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Hydrocarbons degradation by a newly isolated thermophilic

Anoxybacillus sp. with bioemulsifier production and new alkB genes

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38 **Abstract:** Interests in biodegradation of petroleum derived pollutants by thermophilic bacteria have
39 been steadily increasing in recent decade. In this work, a newly isolated thermophilic bacterial strain
40 was isolated from the deep petroleum reservoir and identified as *Anoxybacillus. sp* WJ-4 based on the
41 analysis of the physiological characteristics, 16S rRNA sequencing, GC content and cellular fatty acids.
42 It is the first report that strain WJ-4 can degrade a wide range of hydrocarbons (C8-C22) at 67°C. The
43 production of oligosaccharide-lipid-peptide bioemulsifier was detected. It exhibited the excellent
44 emulsification activity with various oil phases ($EI_{24}>60\%$), and the ability of increasing the cell surface
45 lipophilicity during the degradation, but has no significant impact on the surface tension with reducing
46 from 72.22 mN/m to 52.45 mN/m. Four alkane monooxygenase genes showed high phylogenetic
47 relationship (>95%) with alkB genes from *Geobacillus*. These results indicated that this newly isolated
48 bacterial strain and its bioemulsifier have great potentials in the environmental remediation and
49 petroleum recovery under thermophilic condition.

50 **Key words:** Biodegradation; Bioemulsifier; *Anoxybacillus*; Thermophilic; Cell surface Lipophilicity.

51 **1. Introduction**

52 Petroleum is the predominant energy resource and chemically industrial raw materials in modern
53 society. Spills, leaks, and other releases of petroleum, however, often result in soil and groundwater
54 contamination.¹ Main sources of petroleum contamination include petrochemical industry processes,
55 oil field installations, petroleum plants, liquid fuel distribution and storage devices, transportation
56 equipment for petroleum products.² Pollution from such sources has significant environmental impact
57 and presents human health hazardous. Petroleum is a complex mixture of hydrocarbons and related
58 compounds. It is generally classified into four fractions: saturated alkanes, aromatics, resins, and
59 asphaltenes, the latter two of which consist of polar molecules containing N, S, and O₂. Alkanes can

60 constitute 50% to 95% of crude oil, depending on source. As the main component of fuels and oils, its
61 relative inertness poses ecological problems upon release to the environment. Due to their lack of
62 functional groups as well as poor water solubility, alkanes exhibit both low chemical reactivity and low
63 bioavailability for microorganisms.^{3,4} Various linear, branched and cyclo-alkanes are known to cause
64 respiratory, renal, or central nervous system disorders. As a result, considerable attention has been paid
65 to the treatment of these pollutants.

66 Microbial bioremediation has been evaluated as one option in various polluted environments and
67 is claimed to represent an efficient, economical, and versatile alternative to physicochemical
68 treatments.^{5,6} Hydrocarbon-degrading bacteria are widely distributed in nature. They include members
69 of at least 60 genera of aerobic and five genera of anaerobic bacteria,^{3,7-9} particularly *Acinetobacter*,
70 *Rhodococcus*, *Alcanivorax*, *Bacillus*, *Mycobacterium*, *Pseudomonas* and *Dietzia*, which are among the
71 most well known and studied.¹⁰⁻¹³ However these bacteria generally exhibit good performance only
72 under mesophilic conditions. Few thermophilic hydrocarbon-degrading species have been reported,
73 although many thermophiles have been described.

74 Degradation of petroleum hydrocarbons by thermophilic bacteria has advantages over that by
75 mesophilic or psychrophilic organisms, especially when they are incorporated into biotechnological
76 applications.^{14,15} An increasing attention has been paid directly to this field in recent decade.^{16,17} Thus it
77 is necessary to isolate more thermophilic strains with the capability to degrade these water-insoluble
78 molecules. The majority of thermophilic strains are obtained from hot springs and oil reservoirs. High
79 temperature petroleum reservoirs with temperatures exceeding 50 °C are one type of biotope attracting
80 great interest as sites for the collection and screening of new thermophilic hydrocarbon-degrading
81 bacteria.^{4,14,17,18} Genera collected from these sites have been limited to *Bacillus*, *Geobacillus*,

82 *Thermoactinomyces*, and *Brevibacillus*.^{19,20} Only four studies on hydrocarbon degradation by
83 *Anoxybacillus* have also been reported,²¹⁻²⁴ where the degraded compounds were aromatic. As new
84 thermophilic genus, *Anoxybacillus* could be utilized in a large number of applications as previous
85 description,²⁵ however these applications are currently limited to the hydrolysis of the starch and
86 lignocellulosic biomasses, not the degradation of alkanes.

87 Recognized bacterial mechanisms for enhancing hydrocarbon substrate availability and utilization
88 for simplicity generally divide into (a) those related to the uptake of soluble fractions and (b) those
89 related to the production of surfactants for physical modification of substrates and cell surface
90 lipophilicity;²⁶⁻²⁸ the mechanisms however overlap. Extremely limited water solubility is the most
91 serious hurdle for the uptake alkane as a metabolic substrate, although higher temperatures can
92 contribute to improved mass transfer rates. The uptake of hydrophobic components therefore
93 commonly and frequently involves the production of microbial surfactant molecules as emulsifying
94 agents or cell surface lipophilicity (CSL) altering agents to facilitate ultimate biodegradation.

95 Produced by a wide range of hydrocarbon-degrading microorganisms, microbial surfactants can
96 be grossly classified into two major categories based on molecular weight (MW).^{29,30} Examples of
97 low-MW biosurfactants include glycolipids and lipopeptides, which dominant function is the lowering
98 of the surface/interfacial tension. While examples of high-MW biosurfactants include emulsan, alasan,
99 biodispersan, and extracellular or cell membrane-bound bioemulsifiers (such as exopolysaccharide and
100 lipopolysaccharides), which main function is emulsion stabilization.^{30,31} For many
101 hydrocarbon-degrading microorganisms, extensive changes of the cell surface lipophilicity were
102 detected during growth on hydrocarbons.³² Several reports in the literature find correlation among CSL,
103 microbial surfactant production, and hydrocarbon degrading capability.^{26,30-32}

104 In the present report, a novel thermophilic hydrocarbon-degrading strain WJ-4 was isolated and
105 investigated. This is the first report that this *Anoxybacillus* strain can utilize alkanes (C8 - C22) as the
106 sole source of carbon for growth. With a goal of designing future applications in remediation of crude
107 oil contaminated environments, the alkane-degrading kinetics of strain WJ-4 was characterized. A
108 bioemulsifiers was detected during biodegradation and its structure was analyzed. Emulsification
109 stability was evaluated as well. A tentative comparison of the *alkB* gene from this novel thermophilic
110 strain was also carried out. This paper highlights an important potential use of a novel thermophilic
111 strain for the cleanup of alkane or petroleum polluted environments.

112 2. Experimental

113 2.1. Isolation of microorganisms

114 The brine and oil samples were collected from the reservoir formation located at depths of 1300 to
115 1600 meters subterranean with the temperature of 58-68°C in Longhupao block (latitude 46.798383,
116 longitude 124.350554) of Daqing oilfield. Fifty milliliter of the brines and five grams of crude oil were
117 transferred to the 250 ml flask filled with 50.0 ml of the autoclaved minimal medium supplemented
118 with trace-element solution (0.1%, v/v), and then incubated at 65°C with 180 rpm for 20 days. The
119 minimal medium contains (g/L): NH_4NO_3 3.4, K_2HPO_4 1.5, NaH_2PO_4 , 1.5, MgSO_4 0.3, yeast extract
120 0.3, Glucose 0.3, pH 6.8-7.2; and the composition of the trace-element solution contains (g/L):
121 $\text{MnCl}_2 \cdot 4 \text{H}_2\text{O}$ 0.1, $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ 0.17, $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ 0.02, FeCl_2 0.4, H_2BO_3 0.019, ZnCl_2 0.1,
122 $\text{NaMoO}_4 \cdot 2\text{H}_2\text{O}$ 0.1, vitamin B₁₂ 0.01, Vitamin C 0.01, $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ 0.05. After 20-days incubation,
123 fifty milliliter of the cultured sample was transferred to 50 ml of the above fresh sterilized medium
124 again for the second 20-days incubation at the same condition. Then 100 ul of the final cultured sample
125 was spread onto LB agar plates and incubated at 65°C for 48 h. Pure cultures of each morphologically

