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COMMUNICATION

Construction of tetralin skeletons based on rhodium-catalysed site-selective ring opening of benzocyclobutenols

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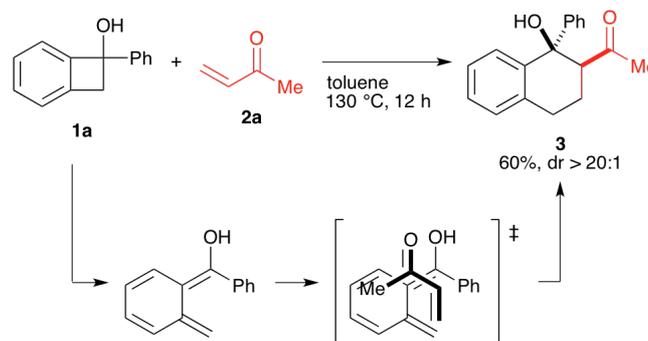
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Tetralins (tetrahydronaphthalenes) are synthesised from benzocyclobutenols based on the rhodium-catalysed site-selective ring opening followed by intermolecular/intramolecular conjugate addition of the resulting arylrhodium species to electron-deficient alkenes. The produced structures make a remarkable contrast with those available from the same compounds under thermal reaction conditions.

Tetralin is a key structural motif found in a wide variety of bioactive molecules including natural products such as heritonine,¹ cycloolivil,² morphine³ and daunorubicin.⁴ The development of efficient methods for the construction of tetralin skeleton has been a subject of intense research. We report herein a new protocol to prepare tetralin derivatives from benzocyclobutenols based on their site-selective ring opening under the catalysis of rhodium.⁵⁻⁸ 2-Hydroxytetralins are stereoselectively synthesised by an insertion reaction of vinyl ketones into the C(sp²)-C(sp³) bond of benzocyclobutenols. The rhodium-catalysed rearrangement of 1-alkenylbenzocyclobutenols furnishes 2-tetralones. The produced structures make a remarkable contrast with those available from the same compounds under thermal reaction conditions.

A [4+2] cycloaddition reaction of *o*-quinodimethanes with alkenes presents one of the most reliable synthetic pathways to tetralin skeletons.⁹ For example, when benzocyclobutenols are heated, a thermal ring opening reaction takes place with outward rotation of the hydroxyl group¹⁰ to stereoselectively furnish the hydroxy-substituted *o*-quinodimethanes. A regioselective [4+2] cycloaddition reaction with electron-deficient alkenes follows in an *endo* fashion.¹¹ Thus, simple heating of a toluene solution of benzocyclobutenol **1a** and methyl vinyl ketone (**2a**) at the refluxing temperature (bath temperature: 130 °C) gave 1-hydroxytetralin **3** in a diastereoselective fashion (Scheme 1). It has been also known that treatment of benzocyclobutenols with butyllithium generates an oxyanion intermediate, which follows an analogous pathway even at -78 °C.¹² In these cases, the C-C double

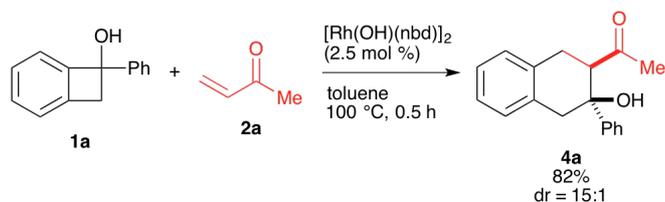
bond is formally inserted into the C(sp²)-C(sp³) bond of the cyclobutene ring.



Scheme 1 Thermal reaction of **1a** and **2a**.

