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Prediction of MXene based 2D tunable band gap semiconductors: GW quasiparticle calculations

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Abstract

MXenes are a large family of layered transition metal carbide/nitride materials that possess a number of desired properties such as flexible chemical composition, high mechanical strength, and excellent structural stability. Although MXene based semiconductors have attracted considerable recent research attention in the search of novel 2D electronic materials, accurate understanding of their electronic properties has not been established. In this work, we carry out fully converged GW quasiparticle calculations for M₂CO₂ (M = Hf, Zr, and Ti) MXene based 2D semiconductors and alloys using newly developed accelerated GW methods. The quasiparticle band gaps of single-layer Hf₂CO₂, Zr₂CO₂, and Ti₂CO₂ are predicted to be 2.45, 2.13, and 1.15 eV, respectively. The narrow band gap of Ti₂CO₂ is attributed to the low energy of Ti 3*d* as compared with the Hf and Zr *d* states. Considering their chemical similarity, it is expected that Hf_{2-2x}Ti_{2x}CO₂ ($0 \le x \le 1$) semiconductor alloy can be continuously tuned from 2.45 to 1.15 eV, offering a unique 2D semiconductor with moderate and tunable gap for future electronics applications.

Keywords: 2D semiconductors, MXenes, quasiparticle band gap, GW method

1. Introduction

MXenes are a new family of layered transition metal carbide/nitride materials which have attracted considerable research interests since nanosheets of Ti₃C₂ MXene were first synthesized.¹⁻⁴ MXenes can be synthesized by exfoliating the "A" laminar component from the MAX matrix phase in HF solution at low temperature. The MAX phase has a general chemical formula $M_{n+1}AX_n$ (n = 1, 2, or 3), where M is an early transition metal (M = Sc, Ti, V, Cr, Zr, Nb, Mo, Hf, Ta or W), A is a Group IIIA or IVA element (A = Al, Ga, In, Si, Ge, Sn, Pb, P, As, S or Cd) and X is C and/or N⁵. The strong M-X bonds have a mixed ionic, metallic and covalent character, while the M-A bond shows typical metallic nature. The disparate bonding strength among the M-A and M-X bonds in the MAX phase makes it possible to selectively etch out the A layers without disrupting the M-X framework. Since the as-synthesized MXenes are typically terminated by O, OH and/or F groups, they are usually denoted as $M_{n+1}X_nT_x$ (n=1, 2, and 3), with T standing for the terminating groups. MXenes combine the advantages of metals and ceramics, exhibiting high chemical stability, excellent electrical conductivity and exceptional mechanical properties. A number of potential applications have been proposed for these emerging layered materials, including Li-batteries⁶⁻⁸, highly conductive optical coating⁹, high volumetric capacitors¹⁰, and hydrogen storage¹¹.

Most MXenes are metallic, but there are also a number of MXene semiconductors reported. For example, Hf₂CO₂, Zr₂CO₂, V₂CF₂, V₂C(OH)₂, Ti₂CO₂, Sc₂C(OH)₂, Sc₂CF₂ and Sc₂CO₂ are found to possess nonzero band gaps^{7,8,12,13}. These layered semiconductors may have some of the most sought-after properties such as having a moderate band gap and being highly stable, among others, opening a new avenue for fabricating next generation 2D electronic devices. For example, Hf₂CO₂ has been shown¹⁴ to have a higher thermal conductivity than that of the much studied phosphorene or MoS₂, and a high carrier mobility comparable to that of phosphorene. The reported band gap of single-layer (which consists of five atomic layers as shown in Fig. 1) Hf₂CO₂ MXene is 1.65 eV¹⁴ using the Heyd-Scuseria-Ernzerhof (HSE) hybrid functional^{15,16} within density functional theory (DFT).

Although the HSE functional has been shown to be able to predict the band gap of a wide range of bulk semiconductors with reasonable accuracy, there are notable exceptions. For example, the measured band gap of FeS₂ is about 0.95 eV¹⁷, but the HSE functional predicts a band gap of FeS₂ of about 2.7 eV. In addition, straightforward applications of the HSE functional

to predict the quasiparticle band gap of 2D materials deserve scrutiny since the dielectric screening properties in 2D materials are fundamentally different from those in bulk materials. A more rigorous approach for calculating the band gap, and more generally, the quasiparticle band structure, of semiconductors is the GW method¹⁸. Therefore, it is imperative to carry out GW calculations for these emerging 2D semiconductors.

