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SSSF for ethanol production using N_2 stripping was mainly conducted by the synergy of hydrolytic enzymes and yeast cells.



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1	Solid simultaneous saccharification and fermentation of rice straw for bioethanol production
2	using nitrogen gas stripping
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1 Abstract

2	A solid simultaneous saccharification and fermentation (SSSF) of rice straw for ethanol production
3	using N2 stripping was developed. The effects of N2 flow rate, yeast inoculation volume, and substrate
4	moisture content on SSSF were investigated. The highest total output of ethanol was 4.78 g, and the
5	highest ethanol yield obtained was 56.3% using an N2 flow rate of 30 mL/min, a yeast inoculation
6	volume of 20 mL, and a moisture content 4.6 mL (water)/g (substrate). The low residuals of ethanol
7	and glucose in the substrate demonstrate that the N_2 carrier gas effectively stripped the produced
8	ethanol out of the packed bed, alleviating the inhibitions caused by evolved glucose and ethanol
9	remaining in the bioreactor. With increases in N_2 flow rate or substrate moisture content, the total
10	output of ethanol and ethanol yield initially increased and then decreased. However, with the increase
11	in yeast inoculation volume, both indexes first rose and then remained relatively constant.
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13	Keywords: solid-state fermentation, bioethanol, gas stripping, rice straw, Saccharomyces cerevisiae
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1 **1. Introduction**

2	Renewable energy sources are a popular topic in the world because of increasing demands for energy
3	sources and the decreasing reserves of fossil fuels. Of all the renewable energy sources, bioethanol is
4	an ideal substitute for liquid fuel because it has a high energy content and originates from biomass. ^{1, 2}
5	Currently, bioethanol is mainly produced from grain and creates competition with food stock grains for
6	people and livestock. ^{3, 4} However, ethanol can also be produced from abundant renewable biomass
7	sources such as agricultural and forestry residues. ⁵
8	The saccharification and fermentation processes of cellulosic materials have been mainly separated
9	for bioethanol production. Previous reports of this two-step method indicate that glucose accumulation
10	may suppress the enzymatic hydrolysis of cellulose. ^{6,7} To solve this problem, a method of
11	simultaneous saccharification and fermentation for ethanol production has been developed. ^{8,9} In this
12	process, the reducing sugar derived from the saccharification of cellulose was immediately utilized by
13	yeast cells for ethanol fermentation. Thus, little glucose was accumulated in the reactor and reduced the
14	issue of glucose inhibiting cellulose saccharification. This operation is also more convenient because
15	the two steps (saccharification and anaerobic fermentation) were integrated, and the efficiency of
16	lignocellulose saccharification was improved. Nevertheless, the accumulated ethanol still inhibited the
17	growth and activity of yeast cells in a noncompetitive manner, and the cellulase activity was also
18	decreased, reducing ethanol yield. ^{10, 11} Thus, a novel ethanol stripping method using a carrier gas was
19	developed to carry the generated ethanol out of the bioreactor to be recovered in an absorber. ¹² Zhang
20	et al ¹³ studied effects of different stripping gas on cell physiology and ethanol production during
21	ethanol fermentation, and indicated that common-purity N_2 was the best choice for ethanol

1	fermentation and cell growth. Later research results indicated that both of the efficiencies of
2	saccharification and fermentation were improved using N_2 as carrier gas. ^{9, 11} The ethanol stripping
3	method has many advantages such as simple operation, safe for the yeast culture, requiring low energy
4	input, and inexpensive capital investment for facilities. ¹⁴ Importantly, the two issues of inhibiting
5	saccharification and fermentation can be alleviated simultaneously, enhancing the efficiencies of
6	saccharification and ethanol fermentation overall. ^{15, 16}
7	Traditional simultaneous saccharification and fermentation processes have generally submerged
8	cellulosic materials in liquid. Thus, an increasing cost is incurred as a result of a decreasing substrate
9	density and a large quantity of waste water requiring treatment. ¹¹ Additionally, more energy is required
10	in this process, which becomes another disadvantage. Solid simultaneous saccharification and
11	fermentation (SSSF) can overcome the above problems because of the many benefits such as high
12	substrate concentration and product yield, simple and controllable operation, less effluent wastewater,
13	and less energy consumption. ¹⁷ Thus, SSSF has vast potential in bioethanol production. ¹⁸ Some studies
14	on SSSF for bioethanol production have been shown that the prehydrolysis of cellulosic materials prior
15	to simultaneous saccharification and fermentation had a positive effect on the overall ethanol yield. ^{19,20}
16	Mohanty et al. ²¹ investigated ethanol production using mahula flowers, and the fermentation time
17	achieving peak ethanol concentration in solid-state fermentation was shorter than that in submerged
18	fermentation. These investigations show that ethanol production through the SSSF of cellulosic
19	materials can effectively reduce the production cost, improving the commercial application of
20	bioethanol. However, most of these studies have not considered the inhibiting factors in ethanol
21	fermentation during SSSF. There are few reports on the SSSF of cellulosic materials coupled with

