

Storage conditions modulate the metabolomic profile of a black raspberry nectar with minimal impact on bioactivity

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2	minimal impact on bioactivity
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12 Abstract

13 Pre-clinical and clinical studies suggest black raspberries (BRBs) may inhibit the 14 development of oral cancer. Lyophilized BRB powder is commonly used in these studies, but 15 processed BRB products are more often consumed. The objective of this work was to understand 16 how storage conditions influence the phytochemical profile and anti-proliferative activity of a 17 BRB nectar beverage. Untargeted UHPLC-Q-TOF-MS based metabolomics analyses demonstrated that large chemical variation was introduced by storage above -20 °C over 60 days. 18 19 However, minimal change in anti-proliferative activity was observed when stored nectar extracts 20 were applied to SCC-83-01-82 premalignant oral epithelial cells. As proof of concept, cyanidin-21 3-O-rutinoside and its degradation product, protocatechuic acid, were administered in different 22 ratios maintaining an equimolar dose, and anti-proliferative activity was maintained. This study 23 shows the utility of metabolomics to profile global chemical changes in foods, while 24 demonstrating that isolated phytochemicals do not explain the complete bioactivity of a complex 25 food product. 26

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29 1. Introduction

Black raspberries (BRBs) are extensively studied for their cancer preventative properties 1,². Their bioactivity has been attributed to their rich phytochemical profile inclusive of anthocyanins, ellagitannins, organic acids, and quercetin among other phenolic compounds ³. It has been hypothesized that these components elicit a complex series of biological responses that result in a net inhibition of cancer growth ⁴. Because of their high concentration of phenolic compounds, BRBs have been the subject of many studies on food-based chemoprevention strategies.

37 Much of the interest in the chemopreventative properties of BRBs has focused on oral cancer $^{5-9}$. The oral cavity presents unique opportunities for chemoprevention through dietary 38 39 means due to direct exposure of tissues to food phytochemicals. Oral cancer is prevalent 40 throughout the world, with higher incidence observed in men and people in less developed regions ^{10,11}. The majority of oral cancers are squamous cell carcinomas (SCCs), and risk factors 41 42 include tobacco use, alcohol consumption, human papillomavirus infection, and chronic periodontal disease ¹². Pre-clinical models have consistently shown reduction of oral SCC 43 incidence and multiplicity using whole freeze-dried BRBs, likely due to engagement of a number 44 of biological mechanisms¹³. A human clinical trial with a BRB-based mucoadhesive gel 45 46 demonstrated the ability of BRBs to reduce the size and severity of precancerous oral lesions⁸, 47 while a relative reduction in the expression of molecular biomarkers indicative of SCC was observed after patients were treated with BRB-based troches for two weeks⁶. These studies 48 49 support a role for BRB-mediated efficacy in oral cancer prevention strategies. 50 Most research using BRBs has been conducted with minimally processed, lyophilized

51 BRB powder. In practical terms, consumers mostly encounter BRBs after they have been

52	incorporated into shelf-stable food products and stored for varying lengths of time, during which
53	the phytochemical profile may be altered ¹⁴ . Research on the stability of the phytochemical
54	profile in BRB-based food products is limited to a short defined list of compounds ^{14–16} , while
55	effects on the global phytochemical profile and bioactivity of these products are unknown.
56	Metabolomics is an emerging approach to chemical analysis in which hundreds to thousands of
57	compounds within a food system are profiled, with the potential to provide new insight into the
58	relationship between food phytochemicals and health outcomes ¹⁷ . The objective of the current
59	study is to use an untargeted metabolomics approach to understand the global differences in the
60	phytochemical profile of a BRB nectar beverage over storage time and temperature variations,
61	and how these changes relate to the growth inhibition activity in an <i>in vitro</i> oral premalignacy
62	model.
63	
64	2. Experimental
65	2.1 Chemicals
66	All solvents were of HPLC-MS grade from Fischer Scientific (Pittsburgh, PA) unless
67	otherwise noted. Cyanidin-3-O-rutinoside (C3R) and protocatechuic acid (PA) standards were
68	from Sigma Aldrich (St. Louis, MO). Cell culture-grade dimethyl sulfoxide and water were also
69	from Sigma Aldrich.
70	
71	2.2 Nectar Processing, Storage, and Sampling
72	Nectar was prepared in the pilot plant facilities located at The Ohio State University

(Columbus, OH) using a formula similar to that described by Gu and colleagues, as shown in
 Table 1 ¹⁶. The BRB powder used was produced from whole BRBs harvested at Stokes Berry

Farm (Wilmington, OH). All components were combined in a high shear mixer for 20 min, and
the nectar was subsequently pasteurized using a MicroThermics UHT/HTSTLab-25HV Hybrid
unit (MicroThermics, Inc., Raleigh, NC, USA). The processing specifications mirrored industry
practices for pasteurization of this product by which the nectar was held at 100 °C (± 1.1 °C) for
23 sec, immediately cooled, and aseptically filled into pre-sterilized 50 mL conical centrifuge
tubes.

81 Nectar was stored at -20 °C, 4 °C, 10 °C, 25 °C, or 35 °C for 60 d with samples (n=4) removed from each condition at 5 d, 10 d, and subsequently in 10 d intervals. Two months has 82 83 been described as an appropriate amount of time to study stability in foods intended for clinical trials, as time is often needed for subject recruitment and intervention ¹⁶. At each time point the 84 85 nectar was centrifuged (1000 x g) for 5 min, partitioned into smaller aliquots, and stored at -8086 °C prior to use. Samples of freshly produced nectar were also stored at -80 °C at the time of 87 production as a t_0 sample. Aerobic plate counts and veast and mold counts were obtained for the 88 35 °C incubated samples at each time point with 3M Petrifilm (3M Company; Maplewood, MN), 89 according to manufacturer instructions.

