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Doping Molecular Organic Semiconductors by Diffusion from the Vapor Phase

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Abstract

Thin films of amorphous small molecule semiconductors are widely used in organic light emitting displays and have promising applications in solar cells and thermoelectric devices. Adding dopants increases the conductivity of organic semiconductors, but high concentrations of dopants can disrupt the film structure, alter the shape of the density of states in the material, and increase the effects of Coulomb interactions on charge transport. Electrical doping of the solution processable hole-transport material 2,2',7,7'-Tetrakis[N,N-di(4-methoxyphenyl)amino]-9,9'-spirobifluorene (spiro-OMeTAD) was studied with 2,3,5,6-Tetrafluoro-7,7,8,8-tetracyanoquinodimethane (F₄TCNQ) as a *p*-type dopant. Infiltration of F₄TCNQ from the vapor phase into films of spiro-OMeTAD provided a route to highly doped films with up to 39 ± 2 mol% doping. Structural characterization confirmed that the films remain amorphous even at the highest doping levels with no apparent phase separation. We quantitatively determined the carrier concentration using UV-Vis spectroscopy to interpret the evolution of the electrical conductivity. Over the range of carrier concentrations ($10^{19} - 10^{20}$ 1/cm³), the electrical conductivity increased no more than linearly with carrier concentration, while the thermopower had a small increase with carrier concentration. The trends in conductivity and thermopower were related to the unique electronic structure of spiro-OMeTAD, which is able to support two carriers per molecule. Temperature-dependent conductivity measurements were used to further analyze the transport mechanism.

Introduction

Small molecule organic semiconductors have become widely used in organic light emitting devices (OLEDs) for information displays.¹ Small molecules can be thermally evaporated to form smooth, glassy layers in devices. Amorphous films of organic semiconductors can also be cast from solvent and are used frequently as electron and hole transport layers in devices such as thin film solar cells.^{2,3} Doping of such transport layers is particularly beneficial to modify their electrical conductivity and to control the energetic line-up of transport levels at interfaces. The role of dopants on the structural ordering of amorphous materials has not been extensively studied,⁴ particularly if there is aggregation induced by charge transfer interactions.

Amorphous small molecule semiconductor films have been found to follow an Arrhenius-type thermally activated hopping conduction model.⁵ The conductivity, σ , is given by Eq. 1 where σ_0 is the conductivity at infinite temperature with all carriers activated, E_a is the hopping activation energy, k_B is the Boltzmann constant, and T is the temperature.

$$\sigma = \sigma_0 \exp\left(\frac{-E_a}{k_B T}\right) \quad (1)$$

While Eq. 1 describes hopping conduction well, it does not provide any insight into the shape of the density of states (DOS) of an amorphous semiconducting material. The DOS of amorphous molecular semiconductors is usually modeled with a Gaussian form, but the changes in the shape of the DOS with doping are not well understood. Arkhipov, et al. proposed a DOS model where a Gaussian distribution of host states is broadened by an exponential tail of trap states.⁶ However, this model is only formulated for dopant concentrations up to $10^{18} - 10^{19} \text{ cm}^{-3}$. Recent research on doped small molecule semiconductors has suggested that, at higher dopant concentrations, dopants no longer act as Coulombic traps.⁷ The charge transfer between the host molecules and dopants causes the ionized hosts and dopants to form integer charge transfer complexes (ICTCs) bound with Coulomb forces. This Coulomb interaction modifies the energy level of the ionized host molecules. At 10 mol% doping, the concentration of ICTCs is high enough that carriers will preferentially hop among the Coulomb-modified ICTC states. These distributions of ICTC states were described by the width of their static disorder, σ_{ICTC} . E_a was found to be related to both σ_{ICTC} and the relaxation energy of the host molecule after ionization.⁷ Kinetic Monte Carlo simulations have also treated the DOS of small molecule semiconductors as two distributions of states, neutral host states and ionized host states. The conductivity of some small molecule semiconductors

increases superlinearly with doping,^{8,9} and these simulations found that a reduction in disorder from adding dopants may explain this superlinear increase.¹⁰

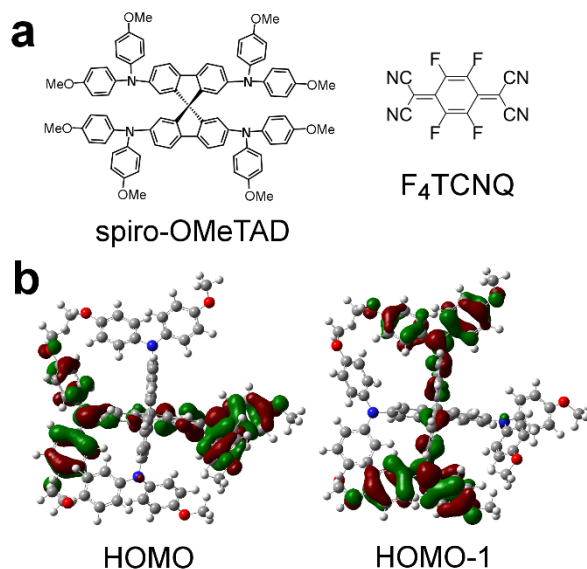


Figure 1. (a) Chemical structure of spiro-OMeTAD and F₄TCNQ and (b) DFT calculations of spiro-OMeTAD's nearly degenerate HOMO and HOMO-1. Perpendicular fluorene units are both seen edge-on in (b).

