

RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This *Accepted Manuscript* will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

1 Synthesis and Properties of Nano-silica Modified Environmentally Friendly 2 Polyurethane Adhesive

3 GUO Jia-Hu^{1, 2*}, LIU Yu-Cun¹, CHAI Tao¹, JING Su-Ming¹, MA Hui¹, QIN Ning¹, ZHOU Hua¹,
4 YAN Tao¹, HE Wei-Ming²

5 1. Chemical Engineering and Environment College, North University of China, Taiyuan 030051, China

6 2. Department of ESH, Science and Technology University of Sichuan Staff, Chengdu 610101, China

7 *Corresponding author: E-mail address: guojiahuphd@163.com

8 **Abstract:** In this study, nano-SiO₂ modified waterborne polyurethane (WPU) adhesive with
9 remarkably low VOC content and high adhesion performance was successfully synthesized via
10 in-situ polymerization. The influence of nano-silica concentration on different properties of the
11 aqueous-based adhesive and the casted films were carefully investigated. When the concentration of
12 nano-SiO₂ was in the range of 2.0% to 2.5%, the performances of the adhesive reached the optimum.
13 Under optimum conditions, the appearance of emulsion was milky white with blue light uniform
14 liquid. The T-peel and shear strength of the nano-modified adhesive showed essential improvement
15 compared with pure-WPU adhesive. The results of shear adhesion failure temperature test showed
16 that the thermo-tolerant temperature of the adhesive was higher than 175°C. The tensile strength,
17 elongation at break and water swelling of the casted films were about 21.3MPa, 890% and 2.7%,
18 respectively. Finally, other properties of the water-based adhesive were characterized by FTIR,
19 SEM, molecular weight, pH, mean particle size and solid content measurement.

20 **Keywords:** Polyurethane; Adhesives; Nano-SiO₂; Health; Storage Stability; Heat Resistance; VOC

21 INTRODUCTION

22 In the past decades, adhesives based on polyurethane, possess a unique combination of
23 properties, including superior adhesion, excellent outdoor durability and outstanding low
24 temperature performance, and have been widely used in leather, footwear, textile, coating, furniture,
25 packaging, construction, automotive, military and other industry fields¹⁻¹⁰. However, it is almost
26 impossible to avoid the release of volatile organic compounds (VOCs) during the usage of

27 traditional solvent-based polyurethane adhesives. These volatile organic compounds usually have
28 toxic and hazardous effects to the human beings. Hence, the studies of green and environmentally
29 friendly water-based adhesives have drawn much attention and a vast amount of researches have
30 been done in the past few years^{3,8,11-16}.

31 From the environmental and healthy point of view, water-borne polyurethane (WPU) adhesives
32 are preferred in most circumstances, due to their lower toxicity compared with solvent-borne
33 products. Unfortunately, the utility of aqueous based polyurethane adhesives were restricted to some
34 extent due to their inferior properties, such as lowered adhesive strength, water resistance,
35 mechanical strength and thermal stability¹²⁻¹⁸. In order to improve the properties mentioned above,
36 significant attempts have been carried out during the last decade via various approaches, such as
37 using different kinds of chain extenders, incorporating different diisocyanates, making and
38 cross-linking nanoparticles^{13,19-26}. It is worth mentioning that nano-modification was regarded as an
39 effective and promising technique in this field, since it has been confirmed that with the
40 incorporation of nano-composites into the polyurethane, the physicochemical properties of the
41 WPU, such as solvent resistance, mechanical strength and thermo tolerance were greatly improved
42 compared to that of conventional pure WPU²⁷⁻³⁵. In particular, polymer/SiO₂ nano-composites
43 exhibit improved mechanical properties and thermal stability compared with pure polymer matrix
44 and polymer micro-composites³⁴⁻³⁷. Therefore, polymer/nano-SiO₂ composites have drawn
45 substantial academic and industrial interest, and have been employed in a variety of applications.
46 The preparation, characterization, properties, and applications of polymer/silica nano-composites
47 have become a quickly expanding field of research in recent years²⁹⁻³⁷.

48 In this work, a series of nano-silica modified aqueous based polyurethane adhesives were
49 successfully synthesized through in-situ polymerization. The synthesized waterborne adhesive was
50 demonstrated by different characterization techniques. And a detailed study of the relationship
51 between nano-SiO₂ concentration and adhesive properties was carried out. The adhesive prepared in
52 this work featured a remarkably low VOC content (less than 0.5%) and exhibited high adhesive

53 performance even when the operating temperature was up to 175°C.

54 **EXPERIMENTAL**

55 **Materials**

56 Isophorone diisocyanate (IPDI, 98% purity) and 1,4-butanediol (BDO, 99% purity) were
57 purchased from Bayer (Germany). Modified nano-silica (99.9% purity), ethylene diamine (EDA, 99%
58 purity) and 1-methyl-2-pyrrolidone (NMP, 99% purity) were purchased from Evonik Degussa
59 (Germany). Poly(1,4-butylene adipate) end capped diol (PBA, number-average molecular weight =
60 2,000 g·mol⁻¹, 98% purity), triethylamine (TEA, 99% purity), dimethylolpropionic acid (DMPA, 99%
61 purity) and acetone (99.5% purity) were purchased from Sinopharm Chemical Reagent (China).
62 Anti-foaming agent BYK-028 was purchased from BYK Additives & Instruments (Germany).
63 Catalyst, naphthenic acids, bismuth salts (NABS, 99.9% purity) was kindly supplied by Shanghai
64 Institute of Organic Chemistry (China). Potassium bromide (99.9% purity) was purchased from
65 Merck (Germany). Distilled water was produced in the laboratory.

66 PBA was dried under vacuum at 110°C and 1-2 mm Hg for 2.5 h before use. Acetone, BDO
67 and NMP were dehydrated with 4-Å molecular sieves for 5 days prior to use. Other chemicals and
68 solvents mentioned above were used as received.

69 **Synthesis of the nano-silica modified WPU**

70 The nano-silica modified WPU dispersions were prepared in a 500mL glass reactor, which
71 equipped with a nitrogen inlet, a temperature sensor, a condenser and a mechanical stirrer. An oil
72 bath was employed for controlling the reaction temperature.

