ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

Communication

Consecutive visible-light photoredox decarboxylative couplings of adipic acid active esters with alkynyl sulfones leading to cyclic compounds

Jingjing Li, Hua Tian, Min Jiang, Haijun Yang, Yufen Zhao and Hua Fu*

Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX 5 DOI: 10.1039/b000000x

Novel and efficient consecutive photoredox decarboxylative couplings of adipic acid active esters (bis(1,3-dioxoisoindolin-2-yl)-substituted hexanedioates) with substituted 1-(2arylethynylsulfonyl)benzenes have been developed under visible-light photocatalysis. The successive photoredox decarboxylative C-C bond formation at room temperature afforded the corresponding cyclic compounds in good yields with tolerance of some functional groups.

- Formation of carbon-carbon bonds is a fundamental and ¹⁵ important chemical transformation in organic synthesis. However, it is a great challenge to develop consecutive formation of two C-C bonds using readily available starting materials under mild conditions. Carboxylic acids and their derivatives widely occur in organic molecules and natural products, and their chemical
- ²⁰ transformations provide diverse compounds.¹ For example, the decarboxylative strategy of carboxylic acids and their derivatives has provided some valuable reactions in organic synthesis,² such as Heck-type reactions,³ allylations,⁴ redox-neutral cross-coupling reactions,⁵ and oxidative arylations.⁶ Recently, visible-
- ²⁵ light photoredox catalysis has attracted much attention, and it has emerged as a powerful activation protocol in new chemical transformations.⁷ Correspondingly, some achievements have been gained on photoredox decarboxylative couplings of carboxylic acids and their derivatives were used as the radical precursors.⁸
- ³⁰ DiRocco and coworkers have developed the use of stable organic acid peroxides activated by visible-light photoredox catalysis in the presence of Ir(III) catalysts, and the reactions effectively achieved the direct methyl-, ethyl-, and cyclopropylation of a variety of biologically active heterocycles.⁹ Tunge and co-
- ³⁵ workers have reported the decarboxylative allylation of amino alkanoic acids and their esters via the dual catalysis of $Ir(ppy)_2(bpy)[BF_4]$ and $Pd(PPh_3)_4$.¹⁰ Organic carboxylic acid active esters¹¹ and free carboxylic acids¹² as the radical precursors have been used in the visible-light photoredox couplings in the
- ⁴⁰ presence of photocatalysts (PC) and radical acceptors (RA) (see Scheme 1a). More importantly, MacMillan and co-workers have developed a pioneering photoredox decarboxylative coupings of *N*-protected α -amino acids.¹³ To the best of our knowledge, there is no report on photoredox decarboxylative coupling of
- ⁴⁵ dicarboxylic acids and their derivatives thus far. Herein, we report consecutive visible-light photoredox couplings of substituted adipic acid active esters (Scheme 1b). Our strategy is

as follows: Treatment of bis(1,3-dioxoisoindolin-2-yl)-substituted hexanedioate (1) under visible-light photocatalysis produces ⁵⁰ radical **D**, and intermolecular coupling (the first coupling) of **D** with alkynyl sulfone (2) provides **F**. Subsequently, photocatalysis of **F** forms radical **J**, and intramolecular coupling (the second coupling) of **J** affords the target product (3).



55 Scheme 1 (a) The previous photoredox decarboxylative couplings of carboxylic acid active esters or free carboxylic acids. (b) Our strategy on consecutive photoredox decarboxylative couplings of adipic acid active esters (1) with alkyne sulfones (2).

