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1	Targeting Vitamin E TPGS-Cantharidin Conjugate Nanoparticles for Colorectal Cancer				
2	Therapy				
3	Shihou Sheng <sup>1</sup> , Tao Zhang <sup>2</sup> , Shijie Li <sup>3</sup> , Jun Wei <sup>4</sup> , Guangjun Xu <sup>4</sup> , Tianhong Sun <sup>5</sup> , Yahong Chen <sup>1</sup> ,				
4 Fengqing Lu <sup>1</sup> , Yongchao Li <sup>6</sup> , Jinghui Yang <sup>7</sup> , Huiqiu Yu <sup>8</sup> , Tongjun Liu <sup>1</sup> *, Gang Han <sup>9</sup> **					
5 6	<ul> <li><sup>1</sup>Department of Colorectalrectal and Anal Surgery, China-Japan Union Hospital of Ji Lin</li> <li>University, ChangChun 130000, China</li> </ul>				
7 8	<ul> <li><sup>2</sup> Department of Gastrointestinal Surgery, China-Japan Union Hospital XinMin District of JiLin</li> <li>University, ChangChun 130000, China</li> </ul>				
9 10	<ul> <li><sup>3</sup> Department of Thyroid Surgery, China-Japan Union Hospital of JiLin University, ChangChun</li> <li>130000, China</li> </ul>				
<sup>4</sup> Department of Neurosurgery, China-Japan Union Hospital of JiLin University, ChangChun 130000, China					
12 13	<sup>5</sup> Department of Operating Room, China-Japan Union Hospital of JiLin University, ChangChun 130000, China				
14 15	<sup>6</sup> Department of Gastrointestinal Surgery, China-Japan Union Hospital of JiLin University, ChangChun 130000, China				
16 17	<sup>7</sup> Department of Hepatopancreatobiliary Surgery, China-Japan Union Hospital of JiLin University, ChangChun, 130000,China				
18 19	<sup>8</sup> Department of Rehabilitation Medicine, China-Japan Union Hospital of JiLin University, ChangChun 130000,China				
20 21	<sup>9</sup> Department of Gastrointestinal Surgery, The Second Hospital of JiLin University,ChangChun 130000,China				
22	2 *corresponding author.Tel/fax:+86 431 89876792				
23	**corresponding author.Tel:+86 18743099891				
24					
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Abstract: A traditional Chinese medicine cantharidin which was previously found to be effective on colorectal cancer cells was translated into nanoparticles for drug delivery to reduce its side effects and enhance its drug efficacy. By further introducing the folate targeting ligands to the nanoparticles, targeting delivery of cantharidin to colorectal cancer is realized. catharidin loaded nanoparticles can increase the cytotoxicity of cantharidin on the folate over-expressed HT-29 cells; Moreover, introducing folate to the nanoparticles can further increase its efficacy, while this obvious enhancement cannot be found on the MCF-7 cells with lower folate receptors.

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35 Key words: canthardin, TPGS, nanoparticles, drug delivery, folate

#### 36 Introduction

37 Nowadays, colorectal cancer has already become the third most commonly diagnosed cancer in 38 both men and women in US excluding skin cancer. Moreover, it is the second leading cause of death in 39 cancer patients in US for both men and women [1,2]. Though hundreds of potential anticancer drugs are 40 available on the market, most of them are not ideal, which can cause serious adverse effects [3]. A very 41 promising candidate is cantharidin, which is a type of terpenoid, a chemical compound secreted by many 42 species of blister beetle [4]. Cantharidin has long been used as a traditional Chinese medicine for the 43 treatment of a variety of cancers, including liver, lung, intestinal, and digestive tract tumors [4,5]. 44 Researchers around the world have found that cantharidin as well as its derivatives demonstrated strong 45 affinity and specificity for protein phosphatase 2A (PP2A). It is reported that the level of PP2A inhibition 46 parallels their cytotoxicity [6]. Recently, cantharidin is also found to be effective against several colorectal 47 cancer cells through inhibition of heat shock protein and Bcl-2 associated athanogene domain 3 48 expressions by blocking heat shock factor 1 binding to promoters [7]. Nevertheless, cantharidin has 49 displayed serious side effects such as dysphasia, hematemesis, and dysuria [8]. Therefore, a potential 50 way of reducing the side effects but remain its activity is urgent.

51 Nanoparticle based drug delivery system (DDS), defined as DDS, with particle diameters of 52 approximately 100 nm or less, is attracting considerable attention worldwide as efficient cancer 53 therapeutics for overcoming some of the limitations of conventional anticancer drug therapy [9-10]. For an 54 efficient DDS, the right choice of a drug carrier is vital. A good example is d-alpha-tocopheryl 55 polyethylene glycol 1000 succinate monoester (TPGS), which is an amiphiphilic PEGylated vitamin E and 56 can work as best surfactant for hydrophobic drugs both via encapsulating the drugs or conjugating with 57 the drugs and then assembling the conjugates into nanoparticular drug delivery systems[11]. What's more, 58 the drug carrier TPGS has almost no toxic issues as it is widely applied in the food and drug industry [12].