73 Firstly, PBA was preheated to 70°C and was kept constant for 30 min, after that, nano-SiO₂ was
74 added into the reactor at high-speed stirring (2,000 rpm). The temperature was then reduced to 60°C.
75 Then, it was slowly reheated to 80°C after the IPDI and NABS catalyst were added. The
76 temperature was kept constant for 4 h under nitrogen atmosphere. Secondly, after the NCO content
77 was reached predetermined value, hydrophilic chain extender DMPA was added into the prepolymer.
78 Then, the temperature was heated to 90°C and was kept constant for 2 h. Thirdly, BDO was added

79 into the reactor and the temperature was kept constant at 90°C for another 2 h. After the period, the
80 mixture was cooled to 55°C and the prepolymer was neutralized with TEA. And acetone was added
81 to reduce the viscosity, if necessary. The polyurethane dispersion of the bulk in water was
82 performed with vigorous stirring (1,000 rpm) for 30 min. Afterwards, EDA was added into the
83 dispersion at 25°C and kept constant for 2 h to complete the chain-extension reaction between the
84 NCO terminated groups of the prepolymer and the amino groups of the EDA. Finally, de-foaming
85 agent BYK-028 was added into the dispersion at 40°C and the product was obtained after acetone
86 was removed under vacuum. The reaction scheme for the synthesis of nano-SiO₂ modified
87 waterborne polyurethane adhesive was illustrated in Figure 1.

88 The basic formulation of the WPU dispersion was 36.0 wt.% PBA, 1.5 wt.% DMPA, 9.9 wt.%
89 IPDI, 0.1 wt.% NABS, 0.4 wt.% BDO, 0.9 wt.% TEA, 47.0-50.0 wt.% distilled water, 0.1 wt.%
90 EDA, 0.1 wt.% BYK-028 and 1.0-3.0 wt.% nano-SiO₂.

91 **Preparation of the polyurethane films**

92 The films were prepared by casting the aqueous dispersion onto a PTFE mould (14cm × 8cm)
93 and drying the dispersion at room temperature for 3 days. The polyurethane films (approximately
94 0.3cm thick) were annealed at 60°C for 12 h and then vacuum dried for 12 h. The vacuum-dried
95 films were stored in desiccators at ambient temperature.

96 **Characterization and performance testing**

97 The viscosities of the aqueous polyurethane dispersions were tested in a Brookfield Viscometer
98 DV-II at 25°C. The molecular weight of the WPU dispersions was obtained by Gel Permeation
99 Chromatography (GPC), using a Waters peristaltic pump model 515 HPLC and a Waters 410
100 refractive index detector. The mean particle size (D_{50}) of the dispersions was measured in a
101 submicron particle sizer PSS Nicomp 380 DLS and the pH values were measured in a pH meter
102 S700-K. Each sample was tested for 5 times, and the values were averaged.

103 The structure of the WPU films were analyzed by a Perkin Elmer 100 Spectrometer, provided
104 with Fourier transform analysis (FTIR). The WPU emulsion was coated on the potassium bromide

105 wafer with a diameter of 13 mm for the analysis. Measurements were carried out by the
106 transmittance mode in the range of 4,000-400 cm^{-1} with a resolution of 2 cm^{-1} .

107 The morphologies of the WPU and nano-SiO₂ dispersed in WPU were observed by a
108 HITACHI S-4700 Scanning Electron Microscopy (SEM) with an accelerating voltage of 10 kV. The
109 samples for the test were prepared by casting the WPU adhesives onto a clean PTFE pan and drying
110 it in a vacuum oven at 60°C for 3 days. All of the specimens were coated with gold in vacuum
111 before the SEM micrographs were carried out.

112 Peel resistance of the aqueous adhesive (i.e. T-peel strength, or 180-degree peel strength) was
113 measured with an Instron 5569 machine at a relative humidity of $50 \pm 2\%$ at $25 \pm 2^\circ\text{C}$. The
114 experiments were carried out according to ASTM D1876-08³⁸. Special test panels about 300 mm
115 wide by 300 mm long were prepared. The adhesive was applied using a roller applicator of 100 μm
116 onto the Aluminum panel and dried at 60°C for 15 min. Then, an Aluminum-Alloy sheet was
117 adhered at a pressure of 20MPa for 15 min to improve the contact. Afterwards, the samples were
118 kept for 7 days at room conditions. Finally, the bonded panels were cut into 25 mm wide test
119 specimens as shown in the standard³⁸. A constant head speed of 250 $\text{mm}\cdot\text{min}^{-1}$ was used in the test.
120 Each specimen was peeled over 125 mm length of the bond line after the initial peak. And the
121 values quoted were the averages of 5 measurements.

122 Lap-shear test was carried out using an Instron 5569 machine at a relative humidity of $50 \pm 2\%$
123 at $25 \pm 2^\circ\text{C}$. The tests were carried out according to ASTM D3163-14³⁹. The synthesized adhesive
124 was deposited using a roller applicator of 100 μm onto the faces of Aluminum and Aluminum-Alloy
125 sheets and dried at 60°C for 15min. Then, the panels were attached at a pressure of 20MPa for 15
126 min to improve the contact. Afterwards, the samples were kept for 7 days at room conditions.
127 Finally, the bonded panels were cut into 25 mm wide test specimens as shown in the standard³⁹. The
128 tests were carried out by loading the specimen to failure at a rate of 10 $\text{mm}\cdot\text{min}^{-1}$ cross head speed.
129 The value of each sample was reported as the average of 5 measurements.

130 Heat-fail temperature showed a limiting temperature above which the adhesive was not to be

131 exposed in service under shear load. In this study, the shear adhesion failure temperature (SAFT)
132 test was employed to define the thermo-tolerant temperature (i.e. maximum usage temperature) of
133 the synthesized waterborne adhesive according to the study of Sardon¹⁵. For each adhesive sample,
134 5 specimens were measured and the values were averaged. The procedure to perform the SAFT test
135 was as follows: the adhesive was spread on the Aluminum panel and Aluminum-Alloy sheet at a
136 thickness of 100 μ m by using a roller applicator. After drying it for 15 min at 60°C, the joint was
137 placed under a pressure of 20MPa for 15 min. Then, the samples were kept at ambient temperature
138 for 7 days. Finally, a lap joint measuring 25 mm \times 25 mm inside two strips of standard substrate
139 was prepared for test. And standard weights (500g) were positioned and the counter was turned on
140 until all specimens failed. SAFT was measured using a temperature ramp of 0.5°C·min⁻¹ from 20°C
141 to 200°C.