Consecutive photoredox decarboxylative couplings of bis(1,3-60 dioxoisoindolin-2-yl)hexanedioate (**1**a) with 1 - (2 phenylethynylsulfonyl)benzene (2a) was selected as the model reaction to optimize conditions including photocatalysts, solvents, atmosphere and time (see Table S1 in Supporting Information). The results showed that the optimal photoredox conditions are as 65 follows: 1.0 mol% [Ru(bpy)₃]Cl₂ as the photocatalyst, 4.4 equiv of diisopropylethylamine (DIPEA) and 3.0 equiv of Hantzsch ester (HE) (relative to amount of 2a) as the radical initiators and reductants in dichloromethane (DCM) at room temperature under argon atmosphere. After obtaining the optimized photoredox 70 conditions, we investigated the substrate scope on consecutive decarboxylative couplings of various bis(1,3-dioxoisoindolin-2yl)-substituted hexanedioates (1) with 1-(2arylethynylsulfonyl)benzenes (2) (Table 1). We first attempted various substituted 1-(2-arylethynylsulfonyl)benzenes (2) using 75 bis(1,3-dioxoisoindolin-2-yl)hexanedioate (1a) as the partner. Different substituted 1-(2-arylethynylsulfonyl)benzenes (2) displayed obviously different reactivity, and substrates 2 containing electron-donating groups gave higher yields than those containing electron-withdrawing groups. For example, alkyne

ChemComm Accepted Manuscri

This journal is $\ensuremath{\mathbb{C}}$ The Royal Society of Chemistry [year]

ChemComm Accepted Manuscrip

sulfones with methyl, ethyl, methoxyl on aryl rings (see **3b-h**) provided higher yields than those with ester groups (see **3p-r**). Alkyne sulfones containing biphenyl and naphthalene (see **3t** and **3u**) also donated the corresponding products in 52% and 66% 5 yields, respectively. Other bis(1,3-dioxoisoindolin-2-yl)-

- substituted hexanedioates (1) were investigated, and *cis* and *trans*-form products were obtained in good yields (*Note:* Z/E ratios were determined by ¹H NMR) (see **3v-ac**). The consecutive visible-light photoredox decarboxylicative couplings showed
- ¹⁰ tolerance of some functional groups including ether (see 3f-h, 3w and 3ac), C-F bond (see 3i), C-Cl bond (see 3j-l, 3x and 3aa), C-Br bond (see 3m-o), esters (see 3p-r and 3y) and amide (see 3s).

Table 1 Consecutive photoredox decarboxylative couplings of bis(1,3dioxoisoindolin-2-yl)-substituted hexanedioates (1) with alkynyl sulfones 15 $(2)^a$



^{*a*} Reaction conditions: under irradiation of visible light and argon atmosphere, bis(1,3-dioxoisoindolin-2-yl)-substituted hexanedioate (1) (0.3 mmol), 1-(2-arylethynylsulfonyl)benzene (2) (0.2 mmol), [Ru(bpy)₃Cl₂] (2 µmol), diisopropylethylamine amine (DIPEA) (0.88 mmol), Hantzsch ester (HE) (0.6 mmol), dichloromethane (DCM) (1.5 mL), temperature (rt, ~25 °C), time (4-16 h) in a sealed Schlenk tube. ^{*b*} Isolated yield. Boc = *tert*-butyloxycarbonyl.

