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# Exceptionally low thermal conductivity of poly(3-hexylthiophene) single nanowires

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## ABSTRACT

Electrical and thermal conductivity measurements were conducted by using microfabricated four-point probe and T-type nanosensors in a vacuum on poly(3-hexylthiophene) (P3HT) single nanowires prepared by the whisker method. The nanowires had a cross-section of only 5 nm x 15 nm and a length of 0.5 - 10  $\mu\text{m}$ . The electrical conductance of nanowires was  $3.3 \times 10^{-7} \text{ } 1/\Omega$  and remained stable even after exposing the wires to ambient conditions for a few days. The thermal conductivity of a nanowire was  $0.055 \pm 0.003 \text{ W/m}\cdot\text{K}$  which was approximately an order of magnitude lower than other previously reported P3HT single nanowires with a diameter of 120 nm and embedded in anodic aluminum oxide matrix. This thermal conductivity was also approximately four times smaller than that of P3HT amorphous thin films ( $\sim 0.2 \text{ W/m}\cdot\text{K}$ ). The exceptionally low thermal conductivity was attributed to increased interface-phonon scattering because heat propagation was confined to an extreme narrow spatial dimension.

**KEYWORDS:** thermal conductivity, nanowire, conducting polymer, electrical conductivity, whisker method

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Organic polymers are attracting considerable attention as plausible thermoelectric materials, because they are semiconductors that exhibit relatively low thermal conductivity. Compared with traditional inorganic thermoelectric materials, organic electronic materials have the noteworthy advantages of being mechanically flexible, low-cost, and solution processable. Given that the thermal conductivities of disorder polymer chains are often an order of magnitude lower than those of silica glass, these polymer chains have been suggested to have thermoelectric applications. A number of conjugated polymers have been studied, including poly(3,4-ethylenedioxythiophene)poly(styrenesulfone) (PEDOT:PSS), [6,6]-phenyl C61-butyric acid methyl ester (PCBM), and poly(3-hexylthiophene) (P3HT) [1-6]. The effective thermal conductivity of solution-derived P3HT thin films with different thickness on substrates has been investigated by applying  $3\omega$  technique [6]. Thermal conductivity was  $0.189 \pm 0.023$  W/m•K for a P3HT layer on glass substrate. Other investigations on highly doped P3HT have reported an average thermal conductivity of 0.48 W/m•k [7]. Optical measurement of thermal conductivity was also performed using fiber aligned frequency domain thermoreflectance for a 600 nm parylene film and a 1600 nm P3HT film [8]. Thermal conductivities of PCBM thin films have further been shown to be independent of temperature above 180 K and to be  $< 0.03 \pm 0.003$  W/m•K at room temperature [4]. Moreover, thermal conductivities varying from  $0.031 \pm 0.005$  W/m•K to  $0.227 \pm 0.014$  W/m•K near room temperature have been measured by time domain

thermoreflectance for PEDOT:PSS, PCBM, P3HT, and P3HT:PCBM blend thin films. The thermal conductivities of conducting polymers are found to be intrinsically lower than those of inorganic materials, which typically range within of 0.1 - 1 W/m•K. Given the advances in organic optoelectronics over the past few years, large electrical conductivity values and Seebeck coefficients can now be found in some conducting polymers. Large ZT values can be expected when combined with low thermal conductivity.

Among such conducting polymers, P3HT is one of the most intensively studied organic semiconductors because of its unique thermal and electrical properties. It possesses excellent solution processability, chemical stability, and high carrier mobility [9-11]. The bandgap energy of P3HT is approximately 1.9 eV, which is in the same range as typical semiconductors. By converting P3HT into nanowires (NWs), a decrease in thermal conductivity of NWs and an improvement in their ZT values are expected because of the reduced dimensionality [5,12,13]. A numerical simulation based on the Holstein small-polaron model has been applied to one-dimensional (1D) P3HT and PEDOT:PSS molecular chains to study thermoelectric properties of molecular NWs [12]. These 1D NWs are excellent thermoelectric materials because narrow and sharp density of states could increase Seebeck coefficient, and thermal conductivity along the molecule stacking direction could be very small because of enhanced interface-phonon scattering [12]. More recently, individual P3HT NWs of three different diameters embedded in a matrix have been

investigated by  $3\omega$  scanning thermal microscopy. Thermal conductivity decreases when NWs diameters are reduced [5]. A thermal conductivity of  $0.5 \pm 0.24$  W/m•K for intrinsic P3HT NWs 120 nm in diameter has been determined by effective medium theory.