59 Folate is a small molecular compound, which is very important for tumor cell proliferation and 60 survival. Several cancer cells were over-expressed folate receptors as much as 200-fold on the cell 61 surfaces compared to the normal cells. The over-expressed receptors are vital to the intake of folate to

the tumor cells [13,14]. Typically, folate receptor is over-expressed in ovarian cancer, breast cancer, head and neck cancer, and some childhood cancers [15]. Given these attributes of folate receptors, folic acid has been conjugated to many delivery systems for cancer therapy including liposomes, polymeric micelles and capsules, upconversion nanoparticles and carbon nanotubes, etc [16-19]. Moreover, previous study revealed that the folate receptor was highly expressed in colorectal cancers [20], which makes it possible to utilize folate for target delivery of drugs for treatment of colorectal cancer.

68 Taking into account the unique anticancer property of cantharidin and the drug delivery 69 technology developed till far, researchers around the world have tried tremendous ways of delivering 70 canthardin or its derivatives directly to the cancer cells for cancer therapy. Zhu, et al reported preparation 71 of cantharidin-loaded solid lipid nanoparticles (CA-SLNs) with oral bioavailability by a film dispersion-72 ultrasonication method [21]. The result showed that CA-SLNs had a sustained release profile without a 73 burst effect and a higher bioavailability than free cantharidin after oral administration. Later, Zhu, et al 74 reported an inclusion complex of cantharidin with β-cyclodextrin for drug delivery [22]. However, the in 75 vitro and in vivo drug efficacy efficacy was not studied. More recently, norcantharidin, a demethyl 76 derivative of cantharidin with lower toxicity was conjugated to polyethylenimine (PEI) and polylysine (PLL) 77 for acid liable drug release by Shen, et al [23]. Though this system showed reasonable acid 78 responsiveness, the drug itself was not the more efficient cantharidin and the polymers used are not FDA 79 approved, making further clinical use harder. Taken together, we here show the first example of rational 80 design of cantharidin loaded TPGS nanoparticles for targeting delivery of cantharidin and effective 81 treatment of colorectal cancer.As cantharidin is an anhydride which can undergo readily reaction with 82 hydroxyl groups. Via a simple one step ring opening reaction of cantharidin with the end hydroxyl group of 83 TPGS, a cantharidin-TPGS conjugate (Can-TPGS, Scheme 1, a) was obtained. Due to the amphiphilic 84 nature of TPGS, this Can-TPGS conjugate can self-assemble in aqueous solution into nanoparticles as 85 effective cantharidin delivery systems (Scheme 1, b). To further increase the efficacy of this system, folate 86 was introduced to the nanoparticles via assembly of FA-TPGS and Can-TPGS together (Scheme 1,c). 87 The novel cantharidin loaded nanoparticles were systematically characterized and studied in vitro on two 88 cancer cell lines, HT-29 (human colorectal cancer cell, folate receptor over-expressed)[24] and MCF7 89 (Human breast cancer, folate receptor low-expressed)[25]. Results showed that catharidin loaded 90 nanoparticles can increase the cytotoxicity of cantharidin on the folate over-expressed HT-29 cells; 91 Moreover, introducing folate to the nanoparticles can further increase its efficacy, while this obvious 92 enhancement cannot be found on the MCF-7 cells.

#### 93 Materials

4-dimethylaminopyridine (DMAP), folic acid, d-alpha-tocopheryl polyethylene glycol 1000
 succinate monoester (TPGS) was purchased from Sigma-Aldrich. Cantharidin was purchased from Santa
 Cruz Biotech. All other chemicals and reagents were used as reagent grade without further purification in
 all of the experiments.

#### 98 Instrumentation

99 <sup>1</sup>HNMR spectra were recorded at ambient temperature on a JEOL JNM LA400 NMR 100 spectrometer. The nanoparticles were analyzed using Transmission electron microscopy (TEM) (TEM; 101 JEOL JEM-1011). The Size and size distribution of the nanoparticles was then determined by dynamic 102 light scattering (DLS) with a vertically polarized He-Ne laser (DAWN EOS, Wyatt Technology, USA). The 103 scattering angle was fixed at 90° and the measurement was carried out at a constant temperature of 25°C. 104 The zeta potential of the nanoparticles prepared here was conducted on a Malvern Zetasizer Nano ZS 105 which was calibrated using a 60 nm polystyrene standard. Nanoparticle samples are prepared by 106 dissolving into de-ionized water and the particle size and zeta potential were measured simultaneously in 107 triplicate.

#### 108 Synthesis of Can-TPGS conjugates

109 TPGS (1.5 g, 1 mmol) was dissolved in 50 mL anhydrous  $CH_2Cl_2$  in a round flask, to which 1 N 110 cantharidin anhydride (0.196 g, 1 mmol) and DMAP (0.122 g, 1 mmol) were added. The reaction mixture 111 was heated to reflux under stirring overnight in an oil bath. After that, the reaction mixture was left cooled 112 down to room temperature and subjected to rotation evaporation. Then, the obtained product was 113 dissolved in 5 ml dimethylformamide (DMF) and subjected to 3 days of dialysis against water in a dialysis 114 bag with a molecular cut-off at 1000 Da.Then the purified product was collected and lyophilized as white 115 powder for further use.

