

# RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This *Accepted Manuscript* will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

1 **Characteristics of estrogenic/antiestrogenic activities during anoxic/aerobic biotreatment process**  
2 **of simulated textile dyeing wastewater**

3 Na Liu, Xuehui Xie\*, Hong Jiang, Fang Yang, Chengzhi Yu, Jianshe Liu\*

4 *College of Environmental Science and Engineering, Donghua University, Shanghai 201620, China;*

5 *State Environmental Protection Engineering Center for Pollution Treatment and Control in Textile*

6 *Industry, Donghua University, Shanghai 201620, China.*

7 Email: liuna900301@163.com, xiexuehui@dhu.edu.cn, 397685268@qq.com, 1160515316@qq.com,

8 hyywsdklts@163.com, liujianshe@dhu.edu.cn.

9 \* Corresponding author: Donghua University, 2999# North Renmin Road, Songjiang District, Shanghai,  
10 China. Tel: +86 2167792535; fax: +86 21 67792522. E-mail: xiexuehui@dhu.edu.cn;

11 liujianshe@dhu.edu.cn;

12

13

14

15

16

17

18

19

20

21

22

23

24 **Abstract**

25 The presence of estrogenic/antiestrogenic chemicals in textile dyeing wastewater is well  
26 demonstrated according to previous studies. However, the characteristics of estrogenic/antiestrogenic  
27 activities during the conventional biological treatment has been poorly investigated. In this study, the  
28 yeast two-hybrid assay (YES) was used to evaluate the agonistic and antagonistic estrogen activity  
29 during the anoxic/aerobic treatment of textile dyeing wastewater. The results indicated that the  
30 estrogenic activity of the textile dyeing wastewater was negligible throughout the anoxic/aerobic  
31 treatment, but the antiestrogenic activity increased obviously after the aerobic treatment. By  
32 fractionating the dissolved organic matter (DOM) in wastewater into different fractions, it was found  
33 that hydrophobic acids (HOA) and hydrophobic neutrals (HON) were the key fractions involved in  
34 increasing antiestrogenic activity of the wastewater during anoxic/aerobic treatment. Furthermore,  
35 fluorescence spectroscopy analysis on wastewater samples and their fractions of soluble organic  
36 compounds suggested that HOA and HON fractions contained more humic/fulvic acid in aerobic  
37 effluent than that in anoxic effluent, which could mask estrogenic activity in aerobic effluent.

38 **Keywords:** estrogenic/antiestrogenic activity; simulated textile dyeing wastewater; anoxic/aerobic  
39 treatment; yeast two-hybrid assay (YES); DOM fractionation

40

41

42

43

44

45

## 46 1. Introduction

47 Biological techniques are ubiquitously used in textile dyeing wastewater treatment in wastewater  
48 treatment plants (WWTPs) for their high-efficiency and low-cost <sup>1</sup>. Although biological treatment  
49 successfully removes the great mass of organic matters resulting in compliance with discharge standard,  
50 some trace organic matters like estrogenic/antiestrogenic chemicals are biodegraded incompletely <sup>2,3</sup>.  
51 The estrogenic/antiestrogenic chemicals have been reported to mimic or antagonize the effect of steroid  
52 hormones, and interfere with the function of the endocrine system through affecting the reproduction  
53 and development of animals <sup>4-6</sup>. Recently, it has been demonstrated that there are  
54 estrogenic/antiestrogenic activities in textile dyeing effluent. Therefore, much more efforts should be  
55 made to investigate the fate of estrogenic/antiestrogenic chemicals in textile dyeing wastewater  
56 treatment <sup>7,8</sup>.

57 Textile dyeing wastewater usually contains variety of dyes and the auxiliaries released from textile  
58 dyeing and printing process <sup>9-11</sup>, which act as endocrine-disrupting compounds (EDCs). A study on 23  
59 commercial textile dyes suggested that over 50% dyes had antiestrogenic effect, and about 13% were  
60 estrogenic <sup>12</sup>. Some dyes can affect the endocrine function at genetic level. For example, Disperse  
61 Yellow 7 and Bismarck Brown Y were able to alter the expression of reproductive-related genes in  
62 western clawed frog <sup>13, 14</sup>. There are also some textile auxiliaries with estrogenic/antiestrogenic  
63 activities such as nonylphenols, which are included in the list of priority substances in the field of water  
64 policy established by the European Parliament <sup>15</sup>.

65 Because not all the estrogenic/antiestrogenic chemicals can be removed effectively by conventional  
66 process <sup>16,17</sup>, the advanced treatment process was implemented for tertiary stage such as ozonation, but  
67 costly and not always applicable <sup>18-20</sup>. In the ultimate treatment for wastewater reclamation, some

68 estrogenic/antiestrogenic chemicals generates during the wastewater disinfection such as chlorination <sup>21</sup>.  
69 Up to now, study on the fate of estrogenic/antiestrogenic chemicals during biological treatment is  
70 highly concerned and conducive to better treatment of textile dyeing wastewater <sup>22-24</sup>. Notably,  
71 dissolved organic matters (DOM) in the textile effluent, containing soluble microbial products and  
72 unknown estrogenic/antiestrogenic chemicals, is changeable along with biological degradation and  
73 synthesis <sup>25, 26</sup>. Therefore, DOM plays an important role in better evaluation of the  
74 estrogenic/antiestrogenic activities of the textile dyeing wastewater.

75 Accordingly, the purpose of this paper was to investigate the change characteristics of  
76 estrogenic/antiestrogenic activities during anoxic/aerobic biotreatment of textile dyeing wastewater,  
77 and further analyze the constituent and degradation products in the wastewater, which might be related  
78 to the estrogenic/antiestrogenic activities.

79

## 80 **2. Materials and Methods**

### 81 **2.1 Lab-scale reactor system and simulation of textile dyeing wastewater**

82 A sequential anoxic-aerobic reactor system was built for simulating the anoxic/aerobic treatment  
83 process of textile dyeing wastewater. The effective volume of anoxic reactor and aerobic reactor were  
84 16 L and 9.6 L, respectively. Each reactor was followed by a sedimentation basin of effective volume 8  
85 L. The initial seed sludge was collected from the returning activated sludge of secondary sedimentation  
86 tank in the sewage treatment plant in Songjiang, Shanghai, China. Firstly, the initial sludge was aerated  
87 for 24 h. Then, the sludge was mixed with water with a proportion of 3:2 (V: V) in both the anoxic and  
88 aerobic reactor. The flow rate of influent was 20 L/d, giving a hydraulic retention times of 12.8 h and  
89 7.68 h for anoxic and aerobic reactor, respectively. Before performing the experiment in this study, the

90 reactor has run for 6 months to achieve stability. During the start-up period of the system, the dissolved  
91 oxygen (DO) of anoxic reactor fluctuated from 0.20 mg/L to 0.50 mg/L, and the DO of aerobic reactor  
92 was between 3.00 mg/L to 5.00 mg/L. Besides, the solids retention times of anoxic and aerobic reactor  
93 were 15 d and 3 d, respectively.