90

91 2.3 Sample Preparation and Analysis for Untargeted Metabolomics

Aliquots (1 mL) of nectar were thawed in a room temperature water bath for 10 min.
Once thawed, 750 µL were deposited into a glass vial followed by 2.25 mL of 0.1% formic acid
in methanol. The mixture was homogenized with a probe sonicator (Branson Ultrasonics;
Danbury, CT) for 10 sec and centrifuged for 5 min at 1000 x g (4 °C). The supernatant was
decanted into a glass vial, and the pellet was extracted twice more with 3 mL of 75% methanol in

97	water with 0.1% formic acid. Aliquots (200 μ L) were deposited into 4 mL glass vials, dried
98	under a stream of nitrogen, and stored at -20 °C until analysis.
99	Dried aliquots were solubilized in 100 μ L 25% methanol in water with 0.1% formic acid
100	and vortexed for 15 sec. Samples were then centrifuged at 21,130 x g for 4 min (4 °C) and placed
101	in the autosampler of a 1290 Infinity II series UHPLC (Agilent Technologies, Santa Clara, CA)
102	maintained at 4 °C until analysis. Samples were injected (5 μ L) onto a 2.1 x 100 mm, 1.8 μ m
103	Acquity HSS T3 column (Waters, Milford, MA) maintained at 40 °C. The mobile phase
104	consisted of A: 0.1% formic acid in water and B: 0.1% formic acid in acetonitrile with a flow
105	rate of 0.5 mL/min. The linear gradient program was as follows: 0% B held for 1 min, increased
106	to 60% B over 5 min, increased to 100% B over 2 min and hold for 1.5 min, immediately
107	switched to 0% B and held for 2 min for a total run time of 11.5 min.
108	Eluent was directed to an Agilent iFunnel 6550 QTOF-MS interfaced with an
109	electrospray ionization (ESI) source operated in negative ion mode. The first minute of flow
110	from the UHPLC was directed to waste. Relevant MS settings were as follows: gas temp 150 °C,
111	drying gas 18 L/min, nebulizer 30 psig, sheath gas temp 350 °C, sheath gas flow 12 L/min, VCap
112	4000 V, nozzle voltage 2000 V, acquisition mode was 2 GHz extended dynamic range with a
113	mass range of 50–1700 m/z. Reference mass solution (Agilent Technologies) was concurrently
114	infused into the source via a dedicated sprayer for continual mass correction. Sample run order
115	was randomized. Quality control samples, composed of equal portions of each nectar sample,
116	were run every 10 samples to monitor instrument performance over the run time (data not
117	shown).
118	

119 2.4 Data Pre-processing and Analysis for Untargeted Metabolomics

120 Raw spectral data was processed using the batch recursive feature extraction algorithm in 121 Profinder (B.08.00, Agilent Technologies). Mass spectral features were picked and binned 122 according to expected isotope patterns, adducts, and charge states. These molecular features were 123 then aligned across all samples, and those appearing in less than three samples per 124 time/temperature group were removed from further analysis. The raw data was then searched 125 against this assembled list in a targeted manner to improve the quality of the data used for 126 multivariate analysis. Further data pre-processing was performed in Mass Profiler Professional 127 (version 14.5, Agilent Technologies), including removal of features present in sample blanks. To 128 remove low quality peaks from the data, an additional abundance filter was applied which required a minimum peak height of 5.0×10^4 in 75% of the samples in at least one 129 130 time/temperature sample group.

131 Multivariate analyses, including principal component analysis and partial least squares 132 regression (PLS), were executed in R (version 3.2.3) with the ropls package using the autofit option to determine the optimal number of components¹⁸. Data were \log_{10} transformed and 133 134 Pareto scaled prior to analysis. PCA is a dimensional reduction technique that allows for analysis 135 of multidimensional data in an easily-visualized space. PLS is a common multivariate modeling technique that builds off the dimensional reduction properties of PCA but in the framework of a 136 linear regression¹⁹. Separate PLS models were constructed for each storage condition. The X 137 138 matrices were composed of features present in 75% of replicates from at least one time point in 139 each storage condition, and the Y matrix was storage time. Performance of the PLS models was 140 assessed using 8-segment cross validation, and statistical significance of each model was 141 determined using permutation tests (n = 100). Features with a variable importance on projection 142 value (VIP) \geq 1 across all successful models were manually reviewed before further analysis.

Similarly, features with a VIP \geq 1 in only the 35 °C samples were also manually reviewed for further analysis. VIP scores are estimates of the relative importance of a chemical feature to a given PLS model, and features with a score \geq 1 are typically considered to be important in the model. A data pre-treatment and analysis summary is shown in Figure 1.

147

148 2.5 Targeted Compound Analysis

149 Cyanidin-3-O-rutinoside (C3R) and protocatechuic acid (PA) were quantified in the 150 nectar samples from t₀ and 60d at 35 °C. Extracts of BRB nectar were obtained as described for 151 the untargeted metabolomics workflow, reconstituted in 5 mL of 5% aqueous formic acid, and 152 filtered through a 0.22 μ m nylon filter. Samples were then injected (0.5 μ L) into an Agilent 1290 153 Infinity II UHPLC coupled to an Agilent 6495 triple quadrupole MS equipped with an ESI 154 source operated in positive and negative ion modes. The mobile phase consisted of A: 5% 155 aqueous formic acid and B: 5% formic acid in acetonitrile. The column and gradient program 156 were identical to that was described here for untargeted analyses. MS parameters included gas 157 temp: 150 °C, gas flow: 18 L/min, nebulizer: 45 psi, sheath gas heater: 375 °C, sheath gas flow: 158 12 L/min, capillary: 3000 V, fragmentor: 350. Quantitation was performed using standard curves 159 constructed from serial dilutions of authentic standards. The transitions used for each compound 160 were as follows: C3R: 595 [M+] \rightarrow 287 (CE = 17 V), PCA: 153 [M-H]⁻ \rightarrow 109 (CE = 10).

