Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/obc

Journal Name

ARTICLE

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/



Haniya Bounar,^a Zhenhua Liu,^a Lin Zhang,^a Xiaoxue Guan,^a Zonglian Yang,^a Peiqiu Liao,^a Xihe Bi,^{*,a} Xingqi Li^{*,a}

An unprecedented silver-catalyzed cascade reaction of tosylmethyl isocyanide (TosMIC) with propargylic alcohols for the efficient synthesis of (*E*)-vinyl sulfones has been developed where TosMIC plays a dual role as both the reactant in the allenylation of propargylic alcohols and the sulfonyl source.

Introduction

Since the introduction of tosylmethyl isocyanide (TosMIC) by van Leusen,¹ this reagent has been of immense use in organic chemistry.² Its broad synthetic application emanates from its isocyano group, acidic α -carbon atom and sulfonyl group acting as both a sulfonyl leaving group and the aid to the enhanced acidity of the α -carbon. Extensive investigations of reactions of TosMIC with various substrates containing unsaturated bonds such as C=O, C=N, and C=C, etc. that offer a large variety of heterocyclic chemical structures (for example, oxazoles,³ imidazoles,⁴ pyrroles⁵ and indoles⁶), have been reported. However, in contrast to the reaction of TosMIC with various substrates containing the aforementioned unsaturated bonds, reports relating to reactions of TosMIC with substrates containing C=C are relatively underdeveloped. To the best our knowledge, there are six known reports of reactions of TosMIC with substrates containing C=C; the reaction between acetylenic esters and TosMIC to afford the corresponding pyrroles (Scheme 1, a)⁷ and the tandem reactions of TosMIC with 1-(2-Iodoary)-2-yn-1-ones to synthesize 4-oxoindeno[1,2-b]pyrroles(Scheme 1, b).⁸ Recently, we realized that TosMIC can react with propargylic alcohols in a peculiar manner where TosMIC acted as allenylative reagent or dual roles, that is, as sulfonyl source and ligand (Scheme 1, c and d).9 As part of our ongoing interest in TosMIC chemistry, especially on the area of its reaction with propargylic alcohols, we herein report an unprecedented silver-catalyzed cascade reaction of TosMIC with propargylic alcohols to access (E)vinyl sulfones (Scheme 1, e). Mechanistic investigations have shown that TosMIC plays a duel role as the reactant in the allenylation of propargylic alcohols and the sulfonyl source. So far, this is the first example where TosMIC has acted as allenylative reagent and sulfonyl source in a cascade reaction.





Results and discussion

Initially, the reaction of 1-phenylprop-2-yn-1-ol **1a** with TosMIC **2b** was repeated according to our reported procedure.9a When we increased the TosMIC from 1.5 equiv to 2.5 equiv on a 10 mol% Ag_2CO_3 as catalyst in 1,4-dioxane at 80 °C for 1 h, the expected product, 4-phenyl-N-(tosylmethyl)buta-2,3-dienamide, was not obtained. After workup and purification, a differently new product characterized as vinyl sulfone **3a**, (*E*)-4-phenyl-3-tosyl-N-(tosylmethyl)but-3-enamide from its spectral and analytical data, was obtained in 61% yield (Scheme 2). The *E*-configuration of product **3a** was verified by its 2D-NOESY spectrum. This reaction represents a new route to vinyl sulfones.

YAL SOCIETY CHEMISTRY

^{a.} Department of Chemistry, Northeast Normal University, 5268 Renmin Street, 130024 Changchun, China; E-mail: <u>lixa653@nenu.edu.cn</u>.

[†] Electronic supplementary information (ESI) available: Experimental procedure, ¹H and ¹³C NMR spectra for all compounds. CCDC 946538. For ESI and crystallographic data in CIF or other electronic format. See DOI: 10.1039/x0xx00000x

ARTICLE



The importance of vinyl sulfones in the pharmaceutical industry¹⁰ as well as in the diverse fields of organic synthesis¹¹ encourages us to carry out an extensive optimization of the reaction conditions. As shown in table 1, a series of solvents including 1,4-dioxane, CH2Cl2, DMF and toluene were examined in the presence of 10 mol% Ag₂CO₃ at 80 °C for 1 h. Although the reaction proceeded smoothly in these solvents, 1,4-dioxane provided a peerless result (Table 1, entries 1-4). Various silver salts such as AgF, AgOAc and AgNO₃ were also evaluated (Table 1, entries 5-7). Among these salts, AgOAc was most effective and AgNO₃ was ineffective. To our delight, when the amount of AgOAc was increased from 10 mol% to 20 mol%, a significant increase in the yield (Table 1, entry 8) was realized. In a control experiment, the necessity of a silver catalyst was demonstrated (Table 1, entry 9). Conditions listed in entry 8, being optimal were therefore selected for further investigations

