# Organic \& Biomolecular Chemistry 

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## Uncatalysed diaryldiazo cyclopropanations on bicyclic lactams: access to annulated prolines

Lawrence Harris, Martin Gilpin, Amber L. Thompson, Andrew R. Cowley and Mark G. Moloney*
The Department of Chemistry, Chemistry Research Laboratory, The University of Oxford, 12 Mansfield Road, Oxford. OX1 3TA.

GlaxoSmithKline Research \& Development Ltd, New Frontiers Science Park, Third Avenue, Harlow, Essex, CM19 5AW.


#### Abstract

The uncatalysed cycloaddition of substituted diaryldiazo compounds onto bicyclic unsaturated lactams derived from pyroglutamic acid efficiently leads to highly functionalised azatricyclononanes. The products are readily elaborated to deprotected pyroglutamate derivatives, providing rapid access to conformationally constrained amino acids and their analogues. Preliminary assessment of antibacterial activity against one Gram positive and one Gram negative organism indicated high levels of efficacy in some cases.


Conformationally constrained amino acid analogues (CCAAs) have attracted considerable interest over recent years for their applications in pharmacologically active compounds, ${ }^{1-3}$ and especially for peptidomimetics ${ }^{4-7}$ and neurotransmitters. ${ }^{3,8-10}$ Key reasons are that their well-defined conformation is likely to suffer a lower entropic cost upon binding to the target receptor, and this is often coupled with a greater degree of selectivity/potency ${ }^{1}$ and metabolic stability. ${ }^{3,10}$ Moreover, different conformationally constrained analogues of the same parent ligand may show selectivity to different receptors or receptor subtypes, ${ }^{9,11}$ and as a result, spirocyclic systems for example have become very important. ${ }^{12}$ Cyclopropanes are popular conformationally restricted spirocyclic motifs, and for this reason the development of novel methodology for cyclopropanation processes has been of significant recent interest. ${ }^{13}$ Cyclopropanation of electron deficient alkenes under catalysed ${ }^{14-19}$ or uncatalysed conditions ${ }^{20-24}$ leading to the synthesis of contiguous chiral centres ${ }^{25}$ and hindered systems ${ }^{26}$ has become widely available. We have shown that bicyclic lactams derived from pyroglutamic acid provide useful scaffolds for modification of the ring periphery by cycloaddition reactions, ${ }^{27}$ as well as nucleophilic additions of organometallics ${ }^{28,}{ }^{29}$ and amines, ${ }^{30}$ and of interest was whether this approach could be extended to include cyclopropyl-
annulated structures, especially since such systems have become of significance as GluR agonists/antagonists. ${ }^{29}$ Recent work by Madalengoitia has accessed such systems using sulfur ylid additions, ${ }^{31-35}$ but we report here that azabicyclo[3.1.0]hexanes may also be formed by direct stereoselective cyclopropanation of bicyclic lactams 1a,b (Scheme 1) using substituted diaryldiazomethanes 2a-d and 9-diazo-9H-fluorenes 3a-f (Figure 1) under thermal conditions; some of this work has appeared earlier in preliminary form. ${ }^{36}$

The required diaryldiazo compounds $\mathbf{2 a} \mathbf{- d}$ and $\mathbf{3 a - f}$ were easily generated either by conversion of the corresponding ketone to the hydrazone followed by yellow mercuric oxide oxidation, or to the tosylhydrazone followed by basic elimination. ${ }^{37}$ They are highly coloured crystalline solids or viscous oils, readily storable at $0{ }^{\circ} \mathrm{C}$, which only decompose at elevated temperature. ${ }^{38}$ These compounds were reacted with unsaturated lactams 1a,b, prepared as previously reported (Scheme 1 and Table 1); ${ }^{27,29}$ earlier work has shown that $\mathbf{1 b}$ is significantly more reactive than $\mathbf{1 a}$ as a result of the presence of the additional electron withdrawing group. $^{39}$ In an initial investigation, reaction of enone $\mathbf{1 a}$ with diphenyldiazomethane 2c in acetone solvent at reflux gave cyclopropane $\mathbf{4 a}$ in $23 \%$ yield, along with the olefin product $\mathbf{6 a}$, in $26 \%$ yield; however, the reaction was sluggish, and did not go to completion even after 1 week (Table 1, entry 1). Pyrazoline 5 was also obtained in $26 \%$ yield, which although indefinitely stable at room temperature, upon heating at reflux in toluene quantitatively yielded cyclopropane $\mathbf{4 a}$; the alternative regiosisomeric pyrazoline was not isolated. Pyrazoline formation, but not cyclopropanation, has been reported using a similar approach in related systems, ${ }^{40,41}$ and DABCO-promoted synthesis of pyrazoles directly from tosylhydrazones and nitroalkenes is known. ${ }^{42}$ By contrast, similar reaction of diazofluorene 3a was much more efficient, giving cyclopropyl adduct 7a in good yield (66\%) as the only isolable product, although the analogous dibromo derivative $\mathbf{3 b}$ gave only a very low return of the expected product 7b (Table 1, entries 2 and 3). The exo-stereochemistry of the isolated cyclopropane products $156 / 4 a$ and $159 / 7$ a was confirmed by single crystal X-ray analysis (Figure 2) ${ }^{43}$ but similar confirmation of the structure of $\mathbf{5}$ was not possible, since satisfactory crystals could not be obtained. Similar carbene additions to fullerenes ${ }^{44}$ and of heteroaryl carbenes ${ }^{38}$ are known.

