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ARTICLE TYPE

High efficiency single-junction semitransparent perovskite solar cells

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Semitransparent perovskite solar cells with a high power conversion efficiency (PCE) close to 7 % and 30% full device transparency have been achieved by implementing a thin perovskite layer and a simple foil compatible layout.

Thin film photovoltaics have attracted much attention as a promising source of renewable energy to reduce the dependencies on fossil and nuclear industries.¹ In the past years, a lot of effort has been devoted to the development of organic photovoltaics (OPV), including new materials and device structures, to provide an alternative to Si-based solar cells. The advantages of OPV include their potential low cost manufacturing, light weight, flexibility and the availability of different materials allowing for colour tuning of the devices.²⁻⁴ Additionally, they can be made semitransparent which allows for specific building integrated elements that could lead to totally self-sustaining buildings at low cost. Even though some breakthroughs in device efficiency has been recently reported, the efficiency of single junction OPV devices is generally below 10 %.⁵⁻⁸ For semitransparent (ST) solar cells, the use of absorbers with a lower bandgap, in order to shift the absorption spectra to the infrared region, the absence of the light reflecting electrode and the implementation of thin active layers compromise even more the power conversion efficiency (PCE).⁹⁻¹² In order to enhance the performance of ST solar cells, in terms of their average visible transparency (AVT) and PCE, several strategies have been implemented including extensive studies about semitransparent electrodes, tandem devices and recently the implementation of more efficient materials. For the case of the transparent electrodes, the most explored material has been thin silver,^{8, 13 14} also in combination with different capping layers,¹⁵⁻¹⁷ or in configurations like Ag grids^{18, 19} or Ag NWs.^{12, 20} PEDOT has also been explored as transparent electrode.^{21, 22} Finally, other materials like thin Al,^{23, 24} thin Au,^{25, 26} ITO^{27, 28}, graphene^{29, 30} and CNT³¹⁻³³ have also been tested with more discrete results. From these studies, considering just single junction devices with AVT higher than 20%, the top performance was set by Chen et al.⁸ 6.2% PCE – 21.2% AVT using a thin silver layer, followed by 5.6% PCE – ~30% AVT¹¹ using thin silver and including an external photonic structure for trapping the UV and NIR sunlight and finally 4.0% PCE – ~50% AVT¹² using ITO nanoparticles and Ag NW mixed electrode. A way to increase the performance of this kind of devices has been the development of tandem OPVs, for example Chen et. al, obtained a 7% PCE - 30% AVT.³⁴ However, the complexity of tandem devices impedes their facile integration in low cost application easier to achieve with single junction devices with improved performances.

An alternative approach to increase the performance of semitransparent photovoltaic devices is to use more efficient materials. Methylammonium lead halide perovskites are an

interesting class of materials that have excellent semiconductor properties and have led to very efficient solar cells.^{35, 36} From the first report by Miyasaki et al.,³⁷ tremendous progress in the performance of methylammonium lead iodide perovskite based solar cells has been achieved.³⁸⁻⁴³ Power conversion efficiencies in excess of 15 % have been obtained.³⁸⁻⁴⁰ Most of the high efficient perovskite solar cells reported until now sandwich the perovskite in between a metal oxide layer such as Al₂O₃, TiO₂ or ZrO₂, and an organic hole transport material. In most cases the organic hole-transporting material (frequently spiro-OMeTAD) is applied on top of the perovskite as a rather thick layer and partially oxidized. The presence of the several hundred nanometer thick partially oxidized hole transport layer leads to parasitic absorption losses reducing the transparency of the active stack. As a consequence this would ultimately limit the achievable transparency of semi-transparent devices. Based on this approach, ST perovskite solar cells achieved 3.5% PCE – ~ 30% AVT.⁴⁴ Their strategy relied on the dewetting of the perovskite film to create ‘perovskite islands’ achieving with this a high transmittance but clearly decreasing the overall PCE after the present voids in the active layer.