161

162 2.6 Extract Preparation for Cell Study

Extraction of the nectar was scaled up from the procedure used in the untargeted metabolomics workflow to ensure sufficient extract mass. Briefly, nectar replicates were pooled and 1 mL aliquots were deposited into glass vials followed by 3 mL of 0.1% formic acid in

166	methanol. The mixture was homogenized with a probe sonicator and centrifuged at 3220 $\times g$ (4
167	°C) for 7 min. The supernatant was decanted into a glass vial, and the pellet was extracted once
168	with 75% aqueous methanol with 0.1% formic acid. The pooled supernatants were dried using a
169	Genevac EZ2 vacuum evaporation system (SP Scientific; Ipswich, United Kingdom) set at 30
170	°C. Remaining water was removed by lyophilization on a Labconco FreeZone 12 Plus system
171	(Kansas City, MO). Nectar extracts were reconstituted in 1:1 DMSO/water, sonicated for 15 sec,
172	and diluted to a concentration of 2 mg extract/mL in cell culture media.
173	
174	2.7 Cell Culture and Growth Inhibition Assay
175	Premalignant human oral epithelial cells (SCC-83-01-82) were maintained in modified
176	minimal essential medium (MEM) with 10% fetal bovine serum and 5% antibiotic/antimycotic
177	solution including penicillin (10,000 U/mL), streptomycin (10,000 U/mL), and amphotericin B
178	(25 μ g/mL) as previously described ^{20,21} . The characteristics of this cell line have been previously
179	described ²² . Cell cultures were incubated at 37 °C in a 90% humidified environment with 5%
180	CO ₂ atmosphere.
181	Cells were seeded at a density of 1000 per well in 96-well plates. After 24 hr, the media
182	was replaced to deliver 200 μ g extract/well or standards of C3G or PA at concentrations ranging
183	from 3–100 μ g/mL using previous work with crude berry product extracts and their isolated
184	components as a guide ²³ . Control samples were composed of an equivalent amount of 1:1
185	DMSO/water diluted in MEM. All samples were incubated for 72 hr.
186	Growth inhibition was determined using a WST-1 assay (Roche; Pleasanton, CA)
187	according to manufacturer instructions. Growth inhibition was calculated as 1 - ((A_{trt} - A_{trt}
188	$_{blank})/(A_{control}-A_{control blank}))$. Treatment blanks were made by incubating sterile media with the

189 corresponding dose of nectar extracts or phytochemicals in identical conditions as the treated 190 cells. Technical replicates were performed in quadruplicate, while biological replicates were 191 performed in triplicate. Cytotoxic activity was evaluated using the Clontech LDH Cytotoxicity 192 Detection Kit (Mountain View, CA) according to manufacturer instructions. Data were analyzed 193 using the generalized linear model procedure in SAS version 9.4. The data were fitted with an 194 ANOVA model with terms corresponding to nectar incubation time, temperature, and their 195 interaction with significance reported at P < 0.05. Differences between treatments were assessed 196 using Tukey's post hoc test with $\alpha = 0.05$.

197

198 **3 Results and Discussion**

199 We report on the phytochemical stability of a BRB nectar over storage using targeted and 200 untargeted metabolomics, and we relate these chemical changes to their bioactive properties on 201 premalignant oral epithelial cell proliferation. The product was a viscous liquid with pH of 3.4 202 and soluble solids reading of 9 °Brix. Microbial growth observed during storage was below the 203 limit of quantitation (data not shown), indicating that any chemical changes incurred over storage 204 were not due to microbial metabolism. Untargeted metabolomics has been used by others to 205 understand the chemistry of foods in several applications including food authentication, effects 206 of different production practices, the dynamics of fermentation processes, and recently, changes in flavor attributes during storage $^{24-26}$. Here we use the technique to understand how the 207 208 chemistry of BRB nectar, as impacted by storage, may relate to the biological activity of the 209 product.

210

Temperatures

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211 3.1 Untargeted Metabolomics Revealed Large Chemical Variation with Elevated Storage 212

213 Full scan UHPLC-MS-QTOF data was acquired for all nectar samples. Following the 214 extraction, alignment, binning, and filtering of peaks in the data, a total of 1,712 molecular 215 features were considered for further analyses. Overall trends across the dataset were visualized 216 using principal component analysis (PCA) autofitted to three components. Only the first two 217 comonents are displayed in Figure 2 to simplify data interpretation, as the third component only 218 explained 3.8% of the variation. The scores plot in Figure 2 indicates that the samples stored at -219 20 °C were relatively stable over 60 days of storage as demonstrated by their close clustering and 220 proximity to the samples from t_0 . Samples stored at higher temperatures for longer amounts of 221 time were further separated from the t₀ samples along the first component, which explained 222 37.7% of the variation, suggesting that considerable chemical variation was introduced with 223 elevated temperature and time.

