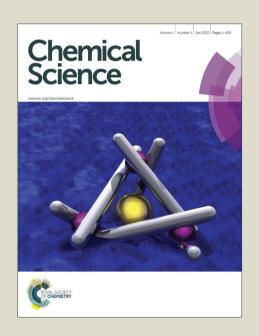
# Chemical Science

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.





### **Chemical Science**

### Edge Article

## A structural view of synthetic cofactor integration into [FeFe]-hydrogenases

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

J. Esselborn<sup>a</sup>, N. Muraki<sup>b†</sup>, K. Klein<sup>c</sup>, V. Engelbrecht<sup>a</sup>, N. Metzler-Nolte<sup>c</sup>, U.-P. Apfel<sup>c\*</sup>, E. Hofmann<sup>d</sup>, G. Kurisu<sup>b\*</sup>, T. Happe<sup>a\*</sup>

[FeFe]-hydrogenases are nature's fastest catalysts for the evolution or oxidation of hydrogen. Numerous synthetic model complexes for the [2Fe] subcluster (2Fe<sub>H</sub>) of their active site are known, but so far none of these could compete with the enzymes. The complex Fe<sub>2</sub>[μ-(SCH<sub>2</sub>)<sub>2</sub>X](CN)<sub>2</sub>(CO)<sub>4</sub><sup>2-</sup> with X=NH was shown to integrate into the apo-form of [FeFe]-hydrogenases to yield a fully active enzyme. Here we report the first crystal structures of the apo-form of the bacteria. [FeFe]-hydrogenase CpI from Clostridium pasteurianum at 1.60 Å and the active semisynthetic enzyme, CpI<sup>ADT</sup>, at 1.63 Å. The structures illustrate the significant changes in ligand coordination upon integration and activation of the [2Fe] complex. These changes are induced by a rigid 2Fe<sub>H</sub> cavity as revealed by the structure of apoCpI, which is remarkably similar to CpI<sup>ADT</sup>. Additionally we present the high resolution crystal structures of the semisynthetic bacterial [FeFe]-hydrogenases CpI<sup>PDT</sup> (X=CH<sub>2</sub>), CpI<sup>ODT</sup> (X=O) and CpI<sup>SDT</sup> (X=S) with changes in the headgroup of the dithiolate bridge in the 2Fe<sub>H</sub> cofactor. The structures of these inactive enzymes demonstrate that the 2Fe<sub>H</sub>-subcluster and its protein environment remain largely unchanged when compared to the active enzyme CpI<sup>ADT</sup>. As the active site shows an open coordination site in all structures, the absence of catalytic activity is probably not caused by steric obstruction. This demonstrates that the chemical properties of the dithiolate bridge are essential for enzyme activity.

### Introduction

[FeFe]-hydrogenases are efficient natural catalysts for both the generation and oxidation of  $\rm H_2^{-1}$ . This reaction is accomplished by the H-cluster, a metal cofactor consisting of a cubane [4Fe4S] cluster (4Fe<sub>H</sub>) connected via a cysteine to an unusual [2Fe] cluster (2Fe<sub>H</sub>). The two iron atoms of the latter, termed distal (Fe<sub>d</sub>) and proximal (Fe<sub>p</sub>) iron are ligated by a total of three CO and two CN- molecules and an aza -dithiolato bridge<sup>2–5</sup>. Four structural characteristics seem to be important for the high activity of the H-cluster<sup>6,7</sup>: a) The cyanide ligands besides having additional effects<sup>8–10</sup> shift the redox potential to more negative values, when compared to all-carbonyl complexes<sup>11</sup>. b) The 4Fe<sub>H</sub>-cluster serves as an intramolecular redox partner<sup>12–15</sup>. c) A

proton donor is present in the dithiolato bridge<sup>16–19</sup>. d) The ligand conformation at the 2Fe<sub>H</sub> subsite features a CO that bridges or semi-bridges the Fe atoms. This leads to an unoccupied coordination site on Fe<sub>d</sub> <sup>3,5,20–22</sup>. For the hydrogenase HydA1 from *Chlamydomonas reinhardtii* three redox states are discussed as part of the catalytic cycle, which can be distinguished by EPR and FTIR spectroscopy. According to this hypothesis the H-cluster cycles from the H<sub>ox</sub> state 4Fe<sub>H</sub><sup>2+</sup>-Fe(I)-Fe(II), via the H<sub>red</sub> state 4Fe<sub>H</sub><sup>2+</sup>-Fe(I)-Fe(I), to the H<sub>sred</sub> state 4Fe<sub>H</sub><sup>4+</sup>-Fe(I)-Fe(I). The redox potentials for these transitions are -400 mV and-470 mV vs. SHE close to the H<sub>2</sub>/H+ redox pair<sup>7</sup>. The crucial proton transfer to and from the active site seems to be accomplished by a proton transfer pathway through the protein towards the central atom of the dithiolato bridge in the 2Fe<sub>H</sub>-subcluster<sup>3,5,23</sup>.

In nature, the 4Fe<sub>H</sub>-cluster and other FeS clusters of the enzyme not specific to [FeFe]-hydrogenases are synthesized by the widespread ISC or SUF systems for FeS cluster synthesis yielding inactive hydrogenases, which lack only the specific 2Fe<sub>H</sub>-subcluster<sup>24</sup>. For the sake of simplicity this pre-form will be referred to as apo-form of [FeFe]-hydrogenases throughout this text. The three maturase enzymes HydE, HydF and HydG are necessary for the synthesis of the 2Fe<sub>H</sub>-cluster and the assembly of the H-cluster within the protein.<sup>25</sup> *In vitro*, chemically synthesized [2Fe] complexes can be bound to the maturase HydF and transferred from there to apo-hydrogenases to form a complete H-cluster.<sup>2</sup> Notably, also in the absence of HydF or

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

<sup>&</sup>lt;sup>a</sup> AG Photobiotechnologie, Fakultät für Biologie und Biotechnologie, Ruhr-Universität Bochum, Universitätsstraße 150, 44801 Bochum, Germany. Email: thomas.happe@rub.de.

b. Laboratory of Protein Crystallography, Institute for Protein Research, Osaka University, Suita, Osaka 565-0871, Japan. Email: gkurisu@protein.osaka-u.ac.

<sup>&</sup>lt;sup>c</sup> Lehrstuhl für Anorganische Chemie I—Bioanorganische Chemie, Fakultät für Chemie und Biochemie, Ruhr-Universität Bochum, Universitätsstraße 150, 44801 Bochum, Germany. Email: ulf.apfel@rub.de.

<sup>&</sup>lt;sup>d</sup>-AG Proteinkristallographie, Fakultät für Biologie und Biotechnologie, Ruhr-Universität Bochum, Universitätsstraße 150, 44801 Bochum, Germany.

