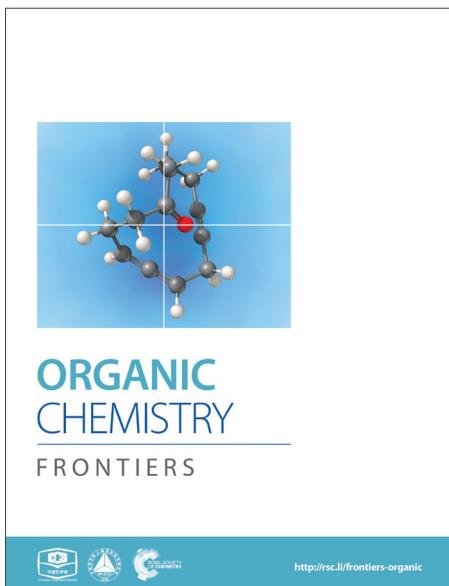
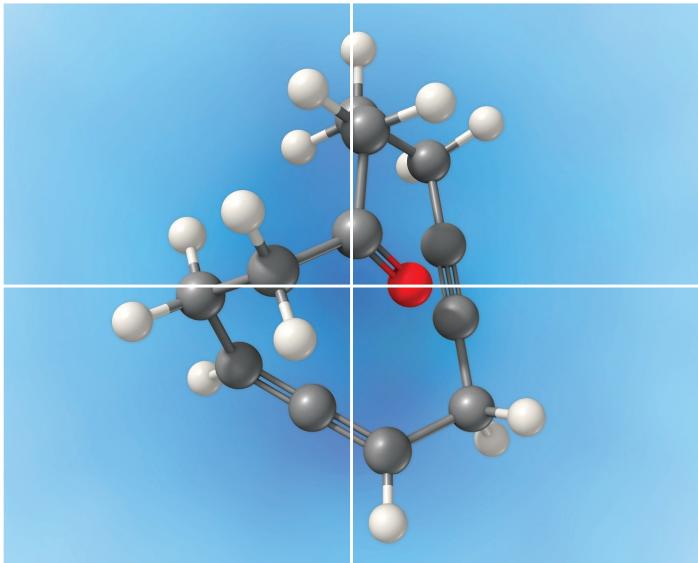


ORGANIC CHEMISTRY FRONTIERS

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Journal Name

ARTICLE

Received 00th January 20xx,
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

A concise total synthesis of sespenine, a structurally unusual indole terpenoid from *Streptomyces*

Yu Sun,^a Zhanchao Meng,^a Pengxi Chen,^a Deliang Zhang,^a Martin Baunach,^b Christian Hertweck,^b and Ang Li^{a,*}

Sespenine is a structurally unusual indole sesquiterpenoid isolated from endophytic *Streptomyces* sp. HKI0595. Herein, we report a ten-step (the longest linear sequence) synthesis of this molecule from commercially available materials, on the basis of our first generation synthesis. Sharpless asymmetric epoxidation and Stille–Miyata coupling were used to construct a functionalized epoxy ester, which underwent Ti(III) mediated reductive radical cyclization to give a *trans*-decalin intermediate with a 2-methoxycarbonylindole side chain. Oxidation of this compound afforded a pair of epimeric 3-hydroxyindolenines, and the major isomer entered a bioinspired cascade of Prins cyclization/Friedel–Crafts/retro Friedel–Crafts under acidic conditions, to furnish the polycyclic core of sespenine. Sespenine analogues bearing different C2 substituents were prepared with similar chemistry. Xiamycin A, a carbazole congener of sespenine, was synthesized from the minor hydroxyindolenine epimer as well.

Introduction

Indole terpenoids have been of increasing interest in the area of chemical synthesis^{1–3} and biosynthesis⁴ because of their fascinating structures and promising biological activities. Recently, we have accomplished total syntheses of a series of indole terpenoids such as those from the anominine, drimentine, mycoleptodiscin, and halalindole families,⁵ during which some useful methods and strategies have been developed.⁶ In 2011, Ding et al. discovered sespenine (**1**, Figure 1), a structurally unusual polycyclic molecule from *Streptomyces* sp. HKI0595,⁷ a bacterial endophyte of the mangrove *Kandelia candel*, together with a number of biogenetically relevant indole sesquiterpenoids such as xiamycin A (**2**) and indosespene (**3**). Soon after, Zhang and coworkers independently isolated several xiamycin A related natural products.⁸ This indole terpenoid family quickly drew attentions from the perspectives of biosynthesis and chemical synthesis.⁹ In 2014, we accomplished the first total synthesis of sespenine (**1**),¹⁰ which featured a Lewis acid promoted conjugate addition and a bioinspired cationic cascade reaction. The Baran group disclosed

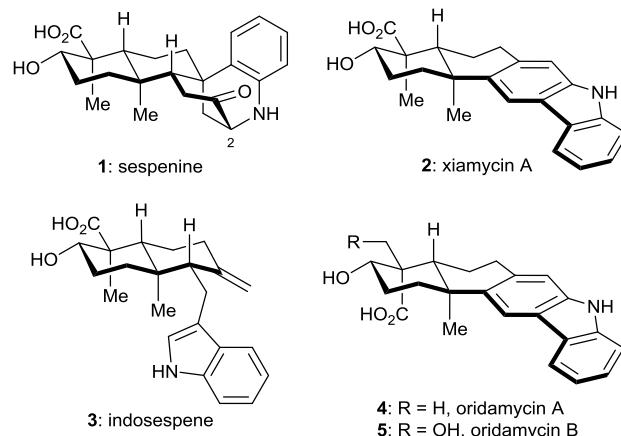


Figure 1. The structures of sespenine and related indole sesquiterpenoids.

elegant syntheses of xiamycin A (**2**) and dixiamycin B; the latter possessing an N–N linkage was assembled through an electrochemical oxidative dimerization of the former.¹¹ We then reported syntheses of xiamycin A, indosespene, oridamycins A and B (**2–5**, Figure 1), and dixiamycin C¹² using two parallel strategies of 6π electrocyclization/aromatization^{13–15} and indole C2 C–H bond activation/Heck annulation.¹⁶ Herein, we report the second generation syntheses of sespenine (**1**) and xiamycin A (**2**) as well as the preparation of sespenine analogues with various substituents at C2.

