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# Synthesis of polypyrrole nanoparticles for constructing full-polymer UV/NIR-shielding film

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**Abstract:** A prerequisite for the selective shielding of solar light is to develop optical materials and coatings, and traditional shielding materials usually combine rare/expensive metal elemental and also have potential heavy-metal pollution after being abandoned. To solve this problem, herein we develop polypyrrole (PPy) nanoparticles as a novel kind of metal-free ultraviolet (UV)/near-infrared (NIR) shielding material. PPy nanoparticles with diameter of ~50 nm are synthesized by a simple solution polymerization route, and they exhibit weak absorption in visible region but strong UV/NIR photoabsorption. Subsequently, PPy nanoparticles are mixed with polyacrylic acid (PAA) resin for the preparation of PPy-PAA full-polymer films. PPy-PAA films exhibit good transparency in visible region (400-780 nm) but can efficiently absorb UV (305-400 nm) and NIR (780-2500 nm) light, for example, 0.34-mm-thick film with 0.05 wt% PPy can transmit 63.1% visible light but shield 47.2% UV and 80.9% NIR light. When this PPy-PAA film coated glass is used as the window of the sealed black box, the interior air temperature of the box goes up from room temperature of 25.0 °C to 29.2 or 33.9 °C in 1500 s under the irradiation of strong solar light (0.3 or 0.5 W cm<sup>-2</sup>). Its temperature elevation (4.2 or 8.9 °C) is remarkably lower compared with that (7.3 or 15.7 °C) from glass slide as window under the other identical condition, resulting from excellent NIR shielding property of PPy. Therefore, PPy nanoparticles have great potential as

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21 a novel UV/NIR shielding material for the development of cost-efficient energy-saving  
22 full-polymer windows without potential heavy-metal pollution.

23 **Keywords:** polypyrrole nanoparticles; synthesis; UV/NIR; shielding film; energy-saving  
24 windows

25

## 26 1. Introduction

27 Optical materials and coatings have played a vital role in selectively shielding solar light  
28 for meeting the growing demand of thermohygro-metric and environmental comfort as well  
29 as improving the energy efficiency of buildings (or automobile).<sup>1, 2</sup> The most ideal optical  
30 materials and coatings should be smart glass which changes between translucent and  
31 transparent when voltage/light/heat is applied. The typical smart glass includes electrochromic  
32 (such as  $\text{WO}_{3-x}$  and  $\text{NiO}_x$ -based),<sup>3, 4</sup> photochromic (such as  $\text{WO}_3$  and  $\text{MoO}_3$ -based),<sup>5, 6</sup>  
33 thermochromic (such as  $\text{VO}_2$ -based)<sup>7</sup> devices, which results in great contributions to the  
34 progress in selectively shielding solar light. However, smart glass is still immature for the  
35 large-scale practical applications in buildings and automobile. In fact, for selectively shielding  
36 solar light in most applications, it would be the most simple and affordable choice to directly  
37 use semi-transparent heat-insulation coating which can absorb/reflect part of solar light. It is  
38 well known that solar light is chiefly concentrated in the wavelength range between 0.2 and  
39  $2.5 \mu\text{m}$ , including ultraviolet (UV,  $0.2\sim 0.4 \mu\text{m}$ , 4% of total energy), visible ( $0.4\sim 0.78 \mu\text{m}$ , 46%  
40 of the total energy), near infrared (NIR,  $0.78\sim 4.5 \mu\text{m}$ , ~50% of total energy) light. For

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41 simultaneously satisfying visual effects and reducing the heating/adverse effect, it is  
42 necessary to develop UV/NIR-shielding coating which can only transmit visible light but cut  
43 off both NIR and UV light. The use of this UV/NIR shielding coating will prevent a  
44 temperature elevation in a room in summer and realize heat-insulating in winter, and it is also  
45 desirable for health care.

46 Currently, the widely used photo-shielding coating should be metal (such as Ag, Cu)  
47 film,<sup>8,9</sup> which can reflect part of visible and IR light. These metal films are usually prepared  
48 by vacuum thermal evaporation or magnetron sputtering, which is too costly and would  
49 hinder large-scale production. To solve this problem, the solution-process of metal-based  
50 nanomaterials and the corresponding nanomaterial-contained polymer film have been  
51 demonstrated to be a good alternative to conventional metal films. The key for this  
52 technology is to obtain efficient UV/NIR shielding nanomaterials. Three kinds of UV/NIR  
53 shielding nanomaterials have been well developed. The first one is noble metal nanoparticles,  
54 such as Ag/Au nanoparticles,<sup>10</sup> but it shows low visible light transparency. The second one is  
55 rare-earth hexaboride nanoparticles, but it can only absorb certain wavelengths of NIR and the  
56 preparation process need the complex high temperature (~1500 °C) and vacuum conditions.<sup>11</sup>  
57 <sup>12</sup> The last kind is metal-oxide semiconductor nanomaterials, which should be the most  
58 studied UV/NIR shielding nanomaterials. For example, both tin-doped indium oxide (ITO)<sup>13</sup>,  
59 <sup>14</sup> and Al-doped zinc oxide (AZO)<sup>15, 16</sup> nanoparticles have been demonstrated to exhibit NIR  
60 photoabsorption ability, but they are famous transparent conductive materials and can only  
61 shield NIR light with wavelength longer than 1500 nm. To broaden NIR photoabsorption  
62 range, tungsten (W)-based nanomaterials have been well developed, including  $W_{18}O_{49}$ <sup>17, 18</sup>  
63 and  $M_xWO_3$  ( $M^{x+} = Na^+, K^+, Rb^+, Cs^+$  and  $NH_4^+$ ).<sup>19-22</sup> These W-based nanomaterials can

64 transmit visible light but shield NIR light, resulting in great contributions to the progress in  
65 shielding materials. It should be noted that all these metal-based materials involve in the  
66 utilization of expensive and/or rare metal (In, W, et al), and the abandon of the  
67 heavy-metal-contained films would deteriorate the environment, both of which would partly  
68 limit their application ranges. Therefore, it is very necessary to develop novel kind of  
69 cost-efficient UV/NIR shielding materials without metal component.

