

RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This *Accepted Manuscript* will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



Room-Temperature Palladium-Catalysed Suzuki-Miyaura Coupling of Arylboronic Acid with Aryl Chlorides†

Received 00th January 20xx,
Accepted 00th January 20xx

Dan Wang,^{†a} Hong-Guan Chen,^{†a} Xin-Chuan Tian,^a Xiao-Xia Liang,^a Feng-Zhen Chen^b and Feng Gao^{*a}

DOI: 10.1039/x0xx00000x

www.rsc.org/

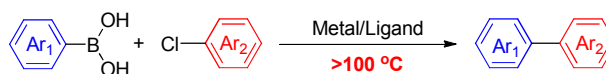
An efficient room-temperature Pd-catalysed Suzuki-Miyaura cross-coupling reaction of arylboronic acids with aryl chlorides is communicated here. The Pd(OAc)₂/NiXantphos catalyst system enables the coupling reaction at room temperature in good to excellent yields (average yield >90%).

The construction of C-C bond is considered to be an important issue in transition-metal catalyzed organic reactions.¹ With the discovery of Suzuki-Miyaura cross-coupling of arylboronic acids with aryl halides in 1981,² a convenient and selective sp² C-C bond forming method was established.³ As boron is similar to carbon in size and electronegativity, both boronic acids and boronate esters are highly nucleophilic, display excellent and broad functional group tolerance. In addition, structurally diverse biaryl motifs obtained from Suzuki reaction are important structural elements of massive natural products, pharmaceuticals, agrochemicals⁴ as well as many bioactive polymers.⁵ It is for these reasons that Suzuki reaction has emerged as a favorite tool in both industrial and academic fields.

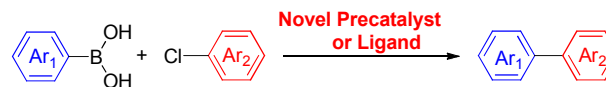
Although aryl chlorides have been proved to be inactive in Suzuki reaction since the very beginning, numerous low-cost compounds available in this class still attract many researchers' attention.⁶ To address this issue, high temperature (>100 °C), along with commercial metal/ligand, was generally employed in early protocols,⁷ which inevitably limits the substrate scope with heat-intolerant heterocyclic aryl halides, such as furans and thiophenes (Scheme 1A); Later, to achieve room-temperature Suzuki coupling reaction with aryl chlorides, some novel ligands/precatalysts were successfully designed and prepared.⁸ However, these studies shared drawbacks like difficult preparation procedures for the ligands/precatalysts (Scheme 1B). It is worth mentioning that some ligand-free researches were highlighted as they achieved room temperature Suzuki reactions in PEG.

Previous Works

A.

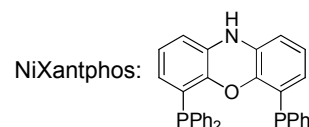
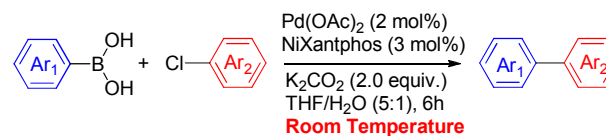


B.



This Works

C.



Scheme 1 Suzuki-Miyaura coupling of aryl chlorides

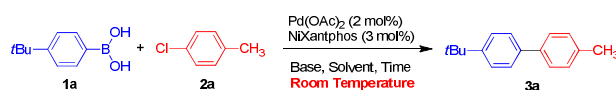
But unfortunately, they were all confined to phenylboronic acid.^{6,9} Therefore, it is still practically and theoretically significant to develop novel synthetic method for room-temperature Suzuki-Miyaura cross-coupling of aryl chlorides.

van Leeuwen's NiXantphos (see Scheme 1 for structure) has been proved to be an efficient ligand in sp³-sp² C-C formation by promoting several arylation reactions,¹⁰ yet its application in sp²-sp² has not been reported. In our preliminary work, we have disclosed the first direct 2-arylation of benzoxazoles catalyzed by Pd/NiXantphos at room temperature.¹¹ Recently, Zhang reported that aryl chlorides could be activated by NiXantphos at room temperature.¹² So we surmised that NiXantphos might be active in room-temperature Suzuki cross coupling of aryl chlorides. Here, we

