

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

COMMUNICATION

Palladium-catalyzed Oxidative CH/CH Cross-coupling of Pyridine *N*-oxides with Five-membered Heterocycles

Cite this: DOI: 10.1039/x0xx00000x

Wei Liu, † Xin Yu, † Yahui Li, † and Chunxiang Kuang* †

Received 00th January 2012,
Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

Using Ag_2CO_3 as an additive, we developed the Pd-catalyzed intermolecular C–H/C–H cross-coupling between pyridine *N*-oxides with five-membered heterocycles such as 1-benzyl-1,2,3-triazoles, thiophenes and furans. This protocol provides an efficient and regioselective approach for the synthesis of unsymmetrical biheteroaryl molecules

A heteroaryl-heteroaryl structural motif is frequently featured at the core of many natural products, pharmaceuticals, and electronic materials.^[1] Carbon–carbon biheteroaryl linkages are generally considered formed through the C–C bond cross-coupling between a heteroarylhalide with a heteroaryl organometallic reagent.^[2] However, the preactivation of heteroaromatic carbon fragments with metal-containing functionalities and halides could involve several synthetic steps. Moreover, several important types of heteroaryl organometallic compounds have proven to be inadequately stable to participate in cross-coupling reactions.^[3–5] For example, the instability and difficult synthesis of 2-pyridyl organometallics severely limits their application.^[6] To solve this drawback, pyridine *N*-oxides have been introduced as widely available and bench-stable substrates for direct cross-coupling reactions.^[7]

At present, C–H/C–H oxidative cross-coupling between two (hetero)arenes is regarded as one of the most attractive strategies for building biheteroaryl linkages without the need for the tedious prefunctionalization of starting materials.^[8] However, this type of C–H/C–H cross-coupling for the

unsymmetrical construction of “heteroaryl–heteroaryl” scaffolds has rarely been studied because of the daunting complexity in the inversion of reactivity and selectivity.^[8] To the best of our knowledge, a few methods have been reported for the synthesis of the structure of 2-pyridinyl-1,2,3-triazoles because of their various biological activities.^[9] However, the metal-catalyzed oxidative cross-coupling of two heteroaryl C–H bonds to form 1-benzyl-5-pyridinyl-1,2,3-triazole molecules remains a challenge. Moreover, only a few investigations of two heteroaryl C–H bonds forming 2-pyridinyl-thiophen(furan) have been reported to date; however, the substrates of pyridine *N*-oxides are limited only to 2-substituted pyridine *N*-oxides and quinoline *N*-oxides.^[10] Additional, direct arylation of the 3-substituted pyridine *N*-oxides and isoquinoline *N*-oxide confronted the problem of regiochemistry because of a competition between C2 and C6 positions.^[7j] With a continuing interest in C–H/C–H cross-coupling of Heterocyclic *N*-oxides^[11], we continued our efforts to expand the application and understanding of the metal-catalyzed cross-coupling between pyridine *N*-oxides and heterocycles. Herein, we illustrate an efficient and site-selective CH/CH cross-coupling of pyridine *N*-oxides with 1-benzyl-1,2,3-triazole, thiophen and furan derivatives.

We initially used the condition similar to that reported^[10a–b] for the oxidative CH/CH cross-coupling reaction of pyridine *N*-oxide (**1a**) and 1-(*p*-methoxybenzyl)-1,2,3-triazoles (**2a**); however, the desired product (**3a**) was observed with low yield of 15% (Table 1, entry 1). From the tests using different oxidants, we found that the corresponding products were obtained with the incomplete ionization of silver salts (Ag_2CO_3 , Ag_2O , and AgOAc). In addition, the oxidant Ag_2CO_3 was more effective than AgOAc , Ag_2O , Ag_2SO_4 , AgNO_3 , $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$, and FeCl_3 (Table 1, entries 2 to 7,) respectively. It was found that 2,6-lutidine^{9c} produced a higher yield than pyridine and 1,10-phenanthroline as ligand (Table 1, entries 7 to 9,). Moreover, we found that the addition of DMSO with dioxane resulted in the highest yield of 83% (Table 1, entry 10) in comparison with other solvents, such as DMSO, DMF, and NMP (entries 10 to 15, respectively), even when the reaction was performed at 160 °C

^a Department of Chemistry, Tongji University, Siping Road 1239, Shanghai 200092, China, E-mail: kuangcx@tongji.edu.cn
Tel: (+86)-21-6598319

^b Key Laboratory of Yangtze River Water Environment, Ministry of Education, Siping Road 1239, Shanghai 200092, China

† Electronic Supplementary Information (ESI) available: characterization data, ¹H and ¹³C NMR spectra. See DOI: 10.1039/b000000x/

(Table 1, entry 16). The optimal yield can be achieved by using 5 mol % of Pd(OAc)₂, Ag₂CO₃ (2 equiv), and 30 mol % of 2,6-lutidine and the desired product (**3a**) formed with complete regioselectivity (Table 1, entry 10). Moreover, no homocoupling product of **2a** was observed.