The present work aimed to investigate electrical and thermal characteristics of individual P3HT NWs with a narrow cross-section of 5 nm x 15 nm in a vacuum using a four-point contact probe and T-type nanosensors. The thermal conductivities of the NWs were determined by a 1D heat-conduction method. Results showed that P3HT NW thermal conductivity was exceptionally lower than those for previously reported thin films and NWs with larger diameters, which can be explained by the increase in phonon boundary scattering caused by reduced dimensionality.

P3HT NWs were synthesized by the whisker method reported in Ref. [14-17], in which cyclohexanone was used as a marginal solvent to prepare P3HT NWs. This preparation step in whisker method was similar to recrystallization in organic reaction. Given the low solubility of polymers, they were only slightly soluble or even insoluble in marginal solvent at room temperature. After heating the polymer solution, solvent solubility increased and the polymer can be completely dissolved. The polymer solution was then slowly cooled to room temperature. Solubility decreased again upon cooling, and the polymer molecules self-assembled to form NWs.

P3HT (regioregularity: 91–94%) with a weight-average molecular weight ( $M_w$ ) of

50–70 kg mol<sup>-1</sup> and a polydispersity (PDI) of  $\approx 2.0$  was used. To fabricate P3HT NWs, 1 wt% P3HT solution in *p*-xylene was heated to 80 °C for 30 min to completely dissolve P3HT. Subsequently, P3HT solution was cooled to room temperature. Then, P3HT solution was stored at room temperature for 48 hours and P3HT NWs were formed. The NW-containing solution was purified by centrifugation for four times (5000 rpm, 30 min). After each centrifugation, the solution containing amorphous P3HT was removed, and pure *p*-xylene was added. More details on the preparation and synthesis of P3HT NWs are found in Ref. [18]. The NWs were 0.5 - 10  $\mu$ m long, 5 nm high, and 15 nm wide, as confirmed by AFM, SEM, and TEM images (Figure 1). NWs with an average diameter up to 32 nm can be fabricated by varying experimental conditions (for example, different temperatures [17] or solvents). Later we found that the thermal conductivity only increased marginally ( $\sim 10\%$ ) even for a NW with a diameter of  $\sim 32$  nm. More details for the synthesis of NWs with different diameters are presented in the ESI section.

A selected area electron diffraction (SAED) pattern of the NWs is displayed in the inset of Figure 1(c). Although the contrast of the SAED pattern is not very high, however, reflections from (1 0 0), (2 0 0), and (3 0 0) can still be clearly observed. It suggests that the alkyl side chains of the P3HT NWs are arranged as lamella morphology. The appearance of the peaks from (100), (200), and (300) in the XRD measurements (Figure S1)

further confirms the conformation of the alkyl side chains of the P3HT NWs. Previous XRD results of P3HT NWs also showed improved crystallinity and orderings of alkyl side chain packing after thermal annealing [18]. Schematic of the idea spatial orientation of the P3HT crystallite from the side view of the b axis of the crystal cell ( $\pi$ - $\pi$  stacking direction) parallel to the NW long axis is illustrated in Figure S2. Although the images of P3HT NWs shown in Figure 1 display fibrillar network structures, single P3HT NWs can be properly separated by spreading diluted nanowires-containing solutions on silicon wafers (Figure S3).