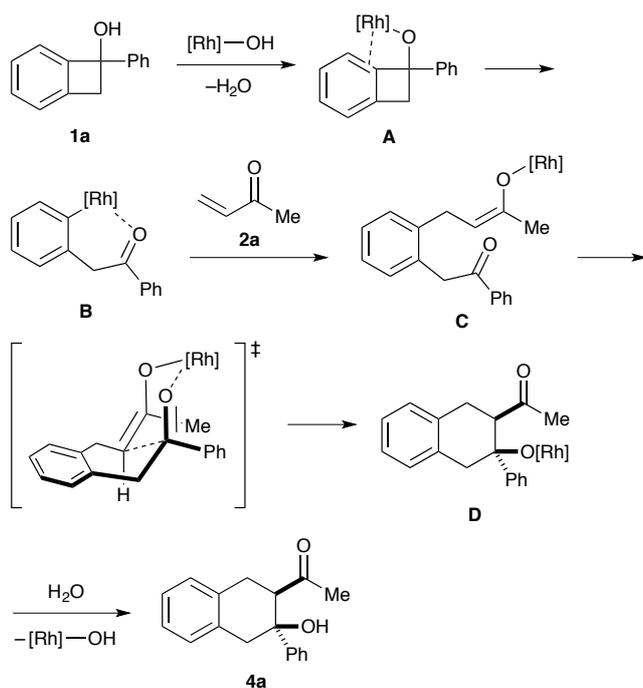
On the other hand, we have reported a different type of ring opening reaction of benzocyclobutenols; a rhodium complex prompts ring opening selectively at the C(sp²)-C(sp³) bond to furnish an *ortho*-aclylmethyl-substituted arylrhodium intermediate.^{5,13,14} The subsequent intermolecular addition across the C-C triple bond of alkynes generates alkenylrhodium species, which then undergoes intramolecular addition onto the carbonyl group to construct dihydronaphthalene frameworks. In a formal sense, a C-C triple bond is inserted into the C(sp²)-C(sp³) bond of the cyclobutene ring in an atom-economical way. This finding led us to explore new synthetic pathways leading to tetralins (tetrahydronaphthalenes) from benzocyclobutenol derivatives.

We first carried out a thermal reaction of **1a** and **2a** at 100 °C for 30 min in the absence of a rhodium catalyst. The thermal ring-opening reaction of **1a** was so slow that 90% of **1a** was recovered. Next, **1a** was reacted with **2a** in the presence of [Rh(OH)(nbd)]₂ under otherwise identical conditions. The C-C double bond of **2a** was successfully inserted into the C(sp²)-C(sp³) bond of **1a** to produce 2-hydroxytetralin **4a** in 82% yield with a trace amount of the minor diastereomer (diastereomeric ratio, dr = 15:1).



Scheme 2. Rhodium-catalysed reaction of **1** and **2**.

A probable mechanistic scenario for the diastereoselective formation of **4a** is depicted in Scheme 3. Initially, the hydroxylic proton of benzocyclobutenol **1a** is exchanged with rhodium to furnish the rhodium benzocyclobutenolate **A**. The benzene ring π -coordinates to the rhodium centre,¹⁵ and the π -coordination is retained during the subsequent β -carbon elimination so that the ipso carbon selectively migrates on rhodium. Thus, the $C(sp^2)$ – $C(sp^3)$ bond is selectively cleaved. The resulting arylrhodium species **B** undergoes conjugate addition across **2a**, which is taking an *s-cis* conformation to allow a six-membered transition state. As a consequence, the (*Z*)-enolate **C** is generated. Then, an intramolecular aldol reaction follows again via a six-membered transition state, for which chair-like conformation is assumed to afford *syn*-aldolate **D** stereoselectively.^{16,17} Protonation of **D** with water, generated in the first step, or with **1a** produces **4a** and the next catalytic cycle starts over.