70 Recently, several kinds of NIR light-driven photothermal nanoagents have been well  
71 developed for biomedical applications, inducing polymer (such as polypyrrole (PPy)),<sup>23, 24</sup>  
72 metal (such as Au),<sup>25, 26</sup> carbon (such as graphene),<sup>27</sup> semiconductors (such as CuS).<sup>28</sup> Our  
73 group has developed W-based ( $W_{18}O_{49}$ <sup>29, 30</sup> and  $Cs_xWO_3$ <sup>31</sup>) nanomaterials, Cu-based ( $CuS$ <sup>32</sup>,  
74  $Cu_9S_5$ <sup>33</sup>,  $Fe_3O_4@Cu_{2-x}S$ <sup>34</sup>) nanomaterials as efficient NIR photothermal agents. All these  
75 photothermal nanoagents exhibit very strong photoabsorption in NIR region. Compared with  
76 W-based and Cu-based nanomaterials, we also found that polypyrrole (PPy) nanoagents have  
77 very broad UV/NIR photoabsorption range and they are metal-free. These features trigger our  
78 interest in developing PPy nanomaterials as a new kind of cost-efficient UV/NIR shielding  
79 agent without potential heavy-metal pollution. In the present work, we have prepared PPy  
80 nanoparticles with diameter of ~50 nm by one-step aqueous dispersion polymerization. PPy  
81 nanoparticles exhibit high transparency in visible region (400-780 nm) but absorb efficiently  
82 UV/NIR light. With PPy nanoparticles and polyacrylic acid (PAA) resin as the mode, flexible  
83 PPy-PAA composite films are prepared by using coating/drying technology. One of typical  
84 PPy-PAA films (thickness: 0.34 mm, PPy content: 0.05 wt%) can transmit 63.1% visible light  
85 (400-780 nm) but shield 47.2% UV and 80.9% NIR light. Importantly, with this PPy-PAA  
86 film coated glass as the window of the sealed box, the interior air temperature exhibits

87 obviously low elevation (4.2 or 8.9 °C) compared with that (7.3 or 15.7 °C) from glass slide  
88 as the window, under the irradiation of solar light with high intensity (0.3 or 0.5 W cm<sup>-2</sup>).

89

## 90 2. Experimental details

### 91 2.1 Materials

92 All of the chemicals are commercially available and were used without further  
93 purification. Iron (III) chloride hexahydrate (FeCl<sub>3</sub>•6H<sub>2</sub>O, 99%), polyvinyl alcohol (PVA,  
94 molecular weight ~1750), pyrrole monomer (98%) were received from Sinopharm Chemical  
95 Reagent Co., Ltd (China). Polyacrylic acid (PAA, molecular weight ~20000) resin was  
96 purchased from HanwHA Corporation.

### 97 2.2 Synthesis of PPy nanoparticles

98 PPy nanoparticles were synthesized by a modified one-step aqueous dispersion  
99 polymerization method.<sup>24</sup> FeCl<sub>3</sub>•6H<sub>2</sub>O (4.6 mmol, 1.2347 g) and PVA (1.5 g) were dissolved  
100 into the deionized water (50 mL) at room temperature under magnetically stirring, forming a  
101 transparent yellow solution. After 1 hour equilibration, the solution was transferred to an  
102 ice-water bath. Subsequently, pyrrole monomer (2.0 mmol, ~140 μL) was added into the  
103 above solution. When pyrrole monomer contacted the oxidizing agent (FeCl<sub>3</sub>), polymerization  
104 reaction proceeded immediately. The mixture solution became black within a few minutes  
105 and was continuously stirred for 4 hours. After the completion of polymerization, the system  
106 was boosted to room temperature naturally. The resulting sample was separated via  
107 centrifugation and was washed three times with hot water to remove impurities. Finally, the  
108 precipitation was dispersed in the deionized water with concentration of 0.5 mg mL<sup>-1</sup>, and the  
109 aqueous dispersion was stored at refrigerator at 4 °C for further use.

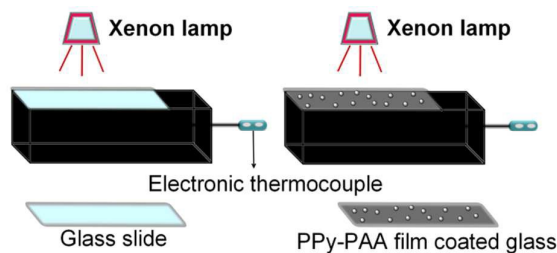
### 110 2.3 Preparation of PPy-PAA films

111 In a typical film synthesis process, PAA resin (2.0 g, ~2.0 mL) was dispersed in the  
112 deionized water (2.0 mL) under magnetically stirring. Then this PAA resin dispersion was  
113 mixed with PPy aqueous dispersion (2.0 mL, 0.5 mg mL<sup>-1</sup>) under stirring, forming a  
114 homogeneous dark colloidal dispersion (slurry). Subsequently, the above slurry with different  
115 volumes (1.0, 1.5, 2.0 mL) was coated on the glass slides, respectively. Then the glass slides  
116 were respectively placed on a hotplate at 50 °C to remove solvent, until PPy-PAA films were  
117 solidified. PPy-PAA full-polymer films with 0.05 wt% PPy and different thickness (~0.34,  
118 0.50, 0.60 mm) could be easily peeled off from the glass slides. For comparison, 0.50  
119 mm-thick PPy-PAA films with different PPy content (0.10 wt%, and 0.125 wt%) were also  
120 prepared by changing the volume (4.0 or 5.0 mL) of suspension containing PPy nanoparticles  
121 with the other identical conditions.

### 122 2.4 Characterization and NIR shielding measurements

123 The morphologies of PPy nanoparticles and PPy-PAA films were investigated by using a  
124 field emission-scanning electron microscopy (FE-SEM, Hitachi S-4800) and transmission  
125 electron microscopy (TEM, JEOL JEM-2100F). The Fourier transform infrared (FT-IR)  
126 spectrum was measured from PPy sample in KBr pellets using an IRPrestige-21 spectrometer  
127 (Shimadzu). UV-vis absorbance spectrum of PPy suspension was recorded on a Shimadzu  
128 UV-2550 UV-visible-NIR spectrophotometer using a quartz cuvette with an optical path of 10  
129 mm. The thickness of PPy-PAA composite films was measured by vernier caliper. In order to  
130 measure its photothermal effect, PPy powder (0.5 mg) was put on the white paper. PPy  
131 sample was irradiated by a Xenon lamp with the intensity of 0.5 W cm<sup>-2</sup>, and its temperature  
132 was real-time recorded by an infrared thermal imaging camera (FLIR A300).

