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Tuning of Photoluminescence Properties and Ultrasensitive Trace Detection in Few Layers MoS₂ by the Decoration of Gold Nanoparticles

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ABSTRACT

We report an easy and inexpensive chemical route for the decoration of few layers MoS₂ with Au nanoparticles (NPs). The Au-NPs are formed on the defective sites of the MoS₂ and localized by non-covalent bond. The NPs act as p-type dopant in MoS₂ layer. An enhancement of the photoluminescence (PL) intensity of the Au-MoS₂ composite with respect to bare few layers MoS₂ has been observed. We also systematically observed blue shift of the excitonic emission as the number and size of the Au-NPs on MoS₂ increases. Both the phenomena have been understood by switching between charged exciton (trion) recombination and neutral exciton recombination. The application aspects of the Au-MoS₂ composite have been demonstrated by using it as a substrate for surface-enhanced Raman scattering (SERS). The SERS measurements show a uniform, reproducible, and strong Raman signal from the adsorbed molecules with concentration down to 10⁻¹² M. Our results provide a method to tune the optical and electronic properties of MoS₂ and the Au-MoS₂ composite might be useful as an efficient SERS substrate for the ultrasensitive detection of biomolecules.

Keywords: Au-MoS₂ hybrid structure, tuning of photoluminescence properties, Raman study, SERS, high enhancement factor.

Introduction:

Layered transition-metal dichalcogenides (LTMDs) have been drawn attention in recent years due to its flamboyant optoelectronic properties.¹⁻³ LTMDs are represented by a general formula of MX_2 where, M is IVB, VB or VIB transition metal and X is sulphur, selenium, or tellurium.⁴ MoS_2 are belonging to the family of LTMDs. Being a zero bandgap material; pristine graphene invokes a constraint on uses of it in visible range of frequencies.⁵ In MoS_2 , Mo layer is sandwiched between two hexagonal S layers. The atomic layers of MoS_2 are held together by weak van der Waals force and due to this reason; layers can be exfoliated very easily with different techniques. Monolayer MoS_2 is a direct bandgap semiconductor of energy ~ 1.9 eV^{3,6,7}, due to which it shows strong photoluminescence.^{1,8} Therefore, monolayer MoS_2 has been used in several optoelectronic applications such as, interband tunnel field effect transistors (FETs),⁹ photodetectors,¹⁰⁻¹² photovoltaics¹³⁻¹⁵ and light emitters.^{3,16} In mono/few layers MoS_2 , due to the presence of strong Coulomb interactions optically generated electron-hole pair form stable exciton states even at room temperature.^{17,18} Therefore, the carrier density has an important role in the optical properties of mono/few layers MoS_2 . Doping is one of the effective techniques to control the carrier density of the monolayer TMDs.¹⁷⁻²¹ Among different doping methods such as application of gate bias voltage using a FET device,¹⁸⁻²⁰ gas physisorption²¹ etc; chemical doping is a simpler, cheaper and convenient technique for controlling the carrier density of layered materials.^{17,22,23} Mouri *et al.*¹⁷ have shown an enhancement and reduction of PL intensity when monolayer MoS_2 is chemically doped by p-type and n-type dopant, respectively.¹⁷ It has also been reported that the Au-NPs attached at the edge sites of monolayer MoS_2 act as p type dopant.^{24,25} However, Bhanu *et al.*²⁵ have observed quenching of PL in Au- MoS_2 hybrid structure, which is inconsistent with the result published in Ref.17. To understand the

optoelectronic properties and their consecutive application of MoS₂ and Au decorated MoS₂ requires a careful investigation on their structure.

In this work, few layers MoS₂ has been prepared from the bulk MoS₂ material by well known micromechanical cleavage or “scotch tape” method.^{1,26,27} A 0D-2D composite structure has been synthesized by immersing the MoS₂ into HAuCl₄ solution. The Au-NPs (0D) are formed at the defective sites of MoS₂ (2D) and the number and size of the Au-NPs increase with the increase of immersing time. The intensity of the excitonic emissions are enhanced as the Au-NPs start growing on the MoS₂ layer. This observation has been attributed to the fact that the Au-NPs act as p type dopant which reduces the intrinsic electrons in the bare sample yields an increase of exciton recombination. In addition to that the excitonic emission energy of composite system are blue shifted compared to the bare MoS₂, which is due to the increase of Fermi energy by the dissociation of charged exciton (trion). Stiffening of E_{12g}¹ and softening of A_{1g} Raman modes with the increase in immersing time has been observed which is consistent with an earlier report.²⁸ The potential applicability of the Au-MoS₂ hybrid structure has been examined by using it as an surface enhanced Raman scattering (SERS) substrate. A uniform and reproducible SERS signals have been observed up to a 10⁻¹² M concentration of Rhodamine 6G (R6G) dye. The enhancement factor we have achieved is sufficient to detect a single molecules. Hence, we conjecture that, the optical properties of few layers MoS₂ can be tuned by decoration of Au-NPs and the Au-MoS₂ hybrid structure can be utilized for ultrasensitive trace detection that might overcome the issues of low cross sections of Raman scattering.

