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Cite this: DOI: 10.1039/c0xx00000x

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Communication

Successful Synthesis of Distinct Dendritic Unimolecular Initiators Suitable for Topologically Attractive Star Polymers.

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Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX 5 DOI: 10.1039/b000000x

Unimolecular initiator end-capped and anchored dendrimers with initiator functionality up to 48 of G4 dendrimer were prepared quantitatively in a single-step reaction. Polymerization of styrene using these initiators yielded multi 10 arm star polymers with mol. weight (M_w) in the order of 10^5 g/mol and polydispersity of arm in the range of 1.01 to 1.09.

In the recent past, major developments in the field of polymer science have resulted from macromolecular engineering. The development of polymers with controlled molecular architecture, 15 complexity and attractive star-like topology (for example, dendritic-linear hybrid multiarm polymers with controlled molecular architecture) is topical in research¹. An executable

protocol based on dendritic growth strategy and living polymerization mechanism could produce such kind of materials. ²⁰ However, the intriguing challenge in achieving this task is to make available suitable multifunctional core molecules.

Many kinds of core molecules have been used for synthesis of star (or) multiarm polymers^{2,3}. Hawker⁴ reported first example of nitroxide based multi-functional initiator for synthesis of three

²⁵ arm star polymers. Subsequently, Frechet^{5a} group described TEMPO functionalized polystyrene for the synthesis of branched and highly branched polymers in a tandem approach and also used a styrenyl-TEMPO monomer which underwent self-condensing living free radical polymerization leading to star
³⁰ polymers^{5b}. Latter, a mesitylene⁶, a cyclodextrin⁷, a porphyrin⁸, silsesquioxane and an octafunctional derivative of resorceinarene

based NMP initiators have been reported^{9,10}. In one report, Miura and co-worker¹¹ used second generation dendritic macro initiators with 12 TEMPO moieties for the synthesis of star and star di-³⁵ block copolymers. Dendrimers functionalized for living polymerization process are meagre^{4,5,11}, and dendrimer functionalized with styrenyl-TEMPO moiety suitable for nitroxide-mediated living radical polymerization leading to multi arm star polymers with number of arm more than 12 is not yet

40 reported.

Herein, we report TEMPO based dendritic unimolecular initiators of two different types having number of initiators up to 48 for the first time. Indeed, these distinct initiators will give similar kind of polymers, but with different topological features

⁴⁵ that could not be expected from other multi-functional initiators reported so far. Moreover, they can be used to produce welldefined multiarm star polymers at nanoscale with pre-determined

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number of arms at higher order.

In this work we used two different styrenyl-TEMPO reagents ⁵⁰ (unimolecular initiators). The unimolecular initiator **1** suitable for end-capping reaction with dendrimers was prepared using a reported procedure¹² and the same **2** suitable for anchoring reaction was prepared according to the Scheme S1 (*see Supporting Information*). The structure and purity of the newly ⁵⁵ prepared styrenyl-TEMPO reagent **2** were confirmed using FT-

⁵⁵ prepared styler PLNIO reagent 2 were commised using PT-IR, ¹H-NMR,¹³C-NMR and HR-MS data and nitrogen content estimation. The reason for end-capping and anchoring of unimolecular initiator on dendrimers is to use such initiators in future for the synthesis of multi arm star polymers with distinct ⁶⁰ topological futures.

The concept of blocked isocyanate chemistry was adapted for end-capping and anchoring of unimolecular initiator on dendrimers. Blocked isocyanate is an adduct usually obtained from an aromatic isocyanate and a phenol. The –NHCOO- group 65 of a blocked isocyanate is thermally weak and at elevated temperature it de-blocks into isocyanate (-NCO) and blocking agent (HO-). Then, the regenerated isocyanate could react with a co-reactant containing aliphatic hydroxyl functional group to form urethane with thermally more stable bonds¹³. The overall 70 reaction can be seen as:



The easy-to-do reaction involving blocked isocyanate group and ⁸⁰ hydroxyl co-reactant of present work is described in Scheme 1¹⁴. The substituent on phenol (blocking agent) can influence seriously the de-blocking temperature; the electron withdrawing substituent reduces the de-blocking temperature. Blocked isocyanates with relatively low de-blocking temperature find ⁸⁵ applications in many chemical processes. In this work, the blocking agent already derived at the surface groups of dendrimer is *p*-chlorophenol which could cleave of easily below 90°C¹⁵, thus the urethane interchange reaction described in Scheme 1 proceeded smoothly at 90°C in DMF in the presence of dibutyltin ⁹⁰ dilaurate catalyst. The styrenyl-TEMPO end-capped and 30

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anchored dendrimers (**3-6** and **7-10**) do not require column purification and it is worth mentioning here that these initiators are found to be highly soluble in 10:1 mixture of styrene and DMF.



Scheme 1 Synthesis of dendritic unimolecular initiators.



Fig. 1 Structure of G4 dendritic unimolecular initiators.

¹⁰ The ¹H-NMR and MALDI-TOF MS spectra of these novel initiators of G4 dendrimer are given in Figure 2 and 3 respectively. The ¹H-NMR spectra show sharp and distinct signals arising from dendritic and reagent part. Moreover, the integration values of hundreds of protons of different types ¹⁵ present in the large globular molecules agree well with the calculated number of protons. One important change observed in the ¹H-NMR spectra that could be treated as confirmation for quantitative conversion into multifunctional initiators is the complete disappearance of a peak at 10.4 ppm corresponding to ²⁰ urethane –NH protons of blocked isocyanate moiety of parent dendrimer. After completion of urethane interchange reaction, the discrepancy in the urethane –NH protons had vanished and all the urethane –NH protons became uniform; thus, they give only one peak at 9.8 ppm. In the MALDI-TOF MS spectra of styrenyl-²⁵ TEMPO end- capped and anchored G4 polyurethane dendrimer,









no peak was observed even with very trace intensity for any undesired products, and only one peak was observed with 100% intensity at m/z 22049.5 and 27814.6 respectively exactly corresponding to $[M+Na]^+$ of respective molecules. The labile

- ⁵ bond formed between styrene and TEMPO moiety was found stable under MALDI-TOF experimental condition and this is not surprising because similar observations have been reported for many unimolecular initiators¹⁶. These analytical examinations confirm that the targeted innovative initiators are structurally
- ¹⁰ pure. Analysis of ¹H-NMR and MALDI-TOF MS spectra (*See Supporting information*) of both such type of initiators based on G1-G3 dendrimers reveals that they also structurally pure.
- As we obtained remarkably good ¹H NMR and MALDI-TOF MS spectra which confirm the defect free structure of dendritic ¹⁵ unimolecular initiators, it was decided to support these data. Size exclusion chromatography combined with multi angle laser light scattering (SEC-MALLS) technique is an ultimate tool to be used to determine the absolute molar mass and polydispersity index (PDI) of the dendrimers. Elution of all the dendritic initiators in ²⁰ the SEC-MALLS resulted narrow disperse signals (Fig. 4) with
- estimated molecular weight close to calculated molar mass (Table 1). Specific refractive index increment (dn/dc) for each



Fig. 4 SEC-MALLS Chromatograms of (a) unimolecular initiator end-40 capped dendrimers and (b) unimolecular initiator anchored dendrimers. dendrimer was determined separately in the same solvent used for SEC experiments and used for molecular weight calculation. The dn/dc value of dendrimer was found decreased with increasing the generation up to G3 and then increased for G4 dendrimer and ⁴⁵ this trend was found to be common for both types of dendritic initiators. The molecular weight data and unimodel chromatograms with narrow distribution clearly support the ¹H NMR and MALDI-TOF MS spectra of the dendrimers.