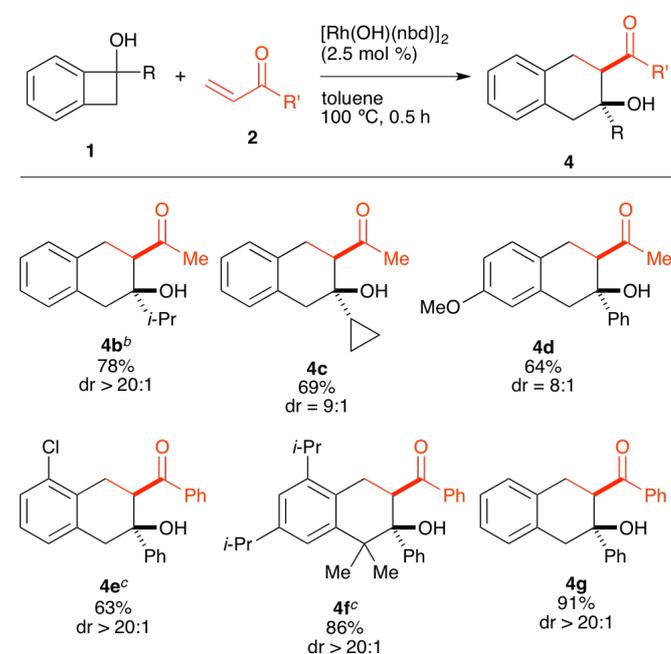


Scheme 3. Proposed mechanism for the formation of **4a**.

The scope of the site-selective insertion reaction is shown in Table 1. Although non-substituted benzocyclobutenol ($R = H$) failed to afford the 2-hydroxytetralin, substituted benzocyclobutenols reacted with vinyl ketones to afford **4b-g** with dr ranging from 8:1 to >20:1. Cyclopropyl, methoxy and chloro groups remained intact under the reaction conditions. The site-selective ring opening was observed even when the migrating ipso sp^2 carbon was obstructed by its *ortho*-

isopropyl group. Phenyl vinyl ketone also successfully engaged in the insertion reaction. On the other hand, alkenes lacking electron-withdrawing groups (e.g. ethylene and styrene) failed to be incorporated. α,β -Unsaturated ketones possessing a substituent at the α - or β -position gave significantly lower yield, presumably due to steric reasons. The use of acrolein and acrylate gave a complex mixture containing the desired 2-hydroxytetralin (<30% yield).

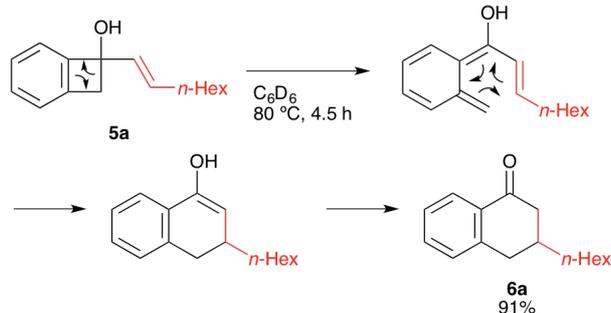
Table 1 Rhodium-catalysed reactions of **1** and **2**.^a



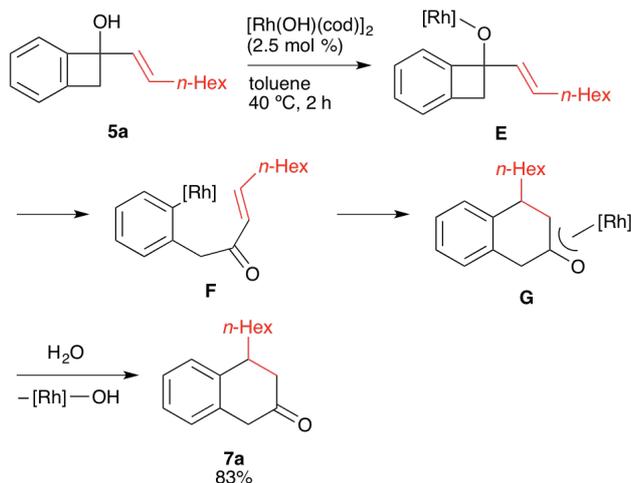
^a Reaction conditions: Benzocyclobutenol **1** (0.20 mmol), vinyl ketone **2** (0.40 mmol, 2 equiv), $[Rh(OH)(nbd)]_2$ (2.5 mol %), toluene (1 mL), 100 °C, 0.5 h. Isolated yields of the major diastereomer were shown. Dr were determined by NMR analysis of the crude reaction mixture. ^b 1 h. ^c $[Rh(OH)(cod)]_2$ (2.5 mol %) was employed as the catalyst.