94 The composition of simulated textile dyeing wastewater was shown in Supplementary Table 1. The  
95 mixture of 5 frequently-used textile dyes was 10 mg/L of Direct Red 28, Direct Yellow 12, Reactive  
96 Black 5, Reactive blue 21 and Reactive Blue 19, respectively, and the total concentration of dyes was  
97 50 mg/L. The starch and inorganic salts were used for providing energy, carbon source, nitrogen source  
98 and other mineral substances.

## 99 **2.2 Sample collection and water quality analysis**

100 Total operation time of the reactor system was over 180 days to reach steady-state conditions, which  
101 was defined by the pH, COD<sub>Cr</sub> within 5% variation in a week. The fresh prepared simulated wastewater  
102 was used as sample influent (named IN). The anoxic effluent (named AN) and aerobic effluent (named  
103 AE) samples were collected from the sedimentation basins. All the three samples were filtered through  
104 0.45 μm microfiltration membrane to remove the insoluble substances, and then stored in 4 °C.  
105 Characteristics of these three samples were listed in Table 1, including pH value, COD<sub>Cr</sub>, BOD<sub>5</sub>, total  
106 organic carbon (TOC) and decolorization ratio. The pH values were measured immediately after  
107 collection. COD<sub>Cr</sub> was analyzed by microwave digestion method after centrifugation. BOD<sub>5</sub> was  
108 measured using 880 digital BOD test apparatus (Jiangfen, China). The TOC was measured with V-CPH  
109 TOC analyzer (Shimadzu, Japan) after filtration. The UV-Visible spectra were detected by a P-300  
110 nanophotometer (Implen, Germany) after filtration. Decolorization of the simulated dyeing wastewater  
111 was analyzed using ADMI (American Dye Manufacturing Institute) color values method<sup>27</sup>.

112 **Table 1.** Characteristics of three samples used in this study

Sample	pH	COD <sub>Cr</sub> (mg/L)	BOD <sub>5</sub> (mg/L)	TOC (mg/L)	Decolorization ratio (%)
IN	7.01	985	487	367.3	-
AN	4.73	442	340	150.6	67%
AE	7.31	162	94	65.3	75%

113

114 **2.3 Fractionation**

115 DOM of the anoxic and aerobic effluent samples was isolated into hydrophobic acids (HOA),  
116 hydrophobic bases (HOB), hydrophobic neutrals (HON) and hydrophilic substances (HIS) with method  
117 described by Huang and Yeh <sup>28</sup>, which was performed with some modification as follows: (1) 500 mL  
118 original effluent was directly passed through the XAD-8 resin column to adsorb hydrophobic bases  
119 (HOB) and hydrophobic neutrals (HON). (2) The column was back-flushed with 200 mL 0.1 M HCl to  
120 obtain HOB fraction and subsequently 100 mL ultrapure water for flushing the residual acid and HOB.  
121 (3) The effluent from Step 1 was adjusted to pH 2 and passed through column again to absorb the  
122 hydrophobic acid (HOA). The effluent of Step 3 was hydrophilic substances (HIS). (4) The column was  
123 back-flushed again with 200 mL 0.1 M NaOH and 100 mL ultrapure water for HOA fraction. (5) The  
124 HON fraction adsorbed XAD-8 resin was eluted with methanol in Soxhlet extractor. After fractionation,  
125 The HON fraction in methanol eluate was dried by vacuum-rotary evaporation. The other fractions  
126 were added with ultrapure water to increase volume to 500 mL.

127 **2.4 Solid-phase extraction and sample concentration**

128 The dried residues of HONs were redissolved in 2 mL methanol. 1 mL methanol solution of HON  
129 was dried under nitrogen stream and dissolved in 250  $\mu$ L DMSO to obtain 1000 fold concentration for  
130 yeast screen assay. 0.2 mL HON methanol solution was also dried and dissolved in 50 mL ultrapure  
131 water for fluorescence spectroscopy analysis.

132 Wastewater samples IN, AN and AE, and their fractions except for HON were solid-extracted  
133 according the method reported by Wu, *et al.* <sup>21</sup>. In brief, 300 mL effluent samples were acidified to pH  
134 2 with 2 M H<sub>2</sub>SO<sub>4</sub> and passed through HLB cartridges (Waters Oasis, America). The organic matters  
135 retained on the cartridge were eluted by 10 mL methanol, 10 mL dichloromethane and 10 mL hexane.  
136 Then 5 mL of each eluate was mixed and dried under the nitrogen stream. The dried residues were  
137 dissolved in 150  $\mu$ L dimethylsulfoxide (DMSO) to obtain 1000-fold concentration for yeast screen  
138 assay. The rest 5 mL of each eluate was also mixed and dried for GC-MS analysis. The  
139 estrogenic/antiestrogenic activities of the concentrated samples were evaluated with the yeast  
140 two-hybrid assay based on yeast cells.

#### 141 **2.5 Estrogenic activity assay**

142 The recombinant yeast cells (*Saccharomyces cerevisiae* Y190) for yeast screen assay was donated  
143 from State Key Joint Laboratory of Environmental Simulation and Pollution Control (Department of  
144 Environmental Science and Engineering, Tsinghua University, PRChina), which contained rat estrogen  
145 receptor ER $\alpha$  and the coactivator TIF2. Estrogenic activity was evaluated by the  $\beta$ -galactosidase  
146 induced by estrogenic samples <sup>21, 29</sup>, which performed as follows: The yeast cells were preincubated  
147 overnight at 30 $\square$ . Then 100  $\mu$ L overnight cells and 20  $\mu$ L DMSO solution containing the samples were  
148 added into the mixture of 400  $\mu$ L SD medium and incubated at 30 $\square$  for 4 h. After incubation, 150  $\mu$ L  
149 yeast culture was used for absorbance at 595 nm. The residual yeast in mixed solution (370  $\mu$ L) were  
150 collected by centrifugation and resuspended in 200  $\mu$ L Z-buff containing 1 g/L Zymolyase 20T for  
151 digestion (15 min, 37 $\square$ C). The enzymatic reaction was started by addition of 40  $\mu$ L 4 g/L  
152 2-nitrophenyl- $\beta$ -D-galactosidase (ONPG) at 30 $\square$ . After 30 min, the reaction was stopped by addition of  
153 100  $\mu$ L of 1M Na<sub>2</sub>CO<sub>3</sub>. Thereafter the solution was centrifuged and 150  $\mu$ L supernatant was taken for

154 absorbance at 405 nm and 570 nm, respectively.