#### 116 Synthesis of FA-TPGS conjugates

117 TPGS (0.15 g,0.1 mmol) was dissolved in 10 mL anhydrous dimethyl sulfoxide(DMSO) in a round flask, to which 1 N folic acid (0.044 g, 0.1 mmol), DCC(0.041 g,0.2 mmol) and DMAP( 0.024 g,0.2 mmol) 118 119 were added. The reaction mixture was kept in ice bath under stirring overnight. After that, the reaction 120 mixture was filtered to remove dicyclohexyl urea. Then the DMSO solution of the FA-TPGS conjugate 121 was put into a dialysis bag with a molecular cut-off at 1000 Da against DMSO for 3 days to remove 122 excess DCC and DMAP. In the meanwhile, DMSO outside the dialysis bag was changed several times. 123 After that, the product was dialyzed against water for 3 days. Then, it was collected and lyophilized to get 124 FA-TPGS powder.

#### 125 Preparation of TPGS NPs and FA-TPGS NPs

For preparation of the blank carrier nanoparticles (TPGS NPs) and FA targeting blank nanoparticles (FA-TPGS NPs), a nano-precipitation method was used [26]. Taking TPGS NPs as an example, briefly, 20 mg of TPGS was dissolved in 2 ml of acetone under vigorously stirring, after that, 15 ml of de-ionized water was added. Then this solution was left stirring at room temperature for volatizing the acetone. Thereafter, the nanoparticles were lyophilized for further use. For making the targeting

nanoparticles, the staring materials become TPGS and FA-TPGS at a ratio of 0.95:0.05. All the othersteps were the same.

#### 133 Preparation of Can-NPs and FA-Can-NPs

For preparation of the cantharidin loaded nanoparticles (Can-NPs) and folate targeting nanoparticles (FA-Can-NPs), a nano-precipitation method was used [26]. Taking Can-NPs as an example, briefly, 20 mg of Can-TPGS conjugates was dissolved in 2 ml of acetone under vigorously stirring, after that, 15 ml of de-ionized water was added. Then this solution was left stirring at room temperature for volatizing the acetone. Thereafter, the nanoparticles were lyophilized for further use. For making the targeting nanoparticles, the staring materials become Can-TPGS and FA-TPGS at a ratio of 0.95:0.05. All the other steps were the same.

#### 141 Preparation rhodamine B loaded nanoparticles of Can/RhB-NPs and FA-Can/RhB-NPs

To prepare fluorescent molecule labeled nanoparticles for further *in vitro* study, rhodamine B(RhB) labeled Can-NPs and FA-Can-NPs called Can/RhB-NPs and FA-Can/RhB-NPs were prepared respectively using the method described above[26]. First, RhB-TPGS conjugate was prepared as previously described for FA-TPGS conjugates. Then the RhB-TPGS conjugate was used together with Can-TPGS or Can-TPGS as well as FA-TPGS to prepare Can/RhB-NPs and FA-Can/RhB-NPs respectively.

#### 148 Cantharidin drug release and Zeta potential monitoring during drug release

149 The hydrolysis of cantharidin from Can-NPs was monitored by <sup>1</sup>H NMR spectroscopy as previously described [23].Briefly, Can-NPs (200 mg) were dissolved in 20 ml PBS (pH=7.4, 10 mM) or 150 151 acetate buffered solution (pH=5.0, 10 mM). The samples were put into the dialysis bag at a molecular 152 cutoff at 1000 Da with shaking. At desirable time intervals, 2 ml of samples was removed from the dialysis 153 bag. The collected polymer was subjected to ultra-centrifugation and washed 3 times by water to remove 154 salt and then lyophilized. Their <sup>1</sup>HNMR spectra were recorded. The degree of hydrolysis was calculated 155 by comparing the peak at 4.7ppm (peak c in cantharidin in Figure 1a) with the peak at 3.65 ppm of PEG (-156  $CH_2$ - $CH_2$ -O-) in TPGS.

To monitor the Zeta potential change in drug release process, 0.5 mL of samples at pH5.0 and pH7.4 respectively were collected at desirable time points and diluted to 3 mL for Zeta potential monitoring. The initial samples prepared at t=0 were set as controls.

#### 160 Cell culture conditions and cell lines

161 The cancer cell lines HT-29 (human colon cancer, folate receptor over-expression) and MCF-7 162 (human breast cancer, folate receptor negative expression) were maintained in McCoy's 5a media and 163 Eagle's minimum essential medium respectively. L-929 cells (mouse fibroblast) were cultured in DMEM. 164 All the culture media were supplemented with 10% FBS (fetal bovine serum), 0.03% l-glutamine and 1% 165 penicillin/streptomycin in 5% CO2 at 37°C in a 95% humidified atmosphere.