Experimental Section

Reagents.

High purity bulk MoS₂ was purchased from SPI Suppliers, gold (III) chloride tetrahydrate (HAuCl₄, 4H₂O) and Rhodamine 6G (R6G) were purchased from Sigma-Aldrich Pvt. Ltd. All chemicals were used as received without further purification. Mili-Q water was used during entire synthesis and experiments.

Preparation of Few layers MoS₂.

Few layers MoS₂ flake was prepared on the SiO₂ (300nm)/Si substrate by micromechanical exfoliation technique.^{1,26} Before deposition of flake Si wafers were freshly cleaned by Piranha solution.

Preparation of Au-MoS₂ hybrid structure.

A 1 mM solution of HAuCl₄ was prepared by mixing proper amount of HAuCl₄ with milli pore water. Four different sets of sample were prepared by just varying the immersing time of MoS₂ flake containing SiO₂/Si substrate into HAuCl₄ solution for 30 min, 60 min, 90 min and 120 min and those samples were marked as S1, S2, S3 and S4, respectively.

Preparation of SERS substrate.

A Stock solution (10⁻³M) of R6G was prepared with desired volume of water and stored in a cold place. Wide range of solutions having different concentration of R6G from 10⁻⁵ to 10⁻¹² M was prepared from stock solution. The Sample S4 was used as SERS substrate. The sample was immersed into R6G solution for two hours and then rinsed with Mili-Q water and then dried with nitrogen flow. Finally, we have performed SERS measurements on this dye adsorbed substrates. The same procedure has been followed for all other concentration of R6G solutions.

Instruments used for various characterizations.

SEM analysis was done using model FEI, Quanta 200 with an accelerating voltage of 30kV. The Raman and PL measurements were performed using a micro Raman set up consisting of spectrometer (Lab RAM HR Jovin Yvon) and a Peltier cold CCD detector. An air cooled argon-ion laser (Ar^+) with wavelength 488 nm and diode laser of wave length 785 nm were used as excitation light source and a 100X objective with numerical aperture (NA) of 0.9 was used to focus the laser on the sample and to collect the scattered light from the sample.

Results and Discussions

Figure 1(a) shows the optical microscope image of the MoS_2 flake on SiO_2/Si substrate. The optical image has been taken using 100X objective. The contrast in the area of the image marked by red circle indicates that the few layers of MoS_2 are present at that region. The number of layers has been confirmed by Raman spectroscopy. Raman spectrum of the sample has been displayed in Figure 1(b). For reference we have also presented the Raman spectrum of bulk MoS_2 . In MoS_2 , E_{2g}^1 phonon mode is originated due to the displacement of Mo and S atoms in the basal plane, whereas A_{1g} phonon mode is due to the vibration of S atom in opposite direction, along the perpendicular to the basal plane (see inset of Figure 1(b)). The peak positions of the E_{2g}^1 and A_{1g} modes are obtained from the fitting of experimental curve using two Lorentzian functions. The Raman modes are highly sensitive to the thickness of the layer.²⁷ In general, the gap between E_{2g}^1 and A_{1g} peaks decreases as the number of layers decreases.²⁷ The gap for our MoS_2 layer is found to be 24.97 cm^{-1} , which ensures that the number of layers is less than 6.²⁷ We have also performed PL measurements for further confirmation of the layer number. Figure 1(c) shows the PL spectra of bulk MoS_2 (orange curve) and few layers MoS_2 (black curve). The large spin-orbit interaction in the few

layers MoS₂ splits the highest valance band located at the K point of the Brillouin zone. Therefore, the direct optical transitions from the top of the spin-split valance bands to the bottom of the conduction band give A1 and B1 peaks.¹ Both the peaks are associated with exciton recombination. In addition to that, PL peak of negative exciton (X⁻, trion) is also present in micromechanical exfoliated MoS₂ layer, because, the layer unintentionally becomes n-type due to defects or substrate effect.^{17,18,29} Therefore, we have fitted the PL spectrum using three Gaussian functions and the peak positions are at 1.80 eV, 1.86 eV and 1.96 eV corresponding to X⁻, A1 and B1, respectively. The energy gap between X⁻ and A1 is ~60 meV, which is consistent with the value reported for mono layer MoS₂.^{17,18,29} The intensity ratio of A1 and B1 is ~1.3, which is close agreement with reported value for 4 layers MoS₂,²⁵ corroborates the number of layers in our material is less than 6 as obtained from Raman measurements.

