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A direct arylation approach towards efficient small molecule organic solar cells

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Three extended molecular chromophores, differing in their central acceptor moiety and specifically designed as electron donor components for small molecule organic solar cells, are synthesized via a two-fold C-H arylation protocol. Upon removal of the side products inherent to the applied direct (hetero)arylation procedure, a record power conversion efficiency of 5.1% is achieved.

Solution-processed bulk heterojunction (BHJ) organic solar cells based on light harvesting and charge transporting conjugated polymers or analogous small molecules have attracted large attention from both academia and industry because of their appealing aesthetical (colour, transparency), mechanical (flexibility, light-weight) and cost features (large area production via simple printing processes).¹ Small molecule organic solar cells (SMOSC's) have recently surpassed the 10% power conversion efficiency (PCE) barrier, thereby becoming a viable substitute for their polymeric counterparts.² In comparison to semiconducting polymers, 'small' molecules display some peculiar beneficial features such as less batch-to-batch variability due to their well-defined structure, more reproducible synthetic protocols, easier purification and a more straightforward understanding of structure-performance relationships.³ Additionally, small molecules more readily allow tuning of the optoelectronic (and other) material properties and they can exhibit fairly high charge carrier mobilities. State of the art polymer and small molecule electron donor materials generally apply the alternating donor-acceptor (D-A) (or 'push-pull') strategy, allowing for a decrease of the HOMO-

LUMO gap and a better overlap with the solar spectrum.¹⁻³ More advanced molecular scaffolds with D-A-D, A-D-A, D-A-A and even more extended architectures are acquiring increased attention to optimize photon collection and charge carrier transport.³ Thiazolo[5,4-*d*]thiazole (TzTz) has emerged as an appealing electron withdrawing building block, showing particular advantages such as easy synthetic access, a planar and rigid structure with strong π - π stacking tendency (promoting mobility) and oxidative stability.⁴ Due to these favourable features, TzTz-based polymer⁵ and small molecule⁶ materials have successfully been applied in organic photovoltaics, with record PCE's of 7.5 and 4.1%, respectively.^{5b,6d}

Low bandgap organic semiconductors are generally synthesized through standard transition metal catalysed cross-coupling reactions to afford (het)aryl-(het)aryl connections, such as Suzuki and Stille reactions, requiring the preparation of organoboron or organotin precursors. Direct (hetero)arylation of activated C-H moieties has emerged as a viable alternative, with specific benefits in terms of (atom) efficiency, sustainability and functional group compatibility, and it has been applied for SMOSC materials with reasonable success.⁷ In previous work, we have demonstrated a facile direct arylation strategy towards diaryl-substituted 2,5-dithienyl-TzTz small molecules with varying electron affinities, particularly emphasizing on monosubstitution towards asymmetric push-pull structures.^{7d} This first generation TzTz chromophores showed a relatively narrow absorption window and SMOSC efficiencies remained below 3%. Prompted by these initial results, more complex structures were pursued, still avoiding traditional cross-coupling protocols. In the present contribution, we report on the combination of an asymmetric triphenylamine-monosubstituted 2,5-dithienyl-TzTz precursor (**TPA-TzTz-T**) with three different brominated central acceptor derivatives – bithiazole (BiTz), quinoxaline (Qx) and isoindigo (II) (Scheme 1), varying the electron affinity, solubility and stacking features – by direct (hetero)arylation. Similar C-H arylation conditions as previously optimized to construct the monosubstituted **TPA-TzTz-T** starting material were

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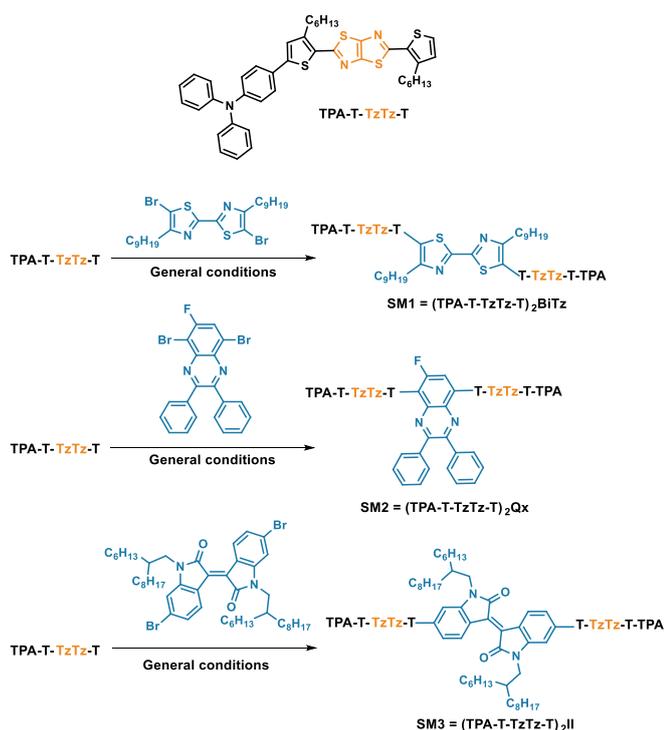
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Electronic Supplementary Information (ESI) available: General experimental methods, MALDI-TOF MS spectra, NMR spectra, UV-Vis spectra (solution), cyclic voltammograms, solar cell optimization tables, EQE spectra, CPE structure, FTPS data and mobility data. See DOI: 10.1039/x0xx00000x

