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Formal [3+3] Annulation of Isatin-derived 2-Bromoenals with 1, 3-Dicarbonyl Compounds Enabled by Lewis Acid/N-Heterocyclic Carbene Cooperative Catalysis

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Abstract A series of novel isatin-derived 2-bromoenals were synthesized and applied for formal [3+3] annulation with 1,3-dicarbonyl compounds enabled by a NHC/Lewis acid cooperative catalysis strategy. This newly developed methodology offers rapid access to functionalized spirooxindole δ -lactones. The newly synthesized isatin-derived 2-bromoenals may be further used as potential electrophilic 1,3-synthons for diversity-oriented synthesis of spirooxindoles.

Over the past two decades, N-heterocyclic carbene (NHC)-catalyzed umpolung (polarity reversal) of aldehydes opened up a new and unique area of unconventional formation of carbon-carbon and carbon-heteroatom bonds *via* various reactive intermediates.¹ Among these intermediates, nucleophilic acyl anion intermediates² and homoenolate equivalents³ generated with NHC catalysis have drawn immense attention. Recently, α , β -unsaturated acyl azoliums **I** have been used as novel and versatile electrophilic 1,3-synthons for the synthesis of various heterocyclic compounds like dihydropyranones and dihydropyridinones since its first discovery in 2006.⁴ Up to now, α , β -unsaturated acyl azoliums **I** can be commonly generated from six types of precursors including ynals,⁵ enals under external oxidative conditions,^{5f, 6} 2-haloenals,⁷ α , β -unsaturated esters⁸ or acyl fluorides,⁹ and *in situ* genereated mixed α , β -unsaturated anhydrides from their corresponding carboxylic acids¹⁰ (Scheme 1).



Scheme 1. Generation of α , β -unsaturated acyl azoliums from various precursors.

Spirooxindoles have emerged as attractive privileged structures and become important synthetic targets due to their prevalence in numerous natural products and synthetic compounds with diverse biological activities. However, the architecture of spirooxindole frameworks has always been a challenging endeavor for organic chemists because it often requires specific designed strategies to install the carbocyclic or heterocyclic moeities at the C3 position of the oxindole core to form a spiro quaternary carboncenter. Therefore, the development of simple and versatile substrates or synthetic approaches for diversity-oriented construction of spirooxindole skeletons is highly desirable.

As a continuation of our long-lasting goal to discover novel methodologies for rapid access to polycyclic indole derivatives,^{3m, 5b, 5f, 7g, 11} we reasoned that isatin-derived α,β -unsaturated acyl azoliums **II** generated from diverse precursors under NHC catalysis can be utilized as versatile electrophilic 1,3-synthons to combine with different bisnucleophiles for divergent synthesis of skeletally diverse spirooxindoles (Scheme 2). During our research on the generation and applications of isatin-derived α,β -unsaturated acyl azoliums **II**, Zhong and co-workers^{6h} very recently reported an oxidative NHC-catalyzed formal [3+3] annulation of imines with isatin-derived enals **3** for rapid access to spirooxindole **4** (Scheme 2). In this reaction, *Z*-isomers **3** are separated from the *Z/E* mixtures **2** which are prepared from substituted isatins **1** within 2 steps using known procedures. However, it is always difficult and timeconsuming to separate *Z*-isomers **3** from their *Z/E* mixtures **2** which are obtained with low stereoselectivity in most cases. On the other hand, enals **3** are

unstable under air, so they should be kept under an inert atmosphere and be used as soon as possible. Since α -bromoenals were reported to be more stable than their corresponding enals and were frequently used as α,β -unsaturated acyl azolium precursors, we tried to synthesize isatin-derived 2-bromoenals 5 via electrophilic addition of Br_2 to Z/E mixtures 2 followed by elimination with a base as a one-pot procedure. Gratifying, introduction of the bromine atom to the α -position of the enals 2 highly enhances the stereoselectivity which may be attributed to the steric effect of more hindered bromine atom. E-isomers 5 were formed as the major products (E/Z =9:1 to >20:1) and proved to be more stable under air (Scheme 2). The pure E-isomers 5 can be easily obtained by recrystallization of the crude products. The structure and stereochemistry of E-isomers 5 were established by spectroscopic analysis and further confirmed by X-ray crystallography of 5d. In order to test the reactivity of the isatin-derived 2-bromoenals 5, 1,3-dicarbonyl compounds 6, good 1,3-bisnucleophiles well-established for formal [3+3] annulations with α , β -unsaturated acyl azoliums I, were used to react with substrates 5 under NHC/base conditions to deliver desired spirooxindole δ -lactones 7 (Scheme 2). Herein, we wish to report the results.



