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## Graphical Abstract

# Efficient catalyst for tandem solvent free enantioselective Knoevenagel-formal [3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder Reactions 

Sylvia Fernandes, ${ }^{\text {a }}$ P. Rajakannu ${ }^{\text {b }}$ and Sujata V. Bhat* ${ }^{\text {a }}$




In this study highly efficient catalyst has been observed for one-pot solvent free enantioselective Knoevenagel-[3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder reactions. Thus, the synthesis of bicyclic tetrahydro-2H-chromen-5(6H)-one and tricyclic octahydro- 2 H -benzo[c]-chromen-1(6H)-one derivatives with enantioselectivity up to ee $98.8 \%$ has been achieved in the presence of chiral LBA, titanium-isopropoxy-(S)-BINOLate under a solvent free condition. The stereochemistry of tricyclic product $\mathbf{1 0}$ has been further supported by single crystal X-ray analysis.

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# Efficient catalyst for tandem solvent free enantioselective Knoevenagel-formal [3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder Reactions 

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In this study highly efficient catalyst has been observed for tandem solvent free enantioselective Knoevenagel-formal [3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder reactions. Thus, the synthesis of bicyclic tetrahydro-2H-chromen-5 6 H )-one and tricyclic octahydro- 2 H -benzo[c]-chromen-1 $(6 \mathrm{H})$-one derivatives with enantioselectivity up to ee $99 \%$ has been achieved in the presence of chiral Lewis acid assisted Brønsted acid, (LBA), titanium-isopropoxy-(S)BINOLate under a solvent free condition. The stereochemistry of tricyclic product $\mathbf{1 0}$ has been further supported by single crystal X-ray analysis.This domino powerful strategy combines both the economic and environmental aspects of organic chemistry, which are necessary for academic and industrial applications.

## Introduction

Several bioactive natural products contain tetrahydro-benzopyran moiety, ${ }^{1}$ some examples being forskolin, ${ }^{2}$ hongoquercin $A,{ }^{3 a}$ zanthosimuline, ${ }^{3 \mathrm{bb}} \quad$ chromazonarol, ${ }^{3 \mathrm{Cl}} \quad \Delta^{9}$-tetrahydrocannabinol (THC), ${ }^{4}$ cannabichromene (CBC). ${ }^{5 \text { a }}$ etc. Asymmetric catalysis of organic reactions to provide enantiomerically enriched products is of central importance to modern synthetic and pharmaceutical chemistry. Therefore, the need for truly efficient and practical asymmetric synthesis has been one of the greatest challenges for synthetic chemists. ${ }^{6}$

A common approach for the synthesis of bicyclic tetrahydro2 H -chromen-5(6H)-one and tricyclic octahydro-2H-benzo[c]-chromen- $1(6 \mathrm{H})$-one rings utilizes Knoevenagel-[3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder reactions. Several catalysts have been reported for these reactions. ${ }^{7-13}$ In previous reports, the reactions between 1,3-dicarbonyls ( $\mathbf{2}$ or $\mathbf{3}$ ) and olefinic aldehydes $\mathbf{4 a}$ or $\mathbf{4 b}$ were evaluated in the presence of catalysts such as $\mathrm{InCl}_{3}{ }^{7}$ EDDA ${ }^{8} p$-TSA, ${ }^{9} \quad \mathrm{BF}_{3} \mathrm{OEt}_{2},{ }^{9} \quad \mathrm{TiCl}_{4},{ }^{9} \quad \operatorname{In}(\mathrm{OTf})_{3},{ }^{9} \quad$ phosphoric acid ${ }^{10} E D D A / \mathrm{ZnCl}_{2},{ }^{11}$ water, ${ }^{12} \mathrm{NaOMe} / \mathrm{MeOH}^{13 \mathrm{a}}$ etc. A recent report utilizes a new hollow-structured ZIF-8-H nanosphere as a catalyst for $[3+3]$ cycloaddition reactions. ${ }^{13 b}$

In previous reports, enantioselective cyclization of 2-alkenyl-1,3-diones promoted by Pd-SPRIX catalyst was evaluated, where isomeric 2,2-dialkyl-6,7-dihydro-2H-chromen-5(3H)-ones were formed in ee up to $88 \% .{ }^{14 \mathrm{a}}$ Similarly, the enantioselective addition of conjugated aldehyde to 1,3-cyclohexadione and cyclopentadione in the presence of organo-catalyst (trimethylsilyl-1,1-diaryl-prolinol) has been achieved, which yielded 4-

[^0]

Scheme 1 Reported asymmetric syntheses of chromanonesin the presence of chiral catalysts
substituted-2-hydroxy or 2-acyloxy-chromenones ${ }^{14 b, c}$ in ee up to 97\% (Scheme 1).