Scanning Electron Microscope (SEM) images of Au-MoS₂ hybrid structures (S2 and S4) are shown in Figures 2(a) and (b), respectively. From the SEM images it is clear that the number and size of Au-NPs are higher in S4 compared to S2 and no Au-NPs are observed on bare SiO₂ substrate. In MoS₂, defective sites are more reactive.²⁴ The AuCl₄⁻ ions in the solution initially attract by the defective sites of the MoS₂ and act as nucleation centre for growth of Au particles.²⁴ So, with the increase of immersing time the number and dimension of the Au-NPs increase as observed in SEM images.

To investigate the role of Au agglomeration on MoS₂, PL measurement has been performed on Au-MoS₂ hybrid structure with excitation wavelength of 488 nm. Figure 3 shows the PL spectra of undoped MoS₂ (black), S1 (red), S2 (blue), S3 (magenta) and S4 (dark yellow) samples. All the spectra have been fitted with three Gaussian functions (for X⁻, A1 and B1). Figure 3(b) shows the integrated intensity of the PL peaks as a function of

immersing time of MoS₂ inside the HAuCl₄. In the as-prepared sample the PL intensity is higher for trion (X⁻) at energy 1.80 eV than the excitonic emission. This is because as-prepared MoS₂ unintentionally becomes n-type due to defects or substrate effect^{18,29,30} where trion recombination is dominant and the result is consistent with the earlier reports.¹⁷ The intensity of the neutral excitonic peaks increase monotonically up to immersing time of 60 min (sample: S3) and then mitigates, whereas the integral intensity of the trion peak remains almost constant. The experimental results are also consistent with earlier reports.¹⁷ This behaviour can be attributed to the fact that in presence of Au-NPs the electronic structure of MoS₂ has been modified. Due to the difference of Fermi energy between MoS₂ (4.7 eV)²⁵ and Au (5.1 eV),²⁵ a band bending takes place in Au-MoS₂ hybrid structure. Upon illumination with a photon of appropriate energy the electron in the excited states of MoS₂ transfer to Au leaving behind a hole, which acts as p-dopant in MoS₂. As a result, the excess carrier in the MoS₂ layer decreases and the excitons recombination became predominant than the formation of trion. Moreover, the trions dissociate to excitons and one of the two electrons in the trion is unbound and placed at the Fermi level of the MoS₂.¹⁸ Therefore, the number of exciton increases with the increase of dipping time, yields the enhancement of intensity of the excitonic emission. Beyond 60 min, the PL intensity decreases because of the excess p doping. Figure 3(c) shows the PL energy shifts towards higher energy with the increase of immersing time of MoS₂ into the HAuCl₄ solution. As the exciton binding energy is large,¹⁸ neutral excitons emit higher energy than trion. The increased number of neutral excitons yields the PL energy shift.²⁹ The relative concentration of charged to neutral excitons can be estimated by the following expression.^{17,18,29}

$$\gamma = I_{X^-} / (I_{X^0} + I_{X^-})$$

where, I_i is the emission intensity of neutral or charged excitons (X^0 or X^-). The estimated values of γ are 0.63, 0.43, 0.16, 0.33 and 0.45 for as-prepared few layers MoS₂, S1, S2, S3 and S4, respectively. The experimental observation reflects that, lower value of γ contributes the intensity enhancement and blue shift of excitonic PL emission.

To enlighten how the adsorbed Au-NPs create impacts on the vibrational mode, Raman measurements have been performed on Au-MoS₂ hybrid structures. Figure 4(a) shows the Raman spectra of all the samples including as prepared MoS₂. The positions of E_{2g}¹ and A_{1g} modes as a function of immersing time of MoS₂ into the HAuCl₄ solution are displayed in Figure 4(b). The A_{1g} mode shows a strong dependence on the dipping time as the carrier concentration changes with dipping time whereas E_{2g}¹ shows less dependence on the carrier concentration. We have observed softening and broadening of the A_{1g} phonon mode as we increase the immersing time, which is due to the interaction of plasmon with A_{1g} phonon. The maximum shift of A_{1g} mode is found to be 4 cm⁻¹, which is consistent with electron doped single layer MoS₂ by Chakraborty *et al.*²⁸

