

Journal of Materials Chemistry B

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

COMMUNICATION

An Amphiphilic Squarylium Indocyanine Dye for Long-term Tracking of Lysosomes

Cite this: DOI: 10.1039/x0xx00000x

Chendong Ji,^a Yang Zheng,^b Jie Li,^a Jie Shen^{*b}, Wantai Yang^a and Meizhen Yin^{*a}

Received 00th January 2012,

Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

A novel amphiphilic squarylium indocyanine (**LysoCy**) is reported for remarkable lysosome tracking in living cells. **LysoCy** performs as a promising lysosome tracker with low cytotoxicity, strong binding affinity and clear subcellular labelling. The long-term dynamics of lysosomes can be revealed by **LysoCy** up to 2 day's culture, while the working duration of commercial dye is no longer than 1 hour. Our work provides a good alternative tool for long-term live cell imaging.

To visualize the subcellular organelles and biological dynamics, organelle-specific markers have attracted much interest for decades.¹⁻⁸ Lysosomes, the vital organelles, are part of endomembrane system and are essential for multiple biological processes such as plasma membrane repair, cell homeostasis, energy metabolism and immune response.⁹ To reveal the morphology and function of lysosomes, the application of fluorescent lysosome tracker draws increasing attention recently.¹⁰⁻¹³

Fluorene¹⁴ and rhodamine¹⁵⁻¹⁷ fluorophores have been used for lysosome marking and imaging. However, these dyes' absorption and emission maxima are below 600 nm which has been thought as disadvantages due to the high exciting energy and the putative autofluorescence of biomolecules. Therefore, fluorochromes whose absorption and emission maxima are in the red and near-infrared (NIR) region (wavelength range from 620 to 1000 nm), have been preferred as proper candidates for bioimaging. Among these dyes, cyanine dyes are ideal candidates owing to the high absorptivity, strong fluorescence, and acceptable photostability.¹⁸ The general cyanine dyes consist of two nitrogen centers, one of which is positively charged and is linked by a conjugated chain of an odd number of carbon atoms to the other nitrogen. ¹⁸ Due to the use of indoline unit as the nitrogen center, the dye is called as indocyanine (ICy). ICys are popular imaging agents¹⁹ or fluorescent probes²⁰ in biological applications. However, to the

best of our knowledge, ICy dyes have not been reported as lysosome tracker for dynamic monitoring.

The most commercial small-molecule trackers, such as neutral red and LysoTracker DND series, performed poorly in long-term tracking due to their high pH dependency.²¹ Other macro-molecule trackers including nanoparticles²² and quantum dots²³ can prolong the imaging period, unfortunately, with high cytotoxicity.²⁴⁻²⁵ Therefore, there is a need to develop a fluorescent ICy dye with low cytotoxicity for long-term lysosome tracking in live cells.

In this work, a novel amphiphilic squarylium indocyanine (SICy) dye (**LysoCy**, Fig. 1), bearing two bromines and two primary amines, was designed, characterized, and explored as a promising fluorescent marker with low cytotoxicity to track the long-term dynamics of lysosomes in live cells.

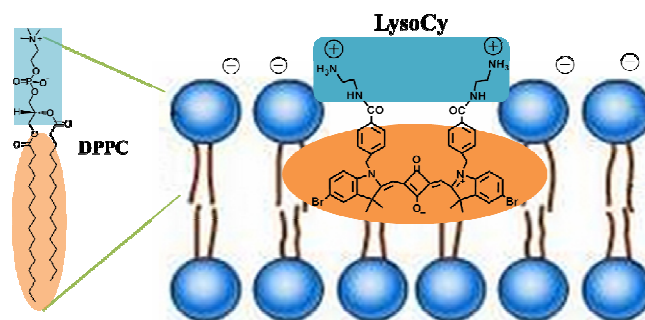
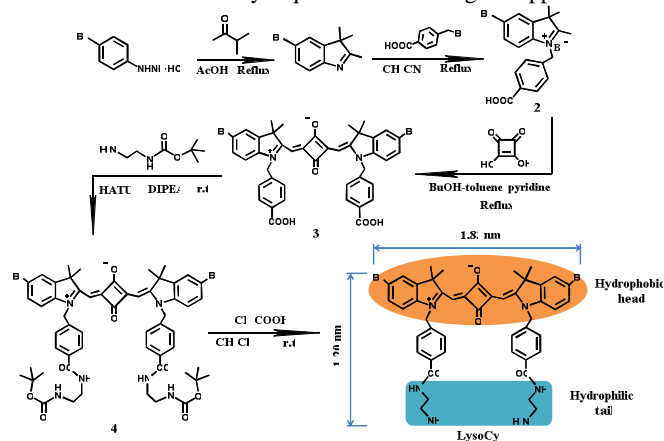