The reagents used for end-capping and anchoring process are 50 dormant species in the TEMPO-mediated polymerization and its effectiveness in molecular weight build-up was confirmed beyond any doubt with variety of monomers¹⁷. Thus, the multifunctional unimolecular initiators reported in this communication could be readily used for the synthesis of 55 dendritic-linear hybrid multiarm polymers. Preliminary work on this line was carried out using the initiator 5 and 6 with an intention to confirm the efficiency of the initiators on tune and controll the arms. Polymerization of styrene carried out at 125°C varying the experimental conditions like monomer/initiator ratio 60 and reaction time yielded high molecular weight multiarm star polymers, the molecular weight (M_w, SEC-MALLS) was in the order of 10⁵ g/mol. The polydispersity index of multiarm star polymers were found to be in the range of 1.37 to 1.39, however the PDI of arm (after hydrolysis of inner polyurethane core) was $_{65}$ highly narrow, in the range of 1.01 - 1.09. This finding i.e., difference in PDI of parent polymers and their hydrolysed into linear chains is consistent with a reported work^{4, 5b, 11, 18}. All the chromatograms were unimodal and are given in the Supporting Information along with molecular weight data. Experimentally 70 calculated number of arms were found agree well with the theoretically calculated values and the results confirm that the number of arms of star polymer can be pre-determined, tuned and controlled accurately using these initiators. Polymerization studies in detail including dependence of molecular size of star 75 polymers and their dn/dc and characterization for topological features at nanoscale would be reported following this communication.

In conclusion, two series of dendritic unimolecular initiators up to fourth generation were prepared in quantitative yield. As these initiators were designed with an intention to produce multiarm star polymers with pre-determined number of arms, their defect-free structures have been established beyond doubt. Use of these initiators in the synthesis of topologically attractive multiarm star polymers enable to tune and control the number of s arms and its PDI value were proved.

Table 1. Molecular weight data of unimolecular initiator end-capped and anchored dendrimers.

Initiator	No. of initiator present	dn/dc	Theoretical Molecular weight (M _n) (g/mol)	SEC-MALLS Molecular weight (M _n) (g/mol)	PDI
3	6	0.1661	2436	2319	1.05
4	12	0.1604	5241	4584	1.09
5	24	0.1106	10852	11220	1.01
6	48	0.1374	22,027	28480	1.00
7	6	0.1701	3157	4493	1.14
8	12	0.1213	6683	5557	1.09
9	24	0.1010	13735	13680	1.29
10	48	0.1591	27792	27910	1.00

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Cite this: DOI: 10.1039/c0xx00000x

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Notes and references

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5 Acknowledgement:

The authors thank the Department of Science of Technology, Government of India for financial support for this work (SERB Sanction Order No. and Date: SR/S1/PC-08/2010 (G).Dt. (13-07-2011)

Electronic Supplementary Information (ESI) available: Experimental ¹⁰ details (materials, measurements, synthetic procedure of Compound **2**), ¹H NMR spectra of compound **2-5** and **7-9**, ¹³C NMR spectra of