Table 1. Condition screening				
F	OH + C≣N ph + C≣N 1a 2b (2.5 ∈	[Ag], Solvent Ts 80 °C, 1 h	H Ph 3a	
Entry	[Ag]	Amount	Solvent	Yield
-		(mol%)		3a (%) ^a
1	Ag ₂ CO ₃	10	CH_2Cl_2	60
2	Ag_2CO_3	10	DMF	58
3	Ag_2CO_3	10	Toluene	59
4	Ag_2CO_3	10	1,4-Dioxane	61
5	AgNO ₃	10	1,4-Dioxane	0
6	ÄgF	10	1,4-Dioxane	55
7	AgOAc	10	1,4-Dioxane	65
8	AgOAc	20	1,4-Dioxane	88
9			1,4-Dioxane	0
^a Isolated yields of the product 3a				

With the optimized conditions in hand (Table 1, entry 8), the substrate scope of propargylic alcohols that could be used in this cascade reaction was investigated (Table 2). A range of substrates bearing various R groups including (hetero)aryl, fused aryl (1b-1p) were converted into the corresponding vinyl sulfones (3b-3p) in synthetically useful yields under the optimized conditions. Moreover, wide variation of the substituents on the benzene ring including electron-donating (e.g., RO and Me) and -withdrawing groups (e.g., Br and Cl) allowed the formation of diversely high functionalized vinyl sulfones (3b-3k) in moderate to excellent yields. It is worth mentioning that several interesting functional groups on the phenyl ring, such as allyloxy or phenylethynyl group, were well tolerated under the optimized reaction condition (30 and 3p), which might provide an opportunity for further

functionalization. However, when the R group was alkyl, the cascade reaction was unsuccessful (3q).

Table 2. Substrate Scope



Subsequently, we turned our attention to the reaction mechanism. To better understand the reaction routes, the following experimental investigations were carried out. As shown in scheme 3. Firstly, when the reaction of 1a with 1.5 equiv of 2b under the optimized conditions for 0.5 h was terminated, the intermediate 4, 4-phenyl-N-(tosylmethyl)buta-2,3-dienamide, was obtained in 94% yield [Scheme 3. Eq. (1)]. Next, the intermediate 4 further reacted with 1.5 equiv of 2b under the optimized reaction for 0.5 h to afford 3a in 96% yield [Scheme 3. Eq. (2)], in which TosMIC might decompose to formaldehyde and a Ts group, probably accompanied by the generation of a cyanide ion (CN⁻).9b,12 Based on the above results and related precedents,^{9,12} a plausible reaction pathway was proposed [Scheme 3. Eq. (3)]. The cascade reaction started with the reaction of propargylic alcohol 1a with TosMIC 2b to yield the intermediate 4 by allenylation of the former in which TosMIC acted as the allenvlative reagent. Next, the stereodefined sulfonylation of 4 took place with the TosMIC as sulfonyl source, thus affording vinyl sulfone 3a.

Journal Name

Journal Name



To investigate the synthetic potential of this method, the reaction of propargylic alcohol **1a** with TosMIC **2b** was carried out on a gram scale. To our delight, the corresponding vinyl sulfone **3a** was obtained in 74% yield. Furthermore, good conversion of **3a** into (*E*)-4-phenyl-3-tosylbut-3-enamide **5** was achieved under palladium-catalyzed condition [Eq. (1)].

$$\begin{array}{c} \textbf{1a} \quad \underbrace{ \begin{array}{c} \textbf{2b} (2.5 \, \text{eq.}) \\ \text{standard conditions} \end{array}}_{\text{g}} \quad \underbrace{ \begin{array}{c} \text{Pd}(\text{OAc})_2 (10 \ \text{mol}\%) \\ \hline \text{DMA:PivOH} (4:1) \\ 0_2 \ 130 \ ^\circ\text{C}, 45 \ \text{h} \end{array}}_{\text{Fh}} \quad \underbrace{ \begin{array}{c} \textbf{Ts} \quad \text{O} \\ \textbf{Ph} \\ \textbf{Ph} \\ \textbf{5}, 80\% \end{array}}_{\text{Fh}} \\ \textbf{NH}_2 \quad (1) \end{array}$$

To expand the scope of this methodology further, propargylic alcohols **6**, 1,3-diphenylpenta-1,4-diyn-3-ol, was prepared by a two-step synthesis. Ethynylbenzene was first subjected to reaction with benzoyl chloride to synthesize the 1,3-diphenylprop-2-yn-1-one under the palladium-catalyzed condition,¹³ followed by reaction with ethynylmagnesium bromide to afford the target product **6**. Under the optimized conditions, the product **7**, (*Z*)-4,6-diphenyl-3-tosyl-*N*-(tosylmethyl)hex-3-en-5-ynamide, was obtained in 37% yield (Scheme 4). The structure of product **7** was unambiguously established by XRD analysis (CCDC 946538).



