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# New $\alpha$-Glucosidase Inhibitors from a Marine Sponge-derived Fungus Aspergillus sp. OUCMDZ-1583 

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[^0]Eighteen new compounds named aspergones $\mathrm{A}-\mathrm{Q}(\mathbf{1}-\mathbf{1 7})$ and $6-O$-demethylmonocerin (18), along with five known analogues (19-23), were isolated from the fermentation broth of Aspergillus sp. OUCMDZ-1583 associated with an unidentified marine sponge XD10410 from the Xisha Islands of China. The structures including the absolute configurations were unambiguously elucidated by spectroscopic, X-ray crystallographic, chemical, and Mosher's methods along with quantum ECD calculation. Compounds $\mathbf{1}, \mathbf{2}, \mathbf{5}, \mathbf{1 0}, \mathbf{1 1}, \mathbf{1 4} \mathbf{- 1 8}$, and 21-23 showed $\alpha$-glucosidase inhibitions with $\mathrm{IC}_{50}$ values of $2.36,1.65,1.30,2.37,2.70,1.36,1.54,2.21,2.26,0.027,1.65,1.19$ and 1.74 mM , respectively (acarbose as positive control, $\mathrm{IC}_{50} 0.95 \mathrm{mM}$ ), among which compound $\mathbf{1 8}$ is 35 times more potent than acarbose. In addition, compounds $\mathbf{1 8}$ and $\mathbf{2 1}$ exhibited inhibitory activity against the influenza $\mathrm{A}(\mathrm{H} 1 \mathrm{~N} 1)$ virus with $\mathrm{IC}_{50}$ values of 172.4 and $175.5 \mu \mathrm{M}$, respectively (ribavirin as positive control, $\mathrm{IC}_{50} 137.3 \mu \mathrm{M}$ ).

Microorganisms continue to play an important role in the search for novel and bioactive compounds for drug development. ${ }^{1,2}$ However, with the deepening of research on the terrestrial microbial natural products, the discovery of new entities from these microorganisms is increasingly difficult due to chemical redundancy. ${ }^{3}$ As a result, many natural product chemists turned their attention to marine counterparts especially marine fungi that are supposed to be a tremendous resource for drug discovery for their special niches. ${ }^{4,5}$ With this trend in mind and as a continuation of our investigations on structurally new and bioactive natural products of marine fungal origin, ${ }^{6-9}$ an endozoic fungus Aspergillus sp. OUCMDZ-1583 was isolated from an unidentified marine sponge XD10410 from the Xisha Islands of China. The EtOAc extract of the fermentation broth showed $\alpha$-glucosidase inhibition with an $\mathrm{IC}_{50}$ value of $0.97 \mathrm{mg} / \mathrm{mL}$ while the $\mathrm{IC}_{50}$ value of acarbose (positive control) was $0.61 \mathrm{mg} / \mathrm{mL}$. Chemical examination of the fermentation broth resulted in the isolation and identification of eighteen new compounds that we named aspergones $\mathrm{A}-\mathrm{Q}(\mathbf{1}-\mathbf{1 7})$ and 6 -O-demethylmonocerin (18) as well as five known compounds, epoxyquinol (19), ${ }^{10} 7$-O-demethylmonocerin (20), ${ }^{11}(+)$-monocerin (21), ${ }^{11,12}$ fusarentin 6-methyl ether (22), ${ }^{11}$ and $6,7-O$-dimethyl-4R-hydroxy-10-epifusarentin (23) ${ }^{13}$ (Table S1, Supporting Information).

## Results and Discussion

Aspergone A (1) was obtained as a brown oil with the molecular formula of $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{3}$ from the HRESIMS peak at $m / z 195.1014[\mathrm{M}+\mathrm{H}]^{+}$. The IR spectrum of $\mathbf{1}$ showed the presence of a lactone carbonyl ( $1777 \mathrm{~cm}^{-1}$ ), and a hydroxy group ( $3377 \mathrm{~cm}^{-1}$ ). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Table 1) along with the HSQC spectrum showed the presence of one triplet methyl ( $\delta_{\mathrm{C/H}} 13.7 / 0.91$ ), three
methylenes including an oxygenated one, six olefinic carbons with four protonated, and one lactone carbonyls $\quad\left(\delta_{\mathrm{C}} \quad 170.1\right)$ These data were similar to the known 5-(E)-but-2-enylidene-3-propyl-5H-furan-2-one (24) ${ }^{14}$ except for an oxygenated methylene ( $\delta_{\mathrm{C} / \mathrm{H}}$ 63.3/4.24) replacement of the corresponding methyl signals, indicating hydroxylation of C-8 in $\mathbf{1}$. This deduction was further evidenced by the COSY cross-peaks from $\mathrm{H}-5\left(\delta_{\mathrm{H}} 5.67\right)$ to $\mathrm{H}_{2}-8\left(\delta_{\mathrm{H}} 4.24\right)$ through H-6 $\left(\delta_{\mathrm{H}} 6.74\right)$ and $\mathrm{H}-7\left(\delta_{\mathrm{H}} 6.04\right)$ and from $\mathrm{H}_{2}-9\left(\delta_{\mathrm{H}} 2.31\right)$ to $\mathrm{H}_{3}-11\left(\delta_{\mathrm{H}} 0.91\right)$ through $\mathrm{H}_{2}-10$ ( $\delta_{\mathrm{H}} 1.57$ ) as well as the HMBC correlations of $\mathrm{H}-3\left(\delta_{\mathrm{H}} 6.96\right)$ to $\mathrm{C}-1\left(\delta_{\mathrm{C}} 170.1\right)$, C-2 ( $\delta_{\mathrm{C}} 134.4$ ), C-4 ( $\delta_{\mathrm{C}} 148.0$ ), and C-9 ( $\delta_{\mathrm{C}} 27.4$ ), H-6 to C-4, H-5 to C-3 ( $\delta_{\mathrm{C}} 136.6$ ) and C-4, and of H-9 to C-1, C-2, and C-3 (Figure 1). The $E$ - and $Z$ - geometries of the $\Delta^{6}$ - and $\Delta^{4}$ - double bonds could be deduced from the large $J_{\mathrm{H}-6, \mathrm{H}-7}$ value ( 16.3 Hz ) (Table 1) and NOE difference experiment (Figure 2), respectively. The $\mathrm{H}-3$ signal ( $\delta_{\mathrm{H}} 6.96$ ) was significant enhanced when $\mathrm{H}-5$ signal ( $\delta_{\mathrm{H}} 5.67$ ) was irradiated.

Aspergone B (2) has the same molecular formula with 1 from the HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $217.0832[\mathrm{M}+\mathrm{Na}]^{+}$and the similar ${ }^{13} \mathrm{C}$ NMR data (Table 1). The difference was the hydroxymethine ( $\delta_{\mathrm{C} / \mathrm{H}} 66.3 / 4.04$ ) and a methyl signals ( $\delta_{\mathrm{C} / \mathrm{H}} 19.0 / 1.84$ ) replaced the corresponding hydroxymethylene and methylene signals of $\mathbf{1}$. HMBC correlations from $\mathrm{H}-3\left(\delta_{\mathrm{H}} 7.15\right)$ to $\mathrm{C}-1\left(\delta_{\mathrm{C}} 171.1\right)$, $\mathrm{C}-2\left(\delta_{\mathrm{C}} 129.7\right)$ and C-4 ( $\delta_{\mathrm{C}} 146.2$ ), $\mathrm{H}_{2}-9\left(\delta_{\mathrm{H}} 2.46,2.53\right)$ to $\mathrm{C}-1, \mathrm{C}-2$ and $\mathrm{C}-3\left(\delta_{\mathrm{C}} 139.3\right)$, and from $\mathrm{H}_{3}-8\left(\delta_{\mathrm{H}} 1.84\right)$ to C-6 ( $\delta_{\mathrm{C}} 124.9$ ) and C-7 ( $\delta_{\mathrm{C}} 136.5$ ) along with the ${ }^{1} \mathrm{H}^{-1} \mathrm{H}$ COSY data of $\mathrm{H}_{2}-9 / \mathrm{H}-10\left(\delta_{\mathrm{H}} 4.04\right) / \mathrm{H}_{3}-11\left(\delta_{\mathrm{H}}\right.$ 1.21) suggested the hydroxy group located at C-10 in 2. The $E$ - and $Z$ - geometries of the $\Delta^{6}$ - and $\Delta^{4}$ double bonds could be deduced from the large $J_{\mathrm{H}-6, \mathrm{H}-7}$ value ( 15.6 Hz ) (Table 1) and NOE difference experiment (Figure 2), respectively. The H-5 signal ( $\delta_{\mathrm{H}} 5.71$ ) was significantly enhanced when $\mathrm{H}-3$ signal ( $\delta_{\mathrm{H}} 7.15$ ) was irradiated. And the absolute configuration of $\mathrm{C}-10$ was determined by the modified Mosher's method. ${ }^{15}$ The $\Delta \delta$ values between (S)-MTPA ester (2b) and (R)- MTPA ester (2a)
clearly indicated the $10 S$-configuration (Figure 3).

Aspergones C (3) and D (4) were initially obtained as a racemic mixture from the zero value of specific rotation and the same molecular formula of $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{3}$ from a HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $219.0988[\mathrm{M}+\mathrm{Na}]^{+}$. Although the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data revealed the presence of the same furan-2(5H)-one nucleus as 1 , the remaining portion was slightly different. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY data (supporting information) and HMBC correlations of H-7 ( $\delta_{\mathrm{H}} 3.96$ ) to C-5 ( $\delta_{\mathrm{C}} 110.4$ ) and $\mathrm{H}-5\left(\delta_{\mathrm{H}} 5.26\right)$ to C-3 ( $\delta_{\mathrm{C}} 137.0$ ) and C-4 ( $\delta_{\mathrm{C}} 149.8$ ) revealed that 3-hydroxybutylidene group in $\mathbf{3}$ and $\mathbf{4}$ replaced the corresponding 4-hydroxybutenylidene group in $\mathbf{1}$. The $Z$-geometry of the $\Delta^{4}$ - double bond was deduced from the NOE enhancements of the $\mathrm{H}-5$ signal after irradiation of the $\mathrm{H}-3\left(\delta_{\mathrm{H}} 6.99\right)$ (Figure 2). Upon chiral chromatography over a CHIRAPAK IA HPLC column, optically pure $\mathbf{3}$ and $\mathbf{4}$ were obtained. The distribution of $\Delta \delta$ values between the $(S)$ - and $(R)$-MTPA esters ( $\mathbf{3 a}$ and $\mathbf{3 b}$ ) indicated the $7 R$-configuration of $\mathbf{3}$ (Figure 3). Therefore, the absolute configuration of $\mathbf{4}$ was determined to be $7 S$.

The molecular formula of aspergone $\mathrm{E}(\mathbf{5})$ was established as $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{3}$ by HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $219.0989[\mathrm{M}+\mathrm{Na}]^{+}$with four degrees of unsaturation. Strong IR absorption at 1769 and $1645 \mathrm{~cm}^{-1}$ implied the presence of lactone carbonyl and double bond functional groups. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY cross-peaks from $\mathrm{H}_{2}-8\left(\delta_{\mathrm{H}} 5.30,5.37\right)$ through H-7 $\left(\delta_{\mathrm{H}} 6.68\right)$, H-6 $\left(\delta_{\mathrm{H}} 6.31\right)$, H-5 $\left(\delta_{\mathrm{H}} 5.41\right)$, H-4 $\left(\delta_{\mathrm{H}}\right.$ 4.98), $\mathrm{H}-3\left(\delta_{\mathrm{H}} 3.97\right), \mathrm{H}-2\left(\delta_{\mathrm{H}} 2.62\right), \mathrm{H}_{2}-9\left(\delta_{\mathrm{H}} 1.84,1.58\right)$ and $\mathrm{H}_{2}-10\left(\delta_{\mathrm{H}} 1.57,1.51\right)$ to $\mathrm{H}_{3}-11\left(\delta_{\mathrm{H}} 0.96\right)$ and HMBC correlations from $\mathrm{H}-2$ and $\mathrm{H}_{2}-9$ to $\mathrm{C}-1$ ( $\delta_{\mathrm{C}}$ 175.9) indicated a 4-(but-1,3-dienyl)-3-hydroxy-2-propylbutyrrolactone structure (figure 2). The Z- geometry of the $\Delta^{5}$-double bond could be determined by the $J_{\mathrm{H}-5, \mathrm{H}-6}$ value ( 10.2 Hz , Table 1 ) and NOE enhancement of H-7 after irradiation of H-4. Further, the NOE enhancements of H-5 and H-9 were observed when

H-3 was irradiated while H-2 signal was enhanced after irradiation of H-4, indicating trans-orientations of both H-2 and H-4 with H-3 (Figure 2). The distribution of $\Delta \delta$ values between $(S)$ - and $(R)$-MTPA esters ( $\mathbf{5 a}$ and $\mathbf{5 b}$ ) indicated the $3 S$-configuration (Figure 3). Thus, the absolute configuration of $\mathbf{5}$ was determined to be $2 S, 3 S$, and $4 R$.

