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## ARTICLE

# Gold(I)-catalyzed [2+2+2] Cycloaddition of Allenamides, Alkenes and Aldehydes: A Straightforward Approach to Tetrahydropyrans

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Allenamides participate as two-carbon components in an intermolecular [2+2+2] cycloaddition with alkenes and aldehydes when treated with catalytic amounts of a phosphite gold complex. The reaction is highly regio- and chemoselective, and works with different types of alkenes, including styrenes, enol ethers or enamides, as well as with aromatic and aliphatic aldehydes. Accordingly, different types of 2,6-disubstituted tetrahydropyrans can be stereoselectively assembled in a single step from commercial or very accessible starting materials.

## Introduction

Transition metal catalyzed [2 + 2 + 2] cycloadditions constitute one of the most attractive methodologies for the construction of six-membered cyclic systems.<sup>1</sup> Despite the significant achievements reported in this field, intermolecular examples involving three different cycloaddition partners are extremely scarce, most probably, because of the chemo- and regioselectivity issues associated with these multicomponent annulations.<sup>2</sup> The few examples so far reported involve the use of Rh, Ru, Nb or Ni catalysts and, at least, one alkyne as cycloaddition component.<sup>2</sup> Curiously, and despite gold catalysis has proven to be very efficient for unveiling novel types of cycloadditions,<sup>3</sup> fully intermolecular [2 + 2 + 2] examples are almost unknown<sup>4</sup> and, to the best of our knowledge, those of three different two-atom components are unprecedented.<sup>5</sup>

Herein, we are pleased to report a fully intermolecular gold-catalyzed [2 + 2 + 2] cycloaddition involving three different  $\pi$ -unsaturated components, namely an allene, an alkene and an aldehyde. The reaction takes place with excellent chemo- and regioselectivity and provides a straightforward and atom-economical entry to tetrahydropyrans (THPs). THPs, and in particular the 2,6-disubstituted counterparts, are privileged scaffolds that are present in a myriad of biologically active molecules (Figure 1).<sup>6</sup> Although many elegant methods have been developed to construct these motifs,<sup>6,7</sup> none of them encompasses the coupling of three readily available components in a single catalytic annulation step.<sup>8</sup>

During the last years, we have developed different types of Au-catalyzed annulations,<sup>9</sup> including a cycloaddition between allenamides and oxoalkenes that affords oxabridged medium-sized carbocycles (Scheme 1, eq 1).<sup>10,11</sup> This annulation was proposed to proceed through the intermediate **I**,<sup>12</sup> which evolves to the product by the sequential formation of species **II** and **III**. On this basis, we then wondered whether it would be

possible to achieve an annulation between the allenamide, alkene and carbonyl units in a fully intermolecular way, a process that would directly afford 2,6-disubstituted THPs like **4** (Scheme 1, eq 2). Despite the process could be viewed as an intermolecular version of the previous annulation, the timely assembly of three different components in a programmed manner is extremely challenging. Indeed, the feasibility of the reaction could be seriously compromised since more simple [2 + 2] adducts of type **5** and **6**,<sup>9c</sup> acyclic products like **7**, or alternative [2+2+2] adducts (**8** / **9**) could be likewise expected.<sup>13</sup>