To explore the utility of the metal-semiconductor composite system as an ultrasensitive trace detection, the Au-MoS₂ hybrid structure have been used as a substrate for surface enhanced Raman spectroscopy (SERS) study. The two well known effects that are involved in SERS; chemical and electromagnetic mechanism (EM) effects. The chemical effect is originated due to the charge transfer between metal nanostructure and the molecule adsorbed at the metal surface. The localized electromagnetic field, originating from the resonant coupling of the incident light with surface plasmons is the origin of EM effect. The EM effect dominates to enhance the Raman signal in the SERS phenomenon.³¹⁻³⁸ The plasmonic coupling effects at the inter particle gap in a cluster of noble-metal particles provide an intensively localized electromagnetic field which is known as hot spots. The Raman active molecules adsorbed at the hot spots can be detected at single-molecule

level.^{31,39-46} A lot of efforts have been focused on the preparation of stable metal clusters with narrow inter particle gaps. DNA has been widely employed as a scaffold to produce specifically dispersed metal nanoparticles and to control the locations of these particles.⁴⁷⁻⁵² Recently, metal nanoparticles decorated graphene and graphene oxide have been used as efficient SERS substrate to detect molecules with nM concentration level.⁵³ In the present work, the Au-NPs are tightly immobilized by MoS₂ and provide the abundant hot spots. R6G has been used as a Raman probe molecule. The normal Raman spectrum of solid R6G dye is presented in Figure 5. The characteristic bands of R6G are observed at 609, 775, 1187, 1305, 1360, 1506, 1572 and 1648 cm⁻¹. Peaks at 609 and 775 cm⁻¹ are due to in plane xanthene ring deformation and out of plane C-H bend, respectively.⁵⁴ The mode at 1187, 1305 and 1360 cm⁻¹ are due to the in plane xanthene ring deformation (C-H bend, N-H bend), the in plane xanthene ring breathing (N-H bend, CH₂ wag) and in plane xanthene ring stretching (C-H bend), respectively.⁵⁴ The bands at 1506, 1572 and 1648 cm⁻¹ are related to in plane xanthene ring stretching (C-H bend, N-H bend), in plane xanthene ring stretching (N-H bend) and in plane xanthene ring stretching (C-H bend), respectively.⁵⁴

The SERS measurements have been performed using S4 sample with R6G solutions having concentrations of 1×10^{-5} , 1×10^{-6} , 1×10^{-9} , and 1×10^{-12} M. The SERS spectra at various concentrations recorded with 488 nm and 785 nm laser are shown in Figures 6 (a) and (b), respectively. The normal Raman spectrum of R6G with 1mM concentration (black curve) does not show any Raman band. But, Raman modes of R6G molecules on Au-MoS₂ substrate are clearly visible up to 10^{-12} M concentration level. The enhancement is higher for 488 nm laser than 785 nm laser, which is expected because 785 nm laser is far from plasmon resonance of Au-NPs. To get a quantitative idea of enhancement strength, the enhancement factor (EF) has been calculated from the experimental data. The estimated value of EF for the modes at 609, 1358, 1575 and 1648 cm⁻¹ is $\sim 10^{10}$, which is sufficient to detect single

molecules.⁵² We strongly believe that, in future the Au-MoS₂ hybrid structure will be useful for ultrasensitive detection of biomolecules and other technologically important fields.

Conclusions

In summary, we have presented a simple technique to realize 0D-2D hybrid structure in few layers MoS₂. The Au-NPs in this hybrid structure exhibit p-doping effect to the MoS₂. The hybrid structure has been studied systematically by PL and Raman measurements. We have demonstrated that the number and size of the Au-NPs play an important role to control the optical and vibrational properties of the MoS₂. The excitonic emission properties are highly modified in the presence of Au-NPs whereas the emission from trion is quite insensitive. The intensity and blue shift of the PL peaks have been understood by the conversion of trion to exciton under illumination of the hybrid structure. We have also shown that the E_{2g}¹ phonon mode shows less dependence on the presence of Au-NPs compared to A_{1g} mode. Finally, we have utilized the hybrid structure as a potential SERS substrate for the detection at single molecule level. Hence, the Au-NPs decorated MoS₂ can provide hybrid nanomaterials for the application of future digital electronics and sensors.