A four-point probe on Si templates was then designed and fabricated to measure the conductivity of P3HT single NW. Templates used in experiments were commercially available four-inch silicon wafers with (001) crystal orientation and n-type background doping. After dicing the Si wafer into 2 cm x 2 cm chips, the surface of Si substrate was passivated in advance using a thermally grown 2000 Å-thick SiO<sub>2</sub> layer to avoid leakage current through the substrate during current–voltage (I-V) measurements. As shown in Figure 2(a), a pattern of cross-finger-type Au/Ti wires with a linewidth of 1 μm, a pitch of 1 μm, and a length of 6 μm were defined on Si chip using e-beam lithography within an area of 1 mm<sup>2</sup>. A drop of diluted P3HT NW in p-xylene solution was placed within electrode patterns. By applying a bias across contact pads, dielectrophoresis force [19-22] drove NW to bridge the electrode gap. After a single NW was selected, a focus ion beam (FIB) was

used to selectively deposit platinum (Pt) to make contacts with finger electrodes.

Suspended Au/Ti nanofilm T-type sensors were fabricated on silicon templates following procedures reported in Ref. [23,24]. Dielectrophoresis technique was also applied to position NWs across the electrode gap. After one end of a single P3HT NW was safely attached onto the center of the nanosensor and the other to the heat sink, the devices were placed in a dry box for a few days and then cleaned with acetone to remove p-xylene residue and NWs. Figure 2(b) and the inset show the SEM image of a T-type nanosensor without NW attached. The schematic, dimensions, and cross-section view of the prepared T-type sensor with a single NW attached are displayed in Figure 2(c) and 2(d). The nanosensor acted simultaneously as a heater and a thermometer to measure the average nanosensor temperature such that the local temperature of the junction and the heat input to P3HT NW can be calculated by solving 1-D steady-state heat conduction model [23].

For all samples tested, four-point I-V data were linear and symmetrical about zero bias. Electrical stability of P3HT NWs was also verified over an extended period. Table I depicts the resistance of a single NW over 6 days in air determined by four-point probe measurements. NWs had an average resistance of  $\sim 3.33 \times 10^7 \Omega$ , which was approximately two order of magnitude lower than P3HT films (with an average resistivity of  $\sim 1.14 \times 10^9 \Omega$ ). The NWs also showed atmospheric stability and low surface defect levels.

Before P3HT NWs were attached onto the nanosensor, resistance-temperature



coefficient  $\beta_{ns}$  ( $K^{-1}$ ) and thermal conductivity  $\lambda_{ns}$  ( $W/m\cdot K$ ) of nanosensor were first be determined. The resistance-temperature relation of nanosensor was measured with a four-point probe, as shown in Figure 2(b), by supplying a small current to nanosensor. The temperature increase of nanosensor in a vacuum caused by self-heating was less than 8.5 mK. Dimensions of nanosensor were determined with a scanning electron microscope and metal film thickness was measured with a Dektak surface profiler. The measured resistance-temperature coefficients of nanosensor are shown in Figure 3. The measured  $\beta_{ns}$  was significantly lower than bulk values ( $\sim 0.039 K^{-1}$ ) of typical metals at 300 K [23,24]. The increase in temperature of nanosensor (volumetric average temperature rise  $\Delta T_v$ ) when subjected to different currents can be determined from the measurement of resistance using the expression  $\Delta T_v = \frac{R-R_0}{\beta_{ns}R_0}$ , where  $R_0$  is the electrical resistance of nanofilm at 273 K and  $\beta_{ns}$  is the resistance-temperature coefficient of the nanosensor obtained from calibration. The volumetric average temperature rise of nanosensor without P3HT NWs as a function of heating rate ( $q=IV$ ) is shown in Figure 4. The thermal conductivity  $\lambda_{ns}$  ( $W/m\cdot K$ ) of nanosensor can be calculated using the expression  $\lambda_{ns} = \frac{l_{ns}^2 q_v}{12\Delta T_v}$  by extracting the slope from a linear least-square fit of experimental data, as shown in Figure 4.  $q_v$  is the volumetric heat generation rate given by  $q_v = \frac{IV}{wtl_{ns}}$ , where  $w$ ,  $t$ , and  $l_{ns}$  are the width, thickness, and length of nanosensor, respectively,  $I$  and  $V$  are heating current and voltage.

