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Design, synthesis and biological evaluation of hydrogen sulfide releasing derivatives of 3-*n*-butylphthalide as potential antiplatelet and antithrombotic agents

Xiaoli Wang^{a,b,c,#}, Linna Wang^{b,d,#}, Xiao Sheng^{b,e,#}, Zhangjian Huang^{b,e}, Tingting Li^{b,d}, Ming Zhang^{a,b}, Jinyi Xu^{b,e}, Hui Ji^{b,d,*}, Jian Yin^{a,b,c,*} and Yihua Zhang^{b,e,*}

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Abstract

In the present study, a series of hydrogen sulfide (H₂S) releasing derivatives (**8a–g** and **9a–f**) of 3-*n*-butylphthalide (NBP) were designed, synthesized and biologically evaluated. The most promising compound **8e** significantly inhibited the adenosine diphosphate (ADP) and arachidonic acid (AA)-induced platelet aggregation *in vitro*, superior to NBP, ticlopidine hydrochloride and aspirin. Furthermore, **8e** could slowly produce moderate levels of H₂S *in vitro*, which could be beneficial for improving cardiovascular and cerebral circulation. Most importantly, **8e** protected against the collagen and adrenaline induced thrombosis in mice, and exhibited greater antithrombotic activity than NBP and aspirin in rats. Overall, **8e** could warrant further investigation for the treatment of thrombosis-related ischemic stroke.

1. Introduction

Ischemic stroke, a leading cause of death and long-term disability worldwide, is a major socioeconomic burden due to the lack of a suitable therapy.¹ Ischemic stroke is characterized by thrombosis initiated by platelet abnormal function, which results in insufficient cerebral blood flow, and further leads to ischemic brain damage.²⁻⁵ Current drug therapies for prevention and treatment of ischemic stroke are primarily dependent on high affinity and single-target drugs. However, a drug directed against a

single molecular target may not be effective in treatment of ischemic stroke because of its multi-factorial pathology. Therefore, there is a significantly unmet medical need for new drugs with multiple actions to combat ischemic stroke.

The racemic 3-*n*-butylphthalide (NBP) was approved by the State Food and Drug Administration (SFDA) of China in 2002 as a new anti-ischemic stroke drug, which has beneficial effects on stroke through multiple actions,⁶⁻⁸ including inhibiting platelet aggregation and thrombosis, improving microcirculation, reducing the area of cerebral infarct and decreasing oxidative damage, etc. Despite these advantages, the clinic application of NBP has been limited due to its moderate potency.^{9,10} To achieve the ideal anti-ischemic stroke drugs, a wide range work of structural modification and related research on NBP have been carried out. Previous studies have demonstrated that the potassium salt of 2-(1-hydroxypropyl)-benzoate (HPBA), a ring-opening derivative of NBP, has an improved aqueous solubility and comparable potency to NBP.¹¹ Recently, we also reported several ring-opening derivatives of NBP as potent anti-ischemic stroke agents.¹²⁻¹⁴

Hydrogen sulfide (H₂S) is considered as a new member of gasotransmitter family, following nitric oxide (NO) and carbon monoxide (CO). Recent studies have revealed that H₂S has a positive effect on cerebral ischemia reperfusion injury. Indeed, augmentation of H₂S production can induce vasodilatation, inhibit platelet aggregation, improve CBF, promote angiogenesis, and protect neurons against oxidative stress and free radicals, benefiting patients with ischemic stroke.¹⁵⁻¹⁸ And a number of H₂S-releasing compounds have been developed and are already undergoing intensive investigation.¹⁹⁻²¹

In this study, we hypothesized that H₂S-releasing derivatives of NBP could display synergistic effects exerted by both NBP and H₂S on platelet aggregation and thrombosis. Thereby, a series of hybrids from HPBA and H₂S were designed, synthesized and evaluated as potential antiplatelet and antithrombotic agents *in vitro* and *in vivo*.

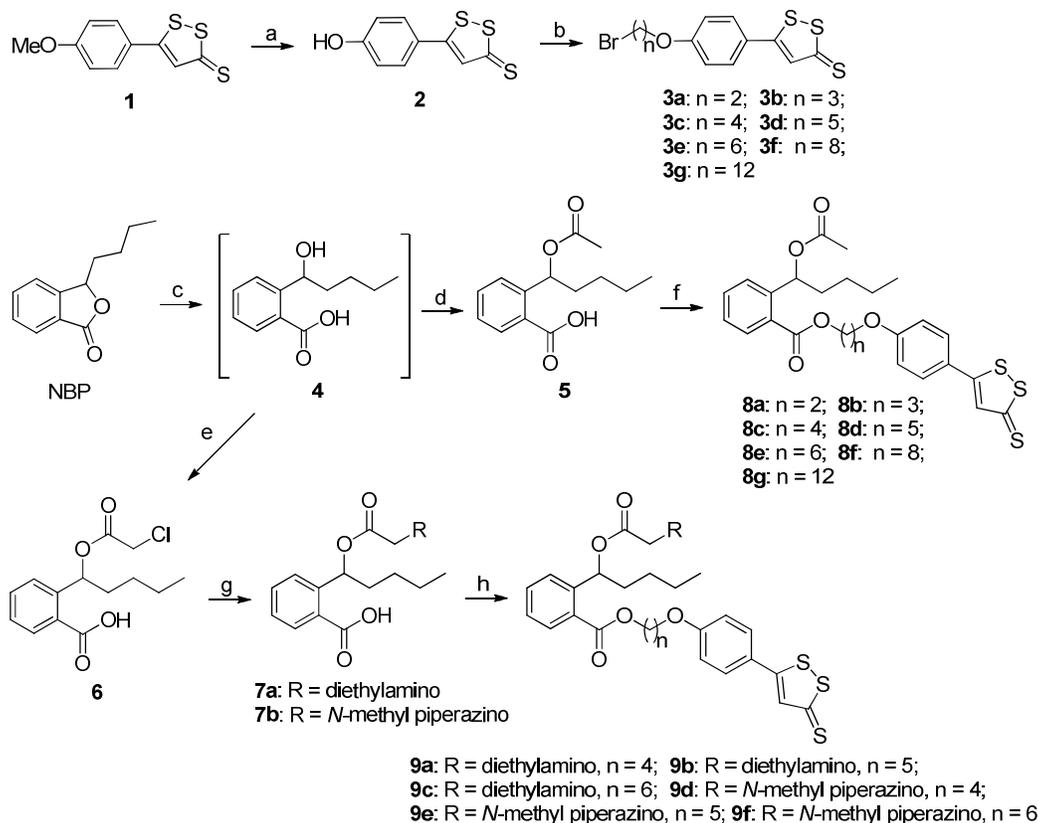
2. Results

2.1. Strategy for the design of H₂S-NBP

It has been demonstrated that conjugation of an H₂S-donor moiety with a “native” molecule enhances therapeutic effect and/or reduces adverse effect of the native molecule.²²⁻²⁴ In view of that ADTOH, 5-(4-hydroxyphenyl)-3*H*-1,2-dithiole-3-thione, a known H₂S-donor, is able to release H₂S *in vitro* and *in vivo*,²⁰ we designed a novel class of compounds through connecting ADTOH to HPBA with different linkers. In addition, amines, such as diethylamine or *N*-methyl piperazine, were introduced to the side chain of HPBA as a substituted acetate linkage, respectively. We expected that the ester bonds of H₂S-NBP would be cleaved *in vivo* by esterases to release HPBA and H₂S, the former would subsequently undergo ring closing to generate NBP; and that both NBP and H₂S would synergistically inhibit platelet aggregation and thrombosis.

2.2. Chemistry

The synthesis of the target compounds **8a–g** and **9a–f** was illustrated in Scheme 1. The H₂S-donor ADTOH (compound **2**) was prepared as described previously²⁶ and converted to the corresponding bromo-substituted ethers **3a–g** by using K₂CO₃ in acetone. NBP was subjected to saponification and sequential acidification to form the ring-opening acid **4**. Treatment of **4** with acetyl chloride or chloroacetyl chloride gave the acylates **5** or **6**. Compound **5** was condensed with **3a–g** to generate the target compounds **8a–g**. Treatment of **6** with diethylamine or *N*-methyl piperazine led to ester **7a** or **7b**, which was then reacted with **3c–d** to provide the target compounds **9a–f**. All of the new compounds were purified by column chromatography and characterized by IR, ESI-MS, ¹H NMR, ¹³C NMR, and HRMS. The individual compounds with chemical purity of > 95% (determined by HPLC analysis) were used for subsequent experiments.