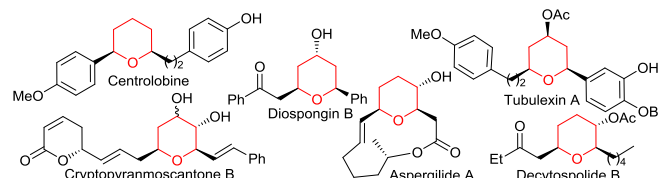
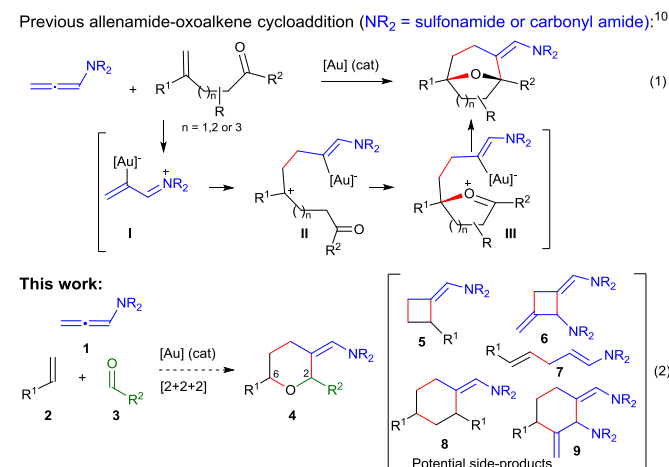


Figure 1 Tetrahydropyran frameworks in biologically active products.

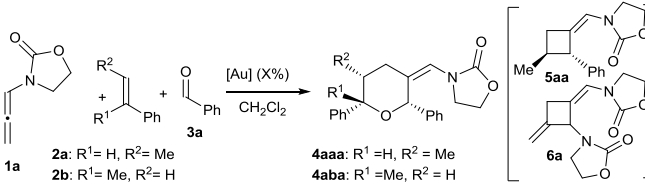


Scheme 1 Previous gold-catalyzed cascade cycloadditions and current proposal.

## Results and discussion

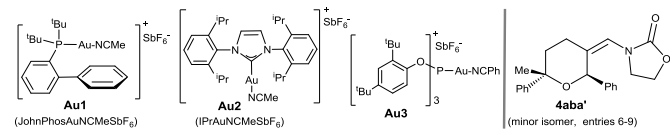
We began our studies by analyzing the reactivity of allenamide **1a** with (*E*)- $\beta$ -methylstyrene (**2a**) and benzaldehyde (**3a**) (Table 1). Initial assays confirmed the expected difficulties for controlling the chemoselectivity of the process. Indeed, despite using an excess of the aldehyde (10 equiv), and adding the allenamide over 2 hours, the gold complex **Au1** induced the formation of the [2 + 2] allenamide dimerization adduct **6a** in 44% yield, together with a minor amount of the cyclobutane **5aa**,<sup>8b</sup> resulting from the [2 + 2] cycloaddition between **1a** and **2a** (entry 1). A [2 + 2 + 2] adduct, eventually identified as the 2,6-*cis* THP **4aaa**, was also detected, but only in trace amounts. Similarly, other frequently used gold catalysts such as Ph<sub>3</sub>PAuNTf<sub>2</sub> or the *NHC*-gold complex **Au2** provided very low yields of the [2 + 2 + 2] adduct **4aaa** (entries 2 and 3), with poor mass recovery balances in all these cases. Interestingly, when using the phosphite-gold complex **Au3**, we observed a significant increase in the global yield of the reaction, which provided **5aa** in 60% yield along with the [2 + 2 + 2] adduct **4aaa** in 21% yield (entry 4). This last yield could be further improved up to 35% by carrying out the reaction at -45 °C (entry 5).<sup>14</sup>

