

Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

Enantioselective organocatalytic oxa-Michael addition of oximes to β -CF₃- β -disubstituted nitroalkenes: efficient synthesis of β -amino- α -trifluoromethyl alcohol

Feng-Lei Liu, Jia-Rong Chen*, Bin Feng, Xiao-Qiang Hu, Li-Hua Ye, Liang-Qiu Lu and Wen-Jing Xiao*

Received (in XXX, XXX) XthXXXXXXXXXX 20XX, Accepted Xth XXXXXXXXXXXX 20XX

DOI: 10.1039/b000000x

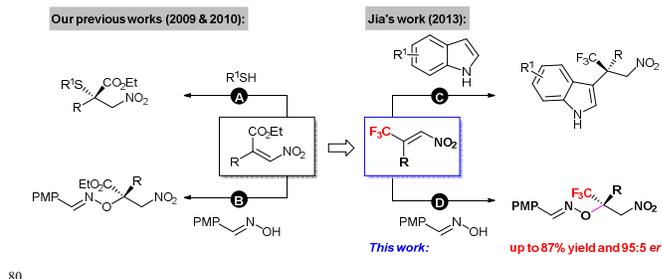
An enantioselective oxa-Michael addition of oxime to β -CF₃- β -disubstituted nitroalkenes catalyzed by a chiral bifunctional cinchona alkaloid-based thiourea has been developed. A variety of trifluoromethylated oxime ethers possessing a tetrasubstituted carbon stereocenter were obtained in good yields with high enantioselectivities.

Trifluoromethylated (CF₃) organic molecules play a primordial role in pharmaceuticals, agrochemicals, and functional materials due to the enhanced metabolic stability, bioavailability, lipophilicity, and binding selectivity upon incorporation of CF₃ group into the parent scaffolds.¹ Accordingly, the development of highly effective methods for the construction of structurally diverse organic molecules bearing a CF₃ moiety has attracted extensive research interest from the synthetic community.² In this context, many elegant examples have been established over the past few decades. Typically, there are two types of strategic construction of trifluoromethylated compounds. One is the use of CF₃-transfer reagents for introduction of CF₃ into organic chemicals,^{2a,3} while the other method is based on the direct functionalization of CF₃-bearing precursors.⁴

Compared with the transition metal-promoted trifluoromethylation reaction, the organocatalytic variants of direct functionalization of CF₃-bearing precursors remain largely unexplored. Recently, Shibata and co-workers have nicely disclosed an enantioselective trifluoromethylation of azamethine imines⁵ and an epoxidation of β -CF₃- β -disubstituted enones⁶ by phase-transfer catalysis, respectively. The Deng group in 2012 reported an unprecedented 9-OH cinchona alkaloid derivatives-catalyzed highly enantioselective isomerization of aryl and alkyl trifluoromethyl imines, affording the corresponding trifluoromethylated amines with high yields and excellent enantiomeric excesses.⁷ Despite these impressive advances, however, the development of efficient organocatalytic methods for the construction of trifluoromethylated quaternary stereogenic centers in a highly enantioselective manner remains a intriguing but challenging task for the organic and medicinal chemists.⁸ Recently, Jia and co-workers documented an elegant nickel-catalyzed asymmetric Friedel-Crafts alkylation reaction between indoles and β -CF₃- β -disubstituted nitroalkenes, leading to the formation of diversely functionalized C-3 substituted indoles with trifluoromethylated all-carbon quaternary stereocenters with high yields and excellent enantioselectivities (Scheme 1C).⁹