155 Some samples with strong absorbance at 595 nm could interfere experimental results, thus the  
156 observation absorbance at 595 nm need to be corrected. The blank samples in YES assay were  
157 composed of 20  $\mu$ L DMSO solution of samples and 500  $\mu$ L ultrapure water. Therefore, the corrected  
158  $OD_{595}$  was calculated in terms of Lambert-Beer's law as following equation (1):

$$OD_{595COR} = OD_{595OBS} - OD_{595BLA} + OD_{595BAC} \quad (1)$$

159 where  $OD_{595COR}$  represents the corrected absorbance of samples at 595 nm;  $OD_{595OBS}$  is the observed  
160 absorbance of samples after 4 h incubation;  $OD_{595BLA}$  is the absorbance of the blank samples;  $OD_{595BAC}$   
161 is the background absorbance of empty test plate.

162 Therefore, the  $\beta$ -galactosidase activity was calculated according to equation (2):

$$U = \frac{1000 \cdot (OD_{405} - 1.75 \cdot OD_{570})}{t \cdot v \cdot OD_{595COR}} \quad (2)$$

163 where U represents the  $\beta$ -galactosidase activity; the  $OD_{405}$  is the absorbance of *o*-nitrophenol after  
164 reaction;  $OD_{570}$  is light scattering after reaction, *t* is time of reaction (min); *v* is volume of culture (mL).

## 165 2.6 Antiestrogenic activity assay

166 The antiestrogenic activity assay was investigated by the inhibitory effect of samples against  
167  $\beta$ -galactosidase activity of E2, also according to the yeast two-hybrid assay<sup>30</sup>. In this assay, 100  $\mu$ L  
168 yeast culture, 20  $\mu$ L DMSO solution containing the concentrated sample and additional E2 was also  
169 added to 400  $\mu$ L SD medium. The final concentration of E2 was 0.77  $\mu$ g/L, which could elicit 40%  
170 submaximal ER agonist response in the absence of antiestrogenic chemicals. For the correction of  
171 absorbance at 595 nm, the blank samples consisted of 10  $\mu$ L DMSO solution of samples, 10  $\mu$ L DMSO  
172 and 500  $\mu$ L ultrapure water. After 4 h incubation, the  $\beta$ -galactosidase activity was determined, the

173 percentage inhibition of concentrated samples to the  $\beta$ -galactosidase reduction was calculated  
174 according to the following equation (3):

$$I_X(\%) = \frac{U_{E2} - U_X}{U_{E2}} \cdot 100\% \quad (3)$$

175 Where  $I_X$  represents the inhibition of concentrated samples to  $\beta$ -galactosidase activity induced by E2;  
176  $U_{E2}$  is the  $\beta$ -galactosidase activity by 0.77  $\mu\text{g/L}$  E2 standard;  $U_X$  is the  $\beta$ -galactosidase activity of E2  
177 and the concentrated sample.

### 178 **2.7 Cytotoxicity assay**

179 Toxic samples can inhibit the growth of yeast cells, which also lead to the inhibition of  
180  $\beta$ -galactosidase activity. Therefore, the toxicity of the sample was measured by the absorbance at 595  
181 nm ( $OD_{595}$ ) after 4 h incubation of yeast culture during the antiestrogenic assay<sup>21,30</sup>. The  $OD_{595}$  value  
182 was also corrected as mentioned in estrogenic assay and then converted to percentage inhibition of the  
183 concentrated sample to yeast growth, as following equation (4):

$$I_c(\%) = \frac{OD_{595b} - OD_{595x}}{OD_{595b}} \times 100\% \quad (4)$$

184 Where the  $I_c$  is the inhibition of the sample to growth of yeast cell;  $OD_{595b}$  is the absorbance after  
185 incubation with E2 and DMSO;  $OD_{595x}$  is the absorbance after incubation with E2 and DMSO solution  
186 containing the concentrated sample. The sample was assessed as toxic when the  $I_c$  was 10% or more  
187 than 10%.

### 188 **2.8 Fluorescence spectroscopy**

189 Fluorescence spectral measurement was conducted using the QuantMaster 40 fluorescence  
190 spectrometer (PTI, America). The fractions of sample AN and AE were adjusted to pH 3 before  
191 measurement. The excitation wavelength range was 200-400 nm, at intervals of 5 nm; the emission

192 wavelength range was 280-550 nm, at intervals of 1 nm. To limit its second-order Raleigh scattering,  
193 the starting wavelength of emission was 20 nm longer than its corresponding excitation wavelength  
194 from beginning to end. After inner-filter correction<sup>31</sup>, data of fluorescence spectra were converted into  
195 the excitation-emission matrixes (EEM). The contour maps were created using Origin 8.0 program with  
196 same scale range of fluorescence intensities.

### 197 **2.9 GC-MS analysis**

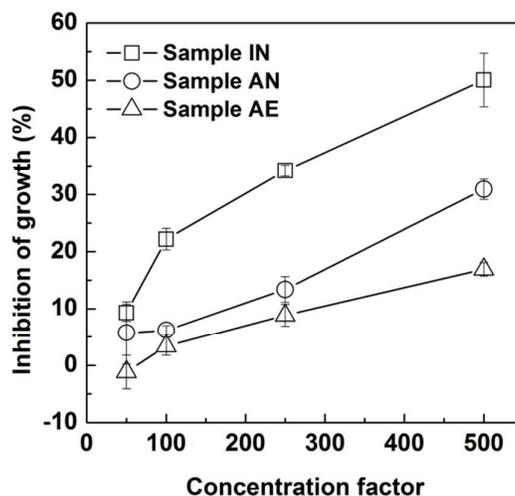
198 The degradation products during anoxic/aerobic biotreatment process was conducted using QP-2010  
199 gas chromatography coupled with mass spectrometer (Shimadzu, Japan). The GC-MS analysis was  
200 performed at ionization voltage 70 eV. The initial Restek column (0.25 mm, 60 m; XTI-5) temperature  
201 remained at 40□ for 10 min, then ascend to 280°C at 10°C min<sup>-1</sup>, and remained for 7 min. The  
202 temperature of injection port kept at 280°C and the GC/MS interface maintained at 300°C. The flow  
203 rate of carrier gas (nitrogen) was 1.0 mL min<sup>-1</sup>. The products were identified based on the mass spectra  
204 and NIST spectral library stored in the computer soft-ware (version 1.10 beta, Shimadzu).

205

## 206 **3. Results and Discussion**

### 207 **3.1 Cytotoxicity of simulated textile dyeing wastewater during anoxic/aerobic treatment**

208 The growth inhibition of samples IN, AN and AE to yeast cells were shown in Figure 1. It could be  
209 seen that sample IN was the most toxic, and its I<sub>c</sub> value reached 29.2% even at 100-fold concentration,  
210 and was over 50% at 500-fold. The growth inhibition of sample AN was higher than that of sample AE  
211 at each concentration, and their I<sub>c</sub> values were 30.1% and 16.9% at 500-fold concentration, respectively.  
212 These results suggested that cytotoxicity of the simulated textile dyeing wastewater decreased  
213 gradually by anoxic/aerobic biological treatment.