<sup>†</sup> Present Addresses: Department of Life and Coordination-Complex Molecular Science, Institute for Molecular Science, National Institutes of Natural Sciences, Okazaki 444-8787, Japan.

any other helper protein, an active H-cluster can be formed spontaneously by bringing together the inactive apohydrogenase and the chemically synthesized [2Fe] complex  $\text{Fe}_2[\mu\text{-}(\text{SCH}_2)_2\text{NH}](\text{CN})_2(\text{CO})_4^{2\text{-}}.^{26}$  While the [2Fe] moiety alone is inactive under physiological conditions, the semisynthetic enzyme shows high catalytic activity, which demonstrates the importance of the protein environment. [2Fe] complexes with variations in the dithiolato bridge and/or the other Fe ligands have recently been shown to integrate into HydA1 as well, but the enzymes were inactive or severely limited in their turnover rates especially if the dithiolato bridge was changed.  $^{2,27}$  The central atom of the dithiolato-moiety seems to influence the redox behavior of the H-cluster either directly or by interfering with the proton transfer to/from the active site.  $^{28}$ 

Structures of the active bacterial [FeFe]-hydrogenases<sup>3,29,30</sup> and the inactive apo-form of HydA1 from *Chlamydomonas reinhardtii*<sup>31</sup> have been solved at high resolution. In this study we aim to expand the knowledge about maturation of [FeFe]-hydrogenases by reporting the crystal structure of the [FeFe]-hydrogenase from *Clostridium pasteurianum* (CpI) in its apoform without the 2Fe<sub>H</sub>-cluster (apoCpI). In addition, we contribute to a deeper understanding of [FeFe]-hydrogenase function by solving the structures of four semisynthetic hydrogenases maturated in vitro with [2Fe] complexes of the kind Fe<sub>2</sub>[ $\mu$ -(SCH<sub>2</sub>)<sub>2</sub>X](CN)<sub>2</sub>(CO)<sub>4</sub><sup>2-</sup>: Fully active CpI<sup>ADT</sup> (X=NH) and its non-active derivatives CpI<sup>PDT</sup>(X=CH<sub>2</sub>), CpI<sup>ODT</sup> (X=O) and CpI<sup>SDT</sup> (X=S).

### Results and discussion

### Only an adt-bridged [2Fe] cluster induces $H_2$ evolution activity in $\mbox{\rm Cpl}$

To compare the structure of active semisynthetic CpI containing the adt-bridged 2Fe<sub>H</sub>-subcluster with the native bacterial hydrogenase and to investigate potential structural aspects of the impaired function of the semisynthetic enzyme derivatives with different dithiolato bridges, apoCpI was maturated with four synthetic [2Fe] clusters. Besides the adt-bridged [2Fe] complex  $(Fe_2[\mu-(SCH_2)_2NH](CN)_2(CO)_4^2)$ , a pdt-bridged [2Fe] complex  $(Fe_2[\mu-(SCH_2)_2CH_2](CN)_2(CO)_4^{2-})$ , an odt-bridged [2Fe] complex  $(Fe_2[\mu-(SCH_2)_2O](CN)_2(CO)_4^{2-})$  and an sdt-bridged [2Fe] complex  $(Fe_2[\mu-(SCH_2)_2S](CN)_2(CO)_4^{2-})$  were synthesized following modified literature procedures<sup>32–38</sup> and used to prepare semisynthetic CpI as described before<sup>26</sup>. Specific hydrogen evolution activities with methylviologen as electron donor were 2874 +/- 262 (µmol H<sub>2</sub>) min<sup>-1</sup> (mg protein)<sup>-1</sup> for Cpl with the adt-bridged 2Fe<sub>H</sub>-cluster (CpI<sup>ADT</sup>), which is in agreement with previously reported values 10,26. Neither for apoCpI nor for the non-natural derivatives CpIPDT, CpIODT or CpI<sup>SDT</sup> could any hydrogen evolution activity be detected above the detection limit of 0.02% of the activity of CpI<sup>ADT</sup>. The same [2Fe] complexes were recently integrated into HydA1. While the odt-bridged and sdt-bridged complexes didn't induce H<sub>2</sub> evolution, 0.9 (µmol H<sub>2</sub>) min<sup>-1</sup> (mg protein)<sup>-1</sup> were reportedly produced by HydA1 with the pdt-bridged 2Fe<sub>H</sub>-cluster. This equals 0.17% of the activity of the same enzyme with the

nature-like adt-bridged  $2Fe_H$ -cluster. As HydA1 is smaller than Cpl, a better accessibility of the active site from the protein surface might promote undirected proton transfer. This could enable slow  $H_2$  production even though the directed proton transfer via the amine of the  $2Fe_H$ -cluster is disrupted.

All forms of CpI were crystallized under strictly anaerobic conditions and the crystal structures of both CpI<sup>ADT</sup> and apoCpI were solved with molecular replacement using the known structure of active, native CpI<sup>3,30</sup> as a search model and refined to 1.63 Å and 1.60 Å resolution respectively (figure 1, table 1). CpI<sup>ADT</sup> was subsequently used as a search model during molecular replacement to determine the structures of CpIPDT, CpI<sup>ODT</sup> and CpI<sup>SDT</sup> at 1.82 Å, 1.73 Å and 1.93 Å resolution respectively (figure 1, table 1). In contrast to already known structures of native CpI<sup>3,30,39</sup>, the spacegroup of the crystals was P2<sub>1</sub> for all five enzymes and the asymmetric units each contained two nearly identical molecules. Of these two molecules, one possesses a more flexible N-terminal domain. (residues 1-90), but at the same time a more rigid active site and thus yields a more reliable electron density in the important Hdomain in all structures. This becomes evident through the slightly lower temperature factors around the active site when compared to the second molecule. Accordingly figures and values given in the text were taken from the former molecule (chain B) if not stated otherwise, while the complete values for both chains of all structures can be found in the supplemental information.

As in vitro maturation of apo-[FeFe]-hydrogenases with synthetic [2Fe] cofactors was described only recently  $^{2,26}$ , we considered the exact structure of the 2Fe<sub>H</sub>-cluster and its environment in the semisynthetic enzyme to be of considerable interest. To minimize model bias of electron density in the active site cavity before starting to refine the 2Fe<sub>H</sub>-subcluster, at least two rounds of refinement of each structure were performed without a 2Fe<sub>H</sub>-cluster in the models.

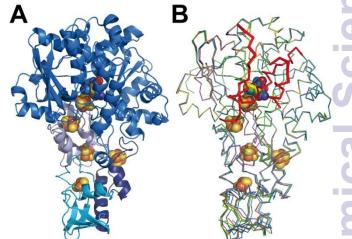


Fig. 1: Structures of unmaturated and semisynthetic [FeFe]-hydrogenases. (A) Cartoon model of [FeFe]-hydrogenase CpI<sup>ADT</sup> with domains in different hues of blue. (B) Overlay of ribbon models of [FeFe]-hydrogenases apoCpI (cyan), CpI<sup>ADT</sup> (marine), CpI<sup>PDT</sup> (magenta), CpI<sup>ODT</sup> (green) and CpI<sup>SDT</sup> (yellow). H domain regions significantly different to apoHydA1 indicated as thicker ribbon and in red in B. FeSc cluster atoms depicted as spheres and colored according to element (Fe=brown, S=beige, O=red, N=blue, C in color of respective protein) in A and B.