Scheme 1. Synthesis of **8a–g** and **9a–f**. *Reagents and conditions:* (a) pyridine hydrochloride, 215 °C, 0.5 h. (b) Br(CH₂)_nBr, K₂CO₃, acetone, reflux, 6–8 h. (c) (i) 2 M NaOH, CH₃OH–H₂O, 50 °C, 0.5 h; (ii) 5% HCl, –10 to 0 °C; (d) CH₃COCl, Et₃N, DMAP, CH₂Cl₂, –10 °C, 5 h; (e) ClCH₂COCl, Et₃N, DMAP, CH₂Cl₂, –10 °C, 5 h; (f) **3a–g**, K₂CO₃, acetone, reflux, 6–10 h; (g) diethylamine or *N*-methyl piperazine, K₂CO₃, acetone, reflux, 6–8 h; (h) **3c–e**, K₂CO₃, acetone, reflux, 4–8 h.

2.2. Antiplatelet aggregation effects *in vitro*.

The individual compounds were evaluated for inhibition of platelet aggregation in rabbit platelet rich plasma (PRP) in response to the adenosine diphosphate (ADP) and arachidonic acid (AA) by using Born's turbidimetric method.²⁷ Ticlopidine hydrochloride (Ticlid) and aspirin, as inhibitors of the ADP- and AA-induced platelet aggregation,^{28,29} respectively, were employed as positive controls, and NBP as reference compound. The inhibitory activity of individual compounds at five different concentrations against the ADP (10 μM) and AA (1.0 mM) induced platelet aggregation. Data were calculated and expressed as the IC₅₀ values.

As shown in Table 1, all the target compounds presented better inhibitory effects than NBP both against the ADP- and AA-induced platelet aggregation. The compounds with 6-carbon alkanes (**8e**, **9c**) displayed the best inhibitory activity. Furthermore, the compounds containing hydrogen atom (**8c–e**) exhibited stronger inhibitory activity than that bearing diethylamino and *N*-methyl piperazino moiety in side chain (**9a–f**). Notably, **8e** (R = H, *n* = 6) was the most potent, whose IC₅₀ value (0.140 mM) on the ADP-induced platelet aggregation was significantly less than that of NBP (0.742 mM) and Ticlid (0.358 mM), and the IC₅₀ value (0.091 mM) on the AA-induced platelet aggregation was also less than that of NBP (0.583 mM) and aspirin (0.136 mM). These data indicated that **8e** was a potent inhibitor of both the ADP- and AA-induced platelet aggregation. Therefore, **8e** was chosen for the following investigations.

Table 1. The IC₅₀ values of **8a–g** and **9a–f** against platelet aggregation *in vitro*.

Compd.	IC ₅₀ (mM)	
	ADP (10 μM)	AA (0.33 mM)
Control	--	--
Ticlid ^a	0.36 ± 0.04	--
Aspirin ^b	--	0.14 ± 0.02
NBP	0.74 ± 0.20	0.58 ± 0.06
8a	0.23 ± 0.09	0.15 ± 0.05
8b	0.20 ± 0.11	0.14 ± 0.06
8c	0.18 ± 0.10	0.14 ± 0.08
8d	0.17 ± 0.13	0.15 ± 0.07
8e	0.14 ± 0.06	0.09 ± 0.01
8f	0.19 ± 0.14	0.16 ± 0.05
8g	0.24 ± 0.12	0.20 ± 0.06
9a	0.23 ± 0.14	0.18 ± 0.03
9b	0.21 ± 0.13	0.17 ± 0.04
9c	0.19 ± 0.11	0.15 ± 0.06

9d	0.71 ± 0.19	0.37 ± 0.10
9e	0.52 ± 0.17	0.23 ± 0.09
9f	0.49 ± 0.14	0.23 ± 0.08

^aTiclopid is an inhibitor of the ADP-induced platelet aggregation; ^bAspirin is an inhibitor of the AA-induced platelet aggregation. IC₅₀ values (a dose achieved 50% inhibition of platelet aggregation) are expressed as mean ± SD (*n* = 6) and analyzed by one-way analysis of variance (ANOVA) followed by *post hoc* Tukey test.

In order to examine whether each moiety (**2**, **3e** and **5**) of **8e** contributes to the inhibitory activity of this compound, we further investigated their corresponding effects on the ADP-induced platelet aggregation *in vitro*. As shown in Fig. 1, compounds **2**, **3e** and **5** inhibited the ADP-induced platelet aggregation by 28.04%, 30.05% and 26.80%, respectively, but each of them was substantially less effective than **8e** (62.90%), suggesting that these moieties may have synergistic effects on the inhibition of **8e** against the ADP-induced platelet aggregation.

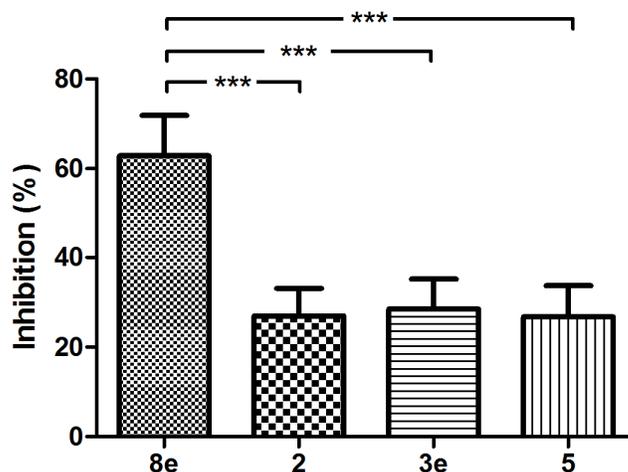


Fig. 1 Inhibition of **8e**, **2**, **3e** and **5** on the ADP-induced platelet aggregation *in vitro*. Rabbit platelet suspensions were preincubated with testing compound (200 μM) at 37 °C for 5 min and exposed to 10 μM of ADP, followed by continually monitoring. Rabbit platelet suspensions that had been treated with vehicle and exposed to ADP were used as positive controls. Data are expressed as mean ± SD of each group (*n* = 6) for two separate experiments. ****P* < 0.001 by one-way ANOVA followed by *post hoc* Tukey test.

To confirm the synergistic effects between these moieties, we then tested the moieties of **8e** in combination. Obviously **8e** had a better antiplatelet activity than NBP together with **2** ($*P < 0.05$, Fig. 2) which support our hypothesis that **8e** affect the platelet aggregation greater than the combination of its mother compound-NBP and H₂S releasing moiety-ADTOH (compound **2**). Although the combination of **3e** and **5** showed weaker inhibition on platelet aggregation than **8e**, no statistical significance was found between them.

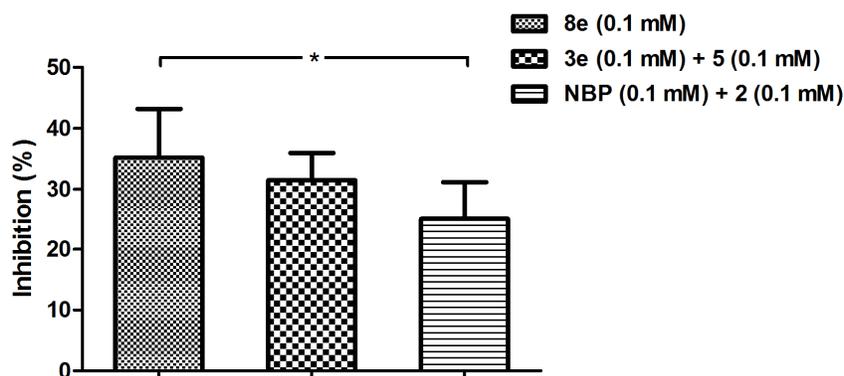


Fig. 2 Inhibition of **8e** and combinations of its moieties on the ADP-induced platelet aggregation *in vitro*. Data are expressed as mean \pm SD of each group ($n = 6$) for two separate experiments. $*P < 0.05$ by one-way ANOVA followed by *post hoc* Tukey test.

2.3. Antithrombotic Activity of **8e** in mice.

We then evaluated the antithrombotic activity of **8e** on a mouse model of thromboembolism, that is, an acute vascular occlusion initiated by intravascular platelet induced aggregation through infusion of a mixture of collagen and adrenaline into the jugular vein.³² Male ICR mice were randomized and treated orally with vehicle, NBP (160 mg/kg), aspirin (160 mg/kg), or **8e** (160 mg/kg and 470 mg/kg, which are equal mass to and equal molar to that of NBP, respectively) for seven days. Two hours after the last administration, individual mice were injected intravenously with a mixture of collagen (0.21 mg/kg) and adrenaline (44.5 μ g/kg) to induce acute systemic vascular thromboembolism. As shown in Fig. 3, treatment with **8e** (470 mg/kg) significantly reduced the onset of hemiplegia and death in mice, and the

percentage of survivors (66.7%) was greater than that of aspirin (60.0%) and NBP (53.3%), while the percentage (60.0%) by treatment with **8e** (160 mg/kg) was slightly greater than that of NBP and comparable to that of aspirin in this model at the same dosage. These results clearly demonstrated that **8e** efficiently inhibited acute systemic intravascular thrombosis in mice leading to protective effects.