142 The tensile properties of the casted polyurethane films were measured at room temperature
143 using an Instron 5569 machine following the specification of ASTM D638-10⁴⁰ with a crosshead
144 speed of 100 mm·min⁻¹. Dumbbell specimens were cut from the casted films by a sheet-punching
145 machine as shown in the standard⁴⁰. The value of each sample was reported as the average of 5
146 measurements.

147 The determination of the VOC content of the synthesized WPU adhesive was used an Agilent
148 6890 N Network Gas Chromatograph System interfaced with a 5973 Network MSD according to
149 the specification of ASTM D3960-13⁴¹. And the VOC content percentage was determined as the
150 follow equation⁴¹:

$$151 \quad VOC \text{ content } (\%) = \frac{M_o}{M_s} \times 100 = \frac{M_v - M_w - M_{ex}}{M_s} \times 100$$

152 Where M_o is the weight of organic volatiles, M_v is the weight of total volatiles, M_w is the weight of
153 water, M_{ex} is the weight of exempt volatile compound and M_s is the weight of solids.

154 For the swelling study, the WPU films (7cm \times 7cm) were immersed in distilled water for 24h
155 at 25°C and the water swelling percentage was determined from the weight increase as follows^{4, 25}:

$$156 \quad \text{Water swelling (\%)} = \frac{W_2 - W_1}{W_1} \times 100$$

157 Where W_1 is the weight of the dried film and W_2 is the weight of the film at equilibrium swelling.

158 The solid content of the WPU was determined from the weight decrease before and after
159 solvent evaporation. About 2-3g waterborne polyurethane dispersion was placed in a weighing
160 bottle, then kept it in an oven at 110°C until constant weight was reached. And the solid content
161 percentage was determined as the follow equation¹⁴:

$$162 \quad \text{Solid content (\%)} = \frac{G_2}{G_1} \times 100$$

163 Where G_1 is the original weight of WPU and G_2 is the dried weight.

164 The value of VOC-content, water-swelling and solid content of each sample was reported as
165 the average of 5 measurements.

166 **RESULTS AND DUSCUSSION**

167 **Physical properties of the nano-silica modified aqueous based polyurethane adhesive**

168 Table 1 shows the physical properties of the nano-SiO₂ modified waterborne polyurethane
169 adhesive. Under ideal conditions, the storage life of the adhesive is more than 9 months, and the
170 appearance of it is milky white with blue light uniform liquid. The pH value of the product is in the
171 range of 6.5-7.5. Number average molecular weight (M_n) and mean particle size (D_{50}) of the WPU
172 adhesive are approximately 55,000 g·mol⁻¹ and 800 nm, respectively. The Brookfield viscosity of
173 the adhesive is about 150 mPa·s (25°C) and the solid content is in the range of 50-52%. The VOC
174 content of the nano-SiO₂ modified water-based polyurethane adhesive is less than 0.5%.

175 **FTIR analysis of nano-silica modified WPU**

176 The typical functional groups of the SiO₂-modified WPU (nano-silica content = 2.0%) are
177 characterized by FTIR as shown in Figure 2. As can be seen in the spectrum, the absence of bands
178 at 2,250-2,270 cm⁻¹ confirmed the absence of a free NCO group in the polymer structure, which
179 indicated the completion of the reaction. The bands observed between 3,300 and 3,600 cm⁻¹ are
180 attributed to the N-H stretching vibration of urethane²². The bands in the range of 2,800-3,000 cm⁻¹

181 are connected to the aliphatic C–H stretching vibration of $-\text{CH}_2$ and $-\text{CH}_3$ ²⁹. The characteristic
182 absorption peak of C=O of urethane is in the unique range of 1,710-1,760 cm^{-1} and the bands at
183 1,020-1,120 cm^{-1} are attributed to the Si–O–C and C–O–C stretching vibration absorptions^{20,31}. The
184 absorption peak around $\sim 470 \text{ cm}^{-1}$ is assigned to the strong Si–O–Si bending vibration from silica,
185 proving that the nano-SiO₂ is successfully incorporated into the polyurethane chains. Other strong
186 characteristic absorption bands of silica are overlapped with the existing bands of WPU groups,
187 such as 1,100-1,020 cm^{-1} (Si–O–Si stretching vibration)³².

188 SEM analysis of nano-silica modified WPU

189 Figure 3 shows the SEM micrographs of the (a) pure-WPU and (b), (c) nano-modified WPU
190 having 2.0 wt.% nano-SiO₂ loading. As can be seen in Figure 3(a), the cross-section of pure-WPU
191 shows a relatively smooth, glassy surface and there are almost no cracks. Comparatively, in Figure
192 3(b), the WPU/SiO₂ nano-composites show much rougher fractured surfaces. It is well known that
193 nanoparticles dispersion and adhesion with the polymer matrix are very important to improve the
194 properties of nano-modified WPU²⁰. Uniform distribution of nano-silica as fillers in the matrix
195 played an important role in the mechanical properties and thermo-tolerant performance of the nano
196 nano-SiO₂ modified WPU²⁷, which will be discussed in the following sections. In Figure 3(c), we
197 cannot find obvious agglomeration of the nano-silica, which indicates the dispersion capability of
198 nano-silica particle in WPU adhesive in this work is pretty good. It is believed that the nano-silica
199 incorporation of nano-SiO₂ groups into the polyurethane chains is kind of hybrid and graft
200 copolymer²⁹, and this statement is proved or partly proved by the SEM image of the WPU as is
201 shown in Figure 3(c).

202 Effect of nano-silica content on the appearance and storage stability of WPU

203 Table 2 shows the effect of nano-SiO₂ concentration on the appearance and storage stability of
204 the WPU adhesive. With the increase of nano-silica content, the appearance of emulsion is
205 gradually changed from micro-blue transparent to milky white with blue light. When the
206 concentration of nano-SiO₂ is in the range of 2.0%-2.5%, the appearance and storage life of the

207 WPU reached the optimum. Furthermore, the results show that nano-silica content has a remarkable
208 impact on the stability of the emulsion. When the nano-silica content increases to 3.0%, the storage
209 life of the WPU is significantly shortened (less than 6 months). The reasons may be that, after the
210 in-situ polymerization, nano-SiO₂ groups incorporate into the polyurethane chains uniformly, which
211 conducive to the formation of a three-dimensional network structure of the polyurethane, at the
212 same time, enhance the hydrophobic of the polyurethane chains¹⁹. Excess nano-SiO₂ may leads to
213 destabilization of the emulsion. Therefore, with the increase of nano-silica, the storage life of WPU
214 is shortened.