One example of a solution-processable small molecule semiconductor that forms amorphous films is 2,2',7,7'-tetrakis[N,N-di(4-methoxyphenyl)amino]-9,9'-spirobifluorene (spiro-OMeTAD) (Figure 1a). This material is commonly used as a hole transport layer in perovskite solar cells^{3,11} and dye sensitized solar cells.^{12,13} In solar cells, spiro-OMeTAD is typically doped with LiTFSI, tert-butylpyridine, and oxygen to a conductivity $\sim 10^{-5}$ S/cm at 12-30 mol% LiTFSI.¹⁴ As shown in Figure 1a, the core of the spiro-OMeTAD molecule has two fluorene units connected by a spiro-bridged carbon atom. Because of this spiro bridge, the two fluorene units are perpendicular to each other, and the molecular structure frustrates crystallization in thin films. Spiro-OMeTAD can be crystallized, leading to a triclinic $P\bar{1}$ structure with only short-range π - π overlap between adjacent molecules.^{15,16} The spiro-bridge also leads to a unique electronic structure relative to other materials. Density functional theory (DFT) calculations using the long-range corrected hybrid functional ω B97X-D reveal that the HOMO and HOMO-1 orbitals of spiro-OMeTAD in the experimental geometry from a single crystal structure are nearly degenerate and localize onto opposite sides of the spiro bridge.^{15,16} Our ω B97X-D/6-31G(d) calculations on an

isolated molecule in vacuum without symmetry constraints show similar localization of the HOMO and HOMO-1 on the asymmetric structure that is likely present in amorphous films (Figure 1b). Such long-range corrected hybrid DFT functionals are helpful to more accurately model the electronic structure of mixed-valence spiro compounds than functionals such as B3LYP that overly delocalize and mix the molecular orbitals.¹⁷ The impact of this near degeneracy on charge transport has not been examined in detail.

Here we examine spiro-OMeTAD as a model system to study charge transport in amorphous small molecule semiconductor films at high carrier concentrations. It has the advantage of being readily cast from solution into amorphous thin films and has a unique electronic structure relative to many molecular semiconductors. We doped spiro-OMeTAD films with small molecule dopant 2,3,5,6-tetrafluoro-7,7,8,8-tetracyanoquinodimethane (F₄TCNQ) by infiltration from the vapor phase leading to a maximum of 39 ± 2 mol% F₄TCNQ films. The electron affinity of F₄TCNQ at 5.24 eV¹⁸ is close to Spiro-OMeTAD's first and second ionization energies, $\approx 5.1 - 5.3$ eV.¹² This close energy match between the dopant and host allowed us to study the effects of spiro-OMeTAD's two hopping sites on the material's electronic transport. We found that Seebeck coefficient and temperature-dependent conductivity measurements are consistent with a Gaussian density of charged states modified by the presence of spiro-OMeTAD²⁺ states.

Results & Discussion

We doped spun-cast spiro-OMeTAD films by diffusing F₄TCNQ vapor into the films, or “vapor doping” them with F₄TCNQ. We and others have previously used this vapor-doping process to dope semiconducting polymers, including PBTTT and P3HT,^{19,20} but we are not aware of its use for molecular systems. Vapor doping differs from methods, such as thermal co-evaporation of host and dopant molecules or spin casting of doped solutions. Casting the neutral semiconductor prior to the doping process eliminates difficulties in the solubilization of the doped form of the semiconductor and its associated counterion that can preclude achieving high levels of doping. For the vapor doping process, the spiro-OMeTAD films were placed face-down in the top of a doping chamber in a N₂ atmosphere glovebox. A small amount of F₄TCNQ was placed in the bottom of the doping chamber. When the doping chamber is heated on a hot plate at 200 °C, the F₄TCNQ sublimates and can diffuse into the film. The glass transition temperature, T_g, of spiro-OMeTAD

is relatively high at 124 °C.²¹ In our apparatus, the substrate temperature is not fixed and after 15 min of heating, the substrate reaches 92.5 °C, and ~106 °C after 90 min of heating, so T_g of spiro-OMeTAD is never exceeded during the doping process (See Supplementary Information for full heating curve).