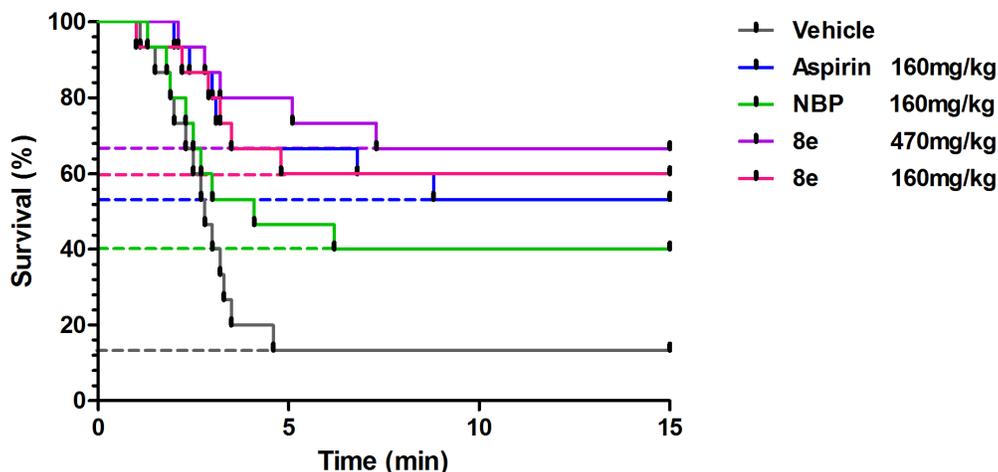


Fig. 3 The survival of mice after injection of 0.21 mg/kg collagen and 44.5 µg/kg adrenaline and subsequent oral administration of **8e**, NBP and aspirin, respectively, at the indicated doses and time points, and the results were expressed as the percentage of mice alive ($n = 15$) as a function of time.

2.4. Antithrombotic Activity of **8e** in Rats.

Parallely, we investigated the antithrombotic activity of **8e** in a rat arterio-venous (A-V) shunt model mimicking arterial thrombosis.³³ SD rats were randomized and treated intragastrically (i.g.) with normal saline (sham group), NBP, aspirin, and **8e**. After a 7-day oral administration of individual compounds, the wet and dry thrombus weights were measured (Fig. 4). All four treatment groups showed significant difference from the sham group on wet weight of thrombus. Treatment with **8e** at dose of 80 or 235 mg/kg, which is equal mass to or equal molar to that of NBP, reduced wet weight by 27.87% and 37.16%, respectively, more effective than treatment with NBP (15.02%), and the potency of the higher dose group was similar to that of aspirin

(36.74%). Similarly, four treatment groups shared significantly difference from sham group on dry weight of thrombus. **8e** at 235 mg/kg reduced the dry weight of thrombus by 32.58%, which was more potent than both NBP (18.72%) and aspirin (30.60%). Treatment with **8e** at 80 mg/kg also effectively reduced the thrombus dry weight by 26.34%, which was higher than NBP but lower than aspirin. These results were consistent with our previous research, and clearly indicated that **8e** was a potent thrombosis inhibitor *in vivo*.

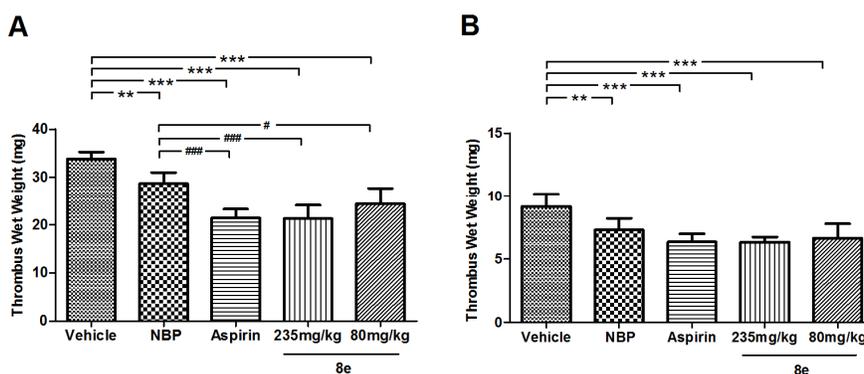


Fig. 4 Effects of **8e**, NBP and aspirin on thrombus wet weights (A) and dry weights (B) in rats. SD rats were treated with the indicated doses of each drug by gavage daily for seven days. Two hours after the last treatment, the rats were subjected to the A-V shunt using a silk thread for 15 min. The wet and dry weights were measured. Data are expressed as mean \pm SD of each group ($n = 12$) and analyzed by ANOVA and post hoc Tukey test. *** $P < 0.001$, ** $P < 0.01$ vs the control group, ### $P < 0.001$, # $P < 0.05$ vs the NBP group.

2.5. Assay of hydrogen sulfide release *in vitro*

The target compounds were designed to be NBP derivatives, which could release H_2S . In order to show that this was indeed the case, the levels of H_2S produced by **8e** were measured in phosphate buffer and rat liver homogenate³⁴, respectively. As shown in Fig. 5, little but still detectable H_2S release was observed when **8e** was incubated with phosphate buffer. However, in liver homogenate, **8e** could produce moderate levels of H_2S peaking at 60 min which remained slightly elevated for a further 30 min, suggesting that **8e** could slowly release H_2S in liver homogenate, which would be expected to increase the duration of intracellular H_2S exposure

leading to the enhanced potency of **8e** relative to NBP. Based on the chemical structure of **8e**, we speculated that it could be firstly metabolized by esterases to release HPBA, which would subsequently undergo ring closure to generate NBP, and H₂S donor ADTOH. Incubation of **8e** in buffer released much smaller quantities of H₂S also implied that the gas released from the parent molecule mainly as a result of metabolic events.

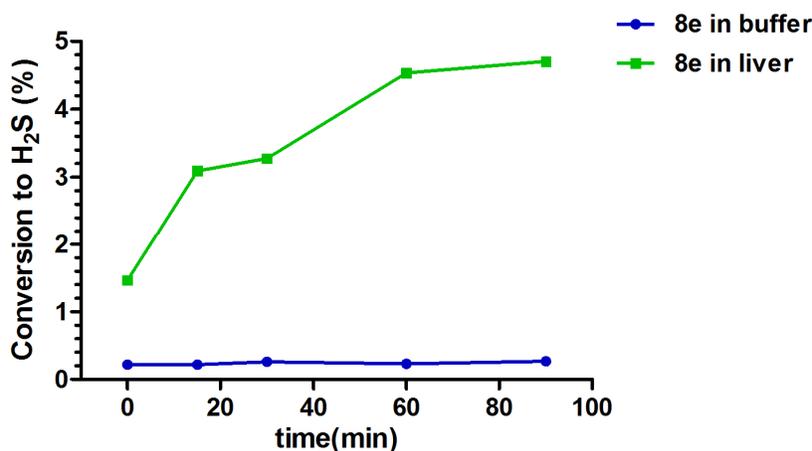


Fig. 5 Release of H₂S from **8e** *in vitro*. **8e** (100 μM) was incubated (37 °C) for 90 min with fresh rat liver homogenate or buffer *in vitro*. H₂S concentration was determined. Data are expressed as accumulate % conversion of **8e** to H₂S at each time point, $n = 6$.

3. Discussion and conclusions

As far as structure-activity relationships are concerned, in general, extension of the carbon chain (CH₂)_{*n*} connecting to H₂S releasing moiety ($n = 2-6$), the platelet aggregation inhibitory activity of the compounds was enhanced, and the activity was declined while $n = 8$. Furthermore, the compounds containing hydrogen atom linking to acetyl group in side-chain exhibited stronger inhibitory activity than that bearing diethylamino and *N*-methylpiperazino moiety in side chain. As a matter of fact, **8e** exhibited the strongest inhibitory activity among all compounds. Interestingly, all of the structural moieties of **8e** substantially inhibited the ADP-induced platelet aggregation but much less than **8e**, suggesting that they may act synergistically.

Moreover, administration of **8e** alone had better antiplatelet aggregation activity than combinations from its corresponding moieties. However, the precise SAR of the target compounds remain further investigated.