However, this process was found to be much more efficient when applied to activated enone 1b (Scheme 1 and Table 1, entries 4-13), giving the expected cyclopropanation products 4b-e and 8a-f as single diastereomers in good yield, even though the starting compound is very prone to dimerisation to give adduct $\mathbf{1 0} .^{27}$ Of interest is that the increase in electron density of the diazo series $\mathbf{2 a} \rightarrow \mathbf{2 d}$ gives significantly increased yields of
cyclopropane adducts (Table 1, entries 4-7), and that in some cases these were obtained along with products of type 6, although the corresponding fluorenyl derivatives 9 were not observed. In this case, the use of room temperature or refluxing acetone conditions was dictated by the stability of the diazo compound partner. These reactions were found to be significantly faster than for lactam 1a, and no pyrazoline adducts analogous to $\mathbf{5}$ were isolated. In contrast to the series of substituted diphenyldiazomethanes, reaction of $\mathbf{1 b}$ with a range of substituted diazofluorenes furnished the product cyclopropanes exclusively (Scheme 1 and Table 1, entries 8-13). Significantly, with unsymmetrical diazo compounds 3b,c,e, two diastereomeric exo-cyclopropanes were obtained in unequal amounts (Table 2, entries 9, 10, 12) although their separation required lengthy chromatography. The best yields were obtained with relatively electro-neutral diazo compounds $\mathbf{3 a}, \mathbf{b}$ and $\mathbf{3 e}$, while a slight decrease is observed for the 2 - and 2,7-substituted bromo-derivatives $\mathbf{3 c}, \mathbf{d}$ and the most electron deficient diazo compound, $\mathbf{3 f}$, produced the lowest yield. In the literature, diazo $\mathbf{3 f}$ is reported to be unstable, ${ }^{45}$ but after two days exposure to the reaction conditions, it was possible to isolate a portion of the unreacted diazo compound after column chromatography of the reaction mixture. The exo-stereochemistry of the structures $\mathbf{8 c}, \mathbf{8 d}$ and $\mathbf{8 e}{ }^{\prime}$ was assigned in several cases using NOE analysis (Figure 3). The two non-crystalline diastereomers $\mathbf{8 b}, \mathbf{b}$, were separable but were distinguishable by ${ }^{1} \mathrm{H}$ NMR spectroscopy only with difficulty. Assignment of stereochemistry in this case required a series of NOE and TOCSY experiments (Figure 3), utilizing firstly, the existence of the key NOE enhancement observed via irradiation of the $\delta 3.5$ singlet for H-6 to identify the proximal aromatic hydrogen, secondly NOE from the aromatic $\mathrm{CH}_{2} \mathrm{OH}$ to flanking $o$-hydrogens, and finally a TOCSY spectrum to identify which of the three possible aromatic rings on which these hydrogens were located. Moreover, of interest was that in the ${ }^{1} \mathrm{H}$ NMR spectra, the signal at $c a$. 83.5 , diagnostic of cyclopropyl H-6, always occurs as a singlet as a result of its orthogonality to $H-5$; a similar feature has been previously reported by Hendrata et al. ${ }^{46}$ This stereochemical assignment was confirmed for products $\mathbf{4 c} \mathbf{- e}, \mathbf{8 a}, \mathbf{8 d}$ and $\mathbf{8 e} \mathbf{e}^{\mathbf{2 4}}$ by single crystal X-ray analysis (Figure 2), ${ }^{43}$ and would appear to be a sterically preferred outcome, since the concave nature of the original template 1b clearly hinders that face. This analysis also clearly shows the proximity between H-6 of the lactam bicycle and the aromatic protons of the substituted cyclopropanes $\mathbf{4 a}$ and 4c-e, as was often detected by nOe analysis.