In order to explore the visible-light photoredox decarboxylative mechanism, reaction of bis(1,3-dioxoisoindolin-2-yl) hexanedioate (1a) with 1-(2-phenylethynylsulfonyl)benzene ²⁰ (2a) was performed in the presence of three equiv of 2,2,6,6tetramethyl-1-piperidinyloxy (TEMPO) as the radical-trapping agent under the standard conditions, and the reaction did not work, which implied that the reaction underwent a free-radical intermediate process. In addition, PhSO₂H as by-product was 25 isolated from coupling 1a and 2a under the standard conditions, and its structure was confirmed by ¹H and ¹³C NMR. Therefore, a plausible mechanism on the consecutive visible-light photoredox decarboxylative couplings is proposed in Scheme 2 according to the results above and the previous references.⁷⁻¹⁴ Irradiation of 30 Ru(bpy)₃²⁺ with visible light leads to the oxidizing excited-state $[Ru(bpy)_3^{2^+}]^*$ $(E_{1/2}^{*II/I} = +0.77 \text{ V vs SCE in MeCN}^{7a})$, and the photoexcited catalyst accepts an electron from DIPEA ($E_{1/2}^{red}$ = +0.65 V vs SCE in MeCN¹⁴) to provide $Ru(bpy)_3^+$, in which DIPEA transforms into A. One electron in $Ru(bpy)_3^+$ transfers to 35 phthalimide in 1 produces radical B regenerating catalyst $\operatorname{Ru}(\operatorname{bpy})_{3}^{2^{+}}$, and subsequent leaving of phthalimide anion (C) and carbon dioxide from B gives radical D. Addition of D to alkyne sulfone (2) leads to radical E, and homolytic cleavage C-S bond in E donates F releasing radical PhSO2. Reaction of PhSO2 with 40 Hantzsch ester (HE) provides PhSO₂H and radical G,^{7a} and treatment of the photoexcited catalyst with G gives $Ru(bpy)_3^+$ and **H**. Similarly, treatment of **F** with $Ru(bpy)_3^+$ produces radical **I** regenerating catalyst $Ru(bpy)_3^{2+}$, and elimination of C and carbon dioxide from I affords radical J. Intramolecular cyclization of J 45 forms radical K. Finally, treatment of K with HE generates the

desired target product (3) liberating G.



Scheme 2 A plausible mechanism on the consecutive visible-light photoredox decarboxylative couplings.



50

Scheme 3 Reaction of ((methylsulfonyl)ethynyl)benzene (2u) with 1a under the standard conditions.

This journal is © The Royal Society of Chemistry [year]

2 | Journal Name, [year], [vol], 00-00

75

80

Reaction of ((methylsulfonyl)ethynyl)benzene (2u) with 1a was attempted under the standard conditions, and the target product (3a) was obtained in 66% yield (Scheme 3).

We attempted 1-(oct-1-ynylsulfonyl)benzene (4) as the ⁵ radical accepter under the standard conditions. Unfortunately, visible-light photoredox decarboxylative coupling of **1a** with **4** gave poor yield (Scheme 4a), so 1-(2alkylethynylsulfonyl)benzenes are not good substrates. Further, visible-light photoredox decarboxylative coupling of pimelic acid ¹⁰ active ester (**6**) with **2a** was performed, and six-membered cycle

7 and internal alkyne 8 were obtained in 31% and 13% yields, respectively (Scheme 4b).



Scheme 4 (a) Visible-light photoredox decarboxylative coupling of 1a with 1-15 (oct-1-ynylsulfonyl)benzene (4). (b) Visible-light photoredox decarboxylative coupling of pimelic acid active ester (6) with 2a.

In summary, we have developed novel and efficient consecutive visible-light photoredox decarboxylative couplings of substituted adipic acid active esters (bis(1,3-dioxoisoindolin-2-

- 20 yl)-substituted hexanedioates) with 1-(2arylethynylsulfonyl)benzenes under the assistant of the photocatalyst [Ru(bpy)₃]Cl₂ and visible-light, in which the starting materials are readily available. Importantly, the reactions were performed at room temperature, and the successive
- ²⁵ photoredox decarboxylative couplings led to formation of two C-C bonds. The present discovery should provide a novel and practical strategy for synthesis of cyclic molecules, and we believe that it will find wide applications in various fields.

Financial support for this work was provided by the National ³⁰ Natural Science Foundation of China (Grant Nos. 21372139 and 21221062), and Shenzhen Sci & Tech Bureau (CXB201104210014A).