Conclusion

In conclusion, we have conceived an unprecedented cascade reaction of tosylmethyl isocyanide (TosMIC) with propargylic alcohols affording the corresponding (E)-vinyl sulfones in synthetically useful yields. TosMIC plays a dual role, that is, as the reactant in the allenylation of propargylic alcohols and the sulfonyl source. To the best our knowledge, TosMIC as allenylative reagent and sulfonyl source in a cascade reaction has not been reported to date. This result opens up a new aspect of synthetic utility of TosMIC and expands TosMIC chemistry.

Experimental section

General information

All reagents were purchased from commercial sources and used without treatment, unless otherwise indicated. The products were purified by column chromatography over silica gel. ¹H NMR and ¹³C NMR spectra were recorded at 25 °C on a Varian 500 MHz and 125 MHz, respectively, and TMS was used as internal standard. Mass spectra were recorded on BRUKER AutoflexIII Smartbeam MS-spectrometer. High resolution mass spectra (HRMS) were recorded on Bruck microTof by using ESI method.

General procedure for synthesis

General procedure for the synthesis of compounds 3a-p (with 3a as an example). To a mixture of 1-phenylprop-2-yn-1-ol (1a) (66 mg, 0.5 mmol) and AgOAc (16.7 mg, 0.1 mmol) in 1,4-dioxane (0.5 mL), 1-isocyanomethanesulfonyl-4-methylbenzene (2b) (244 mg, 1.25 mmol) which was dissolved in 2.0 mL 1,4-dioxane and added in 10 minutes at 80 °C. The reaction mixture was then stirred for 0.5 h-1 h until substrate 1a had been consumed as indicated by TLC. The resulting mixture was concentrated and taken up by dichloromethane. The organic layer was washed with brine, dried over MgSO₄ and concentrated. Purification of the crude product with flash column chromatography (silica gel; petroleum ether: ethyl acetate = 3: 1) gave 3a in 88% yield as a white solid.

Synthesis of compound 5. To a mixture of (*E*)-4-phenyl-3tosyl-*N*-(tosylmethyl)but-3-enamide (3a) (242 mg, 0.5 mmol) in 2.5 mL of N,N-dimethylformamide: pivalic acid (4:1) at 130 oC with oxygen balloon, Pd(OAc)₂ (11.3 mg, 0.05 mmol) was added. The reaction mixture was then stirred for 45 h until substrate 3a had been consumed as indicated by TLC. The resulting mixture was concentrated and taken up by dichloromethane. The organic layer was washed with brine, dried over MgSO₄ and concentrated. Purification of the crude product with flash column chromatography (silica gel; petroleum ether: ethyl acetate = 1: 1) gave 5 in 80% yield as a white solid.

Synthesis of compound 7. To a mixture of 1,3-diphenylpenta-1,4-diyn-3-ol (6) (116 mg, 0.5 mmol) and AgOAc (16.7 mg, 0.1 mmol) in 1,4-dioxane (0.5 mL), 1-isocyanomethanesulfonyl-4methyl -benzene (2b) (244 mg, 1.25 mmol) which was dissolved in 2.0 mL 1,4-dioxane and added in 10 minutes at 80 °C. The reaction mixture was then stirred for 0.5 h-1 h until substrate 1a had been consumed as indicated by TLC. The resulting mixture was concentrated and taken up by dichloromethane. The organic layer was washed with brine, dried over MgSO₄ and concentrated. Purification of the crude product with flash column chromatography (silica gel; petroleum ether: ethyl acetate = 3: 1) gave 7 in 37% yield as a white solid.

 $\label{eq:constraint} \begin{array}{ll} \textbf{(B)-4-phenyl-3-tosyl-N-(tosylmethyl)but-3-enamide} \\ \mbox{White solid, m.p. 157-158 °C; 1H NMR (500 MHz, CDCl_3) δ} \end{array}$

found: 518.0954.

found: 586.0145.

ARTICLE

2.21 (s, 3H), 2.43 (s, 3H), 3.24 (s, 2H), 4.70 (d, J = 6.5 Hz, 2H), 7.15 (d, J = 8.0 Hz, 2H), 7.34 (d, J = 8.0 Hz, 2H), 7.37-7.45 (m, 5H); 7.74 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 8.0 Hz, 2H), 7.94 (s, 1H), 8.02 (t, J = 6.5 Hz, 1H); ¹³C **NMR** (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.1, 60.3, 128.3, 128. 8, 129.0, 129.9, 130.2, 130.4, 130.8, 132.0, 133.1,133.7, 134.8, 142.5, 145.16, 145.21, 167.7; **HRMS** (ESI) m/z calculated for C₂₅H₂₆NO₅S₂ [M+H]⁺: 484.1141, found: 484.1145.