Journal Name ARTICLE

	apoCpl	Cpl <sup>ADT</sup>	CpIPDT	Cpl <sup>ODT</sup>	Cpl <sup>SDT</sup>
A. Crystallographic data		•	•	· · · · · · · · · · · · · · · · · · ·	·
X-ray source	SPring8-BL44XU	SLS-PXII	SLS-PXII	SLS-PXII	SLS-PXII
Space group	P2 <sub>1</sub>	P2 <sub>1</sub>	P2 <sub>1</sub>	P2 <sub>1</sub>	P2 <sub>1</sub>
Unit-cell parameters					
a (Å)	90.06	91.34	87.47	89.66	89.83
b (Å)	71.81	73.65	72.07	72.45	73.13
c (Å)	103.31	103.88	102.71	102.94	103.04
Wavelength (Å)	0.900	0.978	0.979	0.979	0.979
Resolution range (Å)	50.00 - 1.60	48.365 - 1.63	47.58 - 1.82	47.77 - 1.73	48.03 - 1.93
	(1.63 - 1.60) <sup>a</sup>	(1.67 - 1.63)	(1.87 - 1.82)	(1.77 - 1.73)	(1.98 - 1.93)
Total reflections	645133 (31520)	3729315 (129491)	760236 (56288)	929911 (70575)	674928 (47804)
Unique reflections	172056 (8519)	170277 (12470)	112496 (8270)	136702 (10132)	99824 (7353)
Completeness (%)	99.7% (99.5%)	99.9% (99.7%)	99.9% (100.0%)	100.0% (100.0%)	99.9% (99.9%)
R <sub>meas</sub> (%)	5.2% (56.5%)	8.7% (109.1%)	14.3% (98.8%)	8.7% (83.1%)	15.8% (93.6%)
Ι/σ(Ι)	36.8 (3.2)	20.6(2.2)	9.6 (2.0)	14.1 (2.1)	9.9 (2.1)
Correlation coefficient (CC 1/2) <sup>b</sup>	- (-)	99.9 (87.1)	99.7 (77.9)	99.9 (84.1)	99.6 (68.8)
B. Refinement statistics	S				
PDB code	4XDD	4XDC	5BYR	5BYQ	5BYS
Resolution (Å)	1.60	1.63	1.82	1.73	1.93
R <sub>work</sub>	0.14	0.15	0.16	0.15	0.16
R <sub>free</sub>	0.17	0.18	0.19	0.18	0.19
No. Atoms (except H)	10548	10385	9859	9978	9991
Protein	9050	9047	8893	8982	8903
Ligand	72	106	106	106	106
Solvent/ion	1426	1232	860	890	982
RMSD from ideal					
Bond lengths (Å)	0.006	0.013	0.018	0.009	0.014
Bond angles (°)	0.95	1.28	1.48	1.07	1.28
Ramachandran plot					
Most favored (%)	97.67	96.90	96.32	97.23	96.59
Additionally allowed (%)	2.33	3.02	3.68	2.77	3.41
Outliers (%)	0.00	0.09	0.00	0.00	0.00
B factors					
overall	35.0	39.0	32.0	35.0	32.0
2FeH cavity (chain A / chain B)	19.5/22.7	24.5/22.0	19.0/16.8	22.3/19.3	18.6/17.3
2FeH (chain A / chain B)	-/-	23.3/20.7	19.3/17.6	23.1/20.2	18.5/17.3
Average occupancy of 2FeH (chain A/ chain B)	-/-	0.98/0.95	0.94/0.96	0.93/0.96	0.98/0.97

<sup>&</sup>lt;sup>a</sup> Numbers in parenthesis represent values for the highest resolution bin.

<sup>&</sup>lt;sup>b</sup> Correlation coefficient CC (1/2) as defined in Karplus and Diederichs 2012.<sup>55</sup>

Subsequently, starting models of the 2Fe<sub>H</sub>-subcluster based on the structure of native CpI<sup>30</sup> with optimized geometry but adapted composition of the dithiolato moiety<sup>5</sup>, were used. Restraints were applied to all bond distances in the subcluster. We additionally restrained the angles defining the positions of the CO and CN<sup>-</sup> ligands. The position of the bridging CO was not restrained due to its reported flexibility<sup>5</sup> depending on the redox state of the enzyme. Final models of the 2Fe<sub>H</sub> subclusters were verified by inspection of composite omit map.

#### Presumed maturation channel closed in apoCpI crystal structure

The crystal structure of apoCpI reported here is strikingly similar to structures of active CpI, both native and semisynthetic (figure 1). The overall RMSD of the backbone atoms of apoCpI and native  $\mbox{Cpl}^{30}$  is as low as 0.3 Å, while apoCpl and  $\mbox{Cpl}^{\mbox{\scriptsize ADT}}$  display an RMSD of 0.4 Å over all backbone atoms (table S1). Significant differences in side-chain orientation are mainly limited to surface exposed residues with V423 being a notable exception (figure 2). This residue in the central cavity is adapting a different rotamer, supposedly stabilized by one of the water molecules in the binding pocket for 2Fe<sub>H</sub>. As demonstrated earlier31, the structure of apoHydA1 from C. reinhardtii lacking the 2Fe<sub>H</sub>-subcluster shows overall great similarity to the structure of the H-domain of Cpl<sup>3</sup> and DdH<sup>29</sup> with regard to the backbone geometry, but exhibits regions of pronounced differences. Amongst these differences is a channel from the surface to the site of the 2Fe<sub>H</sub>-subcluster, which is only present in apoHydA1<sup>31</sup>. Three regions, which can be understood as plug, lock and lid, block the channel in all known structures of active [FeFe]-hydrogenases (figure S1B), while they are remote in apoHydA1. They presumably shift to close the channel and complete the first sphere of amino acid residues around the Hcluster upon integration of the 2Fe<sub>H</sub>-subcluster in HydA1<sup>31</sup> (figure S1A). However, there has neither been a structure of a maturated [FeFe]-hydrogenase of the short chlorophyta type nor of an unmaturated bacterial type enzyme, which would have allowed for a direct comparison.

The structure of apoCpI presented here shows the three regions 405-423 ("plug"), 437-453 ("lid") and 529-540 ("lock") clearly in a "closed" conformation nearly identical to active CpI (figure 1). Prominently, F417 in direct contact to the 2Fe<sub>H</sub>-subcluster shows minimal deviation in apoCpI when compared to CpIADT (figure 2), while it is moved by 15 Å in apoHydA1. Washed and subsequently dissolved crystals of apoCpI could be maturated with the synthetic adt-bridged [2Fe] cluster to an activity of 1250 nmol H₂/min/crystal, reassuring that the reported closed structure of apoCpI is not a dead-end conformation. This suggests an equilibrium between a "closed" and an "open" state in apoCpI, of which only the former readily crystallizes. Within the regions with striking deviation between apoHydA1 and apoCpl, several glycine residues can be identified, which are highly conserved in a recent sequence alignment of all known [FeFe]-hydrogenase sequences<sup>40</sup> (table S2). These amino acids could function as hinges, as for some of them the "open" or "closed" conformation respectively would imply dihedral angle combinations commonly found only for glycine residues 41 (table S2). Their high degree of conservation thus is another hint that an open and closed form of all [FeFe]-hydrogenases exists.

### Rigid cavity in apoCpI forces the [2Fe] complex to move into its active conformation

Being devoid of the 2Fe<sub>H</sub>-cluster, the active site binding pocket of apoCpl is occupied by seven water molecules and a chloride ion (figure 2) instead. Note that this leaves a water filled cavity of roughly 10 Å diameter in the center of the protein. Nonetheless, residues which are assumed to interact with the cofactor in the active enzyme are shifted only very slightly by 0.1-0.3 Å towards a narrower cavity (table S3, figure 2) and show the same low temperature factors as most of the H-domain (table 1). This exemplifies how the amino acids of seven distinct protein parts (around amino acids 231, 299, 324, 353, 417, 497 and 503; figure 3) coordinate to form a rigid

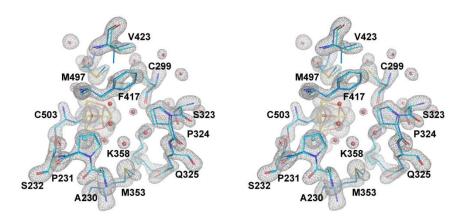


Fig. 2: The central cavity of apoCpl and Cpl<sup>ADT</sup>. Stereo view of a stick model of the central cavity of apoCpl (carbon atoms in cyan) with Fo-Fc simulated annealing omit map contoured at 4σ in the identical orientation as Figure 3. A stick model of amino acids lining the central cavity of Cpl<sup>ADT</sup> is superposed (carbon atoms in marine). Numbering of amino acids as in the structure of native Cpl. Small spheres indicate water atoms (red) and chloride ion (cyan) not present in Cpl<sup>ADT</sup>.