215 **Effect of nano-silica content on the Brookfield viscosity and mean particle size of WPU**

216 Figure 4 shows the Brookfield viscosities (25°C) and mean particle size (D₅₀) of the WPU
217 emulsions with different nano-SiO₂ contents. As shown in the histogram, the viscosity of the WPU
218 dispersion decreases with the increase of nano-silica content and the mean particle size increases
219 with the increase of nano-silica. When the concentration of nano-SiO₂ increases from 0% to 3.0%,
220 the viscosity decreases from 480 mPa·s to 40 mPa·s and the mean particle size of the WPU
221 dispersion increases from 310 nm to 3,590 nm, respectively. In general, the smaller of the particle
222 size is, the higher viscosity of the emulsion will be, and vice versa. Considering the dispersion state
223 of the WPU, when the emulsion particles are relatively small, water-soluble macromolecules are not
224 in the "spherical" crimped state, but in the free-elongation low-energy state. In this condition, the
225 macromolecules are wound and cross-linked with each other, which leads to the significantly
226 increasing of the emulsion viscosity²¹. On the other hand, when the particles are relatively large,
227 particles in the water dispersion are under a relatively ordered arrangement, therefore, the apparent
228 viscosity of the emulsion is relatively low. Thus, with the increase of nano-SiO₂ content, the particle
229 diameter of the emulsion become larger and the viscosity of the WPU become lower.

230 **Effect of nano-silica content on the adhesion properties of WPU**

231 Figure 5 shows the adhesion properties of the waterborne adhesive. T-peel strength and shear
232 strength of the adhesive show a decreasing trend after the prior increase. And this trend is similar to

233 the research of Chen²⁰. When the concentration of the nano-SiO₂ is 2.0%, the T-peel strength and
234 the shear strength reach the maximum, i.e. 4692N/m and 45.7MPa, respectively. The reasons may
235 be that, with the addition of the nano-silica, the rigidity of the polyurethane's hard segment and the
236 cohesion of the polyurethane adhesive are increased. At the same time, the force between
237 nano-silica and polyurethane is greatly enhanced due to the large specific surface area of
238 nanoparticles. Furthermore, there are a large number of active hydroxyl groups on the surface of
239 nano-SiO₂, together with the urethane groups on the surface of the polyurethane, which formed
240 hydrogen bonds²². Therefore, the adhesion properties of the polyurethane adhesive are significantly
241 improved when the concentration of nano-SiO₂ increases from 0% to 2.0%. However, when the
242 nano-silica content further increase to 2.5-3.0%, the adhesion properties become worse. The main
243 reasons may be that, the excess of nano-SiO₂ caused the settlement phenomenon, and thus reduced
244 the adhesion of the adhesive coating.

245 **Effect of nano-silica content on the water resistance and thermal stability of WPU**

246 Figure 6 shows the water swelling and thermo-tolerant properties of the nano-SiO₂ modified
247 adhesive. The water swelling value decreases with increasing the nano-silica concentration as
248 shown in the graph. It means that the water resistance of the films increases with the addition of
249 nano-SiO₂. This might be due to the presence of the hydrophobic Si-O-Si groups of the cross-link
250 structure¹¹. The network structure of nano-SiO₂ increases the density of the film, and mitigates
251 water penetration³³. This barrier effect significantly enhances the water resistance of the films. The
252 water swelling decreases from 12.1% to 2.5% when the nano-silica content increases from 0% to
253 3.0%. And the results are similar to the previous studies^{11, 20}. On the other hand, the thermo-tolerant
254 property of the adhesive is enhanced with the increase of nano-SiO₂ content. The reason may be
255 that, the incorporation of nano-SiO₂ groups into the polyurethane chains increases the resistance of
256 the rearrangement of the molecular chains. Simultaneously, the surface silanol groups of nano-SiO₂
257 react with the NCO groups in the WPU chains, and the formation of stable chemical linkages
258 between the nano-silica and the polyurethane improve the heat resistance of WPU²⁶. When the

259 nano-silica content increases from 0% to 3.0%, the thermo-tolerant temperature of the adhesive is
260 increased from 81°C to 183°C, respectively.

261 **Effect of nano-silica content on the mechanical properties of WPU films**

262 The mechanical properties of the WPU films are evaluated by tensile test and the results are
263 shown in Figure 7. As can be seen from the bar graph, the mechanical properties of the WPU films
264 are greatly improved via added a small amount of nano-SiO₂. When the content of the nano-silica
265 are approximately 1.5% to 2.0%, the elongation and tensile strength at break reach the optimum, viz.
266 890% and 21.3MPa, respectively. This might be due to cross-linking of nanoparticles in the
267 polyurethane chain. Generally, the improvement of properties of nanocomposites as compared with
268 parent polymers is associated with certain structural changed in a polymer matrix due to the small
269 size and unique structure of nanoparticles²⁴. When the content of nano-silica is over 2.0%, the
270 mechanical properties decrease sharply. The reason may be that, when the nano-SiO₂ content is too
271 high, not all of the nanoparticles could be polymerized into the main chain of the polyurethane, but
272 agglomerated^{22-23, 37}. As a result, the mechanical properties decrease. Furthermore, this may also
273 caused storage stability and adhesive properties deteriorated.

274 **CONCLUSIONS**

275 In this study, using nano-SiO₂, IPDI, PBA, DMPA, BDO, TEA, and EDA as main raw
276 materials, a series of nano-silica modified waterborne polyurethane adhesives were successfully
277 obtained via in-situ polymerization. The incorporation of nano-SiO₂ groups into the polyurethane
278 chains was confirmed by means of Fourier Transform Infrared Spectroscopy (FTIR). The viscosity,
279 adhesion properties of the aqueous based polyurethane adhesive, mechanical performances and
280 water resistance of the casted films were also investigated. The results showed that the
281 concentration of nano-SiO₂ had significant impact on these properties.

282 The synthesized aqueous based polyurethane adhesive possessed good storage stability (over 9
283 months). The results of SAFT test showed that the thermo-tolerant temperature of the nano-SiO₂
284 modified adhesive was greatly improved compared with unmodified pure-WPU adhesive. The

285 result of VOC test illustrated that the adhesive featured a remarkably low VOC content. The risk of
286 VOCs to the human health and the environment was significantly reduced compared with
287 conventional solvent-based adhesives. In this respect, the waterborne polyurethane adhesive that
288 prepared in this research can be defined as healthy and environmentally friendly material.