The molecular formula of aspergone $\mathrm{F}(\mathbf{6})$ was established as $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2}$ by HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $203.1037[\mathrm{M}+\mathrm{Na}]^{+}$with an oxygen less than in $\mathbf{5}$. Comparison of ${ }^{13} \mathrm{C}$ NMR data between $\mathbf{6}$ and $\mathbf{5}$ (Table 1) revealed that an olefinic quaternary carbon, an olefinic methine and an oxygenated methylene signals in 6 replaced the corresponding $\mathrm{sp}^{3}$-methine, $\mathrm{sp}^{3}$-methylene and ester carbonyl signals in 5. Two separated ${ }^{1} \mathrm{H}^{-1} \mathrm{H}$ COSY systems of $\mathrm{H}-3 / \mathrm{H}-4 / \mathrm{H}-5 / \mathrm{H}-6 / \mathrm{H}-7 / \mathrm{H}-8$ and $\mathrm{H}-9 / \mathrm{H}-10 / \mathrm{H}-11$ (supporting information) were observed in 6, indicating that the propylidene group replaced the propyl group in 5 . And the key HMBC data from $\mathrm{H}_{2}-1\left(\delta_{\mathrm{H}} 4.27,4.18\right)$ to $\mathrm{C}-2\left(\delta_{\mathrm{C}} 140.2\right), \mathrm{C}-4\left(\delta_{\mathrm{C}} 86.6\right)$ and $\mathrm{C}-9\left(\delta_{\mathrm{C}} 125.7\right)$ and from $\mathrm{H}-9\left(\delta_{\mathrm{H}} 5.36\right)$ to $\mathrm{C}-1\left(\delta_{\mathrm{C}} 70.1\right)$ confirmed the replacement of the carbonyl group in $\mathbf{5}$ by a $-\mathrm{CH}_{2}-$ group in $\mathbf{6}$. The $E$-geometry of the $\Delta^{5}$-double bond was deduced from the large $J_{\mathrm{H}-5, \mathrm{H}-6}$ value ( 15.7 Hz , Table 1) and the NOE enhancement of H-6 ( $\delta_{\mathrm{H}} 6.21$ ) after irradiation of H-4 ( $\delta_{\mathrm{H}} 4.09$ ) (Figure 2). The NOE enhancements of H-5 ( $\delta_{\mathrm{H}} 5.71$ ) and H-9 after irradiation of H-3 ( $\delta_{\mathrm{H}} 4.22$ ) (Figure 2) indicated the trans-orientation of H-3 and H-4 and E-geometry of $\Delta^{2(9)}$-double bond. The distribution of $\Delta \delta$ values between $(S)$ - and $(R)$-MTPA esters ( $\mathbf{6 a}$ and $\mathbf{6 b}$ ) indicated the $3 S$-configuration (Figure 3). Thus, the absolute configuration of $\mathbf{6}$ was determined as $3 S$ and $4 R$.

Aspergones $G(7)$ and $H(8)$ were initially isolated as a racemic mixture and the molecular formula was determined as $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2}$ on the basis of HRESIMS peak at $m / z 203.1038[\mathrm{M}+\mathrm{Na}]^{+}$. The strong UV and IR absorptions at $\lambda_{\max } 223 \mathrm{~nm}$ and $v_{\max } 1703$ and $1614 \mathrm{~cm}^{-1}$ indicated the presence of a
conjugated enone moiety which was further supported by the HMBC correlations from an olefinic proton ( $\delta_{\mathrm{H}-2} 5.87$ ) to a sp ${ }^{2}$-quaternary carbon $\left(\delta_{\mathrm{C}-3} 180.7\right)$ and a carbonyl carbon $\left(\delta_{\mathrm{C}-1} 206.9\right) .{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY (Figure 1) and HSQC data indicated the presence of two isolated spin systems, $\mathrm{CH}(4)-\mathrm{CH}(5)-\mathrm{CH}_{2}(6)-\mathrm{CH}(7)-\mathrm{CH}_{2}(8)$ and $\mathrm{CH}_{3}(11)-\mathrm{CH}_{2}(10)-\mathrm{CH}_{2}(9)$. The key HMBC correlations of H-4 ( $\delta_{\mathrm{H}} 4.45$ ) to $\mathrm{C}-1, \mathrm{C}-2\left(\delta_{\mathrm{C}} 128.9\right), \mathrm{C}-5\left(\delta_{\mathrm{C}} 55.3\right), \mathrm{C}-6\left(\delta_{\mathrm{C}} 31.7\right)$ and $\mathrm{C}-9\left(\delta_{\mathrm{C}} 32.9\right)$ linked these three moieties as 5-allyl-4-hydroxy-3-propylcyclopent-2-en-1-one. When H-4 was irradiated, the $\mathrm{H}_{2}-6\left(\delta_{\mathrm{H}} 2.56,2.21\right)$ signals were enhanced, indicating trans-orientation between $\mathrm{H}-4$ and $\mathrm{H}-5\left(\delta_{\mathrm{H}}\right.$ 2.37). A chiral HPLC separation over a CHIRAPAK IA column afforded the optically pure $d$-isomer (7) and $l$-isomer (8). The distribution of $\Delta \delta$ values between ( $S$ )- and ( $R$ )-MTPA esters ( $\mathbf{7 a}$ and $\mathbf{7 b}$ ) indicated the $4 R$ - configuration (Figure 3). Thus, the absolute configuration of 7 was determined to be $4 R$ and $5 R$. As a consequence, the absolute configuration of $\mathbf{8}$ was determined to be $4 S$ and $5 S$.

Aspergone I (9) showed a HRESIMS peak at $m / z 205.1195[\mathrm{M}+\mathrm{Na}]^{+}$corresponding to the molecular formula $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2}$. Its 1D NMR revealed two methyls ( $\delta_{\mathrm{C}} 17.9 \& 18.1$ ), an oxygenated methylene ( $\delta_{\mathrm{C}} 68.7$ ), six olefinic methines and an oxygenated quaternary carbon $\left(\delta_{\mathrm{C}} 75.0\right)$ (Table 1). ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and HSQC data indicated the presence of $\mathrm{CH}_{3}(8)-\mathrm{CH}(7)-\mathrm{CH}(6)-\mathrm{CH}(5)-\mathrm{CH}(4)-\mathrm{CH}_{2}(3)$ and $\mathrm{CH}(9)-\mathrm{CH}(10)-\mathrm{CH}_{3}(11)$ units (Figure 1). HMBC correlations (Figure 1) from $\mathrm{H}_{2}-1\left(\delta_{\mathrm{H}} 3.45,3.43\right)$ to $\mathrm{C}-2\left(\delta_{\mathrm{C}} 75.0\right), \mathrm{C}-3\left(\delta_{\mathrm{C}} 41.0\right)$ and $\mathrm{C}-9\left(\delta_{\mathrm{C}} 133.3\right)$ and from $\mathrm{H}_{2}-3\left(\delta_{\mathrm{H}} 2.31,2.32\right)$ to $\mathrm{C}-1\left(\delta_{\mathrm{C}} 68.7\right)$, C-2 and C-9 connected these structural moieties as 2-propenylocta-4,6-diene-1,2-diol (Figure 1). The large $J$ values of H-4/H-5 (14.5 Hz), H-6/H-7 (14.4 Hz) and H-9/H-10 (15.5 Hz) (Table 1) indicated all the $\Delta^{4,6,9}$ - double bonds as $E$-geometry. The absolute configuration was assigned using the in situ dimolybdenum CD method. ${ }^{16,17}$ After addition of $\mathrm{Mo}_{2}(\mathrm{OAc})_{4}$ into DMSO solution of 9, a metal complex auxiliary chromophore was generated. Because the contribution from the inherent CD was
subtracted, the Cotton effect observed in the induced CD spectrum originates solely from the chirality of the vic-diol moiety. The negative Cotton effect observed at $320(\Delta \varepsilon-0.64)$ and $400(\Delta \varepsilon$ $-0.10) \mathrm{nm}$ in the induced CD spectrum (Figure 5) revealed the $2 S$ - configuration according to Snatzke's empirical rule. ${ }^{18}$ The structure of 9 was therefore elucidated as ( $2 S, 4 E, 6 E$ )-2-( $E$-propenyl)octa-4,6-diene-1,2-diol.

Aspergone J (10) has the same molecular formula $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2}$ as 9 from a HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $205.1196[\mathrm{M}+\mathrm{Na}]^{+}$. NMR comparison revealed that a methylene and an olefinic methylene of $\mathbf{1 0}$ replaced a methyl and an olefinic methine of $9 .{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum of $\mathbf{1 0}$ further indicated a hexatrienyl and a propyl other than the hexadienyl and propenyl in $\mathbf{9}$, which were also supported by the key HMBC correlations from H-3 ( $\delta_{\mathrm{H}} 5.79$ ) to C-1 ( $\delta_{\mathrm{C}} 69.0$ ), C-2 ( $\delta_{\mathrm{C}} 75.2$ ) and C-9 $\left(\delta_{\mathrm{C}} 40.0\right)$. The $E$-geometries of $\Delta^{3,5}$ - double bonds could be deduced from the $J$ values of $\mathrm{H}-3 / \mathrm{H}-4(14.8 \mathrm{~Hz})$ and H-5/H-6 $(14.8 \mathrm{~Hz})($ Table 1$)$. And the same sign of $[\alpha]_{\mathrm{D}}(-55.3)$ to that of $9(-34.8)$ implied the same $2 R$ - configuration, that is $(2 R, 3 E, 5 E)$-2-propylocta-3,5,7-triene-1,2-diol.

Aspergone K (11) also has the molecular formula $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2}$ based on a HRESIMS peak at $\mathrm{m} / \mathrm{z}$ $205.1195[\mathrm{M}+\mathrm{Na}]^{+}$. The NMR data and ${ }^{1} \mathrm{H}-{ }^{-1} \mathrm{H}$ COSY and HMBC coupling modes along with $[\alpha]_{\mathrm{D}}$ (-64.5) were very similar to those of $\mathbf{1 0}$, indicating almost the same structure. The only difference performed in the $Z$-geometry of $\Delta^{5}$ - double bond from the relatively small $J_{\mathrm{H}-5, \mathrm{H}-6}$ value (11.3 Hz). Thus, structure of $\mathbf{1 1}$ was determined as ( $2 R, 3 E, 5 Z$ )-2-propylocta-3,5,7-triene-1,2-diol.

The molecular formula of aspergone $\mathrm{L}(\mathbf{1 2})$ was also established as $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2}$ by a HRESIMS peak at $m / z 205.1196[\mathrm{M}+\mathrm{H}]^{+}$. 1D NMR, ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and HSQC data indicated an oxygenated methylene, a 5-hydroxyhex-2-en-1-ylidene $\left(\mathrm{CH}_{3}(8)-\mathrm{CH}(7)-\mathrm{CH}_{2}(6)-\mathrm{CH}(5)-\mathrm{CH}(4)-\mathrm{CH}(3)\right)$, a propenyl $\left(\mathrm{CH}(9)-\mathrm{CH}(10)-\mathrm{CH}_{3}(11)\right)$, and an olefinic quaternary carbon. The key HMBC correlations of $\mathrm{H}-3\left(\delta_{\mathrm{H}}\right.$
6.03 ) to $\mathrm{C}-1\left(\delta_{\mathrm{C}} 62.4\right), \mathrm{C}-2\left(\delta_{\mathrm{C}} 135.7\right)$ and $\mathrm{C}-9\left(\delta_{\mathrm{C}} 125.8\right)$ along with $\mathrm{H}_{2}-1\left(\delta_{\mathrm{H}} 4.07\right)$ to $\mathrm{C}-2, \mathrm{C}-3\left(\delta_{\mathrm{C}}\right.$ 126.7) and C-9 connected above structural moieties as 2-propenylocta-2,4-diene-1,7-diol (Figure 2). All the geometries of $\Delta^{2,4,9}$ - double bonds were assigned as $E$ - according to the large $J$ value of H-4/H-5 (15.0 Hz) and H-9/H-10 (16.6 Hz) (Table 1) and the NOESY cross-peak between H-3 and $\mathrm{H}_{2}-1$ (Figure 2). To determine the absolute configuration of 12, the 1-O-t-butyldimethylsilyl (TBDMS) derivative (12a) and the $(R)$ - and $(S)$-MTPA esters of $\mathbf{1 2 a}$ were prepared. The distribution of $\Delta \delta$ values between ( $S$ )- and ( $R$ )-MTPA esters (12aa and 12ab) indicated the $7 R$-configuration (Figure 3). Thus, compound 12 was determined to be ( $7 R, 2 E, 4 E$ )-2-(E-propenyl)octa-2,4-diene-1,7-diol.