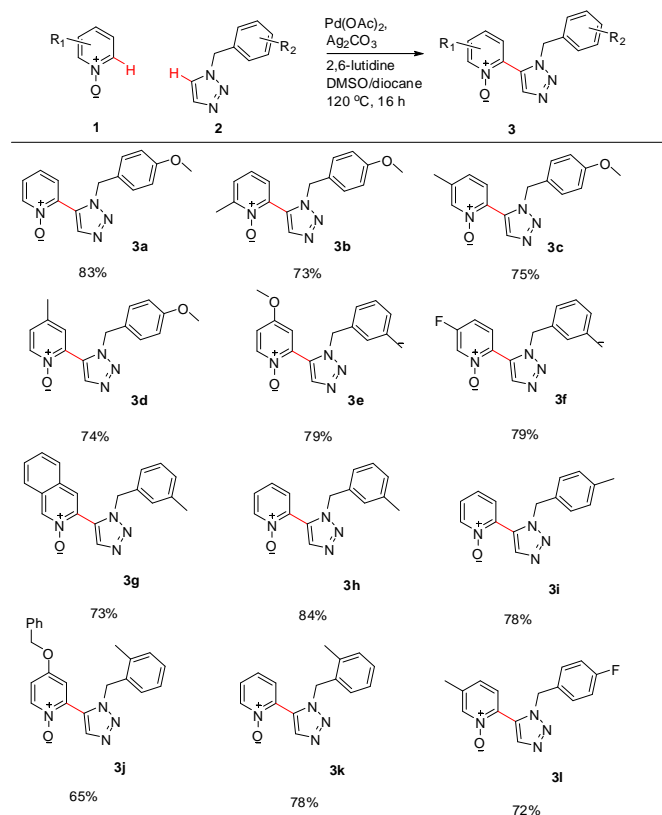
Table 1. Influence of reaction conditions on the reaction of **1a** with **2a**^a

entry	oxidant (2 equiv)	ligand	solvent	yield of 3a (%) ^c
1	Cu(OAc) ₂	pyridine	dioxane	15
2	FeCl ₃	pyridine	dioxane	trace
3	Ag ₂ O	pyridine	dioxane	55
4	Ag ₂ SO ₄	pyridine	dioxane	trace
5	AgNO ₃	pyridine	dioxane	trace
6	AgOAc	pyridine	dioxane	46
7	Ag ₂ CO ₃	pyridine	dioxane	53
8	Ag ₂ CO ₃	Phen	dioxane	45
9	Ag ₂ CO ₃	2,6-lutidine	dioxane	66
10	Ag ₂ CO ₃	2,6-lutidine	5%DMSO/dioxane	83 (65) ^b
11	Ag ₂ CO ₃	2,6-lutidine	10%DMSO/dioxane	51
12	Ag ₂ CO ₃	2,6-lutidine	20%DMSO/dioxane	43
13	Ag ₂ CO ₃	2,6-lutidine	DMSO	36
14	Ag ₂ CO ₃	2,6-lutidine	DMF	43
15	Ag ₂ CO ₃	2,6-lutidine	NMP	47
16	Ag ₂ CO ₃	2,6-lutidine	NMP	trace

^aReaction conditions: **1a** (0.5 mmol), **2a** (0.55 mmol), Pd(OAc)₂ (0.025 mmol), ligand (0.15 mmol) and oxidant (1.0 mmol) in 3 mL of solvent at 120 °C for 16 h. ^bno 2,6-lutidine was added. ^cIsolated yields