## *In vitro* cytotoxic evaluation via MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay

168 HT-29, MCF-7 and L-929 cells were harvested in a logarithmic growth phase and seeded on a 169 96-well plate in 100 µL media at a density of 5000 cells/well and incubated overnight. All the drugs were 170 dissolved in culture media and subjected for use. The cells were treated in three ways separately: 1) for 171 testing the compatibility of the blank micelles, the cells were treated with TPGS NPs and FA-TPGS NPs 172 from 15.6 µg/mL to 500 µg/mL; 2) for testing drug loaded micelles, Can-NPs and FA-Can-NPs with a 173 cantharidin concentration from 0.097 µM to 50 µM were used to treat the cells; 3) for folate competing 174 experiments, the cells were pre-treated with 2 mM sodium folate for 4 h and then the culture media was 175 removed, washed by PBS and new culture media was added. After that, FA-Can-NPs with a final 176 cantharidin concentration from 0.097 µM to 50 µM were added to the plates. After treatment of drugs, the 177 cells were transferred to an incubator at 37 °C for further incubation. At the end of the incubation for 48 h 178 at 37 °C, 10 µL of MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) stock solution (5 179 mg/ml) is added to each culture well and further incubated for 4 hours. 100 µL of acidified isopropyl 180 alcohol-sodium dodecyl sulfate (SDS) solution was added to each well and incubated for another 12 h. 181 After that, the plate was read by a micro-plate reader at a wavelength of 570 nm.

#### 182 Intracellular localization and uptake efficiency study

HT-29 cells were seed onto a glass in 6-wells plate at a density of  $1 \times 10^5$  cells/well in 2 ml culture media overnight before use. Then cells were then treated with Can/RhB NPs and FA-Can/RhB NPs at an equal RhB concentration of 2 µg/ml for 1 h. Then the cells were washed twice with cold PBS, and fixed with 4% formaldehyde. Cell nucleus was stained by DAPI (2-(4-amidinophenyl)-6-indolecarbamidine dihydrochloride).Cells were then thoroughly washed by PBS for three times and observed using confocal laser scanning microscope (CLSM, Olympus FV1000).

189 For flow cytometry study, HT-29 cells were seed onto a glass in 6-wells plate at a density of 2 190 X10<sup>5</sup> cells/well in 2 ml culture media overnight before use. Then cells were treated with Can/RhB-NPs, 191 FA-Can/RhB-NPs at an equal RhB concentration of 5 µg/ml. For FA blocking study, the cells were 192 pretreated with 2 mM FA and washed 3 times by cold PBS and then treated with FA-Can/RhB-NPs. For 193 lower temperature controls, cells were treated with the same procedures but the plates were put at 4 °C 194 NaN<sub>3</sub> mediated inhibition of endocytosis, before treatment of the drugs, cells were instead. For 195 pretreated with 20 mM NaN<sub>3</sub> and washed for 3 times by cold PBS prior to use. Each sample was analyzed on a BD FACSCalibur  $^{TM}$  flow cytometer (BD Biosciences). 196

#### 197 **PP2A inhibition assay**

A million of HT-29 cells were seeded in a 6-well plate and incubated with cantharidin, Can-NPs, FA-Can-NPs at an equal cantharidin concentration of 10  $\mu$ M for 6 h. Then the cells were washed for several times, the cystolic contents were extracted for PP2A activity assay. The PP2A activity is the percentage of residual PP2A activity compared with untreated control groups. PP2A inhibition adds PP2A activity equals to 1.

203

#### 204 Results and Discussion

#### 205 Synthesis of Can-TPGS conjugates

Secreted by blister beetle, most notably by the "Spanish fly"[4], cantharidin works by inhibiting protein phosphatases 1 and 2A (PP1, PP2A)[6]. It has been a long time for people to use blister beetle to treat Molluscum contagiosum virus (MCV) infections and associated warts. Nowadays, cantharidin has also shown potent anticancer activities on many types of human cancer cells [4,5].

210 Although cantharidin possesses potent anti-tumor properties, the clinical application of cantharidin is 211 limited till far due to severe side-effects and its highly toxic nature [8]. Therefore, to find a way to reduce 212 its severe side effects becomes urgent. Recently, drug delivery systems (DDS) using liposomes, 213 biodegradable polymers, inorganic nano-materials such as nanotubes, nanocrystals, etc., have drawn 214 particular audience in the scientific world because they have the ability to reduce systemic toxicity and 215 improve tumor-targeting efficiency of therapeutic agents such as drugs, antibodys, proteins, siRNA and 216 imaging agents [27-31]. For this specific reason, we show here to utilize the most widely accepted TPGS 217 as a drug carrier to prepare a Can-TPGS conjugate. To prove the successful synthesis of the Can-TPGS 218 conjugate, <sup>1</sup>HNMR spectra of cantharidin, TPGS and Can-TPGS were collected in Figure 1(a-c). As 219 shown in Figure 1(a), the typical chemical proton shifts of -CH<sub>3</sub> at 1.24 ppm, -O-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH-O- at 1.79 ppm and -CH<sub>2</sub>-CH-O- at 4.73 ppm were assigned. Figure 1(b) collects the <sup>1</sup>HNMR of TPGS. The 220 221 peak at 3.65 ppm could be assigned to the  $-CH_{2-}$  protons of PEO in TPGS. The lower peaks in the 222 aliphatic region belong to various moieties of vitamin E tails. Figure 1(c) collects the <sup>1</sup>HNMR of Can-TPGS. 223 There appears a peak in 4.7 ppm which is typical chemical shift of -CH<sub>2</sub>-CH-O- in cantharidin. By 224 integrating the peak compared to the PEO in TPGS, we can calculate that there is about 0.9 cantharidin 225 per TPGS, which means ca. 90% of the TPGS is end capped with cantharidin and a drug content of 11.5% 226 weight by weight.