**Journal Name** 

central cavity, perfectly positioned to arrange the ligands of a [2Fe] cluster in its center.

### Crystal structure of semisynthetic CpI<sup>ADT</sup> reveals native-like structure and open coordination site

Superposition of the structure of semisynthetic Cpl<sup>ADT</sup> and the best available crystal structure of native Cpl<sup>30</sup> results in nearly identical structures with an RMSD of 0.3 Å for the main chain atoms. Even with regard to the side chain atom conformations, significant differences between Cpl<sup>ADT</sup> and the native Cpl can only be found in several surface exposed residues, which is surprising given the considerable differences in the crystal packing.

When comparing the important cofactor-peptide interactions in native CpI and the structure of CpI<sup>ADT</sup>, the distances between the atoms of 2Fe<sub>H</sub> and their respective interaction partners in the protein environment show a maximum deviation of 0.17/0.13 Å and an average deviation of 0.06/0.05 Å for chain A/chain B (figure 3, table S4). This is well within the experimental error of crystal structure analysis at the given resolution. Moreover the synthetic 2Fe<sub>H</sub>-cluster itself in the structure of CpI<sup>ADT</sup> compares very well to the *in vivo* synthesized version in native  $\mbox{CpI}^{30}$  (figure 3) and the [FeFe]-hydrogenases DdH<sup>29</sup> and HydA1 (data from XANES/EXAFS)<sup>42</sup> within the error of crystal structures of macromolecules (table S5). This finding confirms data from FTIR and EPR spectroscopic studies, which showed excellent agreement of native and semisynthetic protein for the small [FeFe]-hydrogenase HydA1 from Chlamydomonas reinhardtii<sup>26,28</sup>. For [FeFe]-hydrogenases with additional N-terminal domains like CpI or DdH, a bridging conformation of one CO ligand is assumed to occur only in the H<sub>ox</sub> or CO inhibited state<sup>5,21,29</sup>. In our structure the CO ligand between the Fe atoms is positioned at an angle of 114°/132° (chain A /chain B) between Fed-C-O (table S5, figure 4), which does not indicate terminal binding of the CO to Fed as previously published for a structure of reduced DdH<sup>5</sup>. Thus we understand the here reported structure of  $CpI^{ADT}$  to be mainly in the  $H_{ox}$ state. While in earlier structures of CpI<sup>3,30</sup> a region of low but significant electron density next to Fe<sub>d</sub> was assigned as a water molecule in this particular redox state, in the structure described here the H-cluster of both chains clearly features one coordination site on Fe<sub>d</sub> devoid of electron density (figure 4).

A comparison of the crystal structures of the synthetic adtbridged [2Fe] complex<sup>37</sup> before and after integration into the protein environment as 2Fe<sub>H</sub> illustrates the distortions that the protein forces upon the [2Fe] complex (figure 4). An Fe-S-Fe bridge to the 4Fe<sub>H</sub>-cluster is formed and, as demonstrated earlier, one CO ligand is lost during the process of activation<sup>26</sup>. Another CO ligand shifts into a bridging position between the two Fe atoms and the CO/CN<sup>-</sup> ligands move into an octahedral coordination at each Fe with nearly perpendicular equatorial planes (figure 4). This conformation has been attributed a crucial role in allowing the mixed Fe(I)-Fe(II) valency of the H<sub>ox</sub> state within the catalytic cycle, which is difficult to achieve in isolated [2Fe] clusters<sup>43</sup>. Additionally the new conformation features the open

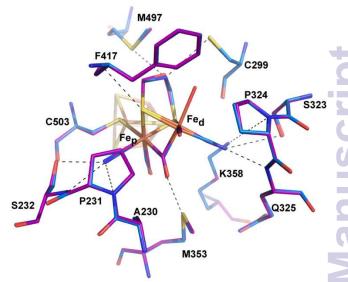
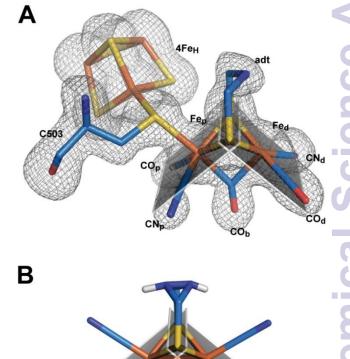


Fig. 3: Comparison of the active site of CpI<sup>ADT</sup> and native CpI. Stick model of the environment of the  $2Fe_{H^-}$ subcluster of CpI<sup>ADT</sup> (carbon atoms in marine) superposed to a stick model of native CpI (PDB ID 3C8Y) $^{30}$  (carbon atoms in magenta). Dashed lines indicate potential interactions between  $2Fe_{H}$  and the protein listed in table S4. Numbering of amino acids as in the structure of native CpI.

Fig. 4: Structure of semisynthetic H-cluster and structural changes in ligand coordination upon integration of  $2Fe_{H^-}$  (A) Stick model of the H-cluster of  $Cpl^{ADT}$  colored according to element with  $F_o$ - $F_c$  simulated annealing omit map contoured at 3.5o. (B) Stick model of the crystal structure of  $Fe_2[\mu-(SCH_2)_2NH](CN)_2(CO)_4^{2-37}$ . The planes in A and B are drawn through the sulfur atoms of the [2Fe] complexes and one of the two Fe atoms each to clarify the coordination geometry of the Fe ligands.



coordination site at  $Fe_d$  trans to the bridging CO (figure 4). This promotes regionselectivity of  $H_2$  binding or hydride formation close to the amine in the adt-bridge, which is believed to be crucial for the mechanism  $^{16,44}$ .

### $\ensuremath{\mathsf{2Fe}_\mathsf{H}}$ subsite structure remains unaltered upon changes in the dithiolato bridge

The structures of all three CpI derivatives with non-natural 2Fe<sub>H</sub> subsites superpose very well with each other and the native CpI, apoCpI and CpI<sup>ADT</sup> with RMSD's for Cα atoms between 0,2 Å and 0,5 Å (table S1). Comparison of the exact positions of amino acids supposedly involved in enzyme function, e.g.amino acids in the proton transfer pathway or around the active site, yielded little differences within the limits of exactness of macromolecular crystallography (figure 5). The average RMSDs of all atoms of selected amino acids were as low as 0.08-0.11 Å when comparing the non-natural derivatives with CpI<sup>ADT</sup>. As significant differences in the degree of maturation were observed for semisynthetic HydA1 with non-natural 2Fe<sub>H</sub> clusters<sup>27</sup>, we allowed variation of the occupancies of the atoms of the 2Fe<sub>H</sub>-subclusters during refinement. According to this rough estimate more than 90% of the molecules in the crystals contained the 2Fe<sub>H</sub> subsite (table 1). Even though the effects of partial occupancy and temperature factor are hardly discernible at the given resolution, we expect these results to be a good lower limit as the calculated temperature factors of the  $2Fe_{H^-}$ clusters and the surrounding amino acids agree well (table 1). As apoCpl crystallizes in a nearly identical structure and an isomorphous unit cell we assume that the high occupancy of the 2Fe<sub>H</sub>-clusters does not result from positive selection during crystal formation, but illustrates the effectiveness of *in vitro* maturation of Cpl with the chosen [2Fe] clusters.