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1	carrier gas stripping of ethanol.
2	In the present study, a packed bed for the SSSF of rice straw coupled with N_2 sparging was designed
3	to improve the saccharification of lignocellulose and bioethanol production. This study aimed to
4	evaluate the performance of SSSF using ethanol gas stripping, and the effects of gas flow rate,
5	inoculation amount of yeast, and substrate moisture content were analyzed.
6	2. Materials and methods
7	2.1 Materials and yeast strain
8	Rice straw collected in the suburb of Chongqing, China was cleaned with tap water and dried at
9	room temperature. The straw was cut into segments 2-4 cm long. These segments were pretreated in
10	1% NaOH solution for 24 h, rinsed with water five times and dried. The treated rice straw was ground
11	up and sieved using a 60-mesh sieve. The rice straw powder was sterilized at 121 $^\circ C$ for 15 min. Yeast
12	strain, Saccharomyces cerevisiae, which has a high activity and high temperature resistance, was
13	purchased from Yichang Angel Yeast Co. Ltd, Hubei Province, China. Prior to inoculation, 2 g of dried
14	yeast powder was mixed with a glucose solution (2 g/L) at the ratio of 1 g to 25 mL and activated at
15	38 °C for 15 min, then at 33 °C for 1.5 h in a water bath, the OD_{630nm} value of the activated yeast
16	solution was 4.05. Cellulase (135 u/mg dw) and cellobiase (≥250 u/g, Novozym188) were purchased
17	from Worthington Biochemical Corporation (New Jersey, US). The culture medium for yeast growth
18	was the following composition: 2 g/L (NH ₄) ₂ SO ₄ , 2 g/L KH ₂ PO ₄ , 0.2 g/L MgSO ₄ , 0.2 g/L CaCl ₂ , and 3
19	g/L of yeast extract. The solution was adjusted to pH 4.8 using acetic acid-sodium acetate buffer
20	solution and monitored using a pH meter.
21	2.2 Packed-bed test system

1	The test system for SSSF of rice straw with N_2 sparge included a packed-bed reactor, an N_2 canister
2	(gas source), a humidifier, a gas flow meter, and an absorber (Fig.1). The cylindrical bioreactor (Ø48
3	mm \times H190 mm) used a working volume of 280 mL and was made of transparent
4	polymethylmethacrylate (5 mm wall thickness). Approximately 54 mL of sterile glass beads (30 mm
5	stacking height) were placed into the bottom of the packed-bed reactor to distribute the carrier gas
6	uniformly. The whole packed-bed bioreactor was fixed in an incubator with a constant temperature of
7	35 ± 2 °C. Before assembling the test system, the components were autoclaved at 121 °C for 15 min.
8	The bioreactor body was sterilized using formalin solution. The assembled test system was cleaned
9	three times with sterile distilled water. During SSSF, the N_2 gas was passed through a humidifier
10	before sparging into the bioreactor. The N_2 flow rate was controlled using a gas valve. The effluent gas
11	from the bioreactor was channeled into an absorber with distilled water to recover the stripped ethanol.
12	Two test systems were employed and run in parallel.
13	2.3 Operational processes
14	A 25 g sterile rice straw powder was mixed with 20 mL acetate buffer solution (pH 4.8) containing 5
15	mL cellobiase and 0.025 g cellulase. The mixture was poured into a conical flask for enzymatic
16	hydrolysis for 24 h at 50 $$ °C in an incubator, and then the activated yeast solution was mixed with the
17	substrate. The inoculated substrate was transferred into the packed-bed bioreactor for SSSF at 35 $$ °C.
18	Nitrogen gas was sparged into the air-tight reactor from the bottom to strip the generated ethanol out of
19	the bioreactor. During SSSF, ethanol concentration in the absorber was detected every 12 hours. At the
20	end of SSSF, distilled water was added into the residual substrate to extract glucose and ethanol, and
21	the supernatant was collected by centrifugation at 9500 \times g to detect the ethanol and glucose