The molecular formula of aspergone $\mathrm{M}(\mathbf{1 3})$ was established as $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{O}_{2}$ by a HRESIMS peak at $m / z 207.1350[\mathrm{M}+\mathrm{Na}]^{+}$, equivalent to $\mathbf{1 2}$ with an additional $\mathrm{H}_{2}$ unit. Comparison of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data between $\mathbf{1 3}$ and $\mathbf{1 2}$ revealed that their structures were only different in the replacement of the propenyl group in 12 with a propyl group in 13. This deduction was further supported by the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY connectivity of $\mathrm{H}_{3}-11 / \mathrm{H}_{2}-10 / \mathrm{H}_{2}-9$ and the key HMBC correlations of $\mathrm{H}_{2}-1\left(\delta_{\mathrm{H}} 3.87\right)$ to C-2 $\left(\delta_{\mathrm{C}} 140.9\right), \mathrm{C}-3\left(\delta_{\mathrm{C}} 124.3\right)$ and C-9 $\left(\delta_{\mathrm{C}} 30.4\right)$. The $E$-geometry of both $\Delta^{2}$ - and $\Delta^{4}$ - double bonds were assigned from the large $J_{\mathrm{H}-4, \mathrm{H}-5}(15.4 \mathrm{~Hz})$ and the NOESY cross-peak of H-3 ( $\delta_{\mathrm{H}} 5.97$ ) with $\mathrm{H}_{2}-1$ (Figure 2). The close specific rotation of $\mathbf{1 3}$ to $\mathbf{1 2}(-36.7 \mathrm{vs}-30.5)$ implied the same $7 R-$ configuration. Thus, compound $\mathbf{1 3}$ was elucidated as ( $7 R, 2 E, 4 E$ )-2-propylocta-2,4-diene-1,7-diol.

Aspergone N (14) had a molecular formula $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}_{5}$ from a HRESIMS peak at $m / z 239.0888$ [ M $+\mathrm{Na}]^{+}$with three degrees of unsaturation. The IR spectrum of $\mathbf{1 4}$ showed the absorption of the double bond $\left(1645 \mathrm{~cm}^{-1}\right)$ and hydroxy ( $3287 \mathrm{~cm}^{-1}$ ). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR along with the HSQC data revealed the presence of a doublet methyl, an oxygenated methylene, four oxygenated methines,
two vicinal olefinic methines, two olefinic quaternary carbons, and five hydroxy groups. This was accounting for two degrees of unsaturation, indicating a cyclic nucleus in $14 .{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY correlations reveal the presence of $\mathrm{CH}(-\mathrm{OH})(3)-\mathrm{CH}(-\mathrm{OH})(4)-\mathrm{CH}(-\mathrm{OH})(5)-\mathrm{CH}(-\mathrm{OH})(6)$, $\mathrm{CH}_{3}(9)-\mathrm{CH}(8)-\mathrm{CH}(7)$, and $\mathrm{CH}_{2}(-\mathrm{OH})(10)$ moieties. HMBC correlations from $\mathrm{H}-7\left(\delta_{\mathrm{H}} 6.34\right)$ to $\mathrm{C}-1$ $\left(\delta_{\mathrm{C}} 133.4\right), \mathrm{C}-2\left(\delta_{\mathrm{C}} 135.8\right)$ and $\mathrm{C}-6\left(\delta_{\mathrm{C}} 67.9\right)$ and from $\mathrm{H}_{2}-10\left(\delta_{\mathrm{H}} 4.18,3.92\right)$ to $\mathrm{C}-1, \mathrm{C}-2$, and $\mathrm{C}-3\left(\delta_{\mathrm{C}}\right.$ 68.5) connected these moieties as 2-hydroxymethyl-1-propenyl-1-cyclohexene-3,4,5,6- tetraol. The large $J_{\mathrm{H}-7, \mathrm{H}-8}(15.7 \mathrm{~Hz})$ suggested $E-\Delta^{7}$-double bond. The relative configurations of C-3-C-6 and the geometry of $\Delta^{7}$-double bond were supported by the x-ray single crystal diffraction (Figure 4). The absolute configurations of all the chiral centers were assigned as $R$ - by ECD calculations of 14 and ent-14 using the time-dependent density functional theory (TD-DFT) method at the B3LYP/6-31G(d) level. ${ }^{19}$ The results showed that the measured CD curve is matched well with the calculated ECD for 14 and opposite to that of ent-14 (Figure 6).

Aspergone O(15) was isolated as a colorless orthorhombic crystal with a molecular formula of $\mathrm{C}_{10} \mathrm{H}_{15} \mathrm{ClO}_{4}$ on the basis of a HRESIMS peak at $m / z 257.0548[\mathrm{M}+\mathrm{Na}]^{+}$, indicating the substitution of -Cl for -OH relative to $\mathbf{1 4}$. In addition, we noted that there was one hydroxy signal less and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and HMBC coupling patterns were similar to 14 except the lack of COSY between 4-OH and 4-CH, supporting the substitution of $4-\mathrm{Cl}$ in 15 for $4-\mathrm{OH}$ in 14 . The large $J_{\mathrm{H}-7, \mathrm{H}-8}(16.2 \mathrm{~Hz})$ indicated $E$-geometry of $\Delta^{7}$-double bond. The similar CD data with 14 (Figure 6 ) indicated ( $3 R, 4 S, 5 S, 6 R$ )configuration of 15 that was further confirmed by single-crystal X-ray diffraction (Figure 4) with a small Flack parameter (0.05(7)) and a heavy chlorine atom in molecule. ${ }^{20,21}$ Thus, aspergone O (15) was identified as $(3 R, 4 S, 5 S, 6 R, 7 E)$-4-chloro-2-hydroxymethyl-1-propenylcyclohex-1-ene-3,5,6triol.

Aspergone $\mathrm{P}(\mathbf{1 6})$ had the molecular formula of $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{O}_{4}$ as determined from its HRESIMS peak at $m / z 221.0780[\mathrm{M}+\mathrm{Na}]^{+}$, equivalent to 14 minus a $\mathrm{H}_{2} \mathrm{O}$ unit. The absence of two hydroxy proton signals and the upfield oxygenated methine carbon signals ( $\delta_{\mathrm{C}} 54.5,54.3$ ) indicated a replacement of two hydroxy groups by an epoxy group. This epoxidation was deduced to occur at C-4 and C-5 according to the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY of $\mathrm{OH}-3 / \mathrm{H}-3 / \mathrm{H}-4 / \mathrm{H} / 5 / \mathrm{H}-6 / \mathrm{OH}-6$ and the HMBC correlations from H-3 ( $\delta_{\mathrm{H}} 4.52$ ) to C-1 ( $\delta_{\mathrm{C}} 131.2$ ) along with H-6 ( $\delta_{\mathrm{H}} 4.45$ ) to C-2 $\left(\delta_{\mathrm{C}} 133.2\right)$, C-4 $\left(\delta_{\mathrm{C}} 54.5\right)$ and C-7 $\left(\delta_{\mathrm{C}}\right.$ 128.3). The large $J_{\mathrm{H}-7, \mathrm{H}-8}(15.8 \mathrm{~Hz})$ suggested $E$-geometry of $\Delta^{7}$-double bond. The absolute configuration of $\mathbf{1 6}$ was defined by chemical transformation into $\mathbf{1 5}$. After adding hydrochloric acid to a methanol solution of $\mathbf{1 6}$, compound 15 was yielded as a major product (Figure S120) which showed the identical $[\alpha]_{\mathrm{D}}$, NMR and MS data with those of natural one, indicating $(3 R, 4 R, 5 S, 6 R)$-configuration of $\mathbf{1 6}$. The result also indicated that compound $\mathbf{1 5}$ might be an artifact under acidic conditions formed in the fermentation process.

The molecular formula of aspergone Q (17) was established as $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{6}$ by a HRESIMS peak at $m / z 265.0680[\mathrm{M}+\mathrm{Na}]^{+}$. The NMR data (Table 1) of $\mathbf{1 7}$ are similar to those of $\mathbf{1 4}$, except for the absence of two hydroxy proton signals and the presence of an additional carbonate carbonyl signal ( $\delta_{\mathrm{C}} 154.8$ ), suggesting that two hydroxy groups of $\mathbf{1 4}$ were esterified to form a cyclic carbonate. HMBC correlations of $\mathrm{H}-5\left(\delta_{\mathrm{H}} 4.81\right)$ and $\mathrm{H}-6\left(\delta_{\mathrm{H}} 5.60\right)$ to C-11 ( $\delta_{\mathrm{C}} 154.8$ ) and H-7 ( $\delta_{\mathrm{H}} 6.42$ ) to C-6 ( $\delta_{\mathrm{C}} 75.5$ ) supported that the 5,6-dihydroxy was carbonated in $\mathbf{1 7}$ (Figure 2). Treatment of $\mathbf{1 7}$ in $1 \%$ aqueous sodium hydroxide yielded compound 14 which showed the identical $[\alpha]_{\mathrm{D}}$, NMR and MS data with the natural one, indicating $(3 R, 4 R, 5 S, 6 R)$ - configuration of 17 .

Compound 18 was found to have the molecular formula $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{6}$ by HRESIMS peak at $\mathrm{m} / \mathrm{z}$ 295.1171 $[\mathrm{M}+\mathrm{H}]^{+}$. Apart from the phenyl motif, the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Table 1) were nearly
identical to the known compound $\mathbf{2 0},{ }^{11}$ indicating the similar structure isomeric in phenyl motif. The HSQC revealed the presence of five aromatic quaternary carbons, one aromatic methine ( $\delta_{C H}$ 108.1/6.60), three oxygenated methines, three methylenes, one methyl ( $\delta_{C H} 14.1 / 0.90$ ), one methoxyl ( $\delta_{C H} 60.9 / 3.97$ ), and one ester carbonyl ( $\delta_{C} 168.2$ ). The key HMBC correlations from a phenolic hydroxy proton $\left(\delta_{\mathrm{HO}-8} 11.5\right)$ to three aromatic quaternary carbons $\left(\delta_{\mathrm{C}-7,8,8 \mathrm{a}} 134.9,155.6,101.3\right)$ and the methoxy protons ( $\delta_{\mathrm{H}} 3.97$ ) to a relative upfield oxygenated aromatic quaternary carbon ( $\delta_{\mathrm{C}-7}$ 134.9) revealed that $O$-methylation occurred at $7-\mathrm{OH}$ in $\mathbf{1 8}$ rather than $6-\mathrm{OH}$ in $\mathbf{2 0}$. The relative configuration was established from the NOE data and the comparison of NMR data with $\mathbf{2 0}$, especially vicinal coupling constants. The small $J_{\mathrm{H}-3, \mathrm{H}-4}(3.0 \mathrm{~Hz})$ (Table 1) along with the NOESY correlations from $\mathrm{H}-10\left(\delta_{\mathrm{H}} 4.10\right)$ to $\mathrm{H}-3\left(\delta_{\mathrm{H}} 5.02\right)$ and $\mathrm{H}-4\left(\delta_{\mathrm{H}} 4.49\right)$ indicated the same cis-orientation of H-3, H-4 and H-10 as in 20. The similar CD spectra between $\mathbf{1 8}$ and $\mathbf{2 0}$ (Figure 5) revealed the same $(3 R, 4 R, 10 S)$-configuration. Because the absolute configuration of $\mathbf{2 0}$ has been resolved by chemical synthesis ${ }^{11}$ and further by $\mathrm{Cu}-\mathrm{K} \alpha$ radiated x -ray single crystal diffraction (Figure S107) with a small Flack parameter $(0.0(3))$ in this paper, the structure of compound 18 could be clearly elucidated as 6-O-demethylmonocerin.