289 And in the following research, a detailed characterization of this nano-modified WPU adhesive
290 will be conducted by using NMR, TEM, AFM, XRD, TG, DTA, DSC, DMA, etc.

291 **ACKNOWLEDGEMENTS**

292 This work was supported by *NSAF (Grant No. U1330131)*. The authors also would like to
293 acknowledge the collaboration of Mr. CAI Xing-Wang and Mr. FANG Wei in this research.

294 **LITERATURE CITED**

- 295 1. Chattopadhyay, D.K. and Raju, K.V.S.N. *Prog. Polym. Sci.* 2007, 32, 352-418.
- 296 2. Sheikhy, H., Shahidzadeh, M., Ramezanzadeh, B. and Noroozi, F. *J. Ind. Eng. Chem.* 2013, 19,
297 1949-1955.
- 298 3. Xiang, L., Ling, H.J., Zhang, Z.Y., Luo, L., Yang, L.T. and Li, L.X. *Polym. Mater. Sci. Eng.*
299 2011, 27, 31-34.
- 300 4. Ji, X., Zhou, Y., Zhang, B., Hou, C. and Ma, G. *ISRN. Polym. Sci.* 2013.
- 301 5. Liu, Z., Yuan, X.H., Beck, A.J. and Jones, F.R. *Compos. Sci. Technol.* 2011, 71, 1529-1534..
- 302 6. Poh, A.K., Sin, L.C., Foon C.S. and Hock, C.C. *J. Adhes. Sci. Technol.* 2014, 28, 1020-1033.
- 303 7. Zain, N.M., Ahmad, S.H. and Ali, E.S. *Appl. Mech. Mater.* 2013, 393, 51-56.
- 304 8. Łągiewczyk, M. and Czech, Z. *Pol. J. Chem. Technol.* 2011, 13, 47-50.
- 305 9. Tyczkowski, J., Krawczyk-Klys, I., Kuberski, S. and Makowski, P. *Eur. Polym. J.* 2010, 46,
306 767-773.
- 307 10. Desai, S.D., Emanuel, A.L. and Sinha, V.K. *J. Polym. Res.* 2003, 10, 275-281.
- 308 11. Rahman, M.M., Hasneen, A., Lee, W.K. and Lim, K.T. *J. Sol-gel Sci. Techn.* 2013, 67,
309 473-479.

- 310 12. Sardon, H., Irusta, L., Fernandez-Berridi, M.J., Lansalot, M. and Bourgeat-Lami, E. *Polymer*.
311 2010, 51, 5051-5057.
- 312 13. Delpech, M.C. and Coutinho, F.M.B. *Polym. Test.* 2000, 19, 939-952.
- 313 14. Garcia-Pacios, V., Iwata, Y., Colera, M. and Martin-Martinen, J.M. *Int. J. Adhes. Adhes.* 2011,
314 31, 787-794.
- 315 15. Sardon, H., Irusta, L., Gonzalez, A. and Fernandez-Berridi, M.J. *Prog. Org. Coat.* 2013, 76,
316 1230-1235.
- 317 16. Ballard, N., Carretero, P. and Asua, J.M. *Macromol. React. Eng.* 2013, 7, 504-514.
- 318 17. Lopez, A., Reyes, Y., Degrandi-Contraires, E., Canetta, E., Creton, C. and Asua, J.M. *Eur.*
319 *Polym. J.* 2013, 49, 1541-1552.
- 320 18. Cakic, S.M., Ristic, I.S., Marinovic-Cincovic, M. and Spirkova, M. *Int. J. Adhes. Adhes.* 2013,
321 41, 132-139.
- 322 19. Petrovic, Z.S., Cho, Y.J., Javni, I., Magonov, S., Yerina, N., Schaefer, D.W., Ilavsky, J. and
323 Waddon, A. *Polymer*. 2004, 45, 4285-4295.
- 324 20. Chen, J.J., Zhu, C.F., Deng, H.T., Qin, Z.N., and Bai, Y.Q. *J. Polym. Res.* 2009, 16, 375-380.
- 325 21. Subramani, S., Lee, J.Y., Kim, J.H., and Cheong, I.W. *Compos. Sci. Technol.* 2007, 67,
326 1561-1573.
- 327 22. Lin, J., Wu, X., Zheng, C., Zhang, P., Huang, B., Guo, N., and Jin, L. *Appl. Surf. Sci.* 2014, 303,
328 67-75.
- 329 23. Sun, D., Miao, X., Zhang, K., Kim, H., and Yuan, Y. *J. Colloid Interf. Sci.* 2011, 361, 483-490.
- 330 24. Badamshina, E., Estrin, Y., and Gafurova, M. *J. Mater. Chem. A.* 2013, 1, 6509-6529.
- 331 25. Wu, Z., Wang, H., Tian, X., Xue, M., Ding, X., Ye, X., and Cui, Z. *Polymer*. 2014, 55, 187-194.
- 332 26. Schroeder, G., and Kurczewska, J. *Pol. J. Chem. Technol.* 2014, 16, 70-74.
- 333 27. Ding, J., and Ye, L. *J. Appl. Polym. Sci.* 2013, 127, 2732-2738.
- 334 28. Jena, K.K., Sahoo, S., Narayan, R., Aminabhavi, T.M., and Raju, K.V.S.N. *Polym. Int.* 2011, 60,
335 1504-1513.

- 336 29. Gao, X., Zhu, Y., Zhao, X., Wang, Z., An, D., Ma, Y., Guan, S., Du, Y., and Zhou, B. *Appl.*
337 *Surf. Sci.* 2011, 257, 4719-4724.
- 338 30. Takai, C., Fuji, M., and Takahashi, M. *Colloid. Surface. A.* 2007, 292, 79-82.
- 339 31. Anastasescu, C., Anastasescu, M., Teodorescu, V.S., Gartner, M., and Zaharescu, M. *J.*
340 *Non-Cryst. Solids.* 2010, 356, 2634-2640.
- 341 32. Xu, J., Shi, W., and Pang, W. *Polymer.* 2006, 47, 457-465.
- 342 33. Mamunya, Y.P., Shtompel, V.I., Lebedev, E.V., Pissis, P., Kanapitsas, A., and Boiteux, G. *Eur.*
343 *Polym. J.* 2004, 40, 2323-2331.
- 344 34. Xu, Q., Sardon, H., Chan, J. M., Hedrick, J. L., and Yang, Y. Y. *Polym. Chem.* 2015.
- 345 35. Zhang, S., Li, Y., Peng, L., Li, Q., Chen, S., and Hou, K. *Compos. Part. A-Appl. S.* 2013, 55,
346 94-101.
- 347 36. Shamsipur, M., Bahrami Adeh, N., Hajitarverdi, M.S., Yazdimamaghani, M., and Zarei, F. *Iran.*
348 *J. Chem. Chem. Eng.* 2013, 32, 1-7.
- 349 37. Zou, H., Wu, S., and Shen, J. *Chem. Rev.* 2008, 108, 3893-3957.
- 350 38. ASTM International. ASTM D1876-08.
- 351 39. ASTM International. ASTM D3163-14.
- 352 40. ASTM International. ASTM D638-10.
- 353 41. ASTM International. ASTM D3960-13.
- 354