**Table 1** Preliminary evaluation of the [2+2+2] cycloaddition.<sup>a,b</sup>



entry	[Au] (mol %)	<b>2</b>	R <sup>1</sup>	R <sup>2</sup>	Conv.	<b>4</b> (%)	<b>5</b> (%)	<b>6</b> (%)
1	<b>Au1</b> (5%)	<b>2a</b>	H	Me	99%	<b>4aaa</b> , 2	<b>5aa</b> , 4	<b>6a</b> , 44
2	Ph <sub>3</sub> PAuNTf <sub>2</sub> (5%)	<b>2a</b>	H	Me	60%	<b>4aaa</b> , 2	<b>5aa</b> , 0	<b>6a</b> , 7
3	<b>Au2</b> (5%)	<b>2a</b>	H	Me	99%	<b>4aaa</b> , 15	<b>5aa</b> , 7	<b>6a</b> , 22
4	<b>Au3</b> (2%)	<b>2a</b>	H	Me	99%	<b>4aaa</b> , 21	<b>5aa</b> , 60	<b>6a</b> , 8
5 <sup>c</sup>	<b>Au3</b> (2%)	<b>2a</b>	H	Me	99%	<b>4aaa</b> , 35	<b>5aa</b> , 37	–
6	<b>Au3</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 98 <sup>d</sup>	–	–
7 <sup>e</sup>	<b>Au3</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 99 <sup>d</sup>	–	–
8	<b>Au1</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 51 <sup>d</sup>	<b>5ab</b> , 17	–
9	Ph <sub>3</sub> PAuNTf <sub>2</sub> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 77 <sup>f</sup>	<b>5ab</b> , 14	–
10	<b>Au2</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 80 <sup>d</sup>	<b>5ab</b> , 6	–
11 <sup>e,g</sup>	<b>Au3</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 98 <sup>h</sup>	–	–
12 <sup>e,i</sup>	<b>Au3</b> (2%)	<b>2b</b>	Me	H	99%	<b>4aba</b> , 98 <sup>j</sup>	–	–

<sup>a</sup> **1a** (1 equiv) added over 2 h to a solution of **2** (2 equiv), **3a** (10 equiv), [Au] (X mol%) and 4 Å MS, in CH<sub>2</sub>Cl<sub>2</sub> at -15 °C, unless otherwise noted. <sup>b</sup> Conversion of **1a** and yields of **4–6** determined by <sup>1</sup>H-NMR of the crude mixture using 1,3,5-(MeO)<sub>3</sub>C<sub>6</sub>H<sub>3</sub> as internal standard (IS). <sup>c</sup> Carried out at -45 °C, (1 h). <sup>d</sup> *dr* (2,6-*cis* : *trans*) = 2 : 1; the major isomer is that drawn. <sup>e</sup> **1a** added in one portion. <sup>f</sup> *dr* 1.5 : 1; <sup>g</sup> Carried out at -78 °C, (1 h). <sup>h</sup> 90% isolated yield, *dr* 3.4:1 (**4aba** : **4aba'**). <sup>i</sup> Carried out in F<sub>3</sub>C-Ph at -25 °C (4 h) <sup>j</sup> 86% isolated yield, *dr* 4.5:1.



At this point, we envisioned that an additional stabilization of the putative carbocationic species of type **II**, resulting from the addition of the alkene to intermediate **I** (Scheme 1), could eventually facilitate its intermolecular capture by the aldehyde. In consonance with this hypothesis, we were pleased to find that the use of  $\alpha$ -methylstyrene (**2b**) instead of  $\beta$ -methylstyrene (**2a**) provided, under otherwise identical conditions, the desired THP in an excellent 98% yield, as a 2:1 mixture of 2,6-*cis* (**4aba**) and 2,6-*trans* (**4aba'**) diastereoisomers (entry 6).<sup>15</sup> The same result was obtained when **1a** was added in one portion (entry 7). Gold catalysts such as JohnPhosAuNCMeSbF<sub>6</sub> (**Au1**), Ph<sub>3</sub>PAuNTf<sub>2</sub> or IPrAuNCMeSbF<sub>6</sub> (**Au2**), also provided the desired [2 + 2 + 2] cycloadduct **5aba** as the major adduct; however, yields and chemoselectivities were significantly lower to those obtained with the phosphite-gold catalyst **Au3** (entry 7 vs 8-10). Moreover, with this latter catalyst the diastereoselectivity could be improved by either performing the reaction at -78 °C (*dr* 3.5:1, 90% isolated yield, entry 11) or by using  $\alpha,\alpha,\alpha$ -trifluorotoluene as solvent (*dr* 4.5:1, 86% yield, entry 12).