^a Department of Chinese Traditional Herbal, Agronomy College, Sichuan Agricultural University, Chengdu 611130, P. R. China
Email: gaofeng@sicau.edu.cn

^b School of bioengineering, Chengdu University, Chengdu 610106, R. R. China

† Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

Table 1 Optimization of Suzuki coupling of 4-tert-butylphenylboronic acid (**1a**) with 1-chloro-4-methylbenzene (**2a**)^a

Entry	Base	Solvent (ratio)	Time	Yield (%) ^b
1	KF	H ₂ O/THF (1:5)	3h	58
2	CsCO ₃	H ₂ O/THF (1:5)	3h	55
3	K ₂ CO ₃	H ₂ O/THF (1:5)	3h	88
4	KF	H ₂ O/THF (1:5)	6h	60
5	CsCO ₃	H ₂ O/THF (1:5)	6h	80
6	K ₂ HPO ₃	H ₂ O/THF (1:5)	6h	85
7	KHCO ₃	H ₂ O/THF (1:5)	6h	90
8	KH ₂ PO ₃	H ₂ O/THF (1:5)	6h	90
9	K ₂ CO ₃	H ₂ O/THF (1:5)	6h	95
10	K ₂ CO ₃	H ₂ O/PhH (1:5)	6h	75
11	K ₂ CO ₃	H ₂ O/DME (1:5)	6h	92
12	K ₂ CO ₃	H ₂ O/Dioxane (1:5)	6h	90

^a Reactions performed using 1.2 equiv of **1a**, 1.0 equiv of **2a** and 2.0 equiv of base on a 0.2 mmol scale.

^b Isolated yields.

are communicating the efficient Pd(OAc)₂/NiXantphos-catalyzed room-temperature Suzuki-Miyaura coupling of arylboronic acids with aryl chlorides (Scheme 1C).

Our research started from the screening of bases (KF, CsCO₃, K₂HPO₃, K₂CO₃, KHCO₃, KH₂PO₃), adopting Pd(OAc)₂/NiXantphos-based catalytic system with H₂O/THF (1:5) as solvent (Table 1). Results showed that with K₂CO₃ as base, 4-tert-butylphenylboronic acid **1a** (1.2 equiv) and 1-chloro-4-methylbenzene **2a** (1.0 equiv) could access to 4-tert-butyl-4'-methylbiphenyl **3a** in 88% yield after 3 h (Table 1, entry 3). And when the reaction time was prolonged to 6h, the conversion rate was significantly improved. Then the scope of solvents (H₂O/THF, H₂O/PhH, H₂O/DME, H₂O/1,4-Dioxane) were tested adopting K₂CO₃ as base at room temperature for 6h. With H₂O/THF (1:5) as solvent, **3a** was afforded in near quantitative yield after 6 h (Table 1, entry 9).