1	concentrations. To avoid ethanol loss in the absorber, whenever the ethanol concentration in the
2	absorber was detected the recovered solution of ethanol was replaced with fresh distilled water. The
3	ethanol concentration in discharged gas from the absorber was also measured to evaluate ethanol loss.
4	2.4 Analytical methods
5	Ethanol concentration in the absorber was detected at an interval of 12 hours using a gas
6	chromatograph (SC-3000, Chongqing, China) equipped with a flame ionization detector and a 1.5 m
7	stainless-steel column packed with porous styrene particles. For operation, N_2 with a 30 mL/min flow
8	rate was used as a carrier gas, H_2 with a 25 mL/min flow rate as a flammable gas, and air with a 150
9	mL/min flow rate as an oxidizing gas. The injector and oven temperatures were set to 160 $^\circ$ C, and the
10	detector temperature was set to 170 °C.
11	The consumption amount of cellulose was calculated by subtracting the weight of the residual rice
12	straw from its initial weight. The substrate was weighed using an electronic analytical balance
13	(Sartorius BP114, Göttingen, Germany). The cellulose content of substrate was determined using Van
14	Soest's method. ²² At the end of SSSF, the glucose content in the residual substrate was measured using
15	the 3, 5-dinitrosalicylic acid method using a UV-Vis spectrophotometer (TU1950, Beijing, China). The
16	pH value of culture medium was detected using a pH meter (Orion 3 Star, Waltham, MA, USA).
17	The recovered ethanol amount refers to the accumulative amount of ethanol stripped out of the
18	bioreactor collected in the absorber and reveals the capacity of N_2 for stripping ethanol out of the
19	packed bed.
20	The total output of ethanol is equal to the sum of the recovered ethanol amount from the absorber
21	and the residual ethanol amount measured from the bioreactor.

1	The theoretical ethanol amount was obtained using the formula:
2	Theoretical ethanol amount (g) = $0.51f \times \text{initial biomass weight} \times 1.111$,
3	where f is the cellulose fraction of dry biomass (g/g) , 0.51 is the conversion factor for glucose to
4	ethanol calculated from the stoichiometry and biochemistry of yeast, and 1.111 is the conversion factor
5	for cellulose to equivalent glucose.
6	The ethanol yield (%) was calculated using the following expression taken from: ²³
7	Ethanol yield $=\frac{\text{total output of ethanol (g)}}{\text{theoretical ethanol amount (g)}} \times 100\%.$
8	3. Experimental results
9	3.1 Effect of gas flow rate
10	The flow rate of N_2 as the carrier gas was set at 10, 20, 30, and 40 mL/min, respectively, and 25 mL
11	of activated yeast solution was inoculated into the substrate. The temperature for the SSSF was
12	controlled at 35 °C, and the results are shown in Fig. 2. The recovered ethanol amount in the absorber
13	gradually increased over time at a fixed gas flow rate (Fig. 2 a), demonstrating that N_2 sparging
14	stripped the evolved ethanol out of the bioreactor. After 96 hours of inoculation, the recovered ethanol
15	amount increased slower than that after 12 h. The decrease in the performance of SSSF can be ascribed
16	to the reducing activities of enzymes and yeast cells as a result of the production of intermediates. ²⁴
17	Fig. 2 b shows that with an increase in N_2 flow rate from 10 to 30 mL/min, the total output of ethanol
18	increased from 1.57 to 2.09 g. However, when the gas flow rate was increased to 40 mL/min, the total
19	output of ethanol was 1.93 g. Thus, the highest total output of ethanol, 2.09 g, was obtained using an N_2
20	flow rate of 30 mL/min. The increase in N_2 flow rate dispersed more ethanol into the gas phase from
21	the fermented substrate, then the ethanol was stripped out of the bioreactor. Consequently, the total