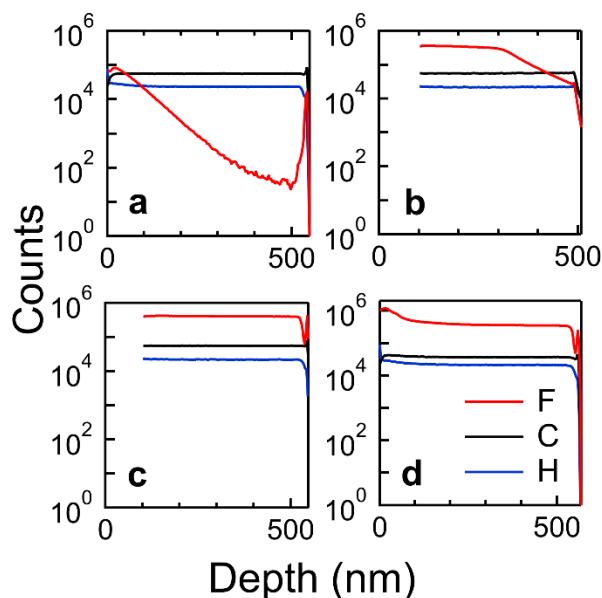


Figure 2. Secondary ion mass spectroscopy plots tracking the depth distribution of fluorine (red), carbon (black), and hydrogen (blue) in spiro-OMeTAD films vapor doped with F₄TCNQ for (a) 15 min, (b) 30 min, (c) 45 min, and (d) 30 min with 10 min annealing. Spikes in the plots at the bottom of the film are an artefact caused by the primary ion beam hitting the insulating quartz substrates.

We determined whether F₄TCNQ had diffused through the full depth of the film during vapor doping using secondary ion mass spectroscopy (SIMS). Fluorine atoms provide a unique marker for the presence of F₄TCNQ; we note that due to the high fluorine signal level, the amount of F₄TCNQ could not be accurately quantified relative to other atoms using our SIMS instrument. In Figure 2a, 15 min of vapor doping leads to a concentration profile where the concentration of F₄TCNQ is greatest near the surface of the film and decreases to the instrument noise level at the bottom of the film. After 30 min of vapor doping, Figure 2b, F₄TCNQ has nearly diffused through the full depth of the film. F₄TCNQ can fully diffuse through the film with a roughly even

concentration profile after 45 min of vapor doping, as seen in Figure 2c. To test whether the uneven concentration profiles caused by vapor doping could be leveled through annealing, we annealed a 30 min vapor doped film for 10 min at 140 °C. This annealing temperature was chosen because it is above spiro-OMeTAD's glass transition temperature of 124 °C but below the temperature at which F₄TCNQ sublimates from films, 150 °C.²² Figure 2d shows that the concentration profile of the annealed film was more even at the bottom of the film than the unannealed film (Figure 2b). Ten minutes was long enough for F₄TCNQ to diffuse through the film without much de-doping but was not long enough for the film to crystallize. In the 45 and 75 min doped films, crystallization was observed around the edges of the film after annealing. Films used for charge transport measurements were annealed to reduce concentration gradients, and partially crystallized areas of annealed films were not measured.

We used grazing incidence wide angle X-ray scattering (GIWAXS) to determine if either spiro-OMeTAD or F₄TCNQ crystallized (or co-crystallized) during the doping procedure. In the scattering from the undoped film, Figure 3a, a broad, nearly isotropic ring of scattering was observed around $q \approx 1.25 \text{ \AA}^{-1}$. This indicates that the film has an amorphous structure, as expected for spiro-OMeTAD, which is known to generally form glassy films. In Figure 2b, the scattering for an 80 min F₄TCNQ vapor doped film showed a similar, nearly isotropic scattering pattern and no signatures of scattering at *d*-spacings known for crystalline form of F₄TCNQ.²³ Because this 80 min doping time is higher than the doping times used for charge transport measurements, this scattering indicates that the doped films retain their amorphous structure throughout the vapor doping process, as opposed to forming a co-crystal or phase separating.