The volumetric average temperature rise of nanosensor as a function of heating rate

with a P3HT NW is depicted in Figure 4. The average temperature rise with a P3HT NW was lower than that without NW. Assuming that thermal contact resistance was negligible [25], the thermal conductivity  $\lambda_{NW}$  of a single P3HT NW at room temperature can be calculated from the equation given in Ref. [22]:

$$\lambda_{NW} = \frac{l_f l_{ns} \lambda_{ns} A_{ns} (l_{ns}^3 q_v - 12 l_{ns} \lambda_{ns} \Delta T_v)}{l_{ns1} l_{ns2} A_f [12 l_{ns} \lambda_{ns} \Delta T_v - q_v (l_{ns1}^3 + l_{ns2}^3)]},$$

where  $l_f$  is the length of P3HT NW between the two connecting points at nanosensor and heat sink,  $A_f$  and  $A_{ns}$  are the cross-sections of the P3HT NW and nanosensor, respectively, and  $l_{ns1}$  and  $l_{ns2}$  are the lengths of the left- and right-hand sides of nanosensor from the junction point. The thermal conductivity of a single P3HT NW with a value of 0.055 W/m•K at room temperature can be obtained by plugging in the corresponding parameters into the equation above. Since the electrical resistance of the nanosensor was determined by a four-point probe technique, the maximum uncertainty of the electrical resistance values is less than 0.005%. The uncertainty in the thermal conductivity arises from measurement errors in voltage, current, temperature, and dimensions of the nanosensor. The width and length of the nanosensor are measured with a scanning electron microscope. The metal film thickness of nanosensor is measured with a Dektak surface profiler (with a resolution of 0.01 nm). The errors produced by the dimension measurements are estimated to be less than 3%. The total error in the thermal conductivity value is estimated to be within 5.5%. Therefore, the thermal conductivity of the P3HT NWs that we measured is  $0.055 \pm 0.003$  W/m•K.

The thermal conductivity obtained from this measurement was compared with previously reported ones for NWs and P3HT thin films. Thermal conductivity measurements were conducted by Muñoz Rojo et al. on P3HT NWs using  $3\omega$  scanning thermal microscopy technique operated in contact mode [5]. The intrinsic thermal conductivity of individual semicrystalline P3HT NW embedded in the anodic aluminum oxide matrix was deduced by solving an equation derived from effective medium theory [26,27]. They observed a reduction of thermal conductivity from  $\sim 2.29 \text{ W/m}\cdot\text{K}$  to  $\sim 0.5 \text{ W/m}\cdot\text{K}$  when the diameter of NWs was reduced from 350 nm to 120 nm. The exceptionally high thermal conductivities in this experiment were attributed to well-aligned polymeric chains in contrast to disordered ones. However, our results showed nearly an order of magnitude lower thermal conductivity when the cross-section of NWs was reduced to 5 nm x 15 nm. Thermal conductivity decreased with decreased NW diameter, indicating that phonon transport through P3HT NW was suppressed because of the strong phonon boundary scattering and/or possible changes in phonon dispersion due to enhanced confinement in 1D structure. Duda and co-workers [3] reported on the thermal conductivity of P3HT thin films as measured by time domain thermoreflectance. They found a thermal conductivity of  $\sim 0.185 \text{ W/m}\cdot\text{K}$  near room temperature, which did not vary with changes in film thickness from 77 nm to 200 nm. Their reported thermal conductivity of P3HT thin film well agreed with previous measurements [6,8]. However, the result was still

approximately four times higher than our semicrystalline P3HT single NW. A parameter free computational model for characterizing thermoelectric figure of merit (ZT value) of single organic molecular chains was developed by Wang et al. [13] using nonequilibrium MD method, Boltzmann transport theory, and deformation potential theory. They estimated ZT values of 1.6 and 1.4 for single P3HT chains with p-doping level of  $4 \times 10^5 \text{ cm}^{-1}$  and n-doping level of  $4 \times 10^5 \text{ cm}^{-1}$ , respectively, assuming a lattice thermal conductivity of  $0.2 \text{ W/m}\cdot\text{K}$ . For the NWs with an exceptionally low thermal conductivity ( $0.055 \text{ W/m}\cdot\text{K}$ ) in this study, one should expect figure of merit larger than 5 with proper doping levels to enhance electrical transport and at the same time to suppress phonon transport with low-dimensional structures.