1	ethanol output increased with increasing N_2 flow rate. However, an N_2 flow rate of 40 mL/min caused a
2	decrease in substrate humidity because of water evaporating in the substrate, resulting in a decrease of
3	total ethanol output from that of 30 mL/min.
4	The residual ethanol amount and residual glucose amount are shown in Fig. 2 c. When the gas flow
5	rate was increased from 10 to 30 mL/min, the ethanol residue in substrate decreased from 0.40 to 0.10
6	g, and plateaued using a flow rate of 40 mL/min to a value of 0.08 g. This result further demonstrates
7	that using N_2 as a carrier gas stripped the evolved ethanol from the SSSF of rice straw out of the
8	bioreactor. Furthermore, the residual glucose amount in the fermented substrate was maintained within
9	the range of 0.83–0.85 g for the different N_2 flow rates studied. Low concentrations of ethanol and
10	glucose remained in the fermented substrate, suggesting that the SSSF of rice straw with N_2 sparge
11	alleviated the issues of ethanol inhibiting glycolysis and glucose inhibiting enzymatic hydrolysis.
12	As depicted in Fig.2 d, the substrate consumption was initially increased slightly and then slightly
13	decreased with increasing N_2 flow rates. The substrate consumption amount initially increased from
14	10.5 to 11.8 g with the increase in N_2 flow rate from 10 to 20 mL/min and then decreased to 10.9 g
15	with an N_2 flow rate of 40 mL/min. The ethanol yield initially increased from 10 to 30 mL/min and
16	then decreased slightly for 40 mL/min. The highest ethanol yield obtained was 25.7% at 30 mL/min,
17	which was equivalent to producing 0.19 g of ethanol per gram of initial substrate. Therefore, it was
18	considered that the optimal N_2 flow rate for SSSF was 30 mL/min in this experiment.
19	3.2 Effect of yeast inoculation amount
20	The yeast inoculation amount has a large influence on the fermentation process. Here, the yeast
21	inoculation amount on SSSF were set at 10, 15, 20, and 25 mL. To maintain the same moisture content

1	of the substrate in each run, the yeast solution was made up to a total volume of 25 mL using sterile
2	distilled water. During SSSF, the flow rate of N2 was controlled at 30 mL/min, and the other operating
3	parameters were the same as those given above.
4	The recovered ethanol amount in all four runs gradually increased with time (Fig. 3 a). The yeast
5	inoculation amount affected the recovered ethanol amount by first showing step increases from 1.46 to
6	2.17 g from 10 to 20 mL of inoculation amount and maintained relatively similar values for 25 mL. For
7	example, 108 h after inoculation, the recovered ethanol increased for the yeast inoculation amount
8	rising from 10 to 20 mL, and then slightly increased to 2.19 g for 25 mL. This result also showed that
9	by increasing the yeast inoculation amount from 10 to 20 mL, the total output of ethanol gradually
10	increased from 1.64 to 2.34 g, and then with an inoculation amount of 25 mL, the total output of
11	ethanol rose slightly to 2.36 g (Fig 2 b).
12	The residual ethanol amount in the fermented substrate using different yeast inoculation amounts
13	were generally maintained between 0.17 and 0.18 g (Fig. 3 c). This result demonstrates that the
14	inoculation amount only slightly affected the residual ethanol amount in the substrate during SSSF
15	coupled with N2 sparge. Furthermore, residual glucose in the fermented substrate was maintained
16	between 0.79 and 0.81 g, confirming that the accumulation of glucose, which inhibits enzymatic
17	hydrolysis, was almost eliminated during SSSF. The substrate consumption amount and ethanol yield
18	for 20 mL yeast inoculation amount was 10.95 g and 27.47%, respectively. For 25 mL, the
19	consumption and yield was slightly higher at 11.02 g and 27.79%, respectively (Fig. 3 d). The highest
20	ethanol yield recorded of 27.79% indicates that 0.21 g ethanol can be produced for one gram of
21	substrate in this experiment.

1	The above results indicate that with the increase in yeast inoculation amount to 20 mL, more yeast
2	cells utilize the monosaccharides derived from the saccharification of rice straw to produce ethanol by
3	anaerobic fermentation. Correspondingly, the hydrolysis of rice straw was partly enhanced because the
4	problem of evolved monosaccharides inhibiting saccharification was alleviated in the process of SSSF.
5	Consequently, the recovered ethanol amount, total output of ethanol and ethanol yield increased.
6	However, a further increase of the yeast inoculation amount to 25 mL produced an ethanol yield rise of
7	only 0.32% higher than that of 20 mL. The saccharification of rice straw did not provide any more
8	monosaccharides for ethanol fermentation by more yeast cells because of the limited enzyme activities.
9	Therefore, in this experiment although both peaks of total output of ethanol and ethanol yield of 2.36 g
10	and 27.79%, respectively, were obtained using 25 mL yeast inoculation amount, the optimal yeast
11	inoculation amount was considered to be 20 mL.
12	3.3 Effect of substrate moisture content
13	In the process of solid-state fermentation, moisture content of the substrate affects the efficiencies of
14	saccharification and ethanol production. ^{25, 26} Here, the substrate moisture content was fixed at 3.6, 4.6,
15	5.6, and 6.6 mL (water)/g (substrate), respectively. The gas flow rate in the experiment was controlled
16	at 30 mL/min, and the yeast inoculation amount was fixed at 20 mL. The other experiment conditions
17	were the same as those given above.
18	The recovered ethanol amounts in the four runs increased gradually over time, and the rate of
19	increase became slower after 84 h (Fig. 4 a). Moreover, by the end of the experiment, the recovered
20	ethanol amount increased from a moisture content of 3.6 to 4.6 mL/g and then decreased with further
21	increases in moisture content. Correspondingly, the total output of ethanol increased from 3.42 to 4.78