All the isolated compounds ( $\mathbf{1 - 2 3}$ ) were evaluated for antiviral activity against the H1N1 flu virus, as well as $\alpha$-glucosidase inhibition by preciously described methods. ${ }^{6,22}$ Only compounds $\mathbf{1 8}$ and 21 exhibited anti-H1N1 activities with $\mathrm{IC}_{50}$ values of $172.4 \mu \mathrm{M}$ and $175.5 \mu \mathrm{M}$, respectively (ribavirin as positive control, $\mathrm{IC}_{50} 137.3 \mu \mathrm{M}$ ), while the other compounds were not active against H1N1. These data suggested that the tetrahydrofuran moiety and the methoxy group located at C-7 in the isocoumarins were required for the antiviral activity against the influenza A H1N1 virus. In addition, compounds $\mathbf{1}, \mathbf{2}, \mathbf{5}, \mathbf{1 0}, \mathbf{1 1}, \mathbf{1 4} \mathbf{- 1 8}$, and $\mathbf{2 1} \mathbf{- 2 3}$ showed $\alpha$-glucosidase inhibition with $\mathrm{IC}_{50}$
values of $2.36,1.65,1.30,2.37,2.70,1.36,1.54,2.21,2.26,0.027,1.65,1.19$ and 1.74 mM (acarbose as positive control, $\mathrm{IC}_{50} 0.95 \mathrm{mM}$ ) (Table 3). The result showed that compound $\mathbf{1 8}$ is 35 times more potent than the positive control, which supports that marine fungi might be a promising source for drug discovery. Compounds $\mathbf{1 - 2 3}$ were also tested for cytotoxicities against A549 and K562 tumor cells by MTT ${ }^{23}$ and MCF-7 cells by CCK- $8^{24}$ methods, respectively. Their antimicrobial activities against Escherichia coli, Enterobacter aerogenes, Bacillus subtilis, Pseudomonas aeruginosa and Candida albicans were also evaluated by an agar dilution method. ${ }^{25}$ However, none of the compounds showed any activities against the tested tumor cells and pathogens.

## Experimental Section

General Experimental Procedures. Optical rotations were measured on a JASCO P-1020 digital polarimeter, and UV spectra were measured on a Beckman DU 640 spectrophotometer. CD data were collected using a JASCO J-715 spectropolarimeter. IR spectra were taken on a Nicolet Nexus 470 spectrophotometer as KBr discs. ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, DEPT, HMQC, HSQC, HMBC, COSY and NOESY spectra were acquired using a JEOL JNM-ECP 600 spectrometer or an Agilent 500 MHz DD2 spectrometer using TMS as an internal standard or residual solvent signals for referencing. HR-ESI-MS spectra were determined using the Q-TOF ULTIMA GLOBAL GAA076 LC mass spectrometer. Semi-preparative HPLC was carried out using an ODS column (YMC-pack ODS-A, $10 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}, 4 \mathrm{~mL} / \mathrm{min}$ ) and a $\pi \mathrm{NAP}$ column (COSMOSIL-pack, $10 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}, 4$ $\mathrm{mL} / \mathrm{min}$ ). Sea salt used was made from the evaporation of seawater collected in Laizhou Bay (Weifang Haisheng Chemical Factory). Thin layer chromatography (TLC) and column chromatography (CC) were performed on plates precoated with silica gel $\mathrm{GF}_{254}(10-40 \mu \mathrm{~m}$, Qingdao Marine Chemical Factory) and Sephadex LH-20 (Amersham Biosciences), respectively. Vacuum-liquid chromatography (VLC) utilized silica gel H (Qingdao Marine Chemical Factory).

Fungal Material and Fermentation. The fungus Aspergillus sp. OUCMDZ-1583 was isolated from a piece of sponge XD10410 from the Xisha Islands of China in August 2010. After it was ground into powder, the sample ( 1 g ) was diluted to $10^{-2} \mathrm{~g} / \mathrm{mL}$ with sterile water, $100 \mu \mathrm{~L}$ of which was deposited on a PDA ( 200 g potato, 20 g glucose, 20 g agar per liter of tap water) plate containing chloramphenicol ( $100 \mu \mathrm{~g} / \mathrm{mL}$ ) as a bacterial inhibitor. A single colony was transferred onto another PDA plate and was identified according to its morphological characteristics and ITS gene sequences (GenBank accession No. KM056275, Supporting Information). A reference culture of Aspergillus sp. OUCMDZ-1583 maintained at $-80^{\circ} \mathrm{C}$ is deposited in our laboratory. The isolate was cultured on slants of PDA medium at $28^{\circ} \mathrm{C}$ for 5 days. Plugs of agar supporting mycelium growth were cut and transferred aseptically to $200 \times 1000 \mathrm{~mL}$ Erlenmeyer flasks each containing 300 mL of liquid medium ( $2 \%$ mannitol, $2 \%$ maltose, $1 \%$ glucose, $1 \%$ monosodium glutamate, $0.3 \%$ yeast extract, $0.05 \%$ corn meal, $0.05 \% \mathrm{KH}_{2} \mathrm{PO}_{4}, 0.03 \% \mathrm{MgSO}_{4}, 1.75 \% \mathrm{Na}_{2} \mathrm{HPO}_{4} \cdot 2 \mathrm{H}_{2} \mathrm{O}, 1.05 \% \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{O}_{7} \cdot \mathrm{H}_{2} \mathrm{O}, 3.3 \%$ sea salt, pH 7.0 ). The flasks were incubated at room temperature under static conditions for 30 days.

Extraction and Isolation. The cultures ( 50 L ) were filtered through cheesecloth to separate the mycelial mass from the aqueous layer. The filtrate was then extracted three times by 3 -fold volumes of EtOAc, while the mycelium was extracted by acetone. After removing acetone by evaporation under vacuum, the obtained aqueous acetone solution was extracted three times with equal volumes of EtOAc. The combined EtOAc extracts were dried under vacuum to produce 28.7 g of extract. The EtOAc extract was subjected to a silica gel VLC column, eluting with a stepwise gradient of $0 \%$, $20 \%, 40 \%, 60 \%, 80 \%$ and $100 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(\mathrm{v} / \mathrm{v})$, to give 20 fraction (fractions $1-20$ ). Fraction $1(1.2 \mathrm{~g})$ was subjected to Sephadex LH-20 chromatography ( $5 \times 200 \mathrm{~cm}$ ) with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (1:1) to afford two subfractions (1.1 and 1.2). Fraction $1.1(106 \mathrm{mg})$ was further
purified by HPLC over a $\pi \mathrm{NAP}$ column $\left(70 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to give compounds $\mathbf{2 3}\left(t_{\mathrm{R}} 11.8 \mathrm{~min}\right.$; 20 mg ) and $21\left(t_{\mathrm{R}} 16.0 \mathrm{~min} ; 24 \mathrm{mg}\right)$. Fraction $1.2(234 \mathrm{mg})$ was subjected to HPLC over ODS ( $55 \%$ $\left.\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield $\mathbf{2 0}\left(t_{\mathrm{R}} 9.5 \mathrm{~min}, 16.3 \mathrm{mg}\right), \mathbf{2}\left(t_{\mathrm{R}} 9.7 \mathrm{~min}, 10.2 \mathrm{mg}\right), \mathbf{1}\left(t_{\mathrm{R}} 12.6 \mathrm{~min}, 7.6 \mathrm{mg}\right)$ and $\mathbf{1 8}\left(t_{\mathrm{R}} 13.8 \mathrm{~min}, 7.6 \mathrm{mg}\right)$. The fraction $2(3.4 \mathrm{~g})$ eluted with petroleum $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(100: 1)$ was subjected to reversed-phase C18 silica column eluting with a stepwise gradient of $30 \%$ to $100 \%$ MeOH in $\mathrm{H}_{2} \mathrm{O}$ to obtain four subfractions (2.1-2.4). Fraction 2.1 ( 247 mg ) was further separated into three subfractions (2.1.1-2.1.3) by Sephadex LH-20 chromatography ( $5 \times 200 \mathrm{~cm}$ ) eluting with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(1: 1, \mathrm{v} / \mathrm{v})$. Fractions 2.1.1 ( 56 mg ) was subjected to HPLC over ODS ( $55 \%$ $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}$ ) to yield fractions 2.1.1.1, which was further purified by HPLC over a $\pi \mathrm{NAP}$ column $\left(50 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to give compound $5\left(t_{\mathrm{R}} 17.0 \mathrm{~min} ; 8.2 \mathrm{mg}\right)$ and racemic mixture of 7 and $8\left(t_{\mathrm{R}} 14.7 \mathrm{~min} ; 16 \mathrm{mg}\right)$ that was further subjected to HPLC over a CHIRAPAK IA column $(75 \%$ $\left.n-\mathrm{C}_{6} \mathrm{H}_{14} / \mathrm{EtOH}, \mathrm{v} / \mathrm{v}\right)$ to give the optically pure $7\left(t_{\mathrm{R}} 5.7 \mathrm{~min} ; 7.1 \mathrm{mg}\right)$ and $\mathbf{8}\left(t_{\mathrm{R}} 5.9 \mathrm{~min} ; 8.3 \mathrm{mg}\right)$. Fraction 2.1.2 ( 78 mg ) was subjected to HPLC over ODS $\left(35 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield racemic mixture of $\mathbf{3}$ and $\mathbf{4}\left(t_{\mathrm{R}} 7.7 \mathrm{~min} ; 24 \mathrm{mg}\right)$ that was further purified by a CHIRAPAK IA column $(75 \%$ $\left.\mathrm{n}-\mathrm{C}_{6} \mathrm{H}_{14} / \mathrm{EtOH}, \mathrm{v} / \mathrm{v}\right)$ to give optically pure $3\left(t_{\mathrm{R}} 6.9 \mathrm{~min} ; 11.3 \mathrm{mg}\right)$ and $4\left(t_{\mathrm{R}} 7.8 \mathrm{~min} ; 10.9 \mathrm{mg}\right)$. Fraction 2.1.3 ( 39 mg ) was subjected to HPLC over ODS $\left(50 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield compound $6\left(t_{\mathrm{R}} 21.3 \mathrm{~min} ; 11.3 \mathrm{mg}\right)$. Fractions $15-20(4.2 \mathrm{~g})$ were combined and re-chromatographed over Sephadex LH-20 $(5 \times 200 \mathrm{~cm}, \mathrm{MeOH})$ to afford four subfractions (Fr. 15.1-Fr. 15.4). Fraction 15.4 ( 706 mg ) was subjected to HPLC over ODS $\left(25 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to give $17\left(t_{\mathrm{R}} 6.2 \mathrm{~min} ; 76 \mathrm{mg}\right)$, $14\left(t_{\mathrm{R}} 4.0 \mathrm{~min} ; 22 \mathrm{mg}\right)$ and subfraction 15.4.1. Fraction $15.4 .1(215 \mathrm{mg})$ was fractionated by silica gel CC using petroleum ether-EtOAc (4:6, v/v) to afford $\mathbf{1 5}(35.5 \mathrm{mg}), \mathbf{1 6}(106.5 \mathrm{mg})$ and $\mathbf{1 9}(5.4$ $\mathrm{mg})$. Fractions 6 and $7(1.3 \mathrm{~g})$ were combined and chromatographed over Sephadex LH-20 ( $5 \times 200$
$\mathrm{cm}, \mathrm{MeOH})$ to afford three subfractions (Fr. 6.1-Fr.6.3). Fraction $6.1(78 \mathrm{mg})$ was subjected to HPLC over ODS ( $\left.55 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield compounds $\mathbf{2 4}\left(t_{\mathrm{R}} 6.7 \mathrm{~min} ; 24.3 \mathrm{mg}\right)$ and $22\left(t_{\mathrm{R}} 9.9\right.$ $\mathrm{min} ; 14.7 \mathrm{mg})$. Fraction $6.2(108 \mathrm{mg})$ was subjected to HPLC over $\mathrm{ODS}\left(50 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}\right.$, v/v) to yield compounds $\mathbf{9}\left(t_{\mathrm{R}} 26.6 \mathrm{~min} ; 18.3 \mathrm{mg}\right), \mathbf{1 0}\left(t_{\mathrm{R}} 27.9 \mathrm{~min} ; 3.7 \mathrm{mg}\right)$ and $\mathbf{1 1}\left(t_{\mathrm{R}} 30.5 \mathrm{~min} ; 4.5 \mathrm{mg}\right)$. Fraction $6.3(94 \mathrm{mg})$ was separated by HPLC over ODS $\left(50 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield compounds $12\left(t_{\mathrm{R}} 11.7 \mathrm{~min} ; 18.3 \mathrm{mg}\right)$ and $\mathbf{1 3}\left(t_{\mathrm{R}} 13.9 \mathrm{~min} ; 3.5 \mathrm{mg}\right)$.