In summary, we designed and synthesized a series of novel H₂S releasing derivatives of NBP, and found that compound **8e** was the most potent in inhibiting the ADP- and AA-induced platelet aggregation *in vitro*, superior to NBP, Ticlid and aspirin. Furthermore, **8e** could produce moderate levels of H₂S *in vitro*, which could be beneficial for improving cardiovascular and cerebral circulation. Most importantly, **8e** protected against the collagen and adrenaline induced thrombosis in mice, and exhibited greater antithrombotic activity in rats. Therefore, our findings may aid in the design of new therapeutic reagents for the treatment of thrombosis-related ischemic stroke and other vascular diseases.

4. Experimental section

4.1. Chemistry

General. Melting points are uncorrected and were measured in open capillary tubes using a Gallenkamp melting point apparatus. ¹H and ¹³C NMR spectral data were obtained from a Bruker Avance 300 MHz spectrometer at 300 K using TMS as an internal standard. MS spectra were recorded on a Mariner Mass Spectrum (ESI). Analytical and preparative TLC were performed on silica gel GF/UV 254, and the chromatograms were conducted on silica gel (200–300 mesh) and visualized under UV light at 254 and 365 nm. The purities of the compounds were characterized by HPLC analysis (LC-10A HPLC system consisting of LC-10ATvp pumps and SPD-10Avp UV detector) and HRMS (Agilent technologies LC/MSD TOF). The target compounds **8a–g** and **9a–f** with chemical purity of > 95% (see the ESI†).

All experiments were approved and performed in accordance with National Institutes of Health Guide for Care and Use of Laboratory Animals (Publication No. 85-23, revised 1985), and all animal experimental protocols were approved by the Animal Research Protection Committee of China Pharmaceutical University, Nanjing, China.

5-(4-Hydroxyphenyl)-3H-1,2-dithiole-3-thione (2). Pyridine hydrochloride (68.0 g, 590 mmol) was added to **1** (25.0 g, 100 mmol) in a dry flask, mixed and then heated to melt at 215 °C under argon protection for 0.5 h. After being cooled to 100 °C, warmed water (200 mL) was added and hot-filtered. The cake was placed in a beaker, and 10%NaOH (300 mL) was added. This was stirred for 4 h, filtered, the cake dissolved in water (3 L), then adjusted to pH 2 with concentrated hydrochloride. The red precipitation was filtered and washed to neutral using water, then dried in a vacuum desiccator to yield a yellow solid (18.0 g, 76.5%), mp 172–174 °C. (found: C, 47.78; H, 2.67; O, 7.08; S, 42.53. Calcd for C₉H₆OS₃: C, 47.76; H, 2.67; O, 7.07; S, 42.50%); ¹H NMR (CDCl₃, 300 Hz, δ): 10.47 (brs, 1H, OH), 7.84 (d, 2H, *J* = 8.8 Hz, ArH), 7.70 (s, 1H, =CH), 6.88 (d, 2H, *J* = 8.8 Hz, ArH).

General procedure for the preparation of compounds 3a–g. To the mixture of **2** (0.226 g, 1.0 mmol), appropriate dibromoalkanes (3.0 mmol), and K₂CO₃ (0.690 g, 5.0 mmol), acetone (20mL) was added under constant stirring and argon protection. It was then heated to 65 °C and refluxed for 6–8 h. After completion of the reaction, (monitored by TLC), the reaction mixture was filtrated, the filtrate was concentrated under reduced pressure. The resulting residue was purified by flash chromatography (PE–acetone 10 : 1 v/v) to give the title compounds (70–83%).

5-(4-(2-Bromoethoxy)phenyl)-3H-1,2-dithiole-3-thione (3a). The title compound was obtained as a reddish brown solid (80%), mp 134–136 °C. IR (cm⁻¹, KBr): ν_{\max} 595, 819, 1184, 1400, 1486, 1599. ¹H NMR (300 Hz, CDCl₃): δ 3.68 (t, 2H, *J* = 6.0 Hz, CH₂Br), 4.36 (t, 2H, *J* = 6.0 Hz, ArOCH₂), 7.00 (d, 2H, *J* = 8.8 Hz, ArH), 7.40 (s, 1H, =CH), 7.63 (d, 2H, *J* = 8.8 Hz, ArH).

5-(4-(3-Bromopropoxy)phenyl)-3H-1,2-dithiole-3-thione (3b). The title compound was obtained as a reddish brown solid (78%), mp 131–133 °C. IR (cm⁻¹, KBr): ν_{\max} 591, 815, 1185, 1400, 1485, 1600. ¹H NMR (300 Hz, CDCl₃): δ 2.32–2.40 (m, 2H, CH₂CH₂Br), 3.62 (t, 2H, *J* = 6.3 Hz, CH₂Br), 4.12 (t, 2H, *J* = 5.8 Hz, ArOCH₂), 7.00 (d, 2H, *J* = 8.7 Hz, ArH), 7.40 (s, 1H, =CH), 7.63 (d, 2H, *J* = 8.7 Hz, ArH).

5-(4-(4-Bromobutoxy)phenyl)-3H-1,2-dithiole-3-thione (3c). The title compound

was obtained as a reddish brown solid (81%), mp 129–131 °C. IR (cm⁻¹, KBr): ν_{\max} 590, 823, 1189, 1400, 1489, 1600. ¹H NMR (300 Hz, CDCl₃): δ 1.95–2.43 (m, 4H, CH₂CH₂CH₂Br), 3.60 (t, 2H, J = 6.4 Hz, CH₂Br), 4.07 (t, 2H, J = 6.0 Hz, ArOCH₂), 6.97 (d, 2H, J = 8.8 Hz, ArH), 7.40 (s, 1H, =CH), 7.62 (d, 2H, J = 8.8 Hz, ArH).

5-(4-((5-Bromopentyl)oxy)phenyl)-3H-1,2-dithiole-3-thione (3d). The title compound was obtained as a reddish brown solid (75%), mp 125–127 °C. IR (cm⁻¹, KBr): ν_{\max} 592, 826, 1184, 1400, 1482, 1600. ¹H NMR (300 Hz, CDCl₃): δ 1.60–1.70 (m, 2H, CH₂CH₂CH₂Br), 1.81–2.45 (m, 4H, CH₂CH₂CH₂CH₂Br), 3.46 (t, 2H, J = 6.6 Hz, CH₂Br), 4.04 (t, 2H, J = 6.2 Hz, ArOCH₂), 6.96 (d, 2H, J = 8.8 Hz, ArH), 7.40 (s, 1H, =CH), 7.61 (d, 2H, J = 8.8 Hz, ArH).

5-(4-((6-Bromohexyl)oxy)phenyl)-3H-1,2-dithiole-3-thione (3e). The title compound was obtained as a reddish brown solid (83%), mp 122–124 °C. IR (cm⁻¹, KBr): ν_{\max} 596, 827, 1189, 1400, 1482, 1599. ¹H NMR (300 Hz, CDCl₃): δ 1.25–1.33 (m, 2H, CH₂(CH₂)₂Br), 1.60–1.70 (m, 2H, ArO(CH₂)₂CH₂), 1.81–2.45 (m, 4H, ArOCH₂CH₂, CH₂CH₂Br), 3.44 (t, 2H, J = 6.7 Hz, CH₂Br), 4.02 (t, 2H, J = 6.3 Hz, ArOCH₂), 6.96 (d, 2H, J = 8.6 Hz, ArH), 7.40 (s, 1H, =CH), 7.60 (d, 2H, J = 8.6 Hz, ArH).

5-(4-((8-Bromooctyl)oxy)phenyl)-3H-1,2-dithiole-3-thione (3f). The title compound was obtained as a reddish brown solid (76%), mp 118–120 °C. IR (cm⁻¹, KBr): ν_{\max} 596, 833, 1184, 1400, 1483, 1598. ¹H NMR (300 Hz, CDCl₃): δ 1.26–1.38 (m, 4H, CH₂CH₂(CH₂)₃Br), 1.56–1.72 (m, 4H, ArO(CH₂)₂CH₂, CH₂(CH₂)₂Br), 1.81–2.45 (m, 4H, ArOCH₂CH₂, CH₂CH₂Br), 3.42 (t, 2H, J = 6.7 Hz, CH₂Br), 4.02 (t, 2H, J = 6.4 Hz, ArOCH₂), 6.96 (d, 2H, J = 8.6 Hz, ArH), 7.40 (s, 1H, =CH), 7.60 (d, 2H, J = 8.6 Hz, ArH).