- compound **2-10**, MALDI-TOF MS spectra of compound **4**, **5**, **8** and **9** and polymerization results of styrene.
- ¹⁵ 1 Polymer Sciences: A comprehensive Referances, Ed. by, K. Matyjaszewski and M. Möller, Vol.6, Macromolecular Architeacture and Soft Nano-Objects, Vol. Ed. A. H. E. Muller, K. L. Wooley, .; Elscvier, Amsterdam, 2012.
- 2 (a) Star and Hyperbranched polymers, M. K. Mishra and S. Kobayashi,
- 1999 and references therein. (b) T. Shibata, S. Kanaoka and S. Aoshima, J. Am. Chem. Soc, 2006, 128, 7497.
- Polymer Sciences: A comprehensive Referances, Vol.6, Macromolecular Architeacture and Soft Nano-Objects, Ed. A. H. E. Muller, K. L. Wooley, .; Elsevier, Amsterdam, 2012. PP 29-109 and references therein.
- 4 C. J. Hawker. Angew. Chem. In. Ed. 1995, 34, 1456.
- (a) R. B. Grubbs, C. J. Hawker, J. Dao and J. M. J. Frechet, *Angew. Chem., Int. Ed.* 1997, 36, 270. (b) C. J. Hawker, J. M. J. Frechet, R. B. Grubbs and J. Dao, *J. Am. Chem. Soc.* 1995, 117, 10763.
- 30 6 G. Chessa, A, Scrivati, U. Matteoli and V. Castelvetro, *Polym.* 2001, 42, 9347.
- 7 T. Kakuchi, A. Narumi and T. Matsuda, *Macromolecules*. 2003, 36, 3914.
- 8 J. B. Beil and S.C. Zimmerman, *Macromolecules*, 2004, 37, 778.
- 35 9 C-H. Lu, J-H Wang, F-C. Chang and S-W. Kuo, *Macromol. Chem. Phys.* 2010, 211, 1339.
- 10 S. Abraham, J. H. Choi, C. S. Ha and I. Kim, J. Polym. Sci. part A:Polym. Chem. 2007, 45, 5559.
- 11 Y. Miura and H. Dote, J. Polym. Sci. part A:Polym. Chem. 2005, 43, 3689.
- 12 C. J. Hawker and J. L. Hedrick, Macromolecules. 1995, 28, 2993.
- 13 G. Sankar and A. Sultan Nasar, J. Polym. Sci. part A:Polym. Chem. 2007, 45, 1557.
- 14 Synthesis of unimolecular initiator end-capped and anchored polyurethane dendrimers (3-6 and 7-10). 0.5 g of blocked isocyanate-terminated dendrimer (G1 or G2 or G3 or G4) and calculated amount of styrenyl-TEMPO reagent 1 (end-capping reagent) or 2 (anchoring reagent) were taken in a two-necked 100mL flask and were dissolved in 10 mL of dry DMF. To this solution was
- added 0.02 g of DBTDL. The reaction was carried out in an oil bath maintained at 90°C for overnight. The reaction mass was added dropwise into 100 mL of water to precipitate the product and it was filtered and dried under vacuum. Then, the curd product was washed in chloroform to remove unreacted styrenyl-TEMPO reagent and then re-precipitated in dry THF by the addition of petroleum ether.
- (3). Yield: 92.4%. ¹H NMR (400 MHz, DMSO-d₆): δ 9.73 (s, 9H), 7.55-6.77 (m, 42H), 5.05 (s, 6H), 4.67 (m, 6H), 4.40 (m, 6H), 4.17 (m, 6H), 1.50-0.50 (m, 108 H). ¹³C NMR (75 MHz, DMSO-d₆): 163.39, 152.02, 147.73, 141.02, 139.29, 127.62, 126.41, 117.23, 147.62, 20.20, 104.47 (m, 64), 41.71 (m, 64), 117.23, 141.02, 139.29, 127.62, 126.41, 117.23, 141.62,
- ⁶⁰ 116.76, 102.29, 101.47, 63.22, 61.46, 42.37, 22.23, 21.46.
 (4). Yield: 92.8%. ¹H NMR (400 MHz, DMSO-d₆): δ 9.79 (s, 21H), 7.91-7.12 (m, 90H), 5.28 (s, 18H), 4.84 (m, 12H), 4.42 (m 12H), 4.09

Communication

(m, 12H), 1.57-1.27 (m, 216H). 13 C NMR (75 MHz, DMSO-d₆): 163.68, 152.52, 147.73, 141.39, 139.47, 127.62, 126.76, 117.02, 116.39, 102.29, 101.16, 63.47, 61.42, 42.67, 22.63, 21.47. Calcd: [M+Na]⁺ m/z = 5227.33. Found: MALDI-TOF-MS [M+Na]⁺ = 5264.6.

(5). Yield: 94.7%.¹H NMR (400 MHz, DMSO-d₆): δ 9.97 (s, 45H), 7.91-6.70 (m, 186H), 5.02 (m, 24H), 4.81 (s, 42H), 4.62-4.59 (m, 24 H), 4.11 (m, 24H), 1.86-1.06 (m, 432H). ¹³C NMR (75 MHz, DMSO-d₆): 163.78, 152.69, 147.93, 141.26, 139.77, 127.02, 126.16, 117.76, 116.52, 102.20, 101.39, 63.22, 61.63, 42.27, 22.63, 21.47. Calcd: [M+Na]⁺ m/z = 10875.2. Found: MALDI-TOF-MS [M+Na]⁺ = 10873.7.