126 distinct colony were selected by repetitive streaking onto solid LB agar medium. In order to isolate the
127 strains with ability of degrading hydrocarbon and biosurfactant production, the selected pure strains
128 were cultured for 10 days in the above sterilized minimal medium with crude oil as carbon source, then
129 the cell-oil-free supernatant of each cultured sample was conducted the oil spreading agar plate and
130 crude oil emulsification test following the described methods.³³

131 *2.2. Characterization of strain*

132 The final selected bacterial strain was characterized according to the standard biochemical tests
133 (morphology and biochemistry) following the Bergey's manual of systematic bacteriology. Growth was
134 determined by measuring the OD₆₀₀ value at different temperature (20-80°C), pH (4-11) and NaCl
135 concentrations (0-25%, g/L). The ability to utilize various carbon sources was examined in minimal
136 medium. Carbon sources were added at final concentration of 0.05% (w/v) with: fructose, glucose,
137 L-rhamnose, sucrose, lactose, molessess, starch, xanthan, n-dodecane and n-hexadecane.

138 G+C content were determined according to Mesbah³⁴ by using HPLC. Nonmethylated
139 Lambda-DNA (Sigma) with GC-content 49.858 mol% was served as an external standard.

140 DNA was extracted from the isolated strain following the instruction of the extraction Kit (MoBio,
141 USA) for phylogenetic analysis. The methods for 16S rRNA gene amplification and sequencing have
142 been reported previously.²³ The phylogenetic analysis was examined by BLAST of the National Center
143 for Biotechnology Information.³⁵ Multiple sequence alignments were carried out using ClustalX 1.8
144 and the neighbor-joining tree was constructed using MEGA Software Version 6.0.

145 Late exponential phase cells were harvested by centrifugation (10000 rpm for 10 min) and washed
146 triple with distilled water for the analysis of the cellular fatty acids. Lipid extraction and cellular fatty
147 acid analyses were performed according to the method of Siristova.³⁶ The relative percentages of the

148 fatty acids were determined from the peak areas of the methyl esters by gas chromatograph (Agilent
149 5890) equipped with a flame-ionization detector. The running method was as follows: injector
150 temperature, 240°C; detector temperature, 240°C; carrier gas (He) flow rate, 10 ml/min. The oven
151 temperature was programmed from 50 to 300 °C at 8°C/min. The results are means of two independent
152 experiments.

153 2.3. Kinetics of Alkanes Degradation

154 Cell harvested from 50 ml LB medium of the selected strain was washed with distilled water triply
155 and transferred to a 250 ml flask with 2 g of ultrafilter-sterile alkane mixture (n-alkane, C8-C22,
156 purity>99%, Sigma, USA) and 90 ml of minimal medium, then incubated at 70°C with 180 rpm for 40
157 days with using the cell-free sterile medium as control. 4 parallel samples were prepared as interval
158 samples for analyzing the biodegradation kinetics, including cell growth, residual alkane, cell surface
159 lipophilicity, surface tension and emulsification.

160 Cell growth was monitored by measuring the dry cell weight. Cell was collected by centrifugation
161 (10000 × g, 10 min) of a 100 ml culture broth and washed with distilled water twice, and then dried by
162 heating at 40°C until constant weight was attained. Surface tension of cell-free samples was measured
163 by the digital tensiometer (KRUESS klot, Germany) using the ring detachment method. Emulsification
164 activity was determined by addition of 5 ml alkanes mixture to 5 ml of the cell-alkanes-free supernatant
165 in a 15 ml graduated tube according to the previous description.³⁷ Adherence of bacteria to
166 hydrocarbons was used as a measurement of cell-surface lipophilicity according to the reported
167 method.³²

168 Alkanes were extracted with dichloromethane in triplicates (>99%, Sigma, USA), collected the
169 organic phase and removed the solvent in rotary evaporator at 30°C, and then weight the residual

170 alkanes. In order to profile the degradation characterization, gas chromatograph equipped with FID
171 detector (Agilent 5890, USA) was applied to detect the changing of each fraction in the alkanes
172 mixture. The GC running program was as follow: injector temperature, 240°C; detector temperature,
173 240°C; carrier gas (He) flow rate, 10 ml/min. The oven temperature was programmed from 50 to
174 300°C at 8°C/min. The relative percentages of the n-alkanes were determined from the peak areas; and
175 the weight of each alkane was calculated by the relative percentage and the total weight of residual
176 alkanes mixture. The results are means of two independent experiments.

177 2.4. Analysis of bioemulsifier

178 Purification of the bioemulsifier was performed by solvent extraction and alcohol precipitation.³⁷

179 ³⁸ The measurement of surface tension and emulsification were applied to testify the surface/interface
180 activities of the obtained materials, and then the chemical characteristics of these materials were
181 further analyzed with or without the hydrolysis. The carbohydrate moiety was detected by phenol
182 sulfuric acid method using glucose as standard.³⁹ Protein moiety was detected by Lowry method using
183 bovine serum albumin (BSA) as calibration standard.⁴⁰ The lipid moiety was detected by
184 dichloromethane-methanol method.⁴¹ The number-average molecular weight (M_n), weight-average
185 molecular weight (M_w) and polydispersity index (PDI) of the biosurfactant were measured by gel
186 permeation chromatography (GPC) using pullulan standards as described previously.^{42,43}

187 Monosaccharide composition was determined according to the method reported by Carrion.³⁸ The
188 hydrolysates were used to identify and quantify the constituent monosaccharides by high-performance
189 liquid chromatography (HPLC) using Aminex HPLC Carbohydrate Analysis Columns HPX-87P
190 (300 × 7.8 mm) and HPX-87C (300 × 7.8 mm) (BioRad) with the commercial sugars as standard for
191 monosaccharide identification. Amino acids were analyzed following the methods from Xia³⁷ with an

192 automatic amino acid analyzer (1100 series, Hewlett Packard, USA). Fatty acid compositions were
193 extracted from the hydrolyzates with ether in triplicates and esterified with methanol at 100°C for 1 h,
194 then subjected to gas chromatograph mass spectrometer analysis.⁴⁴

195 Effect of the environmental factors (salinity, temperature, pH, and metallic ions) on the
196 performance of the emulsification against the crude oil was determined. The concentration of the
197 purified active bioemulsifier tested in emulsification tests was 2000 mg/L.

198 *2.5. Analysis of alkane hydroxylase genes*

199 DNA isolated from the *Anoxybacillus* strain was used as a template for PCR. Amplification of
200 *alkB* fragments of the thermophilic bacteria was carried out with various degenerate oligonucleotide
201 primers as designed in literatures.^{45,46} The purified PCR products of *alkB* genes were cloned using a
202 pGEM-T vector system (Promega). Clones were sequenced with the universal M13 primers in an ABI
203 3700XL genetic analyzer (Applied Biosystems). The preliminary analysis of the new sequences was
204 done with the BLAST program of NCBI. The nucleotide sequences were aligned with homologous
205 sequences retrieved from GenBank with ClustalX 1.8, and a neighbor-joining tree was constructed
206 using MEGA Software Version 6.0.

207 **3. Results and discussion**

208 *3.1. Characterization of bacteria*

209 The well-performed strain WJ-4 is a facultative aerobic, gram-positive, motile, spore-forming,
210 rod-shaped bacterium with a length of 3.9 to 5.6 µm and a width of 0.6 to 0.9 µm (Table 1). Colonies
211 were gray white, umbonate surface and undulate edge. The sporangium was not swollen, whereas the
212 spores were oval and terminal positioned. Growth was observed both on carbohydrates (arabinose,
213 fructose, glucose, L-rhamnose, mannose, sucrose, lactose, molessess, starch, xanthan) and

214 hydrocarbons (n-dodecan, n-hexadecane and xylene). Indol was not produced; nitrate was not reduced;
215 Voges-Proskauer reaction and Methyl-red test were negative. Like the most thermophilic bacilli,
216 catalase reaction was positive for this tested strain. It can survive at wide range temperature from 45°C
217 to 80°C, and the maximum optical density was achieved at 68-72°C. The tolerance to NaCl levels was
218 weakened along with the increasing of the salt concentration, the growth was ceased almost when
219 salinity reaching at 12%. The optimum pH for the growth was 6.0-8.0.