Fig. 1 Chemical structure of DPPC and the schematic diagram of **LysoCy** binding to the DPPC-mimicked lysosome membrane.

The SICy dye **LysoCy** was designed and synthesized in a straightforward way as shown in Scheme 1. 2,3,3-Trimethyl-5-bromo-3h-indole (**1**) was synthesized by Fisher ring reaction between 3-methyl-2-butanone and 4-bromophenylhydrazine hydrochloride in acetic acid. The further quaternary ammoniation with 4-bromomethylbenzoic acid yielded the quaternary salt **2**. The intermediate dye **3** was synthesized by the condensation of **2** and 3,4-dihydroxycyclobut-3-ene-1,2-dione (squaric acid) in toluene/*n*-butanol/pyridine (v/v/v = 1:1:1) solvent mixture. The rigid

squarylium ring in the conjugated center enhances the photostability of the dye.²⁶⁻²⁷ To quantitatively confirm the improved photostability of LysoCy, the time-dependent emission of aqueous LysoCy were measured under irradiation (Supporting Information, Fig. S1). SICy5 and ICy5 were used as controls. As a result, the presence of rigid squarylium ring notably enhances the photostability of the dye. The two carboxylic acid groups in compound **3** reacted with *tert*-butyl *N*-(2-aminoethyl) carbamate to afford **4**. The removal of *tert*-butoxycarbonyl groups in compound **4** gave rise to the desired final product **LysoCy** dye (Scheme 1). Details of the synthetic process and material characterizations are given in the Experimental and Supporting Information (Fig. S2-S12). **LysoCy** has two bromine groups and two primary amines, which contribute to the hydrophobic property and the water solubility, respectively. The water solubility of **LysoCy** is approximate 1 mg/mL at room temperature, which fulfils the water-solubility requirement of biological applications.



Scheme 1. Synthesis of amphiphilic **LysoCy** dye.

The fluorescence quantum yield (Φ_f) was measured by Edinburgh Instruments' FLS 980 spectrofluorimeter at room temperature. The Φ_f of **LysoCy** is $7 \pm 0.2\%$ in water. **LysoCy** is blue under natural light and is red under UV light (Fig. 2A). Fig. 2B shows that **LysoCy** has a maximum absorption at 635 nm and an emission peak at 643 nm. A dye with absorption and emission peaks above 600 nm is ideal for bioimaging due to the advantage of avoiding of cellular autofluorescence.²⁸ The concentration-dependent absorption and fluorescence spectra are shown in Fig. 2C and 2D. When the concentration of aqueous **LysoCy** was higher than 10 μM , the fluorescence peaks appeared a bathochromic shift and a sign of attenuation, while the absorption peaks gradually increased without any shift. The reason can be explained that the **LysoCy** molecules tend to form dimers above the concentration of 10 μM .

Lysosomes has a single-lipid bilayer membrane, the thickness of its hydrophilic part is less than 3.5 nm,²⁹ while **LysoCy** is 1.83 nm long and 1.2 nm wide upon energy minimization by using Gaussian Program (Scheme 1 and Fig. S13). The phospholipid bilayer provides the incorporation space for **LysoCy** molecule. Due to the acidic environment (around pH 5) in lysosome, the primary amine groups of **LysoCy** can be fully protonated as positive charges³⁰⁻³³ to electrostatically interact with the negatively charged membrane components of lysosomes. The mainly conjugated chromophore in **LysoCy** is hydrophobic that may interact with lipid tail of membranes. The bromine substitutions can increase the hydrophobic

performance of **LysoCy**. Therefore, we assume that **LysoCy** dye can bind to the lysosome membrane via both electrostatic and hydrophobic interactions (Fig. 1).