Results and discussion

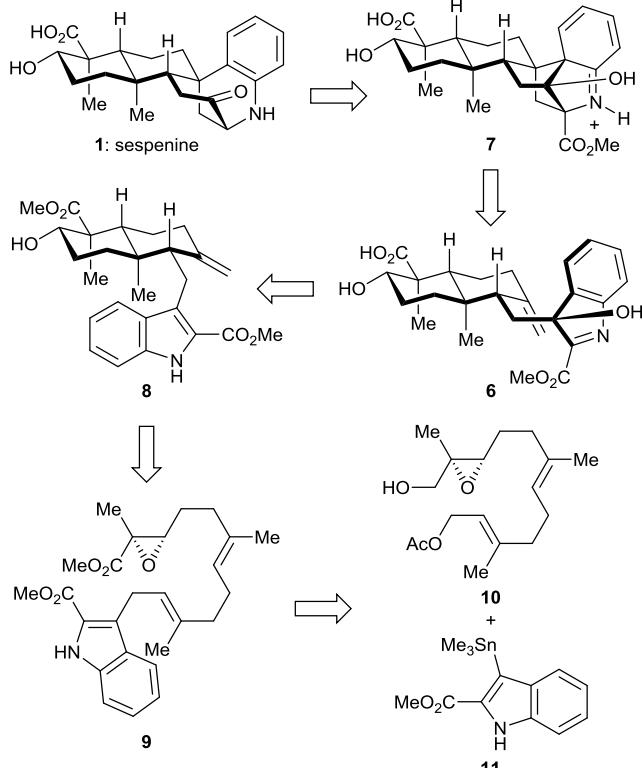
^a Y. Sun, Z. Meng, P. Chen, D. Zhang, Prof. Dr. A. Li
State Key Laboratory of Bioorganic and Natural Products Chemistry, Collaborative Innovation Center of Chemistry for Life Sciences, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China
E-mail: ali@sioc.ac.cn

^b M. Baunach, Prof. Dr. C. Hertweck
Leibniz Institute for Natural Product Research and Infection Biology, HKI, Jena, Germany

† Electronic Supplementary Information (ESI) available: ¹H and ¹³C NMR spectra of all new compounds. See DOI: 10.1039/x0xx00000x

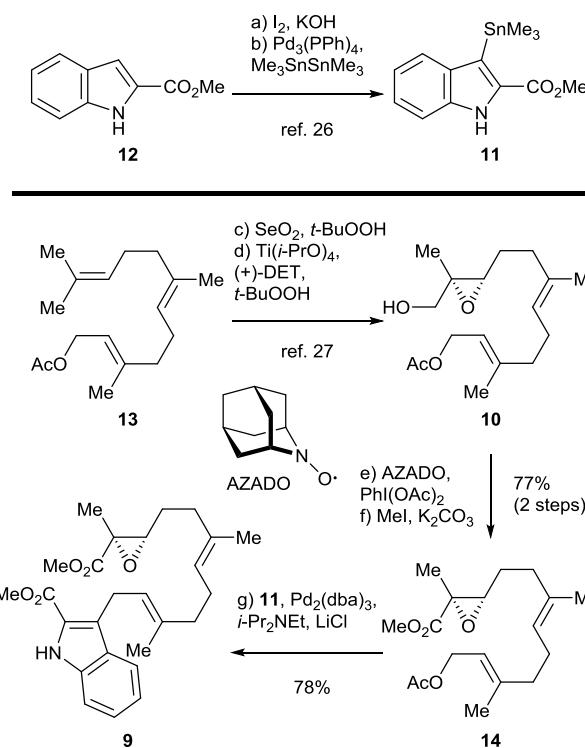
ARTICLE

Journal Name



Scheme 1. Retrosynthetic analysis of sespenine.

The retrosynthetic analysis of sespenine (**1**) is illustrated in Scheme 1. The molecule is traced back to an advanced intermediate **6**, which may undergo a cascade sequence of Prins cyclization/Friedel–Crafts/retro Friedel–Crafts to reach **1**, via the intermediacy of dearomatized compound **7**. It is well preceded that an electron rich arene serves as a suitable nucleophile to trap the carbocation generated through a Prins reaction.^{17,18} The retro Friedel–Crafts reaction from **7** to **1**, which can also be viewed as a retro aldol reaction, would take advantage of the driving force of re-aromatization. Notably, Ding et al. hypothesized a similar reaction cascade in the biosynthetic pathway toward **1**,⁷ and another relevant process was proposed for the biogenesis of aspernomine, a structurally more complex indole diterpenoid, by McWhorter and Tantillo, respectively.¹⁹ A subtle change in our plan of chemical synthesis of **1**, compared with the above mentioned biosynthetic speculations, is the existence of a C2 substituent (a methoxycarbonyl group in the case), which helps stabilizing the 3-hydroxyindolenine intermediate.^{20,21} Compound **6** is expected to arise from precursor **8** via chemoselective oxidation of its indole moiety in the presence of an exocyclic C=C bond. *trans*-Decalin **8** is simplified to α,β -epoxy ester **9**. Ti(III) induced epoxide opening radical reaction, which was pioneered by RajanBabu and Gansäuer,²² has been elegantly exploited by the groups of Barrero, Oltra, Fernández-Mateos, Cuerva, Roy, Reisman, etc. in complex molecule synthesis.²³ Borrowing from precedent¹² in our synthesis of indosespene (**3**), we may rely on a reductive opening of the epoxide followed by radical cascade cyclization catalyzed by Ti(III)