220 The phylogenetic tree (Fig. 1) showed that strain WJ-4 was almost positioned between
221 *Anoxybacillus* genus from one side and *Geobacillus* genus from the other side. The closest (more than
222 98%) sequence relatives found by BLAST search was *Anoxybacillus* species. (KJ842629.1,
223 EU710556.1, KF266689.1), *Geobacillus* species (FJ823100.2, EU087702.1). Although *Anoxybacillus*
224 is phylogenetically close to *Geobacillus* species,^{49,51} strain WJ-4 was related to genus *Anoxybacillus* on
225 the basis of phylogenetic similarity with *Anoxybacillus* species combined with the morphological,
226 physiological, and biochemical properties. The sequence of the 16S rRNA gene of the strain WJ-4 is
227 deposited and available under the GenBank accession number JX673944.1.

228 The G+C content of the genomic DNA for the strain WJ-4 was 44.3 mol%, which was
229 significantly lower than those for the genus *Geobacillus* (48.2-58 mol%),⁵¹ but similar to the closest
230 *Anoxybacillus* relatives (42-57 mol%, *Anoxybacillus amylolyticus* 43.5 mol%; *Anoxybacillus*
231 *voinovskiensis* 43.9 mol%; *Anoxybacillus contaminans* 44.3 mol%). Although the G+C content of
232 DNA was 43.2 mol% for *Geobacillus tepidamans*,⁴⁹ it currently has been reclassified as *Anoxybacillus*
233 *tepidamans*. Therefore, based on the similar percentage range of G+C content between *Geobacillus* and
234 *Anoxybacillus* genus, it is possible that the intimate phylogenetic relationship existed between them.

235 The cellular fatty acids (FAs) of the strain WJ-4 was largely composed of the branched saturated

236 aliphatic acid, and has similar percentage with other thermophilic bacilli containing anteiso-fatty acids
237 as minor components (Table 2). Iso-branched Fatty acids constituted 83.22% of total WJ-4 fatty acids
238 and greatly predominated over anteiso-branched members, and contains iso-branched saturated fatty
239 acids (iso-C15:0 and iso-C17:0) as major fatty acids, especially iso-C15:0 account for 48.79%. The
240 presence of branched FAs is considered to be a means of maintaining membrane fluidity; iso-branched
241 FAs generally have higher melting points, while anteiso-branched FAs typically have lower melting
242 points. This is possible explanation for the thermophilic property of strain WJ-4. The percentage of
243 the iso-C15:0 and iso-C17:0 for other genus representatives were: 48.30% for *Geobacillus*
244 *Stearothermophilus* 5965, 68.59% for *Anoxybacillus contaminans* and 72.8% for *Anoxybacillus*
245 *amylolyticus*, 86.36% for *Anoxybacillus rupiensis* R270.⁵⁰

246 The similar physiological characteristics, G+C content of DNA, fatty acid profile and
247 phylogenetic similarity (98.0-99.0% to the closest relatives) with representatives of the genus
248 *Anoxybacillus* allow us to place the strain WJ-4 in the genus *Anoxybacillus* as the type strain for the
249 novel species. Particularly, it was possible that *Anoxybacillus* sp. WJ-4 has some similar characteristics
250 with *Geobacillus* species.

251 3.2. Degradation kinetics

252 Due to lack of functional groups as well as low water solubility, alkanes exhibit low chemical
253 reactivity and bioavailability for microorganisms. However, some microorganisms possess the
254 metabolic capacity to use these compounds as carbon and energy sources for their growth. In this study,
255 the isolated thermophilic and halotolerant strain WJ-4 was evaluated to degrade the alkanes from C8 to
256 C22. The results of the hydrocarbon degradation kinetic shown in Fig. 2A demonstrated the excellent
257 ability of alkane degradation in this novel strain with maximum growth reaching at 20th day. The

258 degradation rate became flatten after 20 days incubation but totally 58.75% of alkanes were
259 decomposed after 40 days at 70°C under aerobic condition.

260 The thermophilic bacterial genera *Anoxybacillus* and *Geobacillus* were described nearly in the
261 radiation from the Gram-positive genus *Bacillus*,^{51,52} and *Bacillus* and *Geobacillus* has presented the
262 ability of alkane degradation.⁵³ Compared with *Bacillus* and *Geobacillus*, *Anoxybacillus* is a relatively
263 new genus that was proposed in the year 2000.⁵² Few researches on the hydrocarbon degradation by
264 *Anoxybacillus* genus have been reported so that the application of this new genus on the bioremediation
265 and enhanced oil recovery was rare. *Anoxybacillus* strain was able to efficiently degrade the synthetic
266 aromatic hydrocarbons.²¹⁻²⁴ Al-Jailawi et al. found that *Anoxybacillus* was the most predominant genus
267 thermophilic bacteria with aromatic hydrocarbons degradation activities from hydrocarbon
268 contaminated soil in Iraq.²² However, none of literatures has been investigated to explore the possibility
269 and characteristics of alkanes degradation by this relatively new genus, this study therefore was the
270 first report about the alkane biodegradation by *Anoxybacillus* genus.

271 Gas chromatography was applied to present the changing of each component in alkane mixture
272 during degradation, and the residual weight of each alkane was calculated as well. The results in Fig.3
273 showed that this strain can utilize the different chain alkane (C8-C22) with various efficiency,
274 especially that C8 and C9 were almost depleted after the first 10 days degradation. The degradation
275 kinetic of each alkane (in Fig.2B) demonstrated that the degradation rate of the alkanes can be
276 classified into three types based on the rate calculation of the first 10-days degradation. The first type
277 with the degradation rate (higher than 10 mg/d) is limited to C8-C10 with 14.49 mg/d, 14.33 mg/d and
278 11.92 mg/d respectively; for the second type, the degradation rate is between 5 mg/L to 9 mg/L which
279 includes C11-C12 with 8.13 mg/d and 6.47 mg/d; for the third type, the degradation rate is less than 3

280 mg/d for the long chain alkane (C13-C22). Combined the results in Fig. 2A and 2B, it was obvious that
281 the degradation of C13-C22 between 10th and 30th day was facilitated along with the cell surface
282 lipophilicity increasing and reaching 65% of CSL at 30th day, while the cell surface lipophilicity of the
283 strain before 10th day was relatively lower than 20%. Two possible mechanisms of alkane uptaking
284 have been proposed and well proven: direct uptake of the small chain alkane, and the changing of
285 physiochemical properties of cell to alkane.²⁶ In addition, suggested mechanisms for the uptake of
286 hydrophobic contaminants by degrading bacteria include direct contact of substrates with
287 microorganisms having a high CSH and biosurfactant-mediated uptake by microorganisms capable of
288 producing biosurfactant (and bioemulsifier).²⁸ Therefore, the possible mechanism of degradation of
289 *Anoxybacillus* WJ-4 when fed with different molecule alkanes could be explained by the above
290 proposed mechanisms based on the results in Fig. 2 and 3. It hypothesized that this bacterial strain
291 assimilated the small chain alkane (<C10) following the first mechanism (because of the depletion of
292 C8 and C9 in Fig. 2B and 3); and the degradation of the long chain alkane (>C10) conformed to the
293 second mechanism (because of the increasing of cell surface lipophilicity in Fig. 2A).