Notes and references

Key Laboratory of Bioorganic Phosphorus Chemistry and Chemical 35 Biology (Ministry of Education), Department of Chemistry, Tsinghua University, Beijing 100084, P. R. China. Fax: 86 10 62781695; E-mail: fuhua@mail.tsinghua.edu.cn

[†] Electronic supplementary information (ESI) available: General ⁴⁰ procedure for visible-light photoredox synthesis of internal alkynes, characterization data for compounds **3a-ac**, references, and ¹H and ¹³C NMR spectra of compounds **3a-ac**. See http://dx.doi.org/10.1039/b000000x/

- 45 1 (a) P. Gallezot, Chem. Soc. Rev., 2015, 41, 1538; (b) A. J. J. Straathof, Chem. Rev., 2014, 41, 1871.
- For recent reviews, see: (a) J. D. Weaver, A. Recio, III, A. J. Grenning and J. A. Tunge, *Chem. Rev.*, 2011, 111, 1846; (b) N. Rodriguez and L. J. Gooßen, *Chem. Soc. Rev.*, 2011, 40, 5030; (c) W. I. Dzik, P. P. Lange and L. J. Gooßen, *Chem. Sci.*, 2012, 3, 2671; (d)
- 1. DZik, 1.1. Lange and L. J. Goopen,*Chem. Sci.*, 2012,**3**, 2071, (<math>a