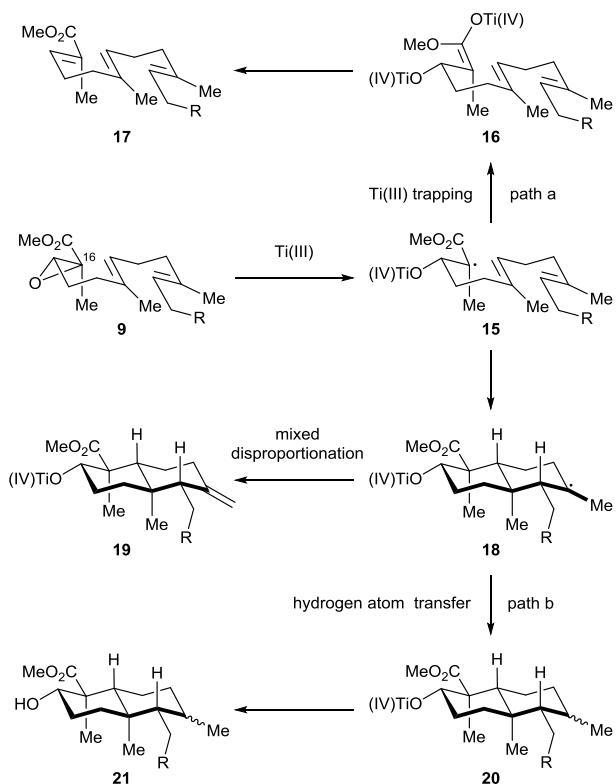


Scheme 2. Synthesis of functionalized α,β -epoxy ester **9**. Reagents and conditions: (e) AZADO (4 mol%), PhI(OAc)₂ (3.0 equiv), CH₂Cl₂, 22 °C, 12 h; (f) MeI (2.0 equiv), K₂CO₃ (1.5 equiv), DMF, 22 °C, 30 min, 77% (2 steps); (g) **11** (1.2 eq), Pd₂(dba)₃ (5 mol%), LiCl (3.0 equiv), *i*-Pr₂NEt (3.0 equiv), DMF, 50 °C, 1 h, 78%.

species in situ generated by Cp₂TiCl₂ and Mn, to prepare **8** from **9**. The indole motif should be tolerated under such conditions. Compound **9** would be disassembled into readily available building blocks allyl acetate **10** and stannane **11**, the recombination of which could be effected through allyl Stille–Miyata coupling.²⁴

The synthesis of sespenine (**1**) commenced with the construction of functionalized epoxy ester **9**, as shown in Scheme 2. We first explored the feasibility of Friedel–Crafts type allylation at the C3 position of commercially available methyl indole-2-carboxylate (**12**). However, these reactions proved to be inefficient under a variety of acidic or basic conditions,²⁵ presumably due to the electron-deficiency of **12**. Stannane **11** was then prepared from **12** via a two step sequence reported by Routier et al.²⁶ Iodination under basic conditions afforded the corresponding C3 iodide, which was converted to **11** through a Pd-catalyzed stannylation [Pd(PPh₃)₄, Me₃SnSnMe₃]. We then synthesized allyl acetate **14**, which later served as a coupling partner of **11**. Farnesyl acetate (**13**) was subjected to a two step process of allylic oxidation (SeO₂, *t*-BuOOH) and Sharpless asymmetric epoxidation [Ti(i-PrO)₄, (+)-diethyl L-tartrate, *t*-BuOOH] to reach epoxy alcohol **10**, according to the known literature.²⁷ Although this compound could be oxidized stepwise to give the corresponding carboxylic acid with good overall efficiency, treatment with catalytic amount of 2-azaadamantane *N*-oxyl (AZADO, Scheme 2) developed by Iwabuchi²⁸ and stoichiometric PhI(OAc)₂ effected the transformation in one

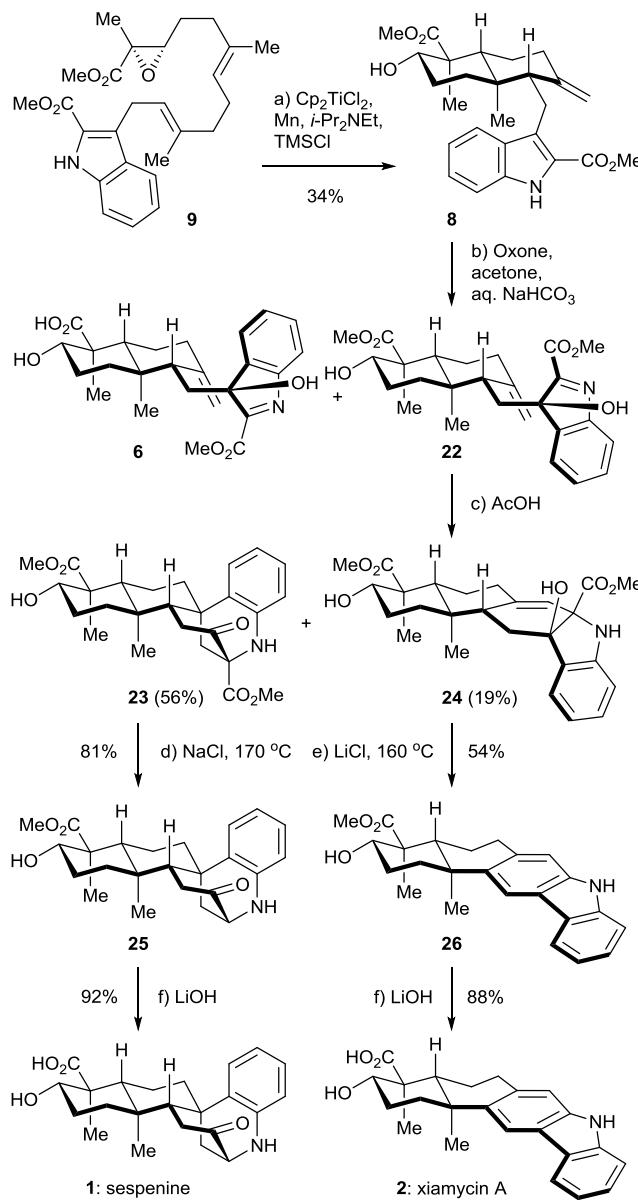
Journal Name



Scheme 3. Postulated pathways of the Ti(III) mediated radical reaction.

pot.²⁹ The resultant carboxylic acid was methylated by using MeI and K₂CO₃ to afford ester **14** in 77% yield for the two steps. A variety of conditions for the allyl Stille–Miyata coupling between **14** and **11** were examined. The conditions [Pd₂(dba)₃, i-Pr₂NEt, LiCl] modified from those described by Hegedus and coworkers proved to be optimal for delivering coupling product **9** (78% yield).²⁴ Notably, the quality of stannane **11** was crucial to this reaction; the freshly prepared stannane was required to ensure a good coupling efficiency.