Scheme 2. Generation and applications of isatin-derived α , β -unsaturated acyl azoliums **II**. We commenced the optimization studies using isatin-derived 2–bromoenal **5a** and ethyl acetoacetate **6a** as model substrates (Table 1). The initial screening of various

carbene precursors **A-F** was discourging as only precatalyst **F** afforded the desired product **7a** in less than 10% yield (entries 1-3). Further examination of a variety of bases and solvents using **F** as the precatalyst revealed that the yield was improved to 42% when the reaction was performed in 1,4-dioxane employing Cs₂CO₃ as a base (entry 12). In recent years, Lewis acids have proved to coordinate with the substrates effectively and thus facilitate the reactions resulting enhanced reaction yields as well as stereoselectivity.^{6f, 6m, 11-12} Inspired by the findings, we tried several commonly used Lewis acids and found that LiCl could enhance the reaction yield to 63% (entry 15). Based on the above observations, the optimal reaction conditions were finally established as the reaction carried out in 1,4-dioxane using Cs₂CO₃ (1.5 equiv) as a base in the presence of 15 mol% of **F** and 1.1equiv of LiCl. **Table 1.** Optimization of the reaction conditions ^{*a*}



Entry	Catalyst	Base	Solvent	Additive	Yield ^b
1	A , B	DBU	THF	None	0
2	C-E	DBU	THF	None	trace
3	\mathbf{F}	DBU	THF	None	<10
4	\mathbf{F}	K_2CO_3	THF	None	23
5	\mathbf{F}	Et ₃ N	THF	None	<10
6	\mathbf{F}	DIPEA	THF	None	21
7	F	<i>t</i> BuOK	THF	None	16
8	F	Cs_2CO_3	THF	None	29
9	F	Cs_2CO_3	DCM	None	11
10	F	Cs_2CO_3	CH ₃ CN	None	trace
11	F	Cs_2CO_3	PhMe	None	<10
12	F	Cs_2CO_3	1,4-dioxane	None	42
13	F	Cs_2CO_3	1,4-dioxane	$\operatorname{Ti}(\operatorname{OPr})_4^c$	20
14	F	Cs ₂ CO ₃	1,4-dioxane	$Yb(OTf)_3^d$	28

15	F	Cs ₂ CO ₃	1,4-dioxane	LiCl ^c	63
^[a] All r	reactions were	e performed in a sealed t	ube on a 0.2 mmol	scale with 1.0 equiv	v of 5a , 2.0 equiv of
6a , 1	15mol% of a	carbene precursor, 1.5	equiv of a base an	d 200 mg of 4 Å N	AS in an anhydrous
solve	ent (3 mL) at	50 °C for 2 h under N ₂ .			
lbl + +					

^[b] Isolated yields based on **5a**.

^[c] 1.1 equiv of a Lewis acid was used.

^[d] 20 mol% of Lewis acid was used. DBU = 1,8-diazabicyclo[5.4.0]-undec-7-ene; Mes = $2,4,6-(CH_3)_3C_6H_2$; DIPEA = N,N-diisopropylethylamine.