Aspergone A (1): brown oil; UV (MeOH) $\lambda_{\text {max }}(\log \varepsilon) 325$ (3.67) and 208 (3.09) nm; IR (KBr) $v_{\max }$ 3377, 2984, 1777, $1715 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $\mathrm{m} / \mathrm{z}$ $195.1014[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{15} \mathrm{O}_{3}, 195.1016$ ).

Aspergone B (2): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-27.8(c \quad 0.1, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 324.8$ (3.63) and 208.6 (3.11) nm; IR (KBr) $v_{\max } 3416,2972,1766,1641 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 217.0832[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{3} \mathrm{Na}, 217.0832$ ).

Aspergone C (3): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-20.6(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 275.2(3.80)$ and 201.6 (3.10) nm; IR (KBr) $v_{\max } 3420,2918,1676,1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 219.0988[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{3} \mathrm{Na}, 219.0992$ ).

Aspergone D (4): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}+21.8(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 275.2$ (3.80) and 201.6 (3.10) nm; IR (KBr) $v_{\max } 3420,2918,1676,1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 219.0988[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{3} \mathrm{Na}, 219.0992$ ).

Aspergone E (5): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-16.8(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 225.2$ (3.78) and 204.3 (3.17); $\mathrm{CD}(c 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 226.5(+1.05)$ and $209(+2.35) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3420$, 2966, 1769, $1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 219.0989$ [M + $\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{3} \mathrm{Na}, 219.0992$ ).

Aspergone F (6): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}+18.7(c \quad 0.1, \mathrm{MeOH})$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 225.5(3.75)$ and $203.6(3.50) ; \mathrm{CD}(c 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 217.5(+3.20)$ and $210.5(-0.87) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3443$, 2966, 1680, $1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 203.1037$ [M + $\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{Na}, 203.1043$ ).

Aspergone G (7): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}+10.2(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 223(3.38)$ and 203.5 (2.99); $\mathrm{CD}(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 235(+1.96)$ and $207(-0.51)$; $\mathrm{IR}(\mathrm{KBr}) v_{\max } 3404,2960$, 1703, $1614 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS m/z 203.1038 $[\mathrm{M}+\mathrm{Na}]^{+}$ (calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{Na}, 203.1043$ ).

Aspergone H (8): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-9.1(c 0.1, \mathrm{MeOH})$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 223$ (3.38) and 203.5 (2.99); $\mathrm{CD}(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 235(-2.06)$ and $207(+0.54) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3404,2960$, 1703, $1614 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 203.1038[\mathrm{M}+\mathrm{Na}]^{+}$ (calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{Na}, 203.1043$ ).

Aspergone I (9): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-34.8(c \quad 0.1, \mathrm{MeOH})$, UV (MeOH) $\lambda_{\max }(\log \varepsilon) 229.5$ (3.45) and $204(2.94) ; \mathrm{CD}(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 233.5(+0.99)$ and $209(+0.40) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3420$, 2918, 1676, $1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 205.1195[\mathrm{M}+$ $\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{Na}, 205.1199$ ).

Aspergone J (10): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-55.3(c 0.1, \mathrm{MeOH})$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 253$ (3.34), 263 (3.57) and 274 (3.38); $\mathrm{CD}(c 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 273(+2.23)$ and $258.5(-0.51)$; $\mathrm{IR}(\mathrm{KBr}) v_{\max }$ 3420, 2949, 1680, $1641 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 205.1196$ [M $+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{Na}, 205.1199$ ).

Aspergone K (11): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-64.5(c 0.1, \mathrm{MeOH})$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 253$ (3.44), 263 (3.48) and 274 (3.38); $\mathrm{CD}(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 265(+2.54)$ and $249(-0.70)$; IR (KBr) $v_{\max }$

3392, 2964, 1676, $1645 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 205.1195$ [M
$+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{Na}, 205.1199$ ).
Aspergone L (12): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}-30.5(c \quad 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 258.5$ (3.79), 269.5 (3.87) and 280 (3.77); $\mathrm{CD}(c 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 276(+2.43)$ and $261.5(-1.10) ; \mathrm{IR}(\mathrm{KBr})$ $v_{\max } 3389,2918,1649,1614 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z$ $205.1196[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{Na}, 205.1199$ ).

Aspergone M (13): brown oil; $[\alpha]^{25}-36.7(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 233$ (3.72), 239 (3.74) and 248.5 (3.57); $\mathrm{CD}(c 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 251.5(+0.26)$ and $234(-0.55)$; $\mathrm{IR}(\mathrm{KBr})$ $v_{\max }$ 3389, 2918, 1649, $1614 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $\mathrm{m} / \mathrm{z}$ $207.1350[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{O}_{2} \mathrm{Na}, 207.1356$ ).

Aspergone $\mathbf{N}$ (14): colorless orthorhombic crystal; mp 123-125; $[\alpha]^{25}{ }_{\mathrm{D}}-73.5$ (c $\left.0.1, \mathrm{MeOH}\right)$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 239.5$ (3.39), 206 (3.03); CD $(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 230.5(-1.02) ; \mathrm{IR}(\mathrm{KBr})$ $v_{\max }$ 3287, 2910, 1645, $1443 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $\mathrm{m} / \mathrm{z}$ $239.0888[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}_{5} \mathrm{Na}, 239.0890$ ).

Aspergone O (15): colorless orthorhombic crystal; mp 117-119; $[\alpha]^{25}{ }_{\mathrm{D}}-12.7(c 0.1, \mathrm{MeOH})$, UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 238(3.58), 206(3.18) ; \mathrm{CD}(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 236.5(-1.67) ; \mathrm{IR}(\mathrm{KBr})$ $v_{\max }$ 3392, 2925, 1657, $1443 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z$ 257.0548 and $259.0518[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{10} \mathrm{H}_{15} \mathrm{O}_{4} \mathrm{ClNa}, 257.0551$ and 257.0522).

Aspergone P (16): white amorphous powder; $[\alpha]^{25}{ }_{\mathrm{D}}+81.0(c 0.1, \mathrm{MeOH})$, $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 238 (3.48), 206.5 (2.90); CD (c 0.1, MeOH) $\lambda_{\max }(\Delta \varepsilon) 232(-1.54) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3350,2921,1660$, $1446 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 221.0780[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\left.\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{Na}, 221.0784\right)$.

Aspergone Q (17): colorless oil; $[\alpha]^{25}{ }_{\mathrm{D}}-11.6(c 0.1, \mathrm{MeOH}), \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon) 238$ (3.36), 205.5 (2.97); CD $(c 0.05, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 231(-1.29) ; \mathrm{IR}(\mathrm{KBr}) v_{\max } 3350,2921,1660,1446 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 265.0680[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{6} \mathrm{Na}, 265.0683$ ).

6-O-Demethylmonocerin (18): brown oil; $[\alpha]^{25}{ }_{\mathrm{D}}+23.3$ (c $\left.0.1, \mathrm{MeOH}\right)$, $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 219 (3.58), 232.5 (3.53), 274 (3.37), 309 (3.02); $\mathrm{CD}(c \quad 0.1, \mathrm{MeOH}) \lambda_{\max }(\Delta \varepsilon) 274(-2.23), 242$ (+0.62), and $212.5(+2.68)$; IR (KBr) $v_{\max } 3335,2956,1660,1468,1376 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data, see Tables 1 and 2; HRESIMS $m / z 295.1171[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{O}_{6}$, 295.1176).

Preparation of 12-O-tert-butyldimethylsilylaspergone L (12a). tert-Butyldimethylsilyl chloride (TBDMSCl) ( 6.7 mg ) was added to a mixture of compound $\mathbf{1 2}(4.1 \mathrm{mg})$ and imidazole $(10 \mathrm{mg})$ in DMF ( 1.5 mL ). The reaction mixture was stirred at r.t. for 2 h . Then the reaction mixture was quenched with $\mathrm{H}_{2} \mathrm{O}$ and extracted three times with EtOAc. The EtOAc layers were combined and separated by semipreparative HPLC on ODS $(100 \% \mathrm{MeOH})$ to afford $\mathbf{1 2 a}\left(t_{\mathrm{R}} 4.4 \mathrm{~min}, 3.4 \mathrm{mg}\right)$ as one product (Figure S119).

12-O-tert-Butyldimethylsilylaspergone $\mathbf{L}(\mathbf{1 2 a}):{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.59(\mathrm{dd}, J=$ $15.4,11.8,1 \mathrm{H}, \mathrm{H}-4), 6.48(\mathrm{~d}, J=16.2,1 \mathrm{H}, \mathrm{H}-9), 6.14(\mathrm{~d}, J=11.2,1 \mathrm{H}, \mathrm{H}-3), 5.73(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-10)$, $5.69(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 4.32(\mathrm{~s}, 2 \mathrm{H}, \mathrm{H}-1), 3.86(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 2.34(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6 \mathrm{a}), 2.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6 \mathrm{~b})$, $1.81(\mathrm{~d}, J=6.6,3 \mathrm{H}, \mathrm{H}-11), 1.21(\mathrm{~d}, J=6.3,3 \mathrm{H}, \mathrm{H}-8), 0.92(\mathrm{~s}, 9 \mathrm{H}, \mathrm{H}-\mathrm{TBDMS}), 0.08(\mathrm{~s}, 6 \mathrm{H}$, H-TBDMS); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 125 \mathrm{MHz}\right) \delta 135.1$ (C, C-2), 130.3 (CH, C-5), 128.9 (CH, C-4), $125.7(\mathrm{CH}, \mathrm{C}-9), 125.6(\mathrm{CH}, \mathrm{C}-10), 124.5(\mathrm{CH}, \mathrm{C}-3), 67.3(\mathrm{CH}, \mathrm{C}-7), 63.8\left(\mathrm{CH}_{2}, \mathrm{C}-1\right), 43.0\left(\mathrm{CH}_{2}\right.$, C-6), $26.0\left(3 \times \mathrm{CH}_{3}, \mathrm{C}-\mathrm{TBDMS}\right), 22.8\left(\mathrm{CH}_{3}, \mathrm{C}-8\right), 19.0\left(\mathrm{CH}_{3}, \mathrm{C}-11\right), 18.4$ (C, C-TBDMS), -5.3 $\left(2 \times \mathrm{CH}_{3}, \mathrm{C}-\mathrm{TBDMS}\right) ;$ ESIMS $m / z 297.2[\mathrm{M}+\mathrm{H}]^{+}$.

Preparation of MTPA Esters. Compound 2 ( 1 mg for each) was reacted with either $R$-(-)- or $S$-(+)-MTPA chloride $(10 \mu \mathrm{~L})$ in anhydrous pyridine $(500 \mu \mathrm{~L})$ for 6 h . The reaction mixture was quenched with $\mathrm{H}_{2} \mathrm{O}$ and extracted three times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layers were combined and separated by semipreparative HPLC on ODS $\left(80 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to afford the $S$-MTPA ester 2a ( $1.2 \mathrm{mg}, t_{\mathrm{R}} 10.2 \mathrm{~min}$ ) and $R$-MTPA ester $\mathbf{2 b}\left(0.8 \mathrm{mg}, t_{\mathrm{R}} 10.3 \mathrm{~min}\right)$, respectively. By the same procedures, the $S$-MTPA esters $\mathbf{3 a}, \mathbf{5 a}, \mathbf{6 a}, \mathbf{7 a}$, and $\mathbf{1 2 a} \mathbf{a}$ and $R$-MTPA esters $\mathbf{3 b}, \mathbf{5 b}, \mathbf{6 b}, \mathbf{7 b}$, and $\mathbf{1 2 a b}$ were prepared, respectively.