224 Partial least squares regression (PLS) was used to further understand how chemical 225 profiles of BRB nectars stored at different temperatures changed over time. A separate model 226 for each storage temperature was generated in which relative feature abundances were regressed 227 against storage time, including t₀ (Table 2). The model for samples stored at -20 °C was of poor quality ($Q^2 = 0.165$; P = 0.03), indicating that storage time was not a strong predictor of 228 229 chemical variation in these samples. This further demonstrated the stability of BRB nectar stored at -20 °C for 60 d. The models for samples stored at 4 °C-35 °C all had a $Q^2 > 0.9$ (P = 0.01), 230 231 which indicated good performance of these models. We focused our analysis on features that 232 were influenced by storage time, regardless of storage temperature, by collating features with 233 $VIP \ge 1$ across all four PLS models. Following a manual data quality review, 73 features were

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234 found to contribute significantly to all four models. Figure 3 displays the mean relative 235 abundance of these features at each time point across storage conditions. Features were clustered 236 using Euclidian distance metric and Ward's linkage rule. The heat map demonstrates that these 237 significant features increased and decreased simultaneously at storage temperatures above -20 238 °C. These relative changes in abundance appeared to be more severe at 25 °C and 35 °C storage, 239 as anticipated. The features in cluster A reflect a pattern of formation by which features were not 240 present at t₀ and were created over time, more so at higher temperatures. At 35 °C, some of these 241 features decreased in abundance before day 60, indicating further degradation of these generated 242 compounds. Cluster B contains features that degraded over time, some which degraded after 20 243 days at 35 °C. The features in cluster C increased in abundance continuously over time with 244 elevated storage temperatures. Many of these features were present at low levels in the t₀ 245 samples. These data demonstrate that above -20 °C, the BRB nectar is a system in dynamic 246 chemical flux over 60 days of storage. 247 Tentative identifications were generated for some features based on plausible database matches from FooDB (www.foodb.ca), a component of the human metabolome database ²⁷. 248 249 Identities were confirmed by authentic standards or by collecting additional MS/MS 250 fragmentation data and comparing to published values when authentic standards were 251 unavailable. These techniques correspond to identification levels 1 and 2, respectively, as

proposed by the Metabolomics Standard Initiative ²⁸. Table 3 displays features that were identified using these methods, all of which have been previously reported in BRBs ²⁹. Catechin and epicatechin are isomeric flavan-3-ols, and their levels have been shown to decline over storage in other products such as apple juice ³⁰, and a variety of blueberry products ³¹. B-type

256 procyanidins are oligomers of catechin and/or epicatechin linked by C-C bonds, and have also

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been reported to be unstable over storage ³¹. The MS/MS fragmentation patterns of the B-type 257 258 procyanidins from the current work closely resembled those reported previously ³². PA is a B-259 ring cleavage product of cyanidin-based anthocyanins that can form as a result of heating or storage, but is also present in fresh BRBs^{29,33}. 260 The lack of plausible database matches for many of the features in clusters A and C in 261 262 Figure 3 led us to hypothesize that these entities may be uncharacterized degradation products of 263 BRB components. The Maillard reaction is a prevalent reaction between reducing sugars and 264 amino acids that occurs over processing and storage of foods. Intermediates in this reaction 265 include reactive carbonyl species that can form adducts with phenolic compounds, such as epicatechin, in food products ³⁴. Kokkinidou and Peterson demonstrated that phenolic-reactive 266 carbonyl species adducts can be decomposed by derivatization with o-phenylenediamine ³⁵. 267 268 When *o*-phenylenediamine was added to BRB nectar extracts, the abundances of 7 features from 269 clusters A and C in Figure 3 were significantly or completely reduced, suggesting that these may

270 be Maillard-related sugar fragmentation-phenolic degradation products.

271

272 3.2 All Extracts of Stored BRB Nectar Inhibited SCC-83-01-82 Cell Growth Similarly

BRB nectar extracts were applied to SCC-83-01-82 premalignant oral epithelial cells to assess the effects of storage time and temperature on their cell growth activities. Extracts of the t₀ nectar samples inhibited cell growth $27.8 \pm 2.8\%$ with inhibition of the stored samples shown in Figure 4. Data were evaluated using a two-way ANOVA model including terms for nectar storage time, temperature, and their interaction. The terms for time (P < 0.01), temperature (P < 0.01), and their interaction (*P*<0.0001) were significant. Few significant differences, however, were seen across time within any one storage condition, except for a 13% difference between

samples stored for 10, 20, and 60 days at 35 °C (Figure 4). Non-significant trends emerged in the

280

281 dataset, but were inconsistent among storage conditions. For example, a non-significant decrease 282 in cell growth activity was observed for nectars stored at 10 °C, but this same trend was not 283 maintained with storage at 25 °C. Thus, the capacity of BRB nectar extracts to inhibit SCC-83-284 01-82 cell growth after 72 hr of incubation was relatively unaffected by nectar storage 285 conditions, despite the large variation seen in the nectar chemical profiles. 286 Few studies have investigated the relationship between storage conditions of berry 287 products, their corresponding chemical profiles, and bioactivity. A decrease in total anthocyanin 288 content was observed over 60 days in blueberry juice produced from two different cultivars and 289 stored at 6 °C and 23 °C. When the anthocyanin fraction of the juice was isolated and applied to 290 HT-29 colorectal adenocarcinoma cells, the authors observed significant decreases in anti-291 proliferative activity after 30–90 days of storage. Only a slight decrease in anti-proliferative 292 activity was noted in samples stored at 23 °C for 60 days, but it was concluded that storage conditions influenced the anthocyanin profiles and biological activities of the juices 36 . The 293 294 untargeted metabolomics approach employed in the present work aims to elucidate the 295 relationship between the chemical profile and biological activity of a berry product in a more 296 comprehensive way. BRBs contain a complex mixture of phytochemicals, thus it is unlikely that 297 any single chemical component can account for the complete bioactivity of the fruit. For 298 example, feeding whole BRB powder, anthocyanin-rich BRB extract, or anthocyanin-deplete 299 extract all suppressed the growth of tumors to an identical amount in a rat model of esophageal

cancer ³⁷. Paudel and colleagues used NMR-based metabolomics to understand the effects of
 BRB cultivar and degree of ripeness on bioactivity in HT-29 colon cancer cells. They observed a
 myriad of biologically active BRB components including anthocyanins, other flavonoids, organic

acids, and ellagic acid derivatives ³⁸. Our data support these findings in that we observed
nominal changes in bioactivity of BRB nectar products with considerably different
phytochemical profiles, further demonstrating that a number of phytochemicals are responsible
for this bioactivity.