We then directed our attention to the Ti(III) mediated radical cascade cyclization for synthesizing **8** from **9**. The conditions [Cp₂TiCl₂, Mn, TMSCl, collidine] developed by Cuerva and coworkers were employed for the initial investigation.^{23c} The anticipated product **8** with an exocyclic C=C bond was obtained with an excellent level of stereochemical control, albeit in less than 20% yield. Over-reduction and deoxygenation, which are illustrated as reaction paths a and b in Scheme 3, respectively, turned out to be severe side reactions accompanied with the desired one. From a mechanistic perspective, the activated C16-O bond of α,β-epoxy ester **9** was cleaved by Ti(III) species generated in situ by Cp₂TiCl₂ and Mn, to initiate an electron deficient radical **15**, which then underwent cascade cyclization. However, this radical could also be trapped by excess of Ti(III) species to form a Ti(IV) enolate **16**,³⁰ similar transformations are well-precedented in the context of Sml₂ chemistry.³¹ Further β elimination of OTi(IV) gave α,β-unsaturated ester **17** as a byproduct. High concentration of Ti(III) in the reaction system increased the formation of **17**. On the other hand, tertiary radical **18** generated from the tandem cyclization may enter a

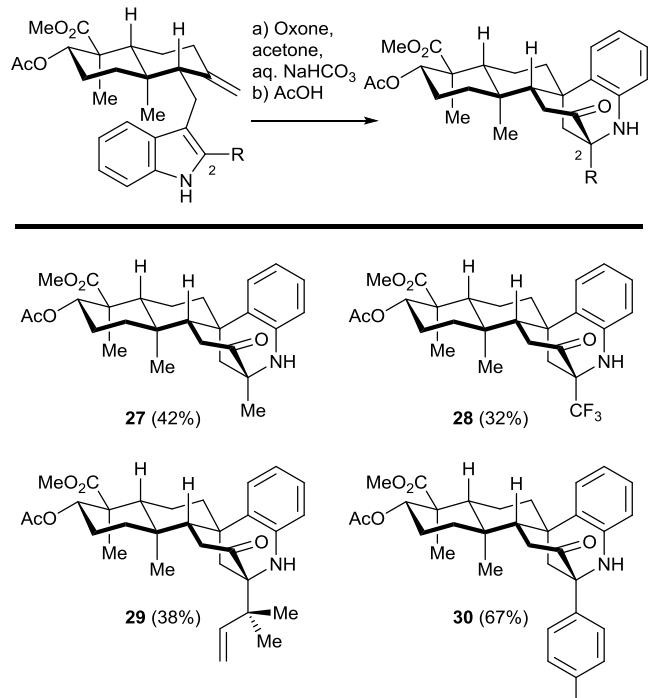
Scheme 4. Total syntheses of sespenine and xiamycin A. Reagents and conditions: (a) Cp₂TiCl₂ (20 mol%), Mn (8.0 equiv), i-Pr₂NEt (6.0 equiv), TMSCl (5.0 equiv), THF, 22 °C, 12 h, 34%; (b) Oxone (2.0 equiv), acetone/saturated aq. NaHCO₃ (1:1), 22 °C, 15 min; (c) AcOH (5.0 equiv), CH₂Cl₂, 22 °C, 5 h, 56% for **23** (2 steps), 19% for **24** (2 steps); (d) NaCl (10 equiv), DMSO/water, 170 °C, 2 h, 81%; (e) LiCl (10 equiv), DMSO/water, 160 °C, 1 h, 51%; (f) aq. LiOH (2.0 M)/THF/MeOH, 50 °C, 10 h, 92% for **1**; 88% for **2**.

Ti(III) mediated mix disproportionation pathway to reach the desired product **19**,^{30a} or unfortunately terminate via a hydrogen atom transfer to form byproduct **21**.³² Inspired by Cuerva's studies of the amine effect in Ti(III) chemistry,³³ we found that i-Pr₂NEt was superior to collidine as an additive for suppressing the over reduction, and strictly anhydrous conditions were critical to inhibit the hydrogen atom transfer. Under our optimized conditions [Cp₂TiCl₂ (20 mol%), Mn, TMSCl, i-Pr₂NEt], compound **8** was isolated in 34% yield (Scheme 4). A suitable initial concentration of Cp₂TiCl₂ (0.08 M) was important as well. Considering the stereochemistry and functionality outcome of this reaction, such efficiency was acceptable and thus allowed us to carry on the synthesis.

1 ARTICLE

2 Journal Name

3 Table 1. Preparation of sespenine analogues through the cationic cascade reaction.



^a Oxone (2.0 equiv), acetone/saturated aq. NaHCO_3 (1:1), 22 °C, 15 min. ^b AcOH (5.0 equiv), CH_2Cl_2 , 22 °C, 1 h.

With compound **8** in hand, we completed the syntheses of sespenine (**1**) and xiamycin A (**2**) through a divergent approach, as depicted in Scheme 4. Treatment of **8** with Oxone in acetone in the presence of aq. NaHCO_3 buffer afforded a C3 epimeric mixture of hydroxyindolenine **6** and **22** in a ca. 2.8:1 ratio. The structure of **22** was indirectly verified by X-ray crystallographic analysis of its bis-acetylated derivative.³⁴ Exposure of this chromatographically inseparable mixture to AcOH for 5 h at ambient temperature provided compounds **23** and **24**, in 56% and 19% yields over the two steps, respectively. The former presumably arose from the devised cascade reaction, while the latter was possibly generated via a straightforward aza-Prins cyclization.^{5b} Krapcho decarboxylation (NaCl, 170 °C) of **23** delivered sespenine methyl ester **25** in 81% yield,³⁵ the structure of which was confirmed by X-ray crystallographic analysis.³⁴ Under these conditions, the other methoxycarbonyl of the molecule remained untouched. Hydrolysis of **25** with aq. LiOH rendered sespenine (**1**) smoothly. Interestingly, under similar Krapcho conditions (LiCl, 160 °C), compound **24** was directly converted into xiamycin A methyl ester **26** in 54% yield. This one pot process may involve sequential ester hydrolysis, β -lactone formation, decarboxylation,³⁶ and dehydrogenative aromatization. The *cis* orientations of the hydroxyl and carboxylic acid functionalities set the geometric basis of β -lactone formation. Basic hydrolysis furnished xiamycin A (**2**) with good efficiency. The spectra and physical properties of the synthetic sespenine and xiamycin A were consistent with those reported for the naturally occurring samples.