Recently, an alternative device layout where the thickness of continuous perovskite layers can be precisely controlled by thermal evaporation while avoiding the use of materials that induce parasitic absorption was developed.⁴⁵ In this layout the conductive polymer poly(3,4-ethylenedioxythiophene):poly(styrenesulfonic acid) (PEDOT:PSS) is deposited on top of the transparent conductive substrate. PEDOT:PSS has been designed for its transparency in the visible part of the spectrum and with layer thicknesses around 100 nm has a transmittance above 90 %. In the first examples the perovskite layer was deposited on top of the PEDOT:PSS and capped with a hole blocking/electron transporting layer ([6,6]-phenyl C₆₁-butyric acid methylester, PCBM₆₀ leading to PCEs of 7.4 %. Recently, You et al., improved the performance of this configuration reaching a maximum PCE of 11.5 %.⁴⁶ When besides the hole blocking layer also a thin electron blocking layer is used in between the PEDOT:PSS and the perovskite layer, the device performance is further improved to reach a PCE of 14.8 %.^{45, 47} Hence, the transmittance of this type of solar cell is almost completely determined by the perovskite layer and the two electrodes. An additional advantage of this layout is the absence of metal oxides which facilitates the manufacturing of the cells and make them compatible with flexible applications and roll to roll (R2R) processing.^{46, 48}

Here we present the development of perovskite semitransparent solar cells reaching 6.4% PCE – 29% AVT and 7.3% PCE – 22% AVT. This successful implementation is based on the robustness of the perovskite evaporation process enabling depositing

continuous layers as thin as 40 nm. The strategy included the implementation of the device layout above described with minimized parasitic absorption and the development of an ultra-thin gold electrode capped with a LiF layer. Such capping layer was intended to protect the gold layer while simultaneously reducing the energy lost, by device specular reflection, which translated in an enhanced device transparency. These results are the best reported to date for semitransparent single junction solar cells, demonstrating their capability for building integrated photovoltaics and other industrial semitransparent applications.

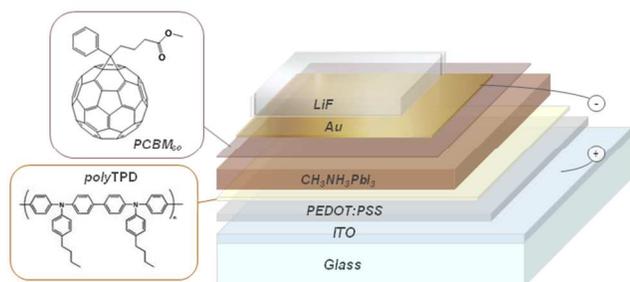


Fig. 1 Schematic layout of the semitransparent solar cell and chemical structures of the organic hole and electron blocking materials.

The architecture for the semitransparent perovskite device explored in this work is presented in Figure 1. In order to achieve a semitransparent device, the top gold electrode was only 6 nm thick. This structure was computationally modelled by employing the transfer matrix formalism^{49, 50} where the optical properties of the materials were obtained from literature or estimated using the method reported by Manificier et al.⁵¹ Figure 2a shows the dependence of the J_{sc} and AVT with the perovskite thickness with and without including a protective LiF layer. Clearly, a higher J_{sc} is theoretically expected for thicker perovskite layers but implying a strongly reduced AVT. Therefore, a trade off needs to be achieved between its inversely related efficiency and sunlight transmission which implies the sacrificing in either efficiency or transparency as seen in recent related publications.⁴⁴ This negative relationship can be partially overcome by exploring optical interference effects inherent to the coherent interaction of sunlight with thin film photovoltaic devices.⁵² In particular, the LiF transparent capping layer besides providing a protection for the underlying gold layer can be used to modify the electric field distribution inside the device. The optical properties of the different device layers determine that the average sunlight reflected out of the device (ARF) is minimized for the combinations of perovskite and LiF thicknesses enclosed in the white square shown in Figure 2b. Equivalent maps were developed for the expected AVT and J_{sc} (Figures 2c and 2d, respectively). From these graphs it was concluded that, for any given thickness of the perovskite layer, a LiF layer around 100 nm thick was optimal for reducing the energy lost by the specular reflection of the device while enabling more transparent and keeping a high J_{sc} as shown in Fig 2a (dashed lines).