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employed (Scheme 1),^{7d} affording the target small molecules **SM1–3** in yields of ~30%.



Scheme 1. Combination of the **TPA-T-TzTz-T** precursor with three different acceptor cores via direct heteroarylation.

The modest reaction yields can be attributed to the formation of multiple side products. By analysing the filtrate after final precipitation of **SM2** by MALDI-TOF mass spectrometry, homocoupling of the central acceptor unit and branching due to the presence of alternative reactive C-H groups were identified as the main side reactions (see Fig. S3). Whereas homocoupling appears in traditional cross-couplings as well and has been shown to have a strong effect on photovoltaic efficiency,⁸ the observed regioselectivity issues are inherent to C-H arylation procedures. The presence of these side products, also in the case of direct arylation polymerizations,⁹ has often been undervalued, especially with respect to their effect on device performance. Small molecules **SM1–3** were carefully purified by standard column chromatography. **SM1** and **SM3** were additionally purified by recycling preparative size exclusion chromatography (prep-SEC). Due to its limited solubility in chloroform, **SM2** was instead precipitated twice in ethyl acetate. Compared to polymers, molecular chromophores are much more versatile probes to evaluate the ‘misconnections’ in direct arylations. In this respect, MALDI is particularly useful, as impurities hardly observable and not identifiable by NMR clearly showed up in the mass spectra of **SM1–3**.

UV-Vis absorption spectra in solution and thin film were gathered for all three compounds to analyse their light-harvesting capacity (Fig. 1 and S10, Table 1). **SM3** displayed the broadest absorption spectra, covering almost the entire

visible range, and the smallest optical HOMO-LUMO gap. The frontier energy levels were estimated via cyclic voltammetry (CV) from the onset of the oxidation and reduction peaks (Fig. S11, Table 1). The HOMO levels are nearly invariant (within 0.09 eV). Since the LUMO's are generally localized on the acceptor parts, a decreasing trend is expected when increasing the electron withdrawing nature of the acceptor building block, which is indeed illustrated in the present series. The narrowest electrochemical HOMO-LUMO gap (1.45 eV) was obtained for the small molecule bearing the strongest acceptor unit (II, **SM3**), in accordance with the optical data.

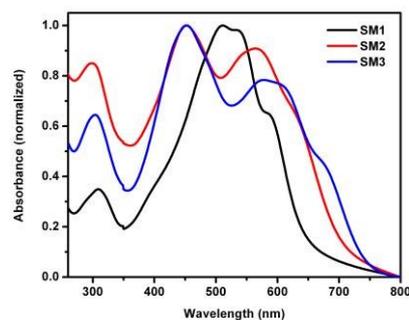


Figure 1. Normalized UV-Vis absorption spectra for the small molecules (in film).

Table 1. Optical and electrochemical properties for the small molecule series **SM1–3**.

	λ_{max}^a (nm) (log ϵ) sol.	λ_{max} (nm) film	E_g^{opt} (eV)	E_{ox} (V) ^c	HOMO (eV)	E_{red} (V) ^c	LUMO (eV)
SM1	492 (4.70)	507	2.17	0.47	-5.33	-1.59	-3.26
SM2	437 (4.78) 527 (4.79)	450 564	1.97	0.43	-5.29	-1.54	-3.32
SM3	449 (5.02) 598 (4.99)	452 578	1.74	0.52	-5.38	-0.93	-3.93

^a In chloroform. ^b Optical HOMO-LUMO gaps, determined from the onset of the UV-Vis spectra in chloroform. ^c Oxidation/reduction onsets.