We then investigated the cationic cascade with indole substrates bearing various C2 substituents, as shown in Table 1. Sterically bulky (reverse prenyl) and electron deficient (trifluoromethyl) groups did not interfere with the tandem reaction, which indicates the wide range of stabilizing groups for 3-hydroxyindolenine intermediates. The overall yield of the two step sequence was mainly determined by the efficiency as well as facial selectivity of the hydroxylation step, and the following cascade reaction was relatively efficient in general. Thus, a number of sespenine analogues (**27–30**) were readily obtained, setting the stage for further biological studies. Notably, the precursors of these compounds were synthesized through a modified version of our first generation approach (construction of an α,β -unsaturated enone, conjugate addition of indole derivatives, and methylenation; see SI for details), instead of the second generation approach (Stille–Miyata coupling and Ti(III)-mediated cyclization) described in this paper. To assemble one specific indosespene (**3**) type structure, the latter route was more efficient because of the shorter linear sequence. However, the former route was superior when a collection of such compounds needed to be synthesized for preliminary biological studies. A considerably large quantity of the enone was prepared first, and a variety of readily available indole derivatives were then added to it, to generate different analogues rapidly. In contrast, protocols for preparing 2-substituted-3-stannylioles varied according to the electronic effect and functional group compatibility, which hampered the collective synthesis via the second generation strategy.

Conclusions

We developed a ten-step (the longest linear sequence) total synthesis of sespenine from commercially available materials based on our first generation synthesis. Stille–Miyata coupling was exploited to construct a functionalized α,β -epoxy ester, and Ti(III) mediated epoxide opening radical cyclization gave a *trans*-decalin intermediate bearing a 2-methoxycarbonylindole side chain. Indole C3 oxidation afforded a pair of epimeric 3-hydroxyindolenines, the major isomer of which then underwent a cascade of Prins cyclization/Friedel–Crafts/retro Friedel–Crafts to furnish the sespenine core. Xiamycin A was synthesized from the minor hydroxyindolenine epimer. Four analogues of sespenine with various C2 substituents were prepared. This work may facilitate the biosynthetic and biological studies of sespenine and related natural products.

Acknowledgements

This paper is dedicated to Prof. Tohru Fukuyama for his contribution to indole chemistry. We thank Prof. Changsheng Zhang and Dr. Qingbo Zhang for helpful discussions. Financial support was provided by Ministry of Science & Technology (2013CB836900), National Natural Science Foundation of China (21525209, 21290180, 21172235, and 21222202),

Journal Name

Shanghai Science and Technology Commission (15JC1400400), and Chinese Academy of Sciences.