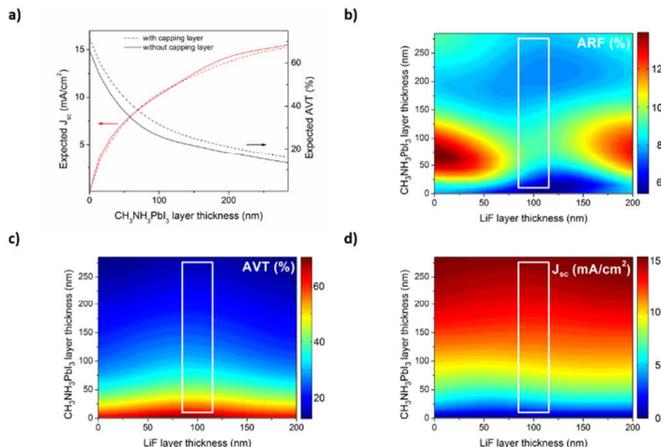


Fig. 2. Optical modelling of the semitransparent perovskite solar cells. a) Negative-related J_{sc} (red lines) and AVT (black lines) for a 6 nm thick gold device. The effect of including a 100 nm LiF capping layer is illustrated (dashed lines). In general, such capping layer modifies the field distribution inside the device and which has an effect on the (b) average reflected sunlight (ARF, 400 – 800 nm), (c) average visible transmission (AVT, 400-800 nm) and (d) short-circuit current density. The white square encloses the combinations of CH₃NH₃PbI₃ and LiF thicknesses that simultaneously reduce the ARF, enhance the AVT and keep a high J_{sc} .

The semitransparent solar cells were prepared by sandwiching the methylammonium lead iodide perovskite between two very thin electron and hole-blocking layers consisting of organic molecules (see Figure 1). First, a 75 nm of poly(3,4-ethylenedioxythiophene):poly(styrenesulfonic acid) (PEDOT:PSS) CLEVIOS P VP Al 4083 from Hereaus was spin-coated on an ITO covered glass substrate. After annealing for 15 minutes at 150 °C, a thin layer of the electron-blocking material poly[N,N'-bis(4-butylphenyl)-N,N'-bis(phenyl)benzidine] (polyTPD) from ADSdyesource was deposited (20 nm) from a chlorobenzene solution (7 mg ml⁻¹) and then annealed at 180 °C during 30 minutes. This annealing step was required in order to fix the polyTPD and prevent its removal when the hole-blocking layer is deposited on the rather thin perovskite layers from the same solvent. To ensure a high purity and a high control of the thickness, the CH₃NH₃PbI₃ layers were prepared by the co-evaporation of the two starting materials PbI₂ and CH₃NH₃I in a high vacuum chamber as described previously^{45, 48}. Four different thicknesses (40 nm, 100 nm, 180 nm and 280 nm) were evaluated. Subsequently, a thin layer (20 nm) of the hole-blocking material [6,6]-phenyl C₆₁-butyric acid methylester (PCBM₆₀) Solenne BV was deposited from a chlorobenzene solution, 10 mg ml⁻¹, using meniscus coating.⁵³ The thickness of the layers was verified using both profilometer and absorbance measurements. For non-transparent devices the described stack was covered by an 70 nm gold layer deposited using vacuum evaporation. In the case of semitransparent devices an alternative top electrode is required that is conductive and has a high transparency. Additionally, the top electrode and its deposition method should be compatible with the device stack. Therefore, only top electrodes prepared by evaporation of metals and dielectrics were considered (Figure S1). The optimum gold layer had a thickness of 6 nm, as it showed good homogeneity, acceptable conductivity and transparency values. This gold layer is considerably thinner than previously reported semitransparent gold layers in ST perovskite devices⁴⁴ which leads to a reduction on the parasitic absorption always induced by metallic layers. As predicted by the optical model, this ultra-thin layer of Au (6nm) was capped by a 100 nm layer of lithium fluoride (LiF) to enhance optically the device.

This LiF layer also protected the cell allowing for easier handling. More details of the device fabrication and characterization are provided in the ESI.†

The robustness of the thermal evaporation to grow perovskite layers in a wide range of thicknesses is crucial to the development of these semitransparent perovskite solar cells. Indeed, the resulting $\text{CH}_3\text{NH}_3\text{PbI}_3$ layers showed a very high crystallinity and uniformity as evidenced by grazing incidence X-ray diffraction (GIXRD) and Scanning Electron Microscopy. As shown in Figure 3a, despite their thicknesses, all the deposited perovskite layers showed a high crystallinity reaching an excellent fit to a one-phase model with a tetragonal cell ($a = 8.80(2)$, $c = 12.57(2)$ Å) and space group $I4/ cm$. Additionally, a high film uniformity is apparent from the SEM pictures presented in Figure 3b.