5-(4-((12-Bromododecyl)oxy)phenyl)-3H-1,2-dithiole-3-thione (3g). The title compound was obtained as a reddish brown solid (75%), mp 113–115 °C. IR (cm⁻¹, KBr): ν_{\max} 597, 833, 1186, 1401, 1483, 1598. ¹H NMR (300 Hz, CDCl₃): δ 1.26–1.38 (m, 12H, 6×CH₂(CH₂)₃Br), 1.43–1.62 (m, 4H, ArO(CH₂)₂CH₂, CH₂(CH₂)₂Br), 1.78–2.09 (m, 4H, ArOCH₂CH₂, CH₂CH₂Br), 3.41 (t, 2H, J = 6.8 Hz, CH₂Br), 4.02 (t, 2H, J = 6.5 Hz, ArOCH₂), 6.96 (d, 2H, J = 8.7 Hz, ArH), 7.40 (s, 1H, =CH), 7.60 (d,

2H, $J = 8.7$ Hz, ArH).

(±)-2-(1-Hydroxypentyl)benzoic acid (4). To a stirred solution of NBP (1.90 g, 10.00 mmol) in $\text{CH}_3\text{OH-H}_2\text{O}$ (20 mL, 1 : 1 v/v) was added NaOH (0.80 g, 20.00 mmol). The reaction mixture was heated at 50 °C for 0.5 h. The solvent was removed under reduced pressure and dissolved in water (20 mL), followed by acidification with 5% HCl to pH 3–4 at –10 to 0 °C. The mixture was extracted with cold Et_2O (10 mL \times 3) and quickly used for the next step without any purification.

(±)-2-(1-Acetoxy)pentylbenzoic acid (5). A solution of acetyl chloride (1.06 mL, 15.00 mmol) in dry CH_2Cl_2 (10 mL) was added dropwise to a mixture of **4** (1.04 g, 5.00 mmol), Et_3N (2.09 mL, 15.00 mmol) and DMAP (0.50 mmol) in CH_2Cl_2 (80 mL) at –10 °C and the solution was left stirring at –10 °C for 5 h. The mixture was acidified with 1 M HCl to pH 2 and stirred for 1 h at room temperature. The organic layer was washed with water, dried, filtered, and evaporated to dryness. The residue was recrystallized from n-hexane to obtain **5** as a white crystal (0.85 g, 68%, over two steps). mp 65–66 °C. MS (ESI): m/z 249.1 [M – H]. IR (cm^{-1} , KBr): ν_{max} 1412, 1691, 1734, 2958, 3450. ^1H NMR (300 Hz, CDCl_3): δ 0.93 (t, 3H, CH_3 , $J = 8.5$ Hz), 1.37–1.42 (m, 4H, $2\times\text{CH}_2$), 1.88–1.91 (m, 2H, CH_2), 2.13–2.33 (m, 3H, COCH_3), 6.61–6.72 (m, 1H, OCHCH_2), 7.37–7.40 (m, 1H, ArH), 7.56–7.62 (m, 2H, ArH), 8.05 (d, 1H, ArH, $J = 8.1$ Hz), 10.98 (brs, 1H, COOH). ^{13}C NMR (75 Hz, CDCl_3): δ 172.0, 166.5, 140.8, 133.1, 130.3, 130.0, 127.1, 125.7, 74.8, 41.0, 36.3, 27.8, 22.4, 13.8.

(±)-2-[1-(2-Chloroacetoxy)pentyl]benzoic acid (6). Compound **6** was synthesized in the same manner as compound **5**. Starting from compound **4** (1.04 g, 5.00 mmol) and 2-chloroacetyl chloride (2.08 mL, 15.00 mmol), compound **6** was obtained as a white crystal (0.93 g, 65%, over two steps). mp 67–68 °C. MS (ESI): m/z 283 [M – H]. IR (cm^{-1} , KBr): ν_{max} 1412, 1691, 1734, 2958, 3450. ^1H NMR (300 Hz, CDCl_3): δ 0.93 (t, 3H, CH_3 , $J = 4.2$ Hz), 1.37–1.42 (m, 4H, $2\times\text{CH}_2$), 1.88–1.91 (m, 2H, CH_2), 4.11–4.32 (m, 2H, COCH_2Cl), 6.71–6.78 (m, 1H, OCHCH_2), 7.36–7.42 (m, 1H, ArH), 7.56–7.62 (m, 2H, ArH), 8.08 (d, 1H, ArH, $J = 8.1$ Hz), 10.89 (brs, 1H, COOH). ^{13}C NMR (75 Hz, CDCl_3): δ 172.0, 166.5, 140.8, 133.1, 130.3, 130.0, 127.1, 125.7, 74.8, 41.0, 36.3, 27.8, 22.4, 13.8.

(±)-2-(1-(2-(Diethylamino)acetoxy)pentyl)benzoic acid (7a). To a solution of **6** (284 mg, 1.0 mmol) and K_2CO_3 (276 mg, 2.0 mmol) in acetone (20 mL) was added diethylamine (0.2 mL, 2.0 mmol), and the solution was heated to 65 °C and refluxed for 4 h. The solution was then filtered, and the filtrate was reconstituted in EtOAc (20 mL) and solvent removed under reduced pressure. The crude was purified by flash chromatography ($CH_2Cl_2/MeOH = 100/1-50/1$, v/v), to give **7a** as a light yellow oil (231 mg, 72%). MS (ESI): m/z 322 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.83 (t, 3H, CH_3 , $J = 7.5$ Hz), 1.02 (t, 6H, $2 \times NCH_2CH_3$, $J = 7.0$ Hz), 1.29–1.37 (m, 4H, $2 \times CH_2CH_3$), 1.87–1.88 (m, 2H, $CHCH_2$), 2.94 (q, 4H, $2 \times CH_2$, $J = 7.0$ Hz), 3.62–3.76 (m, 2H, $COCH_2N$), 6.69 (m, 1H, CH), 7.21–7.26 (m, 1H, ArH), 7.34–7.43 (m, 2H, ArH), 7.80 (d, 1H, ArH, $J = 7.5$ Hz), 10.67 (brs, 1H, COOH).

(±)-2-(1-(2-(4-Methylpiperazin-1-yl)acetoxy)pentyl)benzoic acid (7b). Compound **7b** was synthesized in the same manner as compound **7a**. Starting from compound **6** (284 mg, 1.0 mmol) and *N*-methyl piperazine (0.22 mL, 2.0 mmol), compound **7b** was obtained as a light yellow oil (271 mg, 78%). MS (ESI): m/z 349 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.83 (t, 3H, CH_3 , $J = 7.5$ Hz), 1.29–1.37 (m, 4H, $2 \times CH_2CH_3$), 1.87–1.88 (m, 2H, $CHCH_2$), 2.26 (t, 3H, NCH_3 , $J = 7.0$ Hz), 2.45 (m, 8H, $4 \times NCH_2$), 3.62–3.76 (m, 2H, $COCH_2N$), 6.69 (m, 1H, CH), 7.21–7.26 (m, 1H, ArH), 7.34–7.43 (m, 2H, ArH), 7.80 (d, 1H, ArH, $J = 7.5$ Hz), 10.67 (brs, 1H, COOH).

General procedure for the preparation of the target compounds 8a–g. To a solution of **5** (50 mg, 0.2 mmol) and K_2CO_3 (55 mg, 0.4 mmol) in acetone (10 mL) was added **3a–g** (0.2 mmol), and the solution was heated to 65 °C and refluxed for 6–8 h. The solution was then filtered, and the filtrate was reconstituted in EtOAc (20 mL) and solvent removed under reduced pressure. The crude compounds were purified by flash chromatography ($PE/EtOAc = 10/1-1:1$, v/v), to give the title compounds (67–78%).

(±)-2-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)ethyl 2-(1-acetoxypentyl)benzoate (8a). The title compound was obtained as a reddish brown solid (75%), mp 86–88 °C. MS (ESI): m/z 503.1 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H,

CH₃, $J = 6.9$ Hz), 1.26–1.48 (m, 4H, CH₂CH₂CH₃), 1.74–1.83 (m, 2H, CHCH₂), 2.07 (s, 3H, OOCCH₃), 4.10 (t, 2H, ArOCH₂, $J = 6.2$ Hz), 4.41 (t, 2H, ArCOOCH₂, $J = 6.2$ Hz), 6.53 (dd, 1H, CH, $J = 4.7$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.6$ Hz), 7.27–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.52 (m, 2H, ArH), 7.58 (d, 2H, ArH, $J = 8.6$ Hz), 7.87 (d, 1H, ArH, $J = 7.8$ Hz). ¹³C NMR (75 Hz, CDCl₃): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 36.3, 28.1, 25.4, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for C₂₅H₂₆O₅S₃, [M + H]⁺ 503.0942; found 503.1018.