(*E*)-4-(p-tolyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3b). White solid, m.p. 161-162 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.21 (s, 3H), 2.38 (s, 3H), 2.44 (s, 3H), 3.21 (s, 2H), 4.68 (d, *J* = 6.5 Hz, 2H), 7.14 (d, *J* = 8.0 Hz, 2H), 7.20 (d, *J* = 8.0 Hz, 2H), 7.34 (d, *J* = 8.5 Hz, 4H), 7.72 (d, *J* = 8.5 Hz, 2H), 7.76 (d, *J* = 8.5 Hz, 2H), 7.91 (s, 1H), 8.00 (t, *J* = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.7, 35.2, 60.3, 128.2, 128.8, 129.3, 129.7, 129.9, 130.1, 130.6, 131.7, 133.7; 134.9, 141.5, 142.5, 145.08, 145.10, 167.8; HRMS (ESI) m/z calculated for C₂₆H₂₈NO₅S₂ [M+H]⁺: 498.1403, found: 498.1408.

(E)-4-(4-methoxyphenyl)-3-tosyl-N-(tosylmethyl)but-3-

enamide (3c). White solid, m.p. 165-166 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.21 (s, 3H), 2.42 (s, 3H), 3.20 (s, 2H), 3.83 (s, 3H), 4.67 (d, J = 6.5 Hz, 2H), 6.89 (d, J = 9.0 Hz, 2H), 7.12 (d, J = 8.5 Hz, 2H), 7.33 (d, J = 8.0 Hz, 2H), 7.44 (d, J = 9.0 Hz, 2H), 7.70 (d, J = 8.0 Hz, 2H), 7.75 (d, J = 8.0 Hz, 2H), 7.87 (s, 1H), 8.06 (t, J = 7.0Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.2, 55.4, 60.3, 114.4, 124.7, 128.1, 128.8, 129.75, 129.81, 130.1, 132.6, 133.7, 135.1, 142.0, 144.95, 145.04, 161.7, 168.0; **HRMS** (ESI) m/z calculated for C₂₆H₂₈NO₆S₂ [M+H]⁺: 514.1353, found: 514.1352.

(*E*)-4-(benzo[*d*][1,3]dioxol-5-yl)-3-tosyl-*N*-(tosylmethyl)but-3enamide (3d). White solid, m.p. 133-134 $^{\circ}$ C; ¹H NMR (500

MHz, CDCl₃) δ 2.26 (s, 3H), 2.43 (s, 3H), 3.20 (s, 2H), 4.67 (d, J = 6.5 Hz, 2H), 6.01 (s, 2H), 6.82 (d, J = 8.5 Hz, 1H), 7.00 (d, J = 7.5 Hz, 2H), 7.18 (d, J = 8.0 Hz, 2H), 7.33 (d, J = 8.0 Hz, 2H), 7.73-7.76 (m, 4H), 7.82 (s, 1H), 8.00 (t, J = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.0, 60.3, 101.8, 108.7, 109.8, 126.2, 126.7, 128.1, 128.8, 129.8, 130.1, 130.6, 133.7, 135.0, 142.1, 145.0, 145.1, 148.2, 149.9, 167.8; HRMS (ESI) m/z calculated for C₂₆H₂₆NO₇S₂ [M+H]⁺: 528.1145, found: 528.1148.

(*E*)-4-(2-chlorophenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3e). White solid, m.p. 180-181 °C; ¹H NMR (500 MHz, CDCl3) δ 2.29 (s, 3H), 2.45 (s, 3H), 3.10 (s, 2H), 4.69 (d, *J* = 6.5 Hz, 2H), 7.25-7.30 (m, 5H), 7.33-7.38 (m, 3H), 7.43 (d, *J* = 8.0 Hz, 1H), 7.78 (d, *J* = 8.0 Hz, 2H), 7.81 (d, *J* = 8.5 Hz, 2H), 7.87 (t, *J* = 6.5 Hz, 1H), 8.17 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.6, 21.7, 34.9, 60.4, 127.1, 128.4, 128.9, 129.6, 130.2, 130.5, 131.3, 133.8, 134.5, 134.8, 136.3, 140.1, 145.3, 145.3, 167.7; HRMS (ESI) m/z calculated for C₂₅H₂₅ClNO₅S₂ [M+H]⁺: 518.0867, found: 518.0870.

(*E*)-4-(3-chlorophenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3f). White solid, m.p. 159-160 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.21 (s, 3H), 2.45 (s, 3H), 3.20 (s, 2H), 4.70 (d, *J* = 6.5 Hz, 2H), 7.17

(d, J = 8.0 Hz, 2H), 7.31-7.35 (m, 2H), 7.36-7.41 (m, 3H), 7.42 (s, 1H), 7.74 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 8.0 Hz, 2H), 7.84 (s, 1H), 7.99 (t, J = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 34.9, 60.3, 128.0, 128.3, 128.8, 129.8, 130.0, 130.05, 130.13, 130.2, 130.5, 133.7, 133.8, 134.6, 134.8, 135.1, 140.8, 145.2, 145.4, 167.4; HRMS (ESI) m/z calculated for C₂₅H₂₅ClNO₅S₂ [M+H]⁺: 518.0950,

(*E*)-4-(4-chlorophenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3g). White solid, m.p. 189-190 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.24 (s, 3H), 2.44 (s, 3H), 3.20 (s, 2H), 4.68 (d, *J* = 6.5 Hz, 2H), 7.16 (d, *J* = 7.5 Hz, 2H), 7.34-7.40 (m, 6H), 7.72 (d, *J* = 8.5 Hz, 2H), 7.76 (d, *J* = 8.0 Hz, 2H), 7.86 (s, 1H), 7.99 (t, *J* = 6.0 Hz,1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.7, 35.1, 60.3, 128.3, 128.8, 129.2, 129.9, 130.2, 130.5, 133.8, 131.7, 133.76, 133.79, 134.6, 137.0, 141.0, 145.2, 145.4, 167.6; HRMS (ESI) m/z calculated for C₂₅H₂₅ClNO₅S₂ [M+H]⁺: 518.0977, found: 518.0980.