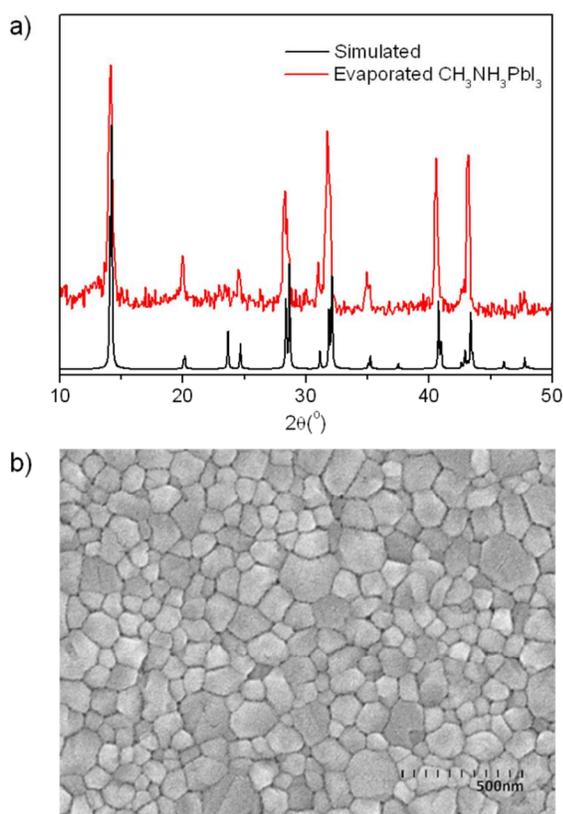


Fig. 3. (a) Grazing incidence X-ray diffraction (GIXRD) pattern of a typical evaporated $\text{CH}_3\text{NH}_3\text{PbI}_3$ film. As reference, it is also added the simulated perovskite pattern with preferred orientation along the (100) and (001) directions. (b) SEM picture of a typical evaporated perovskite layer of 200 nm.

The transmittance of the layers including the glass substrate (device without top electrode), the ST top contact and the completed ST device are depicted in Figure S2 for a perovskite thickness of 100 nm. Clearly, the ST top electrode reduces significantly the transmittance of the complete device. The transmittance spectra for the completed ST device employing different perovskite layer thicknesses are shown in Figure 4a. As expected the transmission increases with reducing perovskite layer thickness, although not completely linearly in function of the film thickness probably due to slight changes in the

perovskite optical properties depending on its particular evaporation process. To highlight the effect of the different perovskite thickness on the transparency of the device, the values for the AVT of the device stacks with and without the ST top electrode are given (inset Figure 4a). Such AVT is taken as the average of the transmittance in the visible region of the spectra leads to good transmittances when the thickness is below 200 nm, showing an AVT of 44% for a perovskite thickness of 100 nm. Thicker perovskite films lead to an important decrease in the transmittance, reducing the AVT value to 19% for films with a thickness of 280 nm.

The completed device with a perovskite layer thickness of 100 nm has an AVT close to 30% which is high enough for many applications. Figure 4c shows the current density versus voltage ($J-V$) characteristics for the described cells with 0.12 cm^2 active area under light intensities of 100 mW cm^{-2} . It is important to mention that the curves are the same under forward and reverse scan directions and as such do not display hysteresis.

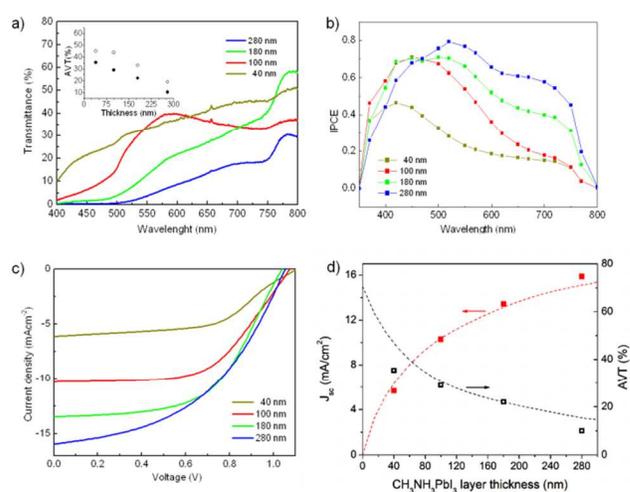


Fig. 4. a) Transmittance spectra through the complete device for different perovskite layer thicknesses. The inset shows the AVT values for the devices with the ST electrode (filled circle) and without (open circles). b) IPCE spectrum and c) $J-V$ characteristics of the best semitransparent devices comprising the Au/LiF electrode for different perovskite thicknesses. d) Comparison between the experimentally obtained AVT (red full squares) and J_{sc} (black empty squares) with the optical modelling (dashed lines).