294 Most of alkane-degrading bacteria secrete diverse surfactants that facilitate emulsification of the
295 hydrocarbon. Zhao et al. elaborated a relationship between the cell surface lipophilicity and the level of
296 biosurfactant production.²⁶ Hassanshahian and Emtiazi reported that a correlation between
297 emulsification activity, cell adherence to hydrocarbon and growth rate of the crude oil degrading
298 bacteria in crude oil media.⁵⁴ Kundu described the effect of biosurfactant production on the cell
299 surface hydrophobicity and the degradation of hetero-aromatic hydrocarbon.²⁸ In this study, cell
300 surface lipophilicity of *Anoxybacillus* cell and the emulsification index was increasing from 5.9% to
301 70.4% and from 0 to 72.2% respectively along with the decrease of the amount of the residual alkanes

302 (Fig. 2A). Therefore, it could be a conclusion that this novel thermophilic and halotolerant
303 *Anoxybacillus* WJ-4 could synthesize the biosurfactant or bioemulsifier to facilitate the degradation of
304 the alkanes.

305 Unlike biosurfactants (that can significantly decrease the surface/interfacial tension),
306 bioemulsifiers are always high molecular biosurfactants that are able to form stable emulsions with
307 hydrophobic materials (usually oil-in-water and less commonly water-in-oil), but hardly reduce the
308 surface or interfacial tension. Fig. 2A presented the surface tension changing of the cell-alkane-free
309 supernatant and indicated that the microbial surfactant produced by this novel strain WJ-4 has no
310 significantly influence on the surface tension with decreasing from 72.22 mN/m to 52.45 mN/m. A 890
311 Da-biosurfactant from *Rhodococcus* could decrease the surface tension from 71 to 29 mN/m.²⁸ Haza
312 found a 1044 Da-surfactin from *Bacillus* could reduce the surface tension from 69.07 mN/m to 30
313 mN/m.⁵⁵ Bao et al.⁵⁶ reported small-molecular lipopeptide biosurfactant has the lowest surface tension
314 of 26.30 mN/m when using crude oil as carbon source. Bioemulsifiers are higher in molecular weight
315 than biosurfactants as they are complex mixtures of heteropolysaccharides, lipopolysaccharides,
316 lipoproteins and proteins.^{30,57-59} They are also known as high molecular weight biopolymers or
317 exopolysaccharides, which possess only emulsifying activity and not surface activity. Similar to
318 biosurfactants, these molecules can efficiently emulsify two immiscible liquids such as hydrocarbons
319 or other hydrophobic substrates even at low concentrations but in contrast are less effective at surface
320 tension reduction.³⁰ In addition, the number-average molecular weight (M_n), weight-average molecular
321 weight (M_w) and polydispersity index (PDI) of the bioemulsifier were measured by gel permeation
322 chromatography (GPC) using pullulan standards. The results showed that the number-average
323 molecular weights (M_n), and weight-average molecular weight (M_w) of the obtained bioemulsifier was

324 between $129,485 \pm 1,827$ Da and $210,000 \pm 1,827$ Da with PDI average values of 1.212, respectively.
325 Therefore, we can infer that the strain WJ-4 produces relatively high molecular bioemulsifier which
326 was not able to reduce the surface tension obviously, but has ability of changing the cell surface
327 lipophilicity and emulsifying the alkanes as evidence showed in Fig. 2A. Large amount of bacterial
328 strains can produce high molecule bioemulsifier which can stabilize the various kinds of emulsion
329 formed with the immiscible phases but cannot decrease surface tension significantly. The bioemulsifier
330 of *Anoxybacillus* strain WJ-4 has similar characteristic with these reported bioemulsifiers.^{31,42} Although
331 numerous literatures on CSL, emulsification and biodegradation were limited to the small molecule
332 biosurfactant (generally like glycolipids and lipopeptides), it was not ignorable and increasingly
333 significant that bioemulsifier (high molecule weight biosurfactant) could be applied efficiently in many
334 fields, especially the hydrocarbons bioremediation.

335 3.3. Analysis of biosurfactant

336 The chromogenic reaction was developed to analyze the composition of bioemulsifier from
337 *Anoxybacillus* sp. WJ-4. Results showed in Fig. 3 that there were the blue-green spot, purplish-red spot
338 and yellow brown spot on silica gel plates when separately using ninhydrin, anthrone and
339 ammonium-perchlorate as color developing reagent, indicating that peptides, glycosides and lipids
340 were constituted with approximate percentage of 6.25%, 53.12% and 40.63% respectively in this
341 bioemulsifier. Thus the bioemulsifier produced by *Anoxybacillus* sp. WJ-4 was a high molecular
342 weight oligosaccharide-lipid-peptide complex in which these moieties could be structurally associated
343 involving either covalent or non-covalent bonds. The oligosaccharide composition of the hydrolysed
344 bioemulsifier consisted of D-glucose, D-galactose, D-mannose and L-rhamnose in approximate molar
345 ratios of 2:1:2:1. GC-MS analysis of lipid fraction showed that hexadecanoic acid and octadecanoic

346 acid were the major fatty acids that account for 89.85% of total fatty acids. Other fatty acids
347 determined at lower extent were decanoic acid (2.38%), dodecanoic (4.25%) acid and tetradecanoic
348 acid (3.52%). The amino acids of the peptides of the bioemulsifier were analyzed and tabulated in
349 Table 3.

350 3.4. Emulsification activity

351 Emulsification property is critical for bioemulsifier to be promising in different environmental and
352 industrial applications. The emulsification activity of *Anoxybacillus* bioemulsifier against various oil
353 phases was investigated at room temperature for 24 hours. Results as shown in Fig.5A that
354 *Anoxybacillus* bioemulsifier can effectively emulsify the different hydrocarbons, and formed stable
355 emulsions with pure alkane (C6-C10, C12 and C16), as well as the mixtures (crude oil, kerosene,
356 paraffin). Furthermore, it had better emulsification activity with the hydrocarbon mixtures than pure
357 alkane with crude oil as the best oil phase, and the weaker emulsion was formed along with the
358 increasing of the alkane chain number.

359 Effectiveness of the bioemulsifier over a wide range of temperature, pH and salinity can endow its
360 broad applications. Exposed to different extreme temperature (<10°C and >80°C) for 2 hours, this
361 bioemulsifier exhibited the high activity (EI>60%) stably over a wide temperature range (Fig. 5B), and
362 a slight decreasing was detected with the treatment higher than 80°C. In pH evaluation, *Anoxybacillus*
363 emulsifier showed the relative stable emulsification performance at pH from 6 to 12 and was
364 significantly inhibited under the extreme acid conditions (Fig. 5C. Previous literatures reported that the
365 emulsifying activity was significantly inhibited at a NaCl concentration greater than 5 %, ⁴² whereas the
366 bioemulsifier from *Anoxybacillus sp* WJ-4 showed a better halotolerance, remaining high
367 emulsification (EI>60%) under the salinity up to 20% (Fig. 5D). In contrast, the commercially

368 chemical surfactants SDS (sodium dodecyl sulfate), Triton X-100 or Tween 80 have no emulsifying
369 activity at NaCl concentrations of 100-120 g/l.⁶⁰ Although the production cost of
370 bioemulsifier/biosurfactant was still high, it is obvious that the trend of the replacement of chemical
371 surfactant by bio-counterpart is being accelerated due to the unparalleled advantages over the
372 commercial petro-derived chemical surfactant and the rapid development of biological methods and
373 purification technology.

374 3.5. Alkane hydroxylase gene analysis

375 Because of the nearest phylogenetic relationship with *Geobacillus* genus,^{25,50} the total DNA of the
376 strain WJ-4 grown on alkanes containing medium was used to amplify *alkB* gene fragments with
377 various degenerate primers targeting the most conserved region of the *Geobacillus alkB* gene.^{45,46} The
378 selected 50 clones were analyzed by DNA sequencing using plasmid primers. The obtained nucleotide
379 sequences were compared to the known sequences of *alkB* geo homologs, as well as with *alkB* genes of
380 various bacteria deposited in GenBank. Four homologs of *alkB* gene were revealed in *Anoxybacillus*
381 strain WJ-4, namely, *alkB-an1* (10 clones), *alkB-an2* (11 clones), *alkB-an3* (8 clones), *alkB-an4* (21
382 clones). The sequenced *Anoxybacillus alkB* fragments were blasted with the homologs available from
383 GenBank. The corresponding phylogenetic trees differ from each other shown in Fig. 6. The genbank
384 access number of four genes is from KR153280 to KR153283.