After establishing the optimal reaction conditions, we carried on to explore the scope of the reaction between various isatin-derived 2-bromoenals **5** with 1,3-dicarbonyl compounds **6** (Table 2). First, a variety of 1,3-dicarbonyl compounds were tested. The β -keto esters **6b-f** with diverse substituted phenyl group underwent efficient annulation to afford products **7b-f** in moderate to good yields (entries 2-6). The β -diketones **6g-i** were also found applicable to this reaction (entries 7-9). However, cyclohexane-1,3-dione **6j** did not work for this reaction (entry 10). Subsequently, isatin-derived 2-bromoenals **5** with diverse substituents on the phenyl ring were further examined using 1-phenylbutane-1,3-dione **6h** as the model substrate (entries 11-15). 2-Bromoenals **5b-f** with electron-donating and electron-withdrawing groups on the phenyl ring were well tolerated to the reaction. Particularly, substrate **5e** with 5-methyl group afforded the product **7m** in 95% yield (entry 14).

I	Br O H Br H H S Bn	$\begin{array}{c} F (15 \text{ mol\%}) \\ Cs_2CO_3 (1.5 \text{ equiv}) \\ Cs_2CO_3 (1.5 \text{ equiv}) \\ \hline Cs_2CO_3 (1.5 \text{ equiv}) \\ \hline LiCl (1.1 \text{ equiv}) \\ \hline 1,4\text{-dioxane} \\ \hline 6 \\ 50 \ ^\circ C, N_2 \end{array}$		2 DR ³
Entry	R ¹ , 5	$R^2, R^3, 6$	Products	Yield $(\%)^b$
1	Н, 5а	Me, OEt, 6a	7a	63
2	Н, 5а	Ph, OEt, 6b	7b	57
3	Н, 5а	(4-OMe)Ph, OEt, 6c	7c	58
4	Н, 5а	Ph, OMe, 6d	7d	80
5	Н, 5а	(4-Me)Ph, OMe, 6e	7e	63
6	Н, 5а	(4-Cl)Ph, OMe, 6f	7f	67
7	Н, 5а	Me, Me, 6g	7g	54
8	Н, 5а	Ph, Me, 6h	7h	80
9	Н, 5а	Ph, Ph, 6i	7i	68
10	Н, 5а	cyclohexane-1,3-dione, 6j		0
11	5-F, 5b	Ph, Me, 6h	7j	62

 Table 2. The reaction scope^a

5-Cl, 5c	Ph, Me, 6h	7k	68	

13	5-Br, 5d	Ph, Me, 6h	71	45
14	5-Me, 5e	Ph, Me, 6h	7m	95
15	7-Cl, 5f	Ph, Me, 6h	7n	45

^a All reactions were performed in a sealed tube on a 0.2 mmol scale with 1.0 equiv of 5, 2.0 equiv of 6, 15mol% of F, 1.5 equiv of Cs₂CO₃, 1.1 equiv of LiCl and 200 mg of 4 Å MS in an anhydrous 1,4-dioxane (3 mL) at 50 °C for 2 h under N₂.

^b Isolated yields based on **5a**.

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Since β -naphthol^{5g} and enolizable carbonyl compounds^{7c} have been successfully applied to the formal [3+3] annulations with α , β -unsaturated acyl azoliums **I**, we tested the feasibility of the annulation between **5a** with β -naphthol and 3-oxo-3-phenylpropanenitrile respectively (Scheme 3). Unfortunately, neither of the substrates was found applicable to this methodology.



Scheme 3. The reactions of 2-bromoenal 5a with β -naphthol or 3-oxo-3-phenylpropanenitrile.

A preliminary enantioselective study of the reaction was then carried out using chiral carbene precursors **G** and **H**. However, the desired product **7a** was obtained in moderate yields but 0% e.e. value for both cases (Scheme 4).





The ring-opening reaction of the spirooxindole δ -lactones was also carried out (Scheme 5). Interesting, when compound **7k** was heated in MeOH, the ring-opening product **8** was obtained in 93% yield with the loss of an acetyl group.



Scheme 5. The ring-opening reaction of the product 7k.