133 To evaluate NIR shielding performance, the transmittance spectra of PPy-PAA films  
134 were measured on a UV-visible-NIR spectrophotometer from 300 to 2500 nm (Shimadzu  
135 UV-3600). To further measure the heat-insulation performance, we constructed two sealed  
136 boxes, as illustrated in Fig. 1, where glass slides or PPy-PAA film-coated glass ( $2 \times 6 \text{ cm}^2$ ) was  
137 used respectively as the window. The environmental air temperature was  $25 \text{ }^\circ\text{C}$ . An adjusted  
138 xenon lamp (PLS-SXE300/300UV, Beijing Perfect Light Co. Ltd., Beijing) was used as the  
139 simulated solar light source and the light intensity was independently calibrated using a  
140 hand-held optical power meter (Newport model 1918-C, CA, USA). For amplifying the  
141 heating effect of solar light, solar light with high intensity ( $0.3$  or  $0.5 \text{ W cm}^{-2}$ ) was used to  
142 illuminate the box through the window, and then the air-temperature in the box was real-time  
143 recorded by using an electronic thermometer that should not be directly illuminated by light.



144 **Fig. 1** Schematic illustration of the sealed black boxes, where only the top facet was covered  
145 by glass slides or PPy-PAA film coated glass.  
146  
147

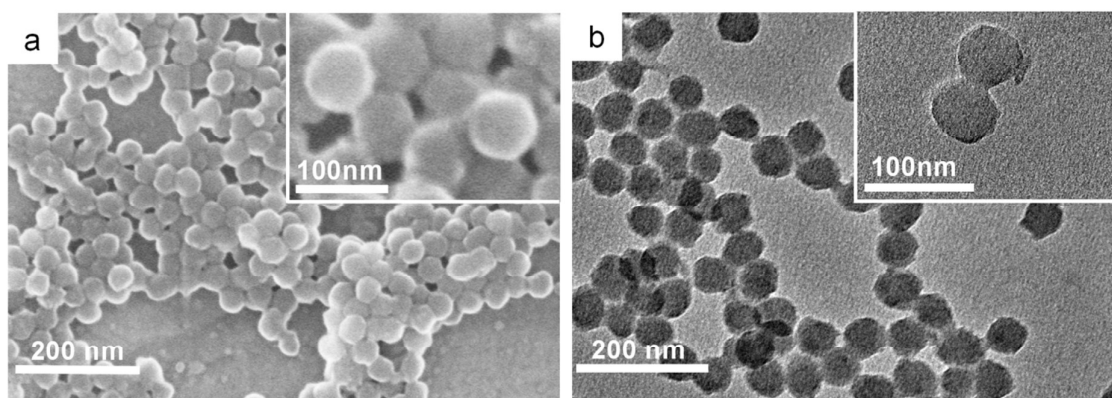
### 148 3. Results and discussion

#### 149 3.1 Preparation and characterization of PPy nanoparticles

150 PPy nanoparticles were synthesized by a modified one-step aqueous dispersion  
151 polymerization with PVA as the stabilizer and  $\text{FeCl}_3$  as the oxidizing agent.<sup>23</sup> The morphology  
152 and size of PPy sample were studied by SEM images (Fig. 2a). Obviously, PPy sample  
153 consists of exclusively spherical nanoparticles with narrow particle size distribution, and the  
154 average size is about 50 nm. In addition, TEM images (Fig. 2b) further confirm the formation

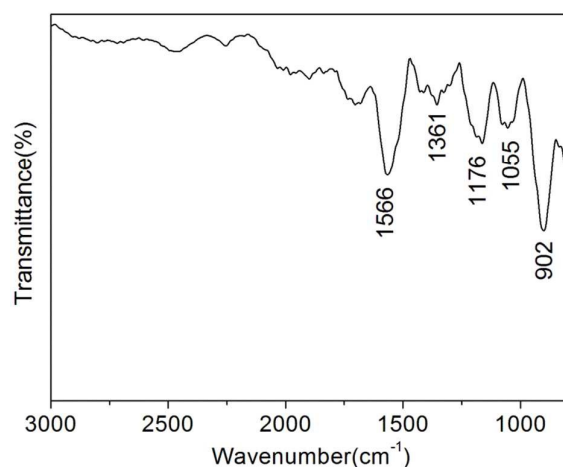


155 of PPy nanospheres with diameter of  $\sim 50$  nm.



156

157 **Fig. 2** SEM (a) and TEM (b) images of PPy nanoparticles.



158

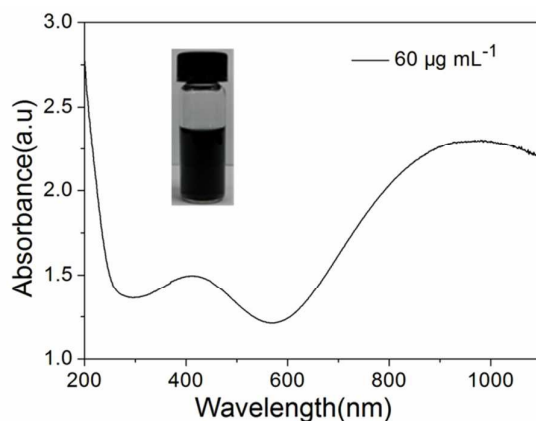
159 **Fig. 3** FTIR spectrum of PPy nanoparticles.

160

161 Subsequently, PPy nanoparticles were identified by FTIR spectrum (Fig. 3). Obviously,  
162 PPy nanoparticle sample exhibits a band at  $1566\text{ cm}^{-1}$ , which is assigned to the stretching  
163 vibration of  $\text{C}=\text{C}$  units.<sup>35</sup> The band at  $1361\text{ cm}^{-1}$  is corresponding to the stretching vibration  
164 of  $\text{C}-\text{N}$  adsorption.<sup>36</sup> The bands at  $1176$  and  $1055\text{ cm}^{-1}$  are attributed to the  $=\text{C}-\text{N}$  in-plane  
165 deformation vibrations. In addition, the broad band near  $902\text{ cm}^{-1}$  is corresponds to  $=\text{C}-\text{N}$   
166 out-plane deformation vibration.<sup>35, 37</sup> Based on the above results, one can confirm the  
167 formation of PPy material.<sup>36</sup>