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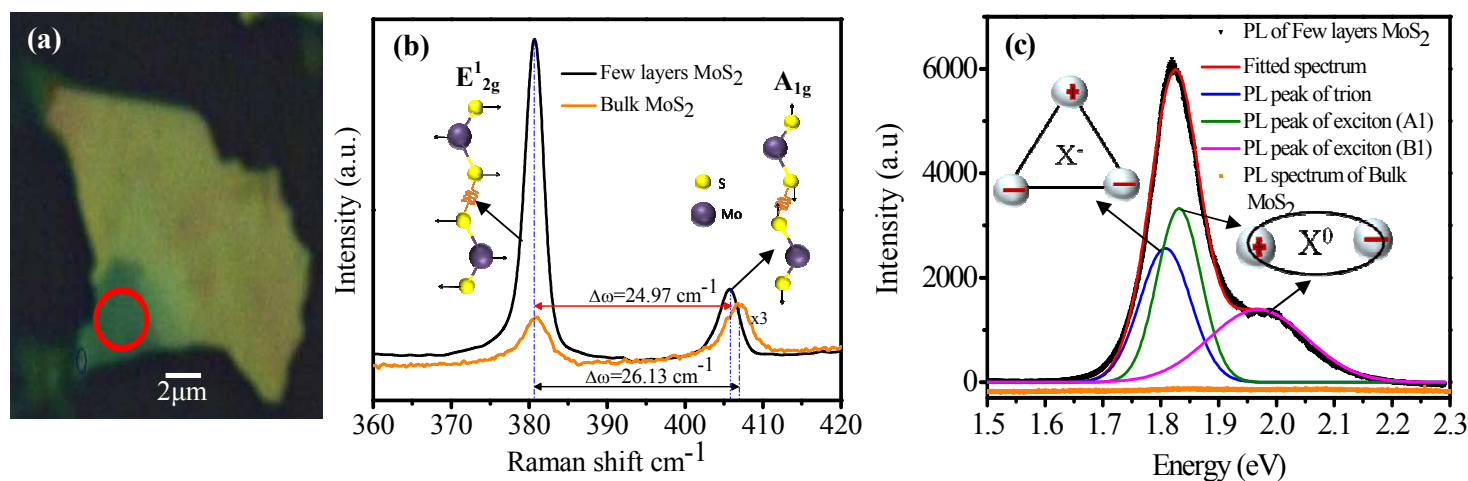


Figure 1: (a) Optical microscope image of few layers MoS₂. The measurements have been done at the place marked by red circle. (b) Raman spectrum of bulk (orange curve) and few layers (black curve) MoS₂. The schematic structure of the atomic displacements of the E_{12g}¹ and A_{1g} are shown in the inset of the figure. (c) PL spectrum of bulk (orange curve) and few layers (black curve) MoS₂. The red line is the result of a fit to the experimental data of few layers MoS₂ with three Gaussians functions for X⁻ (blue), A1 (green), and B1 (magenta). The schematic structure of trion and exciton are shown in the inset