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#### 228 Preparation and characterization of TPGS NPs and FA-TPGS NPs

TPGS is an amphiphilic molecule which can self-assemble into micelles in aqueous solution itself [11]. To introduce the targeting ligand folate to the micelles, TPGS and FA-TPGS were co-assembled. The TPGS NPs and FA-TPGS NPs were characterized by DLS and zeta potential analyzer. As shown in Table 1, the mean diameter of TPGS is  $65.7 \pm 2.5$  nm with a polydispersity index (PDI) of 0.125. TPGS NPs showed a zeta potential of -22.1 ± 1.5 mV. For the FA-TPGS NPs, the mean diameter is 72.4 ± 1.9 nm with a PDI of 0.135. Zeta potential of FA-TPGS NPs was -29.2 ± 3.1 mV, slightly lower than that of
 TPGS NPs.

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#### 37 Preparation and characterization of Can-NPs and FA-Can-NPs

238 Cantharidin is conjugated to TPGS to make a Can-TPGS conjugate which can be further 239 assembled into micelles. Due to the hydrophilic nature of the free carboxyl group in the cantharidin 240 molecule, it will stay with PEG as the outer layer and the vitamin E can form the hydrophobic inner core. 241 Similarly, the size and size distribution of Can-NPs and FA-Can-NPs were characterized by DLS. Data 242 were shown in Figure 2a and Figure 2b and Table 1. Can-NPs and FA-Can-NPs had a mean diameter of 243 114.7 ± 1.2 nm and 130.4 ± 3.2 nm with a poly-dispersity index of 0.104 and 0.216 respectively. FA-Can-244 NPs had a slightly larger diameter possibly due to the introducing of folate ligands onto the surface. From 245 the TEM images, spherical structures of both Can-NPs and FA-Can-NPs can be found with diameters of 246 102.5 nm and 110.7 nm respectively (Figure 2c and Figure 2d). The zeta potential of Can-NPs and FA-247 Can-NPs were -35.6 ± 2.4 mV and - 28.4 ± 3.5 mV. The negative charge is considered beneficial for 248 blood circulation.

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#### 250 Drug release profiles of cantharidin and zeta potential change

251 In order to study the drug release profiles of cantharidin from Can-NPs and FA-Can-NPs, we simplify 252 to study only Can-NPs rather than both of Can-NPs and FA-Can-NPs on the assumption that FA could 253 not change the drug release behavior of cantharidin. Here, a dialysis method was utilized for the drug 254 release study and two pH values at pH5.0 and pH7.4 for the dialysis buffer solution were chosen to 255 simulate the drug release in blood and in the tumor cells respectively due to the extensive report that the 256 pH in the tumor cells is around 5.0, while it is 7.4 in the blood [23].Due to release and liberation of the 257 drug molecule cantharidin into the buffered solution, most of the released cantharidin can go through the 258 dialysis bag. At desirable time points, we collected the polymer solution in the dialysis bag and washed 259 the polymer via ultracentrifugation to remove the salt and residual cantharidin released, and then the 260 nanoparticles were collected and lyophilized for <sup>1</sup>HNMR measurement. By simply integrating the peak of 261 canthardin at 4.7 ppm as compared to 3.65 ppm of PEG in TPGS, we can calculate the drug release 262 percentage. In Figure 3a, results showed that cantharidin was released much faster at pH5.0 than at 263 pH7.4. To be more specific, it can be found that at 48 h, 19% cantharidin was released at pH7.4, while 264 this was 48 % at pH5.0. The drug release difference between the two pH values can be explained by the 265 possible hydrolysis of the linkage of the ester bond at two different pH values. The fact that cantharidin is 266 released much faster at pH5.0 is would be beneficial for drug delivery to cancer cells due to the acidic 267 environment in the cancer cells [32].

268

As cantharidin was linked to TPGS with a bare carboxyl group on the end, there are many carboxyl groups on the surfaces of Can-TPGS NPs. Once more and more cantharidin molecules are released, less

271 and less carboxyl groups would be present at the surfaces of Can-TPGS NPs. Therefore, monitoring the 272 Zeta potential during the drug release process may give some additional information to the drug release 273 process. As shown in Figure 3b, the freshly prepared Can-TPGS NPs at pH5.0 and pH7.4 had a zeta 274 potential of -25.3 mV and -39.6 mV respectively. The difference at the initial stage can be attributed to the 275 pH dependence of Zeta potential. At lower pH values (pH5.0), less carboxyl groups can be de-protonated, 276 therefore higher zeta potential could be found (-25.3 mV at pH5.0 vs -39.6 mV at pH7.4). As the 277 incubation time was prolonged, the drugs with carboxyl group were liberated from the surface of the 278 nanoparticles, the zeta potential became higher and higher both at pH5.0 and pH7.4. However, at pH5.0, 279 the Zeta potential increased much greater than that that at pH7.4, indicating more drugs were released at 280 5.0.

