Tian, Q. X. Guo, J. Shi, M. Zhang and Y. M. Li, *Tetrahedron Lett.* 2014, **55**, 2883; (j) J. S. Zheng, S. Tang, Y. K. Qi, Z. P. Wang and L. Liu, *Nat. Protoc.* 2013, **8**, 2483-2495;
36. L. Spasser, A. Brik, *Angew. Chem. Int. Ed.*, 2012, **51**, 6840-6862.