With these results in hand, we next analyzed the scope of the process (Table 2). In consonance with the performance of  $\beta$ -methylstyrene (**2a**, Table 1, entry 5), the cycloaddition of styrene (**2c**) with **1a** and benzaldehyde provided the desired 2,6-disubstituted THP (**4aca**) in a moderate 37% yield, but with complete 2,6-*cis* selectivity (**5ac** was also isolated in 46% yield). Gratifyingly, use of styrenes with an electron-donating groups (e.g. *p*-MeO or *o*-MeO) allowed to significantly improve the chemoselectivity, so the corresponding THPs, **4ada** and **4aea**, were isolated in good yields (60-65% yield) and with complete 2,6-*cis* diastereoselectivity.

On the other hand, the cycloaddition with  $\alpha$ -phenylstyrene provided the desired THP (**4afa**) in an excellent 86% yield, whereas the use *exo*-methylenes such as 1-methylene-tetrahydronaphthalene allowed an efficient access to spiro-tetrahydropyran derivatives like **4aga**, which was isolated in an excellent 94% yield (*dr* 1.5:1).

Remarkably, cyclic alkene derivatives were also excellent partners for this process. Thus, the cycloadditions of allenamide **1a** and benzaldehyde with 1-phenylcyclohexene, 4-methyl-1,2-dihydronaphthalene or 3-methyl-1H-indene, provided the corresponding THPs (**4aha–4aja**) in good yields (57 – 84% yield) and moderate (**4aha**) or complete (**4aia–aja**) stereoselectivity.<sup>16</sup> X-ray analysis of crystals of **4aha** and **4aia** unambiguously confirmed their structures and relative stereochemistry (Figure 2).<sup>17</sup>

We next explored the use of alternative electron-rich alkenes. Gratifyingly, the cycloaddition could also be performed with enol ethers such as 2-methoxyprop-1-ene or ethoxyethene, to obtain the corresponding cyclic acetals (**4aka–4ala**) with moderate to good yields. Similarly, the cycloaddition between **1a**, **3a** and 1-vinylpyrrolidin-2-one was also feasible, providing the cyclic hemiaminal ether **4ama** in 45% yield and with complete diastereoselectivity.

These annulations are also feasible with other allenamides. For instance, the reaction of  $\gamma$ -methyl-substituted allenamide **1b**

(see Table 2, footnote) with  $\alpha$ -methylstyrene and benzaldehyde provided the [2 + 2 + 2] adduct **4bba**, featuring three new stereogenic centers, in 70% yield and with excellent diastereoselectivity (*dr* 11 : 1).<sup>18</sup> On the other hand, *N*-tosylphenyl allenamides such **1c** were also suitable partners. Thus, the [2 + 2 + 2] adduct **4cfa**, resulting from the cycloaddition of **1c**, benzaldehyde and  $\alpha$ -phenylstyrene was obtained in 77% yield, whereas the adduct **4cka**, from 2-methoxyprop-1-ene, was obtained in 84% yield and, importantly, with complete 2,6-*cis* stereoselectivity.

Remarkably, the scope of the method is not limited to benzaldehyde. Indeed, the reaction of  $\alpha$ -methylstyrene, allenamide **1a** and an aliphatic aldehyde such as pentanal led to the desired adduct, **4abb**, in 97% yield (*dr* 3:1). Other aldehydes such as isobutyraldehyde, cyclopropanecarbaldehyde or pent-4-enal also gave the THPs **4abc-4abe** in excellent yields.  $\alpha,\beta$ -Unsaturated aldehydes such as 2-methylbut-2-enal or methacrolein (**3g**) also participated in the annulation yielding the desired THPs (**4abf**, **4aff**, **4afg**) in yields above 90%. Moreover, the cycloaddition of the  $\gamma$ -substituted allenamide **1b** with an aliphatic aldehyde such as 2-methylbut-2-enal was also feasible, providing **4bbf** in 84% yield (*dr* 10 : 2 : 1).<sup>14</sup>