In summary, we presented electrical and thermal conductivity measurements of P3HT single NW with a cross-section of  $5 \text{ nm} \times 15 \text{ nm}$  using microfabricated suspended nanosensor and four-point probe technique. The electrical conductivity of the NWs remained relatively stable over an extended period. The thermal conductivity observed was exceptionally lower than larger-diameter P3HT NWs. Thermal conductivity was also four times smaller than those of amorphous thin films, suggesting that phonon boundary scattering had an appreciable effect on thermal transport in P3HT NWs when 1D confinement was strong. Our research can stimulate further theoretical investigations on the intriguing effect of size on thermal transport in P3HT NWs with very narrow diameters.

## ACKNOWLEDGMENT

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## REFERENCES

- (1) D.E. Motaung, G.F. Malgas, C.J. Arendse, S.E. Mavundla, D. Knoesen, *Mater. Chem. Phys.*, 2009, **116**, 279-283.
- (2) M. He, J. Ge, Z. Lin, X. Feng, X. Wang, H. Lu, Y. Yang, F. Qiu, *Energ. Environ. Sci.*, 2012, **5**, 8351-8358.
- (3) J.C. Duda, P.E. Hopkins, Y. Shen, M.C. Gupta, *Appl. Phys. Letts.*, 2013, **102**, 251912.
- (4) J.C. Duda, P.E. Hopkins, Y. Shen, M.C. Gupta, *Phys. Rev. Letts.*, 2013, **110**, 015902.
- (5) M. Muñoz Rojo, J. Martin, S. Grauby, T. Borca-Tasciuc, S. Dilhaire, M. Martin-Gonzalez, *Nanoscale*, 2014, **6**, 7858-7865.
- (6) S. Rausch, D. Rauh, C. Deibel, S. Vidi, H.P. Ebert, *Int. J. Thermophys.*, 2013, **34**, 820-830.
- (7) J. Sun, M.-L. Yeh, B.J. Jung, B. Zhang, J. Feser, A. Majumdar, H.E. Katz, *Macromolecules*, 2010, **43**, 2897.

- (8) J.A. Malen, K. Baheti, T. Tong, Y. Zhao, J.A. Hudgins, A. Majumdar, *J. Heat Trans. – T. ASME*, 2011, **133**, 081601.
- (9) M. He, W. Han, J. Ge, Y.L. Yang, F. Qiu, Z.Q. Lin, *Energ. Environ. Sci.*, 2011, **4**, 2894-2902.
- (10) M. He, W. Han, J. Ge, W.J. Yu, Y.L. Yang, F. Qiu, Z.Q. Lin, *Nanoscale*, 2011, **3**, 3159-3163.
- (11) M. He, L. Zhao, J. Wang, W. Han, Y.L. Yang, F. Qiu, Z.Q. Lin, *ACS Nano*, 2010, **4**, 3241-3247.
- (12) Y. Wang, J. Zhou, R. Yang, *J. Phys. Chem. C*, 2011, **115**, 24418-24428.
- (13) D. Wang, W. Shi, J. Chen, J. Xi, Z. Shuai, *Phys. Chem. Chem. Phys.*, 2012, **14**, 16505-16520.
- (14) K.J. Ihn, J. Moulton, P. Smith, *J. Polym. Sci. Pt. B-Polym. Phys.*, 1993, **31**, 735-742.
- (15) W.D. Oosterbaan, V. Vrindts, S. Berson, S. Guillerez, O. Douhéret, B. Ruttens, J. D’Haen, P. Adriaensens, J. Manca, L. Lutsen, D. Vanderzande, *J. Mater. Chem.*, 2009, **19**, 5424-5435.
- (16) S. Berson, R. De Bettignies, S. Bailly, S. Guillerez, *Adv. Funct. Mater.*, 2007, **17**, 1377-1384.
- (17) J. Liu, M. Arif, J. Zou, S. I. Khondaker, L. Zhai, *Macromolecules*, 2009, **42**, 9390-9393.