385 The sequences of *alkB-an1*, *alkB-an2*, *alkB-an3* and *alkB-an4* belong to the phylogenetic clusters
386 were similar with *Geobacillus alkB-geo1*, *Geobacillus alkB-geo2*, *Geobacillus alkB-geo3*, and
387 *Geobacillus alkB-geo4* found in *Geobacillus* genus.^{45,46,53} The nucleotide sequences similarity to the
388 corresponding *Geobacillus alkB* homologs is 97.0, 95.0, 97.0%, and 98.0% for these four *alkB* genes
389 respectively. Although the PCR and sequencing of the alkane gene from *Anoxybacillus* were

390 implemented in this study, it is obviously nonsufficient to explain the alkane degradation mechanism of
391 this novel strain. In order to reveal the metabolic pathway of alkane degradation, the whole genome
392 sequencing, reverse transcription, functional analysis and structure of *alkB*-an series, and the
393 intermediate metabolites will be further investigated.

394 **4. Conclusions**

395 New isolated thermophilic *Anoxybacillus* strain from high temperature petroleum reservoir was
396 evaluated to degrade the hydrocarbon pollutant. The produced oligosaccharide-lipid-peptide
397 bioemulsifier showed the good performance of emulsification ($EI_{24}>60\%$) with various oil phases at
398 different conditions. This bioemulsifier reduced surface tension from 72.22 mN/m to 52.45 mN/m,
399 and increased the cell surface lipophilicity to 65% during hydrocarbon degradation. Four
400 hydrocarbon-hydrolysis genes were detected in this strain, and showed higher sequences similarity
401 ($>95\%$) to the corresponding *Geobacillus* *alkB* homologs. Although the mechanism of degradation and
402 bioemulsifier production need more studies, it has significantly showed the bioemulsifier, cell surface
403 lipophilicity and hydrocarbon monooxygenase has significant effect on the alkane degradation, even
404 under thermophilic condition.

405 **Acknowledgements**

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409 **References**

- 410 [1] B. Gargouri, F. Karray, N. Mhiri, F. Aloui, S. Sayadi. *J. Chem. Technol. Biotechnol.* 2013, **8**, 1-10.
411 [2] J.G. Leahy, R.R. Colwell. *Microbiol. Rev.* 1990, **54**, 305-315.

- 412 [3] F. Rojo. Degradation of alkanes by bacteriaem. *Environ. Microbiol.* 2009, **11**(10), 2477-2490.
- 413 [4] B.M. Dellagnezze, G.V. de Sousa, L.L. Martins, D.F. Domingos, E.E.G. Limache, S.P. de
414 Vasconcellos, G.F. da Cruz, V.M. de Oliveira. *Mar. Pollut. Bull.* 2014, **89**(1), 191-200
- 415 [5] M. Soleimania, M. Farhoudi, J.H. Christensen. *J. Hazard. Mater.* 2013, **254**(15), 372-381.
- 416 [6] S. Fuentes, V. Mendez, P. Aguila, M. Seeger. *Appl Microbiol Biotechnol.* 2014, **98**(11), 4781-4794.
- 417 [7] K. Zengler, H.H. Richnow, R. Rosselló-Mora, W. Michaelis, F. Widdel. *Nature*, 1999, **401**, 266-269.
- 418 [8] R.C. Prince. The microbiology of marine oil spill bioremediation. In: Ollivier, B., Magot, M. (Eds.),
419 Petroleum Microbiology. ASM Press, Washington, DC, 2005, pp. 317-336.
- 420 [9] N. Das and P. Chandran. *Biotechn. Rese. Interna*, 2011, **20**, 1-12.
- 421 [10] Y. Chen, C. Li, Z. Zhou, J. Wen. *Appl. Biochem. Biotechnol.* 2014, **172**(7), 3433-3447.
- 422 [11] J. Zampolli, E. Collina, M. Lasagni, P.D. Gennaro. *AMB Express*. 2014, **4**, 73-82.
- 423 [12] W.P. Wang, Z.Z. Shao. *Nat Commun*, 2014, **5**, 5755-5760.
- 424 [13] E. Ivanova, M.V. Sukhacheva, A.Y. Kanateva, I.K. Kravchenko, A.A. Kurganov.
425 *Microbiology*. 2014, **83**(6), 764-772.
- 426 [14] M.C. Portillo, M. Santana, J.M. Gonzalez. *Naturwissenschaften*. 2012, **99**, 43-53.
- 427 [15] T. Kato, M. Haruki, T. Imanaka, M. Morikawa, S. Kanaya. *J.Biosci. Bioengin.* 2001, **91**(1), 64-70.
- 428 [16] C. Meintanis, K.I. Chalkou, K.A. Kormas, A.D. Karagouni. *Biodegradation*, 2006, **17**(2), 105-111.
- 429 [17] G. Z. Gu, Z. Li, D.F. Zhao, C.C. Zhao. *China. Pet. Process. Petrochem. Tech.* 2013, **15**(2), 82-90.
- 430 [18] N. M. Shestakova, A. V. Korshunova, E. M. Mikhailova, D. Sh. Sokolova, T. P. Tourova, S. S.
431 Belyaev, A. B. Poltarau, T. N. Nazina. *Microbiology*. 2011, **80**(1), 60-69.
- 432 [19] A.I. Slobodkin, G.B. Slobodkin. *Microbiology*, 2014, **83**(3), 255-270
- 433 [20] B. Khazra, S.M. Mousavi, S. Mehrabi, M. Hashemi, S.A. Shojaosadati. *RSC. Adv.* 2015, **5**,

- 434 33414-33422.
- 435 [21] F.J. Deive, A. Dominguez, T. Barrio, F. Moscoso, P. Moran, M.A. Longo, M.A. Sanroman. *J.*
- 436 *Hazard. Mater.* 2010, **182**, 735-742
- 437 [22] M.H. Al-Jailawi, M.S. Mahdi, A.M.A. Fadhil. *Inter. J. Biotechnol.*, 2013, **111**, 275-283.
- 438 [23] A.M.A. Fadhil, M.H. Al-Jailawi, M.S. Mahdi. *Int. J. Adv. Res.*, 2014, **2**(3), 795-805.
- 439 [24] Y.H. Gursahani, S.G. Gupta. *J. Pet. Environ. Biotechnol.* 2011, **111**(2), 1-4.
- 440 [25] K.M. Goh, U.M. Kahar, Y.Y. Chai, C.S. Chong. *Appl. Microbiol. Biotechnol.* 2013, **97**, 1475-1488.
- 441 [26] Z.Y. Zhao, A.S. Jonathan, W.C. Wong. *Bioresour. Technol.* 2011, **102**(5), 3999-4007.
- 442 [27] Y. Emine. *Environ. Technol.*, 2011, **32**(15), 1743-1749.
- 443 [28] D. Kundu, C. Hazra, N. Dandi, A. Chaudhari. *Biodegradation.* 2013, **24**, 775-793.
- 444 [29] L. Fracchia, M. Cavallo, M.G. Martinotti, I.M. Banat (2012). Biosurfactants and
- 445 bioemulsifiers biomedical and related applications-present status and future potentials, biomedical
- 446 science, engineering and technology, Prof. Dhanjoo N. Ghista (Ed.), ISBN: 978-953-307-471-9,
- 447 InTech, DOI: 10.5772/23821.
- 448 [30] C. Uzoigwe, J. G. Burgess, C.J. Ennis, P.K.S.M. Rahman. *Front. Microbiol.* 2015, **6**, 245.
- 449 [31] S.K. Satpute, I.M. Banat, P.K. Dhakephalkar, A.G. Banpurkar, B.A. Chopade. *Biotechnol. Adv.*
- 450 2010, **28**, 436-450
- 451 [32] P. Tribedi, A.K. Sil. *J. Appl. Microbiol.* 2014, **116**(2), 295-303.
- 452 [33] W.J. Xia, H.P. Dong, L. Yu, D.F. Yu. *Colloids Surf A Physicochem Eng Asp.* 2011, **392**(1),
- 453 124-130.
- 454 [34] M. Mesbah, U. Premachandran, W.B. Whitman. *Int. J. Syst. Bacteriol.* 1989, **39**, 159-167
- 455 [35] S.F. Altschul, W. Gish, W. Miller, E.W. Myers, D.J. Lipman. *J. Mol. Biol.* 1990, **215**(3), 403-410.