Overall, it is important to highlight that the current method constitutes one of the very few catalytic approaches that affords THPs featuring fully substituted carbons at the oxygen-adjacent position (e.g. C6).<sup>19</sup> On the other hand, while the above reactions were carried out using a relatively large excess of the aldehyde, gratifyingly, we found that in most of the cases the reaction can be efficiently performed using an allenamide (**1**) / alkene (**2**) / aldehyde (**3**) molar ratio of 1 / 1.2 / 2 (Table 2, footnote *b*, results under parenthesis). Thus, using these conditions, the THPs **4aba**, **4afa**, **4aga**, **4ala**, **4abb**, **4abc**, **4afd**, **4abe**, **4abf** or **4afg** were obtained in yields varying from 60% to 90% (Table 2).<sup>20</sup> Additionally, more complex polycyclic systems like **4aha-4aja** could also be obtained in yields from 45% to 68%.<sup>14</sup>

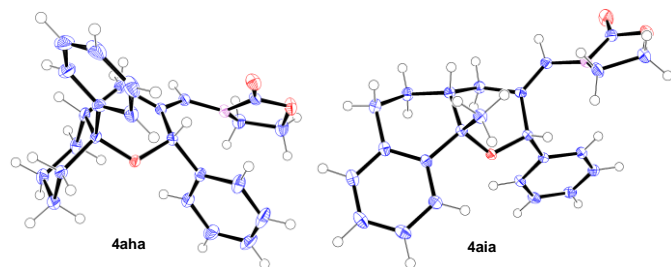
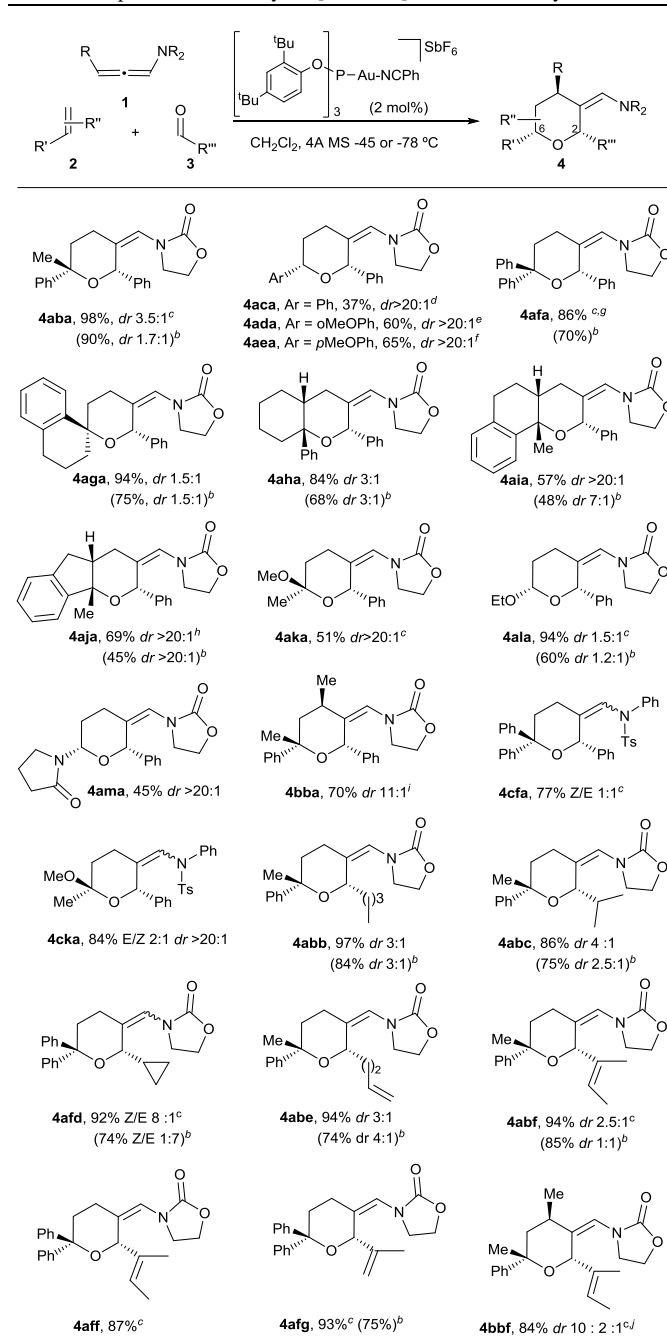