355

Table 1. Physical properties of the nano-silica modified WPU adhesive

Parameter	Remarks
M_n	$55,000 \pm 1,500 \text{ g} \cdot \text{mol}^{-1}$
pH	7.0 ± 0.5
Particle size/ D_{50}	$800 \pm 50 \text{ nm}$
Viscosity(25°C)	$150 \pm 20 \text{ mPa} \cdot \text{s}$
Solid content	$51 \pm 1\%$
VOC content	$0.4 \pm 0.1\%$
Storage stability	$>9 \text{ months}$
Appearance	Milky white with blue light uniform liquid

356

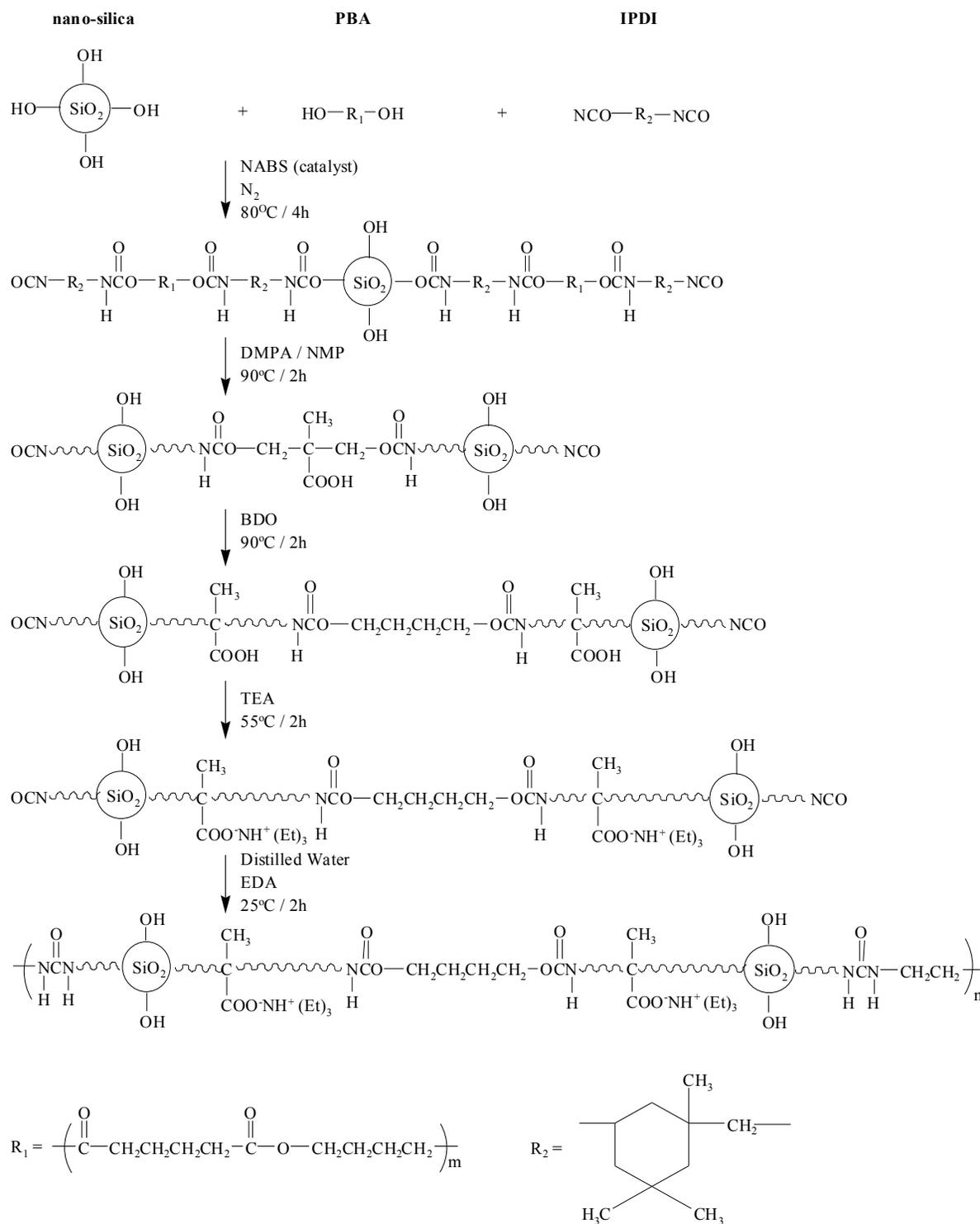
357

358

Table 2. The effect of nano-silica concentration on appearance and storage stability of WPU

Sample	Nano-SiO ₂ /%	Appearance	Storage life
WPU 0	0	Micro-blue transparent	$>9 \text{ months}$
WPU 1	1.0	Pale blue semi-transparent	$>9 \text{ months}$
WPU 2	1.5	Pale blue semi-transparent	$>9 \text{ months}$
WPU 3	2.0	Milky white with blue light	$>9 \text{ months}$
WPU 4	2.5	Milky white with blue light	$>9 \text{ months}$
WPU 5	3.0	Milky white with blue light	$<6 \text{ months}$