In contrast to the reactivities of diaryldiazomethanes and fluorenyldiazomethanes towards 1b, ethyl diazoacetate 11a furnished the 2-pyrazoline 12, which is presumably the
most stable tautomer, and whose regiochemistry was established on the basis of careful analysis of ${ }^{13} \mathrm{C}$ shift data (Scheme 1). This is a similar outcome to the cycloaddition reaction with 4-naphthoxybutenolide. ${ }^{25}$ However, diazomalonates 11b and 11c gave no such reaction, even when used as the solvent, and only dimer $\mathbf{1 0}$ was formed.

Table 1: Products and yields from the reaction according to Scheme 1.

| Entry | Substrate | Diazo compound | Conditions ${ }^{\text {a }}$ | Substituents | Product(s) <br> (Yield,\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1a | 2 c | A | $\mathrm{X}=\mathrm{H}$ | $\begin{gathered} \text { 4a }(23) ; \\ \mathbf{5}(26) ; \mathbf{6 a}(26) \end{gathered}$ |
| 2 | 1a | 3 a | A | $\mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{B}=\mathrm{C}=\mathrm{CH}$ | 7 a (66) |
| 3 | 1 a | 3 c | A | $\mathrm{A}=\mathrm{D}=\mathrm{Br}, \mathrm{B}=\mathrm{C}=\mathrm{CH}$ | $7 \mathrm{~b}(<5)$ |
| 4 | 1b | 2 a | A | $\mathrm{X}=\mathrm{NO}_{2}$ | 4b (0) ${ }^{\text {b }}$ |
| 5 | 1b | 2b | A | $\mathrm{X}=\mathrm{I}$ | 4c (19) ${ }^{\text {b }}$ |
| 6 | 1b | 2 c | A | $\mathrm{X}=\mathrm{H}$ | $\begin{aligned} & \text { 4d (29); } \\ & \text { 6d (46) } \\ & \hline \end{aligned}$ |
| 7 | 1b | 2d | B | $\mathrm{X}=\mathrm{OMe}$ | $\begin{aligned} & \text { 4e (64); } \\ & \text { 6e (16) } \\ & \hline \end{aligned}$ |
| 8 | 1b | 3a | A | $\mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{B}=\mathrm{C}=\mathrm{CH}$ | 8a (95) |
| 9 | 1b | 3b | A | $\begin{gathered} \mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{~B}=\mathrm{C}\left(\mathrm{CH}_{2} \mathrm{OH}\right), \\ \mathrm{C}=\mathrm{CH} \end{gathered}$ | 8b (36) |
|  |  |  |  | $\begin{gathered} \mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{~B}=\mathrm{CH}, \mathrm{C}= \\ \mathrm{C}\left(\mathrm{CH}_{2} \mathrm{OH}\right) \end{gathered}$ | 8b' (36) |
| 10 | 1b | 3 c | A | $\mathrm{A}=\mathrm{Br}, \mathrm{B}=\mathrm{C}=\mathrm{CH}, \mathrm{D}=\mathrm{H}$ | 8c (45) |
|  |  |  |  | $\mathrm{A}=\mathrm{H}, \mathrm{B}=\mathrm{C}=\mathrm{CH}, \mathrm{D}=\mathrm{Br}$ | 8c' ${ }^{\text {(24) }}$ |
| 11 | 1b | 3d | A | $\mathrm{A}=\mathrm{Br}, \mathrm{B}=\mathrm{C}=\mathrm{CH}, \mathrm{D}=\mathrm{Br}$ | 8d (64) |
| 12 | 1b | 3 e | A | $\mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{B}=\mathrm{N}, \mathrm{C}=\mathrm{CH}$ | 8 e (58) |
|  |  |  |  | $\mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{B}=\mathrm{CH}, \mathrm{C}=\mathrm{N}$ | 8e' (39) |
| 13 | 1b | 3f | A | $\mathrm{A}=\mathrm{D}=\mathrm{H}, \mathrm{B}=\mathrm{C}=\mathrm{N}$ | 8 f (41) |
| 14 | 1b | 11a | A | $\mathrm{A}=\mathrm{H}, \mathrm{B}=\mathrm{CO}_{2} \mathrm{Et}$ | 12 (74) |
| 15 | 1b | 11b | A | $\mathrm{A}=\mathrm{B}=\mathrm{CO}_{2} \mathrm{Me}$ | $-{ }^{-6}$ |
| 16 | 1b | 11c | A | $\mathrm{A}=\mathrm{Ph}, \mathrm{B}=\mathrm{CO}_{2} \mathrm{Et}$ | - ${ }^{\text {b }}$ |