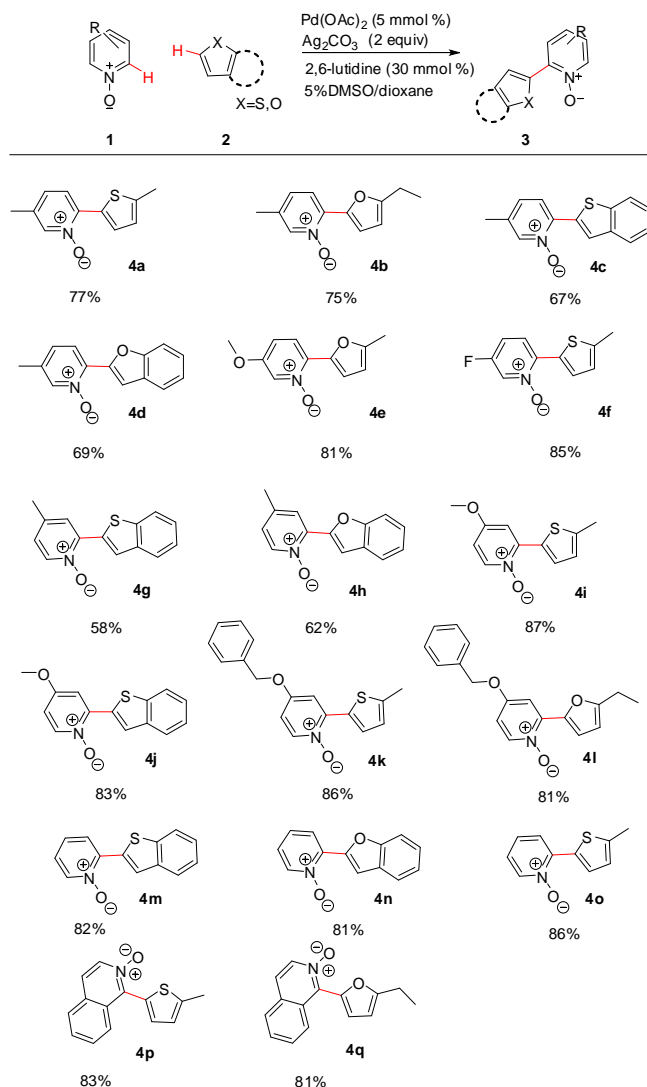
With the optimized conditions (Table 1, entry 10), We then tested the protocol for the cross-coupling of other pyridine *N*-oxides and 1-benzyl-1,2,3-triazoles (Table 2). As outlined, a wide range of various substituted pyridine *N*-oxides, thereby allowing a linkage of 2-methyl (**3b**), 3-methyl (**3c**, **3l**), 4-methyl (**3d**), 4-methoxyl (**3e**), 4-benzyloxyl (**3j**), 3-fluoro(**3f**) gave the desired products in good yields. Furthermore, the above heteroarylation also reacts with other biologically important heteroarene cores, such as isoquinoline *N*-oxide, which only occurred at the C3 position with a yield of 73% (**3g**). No desired cross-coupling product was obtained when 4-nitropyridine and quinoline *N*-oxides were used as coupling partners. Notably, groups at the N-1 position of the triazole ring bearing *p*-methoxybenzyl (**3a** to **3d**), *p*-tolylbenzyl (**3i**), *m*-tolylbenzyl (**3e** to **3h**), *o*-tolylbenzyl (**3j**, **3k**) and *p*-fluorobenzyl (**3l**) can also be used for this reaction. Moreover, no reaction occurred under the same conditions if 1-phenyl-1,2,3-triazole reacts with pyridine *N*-oxide (**1a**).

Table 2. Scope of CH/CH cross-coupling of pyridine *N*-oxides with 1,2,3-triazoles^{a,b}

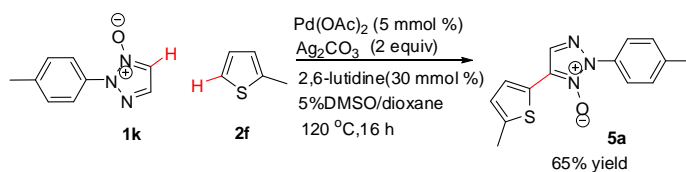


^aReaction conditions: **1** (0.5 mmol), **2** (0.55 mmol), Pd(OAc)₂ (0.025 mmol), Ag₂CO₃ (1 mmol), and 2,6-lutidine (0.15 mmol) in 5%DMSO/1,4-dioxane (3 mL) at 120 °C for 16 h. ^bIsolated yields

