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A New Synthetic Route to 5,6,11,12-tetraarylethynyltetracenes

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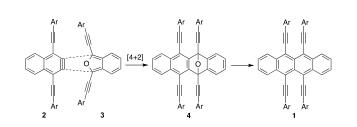
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A new synthetic route to 5,6,11,12-tetrakis(arylethynyl)tetracenes, π -extended rubrenes, was developed via [4+2] cycloadditions of dialkynylisobenzofuran and 1,4-naphthoquinone. Introduction of arylethynyl groups by double nucleophilic additions to tetracenequinone gave sterically congested (arylethynyl)tetracenes after reductive aromatization. The photophysical properties of the newly prepared π -conjugated molecules are also evaluated.

We previously reported a preparation of 5,6,11,12tetraarylethynyltetracene **1**, a new class of π -extended rubrenes, via [4+2] cycloaddition of dialkynylnaphthalyne **2** and dialkynylisobenzofuran **3** (Scheme 1).^{1,2} In this reaction, two alkynyl groups onto the naphthalyne **2** can lower the LUMO energy, allowing the practical construction of the sterically overcrowded structure through their efficient HOMO–LUMO interaction.



Scheme 1 The first syntheses of $\pi\text{-extended}$ rubrenes 1 via [4+2] cycloaddition of naphthalyne and isobenzofuran

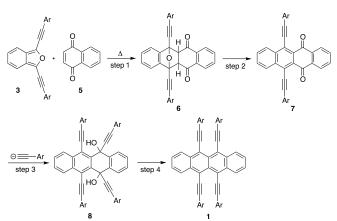
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This approach, however, has a problem in that the yield of the aromatization $(4\rightarrow 1)$ is low or moderate owing to the unexpected reactivities derived from the closely located *peri*ethynyl groups in epoxytetracene **4** under the acidic conditions.³

To solve this problem, we focused on developing a new synthetic route to π -extended rubrene 1 using dialkynylisobenzofuran **3** as a reactive platform.^{4,5} Our second approach is consisting of four-step syntheses, which is depicted in Scheme 2.6 In the first step, the [4+2] cycloaddition of dialkynylisobenzofuran 3 and 1,4-naphthoquinone (5) gives the cycloadduct 6, which is converted to the tetracenequinone 7 by aromatization (step 2). Subsequent introduction of two alkynyl groups by double nucleophilic additions of alkynyl anions (step 3), and reductive aromatization of the resulting diol 8 would produce the target compound 1 (step 4). Along these lines, we now report an efficient synthetic access to π -extended rubrenes possessing various arylethynyl groups at the peri-positions. Moreover, photophysical properties of the newly prepared poly-ethynylated tetracenes are evaluated. Also described is the application of the parent compound **1a** to a cellular imaging agent.



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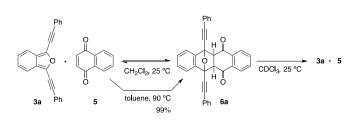
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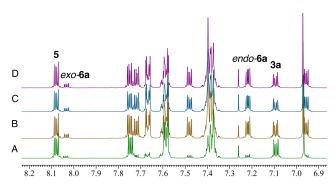
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Scheme 2 New synthetic route to $\pi\text{-extended rubrenes }\mathbf{1}$

Scheme shows the [4+2] cycloaddition of 3 dialkynylisobenzofuran. Upon mixing of isobenzofuran 3a and naphthoquinone 5 (CH_2Cl_2 , r.t.), the new spot corresponded to the cycloadduct 6a was observed by TLC. Further reaction at same temperature, however, did not completely consume the starting materials 3a and 5, indicating their equilibrium with the cycloadduct 6a. Indeed, ¹H NMR analysis revealed that the cycloadduct 6a including endo- and exo-isomer was readily after dissolving the isobenzofuran formed 3a and naphthoquinone $\mathbf{5}$ in $CDCl_3$ at room temperature (see A in Figure 1). After 7 h, the ratio of 3a, 5, exo-6a, and endo-6a almost became constant (see D in Figure 1). The stereochemistry of the exo-6a and endo-6a was tentatively assigned by consideration of the chemical shift of each methine proton.7



After further study on this [4+2] cycloaddition, we were pleased to find that solvent choice is crucial to produce the high yield of the cycloadduct **6a**: when the above mentioned reaction was performed in toluene at 90 °C, the [4+2] cycloadduct **6a** gradually precipitated from the solution due to its low solubility in toluene, affording the essentially pure product **6a** almost in quantitative yield (Scheme 3). Interestingly, the endo isomer **6a** was solely produced under this conditions. By dissolving in CDCl₃ (25 °C, 26 h), the cycloadduct **6a** again underwent the cycloreversion to give the dialkynylisobenzofuran **3a** and 1,4naphthoquinone **(5)**.⁸

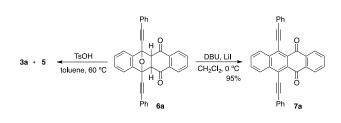


a) A: 5 min, B: 2 h, C: 7 h, D: 15 h

Figure 1 [4+2] Cycloaddition between isobenzofuran 3a and 1,4-naphthoquinone (5) monitored by NMR.