The results show a decrease of the current density when the active layer thickness decreases. This is expected as less light is absorbed and as a consequence fewer charges can be photo-generated. This trend is also seen in the IPCE graph (Figure 4b) where the most notable decrease is observed in the red region of the spectra. The key performance parameter deduced from Figure 4 for the different devices are depicted in Table 1. Additionally, an informative table including average values and standard deviation for the most important parameters can be found in the supporting information, Table S2.

Finally, Figure 4d presents a comparison between the experimentally obtained parameters and the optically modelled ones. A reasonably good fitting was achieved. Just small

discrepancies appeared for the AVT in the case of the 40 and 280 nm thick perovskite devices which could be due to slight changes in their thicknesses or the before-mentioned small variations in the optical properties of the deposited perovskite layers. All devices exhibited a yellowish/brown tonality with considerably good agreement with the optical modelling (Figure S3). The 40 nm thick perovskite device displayed the most neutral color (0.36;0.37) and future works could be addressed to develop optical strategies to tune such tonalities.

During the optical computer simulation the IPCE was modelled as $IPCE = \phi \eta_{abs}(\lambda)$ where $\eta_{abs}(\lambda)$ is related to the efficiency in absorbing the sunlight and the wavelength-independent parameter ϕ is associated to the exciton diffusion efficiency, charge separation efficiency, charge transport efficiency and final charge collection efficiency. The final fitting of this ϕ parameter, after matching the experimental and computer modelled J_{sc} , resulted to be 0.9 which evidences the excellent exciton and charge conductivity properties of the perovskite layers.

Table 1. Most important parameters for the best studied semitransparent solar cells.

	J_{sc} (mA cm^{-2})	V_{oc} (V)	FF	PCE (%)	AVT ¹ (%)	AVT ² (%)
40 nm	5.66	1.037	57.7	3.39	35.4	45
100nm	10.30	1.074	57.9	6.41	29	44
180nm	13.43	1.037	52.5	7.31	22	33
280nm	15.88	1.052	46.	7.73	10	19

¹Average transmittance values through the whole device

²Average transmittance values without the top semitransparent electrode

As a result power conversion efficiencies as high as 6.4% for devices with an AVT of 30% were achieved. This is amongst the highest values reported for semitransparent single-junction cells. Most of the devices lead to quite high short-circuit current densities (J_{sc}) (10 to 16 mA cm^{-2}). The V_{oc} is almost not affected by the thickness of the active layer and remains above 1 V in all the cells. The fill factor (FF) is good for the thin devices, reaching values of 60 %, yet with increasing perovskite layer thickness it decreases to around 45%. The reduction of the FF of the cells with thicker perovskite layers is likely caused by the limited conductivity of the ST top electrode. Comparison experiments were performed with the same perovskite layer yet with a thicker (70 nm) top electrode and these non-transparent devices had slightly higher current densities and FF around 60 % (see Table S2 of the ESI).

The obtained 100 nm or 180 nm perovskite cells represent one of the best performances reported for semitransparent single-junction solar cells achieving power conversion efficiencies as high as 6.4% and 7.3% respectively. A photograph of a typical semitransparent solar cell is shown in Fig.5. Typical samples look yellowish/light brown, depending on the perovskite thickness (see SI, Figure S3). More importantly, this work shows a clear route to develop high performance ST solar cells.



Fig. 5 Photograph of the semitransparent solar cell having a 100 nm perovskite layer resulting in an AVT ~ 30% and a PCE of 6.4 %.

Conclusions

We have successfully prepared high efficient semitransparent solar cells based on methylammonium lead iodide perovskite layers sandwiched in between two organic charge transport layers. A simple cell configuration, which does not require high temperature processes, leads to semitransparent cells with AVT close to 22% and 29 % through the complete device, and high power conversion efficiencies of 7.3% and 6.4% respectively. These results are among the best performances reported to date for single ST solar cells and are fundamentally based on the successful development of thin uniform perovskite layers by thermal evaporation. Additionally, the implementation of an ultra-thin gold layer as electrode enabled minimizing its parasitic absorption and, the introduction of a LiF capping layer was crucial to reduce the energy lost in the device specular reflection enhancing the device transparency without affecting the photon harvesting in the active layer. Furthermore, better performances are expected by implementing these ideas with even better semitransparent electrodes.

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Notes and references

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- †Electronic Supplementary Information (ESI) available: Experimental details on device preparation and characterization, electrode transmittance measurements, comparison cells with thick electrodes, transmission spectra and chromaticity coordinates. See DOI: 10.1039/b000000x/
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TOC

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