Using the above conditions, we also found that the cross-coupling reaction occurred when using thiophenes and furans instead of 1-benzyl-1,2,3-triazoles as substrate. Then we explored the scope of the cross-coupling reactions of pyridine *N*-oxides with thiophenes and furans (Table 3). Diverse decorated products **4a-4q** formed in moderate to good yields, and regioisomeric products were not observed. Notably, the high yields of product **4**, which originates from reaction of 3-substituted pyridine *N*-oxide with an electron-withdrawing group (**4f**) and 4-substituted pyridine *N*-oxide with an electron-donating group (**4e**, **4i** to **4l**), was obtained. Furthermore, thiophene and furan derivatives, namely, including 2-methylthiophene, 2-methylfuran, 2-ethylfuran, benzothiophene, and benzofuran can also be used for this reaction. The desired cross-coupling product was not obtained when 2-methylpyridine and 4-nitropyridine *N*-oxide were used as coupling partners. Surprisingly, the above heteroarylation also proceeded well with isoquinoline *N*-oxide (**4p**, **4q**), but the reaction only occurred at the C1 position of isoquinoline *N*-oxide in good yields. Notably, the desired product was not observed under the above conditions when quinoline *N*-oxide was used as substrate.

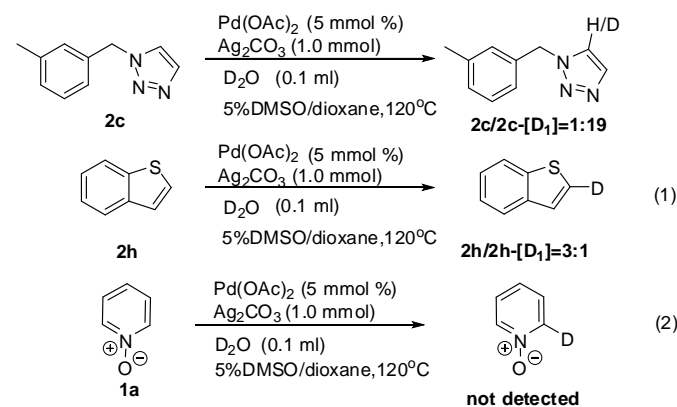
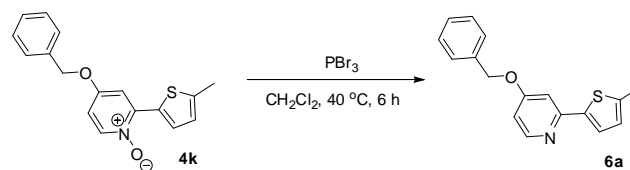
Table 3. Scope of CH/CH cross-coupling of pyridine *N*-oxides with thiophen or furan^{a,b}

^aReaction conditions: **1** (0.5 mmol), **2** (0.55 mmol), Pd(OAc)₂ (0.025 mmol), Ag₂CO₃ (1 mmol), and 2,6-lutidine (0.15 mmol) in 5% DMSO/1,4-dioxane (3 mL) at 120 °C for 16 h. ^bIsolated yields.

**Scheme 1.** C-H/C-H oxidative cross-coupling between 1,2,3-triazole *N*-oxides with 2-methylthiophen

Interestingly, under the same catalytic conditions, the cross-coupling product (**5a**) between 2-substituted 1,2,3-triazole *N*-oxides and 2-methylthiophen was formed with a moderate yield of 65% (Scheme 1).

To gain insight into the reaction mechanism, the H/D exchange control experiments for five-membered heterocycles **2c** and **2h** were performed. As shown in Scheme 2, moderate and high deuterium incorporation were observed respectively. Moreover, no deuterium incorporation occurred when pyridine *N*-oxide (**1a**) was applied to the above system (Scheme 2). The results indicated that the five-membered heterocycles should undergo a substitution reaction with Pd-catalyst to generate palladium intermediate. We proposed that the cross-coupling reaction should proceed through a plausible catalytic cycle similar to the typical mechanism of the cross-coupling reaction of pyridine *N*-oxide.^[11a,12] Moreover, The results from the above experiments demonstrated that, relative to thiophenes, this catalytic system inverted its high compatibility and reactivity in the cross-coupling reaction of 1-benzyl-1,2,3-triazole as substrates. Further studies are needed to understand the mechanism of the difference position cross-coupling of isoquinoline *N*-oxide with five-membered heterocycles.