For the odt-bridged  $2Fe_H$  subsite in HydA1 a less pronounced bridging character of  $CO_b$  compared to other semisynthetic HydA1 enzymes was reported according to FTIR data of the "as isolated" state.  $^{27}$  A very similar state represented a minor part of the mixed population of HydA1 with sdt-bridged  $2Fe_{H}$ -subcluster in the same study. We found an angle of  $145^\circ$  between  $Fe_d$ -C-O of the  $CO_b$  ligand in  $CpI^{SDT}$  which indicates a more terminal than bridging character, but the other structures including  $CpI^{ODT}$  reveal angles suggesting a bridging CO (figure 6, table S5).

Because of the above discussed effects of redox state changes on the CO<sub>b</sub> ligand, a potential dependent FTIR based investigation of the CpI enzyme derivatives would be needed to clarify if this is merely an effect of the redox state at the point of crystal mounting or inherent to the different dithiolato bridge. A detailed comparison of distances between the atoms of the 2Fe<sub>H</sub> subsite and the surrounding amino acids indicates a slightly different position of Fe<sub>d</sub> within the cavity for CpI<sup>PDT</sup> 0.1 Å closer to Ala 230 and further away from Cys 299 (table S4). Besides this, small differences in the dithiolato bridge can be observed. While the bridgehead atom is leaning about 0.2 Å further away from Met 497 in the inactive CpI

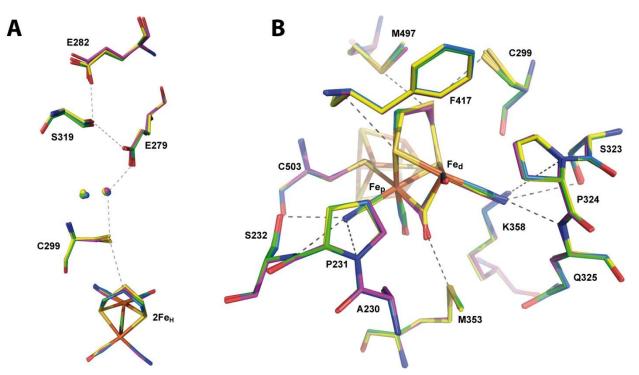
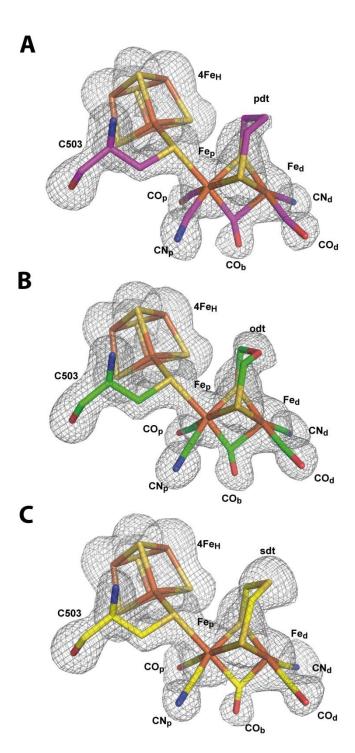


Fig. 5: Comparison of the catalytically important amino acids in CpI derivatives. Stick models of the potential proton transfer pathway (A) and the environment of the 2Fe<sub>H</sub>-subcluster (B) of CpI<sup>ADT</sup> (carbon atoms in marine) superposed to stick models of CpI<sup>PDT</sup> (magenta), CpI<sup>ODT</sup> (green) and CpI<sup>SDT</sup> (yellow). Dashed lines indicate potential proton transfer interactions or potential interactions of 2Fe<sub>H</sub> with the protein as listed in table S4. Numbering of amino acids as in the structure of native CpI.

Journal Name ARTICLE



**Fig. 6: Models of the H-cluster of non-native CpI derivatives.** Stick models of (A) CpI<sup>PDT</sup> (carbon atoms in magenta), (B) CpI<sup>DDT</sup> (carbon atoms in green) and (C) CpI<sup>SDT</sup> (carbon atoms in yellow) with  $F_o$ - $F_c$  simulated annealing omit maps contoured at 3.5 $\sigma$ .

derivatives, the sulfur atom of Cys 299 is pushed back to keep roughly the van-der-Waals distance to the bridgehead atom of the dithiolato bridge in the three structures (figure 5B, table S3). However, the position and geometry of the non-natural 2Fe<sub>H</sub>-clusters do not show any large differences (figure 5, figure 6, table S5) when compared to native CpI or CpI<sup>ADT</sup> and thus do not offer a clear structural explanation for the impaired activity. For CpI<sup>PDT</sup> this is in line with a recent ENDOR and HYSCORE study of HydA1 containing a pdt-bridged 2Fe<sub>H</sub>-subcluster, which

showed very similar spectra in comparison to in vivo maturated DdH. 45 DFT calculations performed for adt-bridged, pdt-bridged and odt-bridged 2Fe<sub>H</sub>-subclusters in CpI also resulted in very similar geometries<sup>30</sup> (table S5). Remarkably, there is no significant electron density in our structures close to Fed at the postulated site of H<sub>2</sub> binding in any of the 2Fe<sub>H</sub>-subclusters (figure 6). Thus binding of an inhibitor to this open coordination site can be ruled out as cause for the quantitative loss of activity. For HydA1 with the pdt-bridged 2Fe<sub>H</sub> cluster no binding of CO to the active site was observed in a recent FTIR based study. 28 Our structure of CpIPDT rules out a rearrangement in the neighboring amino acids as explanation for this behavior. However, once an inhibitory CO is bound to Fed, the distance between the central atom of the dithiolate bridge and the oxygen of CO was reported to be as close as ~ 2.5 Å. 5 While the single hydrogen of an amine bridgehead proposedly points towards C299 and thus away from Fed, the pdt's central methyl group might considerably obstruct binding of CO to Fed through its hydrogen atoms not visible in X-ray crystallography at the given resolution.

### **Conclusions**

We herein report the structure of the [FeFe]-hydrogenase CpI from Clostridium pasteurianum in the unmaturated apo-form and the first high resolution structure of a fully active semisynthetic [FeFe]-hydrogenase along with the structures of three non-natural inactive derivatives of this [FeFe]hydrogenase with changes in the inorganic active site. Surprisingly, the unmaturated apoCpI crystallizes in an overal conformation like the maturated enzyme and not similar to the structure of unmaturated HydA1. The high degree of rigidity of the amino acids in proximity to the H-cluster even in the absence of the 2Fe<sub>H</sub>-subcluster demonstrates how the protein structure is designed to force the 2Fe-cofactor into its highly active form. Semisynthetic CpI<sup>ADT</sup> shows a nearly identical conformation when compared to the native enzyme including the rotated conformation of the 2Fe<sub>H</sub> cofactor with octahedral geometry at both Fe atoms. Unlike previous structures of Cp. the structure of CpI<sup>ADT</sup> presented here displays a completely unoccupied open coordination site at Fe<sub>d</sub>. Non-natural derivatives of the 2Fe<sub>H</sub> subsite with changes in the central aton of the dithiolato bridge can well be incorporated into apoCpI as already reported for apoHydA1, but do not lead to H2 evolution. activity. The structures of the protein matrix of CpIPDT, CpIODT and CpISDT show no clear differences to the highly active CpIADT. Despite their divergence in activity all four different 2Fe<sub>H</sub>subclusters investigated in this study are adapted in theil conformation to the protein matrix when compared to the structures of the free [2Fe] complexes and take up the same typical structure. The proposed site of H2 oxidation at Fed is unoccupied in all structures reported here, which excludes inhibitor binding or steric hindrance as reasons for impaired activity. The structural information gained in this study in combination with previously reported  ${\rm FTIR}^{27,28}$  and  ${\rm EPR}^{4^r}$ spectroscopic data of inactive active site variants of [FeFe]hydrogenases underline the central role the chemistry of the dithiolato bridge plays for enzyme activity. Once the protein

environment has forced the iron ligands into the typical conformation, which is the basis of  $H_2$  evolution, it is solely the reactivity of the central amine, which induces enzyme activity.