In this work, we have carried out first-principles quasiparticle calculations for Hf_2CO_2 within the GW approximation¹⁸ using a local version of the BERKELEYGW package¹⁹. We find that the GW band gap of monolayer Hf_2CO_2 is about 2.45 eV. This value is significantly larger than the value (1.65 eV) predicted by the HSE functional. Since the optimal band gap of semiconductors varies significantly from applications to applications, for example, narrow gap semiconductors may be suitable for infrared or thermoelectric applications whereas power electronics requires wide-gap semiconductors, it is often desirable that the band gap of a class of materials (e.g., 2D MXenes in this work) be tunable to suit different applications. Considering the chemical flexibility of MXenes, it is likely that their band gap can be conveniently tuned through chemical substitutions of the metallic sites with isoelectronic elements. To this end, we have also carried out DFT and GW calculations for single-layer Zr₂CO₂, Ti₂CO₂ and Hf_{2-x}Ti_xCO₂. We find that the quasiparticle band gap of Hf_{2-x}Ti_xCO₂ can be continuously tuned from 2.45 to 1.15 eV with increasing Ti concentration, offering a promising material system for a wide range of applications.

2. Computational details

Structural optimizations and DFT band structure calculations are carried out using the Vienna *ab initio* simulation package (VASP)²⁰. The interaction between valence electrons and the ionic cores is described using the projector augmented wave (PAW) approach²¹. The generalized gradient approximation (GGA)²² functional of Perdew-Burke-Ernzerhof (PBE)²³ is used for DFT calculations. A relatively high cutoff energy of 500 eV for the plane-wave expansion of the wave functions is used to ensure the convergence of the calculations. The structures are optimized until the Hellmann-Feynman forces are smaller than 1.0×10^{-3} eV/Å. The Brillouin zone (BZ) integration is carried out with a uniform $6 \times 6 \times 1$ *k*-grid. The MXene layer is separated from its periodic images by a 20 Å thick vacuum layer to minimize spurious interlayer interactions.

For the quasiparticle band structure calculations within the GW approximation¹⁸, we use a local version of the BERKELEYGW package¹⁹ in which recently developed acceleration

methods^{24,25} are implemented. Normal conserving pseudopotentials²⁶ are used for all GW calculations, and all semicore subshells (i.e., *ns*, *np*, and *nd* semicore electrons, n = 3 for Ti, 4 for Zr, and 5 for Hf) are included as valence electrons. The Hybertsen-Louie generalized plasmonpole model (HL-GPP)¹⁸ is used to extend the static dielectric function to finite frequencies. Including semicore electrons and the use of norm-conserving potential require a rather high plane wave cutoff energy (175 Ry) and a large number of conduction bands have to be included in the GW calculations to ensure that the calculated results are fully converged²⁴. In addition, GW calculations for 2D materials are known to converge extremely slowly²⁷⁻²⁹ with respect to the density of 2D *k*-grid used in the BZ integration due to the analytical behavior of the 2D dielectric function and other relevant quantities as the wave vector *q* approaches 0. Recently, we have developed two methods^{24,25} that aim directly at resolving these difficulties.

The first method²⁴ we developed replaces the cumbersome band-by-band summation in conventional GW calculations with a numerical integration in the energy space. Our method results in a speed-up factor of up to two orders of magnitude for large-scale GW calculations without sacrificing the accuracy. This method has been successfully applied to GW calculations for a wide range of materials^{24,30,31}. The second method we developed²⁵ is specific for 2D GW calculations as we will discuss in more details later.



Fig. 1. Side view (upper) and top view (lower) of the crystal structure of single layer Hf₂CO₂.

3. Results and discussion

3.1 Quasiparticle band structure of single layer Hf₂CO₂



Fig. 2. Atomic-orbital projected band structure of single-layer Hf₂CO₂ calculated using the PBE functional.

Terminal oxygen can adsorb on either fcc or hcp sites. Previous studies have established that the fcc oxygen adsorption has the lowest energy^{13,14,32}. In this work, we adopted the fcc geometry for O atoms. Single-layer Hf₂CO₂ MXene has a hexagonal structure consisting of five atomic layers as shown in Fig. 1. The optimized 2D hexagonal lattice constant is 3.265 Å, and the Hf-C and Hf-O bond lengths are 2.33 Å and 2.10 Å, respectively. Figure 2 shows the PBE band structure superimposed with projections onto atomic orbitals. The valence band maximum (VBM) is placed at zero in Fig. 2. The lower group of valence states is mainly derived from O *p* states with some admixture of Hf *d* states, whereas the upper one has contributions from C *p*, Hf *d*, and O *p* states. The low energy conduction bands, on the other hand, consist mostly of Hf *d* states. The VBM states are located at the Γ point and are primarily composed of C p_x+p_y orbitals hybridized with O p_x+p_y and Hf $d_{xz}+d_{yz}$. The conduction band minimum (CBM) is located at the M point of the hexagonal BZ, consisting mostly of Hf d states. Since the CBM of Hf₂CO₂ MXene is derived from the Hf d states, it is expected that substituting Hf with isoelectronic transition metals with lower d-state energy will result in a reduced band gap as we will discuss later.