(*E*)-4-(4-bromophenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3h). White solid, m.p. 185-186 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.25 (s, 3H), 2.42 (s, 3H), 3.21 (s, 2H), 4.67 (d, *J* = 6.5 Hz, 2H), 7.16 (d, *J* = 8.0 Hz, 2H), 7.30 (d, *J* = 8.0 Hz, 2H), 7.33 (d, *J* = 8.0 Hz, 2H), 7.48 (d, *J* = 8.0 Hz, 2H), 7.72 (d, *J* = 8.0 Hz, 2H), 7.76 (d, *J* = 8.0Hz, 2H), 7.84 (s, 1H), 7.99 (t, *J* = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.0, 60.3, 125.3, 128.3, 128.8, 129.9, 130.2, 131.0, 131.7, 132.2, 133.7; 134.1, 134.6, 141.0, 145.2, 145.3, 167.5; HRMS (ESI)

m/z calculated for $C_{25}H_{24}BrNaNO_5S_2$ [M+Na]⁺: 586.0148,

(*E*)-4-(2-bromo-4-methoxyphenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3i). White solid, m.p.174-175 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.29 (s, 3H), 2.43 (s, 3H), 3.13 (s, 2H), 3.71 (s, 3H), 4.70 (d, *J* = 6.5 Hz, 2H), 6.80-6.82 (m, 1H), 6.94 (s, 1H), 7.26 (d, *J* = 8.0, 2H), 7.35 (d, *J* = 8.0 Hz, 2H), 7.45 (d, *J* = 9.0 Hz, 1H), 7.78 (d, *J* = 8.0 Hz, 2H), 7.83 (d, *J* = 8.5 Hz, 2H), 8.04 (t, *J* = 6.5 Hz, 1H), 8.10 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.0, 56.0, 60.5, 114.7, 114.9, 118.9, 128.4, 128.8, 130.0, 130.1, 132.8, 133.3; 133.8, 134.8, 136.1, 142.6, 145.27, 145.32, 159.0, 167.8; HRMS (ESI) m/z calculated for C₂₆H₂₆BrNO₆S₂ [M+H]⁺: 594.0463. found: 594.0461.

(*E*)-4-(5-bromo-2-methoxyphenyl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3j). White solid, m.p. 169-170 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.20 (s, 3H), 2.43 (s, 3H), 3.10 (s, 2H), 3.88 (s, 3H), 4.66 (d, *J* = 6.5 Hz, 2H), 6.81 (d, *J* = 8.0 Hz, 1H), 7.18 (d, *J* = 8.0 Hz, 2H), 7.28 (d, *J* = 2.0 Hz, 1H), 7.34 (d, *J* = 8.0 Hz, 2H), 7.47 (dd, *J* = 2.5 Hz, *J* = 8.0 Hz, 1H), 7.68 (d, *J* = 8.0 Hz, 2H), 7.77 (d, *J* = 8.0 Hz, 2H), 7.84 (t, *J* = 6.5 Hz, 1H), 8.14 (s, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ 21.5, 21.7, 34.9, 56.0, 60.3, 112.3, 112.9, 123.0, 128.3, 128.9, 129.9, 130.1, 132.4, 133.5, 134.4, 135.0, 137.4, 145.1, 145.2, 156.9, 167.3; HRMS (ESI) m/z calculated for C₂₆H₂₇BrNO₅S₂ [M+H]⁺: 592.0463, found: 592.0465.

(*E*)-4-(2,4-dichlorophenyl)-3-tosyl-*N*-(tosylmethyl)but-3enamide (3k). White solid, 195-196 $^{\circ}$ C; ¹H NMR (500 MHz,

Organic & Biomolecular Chemistry Accepted Manuscrip

Journal Name

Journal Name

CDCl₃) δ 2.32 (s, 3H), 2.45 (s, 3H), 3.10 (s, 2H), 4.70 (d, J = 6.5 Hz, 2H), 7.21 (d, J = 7.5 Hz, 1H), 7.27-7.32 (m, 4H), 7.37 (d, J = 8.0 Hz, 2H), 7.45 (s, 1H), 7.79-7.81 (m, 4H), 7.89 (t, J = 6.5 Hz, 1H), 8.08 (s, 1H); ¹³**C NMR** (CDCl₃, 125 MHz) δ 21.6, 21.7, 35.0, 60.3, 127.6, 128.4, 128.9, 129.0, 129.6, 130.0, 130.3, 131.4, 133.8, 134.6, 135.3, 136.9, 138.9, 145.4, 145.5, 167.6; **HRMS** (ESI) m/z calculated for C₂₅H₂₄Cl₂NO₅S₂ [M+H]⁺: 552.0467, found: 552.0470.