We report herein a highly efficient tandem solvent free enantioselective Knoevenagel-formal[3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder reactions in the presence of the chiral LBA titanium-isopropoxy-(S)-BINOLate (1) (Figure 1).


Figure $1\left[(S)-(-)-\mathrm{BINOLate}_{2} \mathrm{Ti}_{2}(\mathrm{O}-i-\mathrm{Pr})_{4} \mathbf{1}(\mathrm{LBA})\right.$

Table 1 Products of enantioselective reactions in the presence of $[(S)-(-)-\text { BINOLate }]_{2} \mathrm{Ti}_{2}(\mathrm{O}-i-\mathrm{Pr})_{4}^{\mathrm{a}, \mathrm{b}, \mathrm{c}}$

3.

.
路

4c
4.

4a


77

10
7.


8
4b


77


4b
-

9
6.

4c


${ }^{\mathrm{b}}$ Isolated yield
${ }^{\text {c }}$ Enantiomeric excess (ee\%) was obtained in GC analysis using chiral Beta Dex column

Thus, the synthesis of bicyclic tetrahydro-2H-chromen-5(6H)one and tricyclic octahydro- 2 H -benzo[c]-chromen-1 $(6 \mathrm{H})$-one derivatives with enantioselectivity up to $98.8 \%$ has been achieved in the presence of chiral LBA, under a solvent free condition (Scheme 2, Table 1).

## Results and discussion

The chiral catalyst 1 was prepared according to the reported procedure. ${ }^{15}$ Initially, the reaction of 1,3-cyclohexadienone (2) and citral (4a) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ atmosphere was attempted, which went to completion in $1 / 2 \mathrm{~h}$, with $79 \%$ yield. A similar reaction of 1,3-cyclohexadienone (2) and citral (4a) in the presence of chiral catalyst 1 under solvent free condition gave similar yield in the same duration of time. Hence we carried further reactions under solvent free condition.

The reaction between $\alpha, \beta$-unsaturated aldehyde such as citral (4a) ( 17.90 mmol ) and 1,3-cyclohexadienone (2) or dimedone (3) $(17.90 \mathrm{mmol})$ was achieved in the presence of $1(3 \mathrm{~mol} \%)$ at $0{ }^{\circ} \mathrm{C}$ under solvent free condition. The reaction was monitored by GC and TLC analyses and was completed within $1 / 2 \mathrm{~h}$ to give the bicyclic tetrahydro- 2 H -chromen-5(6H)-one derivatives 5 ( $80 \%$ yield and $99 \%$ ee) and 8 ( $77 \%$ yield and $89 \%$ ee respectively (Table 1, entries 1 and 4).

Additional reactions of 1,3-dicarbonyl compounds $\mathbf{2}$ or $\mathbf{3}$ with terpenic olefinic aldehydes such as ( $R$ )-citronellal (4b), ( $R$ )-melonal (4c) were also evaluated on the same reaction scale, which yielded tricyclic octahydro-2H-benzo[c]-chromen-1(6H)-one derivatives 6, 7, 9 and 10 respectively in $\sim 77 \%$ yields and ee in the range of 88 $98 \%$ (Table 1, entries 2, 3, 5, 6). Similarly, the reaction of cyclohexa-1,3-dione (2) with conjugated aldehyde (E)-7-formyl-3-methyl-oct-6-enyl acetate (4d) yielded bicyclic chromenone adduct 11 ( $75 \%$ yield and $98 \%$ ee). The reaction of Meldrum's acid (12) with (R)-citronellal (4b) gave tricyclic [1,3]dioxino[4,5-c]isochromen-1(6H)-one derivative 13 . The reaction of cyclopenta-1,3-dione ) with (R)-citronellal (4b) did not yield tricyclic product. Apparently Knoevenagel adduct intermediate in case of cyclopenta-1,3-dione does not undergo hetero-Diels-Alder reaction under present reaction condition.