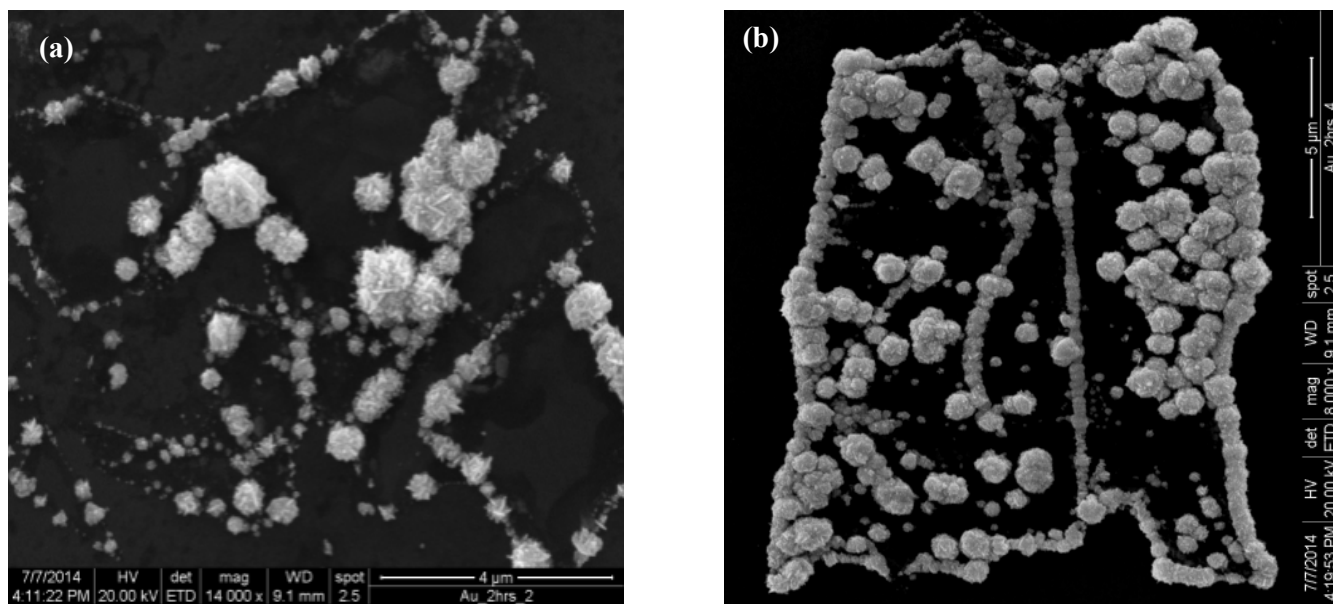


Figure 2: SEM images of (a) Sample S2 (b) Sample S4

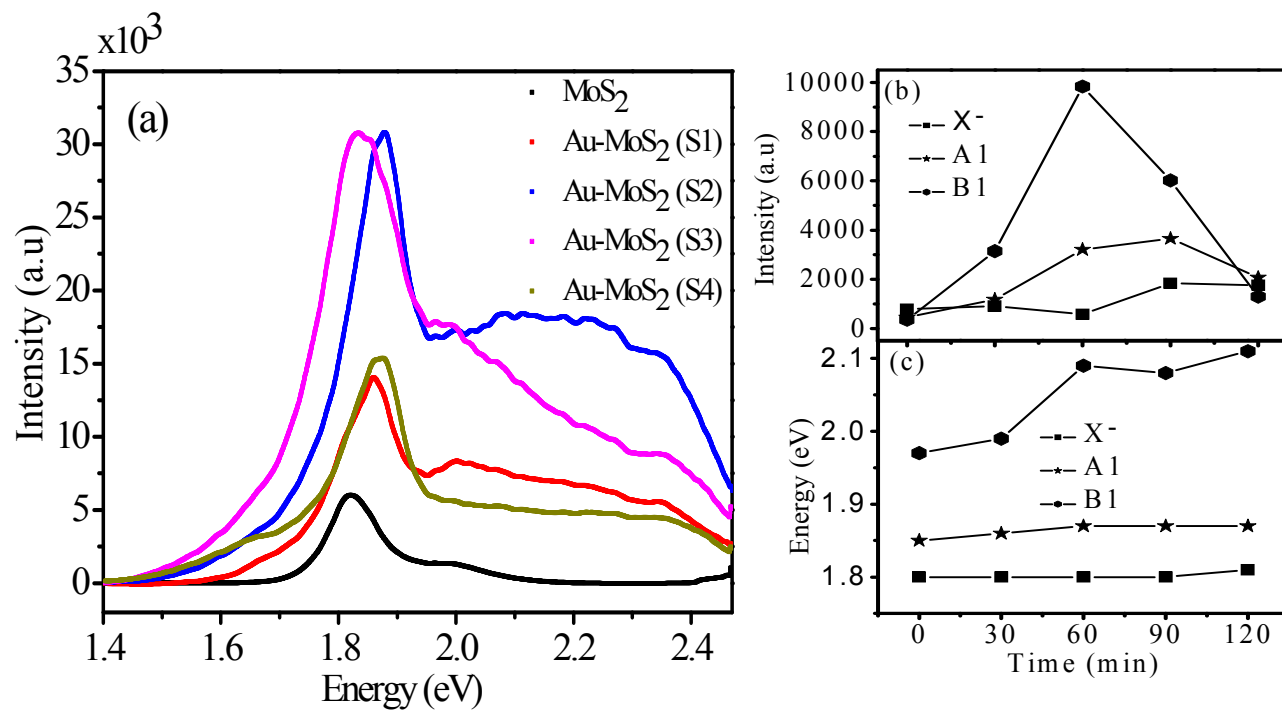


Figure 3: (a) PL spectra of few layers MoS₂ obtained at different dipping time [S1 (30 min), S2 (60 min), S3 (90 min) and S4 (120 min)] inside the HAuCl₄ solution. (b) Intensity and (c) energy of X⁻, A1 and B1 peaks as a function of dipping time.