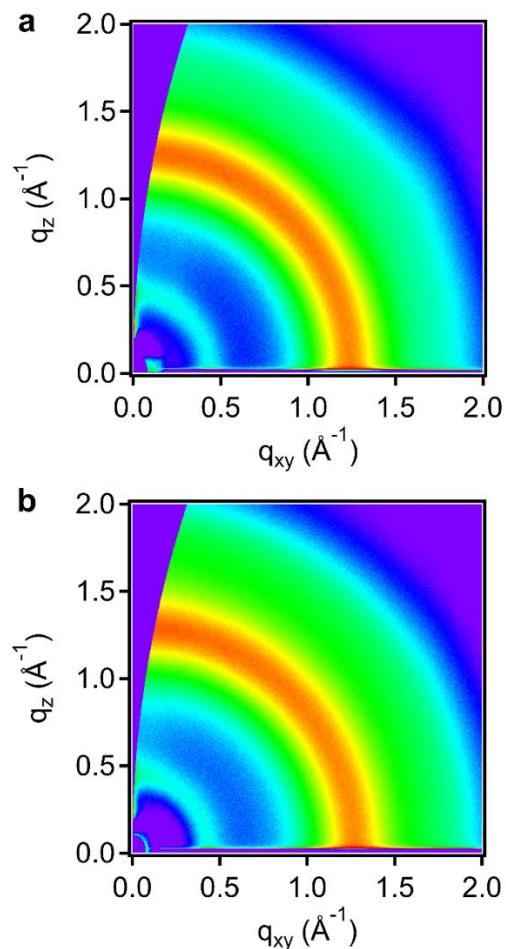


Figure 3. Grazing incidence wide angle X-ray scattering images of (a) neat and (b) 80 min F₄TCNQ vapor doped spiro-OMeTAD.

To understand the nature of the charge transfer process between spiro-OMeTAD and F₄TCNQ, we measured the UV-Vis-NIR absorbance as a function of doping time (Figure 4). The absorbance spectrum of undoped spiro-OMeTAD has an onset around 3 eV with weak sub-optical gap features in the spectrum due to interference effects from the ~500 nm thick film. Upon doping, new features were observed in the spectra that confirm that integer charge transfer (ICT) occurred between F₄TCNQ and spiro-OMeTAD. Two spectra regions have features that can be assigned to the F₄TCNQ anion: the vibronic series between 1.3 – 2 eV and the peak at 2.9 eV.²⁴ The features centered around 0.8 eV and 2.5 eV can be assigned to transitions of spiro-OMeTAD⁺ or spiro-OMeTAD²⁺. Spiro-OMeTAD⁺ and spiro-OMeTAD²⁺ have an additional weak transition centered at 1.77 eV.²⁵ In our spectra, that peak is obscured by the stronger F₄TCNQ anion absorbance. Due

to the degenerate HOMO and HOMO-1 of spiro-OMeTAD, spiro-OMeTAD⁺ still has a transition at 3 eV. At high doping levels, the feature centered around 0.8 eV becomes less symmetric and shows a blueshift. This blueshift suggests that some population of spiro-OMeTAD molecules has been oxidized to the 2+ state by two F₄TCNQ molecules. The energy gap between the HOMO-1 and HOMO levels was calculated to shift from 0.37 to 0.95 eV when spiro-OMeTAD⁺ is oxidized to the 2+ triplet state.²⁶ Because spiro-OMeTAD is a mixed-valence compound with two degenerate oxidation states, spiro-OMeTAD²⁺ may be in a triplet state or an open-shell singlet state.

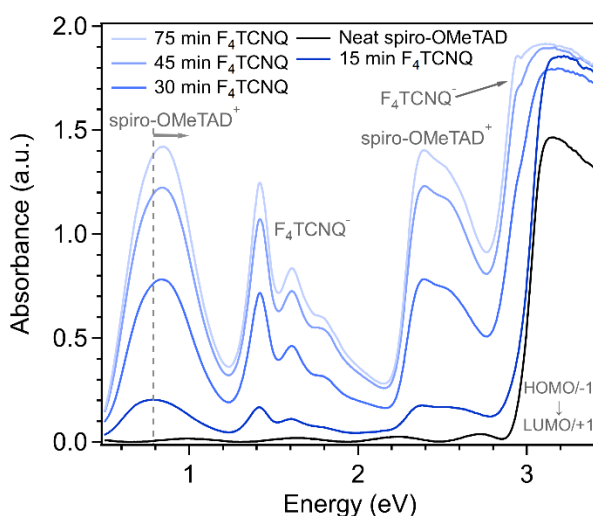


Figure 4. UV-Vis-NIR spectra of undoped (black) and doped (shades of blue) spiro-OMeTAD films. New peaks in the visible and NIR regions of the doped film spectra indicate the presence of F₄TCNQ anions and spiro-OMeTAD cations (labeled in gray) and confirm integer charge transfer in the films.