A plausible catalytic cycle is proposed in Scheme 6. The combination of isatin-derived 2-bromoenal **5a** with NHC **F'** generated upon deprotonation of carbene precursor **F** with a base affords Breslow intermediate **9**. The a^3 - d^3 umpolung of **9** induces the formation of intermediate **10**, which is tautomerized to bromoacyl azolium **11**. The subsequent leaving of the bromide gives rise to the more stable (*E*)-isatin-derived α , β -unsaturated acyl azolium **12**. The stereochemistry of **12** was determined by trapping it with ethanol to only afford (*E*)- α , β -unsaturated ester **13** in 78% yield that is a known compound.¹³ Notably, even if a *E*/*Z* (9:1) mixture of **5a** was applied to this reaction, only *E*-ester **13** was obtained in similar yield. Michael addition of carbon-centered nucleophile **14** derived from **6a** under base/LiCl conditions to acyl azolium **12** generates intermedate **15**, which is followed by intramolecular cyclization to afford spirooxindole δ -lactones **7a** and regenerate NHC **F'**.



Scheme 6. The proposed mechanism.

In summary, a cooperative NHC/Lewis acid-mediated formal [3+3] annulation of various isatin-derived 2-bromoenals **5** with 1,3-dicarbonyl compounds **6** is described. This protocol offers rapid access to functionalized spirooxindole δ -lactones **7**. The newly synthesized isatin-derived 2-bromoenals **5** can be utilized as promising versatile electrophilic 1,3-synthons for divergent construction of diverse spirooxindole skeletons. An enantioselective synthetic protocol from other precursors as well as further applications of isatin-derived 2-bromoenals to the synthesis of diverse spirooxindoles are currently undergoing in our laboratory.

Experimental section

General procedure for the synthesis of 5 from 2: To the solution of compounds 2 (263 mg, 1.0 mmol) in CH₂Cl₂ (3-4 mL) was added Br₂ (62 μ L, 1.2 mmol). The resulting mixture was stirred at 5-10 °C for 15 mins followed by addition of Et₃N (235 μ L, 1.7 mmol). The mixture was further stirred at 5-10 °C until the completion of the reaction as monitored by TLC. Then saturated sodium carbonate aqueous solution (20 mL) was added and the resulting mixture was extracted with CH₂Cl₂ (3 x 10 mL). The organic phase was dried over anhydrous Na₂SO₄, and evaporated under reduced pressure. The residue was purified by chromatography on silica gel using hexane/EtOAc (3:1) as the eluent to afford products **5** as red solids.

General procedure for the synthesis of products 7: To an oven-dried 15 mL glass cylindrical pressure vessel was charged with 2-bromoenals 5 (0.2 mmol), 1,3-dicarbonyl compounds 6 (0.4 mmol), carbene precursor F (8 mg, 0.03 mmol), Cs_2CO_3 (98 mg, 0.3 mmol), LiCl (9.0 mg, 0.22 mmol) and 200 mg of 4 Å MS under N₂ atmosphere. Then anhydrous 1,4-dioxane (3 mL) was added and the vessel was immediately sealed tightly. The resulting mixture was stirred at 50 °C for 2 h. The mixture was cooled to room temperature. The solvent was evaporated under reduced pressure and the residue was purified by chromatography on silica gel to using hexane/EtOAc (5:1) as the eluent afford products 7.