(6). Yield (92.4%).¹H NMR (400 MHz, DMSO-d₆): δ 9.76 (s, 93H), 7.91-6.77 (m, 378H), 5.11 (s, 90H), 4.99-4.95 (m, 48H), 4.62-4.59 (m, 48H), 4.07-4.03 (m, 48H), 2.1-1.06 (m, 864H), ¹³C NMR (75 MHz, DMSO-d₆): 163.38, 152.67, 147.93, 141.20, 139.27, 127.54, 126.16, 117.77, 116.67, 102.20, 101.27, 63.62, 61.13, 42.67, 22.63, 21.17. Calcd: [M+Na]⁺ m/z = 22048.5. Found: MALDI-TOF-MS [M+Na]⁺ = 22048.2.

(7). Yield: 96.5% ¹H NMR (400 MHz, DMSO-d₆): δ 9.69 (s, 9H), 8.15-6.87 (m, 72H), 5.23 (s, 6H), 4.83 (m, 6H), 4.72 (m, 6H), 4.47 (m, 6H), 2.12-0.85 (m, 102H). ¹³C NMR (75 MHz, DMSO-d₆): 166.38, 140.88, 132.89, 130.23, 129.47, 128.29, 128.05, 127.72, 83.04, 66.79, 60.36, 34.20, 29.34, 20.37, 17.05.

(8). Yield: 96.2% ¹H NMR (400 MHz, DMSO-d₆): δ 9.73 (s, 21H), 7.55-6.77 (m, 141H), 5.05 (s, 18H), 4.93 (m, 12H), 4.75 (m, 12H), 4.45 (m, 12H), 1.51-1.11 (m, 204H). ¹³C NMR (75 MHz, DMSO-d₆): 166.88, 140.28, 132.47, 130.03, 129.42, 128.27, 128.02, 127.72, 83.64, 66.09, 60.34, 34.29, 29.24, 20.32, 17.29. Calcd: [M+Na]⁺ m/z = 6682.63. Found: MALDI-TOF-MS [M+Na]⁺ = 6705.2.

- = 6082.65. Found: MALDI-10F-MS [M+Ra] = 6705.2.
 (9). Yield: 94.8%. ¹H NMR (400 MHz, DMSO-d₆): δ 9.96 (s, 45 H), 7.94-7.26 (m, 306), 5.26 (42H), 5.06 (m, 24H), 4.82 (m, 24H), 4.52 (m, 24H), 1.69-0.75 (m, 408H). ¹³C NMR (75 MHz, DMSO-d₆): 166.38, 140.88, 132.89, 130.23, 129.47, 128.29, 128.05, 127.72, 83.04, 66.79, 60.36, 34.20, 29.34, 20.37, 17.05. Calcd: [M+Na]⁺ m/z
- = 13759.2. Found: MALDI-TOF-MS $[M+Na]^+$ = 13757.6. (10). Yield (95.6%).¹H NMR (500 MHz, DMSO-d₆): δ 9.72 (s, 93H),
- 7.74-6.75 (m, 618 H), 5.05 (s, 90H), 4.95-4.91 (m, 48H), 4.75-4.70 (m, 48H), 4.55-4.49 (m, 48H), 1.5-0.5 (m, 816H). 13 C NMR (75 MHz, DMSO-d₆): 166.36, 140.68, 132.83, 130.20, 129.57, 128.28, 128.05, 127.62, 83.94, 66.78, 60.12, 40.41, 34.07, 29.71, 20.36, 17.15. Calcd: [M+Na]⁺ m/z = 27814.3. Found: MALDI-TOF-MS [M+Na]⁺ = 27814.6.
- 15 (a). H. Kothandaraman and A. Sultan Nasar, *Polymer*. 1993, 34, 610.
 b). S. Veerapandian and A. Sultan Nasar, *RSC Adv*. 2015, 5, 3799.
- 16 T. Emrick, W. Hayes and J. M. J. Frechet, *J. Polym. Sci. Part A. Polym.Chem.* 1999, 37, 3748. (b) K. L. Hawker, G. G. Barcly and J. Dao, J. Am. Chem. Soc. 1996, 118, 11467.
 - 17 Hand book of Radical Polymerization; Wiley Interscience: K. Matyjaszewski, T. P. Davis, 2002, pp. 463-522
 - 18 Y. Zhao, X. Shuai, C. Chen and F. Xi, *Macromolecules*. 2004, 37, 8854.

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