Because spiro-OMeTAD and F₄TCNQ undergo integer charge transfer, we used UV-Vis-NIR spectra to estimate the charge carrier concentration in the doped films. In disordered organic semiconductors with hopping conduction, the carrier concentration cannot be measured readily using the Hall effect. Instead, we used the absorbance of F₄TCNQ⁻ to estimate the hole concentration. The area under the F₄TCNQ⁻ peaks between 1.25 – 2.15 eV is proportional to the concentration of F₄TCNQ⁻ anions in the films. We fit the three F₄TCNQ⁻ peaks and the spiro-OMeTAD⁺ transition at 1.77 eV to determine the area of the F₄TCNQ⁻ absorbance. We used

reference spectra of $[\text{Cp}^*_2\text{Co}][\text{F}_4\text{TCNQ}]$ and $[\text{Cp}^*_2\text{Fe}][\text{F}_4\text{TCNQ}]$ solutions to calculate a reference value of the molar extinction coefficient, ϵ of the F_4TCNQ^- optical absorbance. With the F_4TCNQ^- absorbance, film thickness, and molar extinction coefficient, the concentration of F_4TCNQ^- in each film can be calculated. (See SI for full description of fitting methods) The number of F_4TCNQ^- is equivalent to the number of electrons removed from spiro-OMeTAD and spiro-OMeTAD⁺. The UV-Vis spectra cannot easily be used to distinguish the relative amounts of spiro-OMeTAD⁺ and spiro-OMeTAD²⁺, therefore it is difficult to measure how the holes are distributed on the two possible oxidation sites of a spiro-OMeTAD molecule. While some spiro-OMeTAD⁺ polarons could condense to spiro-OMeTAD²⁺ bipolarons, the shape of the NIR peak suggests that both could be present in the doped films.

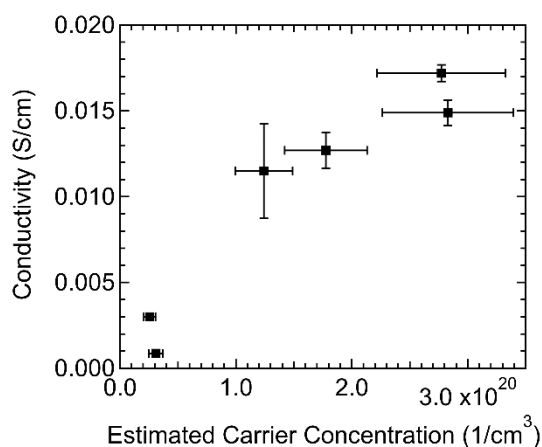


Figure 5. Room temperature conductivity increases with the estimated carrier concentration in the spiro-OMeTAD: F_4TCNQ films. The error in the conductivity is the standard deviation of three measurements taken at different locations of the same film. The error in the carrier concentration is estimated at 20%.

On average, films doped to our maximum doping time of 75 min reached a carrier concentration of $3.3 \pm 0.3 \times 10^{20} \text{ cm}^{-3}$, which is estimated to correspond to a film composition of $39 \pm 2 \text{ mol}\%$ F_4TCNQ based on the concentration of F_4TCNQ^- determined by UV-Vis and the density of spiro-OMeTAD films.²⁷ (see Supplementary Information for details) Because the vapor doping process adds mass to the spiro-OMeTAD films, the thickness or density of the films can change. We found that film thicknesses increased from $\approx 420 \pm 20 \text{ nm}$ for undoped films to ≈ 530

± 30 nm for films doped for 45 and 75 min. With this increase in film thickness, the density of spiro-OMeTAD hopping sites would decrease. Based on a spiro-OMeTAD film density of 1.02 g/cm^3 ,²⁷ the molecule density of spiro-OMeTAD films is $5.02 \times 10^{20} \text{ cm}^{-3}$. Because the HOMO and HOMO-1 of spiro-OMeTAD are nearly degenerate, each molecule has two potential electronic states, giving a hopping site density of $1 \times 10^{21} \text{ cm}^{-3}$.

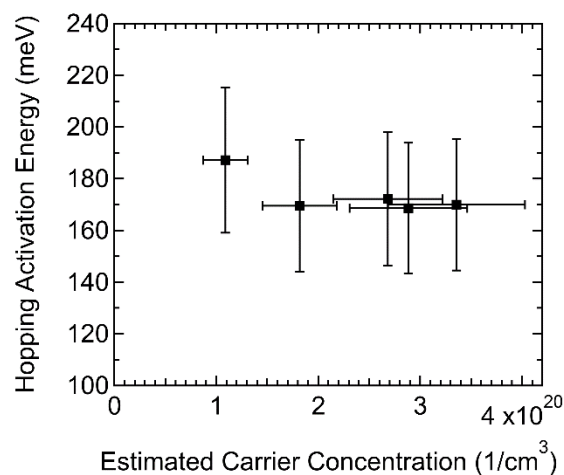


Figure 6. Hopping activation energy (E_a) was roughly constant with carrier concentration in this range. Error in E_a was estimated at 15%, while error in the carrier concentration was estimated at 20%.