Compound 2a: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.93(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 6.57(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 6.04(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-7), 5.65(\mathrm{~d}, J=11.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5), 5.38(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-10), 2.74\left(\mathrm{dd}, J=16.0,8.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 2.67$ $\left(\mathrm{dd}, J=16.0,4.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right), 1.90(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-8), 1.35(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z 411.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 2b: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.62(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 6.54(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 6.03(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-7), 5.53(\mathrm{~d}, J=11.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5), 5.37(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-10), 2.67\left(\mathrm{dd}, J=16.2,8.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 2.62$ $\left(\mathrm{dd}, J=16.2,5.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right), 1.89(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-8), 1.40(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z 411.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 3a: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.85(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 5.26(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 4.89(\mathrm{t}, J$ $=10.2 \mathrm{~Hz}, \mathrm{H}-5), 2.67\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.64\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right), 2.32(\mathrm{t}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H}-9), 1.56(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{H}-10), 1.38(\mathrm{~d}, J=8.3,3 \mathrm{H}, \mathrm{H}-8), 0.97(\mathrm{t}, J=9.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11) ; \operatorname{ESIMS} m / z 413.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 3b: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.94(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 5.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 5.07(\mathrm{t}, J$ $=10.3 \mathrm{~Hz}, \mathrm{H}-5), 2.74\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.71\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right), 2.34(\mathrm{t}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H}-9), 1.60(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{H}-10), 1.31(\mathrm{~d}, J=8.3,3 \mathrm{H}, \mathrm{H}-8), 0.97(\mathrm{t}, J=9.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11) ; \operatorname{ESIMS} m / z 413.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 5a: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.57(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 6.35(\mathrm{t}, J=11.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-6)$,
$5.45(\mathrm{t}, J=10.7 \mathrm{~Hz}, \mathrm{H}-5), 5.42\left(\mathrm{~d}, J=16.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.34\left(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 5.32(\mathrm{~m}, 1 \mathrm{H}$, H-4), $5.18(\mathrm{dd}, J=9.0,6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3), 2.70(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 1.79\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 1.66\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right)$, $1.61\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 1.40\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 0.91(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z 571.1[\mathrm{M}+$ $\mathrm{H}]^{+}$.

Compund 5b: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 6.29(\mathrm{t}, J=11.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-6)$, $5.43(\mathrm{t}, J=10.1 \mathrm{~Hz}, \mathrm{H}-5), 5.37\left(\mathrm{~d}, J=16.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.29\left(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 5.29(\mathrm{~m}, 1 \mathrm{H}$, H-4), $5.18(\mathrm{dd}, J=9.2,5.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3), 2.78(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 1.84\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 1.72\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right)$, $1.63\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 1.48\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 0.95(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z 571.1[\mathrm{M}+$ $\mathrm{H}]^{+}$.

Compound 6a: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.38(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 6.35(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 5.74(\mathrm{dd}, J=$ $14.5,5.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5), 5.69(\mathrm{~s}, \mathrm{H}-3), 5.60(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-9), 5.29\left(\mathrm{~d}, J=15.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.17(\mathrm{~d}, J=$ $\left.9.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 4.66(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4), 4.50\left(\mathrm{~d}, J=12.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-1\right), 4.38(\mathrm{~d}, J=12.6$ $\left.\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-1\right), 2.34\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 2.06\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 0.87(\mathrm{t}, J=7.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z$ $397.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 6b: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.37(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 6.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 5.76(\mathrm{~s}, \mathrm{H}-3)$, 5.73 (dd, $J=14.5,6.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5), 5.66(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-9), 5.28\left(\mathrm{~d}, J=15.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.18(\mathrm{~d}, J=$ $\left.9.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 4.50(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4), 4.50\left(\mathrm{~d}, J=12.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-1\right), 4.39(\mathrm{~d}, J=12.3$ $\left.\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-1\right), 2.34\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 2.16\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 1.0(\mathrm{t}, J=7.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11)$; ESIMS $m / z$ $397.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 7a: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.07(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-2), 5.90(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-4), 5.69(\mathrm{~m}, \mathrm{H}-7)$, $\left.5.25 \mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.12\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 2.63\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.53(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-5), 2.52\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right)$, $2.14\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 2.11\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right), 1.56\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 1.48\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 0.90(\mathrm{t}, J=6.5 \mathrm{~Hz}$,

3H, H-11); ESIMS m/z $397.1[\mathrm{M}+\mathrm{H}]^{+}$.
Compound 7b: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.10(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-2), 5.87(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-4), 5.67(\mathrm{~m}, \mathrm{H}-7)$, $5.23\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-8\right), 5.10\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-8\right), 2.61\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.47\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right), 2.42(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-5)$, $2.34\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-9\right), 2.27\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-9\right), 1.64\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-10\right), 1.58\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-10\right), 0.97(\mathrm{t}, J=6.5 \mathrm{~Hz}$, 3H, H-11); ESIMS m/z $397.1[\mathrm{M}+\mathrm{H}]^{+}$.

Compound 12aa: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.49$ (dd, $\left.J=15.5,11.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4\right), 6.41$ (d, $J$ $=16.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-9), 6.05(\mathrm{~d}, J=11.7 \mathrm{~Hz}, \mathrm{H}-3), 5.71(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-10), 5.52(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 5.22(\mathrm{~m}, 1 \mathrm{H}$, H-7), $4.30(\mathrm{~s}, 2 \mathrm{H}, \mathrm{H}-1), 2.45\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.40\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right), 1.80(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11), 1.35$ (d, $J=6.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-8), 0.92(\mathrm{~s}, 9 \mathrm{H}, \mathrm{H}-\mathrm{TBDMS}), 0.08$ ( $\mathrm{s}, 6 \mathrm{H}, \mathrm{H}-\mathrm{TBDMS}) ;$ ESIMS $m / z 513.1[\mathrm{M}+$ $\mathrm{H}]^{+}$.

Compound 12ab: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) \delta 6.57(\mathrm{dd}, J=15.5,11.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4), 6.44(\mathrm{~d}, J$ $=16.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-9), 6.11(\mathrm{~d}, J=11.7 \mathrm{~Hz}, \mathrm{H}-3), 5.71(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-10), 5.63(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 5.21(\mathrm{~m}, 1 \mathrm{H}$, H-7), $4.32(\mathrm{~s}, 2 \mathrm{H}, \mathrm{H}-1), 2.51\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{a}}-6\right), 2.44\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}_{\mathrm{b}}-6\right), 1.80(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-11), 1.28$ (d, $J=6.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-8$ ), 0.92 ( $\mathrm{s}, 9 \mathrm{H}, \mathrm{H}-\mathrm{TBDMS}$ ), 0.08 ( $\mathrm{s}, 6 \mathrm{H}, \mathrm{H}-\mathrm{TBDMS}$ ); ESIMS $m / z 513.1[\mathrm{M}+$ $\mathrm{H}]^{+}$.

Chemical transformation of 16 to 15 . To compound $16(2 \mathrm{mg})$ in $\mathrm{MeOH}(1 \mathrm{~mL})$ was added 15 $\mu \mathrm{L}$ concentrated $\mathrm{HCl}(37 \%)$. After stirred at r.t for 1 h , the reaction mixture was subjected to HPLC separation over ODS $\left(10 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}\right)$ to yield $\mathbf{1 5}\left(t_{\mathrm{R}} 15.5 \mathrm{~min}, 1.2 \mathrm{mg}\right)$.

Chemical transformation of $\mathbf{1 7}$ to 14 . To compound $\mathbf{1 7}(2 \mathrm{mg})$ in methanol ( 1 ml ) was added 500 $\mu \mathrm{L} \mathrm{NaOH}$ solution (2\%). After stirred at r.t. for 5 h , the reaction was neutralized to pH 7 by 1 MHCl . The obtained mixture was concentrated and then was added to $1.5 \mathrm{~mL} \mathrm{H} \mathrm{H}_{2} \mathrm{O}$. The $\mathrm{H}_{2} \mathrm{O}$ layer was extracted twice with 3 mL EtOAc and the combined EtOAc extracts were concentrated and
purified by HPLC over ODS ( $20 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, \mathrm{v} / \mathrm{v}$ ) to yield $\mathbf{1 4}\left(t_{\mathrm{R}} 4.6 \mathrm{~min}, 1.5 \mathrm{mg}\right)$.

X-ray crystal data for 14 and 15: Colorless crystals of $\mathbf{1 4}$ and $\mathbf{1 5}$ were obtained in $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ $(1: 1, \mathrm{v} / \mathrm{v})$. Crystal data of $\mathbf{1 4}$ were obtained on a Bruker APEX DUO area detector diffractometer with graphite monochromatic $\mathrm{Cu}-\mathrm{K} \alpha$ radiation $(\lambda=1.54178 \AA)$. Crystal data of $\mathbf{1 5}$ were obtained on a Bruker Smart CCD area detector diffractometer with graphite monochromatic Mo-K $\alpha$ radiation ( $\lambda$ $=0.71073 \AA$ ). Crystallographic data for $\mathbf{1 4}$ and $\mathbf{1 5}$ have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC 995364 and 995361, respectively. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal Data for Aspergone $\mathbf{N}$ (14). Monoclinic, $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}_{5} \mathrm{H}_{2} \mathrm{O}$; space group $P 2(1)$ with $a=$ $15.2823(12) \AA, b=4.5964(3) \AA, c=16.3515(14) \AA, V=1143.39(15) \AA^{3}, Z=4, D_{\text {calcd }}=1.361$ $\mathrm{Mg} / \mathrm{m}^{3}, \mu=0.957 \mathrm{~mm}^{-1}$, and $F(000)=504$; unique cell angle $(\beta)=95.4540(10)$; crystal size: $0.21 \times$ $0.14 \times 0.08 \mathrm{~mm}^{3} . T=293(2) \mathrm{K}$. The structure was solved by direct methods (SHELXS-97) and expanded using Fourier techniques (SHELXL-97). The final cycle of full-matrix least-squares refinement was based on 3891 unique reflections $\left(2 \theta<50^{\circ}\right)$ and 292 variable parameters and converged with unweighted and weighted agreement factors of $\mathrm{R} 1=0.1339, \mathrm{wR} 2=0.3268$, and $R=$ 0.2463 for $\mathrm{I}>2 \sigma(I)$ data. Absolute structure parameter is $0.0(12)$ that could not be used for determined the absolute configuration.

Crystal Data for Aspergone O (15). Trigonal, $\mathrm{C}_{10} \mathrm{H}_{15} \mathrm{ClO}_{4}$; space group $R 3$ with $a=24.752(2) \AA$, $b=24.752(2) \AA, c=5.0591(6) \AA, V=2684.3(4) \AA^{3}, Z=1, D_{\text {calcd }}=1.307 \mathrm{Mg} / \mathrm{m}^{3}, \mu=0.313 \mathrm{~mm}^{-1}$, and $F(000)=1116$; crystal size: $0.48 \times 0.37 \times 0.35 \mathrm{~mm}^{3} . T=298(2) \mathrm{K}$. The structure was solved by direct methods (SHELXS-97) and expanded using Fourier techniques (SHELXL-97). The final cycle
of full-matrix least-squares refinement was based on 2066 unique reflections $\left(2 \theta<50^{\circ}\right)$ and 137 variable parameters and converged with unweighted and weighted agreement factors of $\mathrm{R} 1=0.0378$, $\mathrm{wR} 2=0.0787$, and $R=0.0532$ for $\mathrm{I}>2 \sigma(I)$ data. Absolute structure parameter is $0.05(7)$.

## Conclusion

In conclusion, eighteen new compounds (1-18) were isolated from marine sponge-derived Aspergillus sp. strain OUCMDZ-1583. Compounds 5, 14 and 18 displayed comparable or stronger $\alpha$-glucosidase inhibition to acarbose $\left(\mathrm{IC}_{50} 0.95 \mathrm{mM}\right)$ with the $\mathrm{IC}_{50}$ values of $1.30,1.37$ and 0.027 mM , respectively. In addition, the $\alpha$-glucosidase inhibition of fusarentin 6-methyl ether (22) was reported here for the first time with an $\mathrm{IC}_{50}$ value of 1.19 mM .

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Supporting Information Available: Bioassay protocols used, the NMR spectra of compounds 19-23, the ITS gene sequences of Aspergillus sp. OUCMDZ-1583. These materials are available free of charge via the Internet.