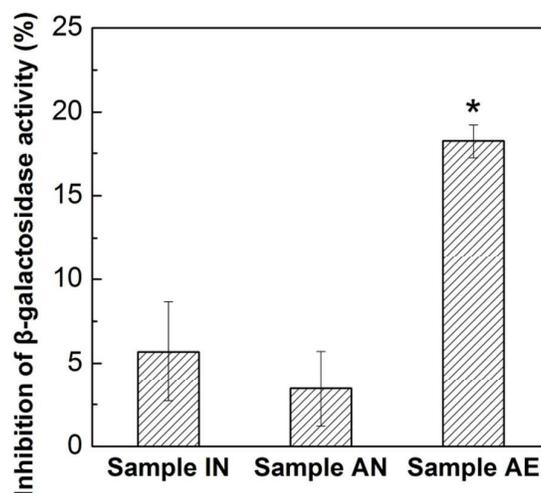
214

215 **Figure 1.** The cytotoxicity of samples IN, AN and AE. Error bars represent the standard deviation  
 216 based on triplicate analyses.

217

### 218 3.2 Estrogenic/antiestrogenic activities of simulated textile dyeing wastewater during 219 anoxic/aerobic treatment

220 In this study, antiestrogenic effects were observed in the samples IN, AN and AE, but estrogenic  
 221 effects were not detected in all samples. To avoid the impact from cytotoxicity, three samples at 50-fold  
 222 concentration assessed as non-toxic ( $I_c$  values < 10%) were chosen to investigate the change of  
 223 antiestrogenic activity during anoxic/aerobic treatment. As shown in Figure 2, the inhibition of  
 224  $\beta$ -galactosidase activity decreased moderately after anoxic reaction, which was from 5.7% down to  
 225 3.5%, but significantly increased to 18.2% after aerobic reaction ( $p < 0.05$ ). This result suggested that  
 226 some active substances could antagonize E2 generated during the aerobic biological treatment.

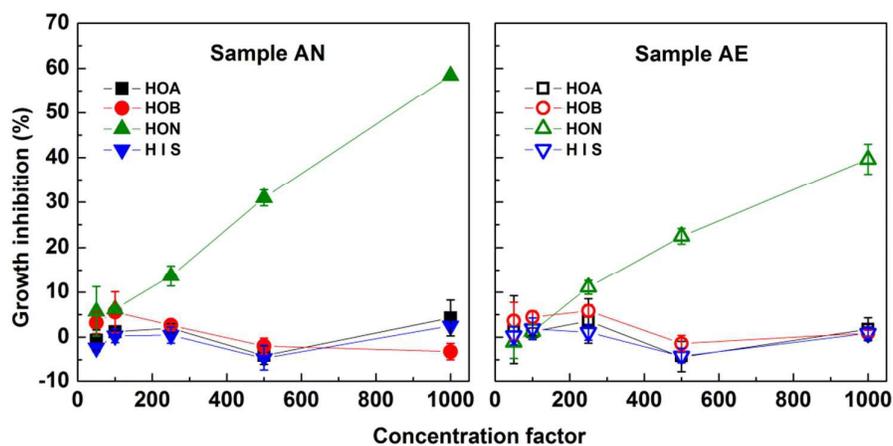


227  
228 **Figure 2.** The antiestrogenic activity of samples IN, AN and AE at 50-fold concentration. Asterisk sign  
229 (\*) indicates that the antiestrogenic activity of sample was significantly different from that of the  
230 sample IN ( $p < 0.05$ ). Error bars represent the standard deviation based on triplicate analyses.  
231

232 Therefore, the simulated textile dyeing wastewater in this study had antiestrogenic activity. It also  
233 has been reported that high antiestrogenic activity was detected in industrial effluent from textile and  
234 dyeing wastewater treatment plants <sup>7</sup>, but not involving the relative study about the characteristics of  
235 the antiestrogen-active substances.

### 236 3.3 Cytotoxicity of different DOM fractions from the anoxic and aerobic effluent

237 For analyzing the main antiestrogen-functional component, sample AN and sample AE were  
238 fractionated into four fractions, including HOA, HOB, HON and HIS. The inhibition of four fractions  
239 from two samples to the growth of yeast cells were shown in Figure 3. HON of each sample had  
240 obvious toxic effect on the yeast cell. Even at 250-fold concentration, the  $I_c$  values of HON were 13.7%  
241 and 11.1% in sample AN and AE, respectively, which was over non-toxic limit. While the other three  
242 fractions of each samples were assessed as non-toxic because their  $I_c$  values were all less than 10%.



243

244 **Figure 3.** The cytotoxicity of four fractions of the sample AN and sample AE at different concentration  
 245 factor. HOA, hydrophobic acids; HOB, hydrophobic bases; HON, hydrophobic neutrals; HIS,  
 246 hydrophilic substances. Error bars represent the standard deviation based on triplicate analyses.

247

248 HON was the only fraction with deep color of four DOM fractions in samples AN and AE, which

249 had strong absorbance in visible spectra of 400-700 nm. This phenomenon suggested that there were

250 some colored matters, including undegraded textile dyestuffs and biodegradation products with

251 chromophoric groups, existing in the HON fraction of both samples. It is possible that the cytotoxicity

252 of HON fraction was mainly related to these colored substances.

### 253 3.4 Antiestrogenic activity of different DOM fractions from the anoxic and aerobic effluent

254 From the Figure 4, DOM fractions from samples AN and AE had different antiestrogenic activities.

255 The  $\beta$ -galactosidase activity inhibition of HOB and HIS of both samples did not increased obviously or

256 change regularly along with the increasing concentration factor, which indicated that these two

257 fractions did not elicit obvious antiestrogenic response. But HOA and HON of both samples could

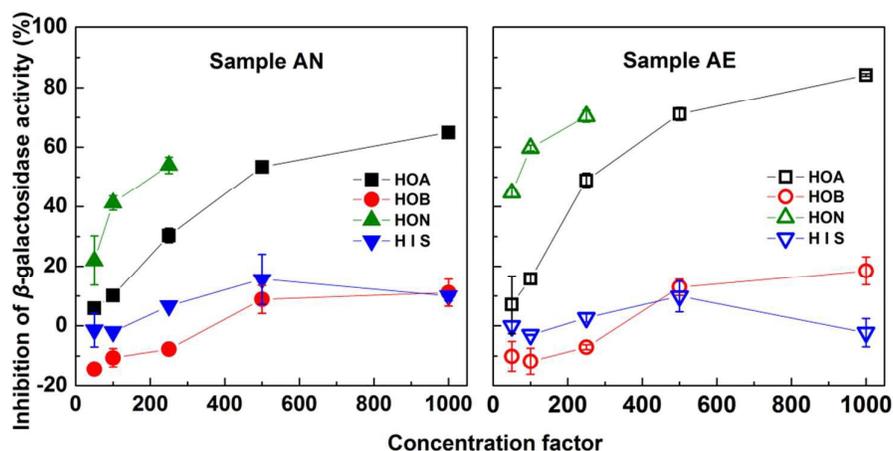
258 cause distinct reduction of  $\beta$ -galactosidase activity. The inhibition of HOA in sample AN to

259  $\beta$ -galactosidase activity increased from 5.9% at 50-fold concentration to 64.9% at 1000-fold

260 concentration, and that in sample AE it ranged from 7.1% to 84.2%. The HON also had strong

261 antiestrogenic activity in sample AN and AE, and their  $\beta$ -galactosidase activity inhibition reached 21.9%

262 and 44.7% at 50-fold concentration, respectively. Because the  $I_c$  value of HON far exceeded the  
 263 non-toxic limit at 500- and 1000-fold concentration, the corresponding data were not shown in Figure 4.  
 264 Overall, the antiestrogenic activity of HOA and HON in sample AE was significantly higher than that  
 265 in sample AN, which was consistent with antiestrogenic activity between un-fractionated sample AN  
 266 and sample AE.