We measured the room temperature conductivity of several samples with different doping levels. We found that the conductivity generally increases with doping level, with a maximum of 0.017 S/cm (Figure 5). This value is two orders of magnitude higher than the conductivity reached by typical LiTFSI doping¹⁴ and similar to the 0.024 S/cm reported for LiTFSI and benzoyl peroxide doping.²⁸ Estimating with $\sigma = qn\mu$, we found that the electrical mobility may decrease but remains on the order of $10^{-4} \text{ cm}^2/\text{V}\cdot\text{s}$. (see Supplementary Information Figure S8) The mobility of these films cannot be measured directly because they are too disordered for Hall effect measurements, and field-effect transistor and space charge limited current (SCLC) mobility measurements are not accurate on highly doped films. For comparison, the Mott-Gurney SCLC mobility of undoped glassy spiro-OMeTAD was $2.3 \times 10^{-4} \text{ cm}^2/\text{V}\cdot\text{s}$.²⁹ Many doped small molecule systems show a superlinear increase in conductivity with doping, such as MeO-TPD:F₆TCNNQ's increase over a dopant:host density ratio range of 0.01 – 0.29.⁹ In our measurements of spiro-OMeTAD:F₄TCNQ,

the increase in conductivity with hole concentration is no greater than linear, which could be attributed to this decrease in hole mobility.

To find the activation energy of hopping, E_a , in this system, we measured the conductivity of films with varying doping levels over a temperature range of at least 140 – 295 K (see Supplementary Information Figure S9). The dependence of the electrical conductivity on temperature followed the Arrhenius relationship of Eq. 1 with a nearly constant $E_a \approx 170$ meV (Figure 6). E_a decreases with increasing doping level and can level off at high doping levels.⁷ The constant (or slightly decreasing) trend of E_a in this carrier concentration range is consistent with E_a leveling off at high doping levels. A broad range of *p*-type materials has been reported to have E_a values 171 – 295 meV.⁷ Our measurements of E_a were on films with greater than 10 mol% doping, but spiro-OMeTAD:F₄TCNQ is among the systems with lower E_a values. For comparison, 2,2',7,7'-tetrakis(*N,N*-diphenylamino)-9,9-spirobifluorene (spiro-TAD):F₆TCNNQ system has similar physical and electronic properties to spiro-OMeTAD:F₄TCNQ and its E_a is 180 meV at 10 mol% doping.³⁰ Both spiro-TAD:F₆TCNNQ and spiro-OMeTAD:F₄TCNQ have very small host IE-dopant EA offsets ~ 0.1 eV. Interestingly the electrical conductivity of spiro-OMeTAD:F₄TCNQ is significantly higher than that of spiro-TAD:F₆TCNNQ despite the comparable value of E_a .

One model proposes that electrical conduction in highly doped small molecule organic semiconductors primarily takes place among host-dopant charge transfer complexes, which form a DOS with width σ_{ICTC} .⁷ Based on data from a wide variety of *p*- and *n*-type host:dopant systems at 10 mol% doping, a relationship between σ_{ICTC} determined by ultraviolet photoemission spectroscopy (UPS) and E_a , as well as the material's carrier density, was determined. Using the reported empirical relationship between E_a and σ_{ICTC} , doped films of spiro-OMeTAD are expected to have a low σ_{ICTC} . Spiro-OMeTAD has a low molecular density in doped films of $5 \times 10^{20} \text{ cm}^{-3}$, similar to other materials that are reported to have a low σ_{ICTC} . The roughly constant E_a of spiro-OMeTAD:F₄TCNQ over this doping range suggests that the σ_{ICTC} disorder in the system is roughly constant over this doping range, as well.