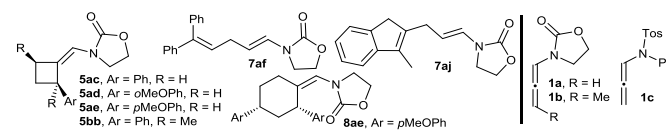
Figure 2 X-Ray structures of **4aha** (left, major isomer) and **4aia** (right).<sup>14</sup>

We next explored some manipulations of the *exo*-enamide moiety of the products (Scheme 2). Thus, THPs like **4afd** or **4aea** can be dihydroxylated to afford the  $\alpha$ -hydroxy aldehydes **11** and **12** in excellent yields and with very good or complete diastereoselectivity (Scheme 2, eq. 1). Moreover, both types of enamides (e.g. **4afd** and **4cfa**) could be easily converted into their corresponding ketones upon ozonolysis (Scheme 2, eq. 2).

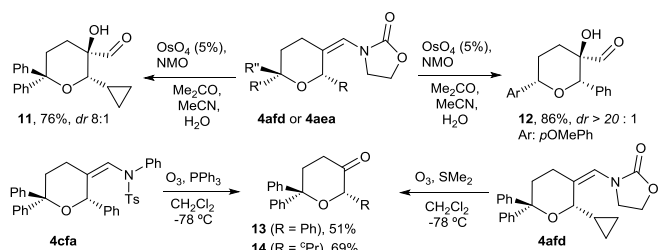
Table 2 Scope of the Au-catalyzed [2 + 2 + 2] intermolecular cycloaddition<sup>a,b</sup>



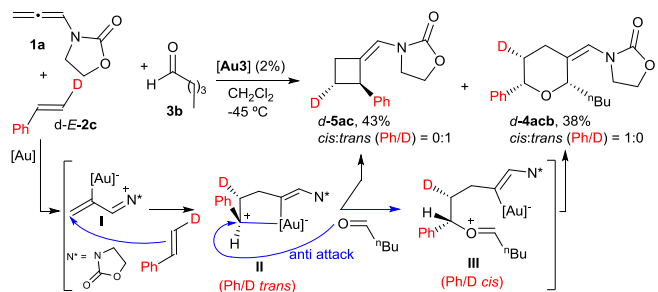
<sup>a</sup> **1** (1 equiv) added to a solution of **2** (2 equiv), aldehyde (10 equiv), [**Au3**] (2 mol%) and 4Å MS, in CH<sub>2</sub>Cl<sub>2</sub> at -45 °C, unless otherwise noted. Conversions > 99% (<sup>1</sup>H-NMR). When a mixture of 2,6-isomers is formed, the major is that drawn. <sup>b</sup> Carried out at -45 °C with a **1** / **2** / **3** molar ratio of 1 / 1.25 / 2. <sup>c</sup> Carried out at -78 °C. <sup>d</sup> 45% of **5ac** was also isolated. <sup>e</sup> 21% of **5ad** was also isolated. <sup>f</sup> Traces of **5ae** and **5af** (5% yield) were also isolated. <sup>g</sup> 5% yield of **7af** was also isolated. <sup>h</sup> 5% yield of **7aj** was also isolated. <sup>i</sup> 17% yield of **5bb** was also isolated. <sup>j</sup> For the structure of the minor isomers, see the ESI.