Notes and references

- 1 Selective examples of chemical synthesis of indole monoterpenoids: (a) H. Muratake and M. Natsume, *Tetrahedron* 1990, **46**, 6331; (b) H. Muratake, H. Kumagami and M. Natsume, *Tetrahedron* 1990, **46**, 6351; (c) V. Vaillancourt and K. F. Albizati, *J. Am. Chem. Soc.* 1993, **115**, 3499; (d) T. Fukuyama and X. Chen, *J. Am. Chem. Soc.* 1994, **116**, 3125; (e) M. Sakagami, H. Muratake and M. Natsume, *Chem. Pharm. Bull.* 1994, **42**, 1393; (f) A. C. Kinsman and M. A. Kerr, *Org. Lett.* 2001, **3**, 3189; (g) A. C. Kinsman and M. A. Kerr, *J. Am. Chem. Soc.* 2003, **125**, 14120; (h) P. S. Baran and J. M. Richter, *J. Am. Chem. Soc.* 2004, **126**, 7450; (i) P. S. Baran and J. M. Richter, *J. Am. Chem. Soc.* 2005, **127**, 15394; (j) P. S. Baran, T. J. Maimone and J. M. Richter, *Nature* 2007, **446**, 404; (k) J. M. Richter, Y. Ishihara, T. Masuda, B. W. Whitefield, T. Llamas, A. Pohjakallio and P. S. Baran, *J. Am. Chem. Soc.* 2008, **130**, 17938; (l) T. J. Maimone, Y. Ishihara and P. S. Baran, *Tetrahedron* 2015, **71**, 3652; (m) S. E. Reisman, J. M. Ready, A. Hasuoka, C. J. Smith and J. L. Wood, *J. Am. Chem. Soc.* 2006, **128**, 1448; (n) A. Chandra and J. N. Johnston, *Angew. Chem. Int. Ed.* 2011, **50**, 7641; (o) R. J. Rafferty and R. M. Williams, *J. Org. Chem.* 2012, **77**, 519; (p) V. Bhat, K. M. Allan, V. H. Rawal, *J. Am. Chem. Soc.* 2011, **133**, 5798; (q) K. M. Allan, K. Kobayashi and V. H. Rawal, *J. Am. Chem. Soc.* 2012, **134**, 1392; (r) A. D. Huters, K. W. Quasdorf, E. D. Styduhar and N. K. Garg, *J. Am. Chem. Soc.* 2011, **133**, 15797; (s) K. W. Quasdorf, A. D. Huters, M. W. Lodewyk, D. J. Tantillo and N. K. Garg, *J. Am. Chem. Soc.* 2012, **134**, 1396; (t) A. D. Huters, E. D. Styduhar and N. K. Garg, *Angew. Chem. Int. Ed.* 2012, **51**, 3758; (u) E. D. Styduhar, A. D. Huters, N. A. Weires and N. K. Garg, *Angew. Chem. Int. Ed.* 2013, **52**, 12422; (v) N. A. Weires, E. D. Styduhar, E. L. Baker and N. K. Garg, *J. Am. Chem. Soc.* 2014, **136**, 14710.
- 2 Selective examples of chemical synthesis of indole sesquiterpenoids: (a) E. J. Velthuisen and S. J. Danishefsky, *J. Am. Chem. Soc.* 2007, **129**, 10640; (b) I. S. Marcos, R. F. Moro, I. P. Costales, P. Basabe, D. Díez, F. Mollinedo and J. G. Urones, *Tetrahedron* 2012, **68**, 7932; (c) I. S. Marcos, R. F. Moro, I. Costales, P. Basabe and D. Díez, *Nat. Prod. Rep.* 2013, **30**, 1509; (d) I. S. Marcos, R. F. Moro, I. Costales, P. Basabe, D. Díez, F. Mollinedo and J. G. Urones, *Tetrahedron* 2013, **69**, 7285; (e) A. Asanuma, M. Enomoto, T. Nagasawa and S. Kuwahara, *Tetrahedron Lett.* 2013, **54**, 4561.
- 3 Selective examples of chemical synthesis of indole diterpenoids: (a) A. B. Smith, III and R. Mewshaw, *J. Am. Chem. Soc.* 1985, **107**, 1769; (b) A. B. Smith, III, T. Sunazuka, T. L. Leenay and J. Kingery-Wood, *J. Am. Chem. Soc.* 1990, **112**, 8197; (c) A. B. Smith, III, N. Kanoh, H. Ishiyama and R. A. Hartz, *J. Am. Chem. Soc.* 2000, **122**, 11254; (d) Y. Zou, J. E. Melvin, S. S. Gonzales, M. J. Spafford and A. B. Smith, III, *J. Am. Chem. Soc.* 2015, **137**, 7095; (e) B. Bradshaw, G. Etxebarria-Jardí and J. Bonjoch, *J. Am. Chem. Soc.* 2010, **132**, 5966; (f) A. E. Goetz, A. L. Silberstein, M. A. Corsello and N. K. Garg, *J. Am. Chem. Soc.* 2014, **136**, 3036; (g) M. Enomoto, A. Morita and S. Kuwahara, *Angew. Chem. Int. Ed.* 2012, **51**, 12833; (h) R. J. Sharpe and J. S. Johnson, *J. Am. Chem. Soc.* 2015, **137**, 4968; (i) R. J. Sharpe and J. S. Johnson, *J. Org. Chem.* 2015, **80**, 9740.
- 4 Selected studies of biosynthesis of indole terpenoids: (a) "Terpene Indole Alkaloid Biosynthesis": S. E. O'Connor and E. McCoy in *Recent Advances in Phytochemistry*, Vol. **40** (Eds.: J. T. Romeo), Elsevier, Amsterdam, 2006, pp. 1–22; (b) M. Baunach, J. Franke and C. Hertweck, *Angew. Chem. Int. Ed.* 2015, **54**, 2604; (c) K. Tagami, C. Liu, A. Minami, M. Noike, T. Isaka, S. Fueki, Y. Shichijo, H. Toshima, K. Gomi, T. Dairi and H. Oikawa, *J. Am. Chem. Soc.* 2013, **135**, 1260; (d) M. Tang, H. Lin, D. Li, Y. Zou, J. Li, W. Xu, R. Cacho, M. E. Hillenmeyer, N. K. Garg and Y. Tang, *J. Am. Chem. Soc.* 2015, **137**, 13724.