267  
 268 **Figure 4.** The antiestrogenic activity of four fractions from the sample AN and sample AE at different  
 269 concentration factor. HOA, hydrophobic acids; HOB, hydrophobic bases; HON, hydrophobic neutrals;  
 270 HIS, hydrophilic substances. Error bars represent the standard deviation based on triplicate analyses.  
 271

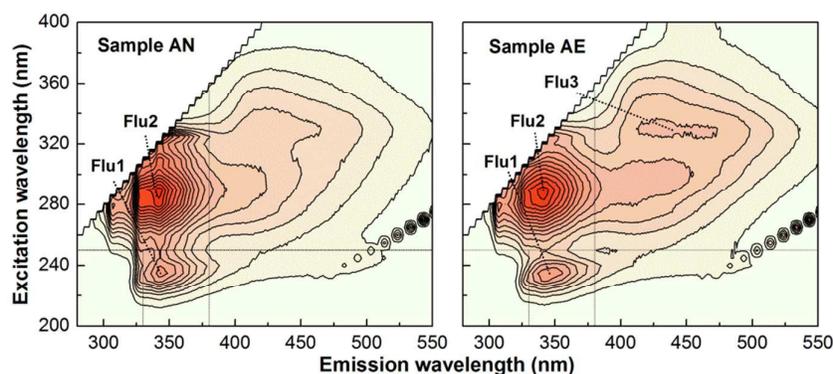
272 Therefore, the main antiestrogen-active fractions were HOA and HON. Until now, a lot of research  
 273 have studied the mechanisms for antiestrogenic effects, including estrogenic receptor antagonists  
 274 and/or interaction<sup>32-34</sup>, sorption by macromolecules<sup>21, 35, 36</sup>, and some other non-specific ways such as  
 275 changes of membrane permeability for estrogenic chemicals<sup>37</sup>. It has been reported that the  
 276 antiestrogenic chemicals such as tamoxifen and raloxifene can completely bind to ER and induce a  
 277 conformational change, which inhibit the transformation of estrogen-dependent genes<sup>33</sup>, and result in  
 278 the antiestrogenic activities. In addition, as in this study, the toxicity of the compound had exceed its  
 279 putative endocrine effects, and the yeast acted as a toxicity biosensor<sup>34</sup>.

280 HOA faction showed inhibition of  $\beta$ -galactosidase induction, but not inhibition of yeast cell growth.

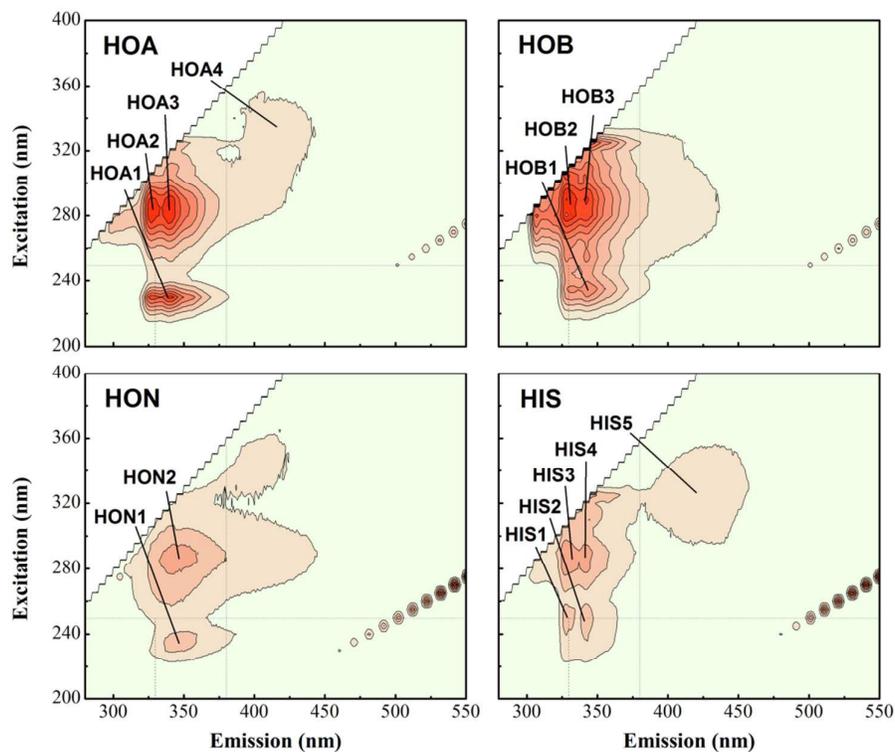
281 Thus, the antiestrogenic activity of HOA may result from the antiestrogenic chemicals affect the  
282 induction mechanism of E2 in yeast such as competitively bind to the ER, and/or the macromolecules,  
283 which can absorb the E2. The probability of high inhibition of  $\beta$ -galactosidase activity of HON in both  
284 samples is mainly related to the high cytotoxicity. In addition, HON may contain molecules in larger  
285 size and micellae such as colloidal organic carbon (COC) which can pass through microfiltration  
286 membrane. Thus these organic matters in HON, which can be intercepted by the XAD-8 resin and  
287 eluted by Soxhlet extraction, may give rise to stronger E2 sorption behavior<sup>36</sup>.

### 288 3.5 Excitation-emission matrix (EEM) fluorescence spectroscopy of different samples and DOM 289 fractions

290 Samples AN and AE and their fractions exhibited different antiestrogenic activity in this study. Thus,  
291 an EEM fluorescence spectroscopy was used to characterize the chemical structures of soluble organic  
292 matters in samples. The contour maps of the results were shown in Figures 5, 6 and 7. The peaks of  
293 different fractions were related to substances with different chemical structures according to the  
294 previous research<sup>38</sup>. The intensity of peak HOB2 recorded was normalized as 1000 arbitrary units  
295 (AU). The location of  $Ex_{max}/Em_{max}$  and intensity of these peaks and their corresponding substances  
296 were listed in Table 2 and 3.