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1: $R_{1}=O H, R_{2}=H, \Delta^{4}-Z$
2: $\mathrm{R}_{1}=\mathrm{H}, \mathrm{R}_{2}=\mathrm{OH}, \Delta^{4}-Z$


5: $\mathrm{R}_{1}=\mathrm{O}, \Delta^{5}-Z, \mathrm{R}_{2}=0-\sqrt{11}$
7: $4 R, 5 R$; 8: $4 S, 5 S$
24: $\mathrm{R}_{1}=\mathrm{R}_{2}=\mathrm{H}, \Delta^{4}-E$


10: $\Delta^{5}-E ; 11: \Delta^{5}-Z$





14: $\mathrm{R}=(\mathrm{R})-\mathrm{OH}, 5 R$ 16: $\mathrm{R}_{1}=\mathrm{OH}, \mathrm{R}_{2}=\mathrm{H}$
15: $\mathrm{R}=(S)-\mathrm{Cl}, 5 S$
19: $R_{1} R_{2}=0$

22: $\mathrm{R}_{1}=\mathrm{R}_{2}=\mathrm{H}, 10 \mathrm{~S}$
23: $\mathrm{R}_{1}=\mathrm{CH}_{3}, \mathrm{R}_{2}=\mathrm{OH}, 10 \mathrm{R}$

18: $\mathrm{R}_{1}=\mathrm{CH}_{3}, \mathrm{R}_{2}=\mathrm{H}$
20: $\mathrm{R}_{1}=\mathrm{H}, \mathrm{R}_{2}=\mathrm{CH}_{3}$
21: $\mathrm{R}_{1}=\mathrm{R}_{2}=\mathrm{CH}_{3}$




HMBC: $\frown$
cosy: -

Figure 1. Key COSY and HMBC correlations of 1-18.




12, 13

Figure 2. Key NOE correlations of compounds 1-7, 12, 13 and 18.


2


6



7


5


12a

Figure 3. $\Delta \delta\left(=\delta_{S}-\delta_{R}\right)$ values for $(S)$ - and ( $R$ )-MTPA esters of 2, 3, 5-7 and 12a.


14


15

Figure 4. ORTEP drawings of 14 and 15.


Figure 5. CD curve for the complex of $\mathbf{9}$ with $\mathrm{Mo}_{2}(\mathrm{OAc})_{4}$ subtracted from the inherent CD .


Figure 6. CD curves of $\mathbf{1 4}, \mathbf{1 5}, 18,20$ and calculated ECD of 14 and ent-14.

Table 1. ${ }^{13} \mathrm{C}$ NMR Data for $\mathbf{1 - 1 8}\left(\delta_{\mathrm{C}} \mathrm{ppm}\right)$

| position | $1{ }^{\text {bc }}$ | $2{ }^{\text {bd }}$ | 3 and $4^{\text {bc }}$ | $5{ }^{\text {bd }}$ | $6^{\text {ad }}$ | 7 and $\mathbf{8}^{\text {bc }}$ | $9^{\text {ac }}$ | $10{ }^{\text {ac }}$ | $11^{\text {ac }}$ | $12{ }^{\text {ac }}$ | $13{ }^{\text {ac }}$ | $14^{\text {ac }}$ | $15^{\text {ad }}$ | $16^{\text {ac }}$ | $17^{\text {ac }}$ | $18^{\text {bc }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 170.1, C | 171.1, C | 170.8, C | 175.9, C | 70.1, $\mathrm{CH}_{2}$ | 206.9, C | 68.7, $\mathrm{CH}_{2}$ | 69.0, $\mathrm{CH}_{2}$ | 68.9, $\mathrm{CH}_{2}$ | 62.4, $\mathrm{CH}_{2}$ | 65.0, $\mathrm{CH}_{2}$ | 133.4, C | 134.8, C | 131,2, C | 126.0, C | 168.2, C |
| 2 | 134.4, C | 129.7, C | 134.1, C | 47.6, CH | 140.2, C | 128.9, CH | 75.0, C | 75.2, C | 75.3, C | 135.7, C | 140.9, C | 135.8, C | 135.4, C | 133.2, C | 140.2, C |  |
| 3 | 136.6, CH | 139.3, CH | 137.0, CH | 78.4, CH | 74.1, CH | 180.7, C | 41.0, $\mathrm{CH}_{2}$ | 140.8, CH | 141.6, CH | 126.7, CH | 124.3, CH | 68.5, CH | 73.3, CH | 65.5, CH | 67.4, CH | 81.5, CH |
| 4 | 148.0, C | 146.2, C | 149.8, C | 79.4, CH | 86.6, CH | 76.7, CH | 125.1, CH | 132.2, CH | 123.6, CH | 127.6, CH | 128.1, CH | 68.7, CH | 69.0, CH | 54.5, CH | 69.8, CH | 74.3, CH |
| 5 | 111.2, CH | 113.8, CH | $110.4, \mathrm{CH}$ | 126.0, CH | 132.9, CH | 55.3, CH | 134.7, CH | 134.2, CH | 128.8, CH | 132.6, CH | 130.8, CH | 68.3, CH | 75.6, CH | 54.3, CH | 77.9, CH | 108.1, CH |
| 6 | 123.8, CH | 124.9, CH | 35.8, $\mathrm{CH}_{2}$ | 135.0, CH | 131.8, CH | 31.7, $\mathrm{CH}_{2}$ | 131.3, CH | 128.3, CH | 130.8, CH | 43.1, $\mathrm{CH}_{2}$ | 43.2, $\mathrm{CH}_{2}$ | 67.9, CH | 73.4, CH | 62.7, CH | 75.5, CH | 155.4, C |
| 7 | 137.1, CH | 136.5, CH | 67.6, CH | 131.1, CH | 136.9, CH | 135.3, CH | 128.6, CH | 137.8, CH | 132.9, CH | 66.6, CH | 66.7, CH | 127.7, CH | 126.9, CH | 128.3, CH | 126.0, CH | 134.9, C |
| 8 | 63.3, $\mathrm{CH}_{2}$ | 19.0, $\mathrm{CH}_{3}$ | 23.4, $\mathrm{CH}_{3}$ | 121.7, $\mathrm{CH}_{2}$ | 118.2, $\mathrm{CH}_{2}$ | 117.4, $\mathrm{CH}_{2}$ | 18.1, $\mathrm{CH}_{3}$ | 117.4, $\mathrm{CH}_{2}$ | 118.5, $\mathrm{CH}_{2}$ | 23.6, $\mathrm{CH}_{3}$ | 23.6, $\mathrm{CH}_{3}$ | 128.0, CH | 129.6, CH | 127.0, CH | 130.2, CH | 155.6, C |
| 9 | 27.4, $\mathrm{CH}_{2}$ | 35.0, $\mathrm{CH}_{2}$ | 27.2, $\mathrm{CH}_{2}$ | 30.4, $\mathrm{CH}_{2}$ | 125.7, CH | 32.9, $\mathrm{CH}_{2}$ | 133.3, CH | 40.0, $\mathrm{CH}_{2}$ | 40.0, $\mathrm{CH}_{2}$ | 125.8, CH | 30.4, $\mathrm{CH}_{2}$ | 19.3, $\mathrm{CH}_{3}$ | 19.3, $\mathrm{CH}_{3}$ | 19.2, $\mathrm{CH}_{3}$ | 19.2, $\mathrm{CH}_{3}$ | 39.1, $\mathrm{CH}_{2}$ |
| 10 | 21.0, $\mathrm{CH}_{2}$ | 66.3, CH | 21.0, $\mathrm{CH}_{2}$ | 19.9, $\mathrm{CH}_{2}$ | 21.9, $\mathrm{CH}_{2}$ | 20.2, $\mathrm{CH}_{2}$ | 135.2, CH | 16.8, $\mathrm{CH}_{2}$ | 16.8, $\mathrm{CH}_{2}$ | 125.7, CH | 22.1, $\mathrm{CH}_{2}$ | 58.1, $\mathrm{CH}_{2}$ | 57.5, $\mathrm{CH}_{2}$ | 56.8, $\mathrm{CH}_{2}$ | 58.2, $\mathrm{CH}_{2}$ | 78.7, CH |
| 11 | 13.7, $\mathrm{CH}_{3}$ | 23.3, $\mathrm{CH}_{3}$ | 13.8, $\mathrm{CH}_{3}$ | 13.9, $\mathrm{CH}_{3}$ | 14.6, $\mathrm{CH}_{3}$ | 14.0, $\mathrm{CH}_{3}$ | $17.9, \mathrm{CH}_{3}$ | 15.2, $\mathrm{CH}_{3}$ | 15.2, $\mathrm{CH}_{3}$ | 19.3, $\mathrm{CH}_{3}$ | 14.6, $\mathrm{CH}_{3}$ |  |  |  | 154.8, C | 38.1, $\mathrm{CH}_{2}$ |

${ }^{\mathrm{a}}$ recorded in DMSO- $d_{6}$. ${ }^{\mathrm{b}}$ recorded in $\mathrm{CDCl}_{3}$. ${ }^{\mathrm{c}}$ measured on a JEOL JNM-ECP 600 spectrometer and $\delta_{\mathrm{C}}$ of C-4a, C-8a, C-12, C-13 and 7-OMe for $\mathbf{1 8}$ were $131.4(\mathrm{C})$, $101.3 \mathrm{C}), 19.2\left(\mathrm{CH}_{2}\right), 14.1\left(\mathrm{CH}_{3}\right)$ and $60.9\left(\mathrm{CH}_{3}\right)$, respectively. ${ }^{\mathrm{d}}$ measured on an Agilent 500 MHz DD 2 spectrometer.

Table 2. ${ }^{1} \mathrm{H}$ NMR Data for $\mathbf{1}-18\left(\delta_{\mathrm{H}} \mathrm{ppm}, J\right.$ in Hz$)$