The required aldehydes $(R)$-melonal ( 4 c ) and ( $R$ )-(E)-7-formyl-3-methyl-oct-6-enyl acetate (4d) were synthesised with minor modifications of reported procedures. ${ }^{16}$

The structures of compounds 5-11, 14 and 15 were identified with IR, ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, GC/MS and HRMS data. In ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum, the olefinic protons of 5 absorbed at $\delta 6.45(d, J=10.12$ $\mathrm{Hz}, 1 \mathrm{H}), 5.18(d, J=10.12 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum of compound 5 showed peaks at $\delta 194.9$ and 172.1 due to $\mathrm{C}-5$ carbonyl and C-8a enol carbon respectively. The olefinic carbons absorbed at $\delta 131.9,123.6,121.6,116.4,110.3$ and ethereal carbon at $\delta 82.4$. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ compound 6 showed absence of olefinic protons. The ${ }^{13} \mathrm{C}$ NMR spectrum of compound 6 showed peaks at $\delta$ 197.8 and 170.5 due to carbonyl and enol carbon respectively. The ${ }^{1} \mathrm{H}$-NMR and ${ }^{13} \mathrm{C}$-NMR of remaining molecules were also in agreement with their structures.

2, $R_{1}=R_{2}=H$, 3, $\mathrm{R}_{1}=\mathrm{R}_{2}=\mathrm{Me}$

6, 7, 9, 10
$\mathrm{n}=0$ or 1

Scheme 2 Stereoselective reactions of cyclohexa-1,3-dienones with Aldehydes in the presence of $\left[(S)-(-)-\mathrm{BINOLate}_{2} \mathrm{Ti}_{2}(\mathrm{O}-i-\mathrm{Pr})_{4}\right.$ (Chiral LBA, 1)


Figure 2 ORTEP diagram of compound 10

The enantiomeric excess (ee\%) of compounds 5-11, 14, 15 was obtained in GC analysis using chiral Beta-Dex column. High stereoselectivity was observed, ee\% ranging from 88.1-98.8\%, for both bicyclic $(5,8,11)$ and tricyclic $(6,7,9,10)$ molecules (Table 1). The structure and stereochemistry of tricyclic compound 10 was further supported by single crystal X-ray analysis, ${ }^{17-19}$ which is shown in Figure 1. It is noteworthy, that the stereochemistry at ring junction positions 3 a and 9 b of analogue $\mathbf{1 0}$ is opposite to that of natural $(R, R)$-THC. ${ }^{4}$ In view of strong analgesic activity of $(S, S)$ daxanabino ${ }^{5 b}$ the present report has more significance.

The 2-(R)-stereochemistry of 7,8-dihydro-2H-chromen-5(6H)one derivatives 5, 8 and 11 was assigned based on (-)-optical rotation and CD spectra. ${ }^{20,21}$

The molecules 7, 10 and 11 are novel to the best of our knowledge.

The probable mechanism for stereoselective formation of (2R)-5,6,7,8-tetrahydro-5-oxo-2H-chromene derivatives is shown in Figure 3. In Intermediate A the bulky substituent (R-) in aldehyde prefers equatorial position, which is converted to condensation product intermediate $\mathbf{B}$. The elimination of water molecule leads to intermediate $\mathbf{C}$, subsequent attack by isopropanol releases (2R)-5,6,7,8-tetrahydro-5-oxo-2H-chromene derivative and the catalyst 1 is regenerated. Similar mechanism for stereospecific formation of tricyclic octahydro-6,6,9-trimethyl-2H-benzo[c]-chromen-1(6H)onederivatives is given in scheme 4. The Aldol condensation of Intermediate A leads to Intermediate B, which undergoes dehydration to give $\mathbf{C}$, the hetero-Diels-Alder reaction, subsequent attack by isopropanol and release of tricyclic octahydro-6,6,9-trimethyl- 2 H -benzo[c]-chromen-1(6H)-one derivative regenerates the catalyst 1.