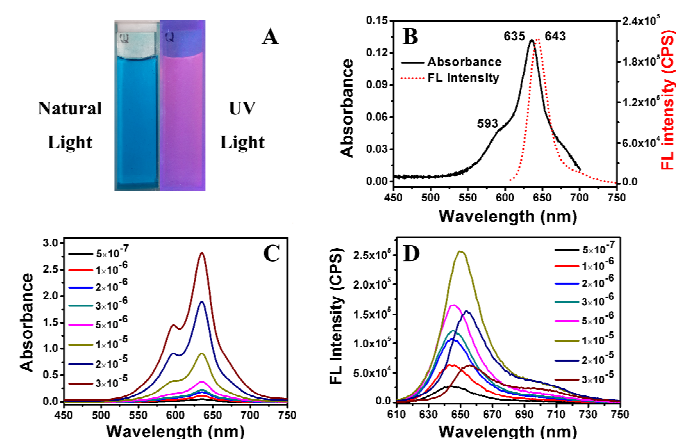


Fig. 2 (A) Colours of aqueous **LysoCy** under natural and UV light. (B) Normalized absorption and fluorescence spectra of **LysoCy** in water (2 μM). Concentration-dependent (C) absorbance and (D) fluorescence spectra of **LysoCy** in water (Excited at 593 nm, slit 5 nm).

To test above hypothesis, we further investigated the interaction between **LysoCy** and lysosome's membrane component. DPPC micelle was chosen as a membrane model for analysis.³⁴ DPPC, consisting of a plasma hydrophilic head and a hydrophobic tail, is one of the components of bio-membranes and has a critical micelle concentration (CMC) of 0.46 nM in buffer solution. Here, the DPPC lipid suspension was prepared under ultrasonic at a high concentration of 10 mM. The interaction between **LysoCy** and DPPC was first studied by spectral analysis. Aqueous **LysoCy** with concentrations of 2 μM were prepared as a blank. Both the absorption and fluorescence intensity of **LysoCy** decreased remarkably after the addition of DPPC solution (Fig. 3A and 3B). These optical changes can be attributed to the aggregation of DPPC micelles-bound **LysoCy**.

Isothermal titration calorimetry (ITC) was further carried out to quantify the interaction between **LysoCy** and DPPC.²⁶ Under neutral condition (pH 7), the measured affinity constant (K_A) of **LysoCy**/DPPC was $2.15 \times 10^5 \text{ M}^{-1}$. Both positive enthalpy change (ΔH) and entropy change (ΔS) values indicate that the interaction between **LysoCy** and DPPC was mainly driven by a hydrophobic effect (Fig. 4).³⁵⁻³⁶ Next, to further mimic the lysosome environment, a similar titration was performed under acidic environment (pH 5). The ΔH under weak acidic condition was below zero, which is the typical sign of an electrostatic interaction.³⁶ The K_A ($1.39 \times 10^6 \text{ M}^{-1}$) was greater compared with that value under neutral condition. These data suggest that the full protonation of primary amines in **LysoCy** under acidic environment can enhance the electrostatic interaction between **LysoCy** and DPPC. Above interactions provide a good potential for **LysoCy** to specifically label the lysosome during live cell imaging.

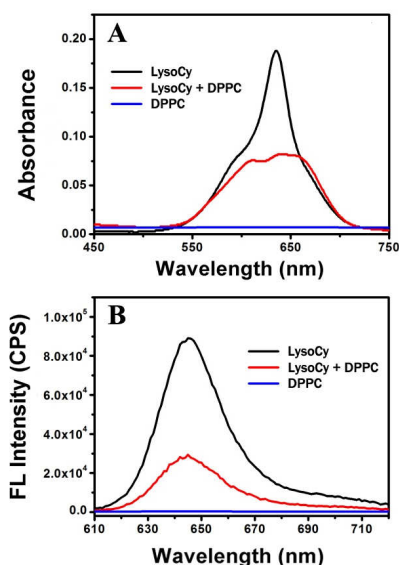


Fig. 3 Absorption (A) and fluorescence spectra (B) of **LysoCy**/DPPC in water. Both of **LysoCy** and DPPC were used at 0.1 mM.