168 The aqueous dispersion containing PPy nanoparticles ( $60\text{ }\mu\text{g mL}^{-1}$ ) exhibits a strong

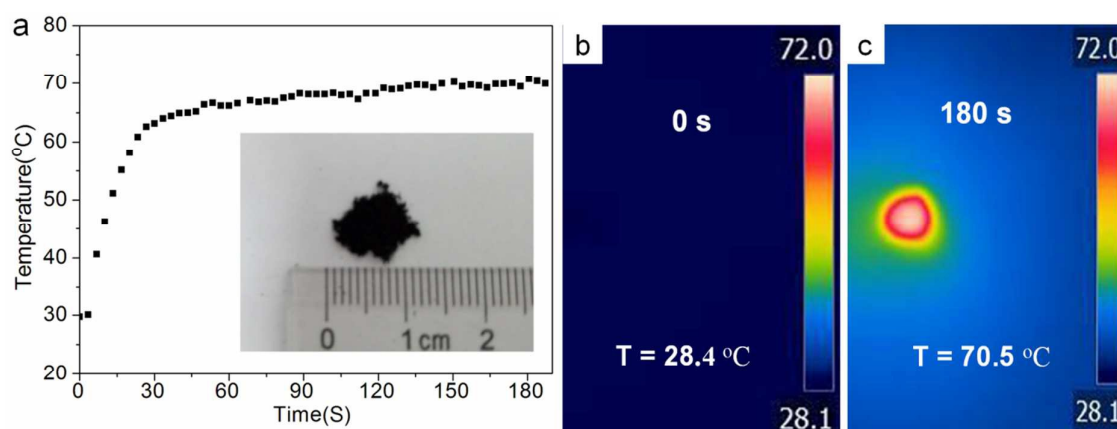
169 black color (the inset in Fig. 4), and has high stability, even remaining unchanged after several  
170 weeks at 4 °C. In addition, the optical property of the aqueous dispersion was studied by using  
171 UV-vis-NIR spectroscopy (Fig. 4). The spectrum is similar to that in the previous reports on  
172 PPy nanoparticles,<sup>23, 24</sup> and it exhibits the short-wavelength absorption edged at  
173 approximately 250 nm. Obviously, the sample has low absorbance in the visible region  
174 (400-650 nm) with the lowest absorbance of 1.21 at 570 nm, indicating the fact that part of  
175 visible light can be transmitted through PPy solution. Importantly, the spectrum shows an  
176 increased absorption with the increase of wavelength from 570 to 988 nm, where the  
177 maximum extinction coefficient at 988 nm was calculated to be  $3.83 \times 10^4 \text{ cm}^2 \text{ g}^{-1}$ . The  
178 absorption intensity in the near-IR region (1000-1100 nm) is also very high. The broad  
179 absorption band from visible region to NIR region is the characteristics of the bipolaronic  
180 metallic state of doped polypyrrole.<sup>23, 24, 38</sup>



181  
182 **Fig. 4** UV-vis-NIR absorption spectrum of aqueous dispersion containing PPy nanoparticles  
183 ( $60 \mu\text{g mL}^{-1}$ ). Inset: Photo of the aqueous dispersion.  
184

185 As a result of their specific photoabsorption, PPy nanoparticles can absorb part of visible  
186 light while absorb strongly NIR light, probably resulting in the efficient photothermal  
187 conversion. To further investigate the photothermal performance, PPy nanoparticle powder  
188 ( $\sim 0.5 \text{ mg}$ ) was directly put on the white paper (the inset in Fig. 5a). The temperature

189 distribution was recorded by thermal imaging camera under the irradiation of simulated solar  
190 light with the intensity of  $0.5 \text{ W cm}^{-2}$  (Fig. 5a). Before the light illumination, both PPy  
191 powder and the white paper remain the room temperature of  $28.4 \text{ }^\circ\text{C}$  (Fig. 5b). When the solar  
192 light is turned on, the maximum temperature of PPy powder goes up rapidly with the  
193 increasing time to 30 s (Fig. 5a), and then exhibits a relatively slow increase to  $70.5 \text{ }^\circ\text{C}$  at 180  
194 s, as vividly shown in the thermographic image (Fig. 5c). The heating rate becomes slow with  
195 the increase of irradiated time, resulting from the faster heat loss at higher temperatures.<sup>29,32</sup>  
196 Interestingly, white paper away from PPy sample remains low temperature of  $\sim 30.5 \text{ }^\circ\text{C}$  (Fig.  
197 5c), although part of them was also irradiated simultaneously. Therefore, one can conclude  
198 that PPy nanoparticles can absorb solar light and convert efficiently it to heat.



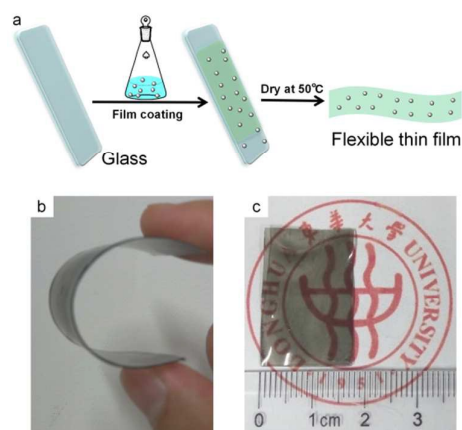
199  
200 **Fig. 5** (a) The temperature elevation of PPy powder on the white paper as a function of  
201 irradiation time with solar light ( $0.5 \text{ W cm}^{-2}$ ), the inset is the photo; (b,c) Thermographic  
202 images at 0 and 180 s irradiation.

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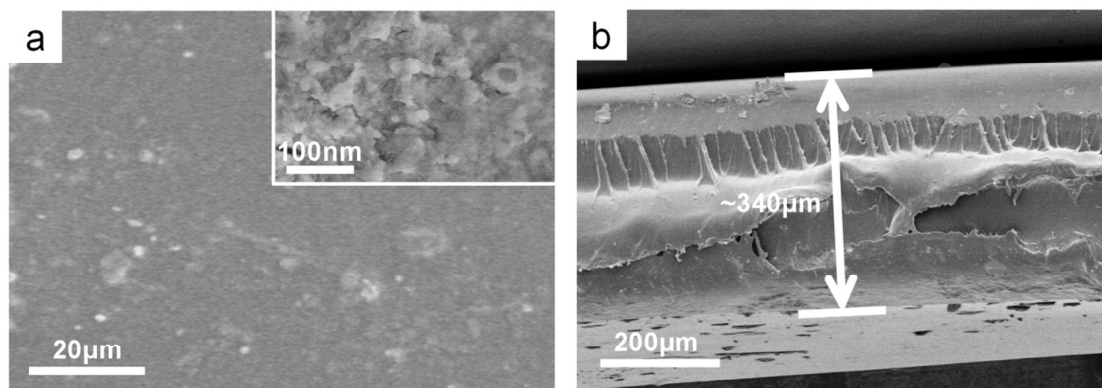
### 204 3.2 Synthesis and NIR shielding property of PPy-PAA composite film

205 For satisfying visual effects but reducing the heating effect from solar light, it is  
206 important to develop UV/NIR shielding film which can transmit only visible light but cut off  
207 UV/NIR light. PPy nanoparticles exhibit good transparency in visible region and strong  
208 absorption in UV/NIR region, which motivates us to investigate their potential in UV/NIR

209 shielding films for buildings or automobile. In the present study, PPy nanoparticles were used  
 210 as the photoabsorption agent, and PAA resin was used as the polymer matrix due to its  
 211 advantages such as non-toxicity and optical transparence. PPy-PAA full polymer films were  
 212 prepared by using coating-drying technology (Fig. 6a), where the slurries of PAA resin  
 213 containing PPy nanoparticles were respectively coated on the glass slides, and then the slides  
 214 were dried at 50 °C to solidify films (Fig. 6a). These films could be easily peeled off from the  
 215 glass slides. Three PPy-PAA composite films with 0.05 wt% PPy and different thickness  
 216 (about 0.34, 0.50, 0.60 mm) were prepared, and two 0.50-mm-thick films with different PPy  
 217 content (0.10 wt% and 0.125 wt%) were also obtained.