1	g with the increase in moisture content from 3.6 to 4.6 mL/g and then decreased to 3.64 g with further
2	increases in moisture content up to 6.6 mL/g (Fig. 4 b).
3	Fig. 4 c shows that with the increase in moisture content from 3.6 to 6.6 mL/g, the residual ethanol
4	amount in the substrate decreased gradually from 0.69 to 0.48 g. Furthermore, the substrate
5	consumption amount decreased from 11.80 to 10.18 g. The ethanol yield first increased to 56.3% with
6	an initial increase in moisture content to 4.6 mL/g and then dropped to 42.8% with further increases in
7	moisture content up to 6.6 mL/g (Fig. 4 d).
8	The highest total output of ethanol and ethanol yield, 4.78 g and 56.3%, respectively, were obtained
9	using a moisture content of 4.6 mL/g with an N_2 flow rate 30 mL/min and a yeast inoculation amount
10	20 mL. This result corresponds to generating 0.46 g of ethanol for every one gram of rice straw.
11	4 Discussion
12	To alleviate the problems of glucose inhibiting saccharification and ethanol inhibiting anaerobic
13	fermentation in the conversion process of cellulosic materials to ethanol, SSSF with carrier gas
14	sparging was used. The effects of gas flow rate, inoculation amount, and substrate moisture content on
15	the performance of SSSF were investigated using rice straw as the substrate. The results indicate that
16	the residual ethanol and glucose in the bioreactor were low throughout. The highest residual ethanol
17	amount found was only 0.69 g in the fermented substrate, corresponding to 0.05 g ethanol per gram of
18	dried residual rice straw. The highest ethanol yield obtained was 56.3%, generating 0.46 g of ethanol
19	per gram of initial substrate, which is higher than those results in other solid-state fermentation
20	studies. ^{25, 27, 28} All these results demonstrate the improved performance of the SSSF system.
21	The mass transfer of ethanol in the SSSF test system was divided into two steps. First, the evolved

1	ethanol dispersed from the solid substrate to the gas phase in the bioreactor. Thus, the ethanol
2	concentration in the gas phase in the bioreactor gradually increased. A high gas flow rate enhanced
3	ethanol mass transfer from the solid substrate to the gas phase because of an increasing concentration
4	difference of ethanol at the gas-solid interface. Therefore, the recovered ethanol amount, total ethanol
5	output and ethanol yield were increased for higher flow rates. However, further increases in gas flow
6	rate above 30 mL/min reduced the recovered ethanol amount. In this case, the high gas flow rate led to
7	a reducing substrate humidity as more water in substrate was stripped out of the bioreactor by the
8	carrier gas, even though the carrier gas was humidified before entering the bioreactor.
9	The second step of ethanol transfer was recovery in the absorber. When ethanol concentration in the
10	absorber is high, the absorption efficiency of ethanol may decrease because the carrier gas may strip
11	ethanol out of the absorber and reduce recovery. Therefore, to avoid losing ethanol, the solution used
12	for ethanol recovery was replaced with fresh distilled water just after ethanol was detected in the
13	absorber. We also evaluated the ethanol loss caused by discharged gas from the absorber and found that
14	the ethanol concentration was low. The highest ethanol loss was about 0.21 g during the SSSF, which
15	accounted for only 4.4% of the total output of ethanol.
16	During SSSF, yeast cells adsorbed on the substrate surface, diffused into the substrate for growth,
17	and formed a biofilm on the surface of the fermented substrate. ²⁹ The formation of a bacterial biofilm
18	on the substrate surface comprises three stages: adsorption, growth, and maturity. ³⁰ Therefore, the
19	growth phase of yeast biofilm was reflected in the recovered ethanol amount. For instance, as shown in
20	Fig. 2 a, 12 h after inoculation, the recovered ethanol amount was low and accumulated slowly over
21	time at a fixed inoculation amount, which indicates that the yeast biofilm was still in the adsorption