${ }^{\mathrm{a}}$ Conditions: A, refluxing acetone; B, acetone at room temperature; ; ${ }^{\text {b }}$ Dimer 10 isolated

The oxazolidine ring of these cyclopropanes could be deprotected in high yields (Table 2), using TFA (10 equiv.) with an equivolumetric amount of water and DCM as solvent, although reactions were often slow and needed vigorous conditions and/or longer reaction times to allow complete consumption of starting material. All products (13a-l) maintained the characteristic singlet $\mathrm{H}-3$ in the range $\delta 3.10-3.66 \mathrm{ppm}$ in their ${ }^{1} \mathrm{H}$ NMR spectra derived from H-6 of their parent cyclopropanes, clearly indicating that the exocyclopropane survived the deprotection procedure. Noteworthy was that all deprotected compounds 13a-l were only sparingly soluble in DCM and $\mathrm{CHCl}_{3}$, in contrast to their parent
cyclopropanes, which were all highly soluble. Several were crystalline solids and single Xray crystallographic analysis of $\mathbf{1 3 a}$ and $\mathbf{1 3 c}$ confirmed the expected exo- stereochemistry (Figure 4), ${ }^{43}$ which also showed that the aryl ring systems of phenyl analogue 13a are orthogonal to the coplanar arrangement enforced in fluorene analogue 13c.

Table 2: Yields of Deprotected Products (Scheme 2) and ${ }^{1} \mathrm{H}$ NMR Shift of $H 3$

| Lactam | R | Ar ${ }^{1}$, Ar $^{2}$ | Product | Yield/\% | $\delta(H 3) / \mathrm{ppm}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 4d | $\mathrm{EtO}_{2} \mathrm{C}$ |  | 13a | 78 | 3.38 |
| 4 e | $\mathrm{EtO}_{2} \mathrm{C}$ |  | 13b | 80 | 3.10 |
| 8 a | $\mathrm{EtO}_{2} \mathrm{C}$ | $2$ | 13c | 92 | 3.14 |
| 8b' | $\mathrm{EtO}_{2} \mathrm{C}$ |  | 13d | 32 | 3.34 |
| 8 c | $\mathrm{EtO}_{2} \mathrm{C}$ | - | 13e | 72 | 3.14 |
| $8 c^{\prime}$ | $\mathrm{EtO}_{2} \mathrm{C}$ | $>_{\mathrm{Br}}$ | 13 f | 58 | 3.15 |
| 8d | $\mathrm{EtO}_{2} \mathrm{C}$ | $>_{B r}$ | 13g | 0 (13e, 59\%) | - |
| 8 e | $\mathrm{EtO}_{2} \mathrm{C}$ | $=N$ | 13h | 76 | 3.51 |
| 8 e | $\mathrm{EtO}_{2} \mathrm{C}$ | $\because V_{0}^{N-}$ | 13i | 67 | 3.49 |
| 8 f | $\mathrm{EtO}_{2} \mathrm{C}$ |  | 13j | 96 | 3.66 |
| 4 a | H | © | 13k | 73 | d 2.54 |
| 7 a | H | $12$ | 131 | 93 | d 2.75 |

However, the deprotection of some cyclopropanes gave different behaviour. For example, the basicity of the aza-fluorene moiety clearly had an effect on the ease of the $\mathrm{N}, \mathrm{O}$-acetal hydrolysis, since aza-fluorene compounds $\mathbf{1 3 h} \mathbf{h} \mathbf{i}$ and bipyridine $\mathbf{1 3 j}$ all required a greater number of equivalents (20-30 equiv.) of TFA for their efficient production from starting material. Additionally, purification of azafluorenes 13h-j by washing out impurities with