218  
 219 **Fig. 6** (a) Schematic illustration of preparation process of the flexible PPy-PAA films. (b, c)  
 220 Typical photographs of 0.34-mm-thick PPy-PAA film with 0.05 wt% PPy.

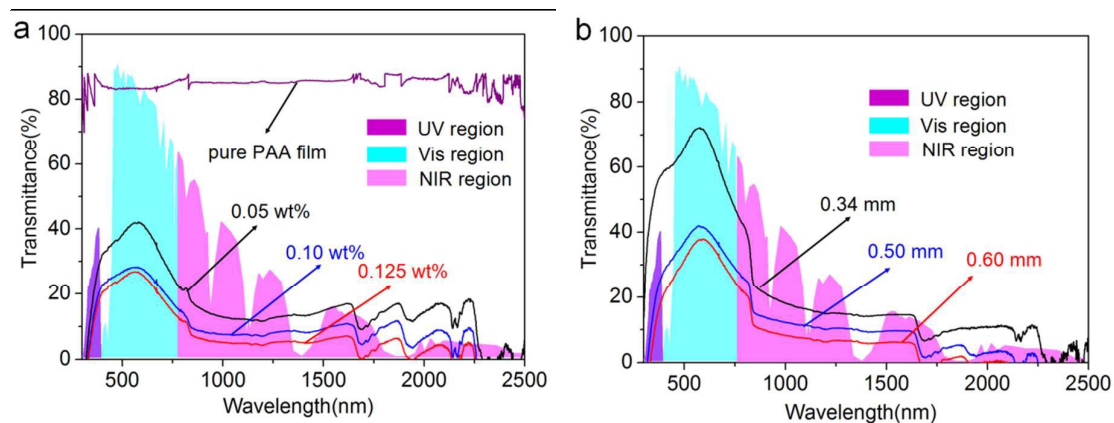


221  
 222 **Fig. 7** Typical surface (a) and cross-section (b) morphologies of 0.34-mm-thick PPy-PAA film  
 223 with 0.05 wt% PPy.

224 All these films have similar appearance, and they are light-black in color. Herein,  
225 0.34-mm-thick PPy-PAA film with 0.05 wt% PPy was used as a model for the subsequent  
226 morphology analysis. It is clear that this film is free-standing and can be reversibly bent at  
227 large angles (0~360°) (Fig. 6b), resulting from the excellent flexibility of PAA matrix.  
228 Furthermore, this PPy-PAA film remains relatively transparent (Fig. 6c), which is favorable to  
229 keep good visual effects when being used as the window. Subsequently, we investigated the  
230 surface and cross-section morphologies by SEM images (Fig. 7). PPy-PAA film has very  
231 uniform and smooth surface without crack (Fig. 7a). From SEM image with higher  
232 magnification (the inset in Fig. 7a), one can find that there are many nanoparticles with good  
233 dispersibility in PAA film, suggesting that PPy nanoparticles was well encapsulated in PAA  
234 matrix. Undoubtedly, this encapsulation will avoid the direct contact between PPy  
235 nanoparticles and air/water, probably conferring the good weatherability of PPy-PAA film. In  
236 addition, the cross-sectional SEM image confirms that the thickness of this PPy-PAA film is  
237 ~0.34 mm (Fig. 7b), which agrees well with the data measured by vernier caliper.

238 The optical properties of these PPy-PAA films were studied by UV-visible-NIR  
239 spectrophotometer. Firstly, we investigated the effects of PPy content (0.05, 0.10, 0.125 wt%)  
240 on the transmittance of 0.50-mm-thick PPy-PAA films (Fig. 8a). For comparison, the  
241 transmittance spectrum of 0.5-mm-thick pure PAA film was also measured. Pure PAA film  
242 exhibits high transmittance ( $\geq 80\%$ ) in the entire UV-vis-NIR region (300-2500 nm),  
243 indicating the very low photoabsorption in the broad wavelength range. In addition, these  
244 three PPy-PAA films exhibit relatively high transmittance in visible region but low  
245 transmittance in UV and NIR region. With the increase of PPy content from 0.05 to 0.125  
246 wt%, the entire transmittance goes down, where the maximum transmittance ( $T_{max}$ ) at 579 nm

247 drops from 41.6% to 26.4%, and the average transmittance ( $T_{adv}$ ) in 1100-2000 nm also  
 248 decreases from  $\sim 15\%$  to  $\sim 6\%$ . In addition, we studied the change of the transmittance with  
 249 film thickness by using PPy-PAA film with 0.05 wt% PPy and different thickness (0.34, 0.50,  
 250 0.60 mm) as the model (Fig. 8b). When the film thickness raises from 0.34 to 0.60 mm, the  
 251 entire transmittance also declines, where  $T_{max}$  at 579 nm drops from 71.9% to 37.5% and  $T_{adv}$   
 252 in the wavelength of 1100-2000 nm goes down from  $\sim 18\%$  to  $\sim 10\%$ . All these results suggest  
 253 that PPy-PAA films can transmit part of visible but absorb strongly NIR light. Undoubtedly,  
 254 this visible transmittance and NIR shielding behavior of PPy-PAA films should be attributed  
 255 to the presence of PPy nanoparticles, since PPy nanoparticles can absorb efficiently NIR light  
 256 and then convert it to heat (Fig. 4,5) while PAA matrix has no obvious photoabsorption in a  
 257 broad wavelength range (300-2500 nm). Furthermore, we also find that the visible  
 258 transmittance and NIR shielding effects can be well controlled by adjusting PPy content  
 259 and/or film thickness. For example, higher PPy content facilitates lower transmittance in NIR  
 260 region, indicating higher NIR shielding effect.



261  
 262 **Fig. 8** Transmittance spectra of PPy-PAA films with (a) 0.50-mm-thickness with different PPy  
 263 content (0, 0.05, 0.10, 0.125 wt%) and (b) 0.05 wt% PPy with different thickness (0.34, 0.50,  
 264 0.60 mm). For comparison, standard solar spectrum is also supplied.