On the other hand, organocatalytic oxa-Michael addition

reactions have been recently established as one of the most straightforward tools for the formation of carbon-oxygen bonds with wide application in the synthesis of natural products.¹⁰ For instance, You and co-workers developed a chiral phosphoric acid-catalyzed asymmetric intramolecular oxa-Michael reaction for desymmetrization of cyclohexadienones, providing the corresponding optically active 1,4-dioxane derivatives with excellent yields and enantioselectivities.¹¹ Shortly after, the Ye group has achieved an asymmetric intramolecular oxa-Michael reaction of cyclohexadienones by using primary amine salt as the catalyst.¹² As a class of valuable “soft” oxygen nucleophiles, oximes have proved to be a new type of suitable oxygen-centered nucleophiles in the asymmetric intermolecular conjugate addition.¹³ In this context, our group have developed two types of organocatalytic asymmetric Michael addition reactions of thiols¹⁴ and oximes¹⁵ to trisubstituted nitroacrylates using chiral bifunctional cinchona alkaloid-based thiourea (Scheme 1A and 1B). To our knowledge, however, the enantioselective oxa-conjugate addition reaction with β -CF₃- β -disubstituted nitroalkenes as Michael acceptors has never been explored probably due to their low reactivity and steric encumbrance. Inspired on the development of bifunctional thiourea catalysts in conjugate additions,¹⁶ we have recently achieved an organocatalytic asymmetric intermolecular Michael addition of oximes to β -CF₃- β -disubstituted nitroalkenes. The reaction afforded the corresponding synthetically and biologically significant chiral CF₃-bearing oxime ethers in high yield with good enantioselectivities (Scheme 1D). Herein, we wish to communicate our detailed results.

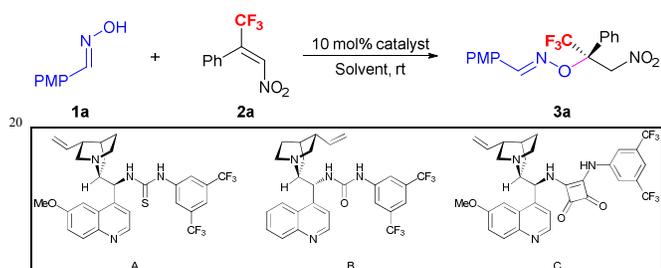


Scheme 1 Enantioselective conjugate addition of β , β -disubstituted nitroalkenes.

At the outset of our studies, we investigated the feasibility of the reaction by choosing 4-methoxybenzaldehyde oxime **1a** and

β -CF₃- β -disubstituted nitroalkene **2a** as the model substrates in the presence of various bifunctional organocatalysts in toluene at room temperature (please see ESI for more details). As shown in Table 1, cinchona alkaloid-based thiourea **A** was identified to be the best catalysts for this reaction, affording the desired product **3a** with moderate yield and enantioselectivity (Table 1, entry 1). With catalyst **A**, we then continued to screen the reaction media and temperature to further improve the efficiency and enantioselectivity. A brief survey of common solvents demonstrated that the reaction in nonpolar solvents provided somewhat superior results (Table 1, entry 1 vs entries 4-8). Further examination of various aromatic solvents showed that mesitylene was the optimal reaction medium and the desired product **3a** was obtained in 91% yield with 92:8 er value (Table 1, entry 10). The enantioselectivity could be slightly improved upon lowering the reaction temperature to 0 °C (Table 1, entry 11).

Table 1 Optimization of the reaction conditions^a



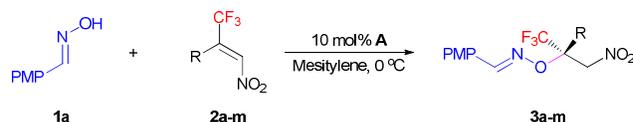
Entry	Solvent	Catalyst	<i>t</i> (h)	Yield ^b (%)	e.r. ^c
1	toluene	A	72	75	88:12
2	toluene	B	96	71	30:70
3	toluene	C	105	54	84:16
4	DCE	A	91	88	54:46
5	Et ₂ O	A	104	54	83:17
6	THF	A	115	8	60:40
7	CH ₃ CN	A	120	46	40:60
8	<i>n</i> -hexane	A	62	69	76:24
9	chlorobenzene	A	50	88	83:17
10	mesitylene	A	50	91	92:8
11 ^d	mesitylene	A	72	75	93:7

^a Unless noted, reactions were carried out with **1** (0.40 mmol), **2a** (0.20 mmol) and catalyst (10 mol%) in solvent (1.0 mL) at rt. ^b Isolated yield. ^c Determined by chiral HPLC. ^d Conducted at 0 °C.