### **Experimental Section**

ApoCpI was expressed with a C-terminally fused strep-tagII in E.c. BL21(DE3)  $\Delta iscR^{46}$  under anaerobic conditions as described earlier  $^{47}$  without coexpression of [FeFe]-hydrogenase specific maturases. Protein purification was achieved by strep-tactin affinity chromatography under strictly anaerobic conditions  $^{48}$  with a 10 mM Tris-HCl buffer with pH 8.0 and 2 mM NaDT and purity was assessed by SDS-PAGE.

 $[Fe_2[\mu\text{-}(SCH_2)_2NH](CN)_2(CO)_4][Et_4N]_2 \quad was \quad synthesized \quad as \\ reported \quad earlier^2. \quad As \quad the \quad purification \quad of \quad [Fe_2[\mu\text{-}(SCH_2)_2NH](CN)_2(CO)_4][Et_4N]_2 \ by \ washing \ with \ hexane^{34} \ did \ not \\ result \quad in \quad clean \quad product, \ the \quad recently \quad described \quad purification \\ procedure \quad for \quad [Fe_2[\mu\text{-}(SCH_2CH_2CH_2S)](CN)_2(CO)_4][Et_4N]_2^2 \quad was \\ adopted \quad for \quad [Fe_2[\mu\text{-}(SCH_2)_2NH](CN)_2(CO)_4][Et_4N]_2. \\ \end{cases}$ 

$$\label{eq:condition} \begin{split} & [\text{Fe}_2[\mu\text{-}(\text{SCH}_2\text{CH}_2\text{CH}_2\text{S}](\text{CN})_2(\text{CO})_4][\text{Et}_4\text{N}]_2, & [\text{Fe}_2[\mu\text{-}(\text{SCH}_2)_2\text{S}](\text{CN})_2(\text{CO})_4][\text{Et}_4\text{N}]_2 & \text{and} & [\text{Fe}_2[\mu\text{-}(\text{SCH}_2)_2\text{O}](\text{CN})_2(\text{CO})_4][\text{Et}_4\text{N}]_2 & \text{were synthesized according to literature procedures}^{33,35,36,38} & \text{and purity of each sample was checked by $^1$H NMR and IR spectroscopy. Samples were stored at $-35^\circ$C under inert atmosphere to avoid any decomposition of the artificial cofactors.} \end{split}$$

Maturation of apoCpl to Cpl<sup>ADT</sup>, Cpl<sup>PDT</sup>, Cpl<sup>ODT</sup> or Cpl<sup>SDT</sup> with a 10fold excess of Fe<sub>2</sub>[µ-(SCH<sub>2</sub>)<sub>2</sub>X](CN)<sub>2</sub>(CO)<sub>4</sub>[Et<sub>4</sub>N]<sub>2</sub> was achieved in a 0.1 M K<sub>2</sub>HPO<sub>4</sub>/KH<sub>2</sub>PO<sub>4</sub> buffer system at pH 6.8 with 2 mM NaDT as described earlier<sup>26</sup> at RT for 1 hour to ensure complete maturation of the sample. The semisynthetic enzymes were subsequently cleaned from leftover [2Fe] complex and buffered again into a 10 mM Tris-HCl buffer with pH 8.0 and 2 mM NaDT by use of a NAP™ 5 (GE Healthcare) size exclusion chromatography column. Enzyme preparations were concentrated using Amicon Ultra centrifugal filters 30K (Millipore) under anaerobic conditions. Success of maturation and quality of purified protein samples of CpI<sup>ADT</sup> were determined by testing their H<sub>2</sub> production activity in vitro with methylviologen as electron donor using an established  $\mathsf{method}^{49}.$  To test for catalytic activity of the non-native semisynthetic enzymes, the same method was applied and additional measurements with 10fold increased protein amount were conducted to lower the limit of detection.

Box-like protein crystals of apoCpI and the semisynthetic hydrogenases were obtained with PEG 3000 or PEG 4000 as precipitant using the hanging drop or sitting drop vapor diffusion method at 277 K under anaerobic conditions within 2-4 days when mixing reservoir solution 1:1 with protein solution (10 mg/ml). The crystallization conditions for the selected crystals of apoCpI were 12% PEG 3000, 0.1 M MES pH 6.5, 0.2 M MgCl<sub>2</sub> in a sitting drop vapor diffusion experiment and cryoprotection was achieved with a final concentration of 15% glycerol in 15% PEG 3000, pH 6.5, 0.2M MgCl<sub>2</sub>. CpI<sup>ADT</sup> crystals selected for diffraction experiments were grown in 11% PEG 4000, 0.1 M MES pH 7.0, 0.2 M MgCl<sub>2</sub> in a hanging drop experiment and protected against formation of ice crystals with paraffin oil. Crystals of the non-native semisynthetic enzymes

were grown by hanging drop vapor diffusion using 0.1 M MES pH 6.0, 0.4 M MgCl<sub>2</sub> and a total of 40% v/v of PEG4000 and glycerol to avoid the need of additional cryo-protection during crystal mounting. In detail the reservoir solutions contained 15% PEG 4000, 25% glycerol for CpIPDT, 19% PEG 4000, 21% glycerol for CpI<sup>ODT</sup> and 21% PEG 4000, 19% glycerol for CpI<sup>SDT</sup>. Maturation capability of crystallized apoCpI was tested by washing a crystal in three fresh drops of its reservoir solution followed by dissolution of the crystal in cold 0.1 M K<sub>2</sub>HPO<sub>4</sub>/KH<sub>2</sub>PO<sub>4</sub> buffer at pH 6.8 with 2 mM NaDT under strictly anaerobic conditions. Maturation was started by addition of 1.5  $Fe_2[\mu-(SCH_2)_2NH](CN)_2(CO)_4[Et_4N]_2$ pmol K<sub>2</sub>HPO<sub>4</sub>/KH<sub>2</sub>PO<sub>4</sub> buffer, pH 6.8, and allowed to proceed for 1 h at 4°C. Subsequently the mixture was transferred completely into a solution of methylviologen, NaDT and phosphate buffer as for standard tests for H2 evolution activity and treated accordingly<sup>49</sup>.

Mounting of protein crystals into CryoLoops™ (Hampton Research) and subsequent flash-freezing in liquid N2 was performed under strictly anaerobic conditions at 298 K. Diffraction data were collected at 100 K at beamline BL44-XU at SPring-8 (Hyogo, Japan) and beamline PXII at the SLS (Villingen, Switzerland) and the data were processed using the software package HKL2000<sup>50</sup> and XDS<sup>51</sup> for apoCpI and the semisynthetic hydrogenases, respectively. Molecular replacement and structure optimization were performed with the software packages CCP452 (apoCpl and CplADT) and PHENIX53 (CplPDT, Cpl<sup>ODT</sup> and Cpl<sup>SDT</sup>) and Coot<sup>54</sup>. At least two final refinement runs were conducted with PHENIX on all structures to improve comparability of the final models. In order to estimate the occupancy of the 2Fe<sub>H</sub>-cluster in the structures of CpI<sup>ADT</sup> and other derivatives, we applied a partial occupancy refinement at the final stage of PHENIX refinement. Simulated annealing omit maps were calculated with PHENIX, omitting the H-cluster with the bridging cysteine residue and the residues around the central cavity as well as all atoms within the central cavity for the semisynthetic [FeFe]-hydrogenases and apoCpl, respectively.