307

308 3.3 C3R and its degradation product PA Equally Contribute to the Bioactivity of BRB Nectar 309 Since the anti-proliferative activity of stored nectars was relatively unchanged despite 310 large changes in chemical profiles, we hypothesized that parent phytochemicals, as well as their 311 degradation products, both contribute to the bioactivity of the product due to similarity in 312 structural motifs. Anthocyanins constitute a large portion of the total polyphenols of BRBs, with C3R as a predominant species $^{39-41}$. Given that PA is a reported degradation product of C3R and 313 314 was identified as an important feature in our PLS models, we focused on these two compounds 315 in a model system as proof-of-concept that parent compounds and their associated degradation 316 products can be complementarily bioactive. 317 To understand how these two related compounds changed in the nectar over time, we

active of the extracted information about their abundances from the untargeted UHPLC-MS-QTOF dataset
(Figure 5). The relative change in abundance over time was greater at higher storage
temperatures for both compounds, consistent with prior findings on anthocyanin degradation ⁴².
C3R and PA were subsequently quantitated at t₀ and 60 d of storage at 35 °C using UHPLCMS/MS and authentic standard curves (Table 4). Interestingly, C3R decreased by 13.1 nmol/mg,
while PA increased by 14.9 nmol/mg during storage, demonstrating that these two bioactive
compounds exchanged in near-equimolar amounts in the BRB nectar. It must be noted that

325 anthocyanins can degrade via a multitude of mechanisms to form several different products apart 326 from PA, while PA can also be a degradation product from other phenolics ⁴² 327 Independently, C3R and PA each inhibited the growth of SCC-83-01-82 cells in a dose-328 dependent manner (Figure 6A, B), with increasing concentrations corresponding with increased 329 growth inhibition. The growth inhibition by C3R is similar to levels previously reported on 330 cyanidin-3-O-glucoside isolated from strawberries, which inhibited the growth of CAL-27 malignant oral cancer cells by approximately 50% at a level of 100 μ g/mL (222 μ mol/L)⁴³. The 331 332 current work further validates the bioactivity of cyanidin-based anthocyanins to inhibit cell 333 growth in human oral cell lines. The anticancer activity of PA against oral cancer has previously been demonstrated in animal models ⁴⁴. While the concentrations we tested *in vitro* are higher 334 335 than those found in the BRB nectar, our results show that SCC-83-01-82 cells respond to 336 individual treatments of C3R or PA in a dose-dependent manner. 337 To demonstrate that BRB phytochemicals and their degradation products can each 338 contribute to the biological activity of the nectar, we delivered doses of equal molarity but 339 differing molar ratios of C3R:PA. The conditions used mirror the equimolar exchange of these 340 two compounds observed in the nectar. As shown in Figure 6C, after a starting dose of 170 341 µmol/L, C3R was reduced by 25% in successive treatments, while in parallel the concentrations 342 of PA were increased in 25% increments to a final treatment dose of 170 µM. A consistent level of growth inhibition was maintained across treatments (P = 0.092 for differences among 343 344 treatments) despite differing molar ratios of C3R:PCA. Consequently, our data demonstrates that 345 the loss in bioactivity of a parent phytochemical (C3R) may be recovered by increased levels of 346 their degradation products (PA) (Figure 6C). Previous studies with other cancer models have 347 found the ortho-dihydroxyphenyl structural element of some anthocyanidins, such as cyanidin, to

be critical for anti-cancer properties of these compounds ⁴⁵. Our data suggest that this structural 348 349 moiety, the main molecular structure maintained between C3R and PA, may also play a role in 350 suppressing the growth of SCC-83-01-82 cells. Partial degradation of cyanidin-3-glucoside in cell culture media has been previously reported, with PA as the primary degradation product 46 . 351 352 While this represents an inherent limitation of studying anthocyanins *in vitro*, it further validates 353 the idea that phytochemical degradation products can maintain active chemical moieties, and 354 thus bioactivity. We speculate that this phenomenon of degrading phytochemicals while 355 maintaining active chemical moieties occurs on a larger scale with other components of the 356 nectar. Additionally, it is plausible that phytochemicals that remain unchanged throughout 357 storage contribute significantly to bioactivity. And while not addressed in the current study, it is 358 also conceivable that the biochemical signaling and activation mechanisms underlying the 359 growth inhibition shifted with changing nectar chemical profiles. In addition, our bioassay was 360 an *in vitro* model with oral cells that can be directly exposed to BRB phytochemicals *in vivo*. 361 Not addressed in this study is the impact that storage-induced changes in BRB phytochemicals 362 affects their bioaccessibility and bioavailability in the remainder of the GI tract, which could 363 have implications for their actions elsewhere in the body.