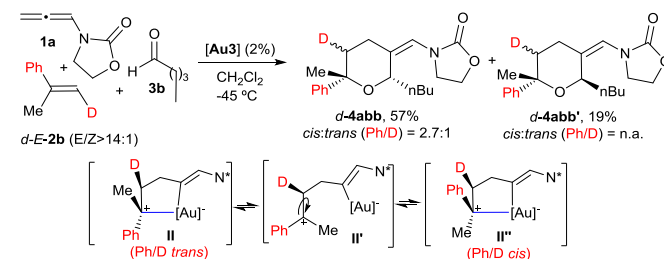
Scheme 2 Functionalization of the *exo*-enamide moiety.

With regard to the mechanism of the annulation, the general proposal indicated in Scheme 1 could also apply for this intermolecular process; however, we found some results that were indicative of a more complex scenario. In particular, it is curious that while the [2 + 2] product (**5aa**) obtained from **1a** and *E*- $\beta$ -methylstyrene retains the *trans* configuration of the alkene, the [2 + 2 + 2] adduct **4aaa**, displays these groups in a *cis* disposition (Table 1, entry 5). Contrary, polycyclic [2 + 2 + 2] adducts like **4aha**, **4aia** or **4aja** retained the configuration of the parent alkene. To shed light on this divergence, we carried out the cycloaddition of **1a** and pentanal with the *trans*-deuterated styrene *d*-**E-2c** (Scheme 3). As expected, the reaction provided a mixture of the [2 + 2 + 2] and [2 + 2] adducts *d*-**4acb** (38% yield) and *d*-**5ac** (43% yield), respectively. Interestingly, *d*-**5ac** incorporates the Ph and the deuterium atom in a *trans* disposition, whereas the [2 + 2 + 2] adduct, *d*-**4acb**, holds these groups in a *cis* arrangement. These results strongly suggest the formation of an intermediate of type **II** that preserves the stereochemical information of the alkene due to an stabilizing electrostatic interaction between the gold atom and the benzylic carbocation (Scheme 4).<sup>21</sup> A subsequent nucleophilic *anti* attack of the carbonyl moiety would lead intermediate **III** and, eventually, to the product *d*-**4acb**. The preferential formation of this THP with the C2 and C6 substituents in *cis* disposition, is in agreement with a transition state that places these groups in equatorial disposition (*Prins*-like cyclization from **III** to **4**). On the other hand, if species **II** collapses to render a [2 + 2] adduct, the Ph and the D atom would retain the initial *trans* arrangement, as observed in *d*-**5ac**.

Scheme 3 Cycloaddition of *d*-**E-2c** and the proposed key intermediate **II**.

We also analysed the cycloaddition with deuterated  $\alpha$ -methylstyrene (*d*-**2b**) as a model for  $\alpha$ -substituted alkenes (Scheme 4). Curiously, the expected [2 + 2 + 2] isomeric adducts *d*-**4abb** and *d*-**4abb'** were obtained as mixtures of

*cis/trans* (Ph/D) isomers. Accordingly, an acyclic carbocation species like **II'** or, alternatively, a fast equilibrium between the Ph/D-*trans* and *cis* intermediates **II** and **II''**, could account for this result.<sup>22,23</sup> Considering this proposal, the exclusive formation of the *cis*-fused polycyclic THPs **4aha**-**4aja**, from cyclic alkene precursors can be easily understood.

Scheme 4 Cycloaddition of *d*-**E-2b** and the proposed key intermediate **II'**.