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References

- For selected reviews, see: (a) D. M. Flanigan, F. Romanov-Michailidis, N. A. White and T. Rovis, *Chem. Rev.*, 2015, **115**, 9307; (b) J. Mahatthananchai and J. W. Bode, *Acc. Chem. Res.*, 2014, **47**, 696; (c) H. U. Vora and T. Rovis, *Aldrichimica Acta*, 2011, **44**, 3; (d) N. Marion, S. Diez-Gonzalez and S. P. Nolan, *Angew. Chem. Int. Ed.*, 2007, **46**, 2988; (e) A. Grossmann and D. Enders, *Angew. Chem. Int. Ed.*, 2012, **51**, 314; (f) X. Bugaut and F. Glorius, *Chem. Soc. Rev.*, 2012, **41**, 3511; (g) P. Dominguez de Maria and S. Shanmuganathan, *Curr. Org. Chem.*, 2011, **15**, 2083; (h) D. C. M. Albanese and N. Gaggero, *Eur. J. Org. Chem.*, 2014, 5631; (i) M. N. Hopkinson, C. Richter, M. Schedler and F. Glorius, *Nature*, 2014, **510**, 485; (j) J. Douglas, G. Churchill and A. D. Smith, *Synthesis*, 2012, **44**, 2295; (k) J. Izquierdo, G. E. Hutson, D. T. Cohen and K. A. Scheidt, *Angew. Chem. Int. Ed.*, 2012, **51**, 11686; (l) D. Enders and T. Balensiefer, *Acc. Chem. Res.*, 2004, **37**, 534; (m) D. Enders, O. Niemeier and A. Henseler, *Chem. Rev.*, 2007, **107**, 5606.
- For selected examples, see: (a) J. Xu, C. Mou, T. Zhu, B.-A. Song and Y. R. Chi, Org. Lett., 2014, 16, 3272; (b) T. Liu, S.-M. Han, L.-L. Han, L. Wang, X.-Y. Cui, C.-Y. Du and S. Bi, Org. Biomol. Chem., 2015, 13, 3654; (c) Z. Rafiński, A. Kozakiewicz and K. Rafińska, Tetrahedron, 2014, 70, 5739; (d) S. M. Langdon, C. Y. Legault and M. Gravel, J. Org. Chem., 2015, 80, 3597; (e) C. G. Goodman and J. S. Johnson, J. Am. Chem. Soc., 2014, 136, 7539; (f) M.-Q. Jia and S.-L. You, ACS Catal., 2013, 3, 622; (g) L.-H. Sun, Z.-Q. Liang, W.-Q. Jia and S. Ye, Angew. Chem. Int. Ed., 2013, 52, 5803; (h) Z. Rafiński, A. Kozakiewicz and K. Rafinska, ACS Catal., 2014, 4, 1404; (i) A. Patra, A. Bhunia and A. T. Biju, Org. Lett., 2014, 16, 4798; (j) J. Zhang, C. Xing, B. Tiwari and Y. R. Chi, J. Am. Chem. Soc., 2013, 135, 8113; (k) K. R. Law and C. S. P. McErlean, Chem.-Eur. J., 2013, 19, 15852; (l) P. M. Lalli, T. S. Rodrigues, A. M. Arouca, M. N. Eberlin, B. A. D. Neto, RSC Adv., 2012, 2, 3201.
- For selected examples, see: (a) C. Burstein and F. Glorius, *Angew. Chem. Int. Ed.*, 2004, 43, 6205; (b) S. S. Sohn, E. L. Rosen and J. W. Bode, *J. Am. Chem. Soc.*, 2004, 126, 14370; (c) M. He and J. W. Bode, *Org. Lett.*, 2005, 7, 3131; (d) A. Chan and K. A. Scheidt, *J. Am. Chem. Soc.*, 2007, 129, 5334; (e) A. Chan and K. A. Scheidt, *J. Am. Chem. Soc.*, 2008, 130, 2740; (f) Y. Li, Z.-A. Zhao, H. He and S.-L. You, *Adv. Synth. Catal.*, 2008, 350, 1885; (g) E. M. Phillips, T. E. Reynolds and K. A. Scheidt, *J. Am. Chem. Soc.*, 2008, 130, 2416; (h) M. Rommel, T. Fukuzumi and J. W. Bode, *J. Am. Chem. Soc.*, 2008, 130, 17266; (i) J.-L. Li, B. Sahoo, C.-G. Daniliuc and F. Glorius, *Angew. Chem. Int. Ed.*, 2014, 53, 10515; (j) K. P. Jang, G. E. Hutson, R. C. Johnston, E. O. McCusker, P. H. Y. Cheong and K. A. Scheidt, *J. Am. Chem. Soc.*, 2014,

136, 76; (k) C. Guo, B. Sahoo, C. G. Daniliuc and F. Glorius, J. Am. Chem. Soc., 2014, 136, 17402; (l) C. Guo, M. Schedler, C. G. Daniliuc and F. Glorius, Angew. Chem. Int. Ed., 2014, 53, 10232; (m) J. Xu, S. Hu, Y. Lu, Y. Dong, W. Tang, T. Lu and D. Du, Adv. Synth. Catal., 2015, 357, 923.