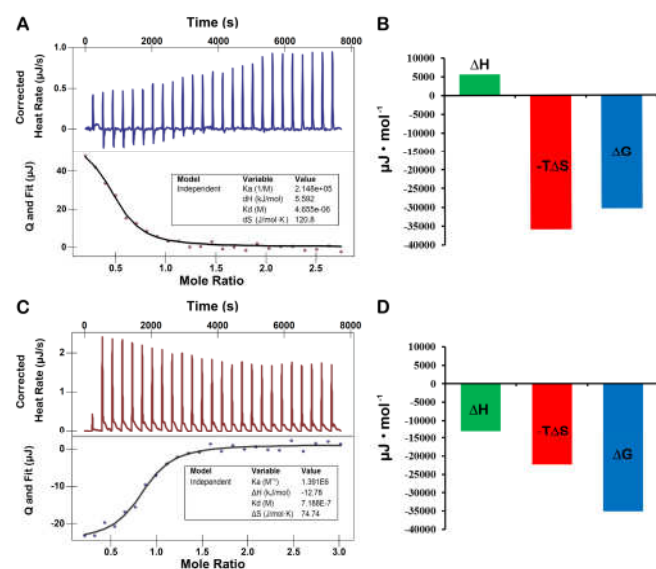


Fig. 4 ITC titration of **LysoCy** and DPPC interaction at pH 7 (A) and pH 5 (C). Thermodynamic parameter of **LysoCy** and DPPC interaction at pH 7 (B) and pH 5 (D).

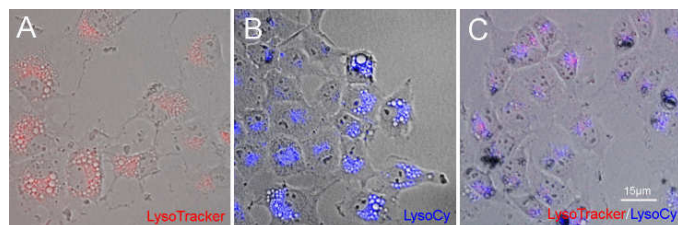


Fig. 5 Fluorescent images of live cells stained with LysoTracker alone (A, red: pseudocolour detected at emission wavelength of 570), **LysoCy** alone (B, blue: pseudocolour detected at emission wavelength of 664) and both of them (C, purple: the overlay of red and blue).

To assay the labelling effects of **LysoCy** on live cells, we chose a commercial acidotropic probe LysoTracker, which efficiently labelled the lysosomes, as a control (Fig. 5A). When live cells were incubated with **LysoCy** at extremely low concentration of 0.1 mM, the lysosomes were clearly labelled (Fig. 5B). To confirm the labelling specificity, a double labelling with both commercial LysoTracker and **LysoCy** was performed and showed well overlapped pattern (Fig. 5C), suggesting well specificity of **LysoCy** for lysosome labelling. Obviously, the fluorescence intensity of **LysoCy** is stronger than that of LysoTracker, indicating that **LysoCy** is more effective than LysoTracker to detect and track lysosomes. Based on above ITC analysis under neutral and acidic conditions, the two weak bases of **LysoCy** are only partially protonated at neutral pH, which makes the dye permeable to cellular membrane. Subsequently the two weak bases of **LysoCy** are further protonated at acidic pH, which leads to the retention and accumulation of **LysoCy** on lysosome membrane. Note that, both commercial LysoTracker and our **LysoCy** can not distinguish the endosome, early or late lysosome.

Long-term tracking of lysosome dynamics was also studied. We performed a real-time monitoring up to two days by incubation of cells with **LysoCy**. The fluorescence of **LysoCy** in lysosomes gradually increased along with the incubation time (Fig. 6A-D'), revealing ideal lysosome dynamics. In contrast, the fluorescence of LysoTracker decreased along with the incubation time (Fig. 6E-H'). The quantified fluorescence intensities of **LysoCy** and LysoTracker labelled cells at a series of time points are shown in Fig. 7. Consistently, the **LysoCy** fluorescence intensities gradually increased along with the continuous cell uptake of the dye, while LysoTracker labelling exhibited declined intensities. Obviously, **LysoCy** has a great potential to be applied as a long-term lysosome tracker with stable labelling and easy manipulation.