(±)-3-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)propyl 2-(1-acetoxypentyl)benzoate (8b). The title compound was obtained as a reddish brown solid (78%), mp 83–85 °C. MS (ESI): m/z 517.1 [M + H]⁺. ¹H NMR (300 Hz, CDCl₃): δ 0.89 (t, 3H, CH₃, $J = 6.9$ Hz), 1.26–1.48 (m, 4H, CH₂CH₂CH₃), 1.74–1.83 (m, 2H, CHCH₂), 1.98–2.04 (m, 2H, ArCOOCH₂CH₂), 2.07 (s, 3H, OOCCH₃), 4.10 (t, 2H, ArOCH₂, $J = 6.2$ Hz), 4.41 (t, 2H, ArCOOCH₂, $J = 6.2$ Hz), 6.53 (dd, 1H, CH, $J = 4.7$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.6$ Hz), 7.27–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.52 (m, 2H, ArH), 7.58 (d, 2H, ArH, $J = 8.6$ Hz), 7.87 (d, 1H, ArH, $J = 7.8$ Hz). ¹³C NMR (75 Hz, CDCl₃): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 36.3, 28.1, 25.4, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for C₂₆H₂₈O₅S₃, [M + H]⁺ 517.1099; found 517.1297.

(±)-4-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)butyl 2-(1-acetoxypentyl)benzoate (8c). The title compound was obtained as a reddish brown oil (72%), MS (ESI): m/z 531.2 [M + H]⁺. ¹H NMR (300 Hz, CDCl₃): δ 0.89 (t, 3H, CH₃, $J = 6.9$ Hz), 1.26–1.48 (m, 4H, CH₂CH₂CH₃), 1.74–1.83 (m, 2H, CHCH₂), 2.00–2.06 (m, 4H, ArCOOCH₂(CH₂)₂), 2.07 (s, 3H, OOCCH₃), 4.10 (t, 2H, ArOCH₂, $J = 6.2$ Hz), 4.41 (t, 2H, ArCOOCH₂, $J = 6.2$ Hz), 6.53 (dd, 1H, CH, $J = 4.7$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.6$ Hz), 7.27–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.52 (m, 2H, ArH), 7.58 (d, 2H, ArH, $J = 8.6$ Hz), 7.87 (d, 1H, ArH, $J = 7.8$ Hz). ¹³C NMR (75 Hz, CDCl₃): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 36.3, 28.1, 25.8, 25.4, 22.5,

21.2, 14.0. HRMS (ESI): m/z calcd for $C_{27}H_{30}O_5S_3$, $[M + H]^+$ 531.1255; found 531.1375.

(±)-5-(4-(3-Thioxo-3*H*-1,2-dithiol-5-yl)phenoxy)pentyl 2-(1-acetoxypentyl) benzoate (**8d**). The title compound was obtained as a reddish brown oil (74%), MS (ESI): m/z 545.3 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.9$ Hz), 1.24–1.47 (m, 4H, $CH_2CH_2CH_3$), 1.67–1.73 (m, 2H, $CHCH_2$), 1.78–2.05 (m, 6H, $ArCOOCH_2(CH_2)_3$), 2.07 (s, 3H, $OOCCCH_3$), 4.06 (t, 2H, $ArOCH_2$, $J = 6.3$ Hz), 4.36 (t, 2H, $ArCOOCH_2$, $J = 6.2$ Hz), 6.54 (dd, 1H, CH , $J = 4.8$ Hz, 8.1 Hz), 6.96 (d, 2H, ArH , $J = 8.8$ Hz), 7.28–7.33 (m, 1H, ArH), 7.39 (s, 1H, $C=CH$), 7.51–7.52 (m, 2H, ArH), 7.59 (d, 2H, ArH , $J = 8.7$ Hz), 7.88 (d, 1H, ArH , $J = 7.8$ Hz). ^{13}C NMR (75 Hz, $CDCl_3$): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 68.0, 64.9, 36.6, 29.7, 28.4, 25.8, 25.4, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for $C_{28}H_{32}O_5S_3$, $[M + H]^+$ 545.1412; found 545.2204.

(±)-6-(4-(3-Thioxo-3*H*-1,2-dithiol-5-yl)phenoxy)hexyl 2-(1-acetoxypentyl) benzoate (**8e**). The title compound was obtained as a reddish brown oil (78%), MS (ESI): m/z 559.4 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz), 1.26–1.45 (m, 4H, $CH_2CH_2CH_3$), 1.47–1.56 (m, 4H, $ArCOO(CH_2)_2CH_2$, $ArO(CH_2)_2CH_2$), 1.70–1.73 (m, 2H, $CHCH_2$), 1.78–1.85 (m, 4H, $ArCOOCH_2CH_2$, $ArOCH_2CH_2$), 2.07 (s, 3H, $OOCCCH_3$), 4.03 (t, 2H, $ArOCH_2$, $J = 6.3$ Hz), 4.34 (t, 2H, $ArCOOCH_2$, $J = 6.6$ Hz), 6.54 (dd, 1H, CH , $J = 4.8$ Hz, 8.1 Hz), 6.96 (d, 2H, ArH , $J = 8.7$ Hz), 7.28–7.33 (m, 1H, ArH), 7.38 (s, 1H, $C=CH$), 7.51–7.53 (m, 2H, ArH), 7.58 (d, 2H, ArH , $J = 8.7$ Hz), 7.88 (d, 1H, ArH , $J = 7.6$ Hz). ^{13}C NMR (75 Hz, $CDCl_3$): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 68.3, 65.1, 36.7, 28.9, 28.6, 28.1, 25.8, 25.7, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for $C_{29}H_{34}O_5S_3$, $[M + H]^+$ 559.1568; found 559.1626.

(±)-8-(4-(3-Thioxo-3*H*-1,2-dithiol-5-yl)phenoxy)octyl 2-(1-acetoxypentyl) benzoate (**8f**). The title compound was obtained as a reddish brown oil (70%), MS (ESI): m/z 587.2 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz),

1.26–1.42 (m, 4H, $\text{CH}_2\text{CH}_2\text{CH}_3$), 1.47–1.68 (m, 12H, $\text{ArCOOCH}_2(\text{CH})_6$), 1.70–1.73 (m, 2H, CHCH_2), 2.06 (s, 3H, OOCCH_3), 4.02 (t, 2H, ArOCH_2 , $J = 6.4$ Hz), 4.32 (t, 2H, ArCOOCH_2 , $J = 6.7$ Hz), 6.54 (dd, 1H, CH, $J = 4.7$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.7$ Hz), 7.28–7.33 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.52 (m, 2H, ArH), 7.59 (d, 2H, ArH, $J = 8.8$ Hz), 7.88 (d, 1H, ArH, $J = 7.7$ Hz). ^{13}C NMR (75 Hz, CDCl_3): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 68.3, 65.1, 36.7, 29.6, 29.3, 29.1, 25.9, 25.8, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for $\text{C}_{29}\text{H}_{34}\text{O}_5\text{S}_3$, $[\text{M} + \text{H}]^+$ 587.2246; found 587.2324.

(±)-12-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)dodecyl 2-(1-acetoxypentyl) benzoate (8g). The title compound was obtained as a reddish brown oil (67%), MS (ESI): m/z 665.2 $[\text{M} + \text{Na}]^+$. ^1H NMR (300 Hz, CDCl_3): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz), 1.26–1.42 (m, 4H, $\text{CH}_2\text{CH}_2\text{CH}_3$), 1.45–1.66 (m, 20H, $\text{ArCOOCH}_2(\text{CH})_{10}$), 1.70–1.73 (m, 2H, CHCH_2), 2.06 (s, 3H, OOCCH_3), 4.02 (t, 2H, ArOCH_2 , $J = 6.4$ Hz), 4.32 (t, 2H, ArCOOCH_2 , $J = 6.7$ Hz), 6.54 (dd, 1H, CH, $J = 4.7$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.7$ Hz), 7.28–7.33 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.52 (m, 2H, ArH), 7.59 (d, 2H, ArH, $J = 8.8$ Hz), 7.88 (d, 1H, ArH, $J = 7.7$ Hz). ^{13}C NMR (75 Hz, CDCl_3): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 68.3, 65.1, 36.7, 29.8, 29.7, 29.6, 29.3, 29.1, 25.9, 25.8, 22.5, 21.2, 14.0. HRMS (ESI): m/z calcd for $\text{C}_{35}\text{H}_{46}\text{O}_5\text{S}_3$, $[\text{M} + \text{H}]^+$ 643.2507; found 643.2601.