| position | $1^{\text {bc }}$ | $2^{\text {bid }}$ | 3 and $4{ }^{\text {bc }}$ | $5{ }^{\text {b] }}$ | $6^{\text {ad }}$ | 7 and $8^{\text {bc }}$ | $9^{\text {ac }}$ | $10^{\text {ac }}$ | $11^{\text {ac }}$ | $12^{\text {ac }}$ | $13^{\text {ac }}$ | $14^{\text {ac }}$ | $15^{\text {ad }}$ | $16^{\text {ac }}$ | $17^{\text {ac }}$ | $18^{\text {bc }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  |  |  |  | $\begin{aligned} & 4.27, \mathrm{~d}(12.5) \\ & 4.18, \mathrm{~d}(12.5) \end{aligned}$ |  | $\begin{aligned} & 3.45, \mathrm{~d}(11.7) \\ & 3.43, \mathrm{~d}(11.7) \end{aligned}$ | $\begin{aligned} & 3.25, \mathrm{~d}(12.0) ; \\ & 3.21, \mathrm{~d}(12.0) \end{aligned}$ | $\begin{aligned} & \hline 3.26, \mathrm{~d}(12.5) ; \\ & 3.22, \mathrm{~d}(12.5) \end{aligned}$ | 4.07, s | 3.87, s |  |  |  |  |  |
| 2 |  |  |  | 2.62, m |  | 5.87, s |  |  |  |  |  |  |  |  |  | 0 |
| 3 | 6.96, s | 7.15, s | 6.99, s | $\begin{aligned} & 3.97 \mathrm{dd} \\ & (7.8,8.9) \end{aligned}$ | $\begin{aligned} & 4.22, \mathrm{dd} \\ & (5.5,5.4) \end{aligned}$ |  | $\begin{aligned} & 2.31, \mathrm{~m} ; \\ & 2.32, \mathrm{~m} \end{aligned}$ | 5.79, d (14.8) | 5.82, d (15.0) | 6.03, d (11.6) | 5.97, d (11.0) | $\begin{aligned} & 4.17, \mathrm{dd}(5.3, \\ & 4.9) \end{aligned}$ | $\begin{aligned} & 4.24, \mathrm{dd} \\ & (10.6,7.5) \end{aligned}$ | $\begin{gathered} 4.52, \mathrm{dd} \\ (5.1,3.2) \end{gathered}$ | $\begin{aligned} & 4.24, \mathrm{dd} \\ & (4.8,3.6) \end{aligned}$ | 5.02, m |
| 4 |  |  |  | $\begin{aligned} & (.98, \mathrm{dd} \\ & (7.8,7.8) \end{aligned}$ | $\begin{aligned} & 4.09, \mathrm{dd} \\ & (5.4,5.4) \end{aligned}$ | 4.45, d (3.0) | $\begin{aligned} & 5.48, \mathrm{dt} \\ & (14.5,7.4) \end{aligned}$ | $\begin{aligned} & 6.23, \mathrm{dd} \\ & (14.8,10.7) \end{aligned}$ | $\begin{aligned} & 6.68, \mathrm{dd} \\ & (15.0,12.4) \end{aligned}$ | $\begin{aligned} & 6.51, \mathrm{dd} \\ & (15.0,11.2) \end{aligned}$ | $\begin{aligned} & 6.26, \mathrm{dd} \\ & (15.4,11.0) \end{aligned}$ | 3.60 , m | $\begin{aligned} & 3.66, \mathrm{dd} \\ & (10.6,10.6) \end{aligned}$ | $\begin{aligned} & 3.35, \mathrm{dd} \\ & (3.2,4.0) \end{aligned}$ | 3.55 , m | 4.49, d (3.0) |
| 5 | $\begin{aligned} & 5.67, \mathrm{~d} \\ & (11.6) \end{aligned}$ | $\begin{aligned} & 5.71, \mathrm{~d} \\ & (11.3) \end{aligned}$ | 5.26, t (8.2) | $\begin{aligned} & 5.41, \mathrm{dd} \\ & (10.2,7.8) \end{aligned}$ | $\begin{aligned} & 5.71, \mathrm{dd} \\ & (15.7,6.7) \end{aligned}$ | 2.37 , m | $\begin{aligned} & 6.08, \mathrm{dd} \\ & (14.5,11.0) \end{aligned}$ | $\begin{aligned} & 6.28, \mathrm{dd} \\ & (14.8,10.7) \end{aligned}$ | $\begin{aligned} & 6.03, \mathrm{dd} \\ & (11.3,12.4) \end{aligned}$ | 5.70, m | $\begin{aligned} & 5.62, \mathrm{dt} \\ & (15.4,7.5) \end{aligned}$ | 3.55 , m | 3.34 , m | $\begin{aligned} & 3.28, \mathrm{dd} \\ & (4.0,1.9) \end{aligned}$ | $\begin{aligned} & 4.81, \mathrm{dd} \\ & (9.6,7.6) \end{aligned}$ | 6.60 s $\quad$ - |
| 6 | $\begin{aligned} & 6.74, \mathrm{dd} \\ & (16.3,11.6) \end{aligned}$ | $\begin{aligned} & 6.52, \mathrm{dd} \\ & (15.6,11.3) \end{aligned}$ | $\begin{aligned} & 2.55, \mathrm{~m} ; \\ & 2.49, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 6.31, \mathrm{dd} \\ & (10.2,11.3) \end{aligned}$ | $\begin{aligned} & 6.21, \mathrm{dd} \\ & (15.7,10.2) \end{aligned}$ | $\begin{aligned} & 2.56, \operatorname{ddd}(14.6,6.0,6.2) \\ & \text { 2.21, ddd (14.6, 7.9, } 7.4) \end{aligned}$ | $\begin{aligned} & 6.02, \mathrm{dd} \\ & (14.4,10.2) \end{aligned}$ | $\begin{aligned} & 6.19, \mathrm{dd} \\ & (14.8,10.2) \end{aligned}$ | $\begin{aligned} & 5.93, \mathrm{dd} \\ & (11.2,11.3) \end{aligned}$ | $\begin{aligned} & 2.24, \mathrm{~m} ; \\ & 2.14, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.19, \mathrm{~m} ; \\ & 2.10, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 4.23, \mathrm{dd} \\ & (4.0,4.0) \end{aligned}$ | $\begin{aligned} & 3.99, \mathrm{dd}(6.5, \\ & 6.5) \end{aligned}$ | $\begin{aligned} & 4.45, \mathrm{dd} \\ & (7.7,1.9) \end{aligned}$ | $5.60, \mathrm{~d}$ (7.6) |  |
| 7 | $\begin{aligned} & 6.04, \mathrm{dt} \\ & (16.3,4.9) \end{aligned}$ | $\begin{aligned} & 5.99, \mathrm{dq} \\ & (15.6,6.8) \end{aligned}$ | 3.96 , m | $\begin{aligned} & 6.68, \text { ddd } \\ & (16.6,11.3, \\ & 10.3) \end{aligned}$ | $\begin{aligned} & 6.32 \text {, ddd } \\ & (17.0,10.2, \\ & 10.2) \end{aligned}$ | 5.75, m | $\begin{aligned} & 5.62, \mathrm{dq} \\ & (14.4,6.9) \end{aligned}$ | $\begin{aligned} & \text { 6.36, ddd, } \\ & \text { (17.1, 10.2, } \\ & 10.2) \end{aligned}$ | $\begin{aligned} & 6.82 \text {, ddd } \\ & (16.7,11.2, \\ & 10.4) \end{aligned}$ | 3.65 , m | 3.62 , m | $\begin{aligned} & 6.34, \mathrm{dd} \\ & (15.7,1.6) \end{aligned}$ | $\begin{aligned} & 6.15, \mathrm{~d} \\ & (16.2) \end{aligned}$ | $\begin{aligned} & 6.36, \mathrm{~d} \\ & (15.8) \end{aligned}$ | $\begin{aligned} & 6.42, \text { br.d } \\ & (16.3) \end{aligned}$ | 10 |
| 8 | $4.24, \mathrm{~d}(4.9)$ | 1.84, d (7.1) | 1.23, d (6.1) | $\begin{aligned} & 5.37, \text { br.d } \\ & \text { (16.6); } 5.30, \\ & \text { br.d (10.3) } \end{aligned}$ | $\begin{aligned} & 5.21 \text { br.d } \\ & \text { (17.0); } 5.07 \\ & \text { br.d (10.2) } \end{aligned}$ | $\begin{aligned} & 5.11, \text { br.d } \\ & \text { (17.0); } 5.03, \\ & \text { br.d (10.5) } \end{aligned}$ | 1.73, d (6.9) | $\begin{aligned} & 5.20, \mathrm{~d}(17.1) ; \\ & 5.06, \mathrm{~d}(10.2) \end{aligned}$ | $\begin{aligned} & 5.24, \mathrm{~d}(16.7) ; \\ & 5.15,(10.4) \end{aligned}$ | 1.03, d (6.1) | 1.02, d (6.2) | $\begin{aligned} & 5.94, \mathrm{dq} \\ & (15.7,6.7) \end{aligned}$ | 5.82, m | $\begin{aligned} & 5.91, \mathrm{dq} \\ & (15.8,6.7) \end{aligned}$ | $\begin{aligned} & 5.93, \mathrm{dq} \\ & (16.3,6.8) \end{aligned}$ | 0 |
| 9 | 2.31, t (7.5) | $\begin{aligned} & 2.53, \mathrm{dd} \\ & (15.0,7.9) ; \\ & 2.46, \mathrm{dd} \\ & (15.0,3.9) \end{aligned}$ | 2.32, t (7.6) | $\begin{aligned} & 1.84, \mathrm{~m} ; \\ & 1.58, \mathrm{~m} \end{aligned}$ | 5.36, t (7.3) | $\begin{aligned} & 2.49, \mathrm{~m} ; \\ & 2.36, \mathrm{~m} \end{aligned}$ | 5.44, d (15.5) | $\begin{aligned} & \text { 1.41, m; } \\ & \text { 1.42, m } \end{aligned}$ | $\begin{aligned} & 1.43, \mathrm{~m} ; \\ & 1.41, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 6.49, \text { dd (16.6, } \\ & 1.5) \end{aligned}$ | 2.07, t (7.3) | $\begin{aligned} & 1.77, \mathrm{dd} \\ & (1.6,6.7) \end{aligned}$ | 1.73,d (5.5) | $\begin{aligned} & 1.76, \mathrm{~d} \\ & (6.7) \end{aligned}$ | $\begin{aligned} & 1.80, \mathrm{dd} \\ & (6.6,1.5) \end{aligned}$ | $\begin{aligned} & 2.56, \text { ddd } \\ & \text { (14.6, 8.7, } \\ & \text { 6.4); 2.13, } \\ & \text { ddd (14.6, } \\ & 6.0,1.4) \end{aligned}$ |
| 10 | 1.57, m | 4.04, m | 1.59, m | $\begin{aligned} & \text { 1.57, m; } \\ & 1.51, \mathrm{~m} \end{aligned}$ | 2.12, m | $\begin{aligned} & 1.56, \mathrm{~m} ; \\ & 1.62, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 5.73, \mathrm{dq} \\ & (15.5,6.6) \end{aligned}$ | $\begin{aligned} & 1.30, \mathrm{~m} ; \\ & 1.18, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.30, \mathrm{~m} ; \\ & 1.18, \mathrm{~m} \end{aligned}$ | 5.72, m | 1.37, m | $\begin{aligned} & 4.18, \mathrm{dd} \\ & (12.0,5.8) ; \\ & 3.92, \mathrm{dd} \\ & (12.0,5.8) \end{aligned}$ | 4.11, s | $\begin{aligned} & 4.16, \mathrm{dd} \\ & (6.0,11.7) ; \\ & 4.13, \mathrm{dd} \\ & (6.0,11.7) \end{aligned}$ | $\begin{aligned} & 4.31, \mathrm{dd} \\ & (12.4,4.7) ; \\ & 3.95, \mathrm{dd} \\ & (12.4,4.7) \end{aligned}$ | $4.10, \mathrm{~m}$ |
| 11 | 0.91, t (7.6) | 1.21, d (6.4) | 0.95, t (7.8) | 0.96, t (7.3) | 0.91, t (7.7) | 0.95, t (7.4) | 1.72, d (6.6) | 0.83, t (7.3) | 0.83, t (6.7) | $\begin{aligned} & 1.76, \mathrm{dd} \\ & (6.8,1.5) \end{aligned}$ | 0.86, t (7.6) |  |  |  |  | $\begin{aligned} & 1.68, \mathrm{~m} ; \\ & 1.55, \mathrm{~m} \end{aligned}$ |
| $3-\mathrm{OH}$ |  |  |  |  | 5.27, d (5.5) |  |  |  |  |  |  | 4.51, d (5.3) | $5.39, \mathrm{~d}$ (7.5) | 5.13, d (5.1) | 5.14, d (4.8) | , |
| $4-\mathrm{OH}$ |  |  |  |  |  |  |  |  |  |  |  | 4.27, d (5.9) |  |  | 5.47, d (5.9) |  |
| $5-\mathrm{OH}$ |  |  |  |  |  |  |  |  |  |  |  | 4.41, d (5.4) | $5.41, \mathrm{~d}(5.6)$ |  |  |  |
| $6-\mathrm{OH}$ |  |  |  |  |  |  |  |  |  |  |  | 4.44, d (4.0) | $5.09, \mathrm{~d}$ (6.5) | 5.16, d (7.7) |  |  |
| $10-\mathrm{OH}$ |  |  |  |  |  |  |  |  |  |  |  | 4.64, t (5.8) | 4.37, t (5.4) | $4.35, \mathrm{t}$ (6.) | 4.91, t (4.9) | (1) |

 $(\mathrm{m}), 0.90(\mathrm{t}, J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 11.5(\mathrm{~s})$ and $3.97(\mathrm{~s}, 3 \mathrm{H})$, respectively. ${ }^{\mathrm{d}}$ measured on an Agilent 500 MHz DD2 spectrometer.

Table 3. $\alpha$-Glucosidase inhibitions of compounds $\mathbf{1 , 2 , 5 , 1 0}, \mathbf{1 1}, 14-18$, and 21-23.

| Compound | Acarbose | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{5}$ | $\mathbf{1 0}$ | $\mathbf{1 1}$ | $\mathbf{1 4}$ | $\mathbf{1 5}$ | $\mathbf{1 6}$ | $\mathbf{1 7}$ | $\mathbf{1 8}$ | $\mathbf{2 1}$ | $\mathbf{2 2}$ | $\mathbf{2 3}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{IC}_{50}(\mathrm{mM})$ | 0.95 | 2.36 | 1.65 | 1.30 | 2.37 | 2.70 | 1.36 | 1.54 | 2.21 | 2.26 | 0.027 | 1.65 | 1.19 | 1.74 |

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