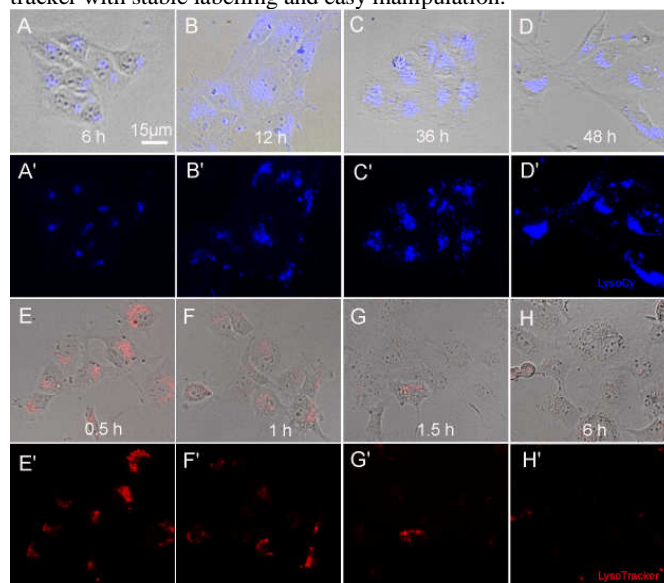


Fig. 6 Lysosome dynamics labelled with **LysoCy** (A-D', blue: pseudocolour detected at emission wavelength of 664) and LysoTracker (E-H', red: pseudocolour detected at emission wavelength of 570).

As good biocompatibility is a significant requirement for the live imaging.³⁷⁻³⁸ We assayed the cytotoxicity of **LysoCy**, and compared with the commercial LysoTracker. **LysoCy** shows low cytotoxicity comparable to the control (Fig. S14). The cell viability of **LysoCy**

was above 80% at all the tested concentration, suggesting the good biocompatibility of LysoCy.

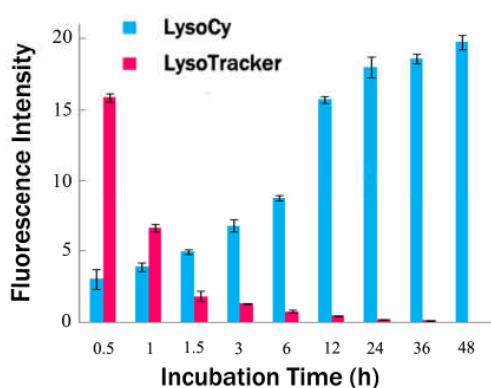


Fig. 7 Quantified fluorescence intensities of LysoCy and LysoTracker at series of time points.

In summary, a novel amphiphilic lysosome tracker **LysoCy** bearing two bromines and primary amines has been synthesized. The dye has absorption and emission in red region, thus avoiding the autofluorescence in bioimaging. The amine groups ensured water solubility and provided an electrostatic binding to the lysosome membrane due to the protonation in acidic aqueous environment. On the other hand, the indole part can insert and interact to the membrane of lysosomes due to a hydrophobic driven effect. In the live cell labelling, **LysoCy** showed remarkable lysosomes tracking ability with low cytotoxicity. The long-term dynamics of lysosomes can be revealed by **LysoCy** up to 2 day's culture. Therefore, a novel classic of SiCy dye was explored as a long-term lysosome tracker for real time monitoring of the cellular dynamics.

This work was financially supported by the 973 program (2013CB127603), the National Science Foundation of China (21174012, 51221002 and 21574009), the Beijing Natural Science Foundation (2142026), Innovation and Promotion Project of Beijing University of Chemical Technology, and the Chinese Universities Scientific Fund (2015NX001 and 2015QC094).

Notes and references

^a State Key Laboratory of Chemical Resource Engineering, Beijing Laboratory of Biomedical Materials, Beijing University of Chemical Technology, 100029 Beijing, China, Email: yinmz@mail.buct.edu.cn.