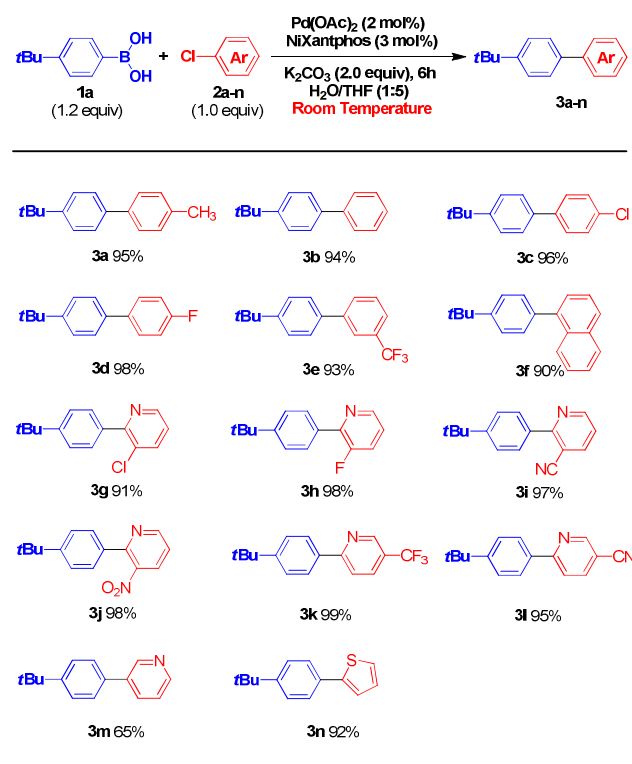
The substrate scope of aryl chlorides **2a-n** (1.0 equiv) with 4-tert-butylphenylboronic acid **1a** (1.2 equiv) was investigated (Scheme 2) under the optimized reaction conditions as Pd(OAc)₂ (2 mol%), NiXantphos (3 mol%), K₂CO₃ (2.0 equiv) in H₂O/THF (1:5) at room temperature for 6 h, and gave corresponding products **3a-n** in yields from 65% to 99%. In general, the reaction could be facilitated by strongly electron-withdrawing groups like trifluoromethyl, cyano group, nitro group, etc. (Scheme 2, **3e**, **3i**, **3j**, **3k**). As is shown in Scheme 2, heteroaryl compounds, mainly 2-phenylpyridines, account for half of the products. It is worth to mention that 2-phenylpyridine derivatives were widely reported to show antitumor and anticancer activity¹³. Since our group has been conducting research on the design and synthesis of anti-cancer molecules with simple structures¹⁴, 2-chloro pyridines were preferred substrates in

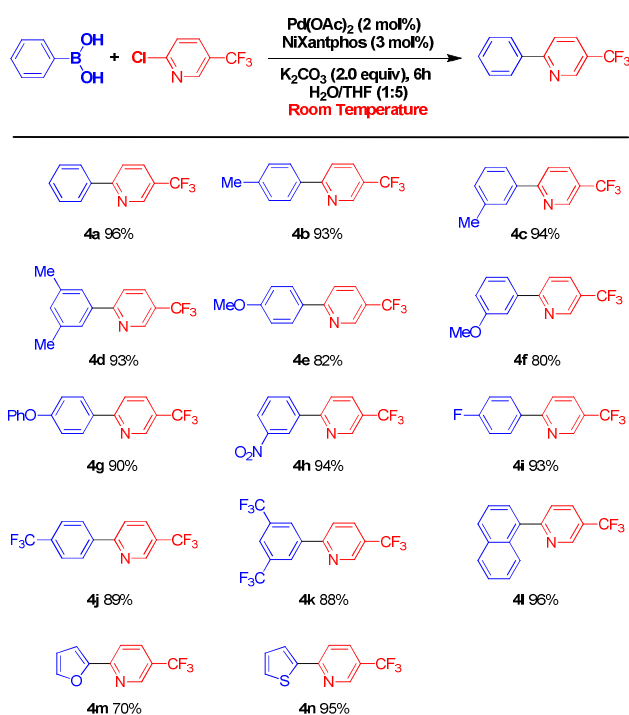
this work.

2-phenyl-5-(trifluoromethyl) pyridine derivatives were reported to have various bioactivities¹⁵, like g-secretase modulation, antimalarial activity, P2X3 receptor antagonism, Glucagon receptor antagonism, etc. Therefore, it was chosen in the scope investigations of arylboronic acids **1b-o** (1.2 equiv) (Scheme 3) under the optimized reaction conditions as Pd(OAc)₂ (2 mol%), NiXantphos (3 mol%), K₂CO₃ (2 equiv) in H₂O/THF (1:5) at room temperature for 6 h, and gave corresponding products **4a-n** in yields from 70% to 96%. It could be seen that the reaction could be suppressed when possessing substrates with strongly electron-donating groups like ether group (Scheme 3, **4e**). Among the coupling products, **4m** and **4n** were afforded from 2-furanboronic acid **1n** and 2-thiophenylboronic acid **1o** with **2k**, respectively, proving the successful application of heteroarylboronic acids.