- 5 Total syntheses of indole terpenoids from our group: (a) M. Bian, Z. Wang, X. Xiong, Y. Sun, C. Matera, K. C. Nicolaou and A. Li, *J. Am. Chem. Soc.* 2012, **134**, 8078; (b) Y. Sun, R. Li, W. Zhang and A. Li, *Angew. Chem. Int. Ed.* 2013, **52**, 9201; (c) Z. Lu, M. Yang, P. Chen, X. Xiong and A. Li, *Angew. Chem. Int. Ed.* 2014, **53**, 13840; (d) S. Zhou, H. Chen, Y. Luo, W. Zhang and A. Li, *Angew. Chem. Int. Ed.* 2015, **54**, 6878; (e) Z. Lu, H. Li, M. Bian and A. Li, *J. Am. Chem. Soc.* 2015, **137**, 13764.
- 6 (a) S. Zhou, D. Zhang, Y. Sun, R. Li, W. Zhang and A. Li, *Adv. Synth. Catal.* 2014, **356**, 2867; (b) X. Xiong, D. Zhang, J. Li, Y. Sun, S. Zhou, M. Yang, H. Shao and A. Li, *Chem. Asian J.* 2015, **10**, 869; (c) H. Yu, C. Wan, J. Han and A. Li, *Acta Chim. Sinica* 2013, **71**, 1488.
- 7 (a) L. Ding, A. Maier, H.-H. Fiebig, W.-H. Lin and C. Hertweck, *Org. Biomol. Chem.* 2011, **9**, 4029; discovery of relevant natural products: (b) L. Ding, J. Münch, H. Goerls, A. Maier, H.-H. Fiebig, W.-H. Lin and C. Hertweck, *Bioorg. Med. Chem. Lett.* 2010, **20**, 6685.
- 8 Q. Zhang, A. Mándi, S. Li, Y. Chen, W. Zhang, X. Tian, H. Zhang, H. Li, W. Zhang, S. Zhang, J. Ju, T. Kurtán and C. Zhang, *Eur. J. Org. Chem.* 2012, 5256.
- 9 (a) Z. Xu, M. Baunach, L. Ding and C. Hertweck, *Angew. Chem. Int. Ed.* 2012, **51**, 10293; (b) M. Baunach, L. Ding, T. Bruhn, G. Bringmann and C. Hertweck, *Angew. Chem. Int. Ed.* 2013, **52**, 9040; (c) M. Baunach, L. Ding, K. Willing and C. Hertweck, *Angew. Chem. Int. Ed.* 2015, **54**, 13279; (d) H. Li, Q. Zhang, S. Li, Y. Zhu, G. Zhang, H. Zhang, X. Tian, S. Zhang, J. Ju and C. Zhang, *J. Am. Chem. Soc.* 2012, **134**, 8996; (e) Q. Zhang, H. Li, S. Li, Y. Zhu, G. Zhang, H. Zhang, W. Zhang, R. Shi and C. Zhang, *Org. Lett.* 2012, **14**, 6142; (f) H. Li, Y. Sun, Q. Zhang, Y. Zhu, S.-M. Li, A. Li and C. Zhang, *Org. Lett.* 2015, **17**, 306; (g) A. H. Trotta, *Org. Lett.* 2015, **17**, 3358.
- 10 Y. Sun, P. Chen, D. Zhang, M. Baunach, C. Hertweck and A. Li, *Angew. Chem. Int. Ed.* 2014, **53**, 9012.
- 11 B. R. Rosen, E. W. Werner, A. G. O'Brien and P. S. Baran, *J. Am. Chem. Soc.* 2014, **136**, 5571.
- 12 Z. Meng, H. Yu, L. Li, W. Tao, H. Chen, M. Wan, P. Yang, D. J. Edmonds, J. Zhong and A. Li, *Nat. Commun.* 2015, **6**, 6096.
- 13 Selected examples of 6π -electrocyclization/aromatization in natural product synthesis: (a) E. A. Anderson, E. J. Alexanian and E. J. Sorensen, *Angew. Chem. Int. Ed.* 2004, **43**, 1998; (b) K. Ohmori, K. Mori, Y. Ishikawa, H. Tsuruta, S. Kuwahara, N. Harada and K. Suzuki, *Angew. Chem. Int. Ed.* 2004, **43**, 3167; (c) T. Suzuki, T. Hamura and K. Suzuki, *Angew. Chem. Int. Ed.* 2008, **47**, 2248; (d) D. L. Sloman, B. Mitasev, S. S. Scully, J. A. Beutler and J. A. Porco, *Angew. Chem. Int. Ed.* 2011, **50**, 2511; (e) D. L. Sloman, J. W. Bacon and J. A. Porco, *J. Am. Chem. Soc.* 2011, **133**, 9952; (f) D. K. Winter, M. A. Endoma-Arias, T. Hudlicky, J. A. Beutler and J. A. Porco, *J. Org. Chem.* 2013, **78**, 7617; (g) A. Fürstner, M. M. Domostoj and B. Scheiper, *J. Am. Chem. Soc.* 2005, **127**, 11620; (h) T. J. Greshock and R. L. Funk, *J. Am. Chem. Soc.* 2006, **128**, 4946; (i) T. J. Greshock and R. L. Funk, *Org. Lett.* 2006, **8**, 2643; (j) R. J. Huntley and R. L. Funk, *Org. Lett.* 2006, **8**, 3403; (k) R. J. Huntley and R. L. Funk, *Org. Lett.* 2006, **8**, 4775; (l) S. T. Staben, J. J. Kennedy-Smith, D. Huang, B. K. Corkey, R. L. LaLonde and F. D. Toste, *Angew. Chem. Int. Ed.* 2006, **45**, 5991.
- 14 Other examples of 6π -electrocyclization/aromatization in natural product synthesis from our group: (a) Z. Lu, Y. Li, J. Deng and A. Li, *Nat. Chem.* 2013, **5**, 679; (b) J. Li, P. Yang, M. Yao, J. Deng and A. Li, *J. Am. Chem. Soc.* 2014, **136**, 16477; (c) M. Yang, J. Li and A. Li, *Nat. Commun.* 2015, **6**, 6445; (d) M. Wan, M. Yao, J.-Y. Gong, P. Yang, H. Liu and A. Li, *Chin. Chem. Lett.* 2015, **26**, 1120.