265

266 Ideally, the photo-shielding film can only transmit visible light but cut off both NIR and



267 UV light. To determine the transmitted efficiency (**TE**) in visible region and shielding  
 268 efficiency (**SE**) in NIR/UV region, we assume that PPy-PAA films are irradiated by a standard  
 269 solar light (AM 1.5, 100 mW cm<sup>-2</sup>), where the standard solar light spectrum are also shown in  
 270 Fig. 8a,b. According to the transmittance spectra and standard solar spectrum (Fig. 8a,b), the  
 271 transmitted intensity ( $I_T$ ) of different light (UV: 305-400 nm, VIS: 400-780 nm, NIR:  
 272 780-2500 nm) can be obtained by the equation (1):

$$273 \quad I_T = \int F(\lambda)T(\lambda)d\lambda \quad (1)$$

274 where  $F(\lambda)$  is the incident photon flux intensity at wavelength  $\lambda$ , and  $T(\lambda)$  is the transmittance  
 275 at wavelength  $\lambda$ . **TE** of different light can easily be obtained by the equation (2):

$$276 \quad \mathbf{TE} = \frac{I_T}{I_{total}} \times 100\% \quad (2)$$

277 where  $I_{total}$  is the total light intensity of different kind of solar light. **SE** of different light is the  
 278 difference of 1 and its corresponding **TE** (such as,  $\mathbf{SE}_{NIR} = 1 - \mathbf{TE}_{NIR}$ ). The data of  $I_T$ , **TE** and  
 279 **SE** were summarized in Table 1.

280  
 281 **Table 1** The data of transmitted intensity ( $I_T$ ), transmitted efficiency (**TE**), shielding  
 282 efficiency (**SE**) and solar energy transmittance selectivity (SETS) for different (UV, Vis, NIR)  
 283 light by different PPy-PAA films.

Type of PPy-PAA film	Transmitted Intensity (mW/cm <sup>2</sup> )			Transmitted Efficiency (TE,%)			Shielding Efficiency (SE,%)			SETS
	UV	VIS	NIR	UV	VIS	NIR	UV	VIS	NIR	
0.50-mm+0.05 wt%	0.848	16.7	6.11	20.6	36.2	14.4	79.4	63.8	85.6	1.22
0.50-mm+0.10 wt%	0.561	11.4	3.52	13.6	24.7	8.28	86.4	75.3	91.7	1.16
0.50-mm+0.125 wt%	0.478	10.4	2.32	11.6	22.6	5.46	88.4	77.4	94.5	1.17
0.34-mm+0.05 wt%	2.16	29.1	8.08	52.8	63.1	19.1	47.2	36.9	80.9	1.44
0.60-mm+0.05 wt%	0.491	14.9	3.38	11.9	32.1	7.95	88.1	67.9	92.1	1.24

284 <sup>a</sup>The total light intensity from standard solar: UV (305-400nm)=4.11mW/cm<sup>2</sup>, VIS (400-780  
 285 nm)=46.2 mW cm<sup>-2</sup>, NIR (780-2500nm)=40.3 mW cm<sup>-2</sup>.

286

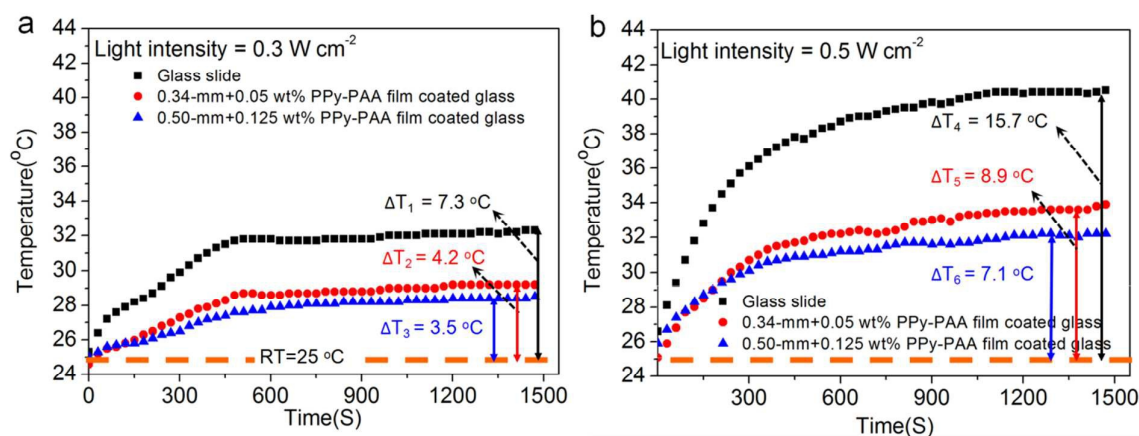
287 For 0.50-mm thick PPy-PAA films with increasing PPy content from 0.05 wt% to 0.125  
 288 wt%, **TE** of visible light goes down from 36.2% to 22.6%, while **SE** of NIR light raises from

289 85.6% to 94.5%. Similarly, when the thickness of 0.05 wt% PPy film goes up from 0.34 to  
290 0.60 mm, **TE** of visible light decreases from 63.1% to 32.1% while **SE** of NIR light increase  
291 from 80.9% to 92.1%. To evaluate the integrated optical performance of these PPy-PAA films,  
292 the solar energy transmittance selectivity (**SETS**) is denoted as the sum of **TE** of visible light  
293 and **SE** of NIR light. Since UV light has very low intensity in solar spectrum, herein we  
294 neglect its effect on **SETS**. **SETS** should have the range of 0-2, where 0 represents an  
295 opposite film that has no transmittance in visible region and no shielding ability in NIR region;  
296 and 2 represents an ideal film that can completely transmit visible but cut off all NIR light.  
297 According to Table 1, 0.34-mm-thick PPy-PAA film with 0.05 wt% PPy can transmit 63.1%  
298 visible light but block 80.9 % NIR light, yielding the highest **SETS** of 1.44. In fact, **SETS** can  
299 be tuned by adjusting PPy content and the film thickness (Table 1).