359

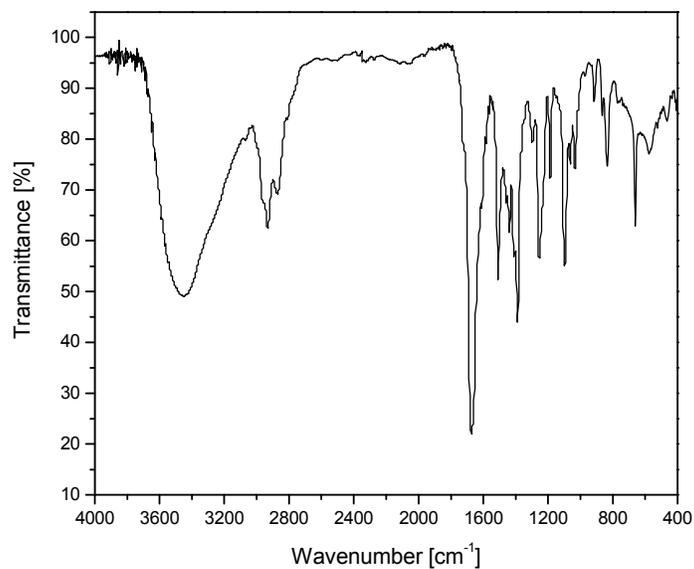


360

361

362

Figure 1. Synthetic scheme of nano-silica modified environmental friendly waterborne polyurethane adhesive

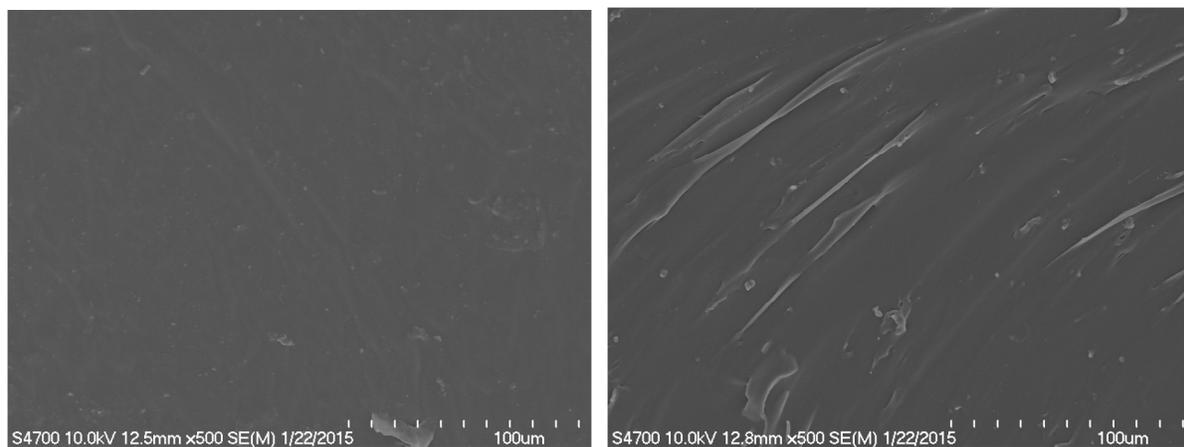


363

364

365

Figure 2. FTIR spectra of the synthesized nano-modified WPU (2.0 wt.% nano-silica)

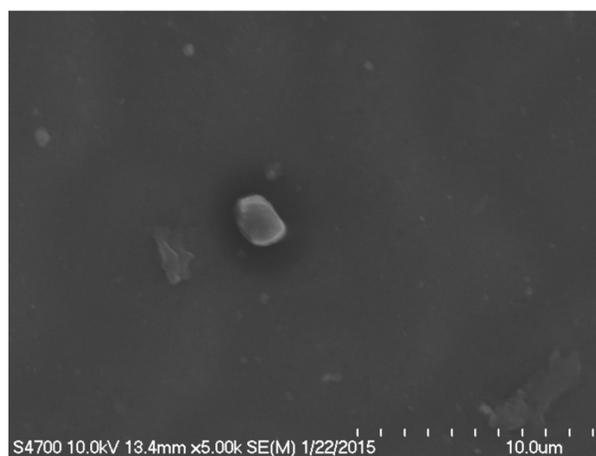


366

367

3(a)

3(b)



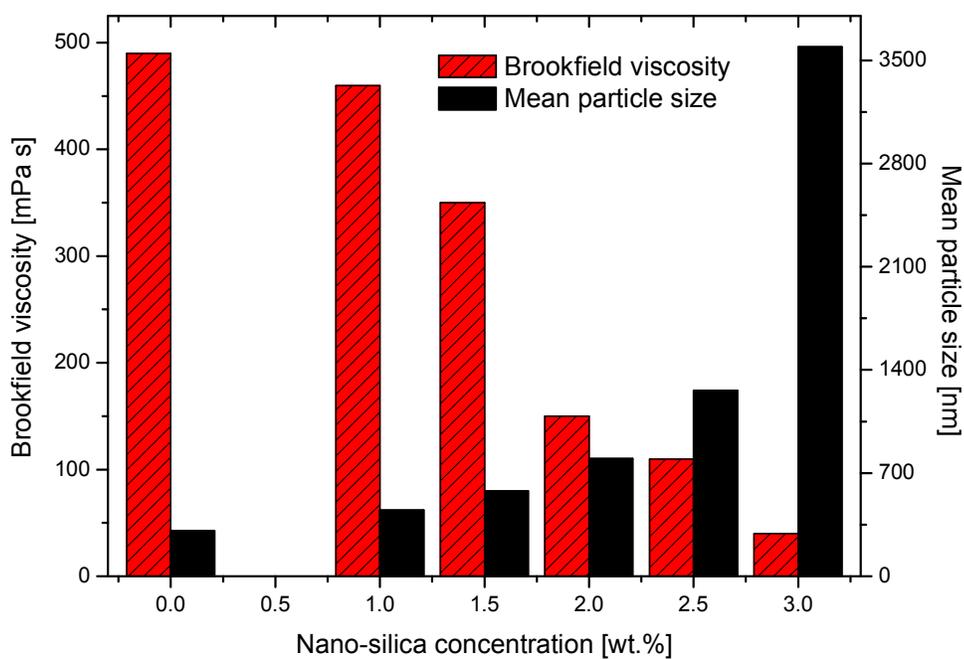
368

369

370

3(c)

Figure 3. SEM images of (a) pure-WPU and (b), (c) nano-modified WPU (2.0 wt.% nano-silica)