Under the optimized conditions, the generality of this reaction was then evaluated using a variety of β -CF₃- β -disubstituted nitroalkenes, and representative results are highlighted in Table 2. In addition to **2a**, the oxa-Michael reaction tolerated a wide range of β -CF₃- β -disubstituted nitroalkenes bearing Me, Cl, Br, and CF₃ at *meta* or *para* positions on the benzene ring, giving the corresponding products with good yields (67-87%) with er values (87:13 to 95:5, Table 2, entries 2-9). Surprisingly, the reaction with nitroalkene **2f** bearing a Cl at *ortho* position of the phenyl ring led to no formation of the corresponding product **3f**, which was also observed by Jia in asymmetric Friedel-Crafts alkylation reaction of indoles with β -CF₃- β -disubstituted nitroalkenes (Table 2, entry 6). Moreover, β -naphthyl, heteroaryl and alkyl substituted β -CF₃- β -disubstituted nitroalkenes also reacted well with 4-methoxybenzaldehyde oxime **1a** under the optimal conditions (Table 2, entries 11-13), while the relatively less sterically hindered alkyl substituted substrate **2m** resulted in a

significant decrease of enantioselectivity (Table 2, entry 13). Fortunately, the use of cinchonidine-derived thiourea catalyst increased the enantiomeric ratio from 73:27 to 85:15 (Table 2, entry 14).

Table 2 Substrate scope of β -CF₃- β -disubstituted nitroalkenes^a



Entry	R	Product	Yield ^b (%)	e.r. ^c
1	Ph (2a)	3a	75	93:7
2	3-MePh (2b)	3b	74	93:7
3	4-MePh (2c)	3c	70	90.5:9.5
4	3,5-Me ₂ Ph (2d)	3d	77	95:5
5 ^d	4-MeOPh (2e)	3e	71	92:8
6	2-ClPh (2f)	3f	n.d. ^e	-
7	3-ClPh (2g)	3g	87	93.5:6.5
8	4-ClPh (2h)	3h	67	89:11
9	4-BrPh (2i)	3i	86	87:13
10	4-CF ₃ (2j)	3j	80	79:21
11	2-naphthyl (2k)	3k	83	91:9
12	3-thienyl (2l)	3l	69	90.5:9.5
13	2-phenylethyl (2m)	3m	71	73:27
14 ^f	2-phenylethyl (2m)	3m	52	85:15

^a Unless noted, reactions were carried out with **1** (0.40 mmol), **2** (0.20 mmol) and **A** (10 mol%) in mesitylene (1.0 mL) at 0 °C for 3 d. ^b Isolated yield. ^c Determined by chiral HPLC. ^d For 4 d. ^e Not detected. ^f With the use of cinchonidine-derived thiourea as the catalyst, see Table S3 in ESI for more details.

The absolute configuration of the product was unambiguously established to be *S* by X-ray crystallographic analysis of **3e**. Based on the experimental results and previous studies,^{14,16c} we proposed a plausible transition state model to account for the stereochemistry of the product. As shown in Fig. 1, catalyst **A** serves as bifunctional catalyst by activating both reaction partners, wherein the tertiary amine deprotonates the oxime while the thiourea moiety coordinates with the nitro group of β -CF₃- β -disubstituted nitroalkene through H-bond interaction, thus leading to the preferential attack of oxygen to the *Si* face of β -CF₃- β -disubstituted nitroalkene.

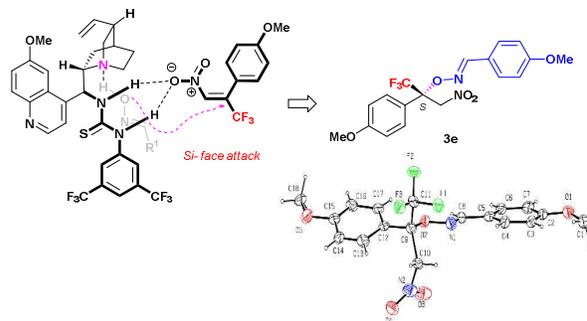
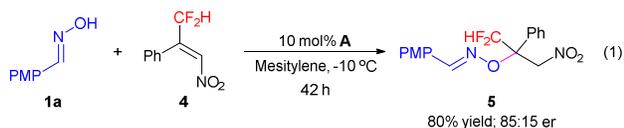