EtOAc, rather than column chromatography, was more effective. Of particular interest was that $2^{\prime}, 7^{\prime}$ 'dibromo-fluorenyl cyclopropane $\mathbf{8 d}$ did not give the expected product $\mathbf{1 3 g}$, but instead gave mono-debrominated diastereoisomer 13e as the major product in $59 \%$ yield, in which the departing bromine atom was the one at $\operatorname{ArC}\left(7^{\prime}\right)$ (structure established by mass spectrometry, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis, and the stereochemistry was elucidated by NOE analysis (Table 2 and Scheme 2)). Of interest is that in Suzuki chemistry, the bromine substituents of dibromofluorenes are notoriously labile, ${ }^{47}$ and it is possible that a trace impurity of metal in the deprotection of the the di-bromo substituted cyclopropane 8d effected the unwanted debromination. By comparison, deprotection of the two mono-bromo diastereomers 18c and 18c' gave the corresponding deprotected products $\mathbf{1 3 e}$ and $\mathbf{1 3 f}$ in good yield.

Conversion of these derivatives to the pyroglutamate products was investigated; both 13c and 131 were subjected to PDC oxidations and direct esterification (Scheme 2) to furnish the desired products $\mathbf{1 4 a}$ and $\mathbf{1 4 b}$ but in poor yields of 37 and $28 \%$ respectively, and attempted optimization with other oxidants (ruthenium tetroxide, chromic oxide) gave no improvement. When these conditions were applied to diphenyl 13a, surprisingly fluorenyl derivative $\mathbf{1 4 a}$ was obtained in $29 \%$ yield, and none of the expected product diphenyl product was obtained. Similar oxidative cyclodehydrogenation reactions has been reported in electron rich polyaromatic systems in the presence of $\mathrm{FeCl}_{3} .{ }^{47-49}$ and more recently such transformations have been shown to have synthetic value. ${ }^{50-52}$ Swern oxidation was attempted in order to avoid this, but instead of giving the aldehyde directly, dimer $\mathbf{1 5}$ was obtained in $66 \%$ yield; single crystal X-ray analysis of 16 allowed determination of the stereochemistry (Figure 4) and indicated that the central six-membered diketopiperazine ring adopts the boat conformation. ${ }^{43}$ Although this type of dimerisation had been previously reported, ${ }^{39}$ the stereochemistry of the observed product was unknown. Ester hydrolysis of the pyroglutaminols was also investigated following a literature procedure; ${ }^{53}$ diaryl-cyclopropane 13b, fluorenyl-cyclopropane 13c, azafluorenyl 13i and diazafluroenyl 13j derivatives all furnished 16 and 17a-c in excellent yield (Scheme 2). The glutamic acid and proline analogues 18a and 18b were also readily obtained by similar hydrolysis of the respective diand mono-esters 14a and 14b (Scheme 2). In the case of the deprotection of hydroxymethyl $\mathbf{8 b}$, the desired product 13d was isolated in only poor yield (38\%) and this was very unsatisfactory given the difficulty of separating $\mathbf{8 b}$ and $\mathbf{8 b}$. Alcohol $\mathbf{8 b}$ was therefore subjected to a TEMPO-catalysed oxidation, ${ }^{54}$ followed by methylation via diazomethane, and found to yield methyl ester 19a quantitatively and cleanly (Scheme 3). Applying these
conditions to the mixture of diastereomers $\mathbf{8 b}$ and $\mathbf{8 b}$ ' gave easily separable diastereomers 19a,b in quantitative yield, and these in turn could be easily deprotected under acidic conditions in excellent yields to give products 20a and 20b respectively (Scheme 3).

In order to access pyrrolidine systems, reduction of bicyclic lactam 8a with lithium aluminium hydride gave diol 21a in $73 \%$ yield after column chromatography (Scheme 4) although this product was found to decompose on standing; immediate double Swern oxidation furnished the di-aldehyde 21b in $88 \%$ yield after column chromatography. The two proton signals at 9.65-9.85 ppm corresponding to each of aldehydes at $\mathrm{C}(2)$ and $\mathrm{C}(4)$ and the corresponding aldehydic carbon signals could also be clearly seen in the ${ }^{13} \mathrm{C}$ NMR spectra. However, a pure sample of dialdehyde 21b was even less stable than the parent diol precursor (21a), and degraded within a matter of hours in air at room temperature. Selective simultaneous oxidation of both aldehyde groups with $\mathrm{NaClO}_{2}$ did not produce diester $\mathbf{2 2}$ as the major product of this reaction but instead gave lactam 23 in low yield (12\%).