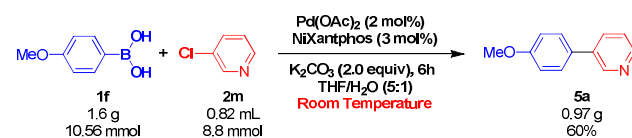
Para, *meta* and *ortho* substituents in aryl chlorides were all well tolerated in the room-temperature Suzuki coupling with 4-tert-butylphenylboronic acid, providing products in yields from 65% to 99%. Similarly, *para* and *meta* substituents in arylboronic acids were well tolerable, and afforded products in yields from 70% to 96%.

We also evaluated the scalability of this method by performing the coupling of 3-chloropyridine **2m** with 4-methoxyphenylboronic acid **1f** on gram scale. The coupling product, 3-(4-methoxyphenyl)pyridine **5a** was isolated in 60% yield. This yield is

Scheme 2 Scope of aryl chloride in the room-temperature Suzuki coupling with 4-tert-Butylphenylboronic acid (**1a**)

Scheme 3 Scope of aryl boronic acid in the room-temperature Suzuki coupling with 2-chloro-5-(trifluoromethyl) pyridine (**2k**)

relatively high, proving that the Pd(OAc)₂/NiXantphos based catalyst could stay active in room-temperature Suzuki cross coupling of aryl chlorides without chelating with Nitrogen atom in pyridine.

Scheme 4 Room-temperature Suzuki coupling of 3-chloropyridine with 4-Methoxyphenylboronic acid on gram scale

In summary, Pd(OAc)₂/NiXantphos-catalyzed Suzuki-Miyaura cross-coupling reaction of arylboronic acids with aryl chlorides were realized at room temperature in high yields (average yield >90%). In our method, a broad range of substrates could be adopted to afford diverse products with minor limitations. The success of this method attributes to the application of commercially available ligand NiXantphos. Next, we will continue our exploration for cross coupling reaction of aryl chlorides using NiXantphos as ligand.

We thank the NSFC (No. 31570341) and Key Technology Support Program of Sichuan Province, China (No. 2015SZ0105) for financial support.

Notes and references

‡ D.W. and H.-G.C. contributed equally.