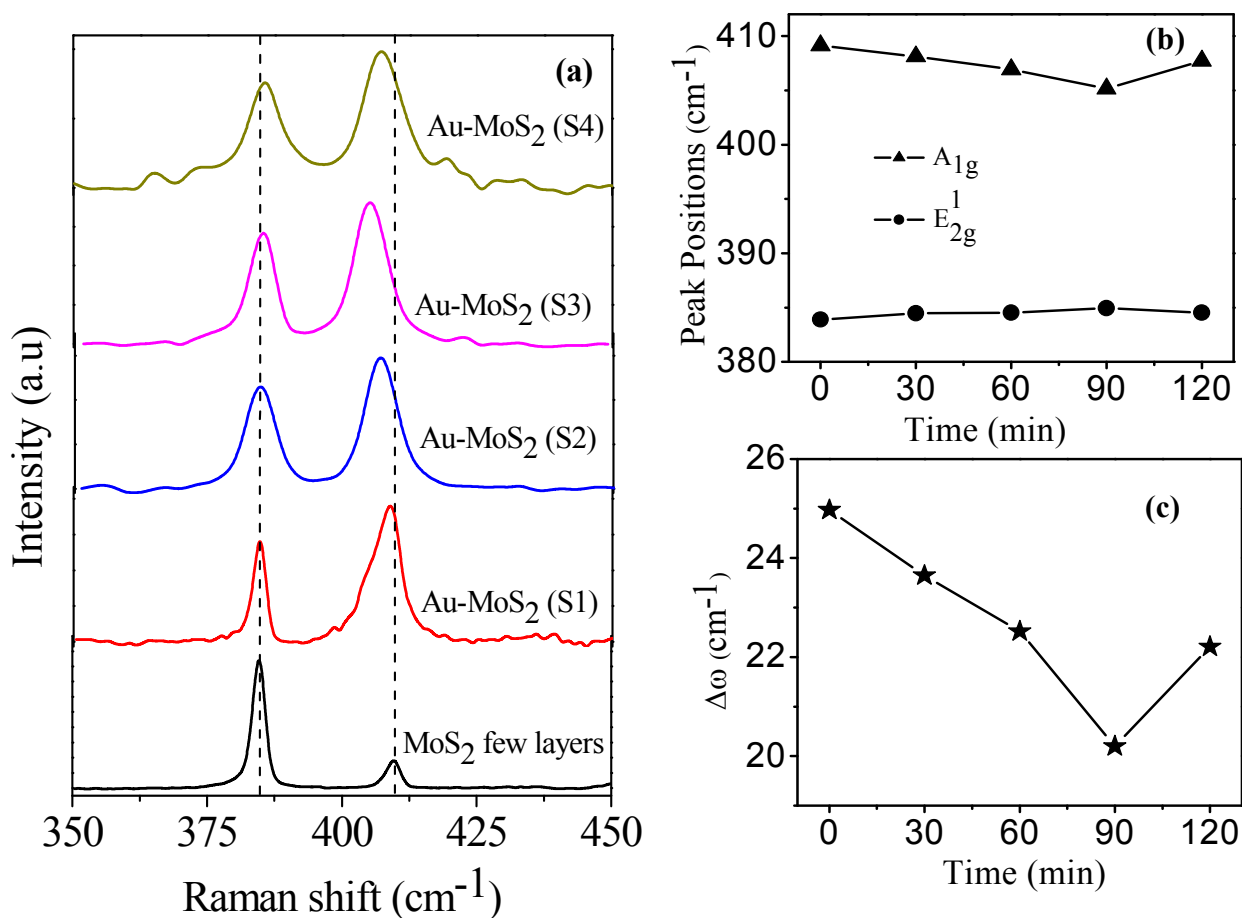


Figure 4: (a) Raman spectra of few layers MoS₂ obtained at different dipping time [S1 (30 min), S2 (60 min), S3 (90 min) and S4 (120 min)] inside the HAuCl₄ solution. (b) variation of E_{2g}¹ and A_{1g} peaks positions as a function of dipping time (c) gap between E_{2g}¹ and A_{1g} modes as a function of dipping time.