364

365 4. Conclusions

We investigated the impact of storage on the phytochemical stability and bioactivity of a BRB nectar product. Our data demonstrate that nectar stored at -20 °C is chemically stable over 60 days, but storage above this temperature introduces large amounts of chemical variation through a variety of mechanisms including cleavage of phenolic compounds and potential adduct formation with reactive carbonyl species. Despite the large chemical variation observed using

371 untargeted metabolomics, storage conditions had minimal impact on the ability of the nectar to 372 differentially inhibit growth in premalignant oral epithelial cells. Exploration of this phenomenon 373 *in vitro* supports our hypothesis that degradation products of bioactive phytochemicals also 374 demonstrate bioactivity, allowing maintenance of growth inhibition capacity, through 375 independent, cooperative, or redundant mechanisms. This work demonstrates that BRBs are a 376 complex mixture of compounds with potential anticancer activities. Assigning functional activity 377 to a single black raspberry compound or metabolite fails to explain and appreciate this fluidity, 378 as different compounds increase and decrease with the dynamics of storage. It remains important 379 to dissect these pleiotropic phytochemical bioactives to fully understand the health benefits and 380 consequences of consuming BRBs and their components. 381 382 Acknowledgements 383 The authors would like to thank David M. Phinney, M.S. for his assistance in processing the 384 nectar utilized in this study. This work was supported by USDA-NIFA National Needs 385 Fellowship (2014-38420-21844), the Lisa and Dan Wampler Endowed Fellowship for Food and 386 Health Research and Foods for Health, a focus area of the Discovery Themes Initiative at The 387 Ohio State University. 388 389 **Abbreviations Used** 390 BRB: black raspberry; SCC: squamous cell carcinoma; C3R: cyanidin-3-O-rutinoside; PA: 391 protocatechuic acid; PLS: partial least squares; PCA: principal components analysis; VIP: 392 variable importance on projection 393

394	Conf	licts of Interest
395	The a	authors have no conflicts of interest to declare.
396		
397	5. Re	ferences
398	(1)	Kula, M.; Krauze-Baranowska, M. Nutr. Cancer 2016, 68, 18–28.
399	(2)	Kresty, L. A.; Mallery, S. R.; Stoner, G. D. J. Berry Res. 2016, 6, 251-261.
400	(3)	Stoner, G. D. Cancer Prev. Res. 2009, 2, 187–194.
401	(4)	Liu, R. H. Am. J. Clin. Nutr. 2003, 78, 517S-520S.
402	(5)	Warner, B. M.; Casto, B. C.; Knobloch, T. J.; Accurso, B. T.; Weghorst, C. M. Oral Surg.
403		<i>Oral Med. Oral Pathol. Oral Radiol.</i> 2014 , <i>118</i> , 674–683.
404	(6)	Knobloch, T. J.; Uhrig, L. K.; Pearl, D. K.; Casto, B. C.; Warner, B. M.; Clinton, S. K.;
405		Sardo-Molmenti, C. L.; Ferguson, J. M.; Daly, B. T.; Riedl, K.; Schwartz, S. J.; Vodovotz,
406		Y.; Buchta, A. J.; Schuller, D. E.; Ozer, E.; Agrawal, A.; Weghorst, C. M. Cancer Prev.
407		<i>Res.</i> 2016 , <i>9</i> , 159–171.
408	(7)	Oghumu, S.; Casto, B. C.; Ahn-Jarvis, J.; Weghorst, L. C.; Maloney, J.; Geuy, P.;

- 409 Horvath, K. Z.; Bollinger, C. E.; Warner, B. M.; Summersgill, K. F.; Weghorst, C. M.;
- 410 Knobloch, T. J. Front. Immunol. 2017, 8.
- 411 (8) Mallery, S. R.; Tong, M.; Shumway, B. S.; Curran, A. E.; Larsen, P. E.; Ness, G. M.;
- 412 Kennedy, K. S.; Blakey, G. H.; Kushner, G. M.; Vickers, A. M.; Han, B.; Pei, P.; Stoner,
- 413 G. D. Clin. Cancer Res. 2014, 20, 1910–1924.
- 414 (9) El-Bayoumy, K.; Chen, K. M.; Zhang, S. M.; Sun, Y. W.; Amin, S.; Stoner, G.;
- 415 Guttenplan, J. B. Chem. Res. Toxicol. 2017, 30, 126–144.
- 416 (10) Petersen, P. E.; Bourgeois, D.; Ogawa, H.; Estupinan-Day, S.; Ndiaye, C. Bull. World

- 417 *Health Organ.* **2005**, *83*, 661–669.
- 418 (11) Ferlay, J.; Soerjomataram, I.; Dikshit, R.; Eser, S.; Mathers, C.; Rebelo, M.; Parkin, D.
- 419 M.; Forman, D.; Bray, F. Int. J. Cancer 2015, 136, E359–E386.
- 420 (12) Bsoul, S.; Huber, M. A.; Terezhalmy, G. T. J. Contemp. Dent. Pract. 2005, 6, 1–16.
- 421 (13) Casto, B. C.; Knobloch, T. J.; Weghorst, C. M. In Berries and Cancer Prevention; Stoner,
- 422 G. D.; Seeram, N., Eds.; Springer Science+Business Media: New York, NY, 2011; pp.
 423 189–207.
- 424 (14) Howard, L. R.; Prior, R. L.; Liyanage, R.; Lay, J. O. J. Agric. Food Chem. 2012, 60,
 425 6678–6693.
- 426 (15) Hager, A.; Howard, L. R.; Prior, R. L.; Brownmiller, C. J. Food Sci. 2008, 73, H134–
 427 H140.
- 428 (16) Gu, J.; Ahn-Jarvis, J.; Riedl, K. M.; Schwartz, S. J.; Clinton, S. K.; Vodovotz, Y. J Agric
 429 Food Chem 2014, 62, 3997–4006.
- 430 (17) Manach, C.; Hubert, J.; Llorach, R.; Scalbert, A. *Mol. Nutr. Food Res.* 2009, *53*, 1303–
 431 1315.
- 432 (18) Thévenot, E. A.; Roux, A.; Xu, Y.; Ezan, E.; Junot, C. J. Proteome Res. 2015, 14, 3322–
 433 3335.
- 434 (19) Kemsley, E. K.; Le Gall, G.; Dainty, J. R.; Watson, A. D.; Harvey, L. J.; Tapp, H. S.;
 435 Colquhoun, I. J. *Br. J. Nutr.* 2007, *98*, 1–14.
- 436 (20) Han, C.; Ding, H.; Casto, B.; Stoner, G. D.; D'Ambrosio, S. M. *Nutr. Cancer* 2005, *51*,
 437 207–217.
- 438 (21) Ding, H.; Han, C.; Guo, D.; Chin, Y. W.; Ding, Y.; Kinghorn, A. D.; D'Ambrosio, S. M.
 439 *Nutr. Cancer* 2009, *61*, 348–356.