(*E*)-4-(naphthalen-1-yl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3). White solid, m.p. 196-197 °C; ¹H NMR (500 MHz, CDCl₃) δ 1.90 (s, 3H), 2.45 (s, 3H), 3.30 (s, 2H), 4.73 (d, *J* = 6.5 Hz, 2H), 7.01 (d, *J* = 8.0 Hz, 2H), 7.36 (d, *J* = 8.0 Hz, 2H), 7.50-7.57 (m, 3H), 7.71 (d, *J* = 8.0 Hz, 2H), 7.81-7.85 (m, 5H), 7.93 (s, 1H), 8.09 (s, 1H), 8.11 (t, *J* = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.2, 21.7, 35.4, 60.3, 126.7, 126.9, 127.5, 128.0, 128.3, 128.6, 129.1, 129.4, 129.8, 130.2, 131.4, 132.9, 133.1, 133.7, 133.9, 134.8, 142.5, 145.1, 145.3, 167.9; HRMS (ESI) m/z calculated for C₂₉H₂₈NO₅S₂ [M+H]⁺: 534.1389, found: 534.1387.

(E)-4-(thiophen-2-yl)-3-tosyl-N-(tosylmethyl)but-3-enamide

(3m). White solid, m.p. 177-178 °C; ¹H NMR (500 MHz, CDCl3) δ 2.29 (s, 3H), 2.44 (s, 3H), 3.29 (s, 2H), 4.64 (d, J = 6.5 Hz, 2H), 7.13-7.17 (m, 3H), 7.34 (d, J = 8.0 Hz, 2H), 7.46 (d, J = 3.5 Hz, 1H), 7.56 (d, J = 5.0 Hz, 1H), 7.69 (d, J = 8.0 Hz, 2H), 7.75-7.80 (m, 3H), 8.03 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.6, 35.3, 60.3, 128.2, 128.6, 128.8, 129.4, 129.7, 130.2, 131.7, 133.7, 133.8, 133.9, 134.8, 135.2, 145.0, 145.2, 166.9; **HRMS** (ESI) m/z calculated for C₂₃H₂₃NNaO₅S₃ [M+Na]⁺: 512.0631, found: 512.0635.

(*E*)-4-(furan-2-yl)-3-tosyl-*N*-(tosylmethyl)but-3-enamide (3n). White solid, m.p. 141-142 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.35 (s, 3H), 2.43 (s, 3H), 3.36 (s, 2H), 4.61 (d, *J* = 7.0 Hz, 2H), 6.50 -6.52 (m, 1H), 6.88 (d, *J* = 3.5 Hz, 1H), 7.19 (d, *J* = 8.0 Hz, 2H), 7.33 (d, *J* = 8.0 Hz, 2H), 7.51 (t, *J* = 6.5 Hz, 1H), 7.56 (s, 1H), 7.67-7.69 (m, 3H), 7.74 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.6, 21.7, 35.2, 60.3, 112.8, 118.7, 128.0, 128.2, 128.7, 129.1, 129.8, 130.1, 133.8, 134.9, 145.0, 145.1, 146.3, 148.4, 167.3; HRMS (ESI) m/z calculated for C₂₃H₂₄NO₆S₂ [M+H]⁺: 474.1040, found: 474.1044.

(E)-4-(4-(allyloxy)phenyl)-3-tosyl-N-(tosylmethyl)but-3-

enamide (30). White solid, m.p. 165-166 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.14 (s, 3H), 2.36 (s, 3H), 3.13 (s, 2H), 4.90 (d, J = 5.0 Hz, 2H), 4.61 (d, J = 7.0 Hz, 1H), 5.23 (d, J = 10.5 Hz, 1H), 5.32 (d, J = 17.0 Hz, 1H), 5.92-6.00 (m, 1H), 6.84 (d, J = 9.0 Hz, 2H), 7.06 (d, J = 8.0 Hz, 2H), 7.26 (d, J = 8.0 Hz, 2H), 7.36 (d, J = 8.5 Hz, 2H), 7.64 (d, J = 8.0 Hz, 2H), 7.68 (d, J = 8.0 Hz, 2H), 7.68 (d, J = 8.0 Hz, 2H), 7.68 (d, J = 8.0 Hz, 2H), 7.80 (s, 1H), 8.00 (t, J = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ 21.5, 21.6, 35.2, 60.3, 68.8, 115.1, 118.2, 124.8, 128.1, 128.8, 129.8, 129.9, 130.1, 132.4, 132.6, 133.7, 135.1, 142.0, 144.9,145.1, 160.7, 168.0; HRMS (ESI) m/z calculated for C₂₈H₃₀NO₆S₂ [M+H]⁺: 540.1515, found: 540.1520.