300 As a result of their moderate visible transmittance and high UV/NIR shielding, PPy-PAA  
301 films should have great potential as the semi-transparent heat-insulation coating of  
302 energy-saving windows for simultaneously satisfying visual effects and reducing the heating  
303 effect from solar light. To evaluate their heat-insulation performances, we constructed two  
304 sealed black boxes with glass slide or PPy-PAA film-coated glass ( $2\times 6\text{ cm}^2$ ) as the window  
305 (Fig. 1). Under the irradiation of the simulated solar light with high intensity ( $0.3$  or  $0.5\text{ W}$   
306  $\text{cm}^{-2}$ ), the air-temperature in the box was real-time recorded, as shown in Fig. 9. Obviously,  
307 the interior air temperature goes up rapidly with irradiated time to 300 s, and then exhibits a  
308 relative flat and reaches a maximum at 1500 s. The maximum temperature elevation ( $\Delta T$ ) is  
309 determined to investigate the heat-insulation performance of the window. When the solar light  
310 intensity is  $0.3\text{ W cm}^{-2}$ , the ordinal glass slide as the blank window confers a temperature  
311 elevation ( $\Delta T_1=7.3\text{ }^\circ\text{C}$ ) (Fig. 9a). For two typical PPy-PAA films (0.34-mm-thick+0.05wt%



312 PPy; 0.50-mm-thick+0.125 wt% PPy) coated glass as the test window, the temperature  
 313 elevation is respectively 4.2 and 3.5 °C (Fig. 9a), which is just about 57.5% and 48.6% of that  
 314 ( $\Delta T_1=7.3$  °C) from glass slide. Similarly, under the illumination of solar light with higher  
 315 intensity ( $0.5 \text{ W cm}^{-2}$ ), the temperature elevation from PPy-PAA films (0.34-mm-thick+0.05wt%  
 316 PPy; 0.50-mm-thick+0.125 wt% PPy) coated glass is 8.9 and 7.1 °C (Fig. 9b), which is just  
 317 56.7% and 45.2 % of that ( $\Delta T_4=15.7$  °C) from glass slide. These facts reveal that the  
 318 temperature elevation from PPy-PAA films coated glass is much lower than that from glass  
 319 slide, with a large difference.



320  
 321 **Fig. 9** The changes of the interior air temperature of the box with the glass slide or PPy-PAA  
 322 film (0.34-mm-thick+0.05wt% PPy; or 0.50-mm-thick+0.125 wt% PPy) coated glass as the  
 323 window, as a function of time under the irradiation of solar light with intensity of (a) 0.3 and  
 324 (b)  $0.5 \text{ W cm}^{-2}$ .

325

326 It is well known that the glass slide has very high transparency in the almost entire solar  
 327 spectrum (300-2500 nm). Thus, almost all solar light can transmit glass slide window and  
 328 reach the interior of the box, resulting in strong heating effect and then large temperature  
 329 elevation. Usually, the large temperature elevation easily makes human uncomfortable and  
 330 then increases the energy consumption of buildings or vehicles. Importantly, the introduction  
 331 of PPy-PAA films results in low temperature elevation, since PPy-PAA film can shield part of

332 visible light and almost all NIR light, as shown in (Fig. 8, Table 1). In addition, although  
333 PPy-PAA films are used as the coating, the interior air temperature will also go up inevitably,  
334 since visible light should be transmitted for satisfying visual effects. But we can expect that  
335 the temperature elevation can be well tuned by changing photo shielding effect, for example,  
336 by adjusting PPy content and the film thickness or using the combination of different films.  
337 More importantly, the present PPy-PAA film has low-cost and is metal-free; and the abandon  
338 of the films will not result in heavy-metal pollution, both of which will be favorable for the  
339 design and development of cost-efficient energy-saving windows without potential pollution.

340

#### 341 **4. Conclusions**

342 PPy nanoparticles with size of  $\sim 50$  nm have been synthesized by a modified one-step  
343 aqueous dispersion polymerization method, and they can transmit part of visible light and  
344 absorb efficiently UV/NIR light. With PPy nanoparticles as the photoabsorbing agent,  
345 PPy-PAA full-polymer films have been prepared and remain the moderate visible  
346 transmittance and efficient NIR shielding. Importantly, when PPy-PAA film coated glass is  
347 used as the window of the sealed box, the interior air temperature exhibits low elevation  
348 ( $3.5\sim 8.9$  °C) compared with ( $7.3\sim 15.7$  °C) from glass slide as window, under the irradiation  
349 of the simulated solar light with high intensity ( $0.3, 0.5$  W cm<sup>-2</sup>). Therefore, PPy-PAA full  
350 polymer films have great potential as novel coating in the application of cost-efficient  
351 energy-saving windows without potential pollution