440	(22)	Lee, H.; Li, D.; Prior, T.; Casto, B. C.; Weghorst, C. M.; Shuler, C. F.; Milo, G. E. Med
441		<i>Biochem.</i> 1997, 13, 419–434.

- 442 (23) Bishayee, A.; Haskell, Y.; Do, C.; Siveen, K. S.; Mohandas, N.; Sethi, G.; Stoner, G. D.
- 443 *Crit. Rev. Food Sci. Nutr.* **2016**, *56*, 1753–1775.
- 444 (24) Cevallos-Cevallos, J. M.; Reyes-De-Corcuera, J. I.; Etxeberria, E.; Danyluk, M. D.;
- 445 Rodrick, G. E. *Trends Food Sci. Technol.* **2009**, *20*, 557–566.
- 446 (25) Ronningen, I.; Miller, M.; Xia, Y.; Peterson, D. G. J. Agric. Food Chem. ePub ahead.
- 447 (26) Ronningen, I.; Peterson, D. G. J. Agric. Food Chem. 2018, 66, 682–688.
- 448 (27) Wishart, D. S.; Jewison, T.; Guo, A. C.; Wilson, M.; Knox, C.; Liu, Y.; Djoumbou, Y.;
- 449 Mandal, R.; Aziat, F.; Dong, E.; Bouatra, S.; Sinelnikov, I.; Arndt, D.; Xia, J.; Liu, P.;
- Yallou, F.; Bjorndahl, T.; Perez-Pineiro, R.; Eisner, R.; Allen, F.; Neveu, V.; Greiner, R.;
 Scalbert, A. *Nucleic Acids Res.* 2013, *41*, 801–807.
- 452 (28) Sumner, L. W.; Samuel, T.; Noble, R.; Gmbh, S. D.; Barrett, D.; Beale, M. H.; Hardy, N.;
- 453 Harnly, J.; Higashi, R.; Kopka, J.; Lane, A. N.; Lindon, J.; Marriott, P.; Nicholls, A. W.;
- 454 Reily, M. D.; Thaden, J. J.; Viant, M. R. *Metabolomics* **2007**, *3*, 211–221.
- 455 (29) Kula, M.; Majdan, M.; Głód, D.; Krauze-Baranowska, M. J. Food Compos. Anal. 2016,
 456 52, 74–82.
- 457 (30) Van Der Sluis, A. A.; Dekker, M.; Van Boekel, M. A. J. S. J. Agric. Food Chem. 2005,
 458 53, 1073–1080.
- 459 (31) Brownmiller, C.; Howard, L. R.; Prior, R. L. J. Agric. Food Chem. 2009, 57, 1896–1902.
- 460 (32) Gu, L.; Kelm, M. A.; Hammerstone, J. F.; Beecher, G.; Holden, J.; Haytowitz, D.; Prior,
- 461 R. L. J. Agric. Food Chem. 2003, 51, 7513–7521.
- 462 (33) Stintzing, F. C.; Carle, R. Trends Food Sci. Technol. 2004, 15, 19–38.

- 463 (34) Totlani, V. M.; Peterson, D. G. J. Agric. Food Chem. 2005, 53, 4130-4135.
- 464 (35) Kokkinidou, S.; Peterson, D. G. Food Funct. 2013, 4, 1093.
- 465 (36) Srivastava, A.; Akoh, C. C.; Yi, W.; Fischer, J.; Krewer, G. J. Agric. Food Chem. 2007,
 466 55, 2705–2713.
- 467 (37) Wang, L.; Hecht, S. S.; Carmella, S. G.; Yu, N.; Larue, B.; Henry, C.; Mcintyre, C.;
- 468 Rocha, C.; Lechner, J. F.; Stoner, G. D. *Cancer Prev. Res.* **2009**, *2*, 84–93.
- 469 (38) Paudel, L.; Wyzgoski, F. J.; Giusti, M. M.; Johnson, J. L.; Rinaldi, P. L.; Scheerens, J. C.;
- 470 Chanon, A. M.; Bomser, J. A.; Miller, A. R.; Hardy, J. K.; Reese, R. N. J. Agric. Food
- 471 *Chem.* **2014**, *62*, 1989–1998.
- 472 (39) Rothwell, J. A.; Perez-Jimenez, J.; Neveu, V.; Medina-Rem??n, A.; M'Hiri, N.; Garc??a-
- 473 Lobato, P.; Manach, C.; Knox, C.; Eisner, R.; Wishart, D. S.; Scalbert, A. *Database* 2013,
 474 2013, 1–8.
- 475 (40) Tulio, A. Z.; Reese, R. N.; Wyzgoski, F. J.; Rtnaldi, P. L.; Fu, R.; Scheerens, J. C.; Miller,
 476 A. R. J. Agric. Food Chem. 2008, 56, 1880–1888.
- 477 (41) Torre, L. C.; Barritt, B. H. J. Food Sci. 1977, 42, 488–490.
- 478 (42) Patras, A.; Brunton, N. P.; O'Donnell, C.; Tiwari, B. K. *Trends Food Sci. Technol.* 2010,
 479 *21*, 3–11.
- 480 (43) Zhang, Y. J.; Seeram, N. P.; Lee, R.; Feng, L.; Heber, D. J. Agric. Food Chem. 2008, 56,
 481 670–675.
- 482 (44) Tanaka, T.; Tanaka, M. J. Exp. Clin. Med. 2011, 3, 27–33.
- 483 (45) Hou, D.-X.; Kai, K.; Li, J.-J.; Lin, S.; Terahara, N.; Wakamatsu, M.; Fujii, M.; Young, M.
- 484 R.; Colburn, N. *Carcinogenesis* **2003**, *25*, 29–36.
- 485 (46) Kay, C. D.; Kroon, P. A.; Cassidy, A. Mol. Nutr. Food Res. 2009, 53, 92–101.