- 456 [36] L. Siristova, K. Melzoch, T. Režank. *Extremophiles*. 2009, **13**(1), 101-109.
- 457 [37] W.J. Xia, Z.F. Du, Q.F. Cui, H. Dong, F.Y. Wang, P.Q. He., Y.C. Tang. *J. Hazard. Mater.* 2014, **276**,
- 458 489-498.
- 459 [38] O. Carrion, L. Delgado, E. Mercade. *Carbohydr. Polym.* 2015, **6**, 1028-1034
- 460 [39] M. DuBois, K. Gilles, J. Hamilton, P. Rebers, F. Smith. *Anal. Chem.* 1956, **28**(3), 350-356.
- 461 [40] O.H. Lowry, N.J. Rosebrough, A.L. Farr., R.J. Randall. *J. Biol. Chem.* 1951, **193**(1), 265-275.
- 462 [41] M.S. Manocha, G. San-Blas, S. Centeno. *J. Gen. Microbiol.* 1980, **117**(1), 147-154.
- 463 [42] C. Zheng, J. He, Y. Wang, M. Wang, Z. Huang. *Bioresour. Technol.* 2011, **102**, 9155-9161
- 464 [43] M.F. Cao, C.J. Song, Y.H. Jin, L. Liu, J. Liu, H. Xie, W.B. Guo, S.F. Wang, *J. Mol. Catal. B*
- 465 *Enzym.*, 2010, **67**(1-2), 111-116.
- 466 [44] S.Z. Yang, D.Z. Wei, B.Z. Mu. *J. Biochem. Biophys. Methods*. 2007, **70**, 519-523.
- 467 [45] T.P. Tourova, T.N. Nazina, E.M. Mikhailova, T.A. Rodionova, A.N. Ekimov, V. Mashukova, A.B
- 468 Poltarau. *Mol. Biol.*, 2008, **42**(2), 217-226.
- 469 [46] L.G. Whyte, T.H. Smits, D. Labbe, B. Witholt, C.W. Greer, J.B. van Beilen. *Appl. Environ.*
- 470 *Microbiol.* 2002, **68**, 5933-5942.
- 471 [47] S. Dulger, Z. Demirbag, A.O. Belduz. *Int J Syst Evol Microbiol.* 2004, **54**, 1499-1503.
- 472 [48] A. Poli, E. Esposito, L. Lama, P. Orlando, G. Nicolaus, F. de Appolonia, A. Gambacorta, B.
- 473 Nicolaus. *Syst. Appl. Microbiol.*, 2006, **29**, 300-307.
- 474 [49] C. Schaffer, W.L. Franck, A. Scheberl, P. Kosma, T.R. McDermott, P. Messner. *Int. J. Syst. Evol.*
- 475 *Microbiol.* 2004, **54**, 2361-2368.
- 476 [50] A. Derekova, C. Sjöholm, R. Mandeva, M. Kambourova. *Extremophiles*, 2007, **11**(4), 577-583
- 477 [51] T.N. Nazina, T.P. Tourova, A.B. Poltarau. *Int. J. Syst. Evol. Microbiol.* 2001, **51**(2), 433-436.

- 478 [52] E. Pikuta, A. Lysenko, N. Chuvilskaya, U. Mendrock, H. Hippe, N. Suzina, D. Nikitin, G.
479 Osipov, K. Laurinavichius. *Int. J. Syst. Evol. Microbiol.* 2000, **50**, 2109-2117
- 480 [53] Y. Nie, C.Q. Chi, H. Fang, J.L. Liang, S.L. Lu, G.L. Lai, Y.Q. Tang, X.L. Wu. *Sci. Rep.* 2014,
481 **4968**, 1-6.
- 482 [54] H. Mehdi, E. Giti. *Int. Biodeter. Biodegr.* 2008, **62**, 170-178.
- 483 [55] C. Hazra, D. Kundu, A. Chaudhari. *RSC. Adv.*, 2014, DOI: 10.1039/C4RA13261K
- 484 [56] M. Bao, Y. Pi, L. Wang, P. Sun, Y. Li, L. Cao. *Environ. Sci.: Processes Impacts*, 2014, **16**, 897
- 485 [57] A. Perfumo, T. J. P. Smyth, R. Marchant, I.M. Banat. 2009. "Production and roles of
486 biosurfactant and bioemulsifiers in accessing hydrophobic substrates," in *Microbiology of*
487 *Hydrocarbons, Oils, Lipids and Derived Compounds*, ed Kenneth N. Timmis (Berlin; Heidelberg:
488 Springer-Verlag), 1502-1512.
- 489 [58] T.J.P. Smyth, A. Perfumo, R. Marchant, I.M. Banat (2010). "Isolation and Analysis of
490 Lipopeptide and high molecular weight biosurfactant," in *Handbook of Hydrocarbon and Lipid*
491 *Microbiology*, ed K. N. Timmis (Berlin; Heidelberg: Springer-Verlag), 3687-3704. doi:
492 10.1007/978-3-540-77587-4_290
- 493 [59] K.K. Sekhon-Randhawa (2014). "Biosurfactants produced by genetically manipulated
494 microorganisms: challenges and opportunities," in *Biosurfactants*, eds N. Kosaric and F. V. Sukan
495 (Boca Raton, FL: CRC Press), 49-67
- 496 [60] A.S. Monteiro, V.S. Domingues, M.V.D. Souza, I. Lula, D.B. Goncalves, E.P. de Siqueira, V.L. dos
497 Santos. *Biotechnol. Biofuels.* 2012, **5**, 29-34.

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521 **Fig. 1.** Phylogenetic relationship based on the 16S rRNA gene sequences between strain WJ-4 and
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539 **Fig.6.** Position of the *Anoxybacillus alkB* homologs on phylogenetic trees based on the nucleotide
540 sequences. Sequences obtained in our study are labeled with bold circle. Numerals indicate the
541 statistical reliability of branching order as determined by bootstrap analysis of 1000 alternative trees.

542 Values exceeding 90% are considered significant.

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Table 1Comparison of the phenotypic characteristics of *Anoxybacillus sp.* WJ-4 and related species

Characteristic	WJ-4	AB04	R270	MR3C	GS5-97
Cell length (um)	3.9-5.6	4.6	3.3-7.0	2.0-2.5	3.9-4.7
Oxygen	Facultative	Aerobic	Aerobic	Facultative	Aerobic
Motility	+	+	+	+	+
Spore shape	Spherical	Spherical	Spherical	Spherical	Oval
Optimal Temperature	68-72	50	55	61	55
Optimal pH	6.5-8.0	7.5-8.5	6.0-6.5	5.6	7.0
Catalase	+	+	+	+	
oxidase	+	+	+	+	
Arabinose	+	+	+	+	ND
Ribose	ND	ND	+	+	ND
Xylose	+	+	+	+	+
Fructose	+	+	+	+	+
Galactose	ND	ND	-	+	+
Mannose	+	+	ND	+	+
Rhamnose	+	-	+	+	-
Sucrose	+	+	-	+	+
Lactose	-	-	-	ND	w
Starch	+	+	+	+	ND
G+C(mol/mol)	44.3	54	41.7	43.5	42.4

+positive; -negative; w weak response; N.D. not determined.

Anoxybacillus ayderensis AB04T,⁴⁷ *Anoxybacillus amylolyticus* MR3C,⁴⁸*Geobacillus tepidamans* GS5-97,⁴⁹ *Anoxybacillus rupiensis* R270.⁵⁰

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Table 2

Cellular fatty acid composition (% w/w) of strain WJ-4 and some relationship closest strains in phylogenetic tree shown in Fig. 1.