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#### 282 In vitro evaluation of TPGS NPs and FA-TPGS NPs

283 TPGSis widely used in pharmaceutical applications. TPGS has shown proven properties to improve 284 bioavailability of poorly absorbed drugs. As a water soluble compound, TPGS is also used as an efficient 285 source of natural vitamin E both for therapeutic purposes and nutrition. To prove that TPGS NPs and FA-286 TPGS NPs are non-toxic to the two cancer cell lines HT-29 and MCF-7, TPGS NPs and FA-TPGS NPs 287 were treated with the cells for 48 h at a concentration ranging from 500 µg/ml to 16.25 µg/ml. As shown in 288 Figure 4a and Figure 4b, we can find that even up to 500 µg/ml, the cell viability for the two cell lines are 289 larger than 90%. And it can be further found that HT-29 and MCF-7 cells displayed no difference in cell 290 viability on the two kinds of nanoparticles, TPGS NPs and FA-TPGS NPs. This can be explained by the 291 fact that the two nanoparticles had no drugs and cells cannot be killed though there is possible targeting 292 effect. From the above results, we can find that TPGS NPs and FA-TPGS NPs are safe enough as drug 293 carriers.

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#### 295 In vitro evaluation of Can-NPs and FA-Can-NPs

296 The anti-cancer therapeutic promise of cantharidin is limited because of its high mammalian toxicity 297 [8]. Recent study has shown that cantharidin displayed considerable toxicity on some colorectal cancer 298 cell lines [7]. To reduce the side effects and further possibly increase the efficacy of cantharidin, here we 299 prepared a Can-TPGS conjugate and self-assembled this conjugate into nanoparticles to deliver 300 cantharidin to the cancer cells. In this way, we believe that the Can-NPs can release the drug cantharidin 301 to kill the cancer cells. Further by targeting the cells with folate ligand, the efficacy of FA-Can-NPs would 302 be higher than that of Can-NPs. To test the targeting effect of folate ligand, here we choose a colorectal 303 cancer cell line, HT-29 and a breast cancer cell line MCF-7 because the former one is reported to have 304 over-expressed folate receptors while the latter one has low-expressed folate receptors [24,25]. Moreover, 305 pretreatment of the cells with 2 mM folate was utilized to block the folate receptors. The results of Can-306 NPs and FA-Can-NPs on two cell lines were shown as in Figure 5a and Figure 5b. In Figure 5a, it was 307 found that cantharidin showed a dose dependent cytotoxicity towards HT-29 cells at a concentration

308 ranging from 0.097 µM to 50 µM to (2-fold dilution). Can-NPs were much more effective than cantharidin 309 almost in all the concentration range. For the targeting nanoparticles, FA-Can-NPs, the cytotoxicity is 310 increased to a greater extent. Pretreatment of the cells with 2 mM folate to block the folate receptors on 311 the cell surfaces can reduce the efficacy of FA-Can-NPs. In this situation, the efficacy of FA-Can-NPs is 312 somewhat the same as non-targeting nanoparticles Can-NPs. The results here demonstrated that the 313 efficacy of the drugs are in the order of FA-Can-NPs > Can-NPs  $\approx$  FA-Can-NPs + FA > cantharidin on 314 HT-29 cells. This can be possibly explained by the targeting effect of the folate ligands and the delivery of 315 drugs to the cancer cells.

To further prove that, MCF-7 cells with low-expression of folate receptors was tested and the results were shown in Figure 5 b. We can find that the efficacy of the drugs was in the order of FA-Can-NPs  $\approx$ FA-Can-NPs + FA  $\approx$  Can-NPs > cantharidin. The results mean that folate targeting ligand has minimum targeting effect on the folate receptor low-expressed cells. However, all the naoparticles are better than the free drug cantharidin, this is possibly due to the intracellular delivery of more drugs to the cells via the nanoparticles.