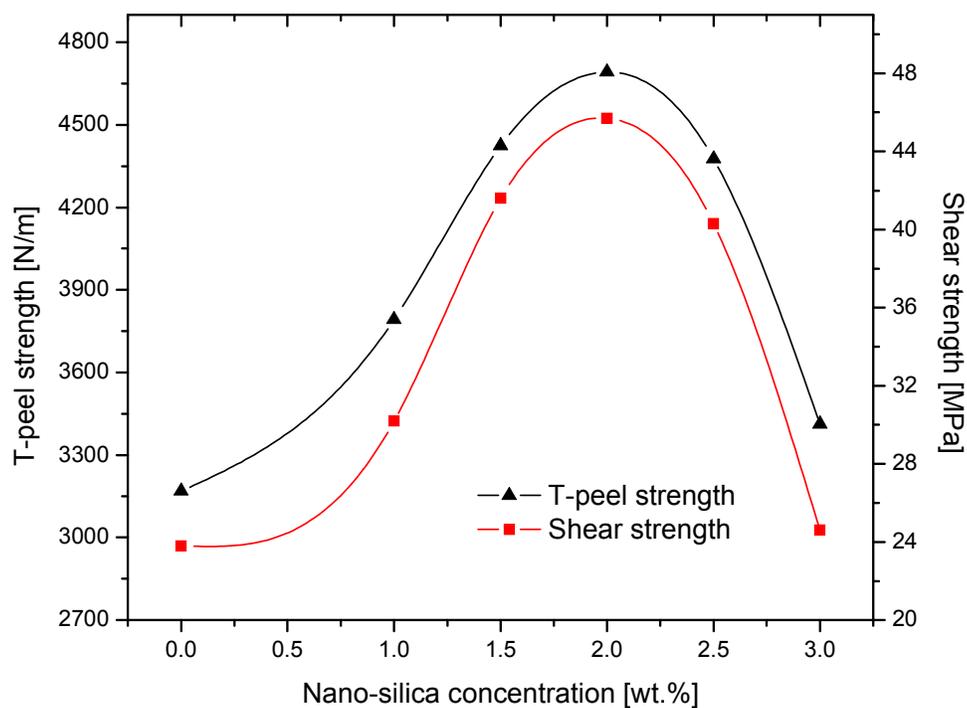


371

372

Figure 4. The effect of nano-silica concentration on Brookfield viscosity (25°C) and mean particle size (D_{50})

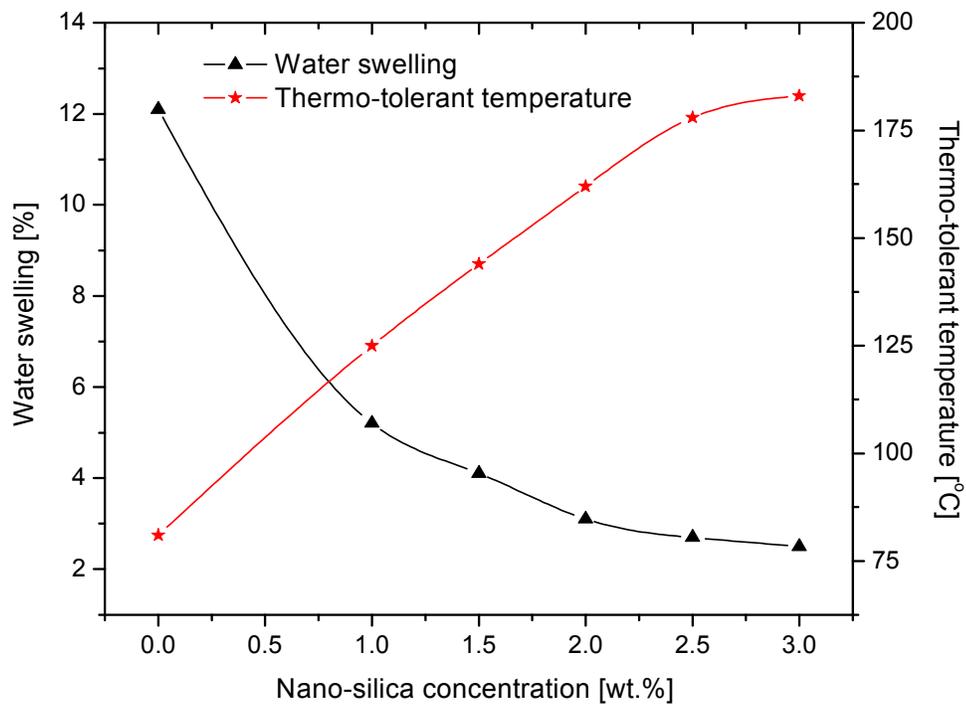
373



374

375

Figure 5. The effect of nano-silica concentration on adhesion properties

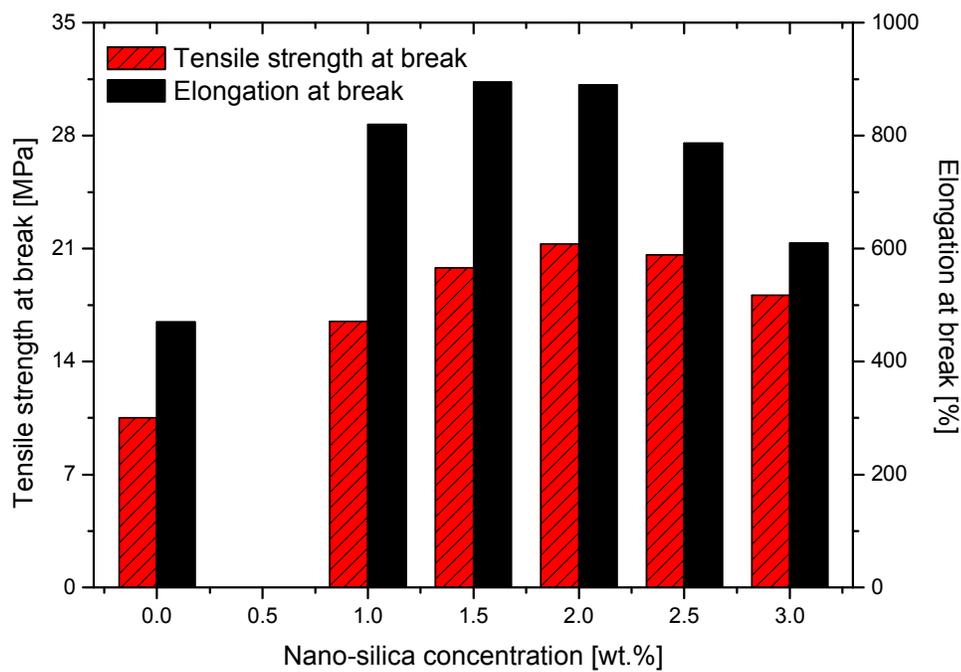


376

377

378

Figure 6. The effect of nano-silica concentration on water resistance and thermo-tolerant properties



379

380

Figure 7. The effect of nano-silica concentration on mechanical properties of WPU films