**Scheme 2.** The H/D exchange control experiments**Scheme 3.** Deoxygenation for biheterocyclic *N*-oxide

Lastly, heterocyclic *N*-oxide **3k** was easily reduced by PBr₃ to generate the corresponding biheterocycles **6a**, indicating that the cross-coupling reaction is practical for the preparation of biheteroaryl molecules (Scheme 3). Additionally, in contrast to these reported the protocol for direct arylation^[7j] of the 3-substituted pyridine *N*-oxides and isoquinoline *N*-oxide, our catalytic system offers complete regioselectivity.

In summary, a highly efficient and regioselective oxidative cross-coupling for pyridine *N*-oxides with five-membered heterocycles through a two-fold C-H activation has been developed. Moreover, this catalytic system shows good compatibility with numerous synthetically relevant functional groups. We hope that this protocol

may provide insights into the synthesis of unsymmetrical biheteroaryl molecules in material and medical chemistry.

We thank the Natural Science Foundation of China (No. 21272174), the Key Projects of Shanghai in Biomedicine (No. 08431902700), and the Scientific Research Foundation of the State Education Ministry for Returned Overseas Chinese Scholars for the support of this research.

Notes and references

- For reviews, see: (a) K. C. Nicolaou, P. G. Bulger and D. Sarlah, *Angew. Chem., Int. Ed.* 2005, **44**, 4442; (b) J. S. Carey, D. Laffan, C. Thomson and M. T. Williams, *Org. Biomol. Chem.* 2006, **4**, 2337.
- (a) J. Hassan, M. Svignon, C. Gozzi, E. Schulz and M. Lemaire, *Chem. Rev.* 2002, **102**, 1359; (b) M. Hapke, L. Brandt and A. Luetzen, *Chem. Soc. Rev.* 2008, **37**, 2782. (c) D. Zhao, J. You and C. Hu, *Chem. Eur. J.* 2011, **17**, 5466;
- For Stille reaction, see: (a) J. K. Stille, *Angew. Chem. Int. Ed.* 1986, **25**, 508; (b) K. C. Nicolaou, T. K. Chakraborty, A. D. Piscopio, N. Minowa and P. Bertinato, *J. Am. Chem. Soc.* 1993, **115**, 4419; (c) M. D. Shair, T. Y. Yoon, K. K. Mosny, T. C. Chou, S. J. Danishefsky, *J. Am. Chem. Soc.* 1996, **118**, 9509.
- For reviews of Suzuki reaction, see: (a) N. Miyaura and A. Suzuki, *Chem. Rev.* 1995, **95**, 2457; (b) N. Miyaura, *Top. Curr. Chem.* 2002, **219**, 11; (c) G. A. Molander and B. Biolatto, *J. Org. Chem.* 2003, **68**, 4302; (d) F. Bellina, A. Carpita and R. Rossi, *Synthesis*. 2004, 2419.
- For Negishi reaction, see: (a) E. Negishi, *Acc. Chem. Res.* 1982, **15**, 340; (b) F. Lopez-Calahorra, M. Martinez-Rubio, D. Velasco, E. Brillas and L. Juli, *Tetrahedron*. 2004, **60**, 285; (c) K. Lee, C. M. Counciller and J. P. Stambuli, *Org. Lett.* 2009, **11**, 1457.
- For development of stable pyridyl boron derivatives, see: (a) A. Bouillon, J. C. Lancelot, S. O. Santos, J. V. Collot, P. R. Bovy and S. Rault, *Tetrahedron*. 2003, **59**, 10043; (b) P. B. Hodgson and F. H. Salingue, *Tetrahedron Lett.* 2004, **45**, 685; (c) D. M. Knapp, E. P. Gillis and M. D. Burke, *J. Am. Chem. Soc.* 2009, **131**, 6961.