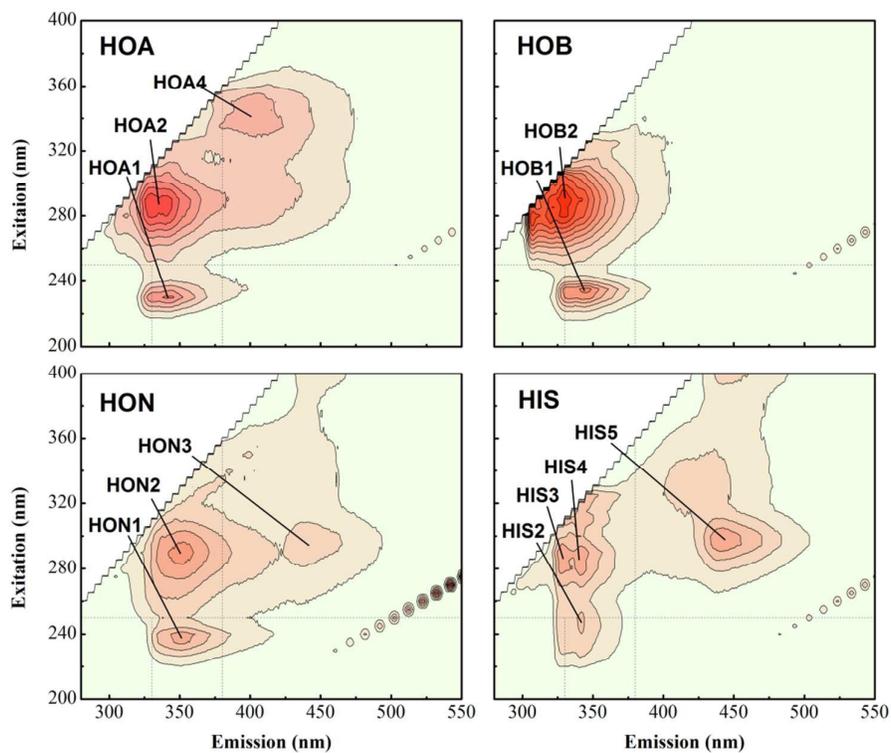


297  
298 **Figure 5.** EEM fluorescence spectra for samples AN and AE.



299

300 **Figure 6.** EEM fluorescence spectra for four fractions of sample AN. HOA, hydrophobic acids; HOB,  
 301 hydrophobic bases; HON, hydrophobic neutral; HIS, hydrophilic substances.



302

303 **Figure 7.** EEM fluorescence spectra for four fractions of sample AE. HOA, hydrophobic acids; HOB,  
 304 hydrophobic bases; HON, hydrophobic neutral; HIS, hydrophilic substances.

305 **Table 2** EEM peaks for samples AN and AE

Peak	Sample AN		Sample AE		Homologous substances
	$E_{x_{max}}/E_{m_{max}}$ (nm/nm)	Intensity at $E_{x_{max}}/E_{m_{max}}$ (AU)	$E_{x_{max}}/E_{m_{max}}$ (nm/nm)	Intensity at $E_{x_{max}}/E_{m_{max}}$ (AU)	
Flu1	235/343	906	235/347	747	Tryptophan-like, aromatic protein
Flu2	290/323	1896	290/341	1480	Soluble microbial by-product-like
Flu3			330/434	488	Humic acid-like

306

307 **Table 3.** EEM peaks for the fractions of sample AN and sample AE

Peak	Sample AN		Sample AE		Homologous substances
	$E_{x_{max}}/E_{m_{max}}$ (nm/nm)	Intensity at $E_{x_{max}}/E_{m_{max}}$ (AU)	$E_{x_{max}}/E_{m_{max}}$ (nm/nm)	Intensity at $E_{x_{max}}/E_{m_{max}}$ (AU)	
HOA1	230/339	506	230/341	250	Tryptophan-like, aromatic protein
HOA2	280/327	530	285/331	355	soluble microbial by-product-like
HOA3	285/341	541			soluble microbial by-product-like
HOA4	325/411	89	335/406	173	humic acid-like
HOB1	235/344	591	235/345	299	Tryptophan-like, aromatic protein
HOB2	280/328	1000	290/328	506	soluble microbial by-product-like
HOB3	290/342	905			soluble microbial by-product-like
HON1	235/348	121	240/352	214	Tryptophan-like, aromatic protein
HON2	285/344	183	290/350	260	soluble microbial by-product-like
HON3			295/438	133	humic acid-like
HIS1	250/328	135			aromatic protein
HIS2	250/342	139	250/342	154	Tryptophan-like, aromatic protein
HIS3	290/327	214	290/327	214	soluble microbial by-product-like
HIS4	290/341	203	290/342	229	soluble microbial by-product-like
HIS5	325/420	94	300/442	226	humic acid-like

308

309 These EEM peaks are related to tryptophan-like aromatic protein, soluble microbial by-product-like

310 or humic acid-like organic compounds according to a location of EEM peaks of many typical

311 chemicals in wastewater<sup>38</sup>. From Table 2, it was found that more soluble microbial by-product-like and

312 aromatic protein-like substances existed in sample AN, and more humic/fulvic acid-like substances

313 existed in sample AE. From Table 3, HOB had the highest content of aromatic protein and soluble

314 microbial by-product and HIS had the highest content of humic/fulvic acid both in sample AN and

315 sample AE. However, these two fractions did not elicit obvious antiestrogenic response. The aromatic  
316 protein in HOA decreased after aerobic treatment, but the antiestrogenic activity increased. Therefore,  
317 it is necessary to find whether some antiestrogen-active substances generated after aerobic reaction.  
318 Furthermore, since the content humic/fulvic acids in HOA and HON both increased after aerobic  
319 reaction, it is possible that these macromolecules mask the estrogenic activity in the bioassay and  
320 exhibit antiestrogenic activity<sup>35,36</sup>.

### 321 **3.6 Products analysis**

322 In this study, antiestrogenic activity also may arise from the mechanisms such as competitive binding,  
323 but not just simple sorption of E2. Some antagonists such as 4-hydroxytamoxifen can competitively  
324 bind to ER leading to a conformational change in receptor, and culminate in inhibition of gene  
325 expression<sup>33</sup>. Therefore, gas chromatography-mass spectra (GC-MS) analysis was carried out to  
326 determine what intermediate and/or degradation products in wastewater samples AN and AE, which  
327 might be related to cytotoxicity or estrogenic/antiestrogenic activities. The detected compounds were  
328 shown in Supplementary Table 2.

329 In GC-MS determination of sample AN, a lot of low-weight-molecule organic acids were detected  
330 by comparison of retention times and mass spectra of standards, including short chain fatty acids  
331 (SCFAs) such as propionic acid, butyric acid, valeric acid, caproic acid, and the isomers such as  
332 isobutyric acid, 2-/3-methylbutyric acid, and the derivatives such as 2-hydracrylic acid. These organic  
333 acids generated by anoxic biodegradation and resulted in low pH of anoxic effluent. Moreover, some  
334 aromatic compounds also could be identified, including aromatic acids such as benzoic and  
335 phenylacetic acids and phenolics such as phenol, m-cresol and 4-methylcatechol.