The present catalytic system provides an attractive protocol to various optically active derivatives of bicyclic chromen-5(6H)-one and tricyclic benzo[c]-chromen-1 $(6 \mathrm{H})$-onein terms of the following features: (i) the catalyst is inexpensive and easily available; (ii) the protocol has a broad scope of substrates; (iii) the reactions show


Figure 3 The probable mechanism of formation of (2R)-5,6,7,8-tetrahydro-5-oxo-2H-chromene derivatives
high enantioselectivity; (iv)short reaction time, reactions are completed within $1 / 2 \mathrm{~h}$; (v) the reaction is environmentally benign because of solvent-free condition; and (vi) low catalyst loading (5 $\mathrm{mol} \%$ ) is sufficient to achieve high yield and optical purity of the products. We hope our findings in this research will stimulate further work on practical asymmetric synthesis of more molecules of biological and pharmacological importance.

## Conclusions

In summary, highly efficient catalyst has been observed for tandem solvent free enantioselective Knoevenagel-formal[3+3] cycloaddition and Knoevenagel-Hetero-Diels-Alder Reactions.


Figure 4 The probable mechanism of formation of octahydro-6,6,9-trimethyl-2H-benzo[c]-chromen-1(6H)-onederivatives

Thus, the synthesis of bicyclic tetrahydro- 2 H -chromen-5(6H)-one and tricyclic octahydro-2H-benzo[c]-chromen-1(6H)-one derivatives with enantioselectivity up to ee 99\% has been achieved in the presence of titanium-isopropoxy-(S)-BINOLate under a solvent free condition. The stereochemistry of tricyclic product 10 has been further supported by single crystal X-ray analysis. The methodology reported herein can be used for the synthesis of various natural and non-natural molecules with chromenone moiety. This one-pot powerful strategy combines both the economic and environmental aspects of organic chemistry which are necessary for academic and industrial applications.

## Experimental

## General procedures

The monitoring of reaction and checking the purity of the products were done using pre-coated silica gel plates (Merck) and visualization using anisaldehyde/ $\mathrm{H}_{2} \mathrm{SO}_{4}$ reagent. FT-IR spectra were recorded on a Perkin-Elmer Spectrum One spectrometer. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Varian spectrometer at 400 MHz and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ at 100 MHz ; $\delta$ in ppm rel. to $\mathrm{Me}_{4} \mathrm{Si}$ as internal standard, $J$ in Hz . Multiplicities is reported as follows: $s=$ singlet, $d$ $=$ doublet, $\mathrm{dd}=$ doublets of doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{m}=$ multiplet, brs = broad singlet.GC-MS was carried on an Agilent instrument, where GC-6890 was coupled with a mass spectrometer MS-5973 N with quadrapole mass detector, using a HP-5 (5\% phenyl methyl siloxane) column. Electrospray ionization and a TOF mass analyser were used for HRMS measurements. The compounds (5-11) showed the required $\mathrm{m} / \mathrm{z}$ : $\left(\mathrm{M}^{+}\right)$values in HRMS. Enantiomeric excess (ee\%) was obtained in GC analysis using chiral Beta Dex 120 column ( $30 \mathrm{~m} \times 0.25 \mu \mathrm{~m} \times 0.25 \mathrm{~mm}$ ). Silica gel (100-200 mesh), which was used for column chromatography (CC) was activated by heating at $120^{\circ}$ for 4 h . All asymmetric reactions were performed under inert atmosphere and at $0^{\circ} \mathrm{C}$.