^b China Agricultural University, 100193 Beijing, China, Email: shenjie@cau.edu.cn

† Electronic Supplementary Information (ESI) available: Synthesis procedures and material characterizations. See DOI:10.1039/b000000x/

- (1) T. Soldati and M. Schliwa, *Nat. Rev. Mol. Cell Biol.*, 2006, **7**, 897-908.
- (2) L. Yuan, L. Wang, B. K. Agrawalla, S.-J. Park, H. Zhu, B. Sivaraman, J. Peng, Q.-H. Xu and Y.-T. Chang, *J. Am. Chem. Soc.*, 2015, **137**, 5930-5938.
- (3) M. Yin, J. Shen, G. O. Pflugfelder and K. Müllen, *J. Am. Chem. Soc.*, 2008, **130**, 7806-7807.
- (4) Y. Liu, K. Li, M.-Y. Wu, Y.-H. Liu, Y.-M. Xie and X.-Q. Yu, *Chem. Commun.*, 2015, **51**, 10236-10239.
- (5) Z. Xu, B. He, W. Wei, K. Lu, M. Yin, W. Yang and J. Shen, *J. Mater. Chem. B*, 2014, **2**, 3079-3086.

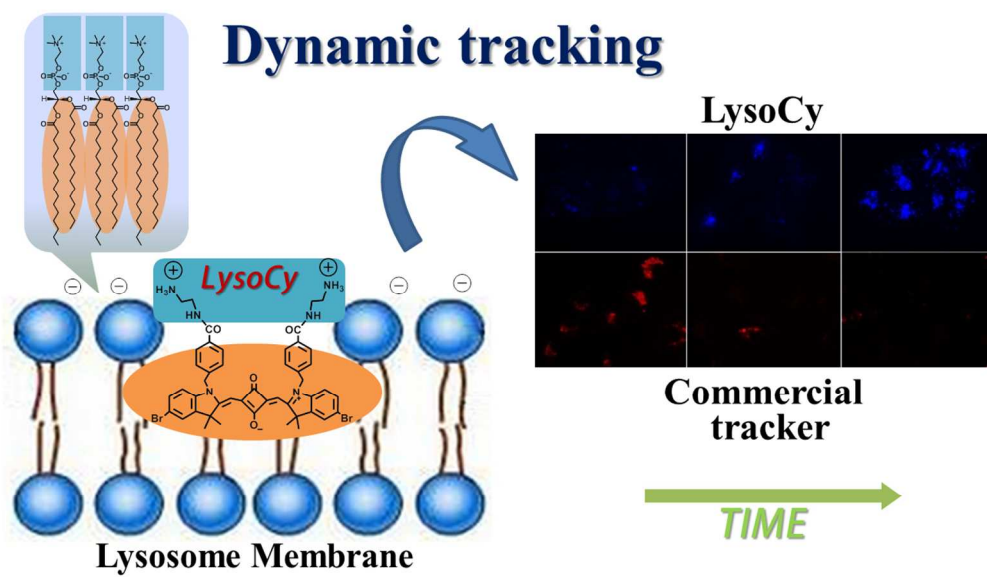
- (6) X. Zhang, K. Wang, M. Liu, X. Zhang, L. Tao, Y. Chen and Y. Wei, *Nanoscale*, 2015, **7**, 11486-11508.
- (7) X. Zhang, X. Zhang, B. Yang, and Y. Wei, *Chinese J. Polym. Sci.*, 2014, **32**, 871-879.
- (8) S. Wan, Y. Zheng, J. Shen, W. Yang and M. Yin, *ACS Appl. Mater. Inter.*, 2014, **6**, 19515-19519.
- (9) C. Settembre, A. Fraldi, D. L. Medina and A. Ballabio, *Nat. Rev. Mol. Cell Biol.*, 2013, **14**, 283-296.
- (10) L. He, C. P. Tan, R. R. Ye, Y. Z. Zhao, Y. H. Liu, Q. Zhao, L. N. Ji and Z. W. Mao, *Angew. Chem.*, 2014, **126**, 12333-12337.
- (11) L. Chen, J. Wu, C. Schmuck and H. Tian, *Chem. Commun.*, 2014, **50**, 6443-6446.
- (12) D.-D. He, W. Liu, R. Sun, C. Fan, Y.-J. Xu and J.-F. Ge, *Anal. Chem.*, 2015, **87**, 1499-1502.
- (13) M. Gao, Q. Hu, G. Feng, B. Z. Tang and B. Liu, *J. Mater. Chem. B*, 2014, **2**, 3438-3442.
- (14) X. Wang, D. M. Nguyen, C. O. Yanez, L. Rodriguez, H.-Y. Ahn, M. V. Bondar and K. D. Belfield, *J. Am. Chem. Soc.*, 2010, **132**, 12237-12239.
- (15) Z. Li, S. Wu, J. Han and S. Han, *Analyst*, 2011, **136**, 3698-3706.
- (16) W. Zhu, X. Chai, B. Wang, Y. Zou, T. Wang, Q. Meng and Q. Wu, *Chem. Commun.*, 2015, **51**, 9608-9611.
- (17) N. B. Yapici, Y. Bi, P. Li, X. Chen, X. Yan, S. R. Mandalapu, M. Faucett, S. Jockusch, J. Ju and K. M. Gibson, *Sci. Rep.*, 2015, **5**, Article number: 8576.
- (18) A. Mishra, R. K. Behera, P. K. Behera, B. K. Mishra and G. B. Behera, *Chem. Rev.*, 2000, **100**, 1973-2012.