- For direct functionalization of pyridine *N*-oxides, see: (a) L.-C. Campeau, S. Rousseaux and K. Fagnou, *J. Am. Chem. Soc.* 2005, **127**, 18020; (b) J.-P. Leclerc and K. Fagnou, *Angew. Chem. Int. Ed.* 2006, **45**, 7781; (c) S. H. Cho, S. J. Hwang and S. Chang, *J. Am. Chem. Soc.* 2008, **130**, 9254; (d) A. Larivee, J. J. Mousseau and A. B. Charette, *J. Am. Chem. Soc.* 2008, **130**, 52; (e) L.-C. Campeau, B.-L. Megan, J.-P. Leclerc, E. Villemure, S. Gorelsky and K. Fagnou, *J. Am. Chem. Soc.* 2008, **130**, 3276; (f) L.-C. Campeau, D. J. Schipper and K. Fagnou, *J. Am. Chem. Soc.* 2008, **130**, 3266; (g) H. Andersson, M. Gustafsson, R. Olsson and F. Almqvist, *Tetrahedron. Lett.* 2008, **49**, 6901; (h) D. J. Schipper, L.-C. Campeau and K. Fagnou, *Tetrahedron*. 2009, **65**, 3155; (i) D. Schipper, J. M. El-Salfiti, C. J. Whipp and K. Fagnou, *Tetrahedron*. 2009, **65**, 4977; (j) L.-C. Campeau, D. R. Stuart, J.-P. Leclerc, M. Bertrand-Laperle, E. Villemure, H.-Y. Sun, S. Lasserre, N. Guimond, M. Lecavallier and K. Fagnou, *J. Am. Chem. Soc.* 2009, **131**, 3291; (k) M. P. Huestis and K. Fagnou, *Org. Lett.* 2009, **11**, 1357; (l) F. Zhang and X. Duan, *Org. Lett.* 2011, **13**, 6102; (m) H. Zhao, R. Wang, P. Chen, T. G. Brian, M. Ming and W. Zhang, *Org. Lett.* 2012, **7**, 1872; (n) T. Jeffery, J. Myers and Jr. Hanna, *Tetrahedron. Lett.* 2012, **53**, 612; (o) X. Chen, C. Zhu, X. Cui and Y. Wu, *Chem. Commun.* 2013, **49**, 6900; (p) G. Li, C. Jia and K. Sun, *Org. Lett.* 2013, **15**, 5198; (q) S. Zhang, L. Liao, F. Zhang and X. Duan, *J. Org. Chem.* 2013, **78**, 2720; (r) Y. Shen, J. Chen, M. Liu, J. Ding, W. Gao, X. Huang and H. Wu, *Chem. Commun.* 2014, **50**, 4292; (s) O. V. Larionov, D. Stephens, A. Mfuh and G. Chavez, *Org. Lett.* 2014, **16**, 864;
- (a) S. You and J. Xia, *Top. Curr. Chem.* 2010, **292**, 165; (b) C. Liu, H. Zhang, W. Shi and A. Lei, *Chem. Rev.* 2011, **111**, 1780; (c) C. S. Yeung, V. Dong, *Chem. Rev.* 2011, **111**, 1215; (d) S. H. Cho, J. Y. Kim, J. Kwak and S. Chang, *Chem. Soc. Rev.* 2011, **40**, 5068; (e) A. E. Wendlandt, A. M. Suess, S. S. Stahl, *Angew. Chem., Int. Ed.* 2011, **50**, 11062; (f) X. Bugaut, F. Glorius, *Angew. Chem., Int. Ed.* 2011, **50**, 7479; (g) W. Han, A. R. Ofial, *Synlett*, 2011, 1951; (i) D. Zhao, J. You and C. Hu, *Chem. Eur. J.* 2011, **17**, 5466. (h) N. Kuhl, M. N. Hopkinson, J. Wencel-Delord, F. Glorius, *Angew. Chem., Int. Ed.* 2012, **51**, 10236;
- (a) D.-K. Kim, K. Joonseop and H.-J. Park, *Bioorg. Med. Chem. Lett.* 2004, **14**, 2401. (b) A. A. Mariam and H. E. Mohamed, *Molecules*. 2009, **14**, 4406.
- (a) D. Zhao, W. Wang, F. Yang, J. Lan, L. Yang, G. Gao and J. You, *Angew. Chem., Int. Ed.* 2009, **48**, 3296; (b) P. Xi, F. Yang, S. Qin, D. Zhao, J. Lan, G. Gao, C. Hu and J. You, *J. Am. Chem. Soc.* 2010, **132**, 1822. (c) A. D. Yamaguchi, D. Mandal, J. Yamaguchi, K. Itami, *Chem. Lett.* 2011, **40**, 555.
- (a) W. Liu, Y. H. Li, Y. Wang, C. X. Kuang, *Org. Lett.* 2013, **15**, 4682. (b) W. Liu, Y. H. Li, B. Xu, C. X. Kuang, *Org. Lett.* 2013, **15**, 2342.
- M. Kentaro, I. Haruka and M. Atsunori, *J. Am. Chem. Soc.* 2004, **126**, 5074.