352

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361

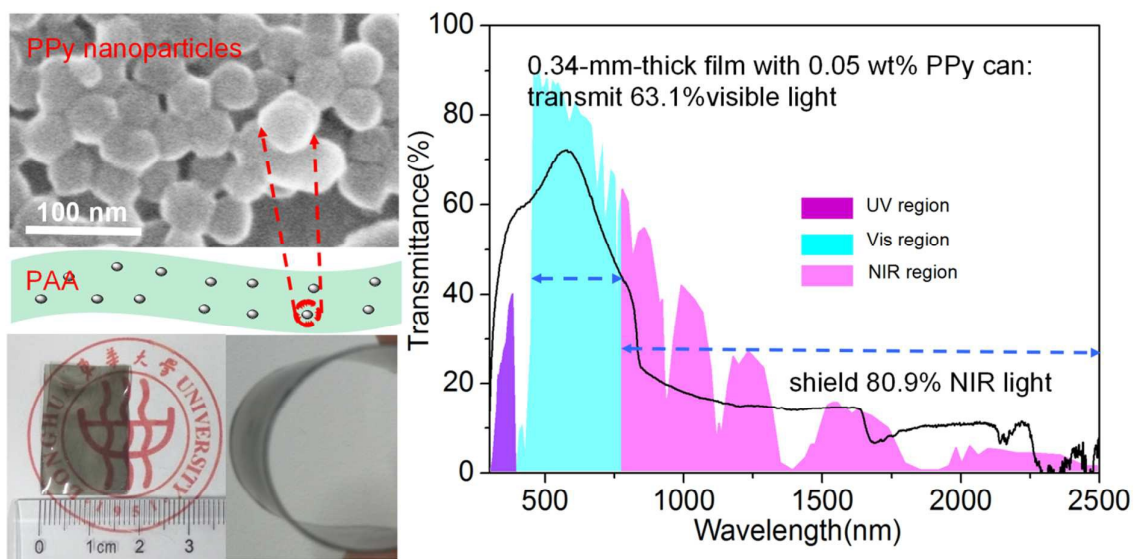
### 362 **Notes and references**

- 363 1. R. Pardo, M. Zayat and D. Levy, *Chem. Soc. Rev.*, 2011, **40**, 672-687.
- 364 2. A. Seeboth, D. Löttsch, R. Ruhmann and O. Muehling, *Chem. Rev.*, 2014, **114**,  
365 3037-3068.
- 366 3. R.-T. Wen, C. G. Granqvist and G. A. Niklasson, *Adv. Funct. Mater.*, 2015, **25**,  
367 3359-3370.
- 368 4. J. Kim, G. K. Ong, Y. Wang, G. LeBlanc, T. E. Williams, T. M. Mattox, B. A. Helms and  
369 D. J. Milliron, *Nano Lett.*, 2015, **15**, 5574-5579.
- 370 5. S. Cong, Y. Tian, Q. Li, Z. Zhao and F. Geng, *Adv. Mater.*, 2014, **26**, 4260-4267.
- 371 6. J. Yao, K. Hashimoto and A. Fujishima, *Nature*, 1992, **355**, 624-626.
- 372 7. B. Fang, Y. Li, G. Tong, X. Wang, M. Yan, Q. Liang, F. Wang, Y. Qin, J. Ding and S.  
373 Chen, *Opt. Mater.*, 2015, **47**, 225-230.
- 374 8. D. Miao, S. Jiang, S. Shang and Z. Chen, *Vacuum*, 2014, **106**, 1-4.
- 375 9. D. Miao, S. Jiang, S. Shang, H. Zhao and Z. Chen, *J. Mater. Sci.: Mater. El.*, 2014, **25**,  
376 5248-5254.
- 377 10. C. L. Tan, S. J. Jang and Y. T. Lee, *Opt. Express*, 2012, **20**, 17448-17455.

- 378 11. T. T. Xu, J.-G. Zheng, A. W. Nicholls, S. Stankovich, R. D. Piner and R. S. Ruoff, *Nano*  
379 *Lett.*, 2004, **4**, 2051-2055.
- 380 12. K. Adachi, M. Miratsu and T. Asahi, *J. Mater. Res.*, 2010, **25**, 510-521.
- 381 13. M. Kanehara, H. Koike, T. Yoshinaga and T. Teranishi, *J. Am. Chem. Soc.*, 2009, **131**,  
382 17736-17737.
- 383 14. A. Llordés, G. Garcia, J. Gazquez and D. J. Milliron, *Nature*, 2013, **500**, 323-326.
- 384 15. D. Miao, S. Jiang, S. Shang, Z. Chen and J. Liu, *Adv. Perform. Mater.*, 2014, **29**, 321-325.
- 385 16. Q. Tan, G. Yu, Y. Liao, B. Hu and X. Zhang, *Colloid Polym. Sci.*, 2014, **292**, 3233-3241.
- 386 17. C. Guo, S. Yin, Y. Huang, Q. Dong and T. Sato, *Langmuir*, 2011, **27**, 12172-12178.
- 387 18. C. Guo, S. Yin, M. Yan, M. Kobayashi, M. Kakihana and T. Sato, *Inorg. Chem.*, 2012, **51**,  
388 4763-4771.
- 389 19. C. Guo, S. Yin, Q. Dong and T. Sato, *CrystEngComm*, 2012, **14**, 7727-7732.
- 390 20. C. Guo, S. Yin, L. Huang and T. Sato, *ACS Appl. Mater. Inter.*, 2011, **3**, 2794-2799.
- 391 21. C. Guo, S. Yin, L. Huang, L. Yang and T. Sato, *Chem. Commun.*, 2011, **47**, 8853-8855.
- 392 22. X. Zeng, Y. Zhou, S. Ji, H. Luo, H. Yao, X. Huang and P. Jin, *J. Mater. Chem. C*, 2015, **3**,  
393 8050-8060.
- 394 23. K. Yang, H. Xu, L. Cheng, C. Sun, J. Wang and Z. Liu, *Adv. Mater.*, 2012, **24**, 5586-5592.
- 395 24. Z. Zha, X. Yue, Q. Ren and Z. Dai, *Adv. Mater.*, 2013, **25**, 777-782.
- 396 25. H. Deng, F. Y. Dai, G. H. Ma and X. Zhang, *Adv. Mater.*, 2015, **27**, 3645-3653.
- 397 26. M. S. Yavuz, Y. Cheng, J. Chen, C. M. Cobley, Q. Zhang, M. Rycenga, J. Xie, C. Kim, K.  
398 H. Song and A. G. Schwartz, *Nat. Mater.*, 2009, **8**, 935-939.
- 399 27. K. Yang, L. Feng, X. Shi and Z. Liu, *Chem. Soc. Rev.*, 2013, **42**, 530-547.
- 400 28. Y. Li, W. Lu, Q. Huang, C. Li and W. Chen, *Nanomedicine*, 2010, **5**, 1161-1171.

- 401 29. Z. Chen, Q. Wang, H. Wang, L. Zhang, G. Song, L. Song, J. Hu, H. Wang, J. Liu and M.  
402 Zhu, *Adv. Mater.*, 2013, **25**, 2095-2100.
- 403 30. W. Xu, Q. Tian, Z. Chen, M. Xia, D. K. Macharia, B. Sun, L. Tian, Y. Wang and M. Zhu,  
404 *J. Mater. Chem. B*, 2014, **2**, 5594-5601.
- 405 31. W. Xu, Z. Meng, N. Yu, Z. Chen, B. Sun, X. Jiang and M. Zhu, *RSC Adv.*, 2015, **5**,  
406 7074-7082.
- 407 32. Q. Tian, M. Tang, Y. Sun, R. Zou, Z. Chen, M. Zhu, S. Yang, J. Wang, J. Wang and J. Hu,  
408 *Adv. Mater.*, 2011, **23**, 3542-3547.
- 409 33. Q. Tian, F. Jiang, R. Zou, Q. Liu, Z. Chen, M. Zhu, S. Yang, J. Wang, J. Wang and J. Hu,  
410 *ACS nano*, 2011, **5**, 9761-9771.
- 411 34. Q. Tian, J. Hu, Y. Zhu, R. Zou, Z. Chen, S. Yang, R. Li, Q. Su, Y. Han and X. Liu, *J. Am.*  
412 *Chem. Soc.*, 2013, **135**, 8571-8577.
- 413 35. M. Omastová, M. Trchová, J. Kovářová and J. Stejskal, *Synthetic. Met.*, 2003, **138**,  
414 447-455.
- 415 36. H. de Oliveira, C. Andrade and C. de Melo, *J. Colloid Interf. Sci.*, 2008, **319**, 441-449.
- 416 37. R. Kostić, D. Raković, S. A. Stepanyan, I. E. Davidova and L. A. Gribov, *J. Chem. Phys.*,  
417 1995, **102**, 3104.
- 418 38. R. B. Bjorklund and B. Liedberg, *Chem. Commun.*, 1986, 1293-1295.

## Graphical Abstract



Polypyrrole (PPy) nanoparticles with diameter of  $\sim 50$  nm were synthesized, and the corresponding flexible PPy-polyacrylic acid (PAA) full-polymer films can transmit visible light but efficiently block UV/NIR light.