Figure 3 compares the PBE and GW band structures of single-layer Hf_2CO_2 . The PBE functional predicts an indirect band gap of 0.99 eV, to be compared with the GW band gap of 2.45 eV. The minimum direct band gap at the M point is about 3.60 eV within the GW approximation. The valence bandwidth is also significantly enhanced with the GW correction. With a moderate band gap of 2.45 eV and other promising properties¹⁴ such as high thermal conductivity and carrier mobility, this MXene based 2D semiconductor holds great potential for future 2D electronics applications.



Fig. 3. DFT-PBE (red) and GW (dotted blue) band structures of single-layer Hf₂CO₂.

increasing *k*-grid density. For example, for silicon (diamond structure with a 2-atom unit cell), a $6 \times 6 \times 6 k$ -grid is sufficient to converge the calculated GW band gap to within 0.01 eV. For 2D materials, however, the convergence is extremely $slow^{27-29}$ due to the analytical behaviors of the 2D dielectric function and other relevant quantities as wave vector \ddot{q} approaches 0. Recently, we have successfully developed a combined subspace sampling and analytical integration technique²⁵ that can drastically reduce the required 2D *k*-grid sampling density for achieving fully converged GW results. Our method is inspired by a recent work²⁹ in which the authors proposed a nonuniform sampling scheme for GW calculations.



Fig. 4. Fitting the self-energy contribution from the mBZ enclosing the Γ point with an analytical function for subsequent integration. Blue dots are the data points used to fit the function (shown with the blue curve). Red squares are additional data points to show the quality of the fitting function.

In our new method, the BZ integration of the self-energy of state $|n\vec{k}\rangle$ near the Γ ($\ddot{q} = 0$) point is replaced with an analytical integration over the mini-BZ (mBZ) enclosing the Γ point,

i.e.,
$$\Sigma_{nk}^{\mathbb{I}} = \sum_{\substack{q \neq 0 \\ q \neq 0}} f_q^{\mathbb{I}} \Sigma(q) + \frac{f_0}{V_{mBZ}} \int_{mBZ} \Sigma(q) dq$$
. We first calculate $\Sigma(q)$ for a few small q-points within the

mini-BZ as shown with blue dots in Fig. 4. These results are then used to fit $\Sigma(q)$ with some analytical form as shown in Fig. 4 with the blue curve. Analytical integration within the mBZ is then carried out using this function. To illustrate the accuracy of the fitting function, we have

also plotted in Fig. 4 $\Sigma(q^{u})$ calculated at other q-points (red squares), showing excellent agreement between the calculated results and the fitting function.



Fig. 5. Calculated GW band gap of single-layer Hf_2CO_2 as a function of k-point sampling density: Comparison between the conventional uniform sampling scheme and the newly developed method.

Using this method, we are able to carry out converged GW calculations for 2D materials using a $6 \times 6 \times 1$ coarse *k*-grid as shown in Fig. 5 in which we compare the performance of our new method with the conventional uniform *k*-point sampling approach. Using our new method, the calculated band gap converges to within 0.02 eV with a very coarse *k*-grid of $6 \times 6 \times 1$. In comparison, the band gap does not seem to converge even with a very fine $24 \times 24 \times 1$ grid using the conventional method. Note that the convergence behavior of the GW results calculated using the uniform sampling approach shown in Fig. 5 is consistent with previous work. For example, it has been shown that a *k*-grid of at least $24 \times 24 \times 1$ is need to properly converged the GW band gap of MoS_2^{33} . Considering the fact that the computational cost of GW calculations (the calculation of the electron polarizability) scales as $O(N_k^2)$ where N_k is the number of *k*-points used to sample the BZ, our method represents a speed-up factor of over two orders of magnitude compared with the conventional method. Combining this method with another method²⁴ we developed in which the conventional band-by-band summation in the GW calculations is

replaced with an energy integration, we can now carry out fully converged GW calculations for 2D materials with a speed-up factor of about three orders of magnitude.