### **Acknowledgements**

We thank the staff at beam line PXII at SLS, Switzerland, and BL44XU at SPring-8, Japan, and at the ESRF, France, for their help during data collection.

### **Notes and references**

### **Funding sources**

This work was supported in part by an International Joint Research Promotion Program, Osaka University, (G.K, T.H.); the Cabinet Office of Japan, through the Funding Program for Next Generation World-Leading Researchers (NEXT Program) (GS016, to G.K.) and the Studienstiftung des deutschen Volkes (J.E.). U.-P.A was supported by the Fonds of the Chemical Industry (Liebig grant) and the Deutsche Forschungsgemeinschaft (Emmy Noether grant AP242/2-1).

T.H. thanks the Cluster of Excellence RESOLV (EXC1069) funded by the Deutsche Forschungsgemeinschaft (DFG) and the

The authors declare no competing financial interest.

#### **Accession Numbers**

Volkswagen Foundation (LigH2t).

Journal Name

The coordinates and structure factors for all structures were deposited with the Protein Data Bank under the following accession numbers: apoCpl: 4XDD, Cpl<sup>ADT</sup>: 4XDC, Cpl<sup>PDT</sup>: 5BYR, Cpl<sup>ODT</sup>: 5BYQ, Cpl<sup>SDT</sup>: 5BYS.

#### **Abbreviations**

2Fe<sub>H</sub>, [2Fe] subcluster of the H-cluster of [FeFe]-hydrogenases; 4Fe<sub>H</sub>, [4Fe4S] subcluster of the H-cluster; adt, aza -dithiolate; apoCpI, CpI lacking only the 2Fe<sub>H</sub>-cluster; CpI, [FeFe]hydrogenase I from Clostridium pasteurianum; CpIADT, apoCpI maturated in vitro with  $(Fe_2[\mu-(SCH_2)_2NH](CN)_2(CO)_4^{2-})$ ;  $CpI^{ODT}$ , apoCpI maturated in vitro with  $(Fe_2[\mu-(SCH_2)_2O](CN)_2(CO)_4^{2-})$ ; CpIPDT, apoCpl maturated in vitro with  $(SCH_2)_2CH_2](CN)_2(CO)_4^{2-})$ ;  $CpI^{SDT}$ , apoCpI maturated in vitro with  $(Fe_2[\mu-(SCH_2)_2S](CN)_2(CO)_4^{2-})$ ; DdH, [FeFe]-hydrogenase from Desulfovibrio desulfuricans; HydA1, [FeFe]-hydrogenase I from Chlamydomonas reinhardtii; odt, oxo -dithiolate; pdt, propanedithiolate; sdt, sulfur-dithiolate.

#### **Supporting Information**

Tables listing and comparing the RMSD of the structures, distances and angles of the 2Fe<sub>H</sub>-subclusters, the distances from 2Fe<sub>H</sub>-subcluster atoms to selected amino acids and the distances of amino acids lining the 2Fe<sub>H</sub>-subcluster cavity, a figure showing the presumed maturation channel in more detail and additional information on the suggested glycine hinges and stereo views of all figures presented in the main article.

### References

- 1 M. Frey, ChemBioChem, 2002, 3, 153-160.
- G. Berggren, A. Adamska, C. Lambertz, T. R. Simmons, J. Esselborn, M. Atta, S. Gambarelli, J.-M. Mouesca, E. Reijerse, W. Lubitz, T. Happe, V. Artero and M. Fontecave, *Nature*, 2013, 499, 66–69.
- 3 J. W. Peters, W. N. Lanzilotta, B. J. Lemon and L. C. Seefeldt, Science, 1998, 282, 1853–1858.
- 4 A. Silakov, B. Wenk, E. Reijerse and W. Lubitz, *Phys. Chem. Chem. Phys.*, 2009, **11**, 6592.
- 5 Y. Nicolet, A. L. de Lacey, X. Vernède, V. M. Fernandez, E. C. Hatchikian and J. C. Fontecilla-Camps, *J. Am. Chem. Soc.*, 2001, **123**, 1596–1601.
- 6 T. R. Simmons, G. Berggren, M. Bacchi, M. Fontecave and V. Artero, *Coord. Chem. Rev.*, 2014, **270–271**, 127–150.
- 7 W. Lubitz, H. Ogata, O. Rüdiger and E. Reijerse, Chem. Rev., 2014, 114, 4081–4148.
- 8 C. A. Boyke, J. I. van der Vlugt, T. B. Rauchfuss, S. R. Wilson, G. Zampella and L. De Gioia, *J. Am. Chem. Soc.*, 2005, **127**, 11010–11018.
- M. Bruschi, C. Greco, L. Bertini, P. Fantucci, U. Ryde and L. D. Gioia, J. Am. Chem. Soc., 2010, 132, 4992–4993.
- 10 P. Knörzer, A. Silakov, C. E. Foster, F. A. Armstrong, W. Lubitz and T. Happe, *J. Biol. Chem.*, 2012, **287**, 1489–1499.