1	stage. Between 84 h and 96 h, the recovered ethanol amount was rising rapidly, which indicates that the
2	yeast biofilm was in the growth stage. Consequently, the biomass of yeast cells increases rapidly in the
3	packing layer, and increased the amount of ethanol generated during the SSSF of rice straw. Then, the
4	increment of the recovered ethanol amount was mitigated and shows that the yeast biofilm was in the
5	maturity stage. However, the intermediates from the enzymatic hydrolysis and ethanol fermentation
6	processes may negatively affect the enzyme activities, which may be another reason that caused the
7	recovered ethanol amount to decrease over time. ³¹ Thus, in conclusion, SSSF relies on the synergy of
8	hydrolytic enzymes and yeast cells. Increasing enzymes loading or yeast inoculation amount cannot
9	enhance the ethanol yield during SSSF. Hence, the ethanol yield did not increase with further rises of
10	yeast inoculation amount in this study (Fig. 3 d).
11	Cellulases catalyzing cellulosic hydrolysis adsorb on the surface of the substrate during solid-state
12	fermentation. ³² A low substrate moisture content affects the adsorption of cellulosic enzymes on the
13	substrate surface and suppresses enzymatic hydrolysis. ³³ Moreover, the growth and metabolism of
14	yeast cells were affected at a low moisture content of substrate because low water levels affect activity.
15	Consequently, a low total output of ethanol and ethanol yield were observed. However, with the
16	increase in moisture content, the adsorption and activities of hydrolytic enzymes were significantly
17	improved, and the growth and activity of yeast cells were enhanced. Therefore, obtaining the highest
18	total ethanol output and ethanol yield (Fig. 4). However, with further increases in moisture content, the
19	adsorption of enzymes on the substrate surface was reduced because of a diluted enzyme concentration.
20	Thus, the resistance to mass transfer was increased because of an increasing thickness of liquid film on
21	the surface of the fermented substrate. ³⁴ These factors caused the total output of ethanol to decrease. In

1	the present study, a significant agglomeration of rice straw using the moisture contents of 5.6 and 6.6
2	mL/g was observed. The high moisture content led to a reduced porosity of the packing layer and an
3	increase in the pressure drop of the bioreactor. ³³ Thus, microbial growth was affected. ³⁵ These effects
4	are another contributing reason as to why the total output of ethanol and ethanol yield decreased with
5	further increases in substrate moisture content during SSSF.
6	To summarize, the SSSF of lignocellulose coupled with the gas stripping of ethanol is a complicated
7	process that can be influenced by many factors such as the activities and loadings of cellulase and yeast
8	cells, the type and pretreatment methods of cellulosic material, the porosity of the packed bed, and the
9	carrier gas flow rate. Although a high ethanol yield of 56.3% in this study was achieved, which was
10	lower than those found in some literatures (Table 1). As shown above the synergy of hydrolytic
11	enzymes, yeast cells, and carrier gas is important to the operation of an SSSF system. Therefore, more
12	research into the enzymolysis mechanism of cellulose and the interaction of multiple factors will
13	further improve the performance of SSSF.
14	4 Conclusions
15	The effects of gas flow rate, yeast inoculation amount, and substrate moisture content on SSSF
16	employing the gas stripping of ethanol were investigated. The low residuals of ethanol and glucose in
17	the substrate demonstrate that the carrier gas effectively strips the evolved ethanol out of the bioreactor
18	during SSSF, alleviating the issues of the produced glucose and ethanol inhibiting enzymatic
19	hydrolysis and fermentation. With increases in gas flow rate and moisture content, the total outputs of
20	ethanol and ethanol yields initially increase and then decrease, whereas with increases in yeast
21	inoculation amount, the total output of ethanol and ethanol yield initially increase and then remain

1	relatively constant. The results reveal that SSSF was mainly conducted by the synergy of hydrolytic			
2	enzymes and yeast cells.			
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7	Reference			
8	1. Sarkar N, Ghosh SK, Bannerjee S, Aikat K. Renew Energy 2012, 37, 19-27.			
9	2. Balat M, Balat H, Öz C. Prog Energy Combust Sci 2008, 34, 551-573.			
10	3. Bai FW, Anderson WA, Moo-Young M. <i>Biotechnol Adv</i> 2008, 26 , 89-105.			
11	4. Wang LJ, Luo ZL, Shahbazi A. <i>Ind Crop Prod</i> 2013, 42 , 280-291.			
12	5. Zaldivar J, Nielsen J, Olsson L. App Microbiol Biotechnol 2001, 56, 17-34.			
13	6. Linde M, Galbe M, Zacchi G. <i>Enzyme Microb Technol</i> 2007, 40 , 1100-1107.			
14	7. Öhgrena K, Burab R, Lesnickic G, Saddlerb J, Zacchi G. Process Biochem 2007, 42, 834-849.			
15	8. Podkaminer KK, Kenealy WR, Herring CD, Hogsett DA, Lynd LR. Biotechnology for Biofuels			
16	2012, 5 , 43-51.			
17	9. Taylor F, Marquez MA, Johnston DB, Goldberg NM, Hicks KB. <i>Bioresour Technol</i> 2010,			
18	101(12) , 4403-4408.			
19	10. Ghose TK, Tyagi RD. Biotechnol Bioeng 1979, 21, 1401-1420.			
20	11. Chen HZ, Liu ZH, Dai SH. Biotechnology for Biofuels 2014,7, 53-65.			
21	12. Walsh PK, Liu CP, Findley M E. Biotech Bioeng Symp 1983, 13(5), 629-647.			