322 To make it clearer, the IC<sub>50</sub> values of cantharidin, Can-NPs, FA-Can-NPs and FA-Can-NPs blocked 323 by free FA on HT-29 and MCF-7 cells at 48 h were listed in Figure 5c and Figure 5d. As shown in Figure 324 5c, the IC<sub>50</sub> values for cantharidin, Can-NPs and FA-Can-NPs and FA-Can-NPs+ FA (blocked by 2 mM 325 folate) were 15.3, 9.2, 3.6 and 8 µM on HT-29 cells at 48 h. Delivery of cantharidin by nanoparticles (Can-326 NPs) can increase the efficacy of cantharidin by ca. 1.7 fold. Further targeting the nanoparticles can 327 increase its efficacy up to approximately 4.3 fold. Therefore, the extra 2.5 fold increase in efficacy can be 328 attributed to the targeting effect of folate. Further, we can take a look at the MCF-7 cell line as shown in 329 Figure 5d. The IC<sub>50</sub> values for cantharidin, Can-NPs and FA-Can-NPs and FA-Can-NPs+ FA (blocked by 330 2 mM folate) were 32.4, 16.5, 13.1 and 14.7 µM respectively at 48 h. One can clearly find that delivery of 331 cantharidin by nanoparticles (Can-NPs) can increase the efficacy of cantharidin by ca. 1.9 fold. However, 332 further targeting the nanoparticles did not show any obvious increase in its efficacy and blocking the cells 333 with 2 mM folate also demonstrated no effect on the enhancement of drug efficacy. This is possibly due to 334 the low-expression of folate receptors on MCF-7 cells.

Moreover, to further show the benefit of delivery cantharidin, we have chosen a mouse fibroblast cells to test the toxicity of cantharidin and Can-NPs. The results were shown in Fig.5e. We can clearly see that both cantharidin and Can-NPs showed less toxicity on this cell lines than on MCF-7 and HT-20 at the same doses, indicating the preference to kill the cancer cells by the Can-NPs.

#### 339 Intracellular uptake of Can-NPs and FA-Can-NPs

To give some insight into the intracellular uptake of Can-NPs and FA-Can-NPs, first the cells were treated with RhB loaded micelles Can/RhB-NPs and FA-Can/RhB-NPs at an equal RhB concentration of 2 µg/ml. After treatment of them for 1 h, the cells were imaged via confocal laser

343 scanning microscope. The results were shown in Figure 6a. The blue fluorescence comes from DAPI in 344 the cell nucleus. The red fluorescence, which comes from RhB and stands for the nanoparticles, diffused 345 into the whole cells and mainly in the cytosol, indicating that Can-NPs entered the cancer cells. It should 346 be noted that at the same 1 h, cells treated with FA-Can-NPs displayed brighter red fluorescence, 347 suggesting more FA-Can/RhB-NPs were in the cells and existence of possible FA targeting of the cancer 348 cells.

349 Free anticancer drugs such as doxorubicin, paclitaxel, cisplatin as well as catharidin are widely 350 believed to enter the cancer cells via passive diffusion [33]. However, the nanoparticles are extensively 351 reported to be internalized by the cancer cells via a different uptake pathway, via so-called endocytosis, 352 which is an energy dependent process. To explain why the nanoparticle formulation of cantharidin is 353 better than the free drug and FA-Can-NPs internalized more drugs than Can-NPs, here we studied 354 intracellular uptake of the drugs via flow cytometry. To track the nanoparticles in this process, we labeled 355 the Can-NPs and FA-Can-NPs with RhB (Can/RhB-NPs and FA-Can/RhB-NPs). Furthermore, to prove 356 the endocytosis of Can-NPs and FA-Can-NPs, 4 °C and 20 mM NaN<sub>3</sub> (ATP depleting agent) were used to 357 inhibit the energy dependent process [34]. As shown in Figure 6b, taking HT-29 as a representative 358 cancer cell line, we can found that FA-Can/RhB-NPs had more than 1.8-fold uptake of nanoparticles 359 compared to the non-targeting nanoparticles of Can/RhB-NPs (control). Blocking the cells with 2 mM 360 folate can greatly reduce the relative uptake of the Can/RhB-NPs to 1.1. The energy inhibition can greatly 361 affect the uptake as is can be seen from rapid reduction of the relative uptake values. It can be further 362 found that 4°C had shown a more profound effect on uptake than pretreatment of cells with 20 mM 363 NaN<sub>3</sub>. The results above clearly revealed that Can/RhB-NPs were internalized by endocytosis and folate 364 targeting can increase the endocytosis efficiency of cantharidin loaded nanoparticles.

365

#### 366 **PP2A inhibition assay**

367 It is generally believed cantharidin exert its anticancer efficacy via a strong affinity and specificity 368 for protein phosphatase 2A (PP2A)[6]. Moreover, scientist around the world have found that he level of 369 PP2A inhibition parallels its cytotoxicity for cantharidin[6]. To find whether Can-NPs and FA-Can-NPs 370 showed the same PP2A inhibition mechanism. A PP2A inhibition assay was studied on the HT-29 cells by 371 treatment of cantharidin, Can-NPs and FA-Can-NPs at 10 µM for 6 h. As shown in Figure 7, compared to 372 the non-treated cells (control), the PP2A activity of the cells treated with cantharidin, Can-NPs, FA-Can-NPs, FA-Can-NPs + FA were 52 %, 41 %, 25% and 33% respectively. Therefore, the nanoparticles 373 374 loaded with cantharidin have shown PP2A inhibition efficacy. In accordance with the cell viability assay, 375 targeting cantharidin nanoparticles increase the inhibition of PP2A, while blocking the cells with 2 mM 376 folate will reduce the inhibition rate.