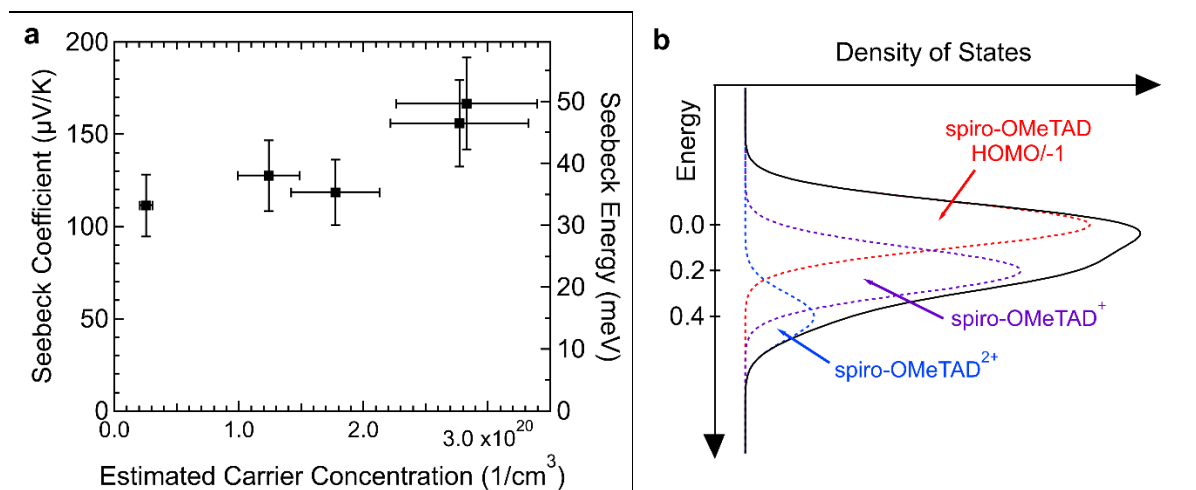


Figure 7. (a) In-plane Seebeck coefficient was generally constant over this carrier concentration range but increased slightly at $\approx 2.8 \times 10^{20} \text{ 1/cm}^3$. Error in the Seebeck coefficient measurement was estimated at 15%, while the estimated error in the carrier concentration was 20%. (b) Schematic the electronic DOS of doped spiro-OMeTAD showing the estimated energies to remove an electron from the neutral molecule, spiro-OMeTAD⁺ and spiro-OMeTAD²⁺. These states are close enough in energy that the DOS of states is likely broadened by ionized states at high levels of doping. This schematic assumes Gaussian widths of 0.1 eV spaced 0.2 eV apart.

We measured the in-plane Seebeck coefficient, S , of the doped spiro-OMeTAD films with varying carrier concentrations at room temperature to assess how the shape of the electronic DOS changes with doping (Figure 7a). Because spiro-OMeTAD has two potential sites for charge hopping, it is of interest to determine if there are signs that both oxidation states are involved in charge transport. The Seebeck coefficient is related to the shape of the DOS through the Mott formalism,³¹

$$S = -\frac{k_B}{e} \int \left(\frac{E - E_F}{k_B T} \right) \frac{\sigma_E}{\sigma} \left(\frac{\partial f}{\partial E} \right) dE \quad (2)$$

where σ_E is the transport function, e is the electronic charge, E is energy, E_F is the Fermi energy, and f is the Fermi function. In hopping semiconductors, Eq. 2 becomes,

$$S = \frac{k_B}{e} \left(\frac{E_{tr} - E_F}{k_B T} \right) \quad (3)$$

where E_{tr} is the transport energy. In a p -type Gaussian DOS, E_{tr} is at an energy slightly above the peak of the DOS. The Seebeck coefficient will increase when the distance between E_{tr} and E_F increases.³²

We found that S was roughly constant in this carrier concentration range with a small increase at high carrier concentrations $\approx 2.8 \times 10^{20} \text{ cm}^{-3}$. The three points below $2 \times 10^{20} \text{ cm}^{-3}$ have an average value of $119 \mu\text{V/K}$, while the values at higher concentrations are 156 and $167 \mu\text{V/K}$. Typically, the Seebeck coefficient of small molecule semiconductors decreases with increasing dopant concentration.⁹ Because the Seebeck constant is related to the distance between E_{tr} and E_{F} by Eq. 3, $E_{\text{tr}}-E_{\text{F}}$ in spiro-OMeTAD:F₄TCNQ may have increased at $\approx 2.8 \times 10^{20} \text{ cm}^{-3}$. The increase in $E_{\text{tr}}-E_{\text{F}}$ could be caused by a change in DOS shape by spiro-OMeTAD²⁺ states. Upon doping, a Gaussian density of spiro-OMeTAD⁺ states forms that overlaps with the Gaussian DOS of spiro-OMeTAD. As the carrier concentration increases, it is possible that bipolarons could form, further extending the tail of the DOS and potentially increasing $E_{\text{tr}}-E_{\text{F}}$. This mechanism of increasing the Seebeck coefficient has been observed and modeled in polymer blends.^{32,33} Figure 7b shows a schematic of how the densities of charged states could look when the spiro-OMeTAD DOS using three Gaussian functions each with a width of 0.1 eV and with centers spaced 0.2 eV apart. The total DOS used was $1 \times 10^{21} \text{ cm}^{-3}$, with a carrier concentration of $3 \times 10^{20} \text{ cm}^{-3}$ and 25% of charged molecules in the bipolaron state. These parameters were an approximation based on typical parameters for amorphous small molecules and the properties of F₄TCNQ:spiro-OMeTAD. Differential pulse voltammetry measurements in solvent have shown that the first three oxidation potentials of spiro-OMeTAD are -5.15 , -5.27 , and -5.48 eV .¹² However, the exact spacing of neutral, 1+, and 2+ states may change in the solid state, especially if the dielectric constant of the film changes with doping. In addition to changing the shape of the DOS, bipolarons have also been associated with an increase in Seebeck coefficient in hopping semiconductors through an increase in vibrational entropy.³⁴