- (18) C.-M. Fu, K.-S. Jeng, Y.-H. Li, Y.C. Hsu, M.-H. Chi, W.-B. Jian, J.-T. Chen, *Macromol. Chem. Phys.*, 2015, **216**, 59-68.
- (19) K. Yamamoto, S. Akita, Y. Nakayama, *Jpn. J. Appl. Phys.*, 1996, **35**, L917-L918.
- (20) K. Yamamoto, S. Akita, Y. Nakayama, *J. Phys. D: Appl. Phys.*, 1998, **31**, L34-L36.
- (21) W.B. Choi, Y.W. Jin, H.Y. Kim, S.J. Lee, M.J. Yun, J.H. Kang, Y.S. Choi, N.S. Park, N.S. Lee, J.M. Kim, *Appl. Phys. Lett.*, 2001, **78**, 1547-1549.
- (22) J. Suehiro, G. Zhou, M. Hara, *J. Phys. D: Appl. Phys.*, 2003, **36**, L109-L114.
- (23) M. Fujii, X. Zhang, H. Xie, H. Ago, K. Takahashi, T. Ikuta, H. Abe, T. Shimizu, *Phys. Rev. Lett.*, 2005, **95**, 065502.
- (24) X. Zhang, H.Q. Xie, M. Fijii, H. Ago, K. Takahashi, K. Ikuta, H. Abe, T. Shimizu, *Int. J. Thermophys.*, 2007, **28**, 33-43.
- (25) P. Kim, L. Shi, A. Majumdar, and P.L. McEuen, *Phys. Rev. Lett.* **2001**, 87, 215502.
- (26) A.I. Persson, Y.K. Koh, D.G. Cahill, L. Samuelson, and H. Linke, *Nano Lett.*, 2009, **9**, 4484-4488.
- (27) M. Muñoz Rojo, S. Grauby, J.M. Rampnoux, O. Caballero-Calero, M.S. Martín-González, S. Dilhaire, *J. Appl. Phys.*, 2013, **113**, 054308.

**Figure Captions**

Figure 1 (a) AFM (b) SEM and (c) TEM images of P3HT NWs. Inset in (c) shows the selected area electron diffraction pattern of the P3HT NWs.

Figure 2 (a) Optical image of cross-finger type four-point probe for measuring electrical conductance of P3HT single NW (b) SEM image of a T-type nanosensor (c) and (d) schematic, dimensions, and cross-section view of fabricated T-type sensor with a single NW attached.

Figure 3 Resistance-temperature coefficient of nanosensor with thickness, width, and length of 60 nm, 547 nm, and 7.72  $\mu\text{m}$ , respectively.

Figure 4 Volumetric average temperature rise versus heating rate measured in a vacuum for the nanosensor with and without the presence of a P3HT single NW.

Table I Resistance of a P3HT single NW exposed in air over an extended period.



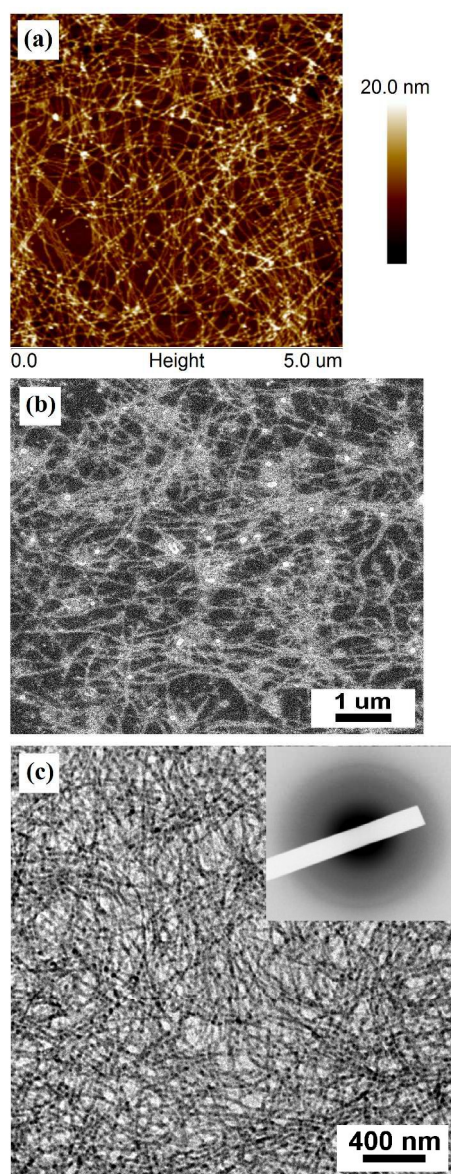


Figure 1 (a) AFM (b) SEM and (c) TEM images of P3HT NWs. Inset in (c) shows the selected area electron diffraction pattern of the P3HT NWs.