General procedure for the preparation of the target compounds 9a–f. To a solution of **7a** or **7b** (0.2 mmol) and K_2CO_3 (55 mg, 0.4 mmol) in acetone (10 mL) was added **3c–e** (0.2 mmol), and the solution was heated to 65 °C and refluxed for 6–8 h. The solution was then filtered, and the filtrate was reconstituted in EtOAc (20 mL) and solvent removed under reduced pressure. The crude compounds were purified by flash chromatography (PE/EtOAc = 5/1–1:1, v/v), to give the title compounds (65–71%).

(±)-4-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)butyl 2-(1-(2-(diethylamino) acetoxypentyl) benzoate (9a). The title compound was obtained as a reddish brown

oil (66%), MS (ESI): m/z 602.2 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.9$ Hz), 1.06 (t, 6H, $2 \times NCH_2CH_3$, $J = 7.2$ Hz), 1.26–1.45 (m, 4H, $CH_2CH_2CH_3$), 1.76–1.88 (m, 2H, $CHCH_2$), 2.00–2.04 (m, 4H, $ArCOOCH_2(CH_2)_2$), 2.68 (q, 4H, $2 \times NCH_2CH_3$, $J = 7.1$ Hz), 3.42 (s, 2H, $OCOCH_2N$), 4.10 (t, 2H, $ArOCH_2$, $J = 6.2$ Hz), 4.41 (t, 2H, $ArCOOCH_2$, $J = 6.2$ Hz), 6.62 (dd, 1H, CH, $J = 5.1$ Hz, 7.8 Hz), 6.96 (d, 2H, ArH, $J = 8.8$ Hz), 7.27–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.52–7.55 (m, 2H, ArH), 7.58 (d, 2H, ArH, $J = 8.8$ Hz), 7.87 (d, 1H, ArH, $J = 7.6$ Hz). ^{13}C NMR (75 Hz, $CDCl_3$): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 47.6, 36.3, 28.1, 25.8, 25.4, 22.5, 21.2, 14.0, 12.2. HRMS (ESI): m/z calcd for $C_{31}H_{39}NO_5S_3$, $[M + H]^+$ 602.1990; found 602.2074.

(±)-5-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)pentyl

2-(1-(2-(diethylamino)acetoxy) pentyl)benzoate (9b). The title compound was obtained as a reddish brown oil (67%), MS (ESI): m/z 616.2 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz), 1.06 (t, 6H, $2 \times NCH_2CH_3$, $J = 7.1$ Hz), 1.26–1.45 (m, 4H, $CH_2CH_2CH_3$), 1.60–1.67 (m, 2H, $CHCH_2$), 1.75–2.04 (m, 6H, $ArCOOCH_2(CH_2)_3$), 2.68 (q, 4H, $2 \times NCH_2CH_3$, $J = 7.1$ Hz), 3.42 (s, 2H, $OCOCH_2N$), 4.05 (t, 2H, $ArOCH_2$, $J = 6.2$ Hz), 4.36 (t, 2H, $ArCOOCH_2$, $J = 6.4$ Hz), 6.62 (dd, 1H, CH, $J = 4.9$ Hz, 8.0 Hz), 6.96 (d, 2H, ArH, $J = 8.8$ Hz), 7.28–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.54 (m, 2H, ArH), 7.58 (d, 2H, ArH, $J = 8.8$ Hz), 7.87 (d, 1H, ArH, $J = 7.6$ Hz). ^{13}C NMR (75 Hz, $CDCl_3$): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 47.6, 36.7, 29.7, 28.7, 28.4, 22.6, 22.5, 21.2, 14.0, 12.2. HRMS (ESI): m/z calcd for $C_{32}H_{41}NO_5S_3$, $[M + H]^+$ 616.2147; found 616.2230.

(±)-6-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)hexyl

2-(1-(2-(diethylamino)acetoxy) pentyl)benzoate (9c). The title compound was obtained as a reddish brown oil (70%), MS (ESI): m/z 630.2 $[M + H]^+$. 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 1H NMR (300 Hz, $CDCl_3$): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz), 1.07 (t, 6H, $2 \times NCH_2CH_3$, $J = 7.1$ Hz), 1.26–1.47 (m, 4H, $CH_2CH_2CH_3$), 1.60–1.66 (m,

2H, CHCH₂), 1.82–1.84 (m, 8H, ArCOOCH₂(CH₂)₄), 2.70 (q, 4H, 2×NCH₂CH₃, *J* = 7.0 Hz), 3.44 (s, 2H, OCOCH₂N), 4.03 (t, 2H, ArOCH₂, *J* = 6.3 Hz), 4.34 (t, 2H, ArCOOCH₂, *J* = 6.6 Hz), 6.62 (dd, 1H, CH, *J* = 5.0 Hz, 7.8 Hz), 6.96 (d, 2H, ArH, *J* = 8.8 Hz), 7.28–7.32 (m, 1H, ArH), 7.38 (s, 1H, C=CH), 7.51–7.54 (m, 2H, ArH), 7.58 (d, 2H, ArH, *J* = 8.8 Hz), 7.87 (d, 1H, ArH, *J* = 7.6 Hz). ¹³C NMR (75 Hz, CDCl₃): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.9, 67.8, 64.6, 47.6, 36.7, 29.7, 28.9, 28.6, 28.1, 25.8, 25.6, 22.4, 14.0, 12.2. HRMS (ESI): *m/z* calcd for C₃₃H₄₃NO₅S₃, [M + H]⁺ 630.2303; found 630.2387.

(±)-4-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)butyl

2-(1-(2-(4-methylpiperazin-1-yl) acetoxy)pentyl)benzoate (9d). The title compound was obtained as a reddish brown oil (71%), MS (ESI): *m/z* 629.2 [M + H]⁺. ¹H NMR (300 Hz, CDCl₃): δ 0.89 (t, 3H, CH₃, *J* = 6.9 Hz), 1.25–1.42 (m, 4H, CH₂CH₂CH₃), 1.54–1.62 (m, 2H, CHCH₂), 1.80–1.86 (m, 4H, ArCOOCH₂(CH₂)₂), 2.28 (s, 3H, NCH₃), 2.48–2.58 (m, 8H, 4×NCH₂), 3.24 (s, 2H, OCOCH₂N), 4.10 (t, 2H, ArOCH₂, *J* = 6.2 Hz), 4.40 (t, 2H, ArCOOCH₂, *J* = 6.4 Hz), 6.62 (dd, 1H, CH, *J* = 5.1 Hz, 7.7 Hz), 6.96 (d, 2H, ArH, *J* = 8.8 Hz), 7.26–7.31 (m, 1H, ArH), 7.39 (s, 1H, C=CH), 7.50–7.51 (m, 2H, ArH), 7.59 (d, 2H, ArH, *J* = 8.8 Hz), 7.86 (d, 1H, ArH, *J* = 8.0 Hz). ¹³C NMR (75 Hz, CDCl₃): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 73.1, 67.8, 64.6, 59.4, 54.8, 52.9, 45.9, 36.6, 29.7, 28.0, 25.4, 22.6, 14.0. HRMS (ESI): *m/z* calcd for C₃₂H₄₀N₂O₅S₃, [M + H]⁺ 629.2099; found 629.2183.

(±)-5-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)pentyl

2-(1-(2-(4-methylpiperazin-1-yl) acetoxy)pentyl)benzoate (9e). The title compound was obtained as a reddish brown oil (70%), MS (ESI): *m/z* 643.2 [M + H]⁺. ¹H NMR (300 Hz, CDCl₃): δ 0.89 (t, 3H, CH₃, *J* = 6.9 Hz), 1.24–1.46 (m, 4H, CH₂CH₂CH₃), 1.52–1.56 (m, 2H, CHCH₂), 1.84–1.92 (m, 6H, ArCOOCH₂(CH₂)₃), 2.27 (s, 3H, NCH₃), 2.47–2.57 (m, 8H, 4×NCH₂), 3.23 (s, 2H, OCOCH₂N), 4.06 (t, 2H, ArOCH₂, *J* = 6.2 Hz), 4.36 (t, 2H, ArCOOCH₂, *J* = 6.4 Hz), 6.62 (dd, 1H, CH, *J* = 5.0 Hz, 7.8 Hz), 6.96 (d, 2H, ArH, *J* = 8.8 Hz), 7.27–7.30 (m, 1H, ArH), 7.39 (s, 1H, C=CH),

7.50–7.51 (m, 2H, ArH), 7.59 (d, 2H, ArH, $J = 8.8$ Hz), 7.87 (d, 1H, ArH, $J = 7.8$ Hz). ^{13}C NMR (75 Hz, CDCl_3): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.2, 67.3, 64.1, 58.5, 53.9, 52.0, 45.0, 35.7, 28.7, 28.0, 27.6, 22.6, 21.6, 14.0. HRMS (ESI): m/z calcd for $\text{C}_{33}\text{H}_{42}\text{N}_2\text{O}_5\text{S}_3$, $[\text{M} + \text{H}]^+$ 643.2256; found 643.2339.