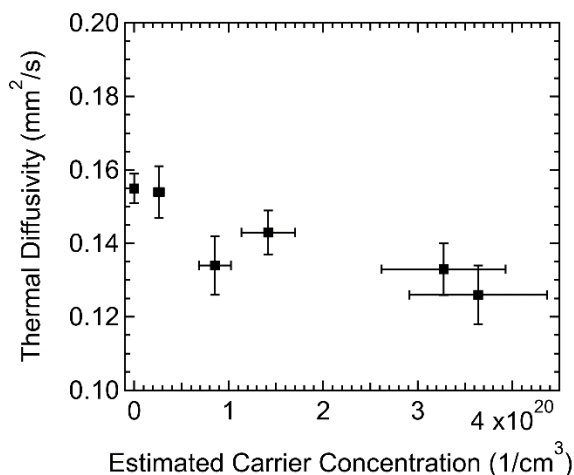


Figure 8. In-plane thermal diffusivity of the samples is roughly constant over this carrier concentration range. The y error is the thermal diffusivity fitting error, and the estimated carrier concentration error is 20%.

To complement these studies, we studied the changes in thermal transport in spiro-OMeTAD films with doping. We measured the in-plane thermal diffusivity of films with varying carrier concentrations with a transient grating technique.³⁵ The transient gradient apparatus uses a visible wavelength laser (515 nm or 2.4 eV) to induce a periodic thermal expansion in the sample. Spiro-OMeTAD⁺ and spiro-OMeTAD²⁺ absorb at this wavelength, so the doped films could be measured without any change in sample preparation. Because undoped spiro-OMeTAD films are transparent to the pump laser, we prepared a dyed spiro-OMeTAD film by blade-coating a film from a solution of Oil Red O and spiro-OMeTAD in chlorobenzene. The thermal diffusivity of this pink-dyed film of insulating spiro-OMeTAD was 0.155 ± 0.004 mm²/s. The thermal diffusivity of F₄TCNQ vapor-doped films, shown in Figure 8, was roughly constant, or slightly decreasing, with carrier concentration in the range measured.

The thermal conductivity can be calculated from the product of the thermal diffusivity, density, and specific heat capacity. We measured the specific heat capacity of spiro-OMeTAD powder with modulated differential scanning calorimetry (0.9 J/kg·°C) and took the density of spiro-OMeTAD films as 1.02 g/cm³.²⁷ With these values, we find an in-plane thermal conductivity for spiro-OMeTAD of 0.14 W/m·K. This thermal conductivity is similar to measurements on other amorphous molecules of 0.14-0.15 W/m·K.³⁶ For comparison, undoped C₆₀ and its derivatives have even lower thermal conductivity of ≈ 0.06 W/m·K,³⁷ while undoped films of semiconducting

polymers can have thermal conductivities of ≈ 0.2 to $2 \text{ W/m}\cdot\text{K}$ depending on the crystallinity and molecular orientation.^{38–41} The thermal conductivity of doped samples will depend on changes in density or heat capacity, but we can expect that the former is unlikely to increase more than $\approx 20\%$ (based on the density of crystalline spiro-OMeTAD) and the latter by a comparable factor. Any expected change due to these factors would be relatively small and would tend to bring the thermal conductivity of the doped samples to the same level as the undoped samples. This result is not surprising given the low electrical conductivity of the doped samples but does show that the intermolecular interactions of the charged molecules likely do not substantially modify the thermal conductivity of the amorphous films. The resulting thermoelectric figure of merit is quite low, $ZT=0.8 \times 10^{-4}$ at room temperature.

Conclusions

We used vapor infiltration to dope spiro-OMeTAD films with F₄TCNQ up to $39 \pm 2 \text{ mol}\%$ F₄TCNQ. Although spiro-OMeTAD can crystallize in neutral form, the doped films remained amorphous. Our results show that this doping method can be used to reach a high conductivity relative to other doped *p*-type small molecules and provides a convenient route to study doping in small molecules. An increase in the Seebeck coefficient at $\approx 2.8 \times 10^{20} \text{ 1/cm}^3$ suggested that the presence of spiro-OMeTAD²⁺ in the films changes the shape of the DOS. Finally, we found that the thermal diffusivity of spiro-OMeTAD is similar to other amorphous organic materials and does not change significantly with doping. Our findings also suggest that degenerate frontier orbitals may change the thermoelectric properties of organic semiconductors and could be a useful molecular design parameter.

Conflicts of Interest

There are no conflicts of interest to declare.

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