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1	13.	Zhang J, Liu HJ, Liu DH. The Chinese Journal of process Engineering 2005, 3, 349-352.
2	14.	Xue C, Zhao J, Liu F, Lu C, Yang ST, Bai FW. Bioresour Technol 2013, 135, 396-402.
3	15.	Sato K, Nakamura K, Sato S. Biotechnol Bioeng 1985, 27, 1312-1319.
4	16.	Liu HS, Hsien-Wen H. Chem Eng Sci 1990, 45(5), 1289-1299.
5	17.	Pandey A. <i>Biochem Eng J</i> 2003, 13 , 81-84.
6	18.	Chen HZ, Qiu WH. Progress in Chemistry 2007, 19(7/8), 1116-1121.
7	19.	Zhao J, Xia LM. Fuel Process. Technol 2009, 90(10), 1193-1197.
8	20.	Vincent M, Pometto III AL, van Leeuwen J H. Bioresour Technol 2014, 158, 1-6.
9	21.	Mohanty SK, Behera S, Swain MR, Ray RC. Appl Energy 2009, 86(5), 640-644.
10	22.	Van Soest PJ. J Ass Offic Agr Chem 1963, 46, 829-835.
11	23.	Zhang J, Chu D, Huang J, Yu Z, Dai G, Bao J. Biotechnol Bioeng 2010, 105(4), 718-728.
12	24.	Krishna C. Crit Rev Biotechnol 2005, 25(1-2), 1-30.
13	25.	Yu J, Zhang X, Tan T. Fuel Process Technol 2008, 89, 1056-1059.
14	26.	Lenz J, Höfer M, Krasenbrink J-B, Hölker U. Appl Microbiol Biotechnol 2004, 65, 9-17.
15	27.	Shi J, Sharma-Shivappa RR, Chinn M, Howell N. Biomass Bioenerg 2009, 33(1), 88–96.
16	28.	Wang R, Ji Y, Melikoglu M, Koutinas A, Webb C. Process Saf Environ 2007, 85(5), 404–412.
17	29.	Yu J, Zhang X, Tian T. J Biotechnol 2007, 129 , 415-420.
18	30.	Garrett TR, Bhakoo M, Zhang Z. Progress in Natural Science 2008, 18, 1049-1056.
19	31.	Chen R, Wang YZ, Liao Q, Zhu X, Xu TF. BMB Rep 2013, 46 (5), 244-251.
20	32.	Matano Y, Hasunuma T, Kondo A. Appl Microbiol Biotechnol 2013, 97, 2231–2237.
21	33.	Couto SR, Sanrom án MÁ. <i>J Food Eng</i> 2006, 76 , 291-302.

1	34.	Auria R, Ortiz I, Villegas E, Revah S. Process Biochem 1995, 30(8), 751-756.
2	35.	Pandey A, Soccol CR, Mitchell D. Process Biochem 2000, 35, 1153-1169.
3	36.	Cha Y L, An GH, Yang J, Moon YH, Yu GD, Ahn JW. Renewable Energy 2015, 80, 259-265.
4	37.	Ahmed IN, Nguyen PLT, Huynh LH, Ismadji S, Ju YH. Bioresource Technology 2013,136, 213-
5		221.
6	38.	Manzanares P,Negro MJ, Oliva JM, S & F, Ballesteros I, Ballesteros M. J Chem Technol
7		Biotechnol 2011,86, 881–887.
8	39.	Gon çalves FA, Ruiz H A, Nogueira CC, Santos ES dos, Teixeira J A, Macedo GR. Fuel 2014,131,
9		66–76.
10	40.	Kumagai A, Kawamura S, Lee SH, Endo T, Jr MR, Mielenz JR. Bioresour. Technol 2014, 162,
11		89–95.
12	41.	Narra M, James P., Balasubramanian V. Bioresour Technol 2015, 179, 331–338.
13	42.	L ópez-Linares JC, Romero I, Cara C, Ruiz E, Castro E, Moya M. J Chem Technol Biotechnol
14		2014, 89 , 104–110.
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Figure Captions Fig.1 Experimental system: 1) nitrogen canister, 2) humidifier, 3) flow meter, 4) packed-bed bioreactor, 5) pH meter and thermometer, 6) ethanol absorber, 7) computer, and 8) incubator. Fig. 2 Effect of gas flow rate on SSSF Fig.3 Effect of yeast inoculation amount on SSSF Fig. 4 Effect of substrate moisture content on SSSF