ARTICLE

(*E*)-4-(3-(phenylethynyl)phenyl)-3-tosyl-*N*-(tosylmethyl)but-3enamide (3p). White solid, m.p. 149-150 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.21 (s, 3H), 2.44 (s, 3H), 3.24 (s, 2H), 4.70 (d, J = 6.5 Hz, 2H), 7.15 (d, J = 8.0 Hz, 2H), 7.36-7.42 (m, 7H), 7.53-7.57 (m, 4H), 7.74 (d, J = 8.5 Hz, 2H), 7.78 (d, J = 8.0 Hz, 2H), 7.89 (s, 1H), 7.96 (t, J = 6.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.5, 21.6, 35.0, 60.4, 88.1, 90.7, 122.7, 124.2, 128.3, 128.4, 128.6, 128.8, 129.1, 129.2, 129.9, 130.2, 131.7, 132.4, 133.3, 133.6, 133.8,134.6, 134.8, 141.6, 145.2, 145.3, 167.5; **HRMS** (ESI) m/z calculated for C₃₃H₃₀NO₅S₂ [M+H]⁺: 584.1567, found: 584.1567.

(*E*)-4-phenyl-3-tosylbut-3-enamide (5). White solid, m.p. 193-194 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.44 (s, 3H), 3.33 (s, 2H), 5.45 (s, 1H), 6.90 (s, 1H), 7.35 (d, *J* = 8.0 Hz, 2H), 7.42-7.47 (m, 3H), 7.59-7.63 (m, 2H), 7.80 (d, *J* = 8.5 Hz, 2H), 7.99 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.7, 35.3, 128.3, 129.1, 130.1, 130.3, 130.7, 132.3, 134.0, 135.0, 142.0, 145.1, 170.3; HRMS (ESI) m/z calculated for C₁₇H₁₈NO₃S [M+H]⁺: 518.0950, found: 518.0954.

(Z)-4,6-diphenyl-3-tosyl-N-(tosylmethyl)hex-3-en-5-ynamide

(7). White solid, m.p. 215-216 °C; ¹H NMR (500 MHz, CDCl₃) δ 2.33 (s, 3H), 2.38 (s, 3H), 3.48 (s, 2H), 4.75 (d, J = 6.5 Hz, 2H), 7.00 (t, J = 6.5 Hz, 1H), 7.19 (d, J = 8.0 Hz, 2H), 7.30 (d, J = 8.0 Hz, 2H), 7.33-7.42 (m, 10H), 7.84–7.88 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.65, 21.70, 31.0, 37.9, 60.5, 86.9, 106.3, 121.9, 128.2, 128.41, 128.43, 128.5, 129.0, 129.2, 129.4, 129.7, 130.1, 131.7, 136.9, 137.0, 137.1, 141.0, 144.3, 145.2, 168.6; HRMS (ESI) m/z calculated for C₃₃H₂₉NO₅S₂ [M+H]⁺: 518.0950, found: 518.0954.

Acknowledgements

We gratefully acknowledge NNSFC (21372038) for financial support.

Notes and references

- (a) A. M. van Leusen, J. Strating, Quart. Rep. Sulfur. Chem., 1970, 5, 67; (b) A. M. van Leusen, G. J. M. Boerma, R. B. Helmholdt, H. Siderius, J. Strating, Tetrahedron Lett., 1972, 2367.
- 2 (a) V. K. Tandon, S. Rai, *Sulfur Rep.*, 2003, 24, 307; (b) D. Van Leusen, A. M. Van Leusen, *Org. React.*, 2001, 57, 417; (c) W. Chen, J. Shao, Z. Li, M. A. Giulianotti and Y. Yu, *Can. J. Chem.*, 2012, 90, 214; (d) M. S. Addie, R. J. K. Taylor, *J. Chem. Soc.*, *Perkin Trans.* 1 2000, 527; (e) I. Collins, *J. Chem. Soc.*, *Perkin Trans.* 1 2000, 2845; (f) T. L. Gilchrist, *J. Chem. Soc.*, *Perkin Trans.* 1 2001, 2491.
- For recent references on synthesis of oxazoles: (a) G. A. Molander, F. A. Wilma, J. G. Ludivine, Org. Lett., 2009, 11, 3830; (b) J. Sisko, A. J. Kassick, M. Mellinger, J. J. Filan, A. Allen, M. A. Olsen, J. Org. Chem., 2000, 65,1516.
- For recent references on synthesis of imidazoles: (a) R. Abbas, E. V. Miranda, A. K. Meysam, *Tetrahedron* 2013, 69, 4199; (b) B. C. Chen, M. S. Bednarz, R. Zhao, J. E. Sundeen, P. Chen, Z. Shen, A. P. Skoumbourdis, J. C. Barrish, *Tetrahedron Lett.*, 2000, 41, 5453.
- 5 For recent references on synthesis of pyrroles: (a) R. Wang, S. Y. Wang, S. J. Ji, *Tetrahedron* **2013**, *69*, 10836; (b) M. Adib, B.