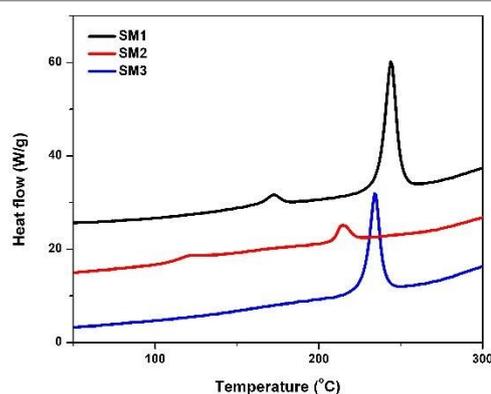


Figure 2. RHC profiles of **SM1–3** (second heating curves, curves shifted vertically for clarity).

The thermal properties of the three small molecules were investigated by rapid heat-cool calorimetry (RHC) (Fig. 2).¹⁰ The obtained results demonstrate a strong impact on the glass transition temperature (T_g), melting temperature (T_m) and melting enthalpy (ΔH_m) upon altering the central acceptor unit. **SM1** displays two melting temperatures (172/244.5 °C),

as well as the highest ΔH_m (30.9 J g^{-1}), indicating a highly crystalline character. **SM2** appears to be significantly less crystalline ($\Delta H_m = 4.1 \text{ J g}^{-1}$, $T_m = 214.1 \text{ }^\circ\text{C}$) and a pronounced T_g ($112.6 \text{ }^\circ\text{C}$) appears. **SM3** falls in between, with a T_m of $234.3 \text{ }^\circ\text{C}$ and a ΔH_m of 21.5 J g^{-1} .

Thin film BHJ OSC's with the traditional architecture glass/ITO/PEDOT:PSS/**SM1–3**:PC₇₁BM/Ca/Al were fabricated to evaluate the photovoltaic features of the synthesized small molecules. Optimal devices were procured from small molecule solutions in chloroform, in combination with PC₇₁BM as the electron acceptor in a 1:1 (wt/wt%) ratio (Table 2, Fig. 3, Table S1–3). Solar cells produced with **SM2** demonstrated highest photovoltaic performance, yielding an open-circuit voltage (V_{oc}) of 0.84 V in combination with a short-circuit current density (J_{sc}) of 9.29 mA cm^{-2} and a fill factor (FF) of 0.55, resulting in an average PCE of 4.25% (and a best device of 4.86%). The performance of **SM1** and **SM3** was somewhat worse, with average PCE's of 1.77 and 2.11%, respectively. Upon addition of 1-chloronaphthalene (CN) as a processing additive, the average PCE of the **SM1** device could be enhanced up to 2.50%, mainly attributable to an increased J_{sc} and FF (Table 2). Unfortunately, a similar approach did not result in further enhancements for **SM2** and **SM3**. The external quantum efficiency (EQE) spectra reflect the difference in performance for the three small molecule:PC₇₁BM OSC's (Fig. S12) and follow the trend observed in the absorption spectra. The **SM2**:PC₇₁BM device shows a maximum EQE of $\sim 55\%$. Noteworthy lower PCE's were obtained when the side products mentioned above were not carefully removed.

The incorporation of interfacial layers, in particular conjugated polyelectrolytes (CPE's), has been documented as a means to boost the inherent I - V properties of BHJ OSC's.¹¹ In previous work, we demonstrated the successful application of a specific ionic copolythiophene (see Fig. S13) as cathodic interlayer.¹² Application of this strategy to the **SM2** device afforded a further performance improvement, with an average PCE of 4.91% (best 5.09%) (Fig. 3, Table 2). The maximum EQE increased up to $\sim 70\%$ (Fig. S12).

Table 2. Photovoltaic output parameters for the (optimized) **SM1–3**:PC₇₁BM SMOSC's.^a

	Additive ^b	V_{oc} (V)	J_{sc} (mA cm^{-2})	FF	Average PCE (%) ^c	Best PCE (%)
SM1	/	0.75	5.80	0.41	1.77	1.96
SM1	CN (1%)	0.76	6.14	0.54	2.50	2.78
SM2	/	0.84	9.29	0.55	4.25	4.86
SM2	CN (1%)	0.86	7.81	0.40	2.68	2.84
SM2^d	/	0.84	10.18	0.57	4.91	5.09
SM3	/	0.86	7.04	0.35	2.11	2.42
SM3	CN (1%)	0.85	6.41	0.32	1.73	1.83

^a Device structure: glass/ITO/PEDOT:PSS/**SM1–3**:PC₇₁BM(1:1)/Ca/Al. All active layers were spin-coated from CHCl₃. ^b CN = 1-chloronaphthalene. ^c Averages taken over 4–8 devices. ^d CPE cathodic interlayer (replacing Ca).