Next examined was the construction of tetralones from 1-alkenylbenzocyclobutenol **5a**, which was easily prepared by addition of alkenyllithium to benzocyclobutenone. When **5a** was simply heated at 80 °C in C_6D_6 for 4.5 h, 1-tetralone **6a** was obtained in 91% yield, as previously reported with analogous substrates (Scheme 4).¹⁸ 4 π -Ring opening of **5a** is followed by 6 π -electrocyclic ring closure to afford **6a**. In sharp contrast, treatment of **5a** with a catalytic amount of $[Rh(OH)(cod)]_2$ at 40 °C for 2 h gave 2-tetralone **7a** in 83% isolated yield.¹⁹ Mechanistically, it is assumed the site-selective cleavage of the $C(sp^2)$ – $C(sp^3)$ bond generates the arylrhodium intermediate **F** with the *ortho* position substituted by an α,β -unsaturated carbonyl group. An intramolecular conjugate addition reaction follows, and the resulting oxa- π -allylrhodium **G** is protonated with water, or with **5a** to give **7a**.

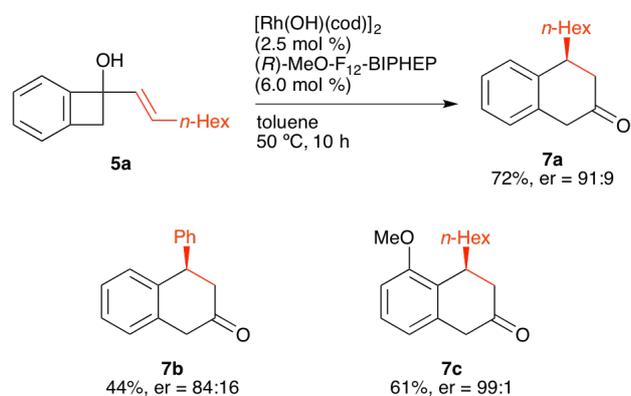
Thus, the pathway of the intramolecular rearrangement reaction of **5a** is also changed by a rhodium complex. In addition, the racemic mixture of **5a** was enantioselectively rearranged when electron-deficient chiral diphosphine (*R*)-MeO-F₁₂-BIPHEP²⁰ was employed as the ligand for rhodium. 2-Tetralone **7a** was obtained in 72% yield with the enantiomeric ratio (er) of 91:9. The reaction conditions were applied to the synthesis of **7b** (er = 84:16) and **7c** (er = 99:1).



Scheme 4. Thermal rearrangement of 5a.



Scheme 5. Rhodium-catalysed rearrangement of 5a.



Scheme 6. Asymmetric rearrangement of 5.

In summary, tetralin skeletons are constructed from benzocyclobutenols based on the rhodium-catalysed site-selective ring opening reaction. Vinyl ketones are site-selectively inserted into the $C(sp^2)-C(sp^3)$ bond of benzocyclobutenols to produce 2-hydroxytetralins. 1-Alkenylbenzocyclobutenols are restructured into 2-tetralones. The obtained tetralins markedly contrast with those given by the conventional thermal reactions.

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Notes and references

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† Electronic Supplementary Information (ESI) available: Detailed experimental procedures, and spectral data for all compounds, including scanned images of 1H and ^{13}C NMR spectra. See DOI: 10.1039/c000000x/.

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