(±)-6-(4-(3-Thioxo-3H-1,2-dithiol-5-yl)phenoxy)hexyl

2-(1-(2-(4-methylpiperazin-1-yl) acetoxy)pentyl)benzoate (9f). The title compound was obtained as a reddish brown oil (65%), MS (ESI): m/z 657.0 $[\text{M} + \text{H}]^+$. ^1H NMR (300 Hz, CDCl_3): δ 0.89 (t, 3H, CH_3 , $J = 6.8$ Hz), 1.25–1.47 (m, 4H, $\text{CH}_2\text{CH}_2\text{CH}_3$), 1.50–1.56 (m, 2H, CHCH_2), 1.80–1.84 (m, 8H, $\text{ArCOOCH}_2(\text{CH}_2)_4$), 2.28 (s, 3H, NCH_3), 2.48–2.58 (m, 8H, $4 \times \text{NCH}_2$), 3.24 (s, 2H, OCOCH_2N), 4.03 (t, 2H, ArOCH_2 , $J = 6.4$ Hz), 4.33 (t, 2H, ArCOOCH_2 , $J = 6.6$ Hz), 6.62 (dd, 1H, CH , $J = 5.0$ Hz, 7.8 Hz), 6.96 (d, 2H, ArH, $J = 8.8$ Hz), 7.28–7.32 (m, 1H, ArH), 7.39 (s, 1H, $\text{C}=\text{CH}$), 7.49–7.50 (m, 2H, ArH), 7.59 (d, 2H, ArH, $J = 8.8$ Hz), 7.87 (d, 1H, ArH, $J = 7.6$ Hz). ^{13}C NMR (75 Hz, CDCl_3): δ 173.1, 170.3, 166.9, 162.3, 143.7, 134.8, 134.6, 132.3, 131.3, 130.1, 128.6, 127.1, 126.2, 125.8, 124.1, 115.5, 92.6, 72.2, 67.3, 64.1, 58.5, 53.9, 52.0, 45.0, 35.7, 28.7, 28.0, 27.6, 24.8, 22.8, 22.4, 14.0. HRMS (ESI): m/z calcd for $\text{C}_{34}\text{H}_{44}\text{N}_2\text{O}_5\text{S}_3$, $[\text{M} + \text{H}]^+$ 657.2412; found 657.2497.

4.2. Antiplatelet aggregation effect *in vitro*.

Blood samples were withdrawn from rabbit carotid artery and mixed with 3.8% sodium citrate solution (9:1, v/v), followed by centrifuging at 500 rpm for 10 min at room temperature. After the resulting platelet-rich plasma (PRP) supernatant was collected, the residue was centrifuged at 3000 rpm for another 10 min at room temperature to obtain platelet-poor plasma (PPP). The PRP was adjusted with PPP in order to obtain platelet counts of $400\text{--}450 \times 10^9$ PL/L. Platelet aggregation was determined by Born's turbidimetric method using a four-channel aggregometer (LG-PABER-I Platelet-Aggregometer, Beijing, China) within 3 h after blood collection. Briefly, PRP (240 μL) was pre-incubated with vehicle, positive control or different concentrations (0.1, 0.2, 0.4, 0.8 and 1.6 mM) of individual compounds (30 μL) for 5 min at 37 $^\circ\text{C}$, followed by the addition of 10 μM of the adenosine

5'-diphosphate sodium salt (ADP, Sigma–Aldrich, USA) or 0.33 mM of the arachidonic acid (AA, Cayman Chemical, USA) respectively to induce the platelet aggregation. The maximum aggregation rate (MAR) was recorded within 5 min at 37 °C. The inhibition rate of the tested compounds on platelet aggregation was calculated with the following formula: Inhibition rate (%) = (100% – MAR of tested compound/MAR of vehicle). The IC₅₀ value of each compound was calculated accordingly.

4.3. Inhibitory effect on the thrombosis in mice.

Male ICR mice (30–35 g) were purchased from B&K Universal Group Limited, Shanghai, China. Animals were kept under standard laboratory conditions and maintained in a 12 h light/dark cycle with free access to food and water. All procedures were performed following institutional approval in accordance with the NIH Guide for the Care and Use of Laboratory Animals. A model of acute systemic vascular thromboembolism induced by infusion of a mixture of collagen and adrenaline was performed as described previously.³² Briefly, mice were divided into groups randomly and treated orally with vehicle dimethyl sulfoxide (DSMO), aspirin (160 mg/kg), NBP (160 mg/kg), or **8e** (160 or 470 mg/kg) by gavage daily for seven days. Two hours after the last administration, the mice were injected intravenously with a mixture of 0.21 mg/kg collagen and 44.5 µg/kg adrenaline. Mice were observed for thrombosis-related death within 15 min. Results are expressed as the percentage of mice alive as a function of time. The protective rates against vehicle are also calculated.

4.4. Antithrombotic activity assay in rats.

Male Sprague-Dawley (SD) rats (250-280 g) were randomized and divided into six groups (*n* = 12 per group): (1) Control group (0.5% CMC-Na), (2) Vehicle group (A-V + 0.5% CMC-Na), (3) A-V + NBP (80 mg/kg) group, (4) A-V + aspirin (80 mg/kg) group, (5) A-V + **8e** (80 mg/kg) group, (6) A-V + **8e** (235 mg/kg) group. The 3-6 groups of rats were treated by gavage with the indicated dose of the compound in saline daily for consecutive 7 days. Subsequently, the rats were anesthetized by intraperitoneal (i.p.) injection of pentobarbital (50 mg/kg). The control group of rats

received a sham surgery while the remaining rats were subjected to the arterio-venous (A-V) shunt procedure.³³ Briefly, individual rats were inserted with an A-V shunt tube that directly connected the right carotid artery with the left jugular vein of rats. The polyethylene tube (14 cm, containing 6 cm long of 4 braided silk threads) was filled with saline before installation. After 15 min-circulation of blood through the shunt tube, both ends of the tubing were pinched, the cotton thread was removed from the shunt tube. The wet weight of the cotton thread was measured immediately and the cotton thread was dried at the room temperature for 6 h, followed by measuring the dry weight. The wet and dry weights of thrombus formed on the cotton threads were determined by subtracting the pre-experiment weight of the dry 6 cm-long threads.

4.5. Assay of hydrogen sulfide release *in vitro*

Compound **8e** (100 μM) were incubated (37° C) for timed intervals (0–90 min) with phosphate buffer or fresh rat liver homogenate (430 μL) and the concentration of released H_2S was measured as described previously.³⁴ After incubation, zinc acetate (1% w/v, 250 μL) was injected followed by trichloroacetic acid (10% w/v, 250 μL) to stop the reaction. Subsequently, *N,N*-dimethyl-*p*-phenylenediamine sulfate (20 μM ; 133 μL) in 7.2 M HCl was added followed by FeCl_3 (30 μM , 133 μL) in 1.2 M HCl. The resulting solution was determined 10 min thereafter using a microplate reader at 670 nm. The H_2S concentration of **8e** was calculated against a standard curve of NaHS (3.125, 6.25, 12.5, 25, 50, 100 μM). Results are expressed as total % conversion of **8e** to H_2S at each time point.

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Notes and references

^aThe Key Laboratory of Carbohydrate Chemistry and Biotechnology, Ministry of Education, School of Biotechnology, Jiangnan University, Lihu Avenue 1800, Wuxi 214122, P. R. China. E-mail: jianyin@jiangnan.edu.cn; Fax: +86-0510-85328229; Tel: +86-0510-85328229 (J.Y.); ^bState Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, P. R. China. E-mail: zyhtgd@163.com; Fax: +86-025-83271015; Tel: +86-25-83271015 (Y.Z.); E-mail: huijicpu@163.com; Fax: +86-25-86635503; Tel: +86-25-86021369 (H.J.); ^cSynergetic Innovation Center of Food Safety and Nutrition, Jiangnan University, Lihu Avenue 1800, Wuxi 214122, P. R. China. ^dDepartment of Pharmacology, China Pharmaceutical University, Nanjing 210009, P. R. China and ^eCenter of Drug Discovery, China Pharmaceutical University, Nanjing 210009, P. R. China.

#These authors contributed equally to this work.

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