ARTICLE

Journal Name

- Lett. 2015, **26**, 272; (e) M. Yang, X. Yang, H. Sun and A. Li, *Angew. Chem. Int. Ed.* DOI: 10.1002/anie.201510568.
- 15 An excellent review of biosynthetic and biomimetic electrocyclizations: C. M. Beaudry, J. P. Malerich and D. Trauner, *Chem. Rev.* 2005, **105**, 4757.
- 16 (a) P. S. Baran and E. J. Corey, *J. Am. Chem. Soc.* 2002, **124**, 7904; (b) E. M. Ferreira and B. M. Stoltz, *J. Am. Chem. Soc.* 2003, **125**, 9578; (c) N. P. Grimster, C. Gauntlett, C. R. A. Godfrey and M. J. Gaunt, *Angew. Chem. Int. Ed.* 2005, **44**, 3125; (d) A. Kong, X. Han and X. Lu, *Org. Lett.* 2006, **8**, 1339; (e) X. Han and X. Lu, *Org. Lett.* 2009, **11**, 2381.
- 17 (a) C. Olier, M. Kaafarani, S. Gastaldi and M. P. Bertrand, *Tetrahedron* 2010, **66**, 413; (b) X. Han, G. R. Peh, P. E. Floreancig, *Eur. J. Org. Chem.* 2013, 1193.
- 18 Application of Prins cyclization in natural product synthesis from our group: J. Deng, S. Zhou, W. Zhang, J. Li, R. Li and A. Li, *J. Am. Chem. Soc.* 2014, **136**, 8185. Also see refs. 5b, 5c and 5e.
- 19 (a) Y. Liu, W. W. McWhorter, Jr. and C. E. Hadden, *Org. Lett.* 2003, **5**, 333; (b) G. A. Ho, D. H. Nouri and D. J. Tantillo, *Tetrahedron Lett.* 2009, **50**, 1578.
- 20 2-substituted 3-hydroxyindolenines as natural product motifs: (a) M. Bittner, M. Silva, E. M. Gopalakrishna, W. H. Watson, V. Zabel, S. A. Matlin and P. G. Sammes, *J. Chem. Soc. Chem. Commun.* 1978, 79; (b) I. R. C. Bick, M. A. Hai and N. W. Preston, *Heterocycles* 1983, **20**, 667; (c) D. Ponglux, S. Wongseripipatana, H. Takayama, M. Kikuchi, M. Kurihara, M. Kitajima, N. Aimi and S. Sakai, *Planta Med.* 1994, **60**, 580.
- 21 Selected examples of syntheses of 2-substituted 3-hydroxyindolenines: (a) B. Witkop and J. B. Patrick, *J. Am. Chem. Soc.* 1951, **73**, 2188; (b) T. Hino, H. Yamaguchi, K. Matsuki, K. Nakano, M. Sodeoka and M. Nakagawa, *J. Chem. Soc. Perkin Trans. 1* 1983, 141; (c) R. Güller and H.-J. Borschberg, *Helv. Chim. Acta* 1993, **76**, 1847; (d) J. Éles, G. Kalaus, A. Léval, I. Greiner, M. Kajtár-Peregy, P. Szabó, L. Szabó and C. Szántay, *J. Heterocycl. Chem.* 2002, **39**, 767; (e) M. Movassaghi, M. A. Schmidt and J. A. Ashenhurst, *Org. Lett.* 2008, **10**, 4009; (f) T. J. Greshock and R. M. Williams, *Org. Lett.* 2007, **9**, 4255; (g) T. J. Greshock and R. M. Williams, *Org. Lett.* 2012, **14**, 6377; (h) S. Liu, J. S. Scotti and S. A. Kozmin, *J. Org. Chem.* 2013, **78**, 8645; (i) S. Han, K. C. Morrison, P. J. Hergenrother and M. Movassaghi, *J. Org. Chem.* 2014, **79**, 473; (j) X. Qi, H. Bao and U. K. Tambar, *J. Am. Chem. Soc.* 2011, **133**, 10050; (k) S. Han and M. Movassaghi, *J. Am. Chem. Soc.* 2011, **133**, 10768; (l) S. Liu and X.-J. Hao, *Tetrahedron Lett.* 2011, **52**, 5640; (m) F. Kolundzic, M. N. Noshi, M. Tjandra M. Movassaghi and S. J. Miller, *J. Am. Chem. Soc.* 2011, **133**, 9104.
- 22 (a) W. A. Nugent and T. V. RajanBabu, *J. Am. Chem. Soc.* 1988, **110**, 8561; (b) T. V. RajanBabu and W. A. Nugent, *J. Am. Chem. Soc.* 1994, **116**, 986; (c) A. Gansäuer, M. Pierobon and H. Bluhm, *Angew. Chem. Int. Ed.* 1998, **37**, 101; (d) A. Gansäuer, H. Bluhm, M. Pierobon, *J. Am. Chem. Soc.* 1998, **120**, 12849; (e) A. Gansäuer and S. Narayan, *Adv. Synth. Catal.* 2002, **344**, 465; (f) A. Gansäuer, J. Justicia, C.-A. Fan, D. Worgull and F. Piestert, *Top. Curr. Chem.* 2007, **279**, 25.
- 23 (a) A. Fernández-Mateos, E. Martín de la Nava, G. Pascual Coca, A. Ramos Silva and R. Rubio González, *Org. Lett.* 1999, **1**, 607; (b) A. F. Barrero, J. M. Cuerva, M. M. Herrador and M. V. Valdivia, *J. Org. Chem.* 2001, **66**, 4074; (c) J. Justicia, A. Rosales, E. Buñuel, J. L. Oller-López, M. Valdivia, A. Haïdour, J. E. Oltra, A. F. Barrero, D. J. Cárdenas and J. M. Cuerva, *Chem. Eur. J.* 2004, **10**, 1778; (d) J. Justicia, J. E. Oltra and J. M. Cuerva, *J. Org. Chem.* 2004, **69**, 5803; (e) J. Justicia, J. L. Oller-López, A. G. Campaña, J. E. Oltra, J. M. Cuerva, E. Buñuel and D. J. Cárdenas, *J. Am. Chem. Soc.* 2005, **127**, 14911; (f) J. Justicia, J. E. Oltra and J. M. Cuerva, *J. Org. Chem.* 2005, **70**, 8265; (g) A. F. Barrero, J. F. Quílez del Moral, E. M. Sánchez and J. F. Arteaga, *Eur. J. Org. Chem.* 2006, **7**, 1627; (h) J. M. Cuerva, J. Justicia, J. L. Oller-López and J. E. Oltra, *Top. Curr. Chem.* 2006, **264**, 63; (i) J. Justicia, L. Álvarez de Cienfuegos, A. G. Campaña, D. Miguel, V. Jakoby, A. Gansäuer and J. M. Cuerva, *Chem. Soc. Rev.* 2011, **40**, 3525; (j) S. P. Morcillo, D. Miguel, A. G. Campaña, L. Álvarez de Cienfuegos, J. Justicia and J. M. Cuerva, *Org. Chem. Front.* 2014, **1**, 15; (k) S. P. Morcillo, D. Miguel, S. Resa, A. Martín-Lasanta, A. Millán, D. Choquesillo-Lazarte, J. M. García-Ruiz, A. J. Mota, J. Justicia, and J. M. Cuerva, *J. Am. Chem. Soc.* 2014, **136**, 6943; (l) J. Y. Cha, J. T. S. Yeoman and S. E. Reisman, *J. Am. Chem. Soc.* 2011, **133**, 14964; (m) J. T. S. Yeoman, V. W. Mak and S. E. Reisman, *J. Am. Chem. Soc.* 2013, **135**, 11764. (n) P. K. Mandal, G. Maiti and S. C. Roy, *J. Org. Chem.* 1998, **63**, 2829; (o) S. C. Roy, K. K. Rana and C. Guin, *J. Org. Chem.* 2002, **67**, 3242.
- 24 L. Del Valle, J. K. Stille and L. S. Hegedus, *J. Org. Chem.* 1990, **55**, 3019.
- 25 (a) Z. Liu, L. Liu, Z. Shafiq, D. Wang and Y.-J. Chen, *Lett. Org. Chem.* 2007, **4**, 256; (b) X. Zhu and A. Ganesan, *J. Org. Chem.* 2002, **67**, 2705; (c) M. Westermaier and H. Mayr, *Org. Lett.* 2006, **8**, 4791.
- 26 A. Bourderioux, A. Ouach, V. Bénéteau, J. Mérou and S. Routier, *Synthesis* 2010, 783.
- 27 J. Lan, Z. Liu, H. Yuan, L. Peng, W. Li, Y. Li, Y. Li and A. C. Chan, *Tetrahedron Lett.* 2000, **41**, 2181
- 28 (a) M. Shibuya, M. Tomizawa, I. Suzuki and Y. Iwabuchi, *J. Am. Chem. Soc.* 2006, **128**, 8412; (b) Y. Iwabuchi, *Chem. Pharm. Bull.* 2013, **61**, 1197.
- 29 T. Kuranaga, Y. Sesoko, K. Sakata, N. Maeda, A. Hayata and M. Inoue, *J. Am. Chem. Soc.* 2013, **135**, 5467.
- 30 (a) J. Justicia, T. Jiménez, S. P. Morcillo, J. M. Cuerva and J. E. Oltra, *Tetrahedron* 2009, **65**, 10837; (b) C. Hardouin, F. Chevallier, B. Rousseau, E. Doris, *J. Org. Chem.* 2001, **66**, 1046.
- 31 K. C. Nicolaou, S. P. Ellery and J. S. Chen, *Angew. Chem. Int. Ed.* 2009, **48**, 7140.
- 32 (a) J. M. Cuerva, A. G. Campaña, J. Justicia, A. Rosales, J. L. Oller-López, R. Robles, D. J. Cárdenas, E. Buñuel, J. E. Oltra, *Angew. Chem. Int. Ed.* 2006, **45**, 5522; (b) T. V. Chciuk and R. A. Flowers, II, *J. Am. Chem. Soc.* 2015, **137**, 11526.
- 33 M. Paradas, A. G. Campaña, T. Jiménez, R. Robles, J. E. Oltra, E. Buñuel, J. Justicia, D. J. Cárdenas and J. M. Cuerva, *J. Am. Chem. Soc.* 2010, **132**, 12748.
- 34 CCDC 987752 (bis-acetylated **22**) and 949307 (**25**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- 35 A. P. Krapcho, *Arkivoc* 2007, (ii), 54.
- 36 A. Pommier, J.-M. Pons, *Synthesis* 1993, 441.