- 11 G. A. N. Felton, C. A. Mebi, B. J. Petro, A. K. Vannucci, D. H. Evans, R. S. Glass and D. L. Lichtenberger, *J. Organomet. Chem.*, 2009, **694**, 2681–2699.
- 12 J. M. Camara and T. B. Rauchfuss, Nat. Chem., 2012, 4, 26–30.
- 13 C. Tard, X. Liu, S. K. Ibrahim, M. Bruschi, L. D. Gioia, S. C. Davies, X. Yang, L.-S. Wang, G. Sawers and C. J. Pickett, *Nature*, 2005, **433**, 610–613.
- 14 C. Greco, Inorg. Chem., 2013, 52, 1901-1908.
- 15 C. Greco, M. Bruschi, P. Fantucci, U. Ryde and L. De Gioia, *Chemphyschem Eur. J. Chem. Phys. Phys. Chem.*, 2011, **12**, 3376–3382.
- 16 M. Bruschi, C. Greco, M. Kaukonen, P. Fantucci, U. Ryde and L. De Gioia, *Angew. Chem. Int. Ed.*, 2009, **48**, 3503–3506.
- 17 M. E. Carroll, B. E. Barton, T. B. Rauchfuss and P. J. Carroll, J. Am. Chem. Soc., 2012, 134, 18843–18852.
- 18 S. Ezzaher, A. Gogoll, C. Bruhn and S. Ott, Chem. Commun., 2010, 46, 5775–5777.
- N. Wang, M. Wang, L. Chen and L. Sun, *Dalton Trans.*, 2013,
   42, 12059–12071.
- 20 C.-H. Hsieh, O. F. Erdem, S. D. Harman, M. L. Singleton, E. J. Reijerse, W. Lubitz, C. V. Popescu, J. H. Reibenspies, S. M. Brothers, M. B. Hall and M. Y. Darensbourg, J. Am. Chem. Soc., 2012.
- 21 W. Roseboom, A. L. Lacey, V. M. Fernandez, E. C. Hatchikian and S. P. J. Albracht, *JBIC J. Biol. Inorg. Chem.*, 2006, **11**, 102–118.
- 22 A. Silakov, C. Kamp, E. Reijerse, T. Happe and W. Lubitz, Biochemistry (Mosc.), 2009, 48, 7780–7786.
- 23 A. J. Cornish, K. Gartner, H. Yang, J. W. Peters and E. L. Hegg, J. Biol. Chem., 2011, 286, 38341–38347.
- 24 D. W. Mulder, D. O. Ortillo, D. J. Gardenghi, A. V. Naumov, S. S. Ruebush, R. K. Szilagyi, B. Huynh, J. B. Broderick and J. W. Peters, *Biochemistry (Mosc.)*, 2009, 48, 6240–6248.
- 25 E. M. Shepard, F. Mus, J. N. Betz, A. S. Byer, B. R. Duffus, J. W. Peters and J. B. Broderick, *Biochemistry (Mosc.)*, 2014, 53, 4090–4104.
- 26 J. Esselborn, C. Lambertz, A. Adamska-Venkatesh, T. Simmons, G. Berggren, J. Noth, J. Siebel, A. Hemschemeier, V. Artero, E. Reijerse, M. Fontecave, W. Lubitz and T. Happe, Nat. Chem. Biol., 2013, 9, 607–609.
- 27 J. F. Siebel, A. Adamksa-Venkatesh, K. Weber, S. Rumpel, E. J. Reijerse and W. Lubitz, *Biochemistry (Mosc.)*, 2015, **54**, 1474–1483.
- 28 A. Adamska-Venkatesh, D. Krawietz, J. Siebel, K. Weber, T. Happe, E. Reijerse and W. Lubitz, J. Am. Chem. Soc., 2014, 136, 11339–11346.
- 29 Y. Nicolet, C. Piras, P. Legrand, C. E. Hatchikian and J. C. Fontecilla-Camps, *Structure*, 1999, **7**, 13–23.
- A. S. Pandey, T. V. Harris, L. J. Giles, J. W. Peters and R. K. Szilagyi, J. Am. Chem. Soc., 2008, 130, 4533–4540.
- 31 D. W. Mulder, E. S. Boyd, R. Sarma, R. K. Lange, J. A. Endrizzi, J. B. Broderick and J. W. Peters, *Nature*, 2010, 465, 248–251.
- 32 E. J. Lyon, I. P. Georgakaki, J. H. Reibenspies and M. Y. Darensbourg, *Angew. Chem. Int. Ed.*, 1999, **38**, 3178–3180.
- 33 A. L. Cloirec, S. C. Davies, D. J. Evans, D. L. Hughes, C. J. Pickett, S. P. Best and S. Borg, *Chem. Commun.*, 1999, 22, 2285–2286.
- 34 M. Schmidt, S. M. Contakes and T. B. Rauchfuss, *J. Am. Chem. Soc.*, 1999, **121**, 9736–9737.
- 35 L.-C. Song, Z.-Y. Yang, Y.-J. Hua, H.-T. Wang, Y. Liu and Q.-M. Hu, *Organometallics*, 2007, **26**, 2106–2110.
- 36 J. Windhager, H. Görls, H. Petzold, G. Mloston, G. Linti and W. Walgand, Fur. J. Joseph Cham. 2007, 2007, 4462, 4471
- Weigand, *Eur. J. Inorg. Chem.*, 2007, **2007**, 4462–4471. 37 H. Li and T. B. Rauchfuss, *J. Am. Chem. Soc.*, 2002, **124**, 726-
- 727. 38 L.-C. Song, Z.-Y. Yang, H.-Z. Bian, Y. Liu, H.-T. Wang, X.-F. Liu
- and Q.-M. Hu, Organometallics, 2005, 24, 6126–6135.
  B. J. Lemon and J. W. Peters, Biochemistry (Mosc.), 1999, 38, 12969–12973.
- 40 M. Winkler, J. Esselborn and T. Happe, *Biochim. Biophys. Acta BBA Bioenerg.*, 2013, **1827**, 974–985.

- 41 S. C. Lovell, I. W. Davis, W. B. Arendall, P. I. W. de Bakker, J. M. Word, M. G. Prisant, J. S. Richardson and D. C. Richardson, *Proteins Struct. Funct. Bioinforma.*, 2003, **50**, 437–450.
- 42 C. Lambertz, P. Chernev, K. Klingan, N. Leidel, K. G. V. Sigfridsson, T. Happe and M. Haumann, *Chem. Sci.*, 2014, **5**, 1187–1203.
- 43 T. Liu and M. Y. Darensbourg, *J. Am. Chem. Soc.*, 2007, **129**, 7008–7009.
- 44 V. Fourmond, C. Greco, K. Sybirna, C. Baffert, P.-H. Wang, P. Ezanno, M. Montefiori, M. Bruschi, I. Meynial-Salles, P. Soucaille, J. Blumberger, H. Bottin, L. De Gioia and C. Léger, *Nat. Chem.*, 2014, 6, 336–342.
- 45 A. Adamska-Venkatesh, T. R. Simmons, J. F. Siebel, V. Artero, M. Fontecave, E. Reijerse and W. Lubitz, *Phys. Chem. Chem. Phys.*, 2015, **17**, 5421–5430.
- 46 M. K. Akhtar and P. R. Jones, *Appl. Microbiol. Biotechnol.*, 2008, **78**, 853–862.
- 47 J. M. Kuchenreuther, C. S. Grady-Smith, A. S. Bingham, S. J. George, S. P. Cramer and J. R. Swartz, *PLoS ONE*, 2010, **5**.
- 48 G. von Abendroth, S. Stripp, A. Silakov, C. Croux, P. Soucaille, L. Girbal and T. Happe, *Int. J. Hydrog. Energy*, 2008, 33, 6076–6081.

- 49 A. Hemschemeier, A. Melis and T. Happe, *Photosynth. Res.*, 2009, **102**, 523–540.
- 50 Z. Otwinowski and W. Minor, *Macromol. Crystallogr. Pt A*, 1997, **276**, 307–326.
- 51 W. Kabsch, Acta Crystallogr. D Biol. Crystallogr., 2010, 66, 125–132.
- 52 M. D. Winn, C. C. Ballard, K. D. Cowtan, E. J. Dodson, P
  Emsley, P. R. Evans, R. M. Keegan, E. B. Krissinel, A. G. W.
  Leslie, A. McCoy, S. J. McNicholas, G. N. Murshudov, N. S.
  Pannu, E. A. Potterton, H. R. Powell, R. J. Read, A. Vagin and
  K. S. Wilson, *Acta Crystallogr. D Biol. Crystallogr.*, 2011, 67,
  235−242.
- 53 P. D. Adams, P. V. Afonine, G. Bunkóczi, V. B. Chen, I. W. Davis, N. Echols, J. J. Headd, L.-W. Hung, G. J. Kapral, R. W. Grosse-Kunstleve, A. J. McCoy, N. W. Moriarty, R. Oeffner, R. J. Read, D. C. Richardson, J. S. Richardson, T. C. Terwilliger and P. H. Zwart, Acta Crystallogr. D Biol. Crystallogr., 2010, 66, 213– 221.
- 54 P. Emsley, B. Lohkamp, W. G. Scott and K. Cowtan, *Acta Crystallogr. D Biol. Crystallogr.*, 2010, **66**, 486–501.
- 55 P. A. Karplus and K. Diederichs, *Science*, 2012, **336**, 1030-1033