486	Figure 1. Summary of untargeted metabolomics data pre-treatment and analysis.
487	
488	Figure 2. Principal component analysis of all samples colored by storage temperature and
489	labeled according to length of storage.
490	
491	Figure 3. Heat map of molecular features with VIP>1 in PLS models for storage at 4–35 °C.
492	Features were clustered using Euclidian distance metrics and Ward's linkage rule. * Denotes
493	potential Maillard-related sugar fragmentation-phenolic degradation products determined after
494	derivatization with o-phenylenediamine.
495	
496	Figure 4. Growth inhibition of SCC-83-01-82 cells by extracts of BRB nectar stored at
497	increasing temperatures. ANOVA terms for storage time, temperature, and their interaction were
498	significant ($P < 0.01$). Only significant differences within each storage temperature are denoted
499	(*).
500	
501	Figure 5. Averaged relative abundances of C3R and PA over time in each storage condition.
502	
503	Figure 6. Dose-response relationship between increasing levels of C3R (A) and PA (B) and
504	growth inhibition of SCC-83-01-82 cells. (C) When the ratio of C3R and PA were varied in
505	equimolar solutions (molarity on right y-axis), growth inhibition (left y-axis) was maintained (P
506	= 0.092 for differences among treatments).
507	

Ingredients	% Wet Basis
Water	89.9
Sucrose	3.0
Pectin	0.5
Corn Syrup	1.0
BRB Powder	5.6
TOTAL	100.0

Table 1. BRB nectar beverage formulation

 Table 2. PLS model cross validation results

Storage Temperature	Q ²	$P Q^2$	RMSEE
-20 °C	0.165	0.03	6.98
4 °C	0.948	0.01	1.04
10 °C	0.980	0.01	1.58
25 °C	0.919	0.01	0.997
35 °C	0.985	0.01	2.05

Table 3. Level 1 and 2 identified compounds from list of features with VIP>1 across all four PLS models

Compound Name	Molecular Formula	Retention time (min)	[M-H] ⁻	Δppm	Heat map cluster
Epicatechin ¹	C ₁₅ H ₁₄ O ₆	3.6	289.0718	0	В
Catechin ¹	$C_{15}H_{14}O_{6}$	3.3	289.0718	0	В
B-type procyanidin dimer A^2	$C_{30}H_{26}O_{12}$	3.6	577.1344	1	В
B-type procyanidin dimer B^2	$C_{30}H_{26}O_{12}$	3.4	577.1398	8	В
Protocatechuic acid ¹	$C_7H_6O_4$	2.9	153.0196	2	С

¹ Level 1 identified features, ² Level 2 identified features

TimeC3R(d)(μg/mg extract)		PCA (µg/mg extract)
0	8.02 ± 0.20	$3.27 \times 10^{-2} \pm 8.1 \times 10^{-3}$
60	$0.20 \pm 5.7 \times 10^{-4}$	2 33 ± 0 13

Table 4. Quantitative analysis of C3R and PA in nectar from t_0 and 60 days at 35 $^{\circ}\mathrm{C}$





127x169mm (300 x 300 DPI)



Figure 2. Principal component analysis of all samples colored by storage temperature and labeled according to length of storage.

101x101mm (300 x 300 DPI)



Figure 3. Heat map of molecular features with VIP>1 in PLS models for storage at 4–35 °C. Features were clustered using Euclidian distance metrics and Ward's linkage rule. * Denotes potential Maillard-related sugar fragmentation-phenolic degradation products determined after derivatization with o-phenylenediamine.

134x101mm (300 x 300 DPI)



Figure 4. Growth inhibition of SCC-83-01-82 cells by extracts of BRB nectar stored at increasing temperatures. ANOVA terms for storage time, temperature, and their interaction were significant (P<0.01). Only significant differences within each storage temperature are denoted (*).

69x38mm (300 x 300 DPI)



Figure 5. Averaged relative abundances of C3R and PA over time in each storage condition.

77x29mm (300 x 300 DPI)



Figure 6. Dose-response relationship between increasing levels of C3R (A) and PA (B) and growth inhibition of SCC-83-01-82 cells. (C) When the ratio of C3R and PA were varied in equimolar solutions (molarity on right y-axis), growth inhibition (left y-axis) was maintained (P = 0.092 for differences among treatments).





39x21mm (300 x 300 DPI)