- 377
- 378 Conclusion

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Taken together, we have shown here rational design of nanoparticles loaded with cantharidin via a Can-TPGS conjugate for drug delivery. By further introducing the folate targeting ligand, we have shown here folate targeting nanoparticles FA-Can-NPs loaded with cantharidin. The drug conjugate was thoroughly characterized and two kinds of nanoparticles were systematically studied *in vitro*. We have found that folate targeting nanoparticles with cantharidin to kill colorectal cancer via a PP2A dependent way. Further *in vivo* evaluation of this system is on-going and this cantharidin loaded nanoparticles may find the potential use in the future.

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**Scheme 1.** Preparation of the folate targeted nanoparticles loaded with cantharidin. (a) Synthesis of the cantharidin TPGS conjugates (Can-TPGS);(b) Self-assembly of the Can-TPGS conjugates to cantharidin loaded nanoparticles (Can-NPs); (c) Self-assembly of the Can-TPGS conjugates with folate-TPGS (FA-TPGS) to folate targeting nanoparticles (FA-Can-NPs).



Figure 1. 1HNMR spectra of cantharidin (a), TPGS (b) and Can-TPGS conjugate(c)



**Figure 2.** Characterization of cantharidin loaded nanoparticles Can-NPs (a,c) and folate targeting nanoparticles FA-Can-NPs (b,d) by DLS(a,b) and TEM (c,d). Can-NPs showed a mean diameter of 114.7 nm with a polydispersity index of 0.104 by DLS (a) and a mean diameter of 102.5 nm by TEM (c). Similarly, FA-Can-NPs showed a mean diameter of 130.4 nm with a polydispersity index of 0.216 by DLS (b) and a mean diameter of 110.7 nm by TEM (d).



**Figure 3**. Representative drug release of cantharidin from Can-NPs studied by <sup>1</sup>HNMR at pH5.0 and pH7.4. Experimental details were shown in the text. Data were shown as mean value  $\pm$  standard deviation (n=3)(a). To give an insight into the nanoparticle change, the zeta potential of the nanoparticles were monitored during this process and the results were listed in (b).



**Figure 4.** *In vitro* cyototoxicity of TPGS NPs and FA-TPGS NPs on HT-29 (a) and breast cancer MCF-7 (b) cells at 48 h (b). Data were shown as mean value ± standard deviation (n=4).



**Figure 5.** Representative *In vitro* cytotoxicity of cantharidin loaded nanoparticles Can-NPs and folate targeting nanoparticles with cantharidin (FA-Can-NPs) on Breast cells (a) and HT-29 cells (b) at 48 h and representative cytotoxicity of FA-Can-NPs via blocking the folate receptor by pre-incubating the cells with 2 mM folate for 1 h. The IC<sub>50</sub> values of cantharidin, Can-NPs, FA-Can-NPs and FA-Can-NPs blocked by free 2 mM FA on breast cancer MCF-7 (c) and HT-29 cells (d) at 48 h were collected. Data were shown as mean value  $\pm$  standard deviation (n=3). ).\*\*\* indicates p<0.001, \*\* indicates p<0.01. To further show the preference to kill the cancer cells, L-929 cells were treated with cantharidin and Can-NPs, MTT assay was performed (e).



**Figure 6** Intracellular uptakes of Can-NPs by HT-29 cancer cells. Representative confocal laser scanning images of Can/RhB-NPs and FA-Can/RhB-NPs with an equal RhB concentration of 2  $\mu$ g/ml at 1 h (a). After treatment of Can/RhB-NPs and FA-Can/RhB-NPs for 1 h, cells were thoroughly washed and fixed by with 4% formaldehyde. Cell nucleus was stained by DAPI (blue fluorescence) and then observed using confocal laser scanning microscope (CLSM, Olympus FV1000).The red fluorescence comes from RhB. The scale bar is 40  $\mu$ m. The relative uptake study of Can-NPs, FA-Can-NPs and FA-Can-NPs blocked by 2 mM FA at 4 h via flow cytometry was shown in (b). Cells were treated with the drug cantharidin loaded micelles at 1  $\mu$ M. RhB was 5  $\mu$ g/ml. Results were shown as relative to Can-NPs at 37 °C for 4 h. Data were shown as mean value ± standard deviation(n=3). ).\*\*\* indicates p<0.001, \*\* indicates p<0.01.



**Figure 7.** The PP2A activity of HT-29 cells treated with cantharidin, Can-NPs, FA-Can-NPs and FA-Can-NPs blocked by 2 mM FA. The non-treated cells were set as control. Data were shown as mean value  $\pm$  standard deviation (n=3).\*\*\* indicates p<0.001, \*\* indicates p<0.01.

Table 1. Physical parameters of the nanoparticles prepared	

Code	DLS (nm)	PDI	Zeta Potential(mV)
TPGS NPs	65.7 ± 2.5	0.125	-22.1 ± 1.5
FA-TPGS NPs	72.4 ± 1.9	0.135	-29.2 ± 3.1
Can-NPs	114.7 ± 1.2	0.104	-35.6 ± 2.4
FA-Can-NPs	130.4 ± 3.2	0.216	- 28.4 ± 3.5