General procedure for synthesis of compounds 5-11, 14 and 15
A mixture of cyclohexa-1,3-dione (2) or 5,5-dimethyl-cycohexa-1,3-dione (3) or Meldrum's acid (12), ( 17.90 mmol ), terpenic aldehyde (4a-4d, 17.90 mmol ) and catalyst (1) ( $3 \mathrm{~mol} \%$ ) was stirred at $0{ }^{\circ} \mathrm{C}$ under solvent free condition for 30 min under $\mathrm{N}_{2}$ atmosphere. After completion of the reaction (monitored by TLC/ GC analysis), the reaction mixture was diluted with hexane: $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (1:1, 5 mL ) and filtered. The filtrate was evaporated and the residue was purified by silica gel column chromatography to afford the pure products.
(R)-2-methyl-2-(4-methylpent-3-enyl)-7,8-dihydro-2H-
chromen-5(6H)-one (5). Yellow oil; $[\alpha]_{D}=-65.2^{\circ}(c=1$ in MeOH), Chiral GC analysis: 33.79 (major), 33.19 (minor) min, ee 99\%; IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2939, 1647, 1603, 1428, 1381, 1333, 1285, 1257, 1184, 1111, 1081, 967, 864, 787; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}, \mathrm{TMS}\right):$ $\delta$ (PPM) $6.45(\mathrm{~d}, J=10.12 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{~d}, J=10.12 \mathrm{~Hz}, 1 \mathrm{H}), 5.09-$ $5.07(\mathrm{~m}, 1 \mathrm{H}), 2.42-2.39(\mathrm{~m}, 4 \mathrm{H}), 2.01-1.94(\mathrm{~m}, 4 \mathrm{H}), 1.74-1.71(\mathrm{~m}$, $1 \mathrm{H}), 1.68(\mathrm{~s}, 3 \mathrm{H}), 1.59(\mathrm{~s}, 3 \mathrm{H}), 1.68-1.53(\mathrm{~m}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta$ (PPM) 194.9, 172.1, 131.9, 123.6, 121.6, 116.4, 110.3, 82.4, 41.7, 36.4, 28.6, 27.4, 25.7, 22.5, 20.6, 17.6; GC-MS (m/z) : 246, 231, 213, 203, 190, 175, 163, 155, 147, 135, 122, 107, 99, 91, 77, 69, 55, 41; HRMS (EI') calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{2}$ 246.1620, found $m / z 246.1622$.