L. J. Goonßen, N. Rodríguez and K. Gooßenn, Angew. Chem. Int. Ed., 2008, 47, 3100.

- 3 (a) A. G. Myers, D. Tanaka and M. R. Mannion, J. Am. Chem. Soc., 2002, **124**, 11250; (b) P. Forgione, M. C. Brochu, M. Stonge, K. H.
- ⁵⁵ Thesen, M. D. Bailey and F. Bilodeau, J. Am. Chem. Soc., 2006, **128**, 11350; (c) A. Maehara, H. Tsurugi, T. Satoh and M. Miura, Org. Lett., 2008, **10**, 1159; (d) P. Hu, J. Kan, W. Su and M. Hong, Org. Lett., 2009, **11**, 2341.
- 4 D. K. Rayabarapu and J. A. Tunge, *J. Am. Chem. Soc.*, 2005, **127**, 13510.
- 5 (a) L. J. Gooβen, G. Deng and L. M. Levy, *Science*, 2006, **313**, 662;
 (b) L. J. Gooβen, N. Rodríguez, B. Melzer, C. Linder, G. Deng and L. M. Levy, *J. Am. Chem. Soc.*, 2007, **129**, 4824; (c) G. Hu, Y. Gao, Y. Zhao, *Org. Lett.*, 2014, **16**, 4464.
- 6 (a) A. Voutchkova, A. Coplin, N. E. Leadbeater and R. H. Crabtree, *Chem. Commun.*, 2008, 6312; (b) C. Wang, I. Piel and F. Glorius, J. *Am. Chem. Soc.*, 2009, **131**, 4194; (c) L. J. Goonβen, N. Rodríguez, P. P. Lange and C. Linder, *Angew. Chem. Int. Ed.*, 2010, **49**, 1111.
- For selected reviews and books on visible-light photoredox catalysis,
 see: (a) C. K. Prier, D. A. Rankic and D. W. C. MacMillan, *Chem. Rev.*, 2013, 113, 5322; (b) T. P. Yoon, M. A. Ischay and J. Du, *Nat. Chem.*, 2010, 2, 527; (c) J. M. R. Narayanam and C. R. J.
 Stephenson, *Chem. Soc. Rev.*, 2011, 40, 102; (d) J. W. Tucker and C.
 R. J. Stephenson, J. Org. Chem., 2012, 77, 1617; (e) K. Zeitler,
 - Angew. Chem. Int. Ed., 2009, 48, 9785; (f) L. Shi and W. Xia, Chem. Soc. Rev., 2012, 41, 7687; (g) J. Xuan and W.-J. Xiao, Angew. Chem. Int. Ed., 2012, 51, 6828; (h) D. P. Hari and B. König, Angew. Chem. Int. Ed., 2013, 52, 4734; (i) Y. Xi, H. Yi and A. Lei, Org. Biomol. Chem., 2013, 11, 2387; (j) Chemical Photocatalysis (Ed.: B. König), De Gruyter, Stuttgart, 2013.
- 8 J. Xuan, Z.-G. Zhang and W.-J. Xiao, Angew. Chem. Int. Ed., 2015, 54, 15632.
- Using carboxylic peroxides acids as the radical precursors, see: D. A. DiRocco, K. Dykstra, S. Krska, P. Vachal, D. V. Conway and M. Tudge, *Angew. Chem. Int. Ed.*, 2014, 53, 4802.
- Using amino alkanoic acids and esters as the radical precursors, see: S. B. Lang, K. M. O'Nele and J. A. Tunge, *J. Am. Chem. Soc.*, 2014, 136, 13606.
- Using carboxylic acid active esters as the radical precursors, see: (a)
 D. H. R. Barton, D. Crich, Y. Hervé, P. Potier and J. Thierry, *Tetrahedron*, 1985, 41, 4347; (b) D. H. R. Barton, D. Bridon, Y. Hervé, P. Potier, J. Thierry and S. Z. Zard, *Tetrahedron*, 1986, 42, 4983; (c) K. Okada, K. Okamoto, N. Morita, K. Okubo and M. Oda, *J. Am. Chem. Soc.*, 1991, 113, 9401; (d) K. Okada, K. Okubo, N.
- Morita and M. Oda, *Tetrahedron Lett.*, 1992, **33**, 7377; (e) M.
 Zlotorzynska and G. M. Sammis, *Org. Lett.*, 2011, **13**, 6264; (f) M. J.
 Schnermann and L. E. Overman, *Angew. Chem. Int. Ed.*, 2012, **51**, 9576; (g) G. L. Lackner, K. W. Quasdorf and L. E. Overman, *J. Am. Chem. Soc.*, 2013, **135**, 15342; (h) J. Xie, P. Xu, H. Li, Q. Xue, H.
 Jin, Y. Cheng and C. Zhu, *Chem. Commun.*, 2013, **49**, 5672; (i) J.
- Jin, Y. Cheng and C. Zhu, *Chem. Commun.*, 2013, **49**, 5672; (i) J. Yang, J. Zhang, L. Qi, C. Hu and Y. Chen, *Chem. Commun.*, 2015, **51**, 5275; (j) H. Huang, G. Zhang and Y. Chen, *Angew. Chem. Int. Ed.*, 2015, **54**, 7872; (k) H. Tan, H. Li, W. Ji and L. Wang, *Angew. Chem. Int. Ed.*, 2015, **54**, 8374.
- ¹⁰⁵ 12 Using carboxylic acids as the radical precursors, see: (a) J. C. T. Leung, C. Chatalova-Sazepin, J. G. West, M. Rueda-Becerril, J.-F. Paquin and G. M. Sammis, *Angew. Chem. Int. Ed.*, 2012, **51**, 10804; (b) J. Liu, Q. Liu, H. Yi, C. Qin, R. Bai, X. Qi, Y. Lan and A. Lei, *Angew. Chem. Int. Ed.*, 2014, **53**, 502.
- 110 13 Using N-Boc α-amino acids as the radical precursors, see: (a) A. Noble and D. W. C. MacMillan, J. Am. Chem. Soc., 2014, 136, 11602; (b) L. Chu, C. Ohta, Z. Zuo and D. W. C. MacMillan, J. Am. Chem. Soc., 2014, 136, 10886; (c) Z. Zuo and D. W. C. MacMillan, J. Am. Chem. Soc., 2014, 136, 5257; (d) Z. Zuo, D. T. Ahneman, L.
- Chu, J. A. Terrett, A. G. Doyle and D. W. C. MacMillan, *Science*, 2014, 345, 437.
 C. Gro, J. Li, J. Xu, H. Yang and H. Fu, *Chem. Commun.* 2016, 52
 - 14 C. Gao, J. Li, J. Yu, H. Yang and H. Fu, *Chem. Commun.*, 2016, **52**, 7292.

This journal is © The Royal Society of Chemistry [year]