- (19) M. E. Lobatto, C. Calcagno, A. Millon, M. L. Senders, F. Fay, P. M. Robson, S. Ramachandran, T. Binderup, M. P. Paridaans and S. Sensarn, *ACS Nano*, 2015, **9**, 1837-1847.
- (20) S.-Y. Lim, K.-H. Hong, D. I. Kim, H. Kwon and H.-J. Kim, *J. Am. Chem. Soc.*, 2014, **136**, 7018-7025.
- (21) R. P. Haugland, *A Guide to Fluorescent Probes and Labeling Technologies*, 10th ed Molecular Probes, Eugene, OR, 2005.
- (22) H. Shi, X. He, Y. Yuan, K. Wang and D. Liu, *Anal. Chem.*, 2010, **82**, 2213-2220.
- (23) L. W. Zhang and N. A. Monteiro-Riviere, *Toxicol. Sci.*, 2009, **110**, 138-155.
- (24) L. R. Farcial, C. Ubaldi, D. Mehn, G. Giudetti, P. Nativo, J. Ponti, D. Gilliland, F. Rossi and A. Bal-Price, *Nanotoxicology*, 2012, **7**, 1095-1110.
- (25) Y.-P. Ho and K. W. Leong, *Nanoscale*, 2010, **2**, 60-68.
- (26) J. Li, K. Guo, J. Shen, W. Yang and M. Yin, *Small*, 2014, **10**, 1351-1360.
- (27) Y. He, W. Zhou, F. Wu, M. Li, E. Wang, *J. Photoch. Photobio. A*, 2004, **162**, 463-471.
- (28) J.-C. Yu, Y.-L. Chen, Y.-Q. Zhang, X.-K. Yao, C.-G. Qian, J. Huang, S. Zhu, X.-Q. Jiang, Q.-D. Shen and Z. Gu, *Chem. Commun.*, 2014, **50**, 4699-4702.
- (29) J. F. Nagle and S. Tristram-Nagle, *Biochim. Biophys. Acta* 2000, **1469**, 159-195.
- (30) H. Hall Jr, *J. Am. Chem. Soc.*, 1957, **79**, 5441-5444.
- (31) J. Li, C. Ji, W. Yang and M. Yin, *Analyst*, 2013, **138**, 7289-7293.
- (32) X. Hu, H. Wang, J. Yang, W. Liu, W. Wang, *J. Appl. Polym. Sci.*, 2014, **131**, 40468.
- (33) Y. Li, J. Yang, B. Xu, F. Gao, W. Wang and W. Liu, *ACS Appl. Mater. Inter.*, 2015, **7**, 8114-8124.
- (34) M. Zong and Y. Gong, *Chinese J. Polym. Sci.*, 2011, **29**, 53-64.
- (35) P. D. Ross and S. Subramanian, *Biochemistry*, 1981, **20**, 3096-3102.

Journal Name

(36) P. Sen, S. Fatima, B. Ahmad and R. H. Khan, *Spectrochim Acta A*, 2009, **74**, 94-99.

(37) H. Wang, D. Sun, N. Zhao, X. Yang, Y. Shi, J. Li, Z. Su, G. Wei, *J. Mater. Chem. B*, 2014, **2**, 1362-1370.

(38) K. Liu, Z. Xu, M. Yin, W. Yang, B. He, W. Wei and J. Shen, *J. Mater. Chem. B*, 2014, **2**, 2093-2096.



A novel amphiphilic squarylium indocyanine (LysoCy) is reported for remarkable lysosome tracking in living cells.

220x134mm (150 x 150 DPI)