Synthesis of ( $\pm$ )-2-methyl-2-(4-methylpent-3-enyl)-7,8-dihydro-2H-chromen-5(6H)-one (5)using ethylenediamine diacetate as catalyst. To a mixture of 1,3-cyclohexadione, $2(2 \mathrm{~g}, 17.9 \mathrm{mmol})$ and citral $4 \mathrm{a},(2.73 \mathrm{~g}, 17.9 \mathrm{mmol})$ was added catalyst EDDA ( 5 mol $\%$ ) and stirred at room temperature for 1 h . After completion of the reaction, the reaction mixture was purified by column chromatography using $n$-hexane/ EtOAc (4:1) as eluent to provide $5(3.31 \mathrm{~g}, 75.4 \%)$ as a yellow oil. $[\alpha]_{\mathrm{D}}=0$.
(6aS,9R,10aS)-6,6,9-trimethyl-2,3,4,6,6a,7,8,9,10,10a-dehydro-1H-benzo[c]chromen-1-one (6). Colourless solid; $[\alpha]_{D}=-100.1^{\circ}(c$ $=1$ in MeOH ), Chiral GC analysis 36.61 min (major), 37.78 min (minor), ee\% 97); IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2920, 2847, 1649, 1598, 1447, $1379,1273,1252,1192,1128,999,850,775,693 ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta$ (PPM) $2.80-2.77$ (ddt, $J=11.5,3,1.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.43-2.18 (m,4 H), 2.13-2.06 (dt, J= 11.08, $2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.92-1.86$ $(\mathrm{m}, 2 \mathrm{H}), 1.86-1.78(\mathrm{~m}, 1 \mathrm{H}), 1.78-1.72(\mathrm{~m}, 1 \mathrm{H}), 1.61-1.50(\mathrm{~m}, 1 \mathrm{H})$, $1.33(\mathrm{~s}, 3 \mathrm{H}), 1.29-1.23(\mathrm{td}, \mathrm{J}=11.3,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.05(\mathrm{~s}, 3 \mathrm{H}), 1.12-$ $0.95(\mathrm{~m}, 2 \mathrm{H}), 0.91-0.89(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.49(\mathrm{q}, J=11.5 \mathrm{~Hz}, 1 \mathrm{H})$; ${ }^{13}$ C NMR ( $100 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta$ (PPM) 197.8, 170.5, 114.7, 80.3, 48.6, $38.8,37.5,35.5,33.6,32.3,29.6,27.5,27.2,22.5,20.2,19.4 ;$ GCMS (m/z) : 248, 233, 219, 205, 192, 177, 163, 150, 137, 123, 109, 95, 81, 68, 55, 41; HRMS (EI') calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{2}$ 248.1776, found $m / z 248.1774$.
(1R,3aS,9bS)-1,4,4-trimethyl-1,2,3,3a,4,7,8,9b-octahydrocyclo-penta[c]chromen-6(6H)-one (7). Colourless liquid. $[\alpha]_{D}=-86.6^{\circ}(c$ $=1, \mathrm{MeOH}$ ); Chiral GC analysis 34.23 (major), 33.88 (minor) min, (ee\% 88); IR $\left(\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}\right): 2945,2869,1651,1582,1455,1428$, 1379, 1334, 1278, 1226, 1192, 1113, 1064, 1004, 991, 848, 629; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}, \mathrm{TMS}\right): \delta(\mathrm{PPM}) 2.41-2.34$ ( dt, $J=11.4,5 \mathrm{~Hz}, 1 \mathrm{H}), 2.35-2.30(\mathrm{~m}, 2 \mathrm{H}), 2.29-2.26(\mathrm{~m}, 2 \mathrm{H}), 2.04-1.98$ $(\mathrm{m}, 1 \mathrm{H}), 1.91-1.87(\mathrm{~m}, 1 \mathrm{H}), 1.86-1.81(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.55(\mathrm{~m}, 2 \mathrm{H})$, $1.41-1.39(\mathrm{~d}, \mathrm{~J}=6.16 \mathrm{~Hz}, 3 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H}), 1.13(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta$ (PPM) 197.5, 171.2, 115.4, 81.5, 53.5, 42.6, 37.6, 36.6, 34.4, 29.7, 28.4, 24.8, 23.4, 20.4, 20.1; GC purity :
90.0\%; GCMS (m/z): 234, 219, 191, 177, 163, 151, 137, 125, 109, 91, 79, 67, 55, 41; HRMS (EI') calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{2}$ 234.1620, found $\mathrm{m} / \mathrm{z} 234.1622$.
( R )-2,7,7-trimethyl-2-(4-methylpent-3enyl)-7,8-dihydro-2H-
chromen-5(6H)-one (8). Yellow oil. $[\alpha]_{D}=-66.2^{\circ}(c=1, \mathrm{MeOH})$; Chiral GC analysis) 33.86 (major), 32.39 (minor) min, (ee\% 89); IR $\left(\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}\right): 2961,1680,1606,1449,1382,1331,1287,1217$, 1198, 1165, 1112, 1077, 1034, 936, 889; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right.$, TMS): $\delta(P P M) 6.44(d, J=10 H z, 1 H), 5.17(d, J=10 H z, 1 H), 5.11-$ $5.07(\mathrm{~m}, 1 \mathrm{H}), 2.29-2.23(\mathrm{~m}, 4 \mathrm{H}), 2.08-2.01(\mathrm{~m}, 2 \mathrm{H}), 1.76-1.69(\mathrm{~m}, 1 \mathrm{H})$, $1.67(\mathrm{~s}, 3 \mathrm{H}), 1.57(\mathrm{~s}, 3 \mathrm{H}), 1.56-1.54(\mathrm{~s}, 1 \mathrm{H}), 1.37(\mathrm{~s}, 3 \mathrm{H}), 1.08(\mathrm{~s}, 3 \mathrm{H})$, 1.06 (s,3H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta(\mathrm{PPM}) 194.4,170.6$, 131.8, 123.6, 121.4, 116.2, 109.0, 82.5, 50.3, 42.4, 41.7, 32.1, 28.6, 28.2, 27.5, 25.7, 22.5, 17.6; GCMS (m/z): 274, 259, 241, 231, 218, 205, 191, 175, 165, 157, 149, 141, 129, 121, 107, 91, 77, 69, 55, 41; HRMS ( $\mathrm{EI}^{+}$) calcd for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{O}_{2}$ 274.1933, found $m / z$ 274.1930.
(6aS,9R,10aS)-3,3,6,6,9-pentamethyl-2,3,4,6,6a,7,8,9,10,10a-decahydro-1H-benzo $[c]$ chromen-1-one(9). Yellow oil. $[\alpha]_{D}=-$ $101.5^{\circ}$ ( $c=1, \mathrm{MeOH}$ ); Chiral GC analysis 34.33 (major), 33.78 (minor) min, (ee\% 89); IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2957, 2868, 1644, 1606, 1455, 1381, 1360, 1278, 1236, 1165, 1088, 1031, 1013, 941, 865, $665 ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}, \mathrm{TMS}\right): \quad \delta$ (PPM) 2.85 (ddt, $J=$ $11.2,3,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.22-2.18(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.18(\mathrm{~m}, 1 \mathrm{H}), 1.84-1.78$ $(\mathrm{m}, 1 \mathrm{H}), 1.78-1.72(\mathrm{~m}, 1 \mathrm{H}), 1.61-1.50(\mathrm{~m}, 1 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.26(\mathrm{td}, \mathrm{J}=$ $11.2,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.12-0.98(\mathrm{~m}, 2 \mathrm{H}), 1.05(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{~s}, 3 \mathrm{H}), 1.03$ $(\mathrm{s}, 3 \mathrm{H}) .0 .90(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.49(\mathrm{q}, J=11.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz, TMS): $\delta($ PPM $) 197.9,175.3,168.8,113.2,80.4,51.4,48.8$, 45.4, 43.3, 43.2, 38.6, 35.6, 33.5, 32.3, 31.6, 29.3, 28.1, 27.6, 27.3, 27.2, 22.5, 19.5; GCMS (m/z) : 276, 261, 243, 220, 205, 192, 177, 165, 153, 134, 123, 109, 91, 81, 69, 55, 41; HRMS (EI') calcd for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{2}$ 276.2089, found $m / z 276.2091$.