Finally, we carried out the above cycloadditions of Schemes 4 and 5 using the *NHC*-gold catalyst **Au2**, instead of **Au3**. Not unexpectedly, lower chemoselectivities and yields of the corresponding [2 + 2 + 2] adducts were obtained in both cases but, interestingly, the stereochemistry of each deuterated cycloadduct (*d*-**4acb**, *d*-**5ac** and *d*-**4abb**), turned out to be identical to that obtained with **Au3**.<sup>14</sup> Thus, the  $\sigma$ -donor or  $\pi$ -acceptor characteristics of the ligand at gold do not seem to significantly affect the nature of the intermediate of type **II**.

## Conclusions

In summary, we have developed a gold-catalyzed fully intermolecular [2 + 2 + 2] cycloaddition that constitutes one of the few transition metal catalyzed annulations involving three different  $\pi$ -unsaturated components. The process shows a high scope with regard to the alkenes and aldehydes that can be used, and provides an efficient, atom-economical and stereoselective access to a variety of 2,6-disubstituted THPs from very accessible or even commercially available materials.

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## Notes and references

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† HF and IV equally contributed to this work. Electronic Supplementary Information (ESI) available: [Characterization data and experimental procedures]. See DOI: 10.1039/b000000x/

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- 15 (a) The [2 + 2] adducts **5ab**, **6a**, or other side-products, were not detected in the crude mixture (<sup>1</sup>H-NMR); (b) The structure and relative stereochemistry of both THP isomers (**4aba** / **4aba'**) were established by NMR and, additionally, those of the major isomer (**4aba**), with the Ph groups in *cis*, were further confirmed by X-ray (CCDC 1038447).<sup>14</sup>
- 16 The ring fusion was exclusively *cis* in all these polycyclic systems. Therefore, *dr* refers to the substituents at the 2,6-THP positions.
- 17 CCDC 1038448 (for **4aba**) and CCDC 1038449 (for **4aia**).<sup>14</sup>
- 18 An  $\alpha$ -alkyl-substituted allenamides such as 3-(buta-2,3-dien-2-yl)oxazolidin-2-one provided a complex mixture of products.
- 19 (a) Previously reported approaches are essentially limited to the formation of THPs with monosubstituted C2 or C6 carbons,<sup>6a</sup> something that significantly facilitates a high stereoselection; For isolated catalytic examples yielding products with fully substituted-C2 or C6 carbons, see: (b) M. Jacolot, M. Jean, N. Levoine and P. van de Weghe, *Org. Lett.*, 2012, **14**, 58; (c) M. P. Castaldi, D. M. Troast and J. A. Porco Jr, *Org. Lett.*, 2009, **11**, 3362; (d) J. S. Yadav, B. V. Subba Reddy, G. G. K. S. Narayana Kumar and S. Aravind *Synthesis*, 2008, 395; (e) When a diastereoisomeric mixture is formed (Table 2) the 2,6-*cis* and *trans* isomers could be usually separated by standard chromatography.<sup>14</sup>
- 20 **4aba** can even be obtained using a 1:1:1 ratio in 86% yield (*dr* 1.8:1).
- 21 (a) A. Z. Gonzalez, D. Benitez, E. Tkatchouk, W. A. Goddard and F. D. Toste, *J. Am. Chem. Soc.*, 2011, **133**, 5500; (b) See also ref 12a.
- 22 The preferential formation of the THP **4abb**, is also in agreement with a preferred *Prins*-like transition state that holds the bulkier groups at C2 and C6 in equatorial disposition.
- 23 Similarly, the cycloaddition of an electron-rich styrene such as *trans*-deuterated *p*-methoxystyrene (*d-E-2e*) with **1a** and benzaldehyde provided *d-4aea* (64% yield) and traces of the [2 + 2] adduct *d-5ae*, both as almost equimolar mixtures of *cis* and *trans* (*pMeOPh* / *D*) isomers.<sup>14</sup> Thus, an intermediate of type **II'** (or an equilibrium between the *cis* and *trans* cyclic isomers **II**, Scheme 4) might also operate in this case.