336 It has been reported that some weak acids such as propionic, butyric, caproic and benzoic acids can  
337 inhibit the fermentation rate of the *Saccharomyces cerevisiae*<sup>39</sup>. From the study of Wattanadilok et al.,  
338 the phenylacetic acid had the growth inhibitory effect on test seven yeasts<sup>40</sup>. With regard to the two  
339 detected phenolic compounds, m-cresol was the most active cresol isomer in antifungal activity to the  
340 *Fusarium verticillioides*<sup>41</sup>, and 4-methylcatechol was found to be able to strongly decrease the growth  
341 rate of *Debaryomyces hansenii*<sup>42</sup>. These studies indicated that the detected compounds may have  
342 inhibitory/cytotoxic effect on the tested yeast *Saccharomyces cerevisiae* Y190.

343 While less organic acids were detected in the sample AE, which suggested that low-weight-molecule  
344 organic acid were degraded under aerobic conditions. Noteworthy, the p-phenetidine and phthalic acid  
345 gave peaks at 15.1 and 20.2 min in all the samples. P-phenetidine can be cleaved from direct yellow 12  
346 and assigned to priority substance because it may cause sensitization by skin contact<sup>43</sup>. Recent studies  
347 have shown that several fungi strains are able to degrade the direct blue 19 and phthalic acid was  
348 identified from the accumulated degradation products<sup>44</sup>. Thus it is possible that the phthalic acid  
349 identified in this study may be also generated from the degradation of direct blue 19. Pavan B. et al first  
350 demonstrated that phthalic acid can bind to the estrogenic receptor with high affinity and mimics  
351  $17\beta$ -estradiol actions in WISH cells<sup>45</sup>. Furthermore, recent studies of phthalic acid found that it can  
352 cause a general significant increase of vitellogenin (vtg) protein in both sexes and induce significant  
353 increase of ER $\alpha$  gene expression<sup>46</sup>. The reason that discrepancy between YES biological assay and  
354 GC-MS analysis lie in two aspects: the concentration level of trace target product - phthalic acid, which  
355 can be further degraded by aerobic biodegradation, was too low to be detected; the complicated sample  
356 may elicit the comprehensive biological effect, but not estrogenic effect.

357 However, the antiestrogenic chemical has not been identified in this study, which might mainly on  
358 account of the complexity and limitation of identification means. Therefore, potential  
359 estrogenic/antiestrogenic chemicals should be identified and characterized by more comprehensive  
360 detection methods in future study.

361

#### 362 **4. Conclusions**

363 In this study, the estrogenic/antiestrogenic activities during the anoxic/aerobic treatment were  
364 investigated by yeast two-hybrid assay. The results showed that estrogenic activity were not detected  
365 during the whole treatment process. However, the antiestrogenic activity decreased via anoxic  
366 treatment, but increased significantly after aerobic treatment. Among four fractions, hydrophobic acids  
367 (HOA) and hydrophobic neutrals (HON) played important roles in increasing antiestrogenic activity  
368 during anoxic/aerobic treatment. In addition, more humic/fulvic acid indicated antiestrogenic activity  
369 were found in HOA and HON fractions in aerobic effluent than that in anoxic effluent. In future, more  
370 different biotoxicity indicators, such as acute toxicity and genotoxicity of samples are needed to be  
371 detected, and the real toxicity of samples are expected to be revealed more comprehensively and  
372 completely.

373

#### 374 **Acknowledgements**

375 The authors acknowledge the financial support by the National Natural Science Foundation of China  
376 (21377023, 51508083), the Fundamental Research Funds for the Central Universities (2232015D3-22)  
377 and Chinese Universities Scientific Fund (CUSF-DH-D-2015040). This work was partially supported  
378 by Shanghai Leading Academic Discipline Project (B604). The authors would like to thank State Key

379 Joint Laboratory of Environmental Simulation and Pollution Control, Department of Environmental  
380 Science and Engineering, Tsinghua University, Beijing 100084, PRChina.

381

382 **References**

383 1 V. Prigione, V. Tigini, C. Pezzella, A. Anastasi, G. Sannia and G. C. Varese, *Water research*, 2008, **42**,  
384 2911-2920.

385 2 A. C. Johnson and J. P. Sumpter, *Environmental Science & Technology*, 2001, **35**, 4697-4703.

386 3 D. Stalter, A. Magdeburg, M. Wagner and J. Oehlmann, *Water Res.*, 2011, **45**, 1015-1024.

387 4 S. Jobling, M. Nolan, C. R. Tyler, G. Brighty and J. P. Sumpter, *Environmental Science & Technology*,  
388 1998, **32**, 2498-2506.

389 5 E. F. Orlando and L. J. Guillette, Jr., *Environ. Res.*, 2007, **104**, 163-173.

390 6 C. P. Silva, M. Otero and V. Esteves, *Environ. Pollut.*, 2012, **165**, 38-58.

391 7 Y. X. Fang, G. G. Ying, J. L. Zhao, F. Chen, S. Liu, L. J. Zhang and B. Yang, *Environ. Toxicol. Chem.*,  
392 2012, **31**, 1273-1282.

393 8 P. Pothitou and D. Voutsas, *Chemosphere.*, 2008, **73**, 1716-1723.

394 9 E. J. *Colour in dyehouse effluent*, Society of Dyers and Colourists, Bradford, UK, 1995.

395 10 C. Pearce, *Dyes and Pigments*, 2003, **58**, 179-196.

396 11 G. Eremektar, H. Selcuk and S. Meric, *Desalination*, 2007, **211**, 314-320.

397 12 I. Bazin, A. Ibn Hadj Hassine, Y. Haj Hamouda, W. Mnif, A. Bartegi, M. Lopez-Ferber, M. De  
398 Waard and C. Gonzalez, *Ecotoxicol. Environ. Saf.*, 2012, **85**, 131-136.

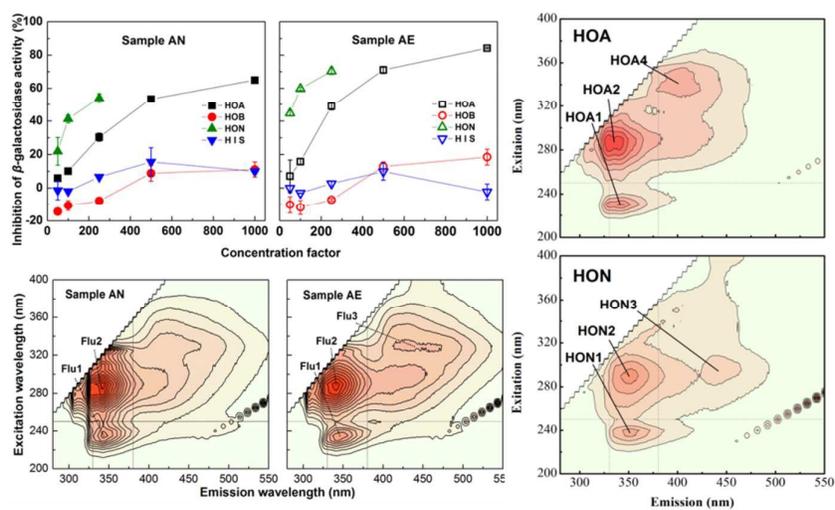
399 13 J. Mathieu-Denoncourt, C. J. Martyniuk, S. R. de Solla, V. K. Balakrishnan and V. S. Langlois,  
400 *Environ. Sci. Technol.*, 2014, **48**, 2952-2961.