1 Table caption

2

3 Table 1 Comparison of conversion efficiency of cellulose to ethanol





Fig. 2 (b) Recovered ethanol amount (c) 2.5 2.0 1.5 2.0 2.0 2.0 2.5 2.4 а Total output of ethanol (g)7.27.27.381.41.51.61.71.6 b 1.5 0.0 1.4 48 60 72 Time (h) 12 24 36 84 96 108 120 0 10 20 30 40 Flow rate of gas (mL/min) 0.45 0.95 13 Substrate consumption amount (g) 34 Residual ethanol amount (g) 0.35 0.25 0.10 0.15 0.10 0.90 (d) 0.90 (d) 0.90 (e) 0.9 d С 12 32 11 30 10 9 8 7 20 6 0.05 18 5 0.50 20 30 Flow rate of gas (mL/min) 10 20 30 40 40 10 Flow rate of gas (mL/min)





Fig. 4 5.0 5.2 5.0 (**b** 4.5 а b Total output of ethanol (g) 3.6 mL water/g substrate
4.6 mL water/g substrate
5.6 mL water/g substrate 4.8 4.03.53.02.51.51.51.00.5 4.6 - 6.6 mL water/g substrate 4.4 4.2 4.0 3.8 3.6 3.4 3.2 0.0 3.0 12 24 48 60 72 96 3.6 4.2 4.8 5.4 6.0 6.6 36 84 108 120 0 Time (h) Moisture content (mL water/g substrate) 0.75 Substrate consumption amount (g) 6 01 11 71 91 91 60 Residual ethanol amount (g) 0.60 0.60 0.55 0.40 0.45 d С 50 10 0.35 0.30 8 3.6 4.2 4.8 5.4 6.0 6.6 Moisture content (mL water/g substrate) A.6 4.2 4.8 5.4 6.0 6. Moisture content (mL water/g substrate) 3.6 6.6 3.6

Table 1 Comparis	on of conversio	n efficiency of	cellulose to ethanol
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			The highest	The corresponding	
Substrate	Pretreatment conditions	Fermentation conditions	theoretical ethanol	Ethanol content	Ref
			yield (%)	(g/L)	
Mr. d	1.5 M NaOH with stirring at 120 rpm	Liquid-state saccharification and fermentation at 42 $$ ${\rm C}$	0.6.204	29.5	36
Miscantnus	and heated to 150 °C for 30 min	with shaking at 150 rpm	86.3%		
		anaerobic condition in an orbital shaker (150 rpm,37 °C)		43.7	37
Paper bark tree	Subcritical water at 180 °C for 30min	for 120 h	91.25%		
a .	Semi-continuous liquid-state simultaneous			10.4	
Corn stover	Steam explosion at 200 °C for 4min	saccharification and fermentation(37 °C for 60 h)	52.1%	40.6	23
Olive tree	liquid hot water pretreated at 210 °C	Liquid-state simultaneous saccharification and	2004	24.0	20
pruning	with magnetic agitation	fermentation at 35 °C for 72 h and 150 rpm	38%	24.9	38
	sequential alkaline hydrogen peroxide		89.15	9.32	39
mature coconut	(Alk-H2O2)–sodium hydroxide	Semi-simultaneous saccharification and fermentation at 30			
fibre	(NaOH)	C for 40 h			
	steam treatment (150 °C for 2 h,) with	yeast-based simultaneous saccharification and	63.4% (calculated		10
Hinoki cypress	wet disk milling	fermentation at 58 °C with shaking at 125 rpm	value)		40
	dilute acid pretreatment, then				
Rice straw	delignification with 0.5% NaOH at	Simultaneous saccharification and fermentation with	84.6%	24.63	41
	121 °C for 30 min	agitation at 120 rpm for 72 h at 42 °C			
	liquid hot water pretreatment at 217 $^{\circ}$	Liquid-state simultaneous saccharification and			10
Rapeseed straw	for 42 min)	fermentation in an orbital shaker at 150 rpm.	66.6%	17.2	42
		Solid-state Simultaneous saccharification and			
Rice straw	1% NaOH solution for 24 h	fermentation at 35 °C for 108 h	56.3%	Equivalent to 21.05	This work