3.2 M₂CO₂ (M = Hf, Zr, and Ti): Structural and electronic properties

We now investigate the possibility of tuning the band gap of M_2CO_2 MXene based 2D semiconductors. As we have shown in Fig. 2, the CBM state of Hf_2CO_2 is mainly derived from Hf *d* states. Therefore, it is plausible the band gap of Hf_2CO_2 can be tuned by replacing Hf with isoelectronic transition metals such as Ti or Zr. In particular, since the 3*d* level of Ti atom is significantly lower than the Hf 5*d* level, replacing Hf with Ti may result in significantly lowered CBM thus reducing the band gap. In addition, due to their chemical similarity, it is expected that semiconductor alloys of M_2CO_2 (M = Hf, Zr, and Ti) can be synthesized, providing a MXene based 2D material system with tunable band gap.

Table I. Formation energies (eV/f.u.), lattice constants (Å) and band gaps (eV) of single-layer Hf_2CO_2 , Zr_2CO_2 , Ti_2CO_2 and (HfTi)CO₂ MXene semiconductors.

	$\mathrm{Hf_2CO_2}$	Zr ₂ CO ₂	Ti ₂ CO ₂	(HfTi)CO ₂
Formation Energy	-8.17	-7.28	-6.67	-7.20
Lattice constant	3.265	3.322	3.034	3.161
DFT-PBE gap	0.99	0.88	0.26	0.55
HSE gap	1.59	1.45	0.90	1.28
GW gap	2.45	2.13	1.15	1.80

Before investigating their electronic structures, we first evaluate the stability of M₂CO₂ (M = Hf, Zr, and Ti) by calculating their formation energies. We define the formation energy as the difference in total energy between single-layer M₂CO₂ and stable competing subsystems, i.e., solid metal and CO₂ gas molecules, i.e., $\Delta E = E(M_2CO_2) - [E(metal) + E(CO_2)]$. Table I shows the calculated formation energies and lattice constants for the four single layer MXenes. Although the formation energy (absolute value) of Ti₂CO₂ is smaller than that of Hf₂CO₂, a formation energy of -6.67 eV per formula unit (f.u.) still suggests that Ti₂CO₂ is thermodynamically stable and synthetically viable.

Figure 6 shows the DFT-PBE band structures of Zr_2CO_2 and Ti_2CO_2 . The DFT-PBE band gap of Ti_2CO_2 (0.26 eV) is indeed much smaller than that of Hf₂CO₂ as expected. Replacing Hf

with Zr, however, does not lead to significant changes to the band structure. The PBE band gap of Zr_2CO_2 is about 0.88 eV, which is only 0.1 eV smaller than that of Hf_2CO_2 as shown in Table I. This is because 4*d* and 5*d* transition metals share similar physical properties such as *d* level positions and atomic radii. Zr_2CO_2 has the largest lattice constant among the three systems. This is expected since Zr has the greatest ionic radius. We have also carried out GW calculations for the band gap of Zr_2CO_2 and Ti_2CO_2 . Including GW self-energy corrections increases the band gap of Ti_2CO_2 to 1.15 eV and that of Zr_2CO_2 to 2.13 eV. These results suggest an interesting possibility of continuously tuning the band gap of these 2D semiconductors from 2.45 to 1.15 eV through alloying.



Fig. 6. Atomic-orbital projected band structure of single-layer Ti_2CO_2 (upper) and Zr_2CO_2 (lower) calculated using the PBE functional.

3.3 Continuous band gap tuning of Hf_{2-x}Ti_xCO₂ 2D semiconductor alloys

Isoelectronic substitution minimally disturbs the chemical bonding of a material while offering the possibility of significantly modifying the low-energy (near-edge) electronic states. The quasiparticle band gap of Hf₂CO₂ is predicted to be 2.45 eV whereas that of Ti₂CO₂ is about 1.15 eV. Therefore, alloying Hf₂CO₂ with Ti₂CO₂ may provide a material system with a widely tunable band gap. To this end, we have calculated the electronic properties of Hf_{2-2x}Ti_{2x}CO₂ ($0 \le x \le 1$) alloys. We create a 4×4×1 supercell containing 80 atoms for studying the alloy systems.

Except for the end systems, for each Ti concentration x, several inequivalent structures are randomly created and their structures are fully optimized.



Fig. 7. A few structural models for the HfTiCO₂ 2D random alloy used in this work.