## (1R,3aS,9bS)-1,4,4,7,7-pentamethyl-1,2,3,3a,4,7,8,9b-

octahydrocyclo-penta[c]chromen-9(6H)-one (10). Yellow oil. $[\alpha]_{D}$ $=-89.6^{\circ}(c=1, \mathrm{MeOH})$, Chiral GC analysis 33.55 min (major), 34.23 $\min$ (minor), (ee\% 99); IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2950, 1728, 1651, 1586, 1468, 1378, 1230, 1118, 1024, 938, 860, 757; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}, \mathrm{TMS}): \delta($ PPM $) 2.29-2.10(\mathrm{~m}, 4 \mathrm{H}), 2.08-1.99(\mathrm{~m}, 1 \mathrm{H}), 1.99-$ $1.91(\mathrm{~m}, 1 \mathrm{H}), 1.89-1.81(\mathrm{~m}, 1 \mathrm{H}), 1.73-1.57(\mathrm{~m}, 2 \mathrm{H}), 1.44(\mathrm{~d}, \mathrm{~J}=6 \mathrm{~Hz}$, $3 \mathrm{H}), 1.38(\mathrm{~s}, 3 \mathrm{H}), 1.22-1.13(\mathrm{~m}, 2 \mathrm{H}), 1.16(\mathrm{~s}, 3 \mathrm{H}), 1.07(\mathrm{~s}, 3 \mathrm{H}), 1.04$ $(\mathrm{s}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta($ PPM $) ~ 198.4, ~ 197.6, ~ 169.4$, 167.5, 114.1, 81.6, 78.3, 53.7, 51.8, 43.5, 42.5, 42.8, 36.4, 34.6,31.7, 28.7, 28.5, 27.9, 25.0, 23.7, 20.4; GCMS ( $\mathrm{m} / \mathrm{z}$ ) : 262, 247, 219, 205, 191, 179, 165, 151, 109, 91, 83, 67, 55, 41; HRMS ( $\mathrm{El}^{+}$) calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{2}$ 262.1933, found $m / z 262.1930$.
(R)-3-methyl-5-((R)-3-methyl-5-oxo-5,6,7,8-tetrahydro-2H-chromen-2-yl)pentyl acetate (11). Yellow oil. $[\alpha]_{D}=-90.5^{\circ}(c=1$, MeOH ), Chiral GC analysis 33.88 min (major), 32.39 min (minor), (ee\% 98); IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2920, 2847, 1729, 1649, 1598, 1447, 1379, 1273, 1252, 1192, 1128, 999, 850, 775, 693; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta(\mathrm{PPM}) 6.21(\mathrm{~s}, 1 \mathrm{H}), 4,75$ (dd, $J=10.8 \mathrm{~Hz}, 3.5 \mathrm{~Hz}$, $1 \mathrm{H}), 4.18-4.12(\mathrm{~m}, 4 \mathrm{H}), 2.42-2.35(\mathrm{~m}, 4 \mathrm{H}), 2.05(\mathrm{~s}, 3 \mathrm{H}), 1.99-1.93$ $(\mathrm{m}, 2 \mathrm{H}), 1.72(\mathrm{~s}, 3 \mathrm{H}), 1.68-1.62(\mathrm{~m}, 1 \mathrm{H}), 1.62-1.53(\mathrm{~m}, 2 \mathrm{H}), 1.50-$ $1.38(\mathrm{~m}, 2 \mathrm{H}), 0.95-0.90(\mathrm{dd}, J=6.4 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , TMS): $\delta$ (PPM) 195.1, 171.2, 169.9, 126.1, 112.5, 80.9, 62.7, 36.4, $35.4,35.1,31.3,30.3,29.7,27.9,26.3,20.9,19.5,19.1 ;$ GCMS $(m / z): 306,291,263,189,163,121,107,91,77,56,43 ;$ HRMS (EI ${ }^{+}$) calcd for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{O}_{4} 306.1831$ found $\mathrm{m} / \mathrm{z}$ 306.1833.
( $6 a S, 9 R, 10 a S$ )-6a,7,8,9,10,10a-hexahydro-3,3,6,6,9-pentamethyl-[1,3]dioxino[4,5-c]isochromen-1(6H)-one (13). Yellow oil. $[\alpha]_{\mathrm{D}}=-40.2^{\circ}(c=1, \mathrm{MeOH})$, Chiral GC analysis 33.77 min (major), 32.76 min (minor), (ee\% 83); IR ( $\mathrm{CHCl}_{3}, \mathrm{~cm}^{-1}$ ): 2925, 2855, $1736,1726,1620,1455,1390,1280,1155,994,965,891,731 ;{ }^{1} \mathrm{H}-$ NMR (CDCl $3,400 \mathrm{MHz}, \mathrm{TMS}$ ): $\delta($ PPM $) 2.65$ (dd, $J=5.6,18.4 \mathrm{~Hz}$, $1 \mathrm{H}), 2.08-2.01(\mathrm{~m}, 1 \mathrm{H}), 1.85-0.95(\mathrm{~m}, 6 \mathrm{H}), 1.68(\mathrm{~s}, 3 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H})$, $1.41(\mathrm{~s}, 3 \mathrm{H}), 1.26(\mathrm{~s}, 3 \mathrm{H}), 0.91(\mathrm{~d}, \mathrm{~J}=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.78-0.62(\mathrm{~m}, 1 \mathrm{H}){ }^{13} \mathrm{C}$ NMR (100 MHz, TMS): $\delta$ (PPM) 171.4, 167.3, 124.4, 85.9, 47.4, 42.3, 37.2, 36.3, 32.0, 31.8, 28.7, 27.6, 25.9, 23.5, 22.3. GCMS $\left(\mathrm{m} / \mathrm{z}\right.$ ) : 280,; HRMS ( $\mathrm{EI}{ }^{+}$) calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{4} 280.1704$ found $\mathrm{m} / \mathrm{z}$ 280.1703.