Mohammadi, E. Sheikhi, H. R. Bijanzadeh, *Chinese Chem.* Lett., **2011**, *22*, 314; (c) L. Zhang, X. Xu, W. Xia, Q. Liu, Adv. Synth. Catal., **2011**, *353*, 2619; (d) A. Baeza, J. Mendiola, C. Burgos, J. Alvarez-Builla, J. J. Vaquero, Eur. J. Org. Chem., **2010**, 5607; (e) V. Este'vez, M. Villacampa, J. Carlos Mene 'ndez, Chem. Soc. Rev., **2014**, *43*, 4633; (f) L. Zhang, X. Xu, Q. Shao, L. Pan, Q. Liu, Org. Biomol. Chem., **2013**, *11*, 7393; (g) R. Wang, S. Y. Wang, S. J. Ji, Org. Biomol. Chem., **2014**, *12*, 1735; (h) R. Sharma, K. Kumar, M. Chouhan, V. Grover, V. A. Nair, *RSC Adv.*, **2013**, *3*, 14521; (i) J. Liu, Z. Fang, Q. Zhang, Q. Liu, X. Bi, Angew. Chem. Int. Ed., **2013**, *52*, 6953.

- For recent references on synthesis of indoles: (a) J. Campo, M. Garc'ıa-Valverde, S. Marcaccini, M. J. Rojoa, T. Torroba, Org. Biomol. Chem., 2006, 4, 757; (b) H. Wang, Y. L. Zhao, C. Q. Ren, A. Diallo, Q. Liu, Chem. Commun., 2011, 47, 12316; (c) Y. Li, X. Xu, H. Shi, L. Pan, Q. Liu, J. Org. Chem., 2014, 79, 5929.
- 7 (a) M. Adib, B. Mohammadi, E. Sheikhi, H. R. Bijanzadeh, *Chinese Chem. Lett.*, **2011**, *22*, 314; (b) A. Alizadeh, H. Masrouri, S. Rostamnia, F. Movahedi, *Helv. Chim. Acta* **2006**, *89*, 923; (c) H. Saikachi, T. Kitagawa, H. Sasaki, *Chem. Pharm. Bull.*, **1979**, *27*, 2857; (d) S. Kamijo, C. Kanazawa, Y. Yamamoto, J. Am. Chem. Soc., **2005**, *127*, 9260; (e) O. V. Larionov, A. Meijere, *Angew. Chem. Int. Ed.*, **2005**, *44*, 5664.
- 8 Q. Cai, F. Zhou, T. Xu, L. Fu, K. Ding, Org. Lett., **2011**, 13, 340.
- 9 (a) J. Q. Liu, Z. H. Liu, N. N. Wu, P. Q. Liao, X. H. Bi, Chem. Eur. J., 2014, 20, 2154; (b) J. Q. Liu, Z. H. Liu, P. Q. Liao, X. H. Bi, Org. Lett., 2014, 16, 6204.
- (a) D. C. Meadows, J. Gervay-Hague, *Med. Res. Rev.*, **2006**, *26*, 793;
 (b) N. Suree, M. E. Jung, R. T. Clubb, *Mini-Rev. Med. Chem.*, **2007**, *7*, 991;
 (c) M. M. M. Santos, R. Moreira, *Mini-Rev. Med. Chem.*, **2007**, *7*, 1040.
- (a) T. Nishimura, Y. Takiguchi, T. Hayashi, J. Am. Chem. Soc., 2012, 134, 9086; (b) J. N. Desrosiers, W. S. Bechara, Org. Lett., 2008, 10, 2315; (c) J. N. Desrosiers, A. Charette, Angew. Chem., 2007, 119, 6059; Angew. Chem. Int. Ed., 2007, 46, 5955; (d) N. S. Simpkins, Sulphones in Organic Synthesis; Pergamon Press: Oxford, 1993; (e) P. L. Fuchs, T. F. Braish, Chem. Rev., 1986, 86, 903; (f) K. Tanaka, A. Kaji, In The Chemistry of Sulphones and Sulphoxides; S. Patai, Z. Rappoport, C. J. M. Stirling, Eds.; Chichester: Wiley, 1988, Chapter 15; (g) J. E. Ba¨ckvall, R. Chinchilla, C. Na jera, M. Yus, Chem. Rev., 1998, 98, 2291.
- 12 (a) J. Peng, J. Zhao, Z. Hu, D. Liang, J. Huang, Q. Zhu, Org. Lett., 2012, 14, 4966; (b) G. Vishnu, P. Srivastava, L. Dhar, S. Yadav, Tetrahedron Lett., 2011, 52, 4622.
- 13 D. A. Alonso, C. Najera, M. C. Pacheco, J. Org. Chem., 2004, 69, 1615.