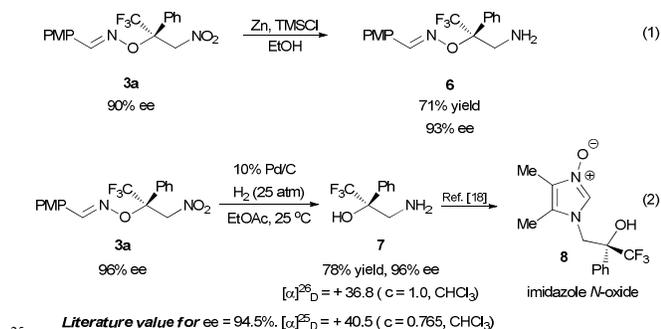
Fig. 1 X-ray crystal structure of product (*S*)-**3e** (thermal ellipsoids are set at 30% probability, Flack parameter = -0.6 (8) and the proposed transition state.

Moreover, the CF₂H moiety is also of key interest in the fields of medicines, agrochemicals, and material chemistry.

Therefore, we also preliminarily examined β -CF₂H- β -disubstituted nitroalkenes as Michael acceptors. For example, the 4-methoxybenzaldehyde oxime **1a** can also react well with β -CF₂H- β -disubstituted nitroalkene **4**, affording the corresponding product **5** in 80% yield with 85:15 er (eqn (1)).



Finally, the synthetic utility of the addition adducts **3** obtained from this transformation was demonstrated in Scheme 3. For example, the adduct **3a** can be conveniently converted into the amine **6** in 71% yield with 93% ee by the reduction of nitro group in the presence of zinc powder [Scheme 2, eqn (1)]. Moreover, the synthetically useful β -amino- α -trifluoromethyl tertiary alcohol **7** could also be expediently obtained in 78% yield with 96% ee by cleavage of the weak N-O bond and reduction of the nitro group [Scheme 2, eqn (2)]. The absolute configuration of **7** was also confirmed by comparison its optical rotation with literatures ($[\alpha]_D^{26} = +36.8$ ($c = 1.0$, CHCl₃)).¹⁷ Importantly, the β -amino- α -trifluoromethyl alcohols of type **7** represent a family of valuable building blocks for the construction of biologically potential chiral CF₃-containing heterocycles. For example, further elaboration of **7** would afford the bioactive imidazole *N*-oxide **8** according to Heimgartner's procedure [Scheme 2, eqn (2)].¹⁸



Scheme 2 Synthetic transformations of product **3a**.

In conclusion, we have developed the first example of organocatalytic asymmetric intermolecular oxa-Michael addition of oximes to β -CF₃- β -disubstituted nitroalkenes, which allows an efficient synthesis of various structurally diverse CF₃-containing oxime ethers with high yields and enantioselectivities. Notably, the products could be easily transformed into synthetically useful chiral building blocks such as β -amino- α -trifluoromethyl tertiary alcohol.²⁰ The reaction advantageously enriches and complements the existing strategies for the construction of chiral CF₃- and CF₂H-containing molecules.

We are grateful to the National Natural Science Foundation of China (No. 21272087, 21202053 and 21232003) and the National Basic Research Program of China (973 program 2011CB808603) for support of this research. We also thank Dr. Xiang-Gao Meng for expert crystallographic analysis.