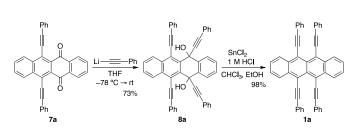
Scheme 4 shows the conversion of the [4+2] cycloadduct **6a** to tetracenequinone **7a**. Upon heating of cycloadduct **6a** in the

presence of TsOH at 60 °C, the cycloreversion occurred quickly, and the aromatized product **7a** was not obtained at all.^{9,10} On the other hand, treatment of the cycloadduct **6a** with LiI and DBU at low temperature (CH₂Cl₂, 0 °C)¹¹ underwent the clean aromatization without invoking the cycloreversion to give the tetracenequinone **7a** in 95% yield.

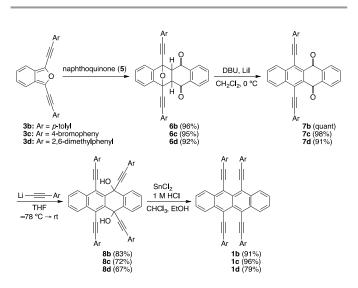


Scheme 4 Aromatization of cycloadduct 6a to tetracenequinone 7a

Further transformation of the tetracenequinone **7a** to π extended rubrene **1a** was achieved through double nucleophilic additions of phenylethynyllithium, followed by Sn^{II}-mediated reductive aromatization (Scheme 5). Importantly, the nucleophilic addition of alkynyllithium to **7a** occurred cleanly by warming the reaction to room temperature, in spite of the high steric hindrance between incoming nucleophile and proximal alkynyl groups.



Scheme 5 Transformation of tetracenequinone 7a to π -extended rubrene 1a



Scheme 6 Preparation of π -extended rubrenes 1b–1d

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In a similar manner, the substituted derivatives **1b** and **1c**, having four *p*-tolylethynyl or (4-bromophenyl)ethynyl groups at both *peri*-positions, were efficiently synthesized by this four-step sequence including the tetracenequinones **7b** and **7c** as key intermediates (Scheme 6).

It should be noted that the developed method has high synthetic potential in that the sterically congested derivative **1d** possessing four 2,6-xylylethynyl groups on the tetracene core was easily accessible in good yield. This is a sharp contrast from our previous method by acid-promoted aromatization of the epoxy tetracene **4d** (Ar: 2,6-xylyl), where the product **1d** was obtained in poor yield, and a sizable amount of furan (structure not shown) was produced.¹

To evaluate the photophysical properties, UV–Vis spectra of π -extended rubrenes **1a–1d** were measured in chloroform (Figure 2). The π -extended rubrene **1a** has its absorption maximum at 640 nm, which was greatly red-shifted over 100 nm from that of the parent rubrene, indicating effective π -extension by the existence of four phenylethynyl groups on the tetracene core. The π -extended rubrenes **1b** and **1c** with para-substitution denoted the similar tendency of **1a**, whereas the absorption maximum of the sterically congested derivative **1d** was slightly blue-shifted (623 nm).

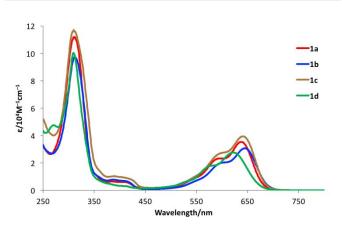
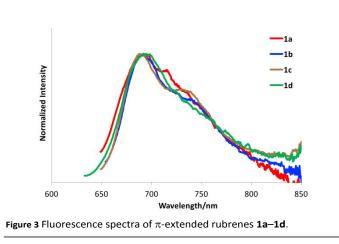


Figure 2 UV–Vis absorption spectra of π -extended rubrenes **1a–1d**.



Fluorescent spectra were also measured in chloroform (Figure 3). The π -extended rubrenes **1a–1d** showed the fluorescent maximum peaking at around 690 nm, which were excited at their absorption maximum. Larger Stokes shift was observed in **1d** (1620 cm⁻¹) compared to that of **1a** (1200 cm⁻¹). The absolute fluorescent quantum yields of these π -extended derivatives were nearly 10%, which were lower than that of the parent rubrene.

Finally, preliminary investigation of cellular imaging using π extended rubrene was performed by treating the HeLa cells with **1a** for 30 min at 37 °C. Fluorescence signals from cells upon excitation with 620 nm indicate a future applicability of π extended rubrene as bioimaging probe.

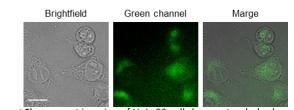


Figure 4 Fluorescent imaging of HeLaS3 cells by π -extended rubrene. The cells were treated with 100 μ M of **1a** for 30 min at 37°C and analyzed by fluorescent microscopy. Green channel: $\lambda_{em} = 620$ nm, $\lambda_{ex} = 700$ nm. Scale bar: 20 μ m.

Conclusions

In conclusion, [4+2] cycloaddition of dialkynylisobenzofuran and 1,4-naphthoquinone allowed rapid construction of alkynylated tetracenequinones, which were amenable to transformation en route to tetrakis(arylethynyl)tetracenes. Further study on application of these attractive π -conjugated molecules to organic electronics materials and fluorescent probes are under active investigation in our laboratories.

Acknowledgement

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Conflicts of interest

There are no conflicts to declare.

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