418x1014mm (96 x 96 DPI)

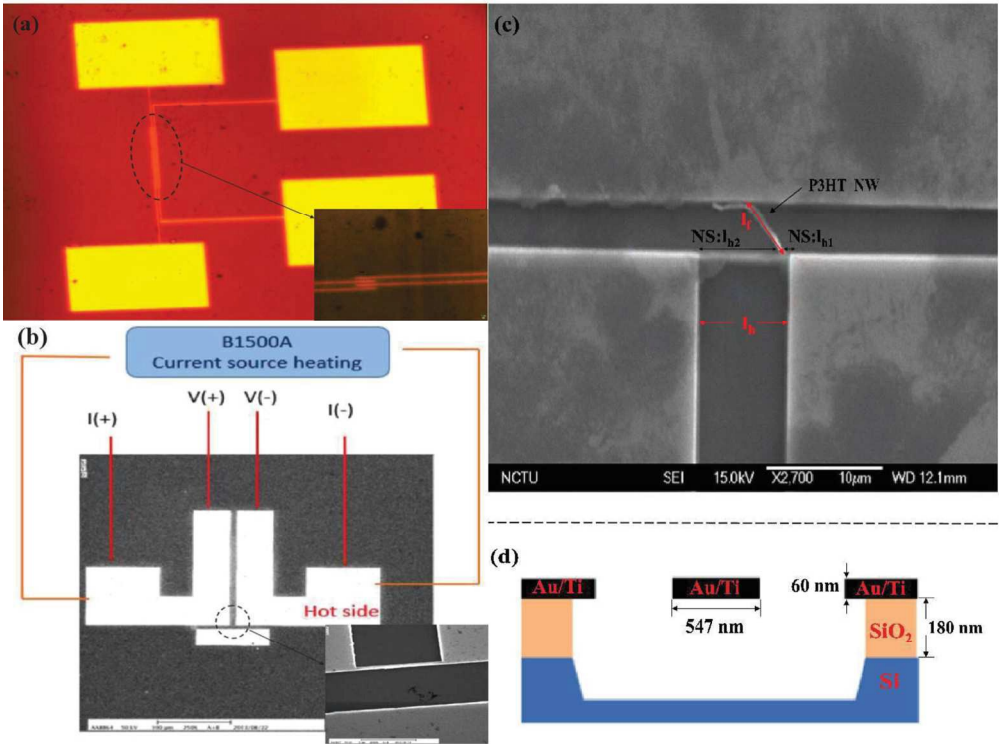


Figure 2 (a) Optical image of cross-finger type four-point probe for measuring electrical conductance of P3HT single NW (b) SEM image of a T-type nanosensor (c) and (d) schematic, dimensions, and cross-section view of fabricated T-type sensor with a single NW attached.  
254x190mm (150 x 150 DPI)