Notes and references

⁴⁵ Key Laboratory of Pesticide & Chemical Biology, Ministry of Education; College of Chemistry, Central China Normal University, 152 Luoyu Road,

Wuhan Hubei 430079, China; Collaborative Innovation Centre of Chemical Science and Engineering (Tianjin), China. E-mail: chenjiarong@mail.ccnu.edu.cn; wxiao@mail.ccnu.edu.cn; Fax: +86 27 67862041; Tel: +86 27 67862041;

[†] Electronic Supplementary Information (ESI) available: Experimental procedures and compound characterisation data, including X-ray crystal structures of **3e**. For ESI and crystallographic data in CIF or other electronic format See DOI: 10.1039/b000000x/

- 1 (a) P. Jeschke, *ChemBioChem*, 2004, **5**, 571; (b) C. Isanbor and D. O'Hagan, *J. Fluorine Chem.*, 2006, **127**, 303; (c) K. Muller, C. Faeh and F. Diederich, *Science*, 2007, **317**, 1881; (d) W. K. Hagmann, *J. Med. Chem.*, 2008, **51**, 4359; (e) T. Furuya, A. S. Kamlet and T. Ritter, *Nature*, 2011, **473**, 470.
- 2 For selected reviews, see: (a) J.-A. Ma and D. Cahard, *Chem. Rev.*, 2008, **108**, PR1; (b) Y. Zheng and J.-A. Ma, *Adv. Synth. Catal.*, 2010, **352**, 2745; (c) O. A. Tomashenko and V. V. Grushin, *Chem. Rev.*, 2011, **111**, 4475; (d) A. Studer, *Angew. Chem. Int. Ed.*, 2012, **51**, 8950; (e) T. Besset, C. Schneider and D. Cahard, *Angew. Chem. Int. Ed.*, 2012, **51**, 5048; (f) H. Liu, Z.-H. Gu and X.-F. Jiang, *Adv. Synth. Catal.*, 2013, **355**, 617; (g) H.-R. He, Y.-Y. Huang and F. Verpoort, *Acta Chim. Sinica*, 2013, **71**, 700.
- 3 (a) Y. Itoh and K. Mikami, *Tetrahedron*, 2006, **62**, 7199; (b) D. A. Nagib, M. E. Scott and D. W. MacMillan, *J. Am. Chem. Soc.*, 2009, **131**, 10875; (c) D. A. Nagib and D. W. MacMillan, *Nature*, 2011, **480**, 224; (d) G. Valero; X. Companyo and R. Rios, *Chem. Eur. J.*, 2011, **17**, 2018; (e) J.-J. Dai, C. Fang, B. Xiao, J. Yi, J. Xu, Z.-J. Liu, X. Lu, L. Liu and Y. Fu, *J. Am. Chem. Soc.*, 2013, **135**, 8436; (f) S. Seo, J. B. Taylor and M. F. Greaney, *Chem. Commun.*, 2013, **49**, 6385; (g) E. Pair, N. Monteiro, D. Bouyssi and O. Baudoin, *Angew. Chem. Int. Ed.*, 2013, **52**, 5346.
- 4 (a) K. Mikami, Y. Itoh and M. Yamanaka, *Chem. Rev.*, 2004, **104**, 1; (b) J. Nie, H.-C. Guo, D. Cahard and J.-A. Ma, *Chem. Rev.*, 2011, **111**, 455; (c) J.-H. Lin and J.-C. Xiao, *Eur. J. Org. Chem.*, 2011, 4536; (d) F. Grellepois, *J. Org. Chem.*, 2013, **78**, 1127.
- 5 H. Kawai, A. Kusuda, S. Nakamura, M. Shiro and N. Shibata, *Angew. Chem. Int. Ed.*, 2009, **48**, 6324.
- 6 H. Kawai, S. Okusu, Z. Yuan, E. Tokunaga, A. Yamano, M. Shiro and N. Shibata, *Angew. Chem. Int. Ed.*, 2013, **52**, 2221.
- 7 Y. Wu and L. Deng, *J. Am. Chem. Soc.*, 2012, **134**, 14334.
- 8 (a) E. J. Corey and A. Guzman-Perez, *Angew. Chem. Int. Ed.*, 1998, **37**, 388; (b) C. J. Douglas and L. E. Overman, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 5363; (c) C. Jiang and B. M. Trost, *Synthesis*, 2006, 369; (d) C. Hawner and A. Alexakis, *Chem. Commun.*, 2010, **46**, 7295; (e) N. Hara, R. Tamura, Y. Funahashi and S. Nakamura, *Org. Lett.*, 2011, **13**, 1662; (f) Y.-L. Liu, T.-D. Shi, F. Zhou, X.-L. Zhao, X. Wang and J. Zhou, *Org. Lett.*, 2011, **13**, 3826; (g) C. G. Kokotos, *J. Org. Chem.*, 2012, **77**, 1131; (h) Y.-L. Liu, X. Wang, Y.-L. Zhao, F. Zhu, X.-P. Zeng, L. Chen, C.-H. Wang, X.-L. Zhao and J. Zhou, *Angew. Chem. Int. Ed.*, 2013, **52**, 13735.
- 9 J.-R. Gao, H. Wu, B. Xiang, W.-B. Yu, L. Han and Y.-X. Jia, *J. Am. Chem. Soc.*, 2013, **135**, 2983.
- 10 (a) C. F. Nising and S. Brase, *Chem. Soc. Rev.*, 2008, **37**, 1218; (b) C. F. Nising and S. Brase, *Chem. Soc. Rev.*, 2012, **41**, 988.
- 11 Q. Gu, Z.-Q. Rong, C. Zheng and S.-L. You, *J. Am. Chem. Soc.*, 2010, **132**, 4056.
- 12 W. Wu, X. Li, H. Huang, X. Yuan, J. Lu, K. Zhu and J. Ye, *Angew. Chem. Int. Ed.*, 2013, **52**, 1743.
- 13 (a) C. D. Vanderwal and E. N. Jacobsen, *J. Am. Chem. Soc.*, 2004, **126**, 14724; (b) S. Bertelsen, P. Diner, R. L. Johansen and K. A. Jorgensen, *J. Am. Chem. Soc.*, 2007, **129**, 1536; (c) P. Diner, M. Nielsen, S. Bertelsen, B. Niess and K. A. Jorgensen, *Chem. Commun.*, 2007, 3646.
- 14 H.-H. Lu, F.-G. Zhang, X.-G. Meng, S.-W. Duan and W.-J. Xiao, *Org. Lett.*, 2009, **11**, 3946.
- 15 F.-G. Zhang, Q.-Q. Yang, J. Xuan, H.-H. Lu, S.-W. Duan, J.-R. Chen and W.-J. Xiao, *Org. Lett.*, 2010, **12**, 5636.
- 16 For selected reviews, see: (a) A. G. Doyle and E. N. Jacobsen, *Chem. Rev.*, 2007, **107**, 5713; (b) S. J. Connon, *Chem. Commun.*, 2008, 2499; (c) O. V. Serdyuk, C. M. Heckel and S. B. Tsogoeva, *Org. Biomol. Chem.*, 2013, **11**, 7051; for selected examples, see: (d) T. Okino, Y. Hoashi and Y. Takemoto, *J. Am. Chem. Soc.*, 2003, **125**,