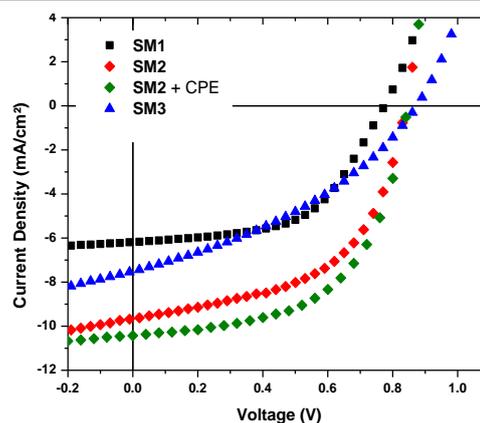


Figure 3. J - V curves under illumination for the (average-performance) solar cell devices based on **SM1–3** (processed from optimal solvents).

Mixing of an organic semiconductor with a fullerene gives rise to interfacial charge transfer (CT) states, determining the V_{oc} of OSC's.¹³ The absorption of the CT state can be directly probed by Fourier transform photocurrent spectroscopy (FTPS). FTPS experiments were done for the **SM1–3**:PC₇₁BM solar cells (see Fig. S14) and the observed E_{CT} values nicely follow the V_{oc} trend for the prepared devices. Atomic force microscopy (AFM) measurements were performed to investigate the active layer film topographies and to portray the influence of the processing additive. As illustrated in Fig. 4, upon addition of CN to the **SM1**:PC₇₁BM blend solution, a significant alteration of the morphology appears, with formation of specific needle-shaped structures, apparently responsible for an increase in J_{sc} and FF. The addition of the additive to the **SM2** solution resulted in a noticeable decrease in efficiency (the average PCE dropped from 4.25 to 2.68%), possibly induced by a non-ideal mixing of the donor and acceptor components. The (strong) reduction in FF and J_{sc} suggests a decreased charge transport capacity. For **SM3**, CN did not have a large impact on either the layer topography or solar cell performance.

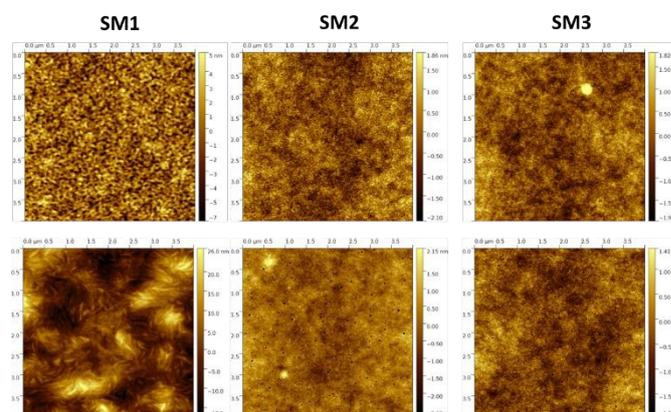


Figure 4. AFM (topography) images ($4 \times 4 \mu\text{m}$) of the photoactive layers of the **SM1–3**:PC₇₁BM solar cells prepared without (upper row) and with (bottom row) CN (1%) as a processing additive.

Finally, field-effect transistors were prepared from the pure chromophores to acquire the mobility data of the small

molecule series. The extracted charge carrier mobilities were found to be $\sim 2 \times 10^{-4}$, $\sim 1 \times 10^{-4}$ and $\sim 6 \times 10^{-5}$ cm² V⁻¹ s⁻¹ (transfer and output curves in Fig. S15) for **SM1**, **SM2** and **SM3**, respectively. The lower charge carrier mobility might partly explain the modest performance for the SMOSC based on **SM3**, even though the absorption features are most favourable. For the **SM2** and **SM3** devices, a similar trend was observed by photo-CELIV (photo-induced charge extraction by linearly increasing voltage; Fig. S16), probing mobility in the appropriate direction of the solar cell mode.

In conclusion, three π -expanded D'A'DADA'D'-type molecular chromophores were synthesized via a two-fold C-H arylation approach and implemented as electron donors in organic solar cells. Their physicochemical and optoelectronic material characteristics were investigated and a PCE of 5.1% was achieved for the **TPA-T-TzTz-T-Qx-T-TzTz-T-TPA** small molecule, a record performance for molecular chromophores prepared via direct arylation.^{7,9} Despite the success of the approach, the current limitations of the C-H (hetero)arylation protocols should not be ignored. If one desires to employ sustainable direct arylation procedures to reduce the 'synthetic complexity'¹⁴ of advanced organic semiconductors, further efforts should be devoted to the optimization of the C-H regioselectivity in order to increase reaction yields and simplify compound purification. In this respect, related recent efforts (e.g. catalyst optimization) in the field of direct arylation polymerizations might provide a first entry.⁹

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TOC entry

Direct arylation affords organic semiconductors enabling high solar cell efficiency upon careful removal of the ‘miss-coupled’ side products.