- 4. K. Zeitler, Org. Lett., 2006, 8, 637.
- For selected examples, see: (a) Z.-Q. Zhu, X.-L. Zheng, N.-F. Jiang, X. Wan and J.-C. Xiao, *Chem. Commun.*, 2011, **47**, 8670; (b) D. Du, Z. Hu, J. Jin, Y. Lu, W. Tang, B. Wang and T. Lu, *Org. Lett.*, 2012, **14**, 1274; (c) J. Mahatthananchai, J. Kaeobamrung and J. W. Bode, *ACS Catal.*, 2012, **2**, 494; (d) F. Romanov-Michailidis, C. Besnard and A. Alexakis, *Org. Lett.*, 2012, **14**, 4906; (e) B. Zhou, Z. Luo and Y. Li, *Chem.-Eur. J.*, 2013, **19**, 4428; (f) Y. Lu, W. Tang, Y. Zhang, D. Du and T. Lu, *Adv. Synth. Catal.*, 2013, **355**, 321; (g) J. Kaeobamrung, J. Mahatthananchai, P. Zheng and J. W. Bode, *J. Am. Chem. Soc.*, 2010, **132**, 8810.
- For selected examples, see: (a) S. D. Sarkar and A. Studer, Angew. Chem. Int. Ed., 2010, 49, 9266; (b) A. Biswas, S. D. Sarkar, R. Fröhlich and A. Studer, Org. Lett., 2011, 13, 4966; (c) Z.-Q. Rong, M.-Q. Jia and S.-L. You, Org. Lett., 2011, 13, 4080; (d) B. Wanner, J. Mahatthananchai and J. W. Bode, Org. Lett., 2011, 13, 5378; (e) A. G. Kravina, J. Mahatthananchai and J. W. Bode, Angew. Chem. Int. Ed., 2012, 51, 9433; (f) S. Bera, R. C. Samanta, C. G. Daniliuc and A. Studer, Angew. Chem. Int. Ed., 2014, 53, 9622; (g) Z.-Q. Liang, D.-L. Wang, H.-M. Zhang and S. Ye, Org. Lett., 2015, 17, 5140; (h) D. Xie, L. Yang, Y. Lin, Z. Zhang, D. Chen, X. Zeng and G. Zhong, Org. Lett., 2015, 17, 2318; (i) S. Hu, B. Wang, Y. Zhang, W. Tang, M. Fang, T. Lu and D. Du, Org. Biomol. Chem., 2015, 13, 4661; (j) S. R. Yetra, S. Mondal, S. Mukherjee, R. G. Gonnade and A. T. Biju, Angew. Chem. Int. Ed., 2016, 55, 268; (k) J.-H. Mao, Z.-T. Wang, Z.-Y. Wang and Y. Cheng, J. Org. Chem., 2015, 80, 6350; (l) H. Lu, J.-Y. Liu, C.-G. Li, J.-B. Lin, Y.-M. Liang and P. Xu, Chem. Commun., 2015, 51, 4473; (m) S. Bera, C. G. Daniliuc and A. Studer, Org. Lett., 2015, 17, 4940.
- For selected examples, see: (a) F.-G. Sun, L.-H. Sun and S. Ye, *Adv. Synth. Catal.*, 2011, 353, 3134; (b) C. Yao, D. Wang, J. Lu, T. Li, W. Jiao and C. Yu, *Chem.-Eur. J.*, 2012, 18, 1914; (c) S. R. Yetra, T. Kaicharla, S. S. Kunte, R. G. Gonnade and A. T. Biju, *Org. Lett.*, 2013, 15, 5202; (d) H.-R. Zhang, Z.-W. Dong, Y.-J. Yang, P.-L. Wang and X.-P. Hui, *Org. Lett.*, 2013, 15, 4750; (e) S. R. Yetra, T. Roy, A. Bhunia, D. Porwal and A. T. Biju, *J. Org. Chem.*, 2014, 79, 4245; (f) H.-M. Zhang, W.-Q. Jia, Z.-Q. Liang and S. Ye, *Asian J. Org. Chem.*, 2014, 3, 462; (g) D. Jiang, S. Dong, W. Tang, T. Lu and D. Du, *J. Org. Chem.*, 2015, 80, 11593; (h) C.-L. Zhang, D.-L. Wang, K.-Q. Chen and S. Ye, *Org. Biomol. Chem.*, 2015, 13, 11255.
- For selected examples, see: (a) Z. Zhang, X. Zeng, D. Xie, D. Chen, L. Ding, A. Wang, L. Yang and G. Zhong, *Org. Lett.*, 2015, **17**, 5052; (b) S. J. Ryan, L. Candish and D. W. Lupton, *J. Am. Chem. Soc.*, 2009, **131**, 14176; (c) J. Cheng, Z. Huang and Y. R. Chi, *Angew. Chem. Int. Ed.*, 2013, **52**, 8592; (d) L. Candish, A. Levens and D. W. Lupton, *J. Am. Chem. Soc.*, 2014, **136**, 14397.
- For selected examples, see: (a) S. J. Ryan, L. Candish and D. W. Lupton, J. Am. Chem. Soc., 2011, 133, 4694; (b) L. Candish and D. W. Lupton, J. Am. Chem. Soc., 2012, 135, 58; (c) S. Pandiancherri, S. J. Ryan and D. W. Lupton, Org. Biomol. Chem., 2012, 10, 7903; (d) S. J. Ryan, A. Stasch, M. N. Paddon-Row and D. W. Lupton, J. Org. Chem., 2012, 77, 1113; (e) L. Candish, C. M. Forsyth and D. W. Lupton, Angew. Chem. Int. Ed., 2013, 52, 9149.
- 10. X.-Y. Chen, Z.-H. Gao, C.-Y. Song, C.-L. Zhang, Z.-X. Wang and S. Ye, Angew. Chem. Int.