- 401 14 J. J. Soriano, J. Mathieu-Denoncourt, G. Norman, S. R. de Solla and V. S. Langlois, *Environmental*  
402 *science and pollution research international*, 2014, **21**, 3582-3591.
- 403 15 E. C. European Parliament, *Off. J. Eur. Commun*, 2001, **331**, 1-5.
- 404 16 A. C. Johnson, H. R. Aerni, A. Gerritsen, M. Gibert, W. Giger, K. Hylland, M. Jurgens, T. Nakari, A.  
405 Pickering, M. J. Suter, A. Svenson and F. E. Wettstein, *Water Res.*, 2005, **39**, 47-58.
- 406 17 L. Clouzot, P. Doumenq, P. Vanloot, N. Roche and B. Marrot, *Journal Of Membrane Science*, 2010,  
407 **362**, 81-85.
- 408 18 T. Schiliro, A. Porfido, F. Spina, G. C. Varese and G. Gilli, *Sci. Total Environ.*, 2012, **432**, 389-395.
- 409 19 G. Bertanza, M. Papa, R. Pedrazzani, C. Repice, G. Mazzoleni, N. Steimberg, D. Feretti, E. Ceretti  
410 and I. Zerbini, *Science Of the Total Environment*, 2013, **458**, 160-168.
- 411 20 G. Bertanza, M. Papa, R. Pedrazzani, C. Repice and M. Dal Grande, *Water Science And Technology*,  
412 2013, **68**, 567-574.
- 413 21 Q. Y. Wu, H. Y. Hu, X. Zhao and Y. X. Sun, *Environmental Science & Technology*, 2009, **43**,  
414 4940-4945.
- 415 22 A. Joss, H. Andersen, T. Ternes, P. R. Richle and H. Siegrist, *Environmental Science & Technology*,  
416 2004, **38**, 3047-3055.
- 417 23 G. G. Ying, R. S. Kookana and A. Kumar, *Environmental Toxicology And Chemistry*, 2008, **27**,  
418 87-94.
- 419 24 M. Muller, S. Combalbert, N. Delgenes, V. Bergheaud, V. Rocher, P. Benoit, J. P. Delgenes, D.  
420 Patureau and G. Hernandez-Raquet, *Chemosphere.*, 2010, **81**, 65-71.
- 421 25 J. Kang, G. Du, X. Gao, B. Zhao and J. Guo, *Water Environ. Res.*, 2014, **86**, 223-231.
- 422 26 D. J. Barker and D. C. Stuckey, *Water Research*, 1999, **33**, 3063-3082.

- 423 27 Allen W. , Prescott W. B. , Derby R. E. , Garland C. E. , Peret J. M. and S. M., *Determination of*  
424 *color of water and wastewater by means of ADMI color values [M]*, Eng Ext Ser, In Proceeding of  
425 28th Industrial Waste Conference, West, 1973.
- 426 28 W. J. Huang and H. H. Yeh, *Journal Of Environmental Science And Health*, 1997, **32**, 2311-2336.
- 427 29 J. Nishikawa, K. Saito, J. Goto, F. Dakeyama, M. Matsuo and T. Nishihara, *Toxicology And Applied*  
428 *Pharmacology*, 1999, **154**, 76-83.
- 429 30 J. Jung, K. Ishida and T. Nishihara, *Life Sci.*, 2004, **74**, 3065-3074.
- 430 31 P. Westerhoff, W. Chen and M. Esparza, *Journal Of Environmental Quality*, 2001, **30**, 2037-2046.
- 431 32 J. I. MacGregor and V. C. Jordan, *Pharmacological Reviews*, 1998, **50**, 151-196.
- 432 33 D. M. Lonard and C. L. Smith, *Steroids*, 2002, **67**, 15-24.
- 433 34 N. Garcia-Reyero, E. Grau, M. Castillo, M. J. L. De Alda, D. Barcelo and B. Pina, *Environmental*  
434 *Toxicology And Chemistry*, 2001, **20**, 1152-1158.
- 435 35 H. Yamamoto, H. M. Liljestrand and Y. Shimizu, *Environmental Science & Technology*, 2004, **38**,  
436 2351-2358.
- 437 36 R. D. Holbrook, N. G. Love and J. T. Novak, *Environmental Science & Technology*, 2004, **38**,  
438 3322-3329.
- 439 37 J. Janosek, M. Bittner, K. Hilscherova, L. Blaha, J. P. Giesy and I. Holoubek, *Chemosphere.*, 2007,  
440 **67**, 1096-1101.
- 441 38 W. Chen, P. Westerhoff, J. A. Leenheer and K. Booksh, *Environmental Science & Technology*, 2003,  
442 **37**, 5701-5710.
- 443 39 M. M. Ferreira, M. C. LoureiroDias and V. Loureiro, *International Journal Of Food Microbiology*,  
444 1997, **36**, 145-153.

- 445 40 R. Wattanadilok, P. Sawangwong, C. Rodrigues, H. Cidade, M. Pinto, E. Pinto, A. Silva and A.  
446 Kijjoa, *Marine Drugs*, 2007, **5**, 40-51.
- 447 41 J. S. Dambolena, A. G. Lopez, J. M. Meriles, H. R. Rubinstein and J. A. Zygodlo, *Food Control*,  
448 2012, **28**, 163-170.
- 449 42 L. C. Duarte, F. Carvalheiro, I. Neves and F. M. Girio, *Applied Biochemistry And Biotechnology*,  
450 2005, **121**, 413-425.
- 451 43 B. J. Bruschweiler, S. Kung, D. Burgi, L. Muralt and E. Nyfeler, *Regul. Toxicol. Pharmacol.*, 2014,  
452 **69**, 263-272.
- 453 44 B. Perlatti, M. da Silva, J. B. Fernandes and M. R. Forim, *Bioresource Technology*, 2012, **124**,  
454 37-44.
- 455 45 B. Pavan, C. Biondi, M. E. Ferretti, L. Lunghi and G. Paganetto, *Toxicology Letters*, 2001, **118**,  
456 157-164.
- 457 46 F. Maradonna, M. Evangelisti, G. Gioacchini, B. Migliarini, I. Olivotto and O. Carnevali, *Toxicol. In*  
458 *Vitro*, 2013, **27**, 84-91.
- 459

## Graphical Abstract



HOA and HON were key fractions involved in increasing antiestrogenic activity and humic/fulvic acid in them could mask estrogenic activity.