- 1 K. Sambasivarao, L. Kakali and K. Dhurke, *Tetrahedron*, 2002, **58**, 9633.
- 2 N. Miyaura, T. Yanagi and A. Suzuki, *Synth. Commun.*, 1981, **11**, 513.
- 3 (a) K. C. Nicolaou, P. G. Bulger and D. Sarlah, *Angew. Chem. Int. Edit.*, 2005, **44**, 4442; (b) C. You, P. Hui, Y. X. Pi, T. Meng, Z. Y. Lian, M. Q. Yan, Y. Liu, S. H. Liu and G. A. Yu, *Org. Biomol. Chem.*, 2015, **13**, 3236.
- 4 (a) A. Suzuki, *J. Org. Chem.*, 1999, **576**, 147; (b) L. A. Thompson and J. A. Ellman, *Chem. Rev.*, 1996, **96**, 555; (c) S. P. Stanforth, *Tetrahedron*, 1998, **54**, 263; (d) M. C. Kozlowski, B. J. Morgan and E. C. Linton, *Chem. Soc. Rev.*, 2009, **38**, 3193.
- 5 (a) D. A. Horton, G. T. Bourne and M. L. Smythe, *Chem. Rev.*, 2003, **103**, 893; (b) T. Guo, A. E. P. Adang, R. E. Dolle, G. Dong, D. Fitzpatrick, P. Geng, K. K. Ho, S. G. Kultgen, R. Liu, E. McDonald, B. F. McGuinness, K. W. Saionz, K. J. Valenzano, N. C. R. van Straten, D. Xie and M. L. Webb, *Bioorg. Med. Chem. Lett.*, 2004, **14**, 1713; (c) R. Faghii, W. Dwight, J. B. Pan, G. B. Fox, K. M. Krueger, T. A. Esbenschade, J. M. McVey, K. Marsh, Y. L. Bennani and A. A. Hancock, *Bioorg. Med. Chem. Lett.*, 2003, **13**, 1325; (d) E. A. Jefferson, P. P. Seth, D. E. Robinson, D. K. Winter, A. Miyaji, L. M. Risen, S. A. Osgood, M. Bertrand and E. E. Swayze, *Bioorg. Med. Chem. Lett.*, 2004, **14**, 5257; (e) G. T. Wang, S. Wang, R. Gentles, T. Sowin, S. Leitza, E. B. Reilly and T. W. von Geldern, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 195.
- 6 L. Yin, Z. Zhang and Y. Wang, *Tetrahedron*, 2006, **62**, 9359.
- 7 (a) M. B. Mitchell and P. J. Wallbank, *Tetrahedron Lett.*, 1991, **32**, 2273; (b) W. Shen, *Tetrahedron Lett.*, 1997, **38**, 5575; (c) A. F. Littke and G. C. Fu, *Angew. Chem. Int. Edit.*, 1998, **37**, 3387; (d) A. Zapf and M. Beller, *Chem. Eur. J.*, 2000, **6**, 1830; (e) A. Zapf, A. Ehrentraut and M. Beller, *Angew. Chem. Int. Edit.*, 2000, **39**, 4153; (f) J. H. Li and W. J. Liu, *Org. Lett.*, 2004, **6**, 2809; (g) R. K. Arvela and N. E. Leadbeater, *Org. Lett.*, 2005, **7**, 2101.
- 8 (a) L. Botella and C. Nájera, *Angew. Chem. Int. Edit.*, 2002, **41**, 179; (b) C. W. K. Gstöttmayr, V. P. W. Böhm, E. Herdtweck, M. Grosche and W. A. Herrmann, *Angew. Chem. Int. Edit.*, 2002, **41**, 1363; (c) O. Navarro, R. A. Kelly and S. P. Nolan, *J. Am. Chem. Soc.*, 2003, **125**, 16194; (d) N. Marion, O. Navarro, J. Mei, E. D. Stevens, N. M. Scott and S. P. Nolan, *J. Am. Chem. Soc.*, 2006, **128**, 4101; (e) T. Kinzel, Y. Zhang and S. L. Buchwald, *J. Am. Chem. Soc.*, 2010, **132**, 14073; (f) M. T. Chen, D. A. Vicic, M. L. Turner and O. Navarro, *Organometallics*, 2011, **30**, 5052; (g) F. Rajabi and W. R. Thiel, *Adv. Synth. Catal.*, 2014, **356**, 1873; (h) J.-Y. Lee, D. Ghosh, J.-Y. Lee, S.-S. Wu, C.-H. Hu, S.-D. Liu and H. M. Lee, *Organometallics*, 2014, **33**, 6481; (i) R. Ghosh, N. N. Adarsh and A. Sarkar, *J. Org. Chem.*, 2010, **75**, 5320; (j) T. E. Barder, S. D. Walker, J. R. Martinelli and S. L. Buchwald, *J. Am. Chem. Soc.*, 2005, **127**, 4685; (k) D. Liu, W. Gao, Q. Dai and X. Zhang, *Org. Lett.*, 2005, **7**, 4907; (l) C. M. So, C. C. Yeung, C. P. Lau and F. Y. Kwong, *J. Org. Chem.*, 2008, **73**, 7803; (m) S. R. Dubbaka and P. Vogel, *Org. Lett.*, 2004, **6**, 95; (n) D. W. Old, J. P. Wolfe, S. L. Buchwald, *J. Am. Chem. Soc.*, 1998, **120**, 9722; (o) A. Konovets, A. Penciu, E. Framery, N. Percina, C. G. Henry and D. Sinou, *Tetrahedron Lett.*, 2005, **46**, 3205; (p) M. Ueda, M. Nishimura and N. Miyaura, *Synlett*, 2000, **2000**, 0856; (q) C. Zhang, J. Huang, M. L. Trudell and S. P. Nolan, *J. Org. Chem.*, 1999, **64**, 3804-3805; (r) A. Kumar, G. K. Rao, F. Saleem, R. Kumar and A. K. Singh, *J. Hazard. Mater.*, 2014, **269**, 9; (s) S. Li, Y. Lin, J. Cao and S. Zhang, *J. Org. Chem.*, 2007, **72**, 4067; (t) G. A. Molander and B. Biolatto, *J. Org. Chem.*, 2003, **68**, 4302; (u) X. Bei, H. W. Turner, W. H.