- 12672; (e) B. Vakulya, S. Varga and T. Soos, *J. Org. Chem.*, 2008, **73**, 3475; (f) D.-R. Li, A. Murugan and J. R. Falck, *J. Am. Chem. Soc.*, 2008, **130**, 46; (g) M. Tsakos and C. G. Kokotos, *Tetrahedron*, 2013, **69**, 10199.
- 5 17 (a) F. Tur and J. M. Saa, *Org. Lett.*, 2007, **9**, 5079-5082; (b) C. Mioskowski, G. Solladie, *Tetrahedron*, 1973, **29**, 3669.
- 18 G. Młostoń, E. Obijalska and H. Heimgartner, *J. Fluorine Chem.*, 2011, **132**, 951.
- 19 CCDC 963303 (**3e**) contains the supplementary crystallographic data
10 for this paper. These data can be obtained free of charge from The
Cambridge Crystallographic Data Centre via
www.ccdc.cam.ac.uk/daa_request/cif.
- 20 According to one of referees' suggestions, we have also tested the
15 asymmetric Michael addition of thiols (ArSH) to β -CF₃- β -alkyl-
disubstituted nitroalkenes. At the current stage, however, all the
attempts with various chiral bifunctional cinchona alkaloid-based
thioureas lead to only moderate enantioselectivity while with good
yield. Please see the ESI for more details.