X-ray single crystal structure determination of molecule 10. A suitable crystal of size $0.06 \times 0.18 \times 0.24 \mathrm{~mm}^{3}$ was mounted on a diffractometer for unit cell determination and three dimensional intensity data collection. 800 frames in total were collected at 150 K with the exposure time of 16 s per frame. Data integration, indexing and absorption correction were followed by structure solution using the programs in a WinGX module. ${ }^{17}$ The structure was solved by direct methods (SIR-92) ${ }^{18}$ and the final refinement of the structure was carried out using full least-squares methods on $\mathrm{F}^{2}$ using SHELXL-97. ${ }^{19}$ Unit cell determination using both high and low angle reflections reveals that compound $\mathbf{1 0}$ crystallizes in a monoclinic $\mathrm{P} 2_{1} / \mathrm{c}$ space group. Non-hydrogen atoms were refined anisotropically. C-H hydrogen atoms were placed in geometrically calculated positions by using a riding model.

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## Notes and references

$\ddagger$ Crystallographic data -Unit cell determination using both high and low angle reflections revealed that compound $10, \mathrm{C}_{17}$ $\mathbf{H}_{26} \mathbf{O}_{2}$; 262.38;crystallizes in a, monoclinic $\mathrm{P} 2_{1} / \mathrm{c}$ space group; Unit cell dimensions $a=5.9897(17)$, alpha $=90^{\circ}, b=12.612(4)$, beta $=96.062^{\circ}(5), c=19.986(6) \AA$ A , gamma $=90^{\circ}$; Volume 1501.3(8) A^3
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