Fatty acid	WJ-4	5965	GS5-97	MR3C	R270	LGM	3721
13:0 iso	-	-	0.4	-	-	-	-
14:0 iso	-	0.20	0.60	-	-	-	0.30
14:0	2.87	0.20	4.10	-	0.30	3.00	1.10
15:0 iso	48.79	22.90	44.33	41.2	52.81	52.00	16.90
15:0 ai	1.32	5.10	6.60	2.13	1.64	7.00	2.40
15:0	0.32	1.70	-	0.10	0.31	-	2.90
16:0 iso	5.68	7.30	3.20	7.00	2.01	5.00	15.20
16:0 ai	-	-	-	0.12	-	-	-
16:0	12.65	14.0	15.10	6.3	5.44	11.00	18.40
17:0 iso	28.45	25.40	15.00	31.60	33.55	12.00	29.90
17:0 ai	4.02	8.10	6.10	0.70	3.94	7.00	6.40
17:0	-	6.30	-	-	-	-	2.50
18:0 iso	0.30	0.7	0.60	1.30	-	-	1.30
18:1	-	-	-	0.70	-	-	0.10
18:0	-	2.2	-	1.90	-	-	1.40
19:0 iso		0.3					0.3
19:0		-					0.1

Anoxybacillus amylolyticus MR3C,⁴⁸ *Geobacillus tepidamans* GS5-97,⁴⁹

Anoxybacillus rupiensis R270, *Anoxybacillus contaminans* LGM,

Geobacillus Stearothermophilus 5965, *Geobacillus Stearothermophilus* 3721.⁴²

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Table 3Protein composition of the bioemulsifier from *Anoxybacillus* strain

Amino acid	Concentration (mg/ml)	Amino acid	Concentration (mg/ml)
Asp	6.126	Gly	3.236
Thr	2.781	Ala	6.519
Ser	2.031	Leu	2.773
Glu	7.231	Phe	1.121
Pro	1.142	Lys	2.452
Val	2.828	Ile	1.859

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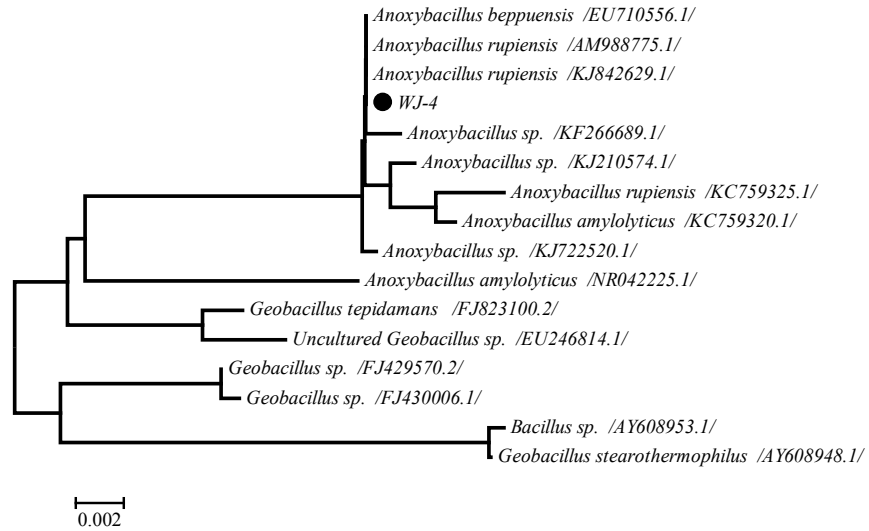


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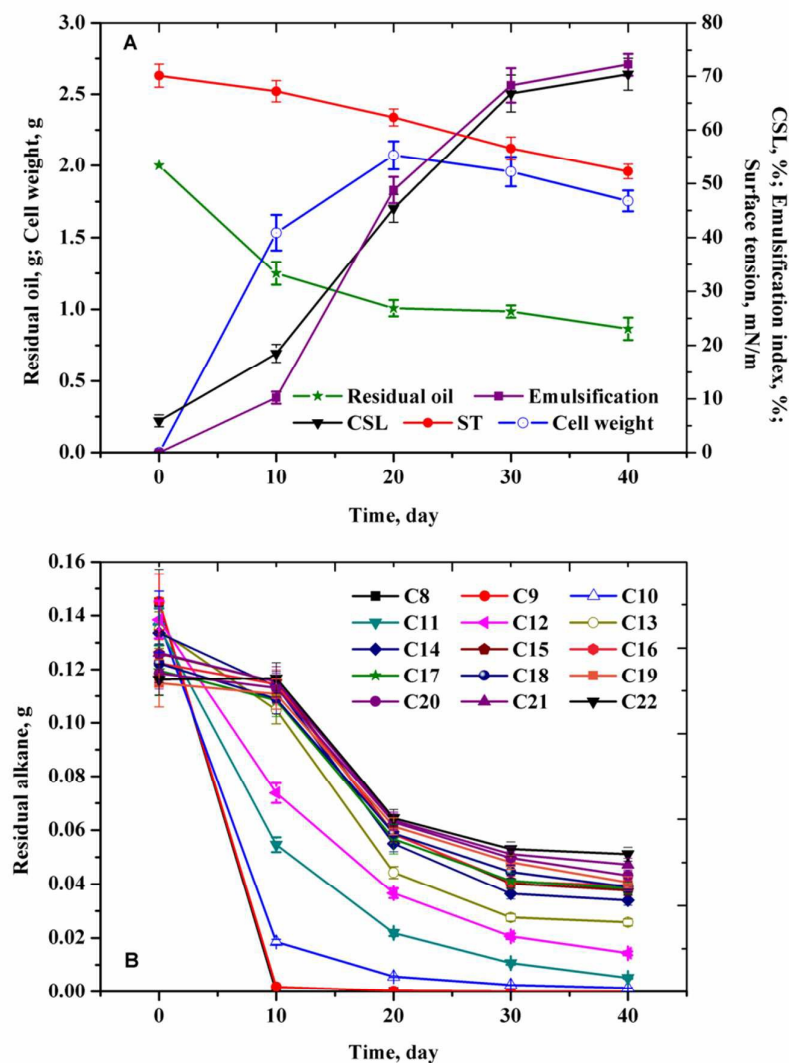