- Weinberg and A. S. Guram, *J. Org. Chem.*, 1999, **64**, 6797; (v) K. W. Anderson and S. L. Buchwald, *Angew. Chem. Int. Ed.*, 2005, **44**, 6173; (w) K. L. Billingsley, K. W. Anderson and S. L. Buchwald, *Angew. Chem. Int. Ed.*, 2006, **45**, 3484.
- 9 (a) W. H, C. Liu, and Z. L. Jin, *Org. Lett.*, 2007, **9**, 4005; (b) W. H, C. Liu, and Z. L. Jin, *Adv. Synth. Catal.*, 2008, **350**, 501.
- 10 (a) T. Jia, A. Bellomo, K. E. L. Baina, S. D. Dreher and P. J. Walsh, *J. Am. Chem. Soc.*, 2013, **135**, 3740; (b) J. Zhang, A. Bellomo, A. D. Creamer, S. D. Dreher and P. J. Walsh, *J. Am. Chem. Soc.*, 2012, **134**, 13765; (c) S.-C. Sha, J. Zhang, P. J. Carroll and P. J. Walsh, *J. Am. Chem. Soc.*, 2013, **135**, 17602; (d) A. Bellomo, J. Zhang, N. Trongsiriwata and P. J. Walsh, *Chem. Sci.*, 2013, **4**, 849; (e) M. Li, B. Yücel, J. Adrio, A. Bellomo and P. J. Walsh, *Chem. Sci.*, 2014, **5**, 2383.
- 11 F. Gao, B. S. Kim and P. J. Walsh, *Chem. Commun.*, 2014, **50**, 10661.
- 12 J. Zhang, A. Bellomo, N. Trongsiriwat, T. Jia, P. J. Carroll, S. D. Dreher, M. T. Tudge, H. Yin, J. R. Robinson, E. J. Schelter and P. J. Walsh, *J. Am. Chem. Soc.*, 2014, **136**, 6276.
- 13 (a) R. Cao, J. Jia, X. Ma, M. Zhou and H. Fei, *J. Med. Chem.*, 2013, **56**, 3636; (b) C. Gabbiani and L. Messori, *Anticancer Agents Med. Chem.*, 2011, **11**, 929; (c) S. F. Ralph, *Curr. Top. Med. Chem.*, 2011, **11**, 572; (d) D. Fan, C. T. Yang, J. D. Ranford, J. J. Vittal and P. F. Lee, *Dalton Trans.*, 2003, **17**, 3376.
- 14 X.-X. Chen, F. Gao, Q. Wang, X. Huang and D. Wang, *Fitoterapia*, 2014, **92**, 111.
- 15 (a) S. Sato, S. Matsuda, C. Matsumura, M. Itotani, T. Shinohara, S. Fujita, Y. Sakurai, K. Tai, T. Fukushima, N. Kanemoto and T. Okamoto, WO2015/8872 A1, 2015; (b) G. D. Heffernan, D. P. Jacobus, K. W. Saionz, G. A. Schiehsler, H.-M. Shieh and W. Zhao, US2014/135320 A1, 2014; (c) M. Mori, K. Fujii, M. Inui, T. Baba, Y. Onishi and A. Aoyagi, EP2700643 A1, 2014; (d) M. Mori, K. Fujii, M. Inui, T. Baba, Y. Onishi and A. Aoyagi, WO2012/144478 A1, 2012; (e) C. S. Burgey, Z. J. Deng, D. N. Nguyen, D. V. Paone, C. M. Potteiger and J. P. Vacca, WO2009/058298 A1, 2009; (f) C. Y. Ho, WO2009/052341 A1, 2009; (g) R. M. Kim, E. R. Parmee, Q. Tan, A. R. Lins, J. Chang and C.-M. Yang, WO2007/111864 A2, 2007.

The first room-temperature Suzuki-Miyaura coupling of arylboric acid with aryl chlorides catalyzed by Pd(OAc)₂/NiXantphos-based system is communicated.