Ed., 2014, 53, 11611.

- 11. Y. Zhang, Y. Lu, W. Tang, T. Lu and D. Du, Org. Biomol. Chem., 2014, 12, 3009.
- For selected examples, see: (a) D. E. A. Raup, B. Cardinal-David, D. Holte and K. A. Scheidt, *Nat Chem*, 2010, 2, 766; (b) D. T. Cohen and K. A. Scheidt, *Chem. Sci.*, 2012, 3, 53; (c) D. T. Cohen, B. Cardinal-David and K. A. Scheidt, *Angew. Chem. Int. Ed.*, 2011, 50, 1678; (d) B. Cardinal-David, D. E. A. Raup and K. A. Scheidt, *J. Am. Chem. Soc.*, 2010, 132, 5345; (e) J. Dugal-Tessier, E. A. O'Bryan, T. B. H. Schroeder, D. T. Cohen and K. A. Scheidt, *Angew. Chem. Int. Ed.*, 2012, 51, 4963; (f) A. M. ElSohly, D. A. Wespe, T. J. Poore and S. A. Snyder, *Angew. Chem. Int. Ed.*, 2013, 52, 5789; (g) G.-T. Huang, T. Lankau and C.-H. Yu, *Org. Biomol. Chem.*, 2014, 12, 7297; (h) J. Mo, X. Chen and Y. R. Chi, *J. Am. Chem. Soc.*, 2012, 134, 8810; (i) J. Qi, X. Xie, R. Han, D. Ma, J. Yang and X. She, *Chem.-Eur. J.*, 2013, 19, 4146; (j) Z. Xiao, C. Yu, T. Li, X.-S. Wang and C. Yao, *Org. Lett.*, 2014, 16, 3632.
- 13. C. Palumbo, G. Mazzeo, A. Mazziotta, A. Gambacorta, M. A. Loreto, A. Migliorini, S. Superchi, D. Tofani and T. Gasperi, *Org. Lett.*, 2011, **13**, 6248.



Several novel isatin-derived 2-bromoenals were applied for formal [3+3] annulation with 1,3-dicarbonyl compounds enabled by NHC/Lewis acid cooperative catalysis.