Figure 7 shows a few structural models that we created for the HfTiCO₂ random alloy. We mention the special quasirandom structure (SQS) approach³⁴ is a well-established approach for generating finite structural models that best mimic the radial correlation functions of perfectly random structures, and ideally one should adopt this approach for creating structural models for random alloys. Since we find that the calculated band gap does not depend sensitively on the specific structures (with a variation of about ± 0.1 eV), we did not attempt to create SQS for these random alloys.

On the other hand, we have noticed that Wong and collaborators³² recently carried out a detailed study on the structural and thermodynamic properties of these random alloys. They concluded that $Ti_{2(1-x)}Zr_{2x}CO_2$, $Ti_{2(1-x)}Hf_{2x}CO_2$, and $Zr_{2(1-x)}Hf_{2x}CO_2$ exist as disordered solid

solutions. These results are expected since isoelectronic ions usually mix rather well as they have the same valency, especially when their ionic radii do not differ significantly as suggested by Hume–Rothery rules. Other MXene alloys have also been investigated theoretically³⁵. Admittedly, the stability of a material shall be ultimately determined by experiments. In recent years, experimentalists have been very successfully (and sometimes unexpectedly) in synthesizing metastable compounds. In particular, chemical vapor deposition (CVD) has been shown to be a powerful method for synthesizing 2D materials and heterostructures³⁶.

Figure 8 (a) shows that the lattice constant of $Hf_{2-2x}Ti_{2x}CO_2$ semiconductor alloy varies nearly linear with the Ti concentration *x*, obeying Vegard's law. Interestingly, the calculated DFT-PBE band gap of $Hf_{2-2x}Ti_{2x}CO_2$ as a function of Ti concentration also shows a nearly linear dependence [Fig. 8 (b)], suggesting a very small band gap bowing parameter for this system. For a given Ti concentration, different structures show slightly different band gap with a deviation of ±0.1 eV from their averaged values. The HSE and GW results for three Ti concentrations are also shown in Fig. 8 (b) for comparison. Our results suggest that the quasiparticle band gap of $Hf_{2-2x}Ti_{2x}CO_2$ alloy can be continuously tuned from 2.45 eV to 1.15 eV.



Fig. 8. Lattice constant (left) and band gap (right) of single-layer $Hf_{2-2x}Ti_{2x}CO_2$ alloy as functions of Ti concentration *x*. The band gaps are calculated within the DFT-PBE approach for all Ti concentrations and using the HSE and GW approaches for selected concentrations.

Isoelectronic substitutions may be achieved beyond single pair of elements. Since the band gap of Zr_2CO_2 is just slightly smaller than that of Hf_2CO_2 , alloys of these two semiconductors may not offer significant band gap tunability. However, considering the fact that Zr_2CO_2 has a

larger lattice constant than Hf_2CO_2 , whereas Ti_2CO_2 has the smallest lattice constant, it may be beneficial to synthesize ternary alloys $Hf_{2-2x-2y}Zr_{2x}Ti_{2y}CO_2$ to minimize the lattice strain. In addition, co-doping with d^1 and d^3 pairs (e.g., Y and Nb, or Sc and Ta) may also be considered. We would like to mention mechanical strain can also be used to tune the band gap of semiconductors. For example, we find that the band gap of Hf_2CO_2 reduces at a rate of about 0.1 eV per 1% compressive 2D lattice strain. It may be desirable to combine mechanical strain with chemical substitution to realize wide band gap tunability in certain applications.

4. Summary

 M_2CO_2 (M = Hf, Zr, and Ti) MXene based tunable band gap 2D semiconductors are investigated using DFT and GW approaches. Recently developed accelerated GW methods are employed in this study. The quasiparticle band gap of single-layer Hf₂CO₂ is predicted to be 2.45 eV, which is significantly different from about 1.59 eV predicted using the HSE functional, suggesting the importance of accurate quasiparticle calculations for predicting the electronic properties of 2D semiconductors. The conduction bands of Hf₂CO₂ are mainly derived from Hf *d* states. This points to the possibility of tuning the band gap through Ti substitutions of Hf since Ti has lower *d* levels. Indeed, the calculated quasiparticle band gap of Ti₂CO₂ is about 1.15 eV, and we show that the quasiparticle band gap of Hf_{2-2x}Ti_{2x}CO₂ ($0 \le x \le 1$) semiconductor alloy can be continuously tuned from 2.45 to 1.15 eV. Introducing Zr into the alloy system is expected to reduce the lattice strain while leaving the fundamental gap nearly unchanged. We believe that M₂CO₂ MXene based semiconductors and their alloys, which combine advantages such as structural stability, chemical flexibility, and electronic tunability, offer a unique material for future 2D electronics applications.

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