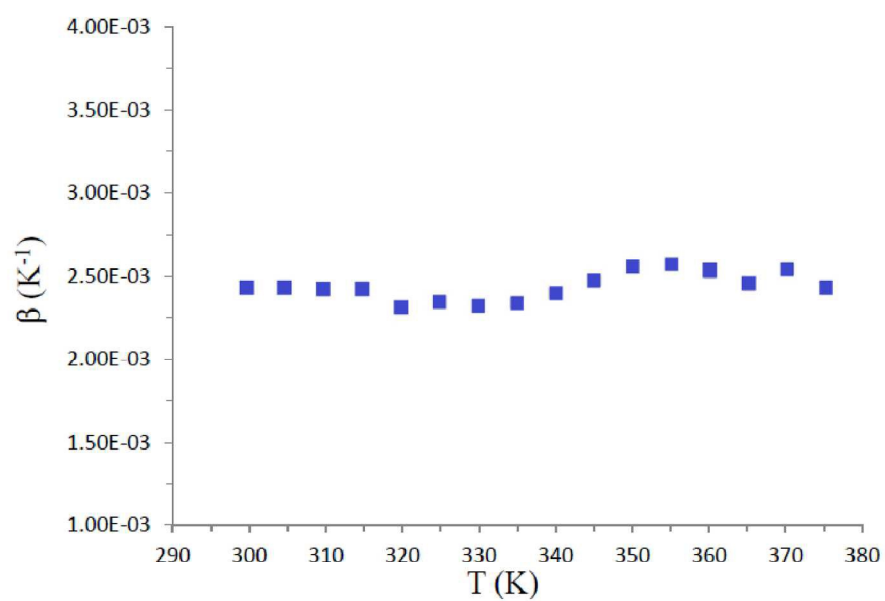


Figure 3 Resistance-temperature coefficient of nanosensor with thickness, width, and length of 60 nm, 547 nm, and 7.72  $\mu\text{m}$ , respectively.  
297x209mm (150 x 150 DPI)

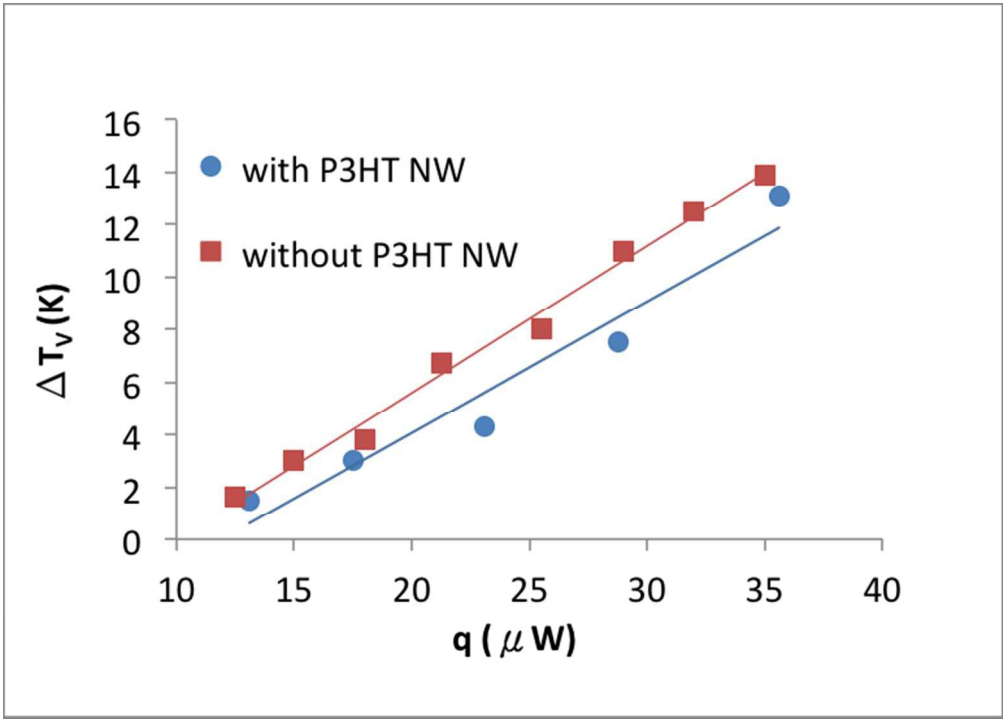


Figure 4 Volumetric average temperature rise versus heating rate measured in a vacuum for the nanosensor with and without the presence of a P3HT single NW.  
130x93mm (150 x 150 DPI)

Day	Day 1	Day 2	Day 3	Day 4	Day 5	Day 6
Resistance( $\Omega$ )	$3.33 \times 10^7$	$3.33 \times 10^7$	$3.32 \times 10^7$	$3.28 \times 10^7$	$3.42 \times 10^7$	$3.32 \times 10^7$

Table I Resistance of a P3HT single NWs exposed in air over an extended period.  
264x63mm (300 x 300 DPI)