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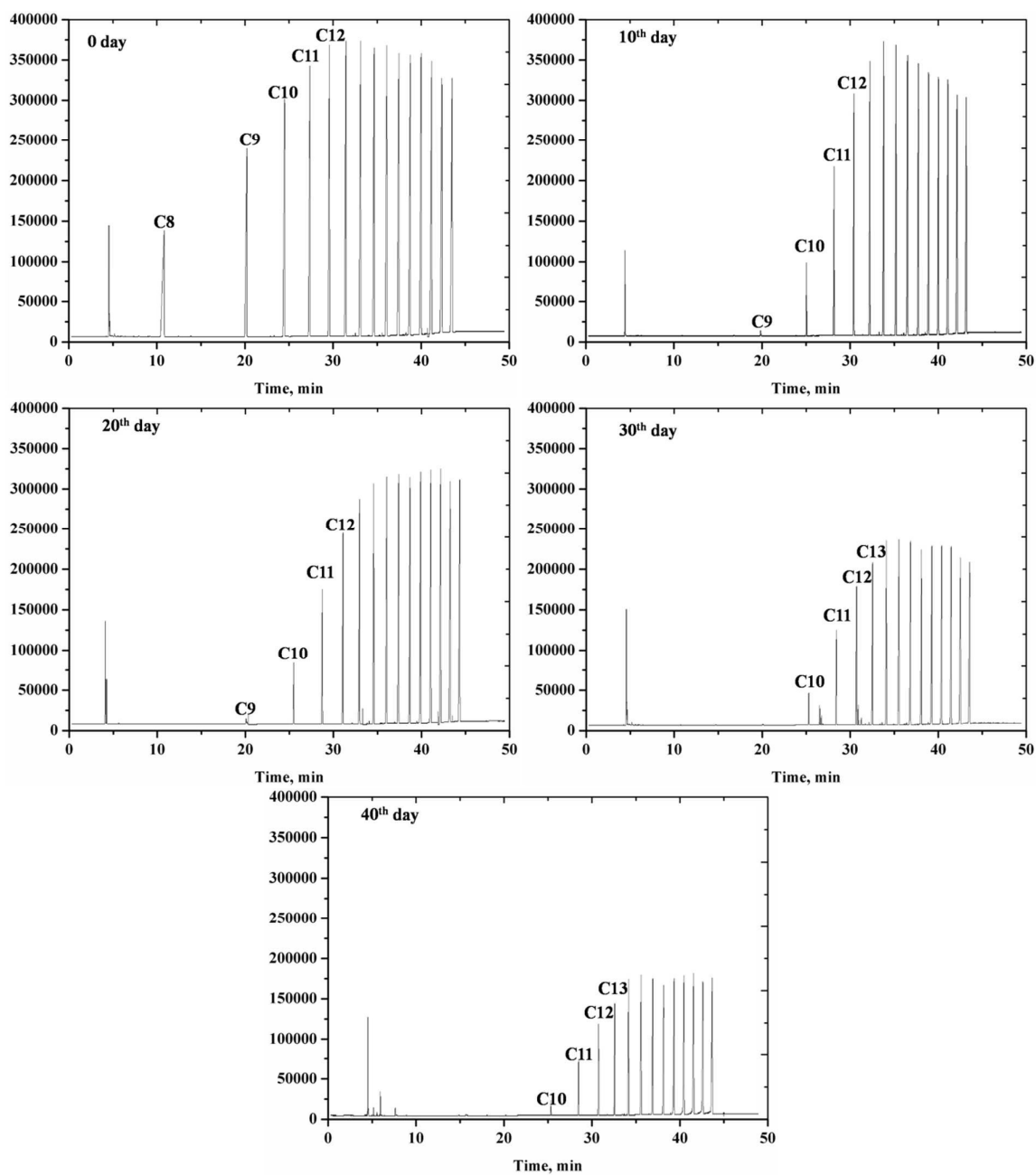


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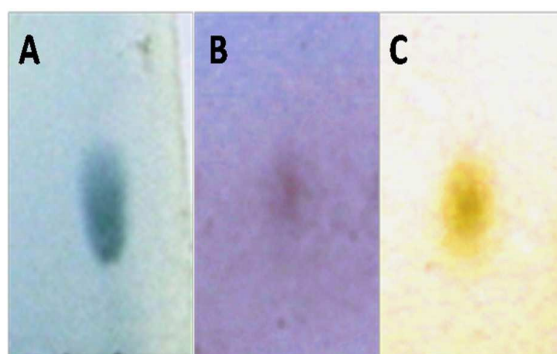


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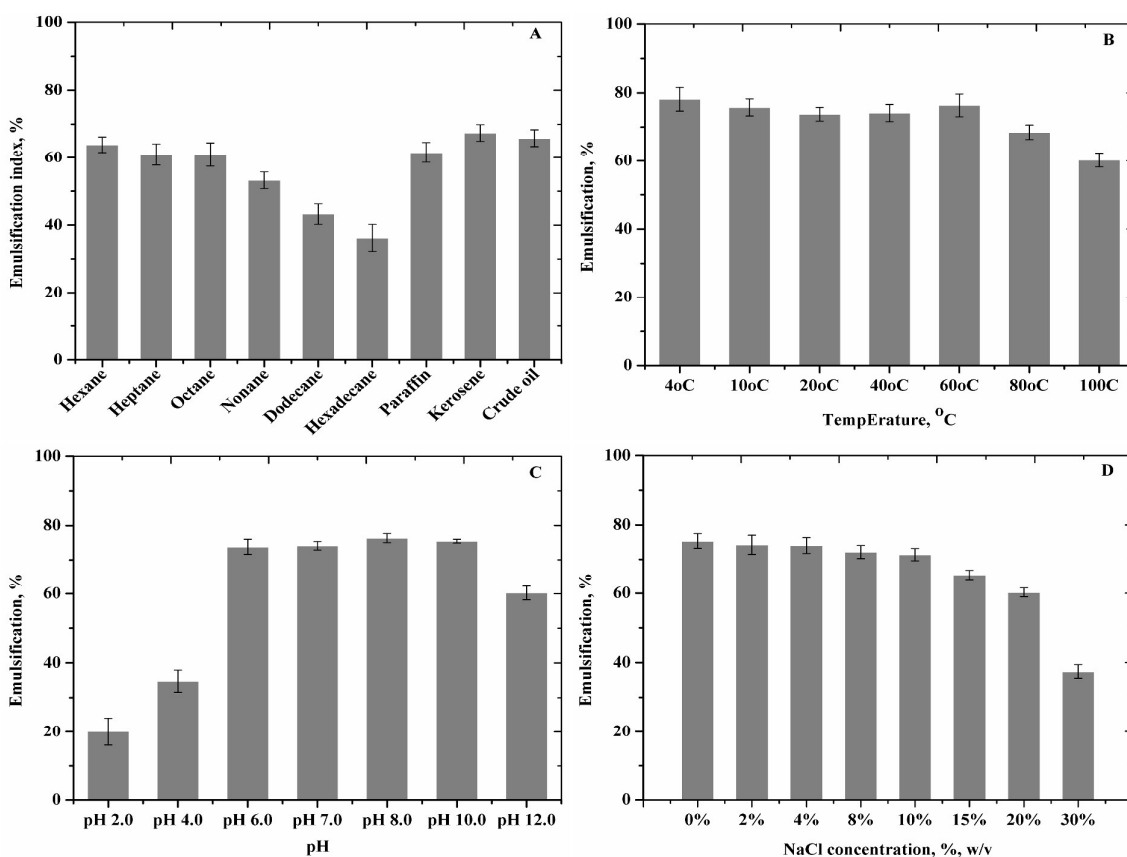


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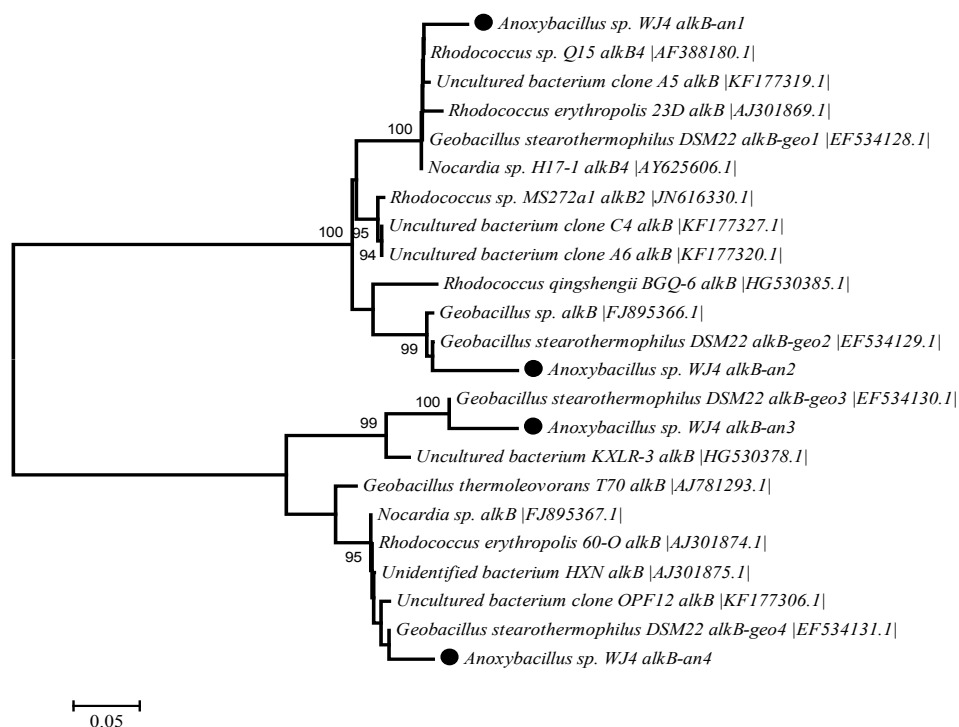


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