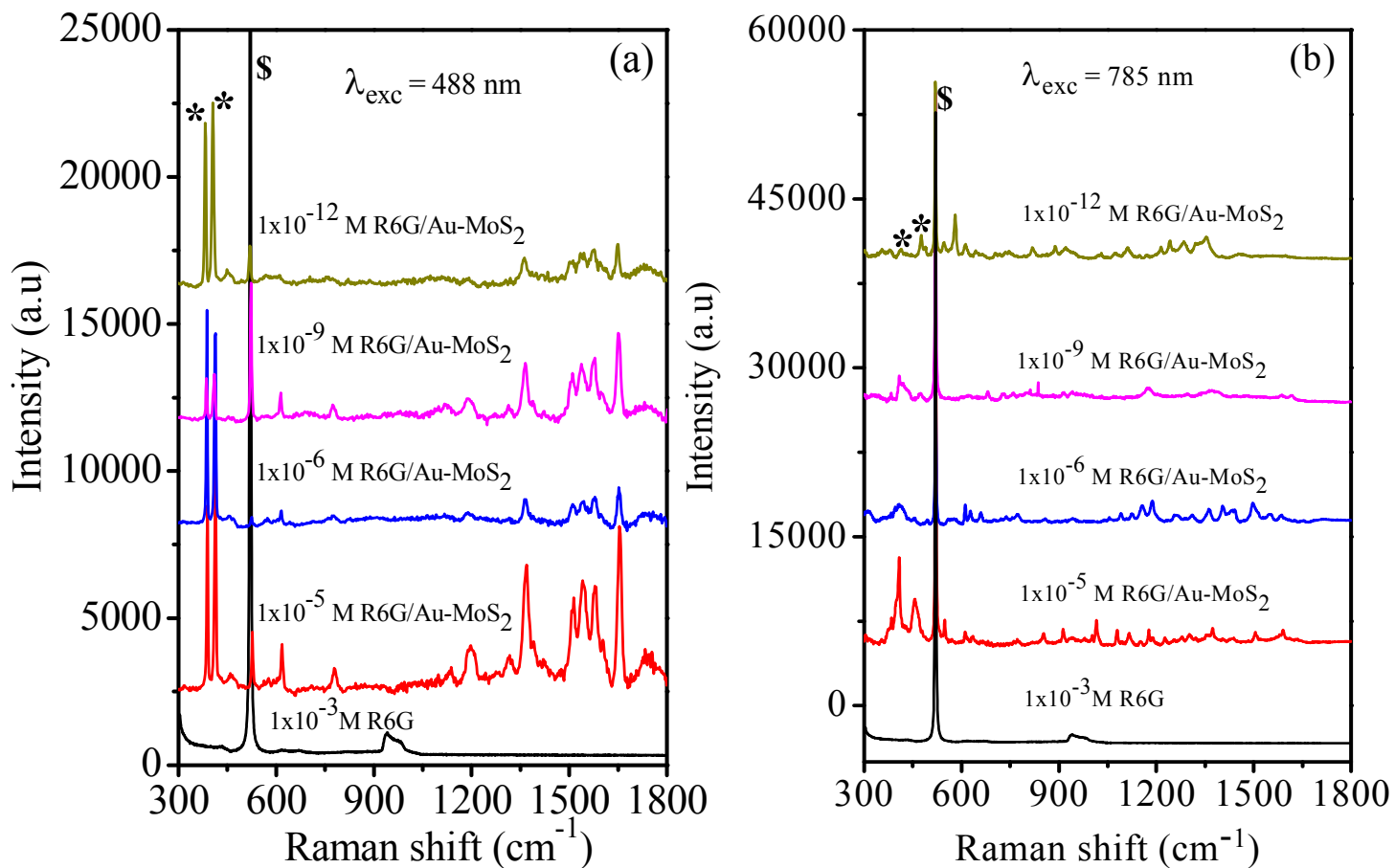
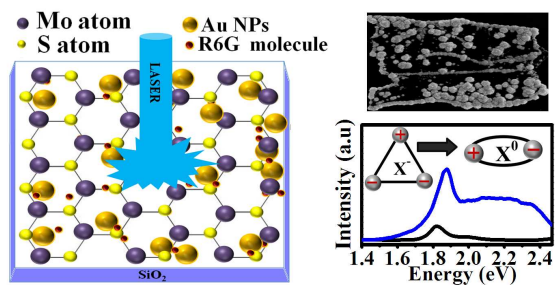


Figure 6: Raman spectra of 1 mM R6G solution (black) SERS spectra of R6G at concentrations of 1×10^{-5} M (red), 1×10^{-6} M (blue), 1×10^{-9} M (magenta), 1×10^{-12} M (dark yellow) with excitation wavelength (a) 488 nm and (b) 785 nm. The asterisk (*)-marked peaks are from MoS₂ and \$ marked peak is from the Si substrate.

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Tuning of Photoluminescence Properties and Ultrasensitive Trace Detection in Few Layers MoS₂ by the Decoration of Gold Nanoparticles

Shib Shankar Singha, Dipanjan Nandi, Achintya Singha*



Tuning optical properties and ultrasensitive molecular detection by Au nanoparticles ornamented MoS₂.