Table 3: Yields of BOC Protected Alcohols (Scheme 5)

| Substrate |  | Product |  | By-product |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{R}=\mathbf{E t O}_{\mathbf{2}} \mathbf{C}$ | Ar$^{1}, \mathrm{Ar}^{2}$ | $\mathbf{R}=\mathbf{\mathbf { C H } _ { 2 } \mathbf { O H }}$ | Yield(/\%) |  |

An alternative approach was therefore sought, and although reaction of pyroglutaminol $\mathbf{1 3 c}$ with $\mathrm{LiAlH}_{4}$ followed by BOC protection gave the product aminodiol $\mathbf{2 4} \mathbf{c}$ in poor ( $12 \%$ ) yield, a better approach was found to be diborane reduction followed by BOCderivatisation, which for 13a-c,h-j successfully gave the fully reduced diols 24a-f (Scheme 5 and Table 3), often along with under-reduced lactams 25a-f. However, lactam 13j yielded no product after workup from its reduction, possibly due to its polarity. In contrast to the mediocre yields of $\mathbf{1 3}$, reduction of the diphenyl-substituted $\mathbf{1 3 k}$ proceeded relatively cleanly to give $\mathbf{2 4 g}$ in an $82 \%$ yield and no starting material was observed or recovered after 2 days under the reaction conditions. In all cases, rotameric equilibration in the NMR spectra was evident, and to confirm this, both ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of diol $\mathbf{2 4 c}$ were obtained at 383 K in $d-6$ toluene as a solvent; separated rotameric peaks were seen to have partially coalesced in the ${ }^{1} \mathrm{H}$ NMR spectrum while no rotameric doubling was observed in the ${ }^{13} \mathrm{C}$ NMR spectrum at this temperature.

TEMPO-catalysed oxidation of diol $\mathbf{2 4 c}$ (Scheme 5), followed by direct column chromatography eluting with $\mathrm{DCM} / \mathrm{MeOH} / \mathrm{AcOH}$ (90:5:4), gave diacid 26a after chromatography. However, for ease of purification and characterisation, diacid 26a was immediately converted to the dimethyl-ester 26b with diazomethane, which was easily obtained from column chromatography in yields varying from 34-65\%, and re-converted to pure diacid 26a by near-quantitative alkaline hydrolysis (96\%). Exposure of diacid 26a to dry TFA in DCM for 3 hours gave clean deprotection to give diacid 27 in 59\% yield. However, application of these TEMPO conditions to diaryls 24a and 24b gave complete consumption of starting material but no concomitant formation of the expected products, while 24d led to mostly recovered starting material (79\%), although use of excess quantities of oxidising reagents gave not only the desired product 28a in only $4 \%$ yield, but also $\mathrm{C}(2)$-mono-oxidised 28b in a $30 \%$ yield after column chromatography, and a trace quantity ( $<2 \%$ ) of $\mathrm{C}(4)$-monooxidised 28c (Scheme 6). This product distribution implies that oxidation of the $\mathrm{C}(4)$-primary alcohol of diol $\mathbf{2 4 d}$ is slower than oxidation of the $\mathrm{C}(2)$-primary alcohol. Similar oxidation of diastereoisomer 24 e yielded an analogous result, although only the $\mathrm{C}(2)$ mono-oxidised product $\mathbf{2 8 e}$ was isolated with a significant yield ( $7 \%$ ). On the premise that slower oxidation of $\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}$ was likely, it was expected that the attempted single oxidation of the $\mathrm{C}(2)$ primary alcohol of proline-analogue precursor $\mathbf{2 4 g}$ would give a cleaner result (Scheme 7); in fact, ester 29a was obtained in $68 \%$ yield after column chromatography. Near quantitative hydrolysis, furnishing acid 29b and subsequent BOC-deprotection (via HCl in dry ether) yielded the final proline templated $\mathbf{2 9}$ c.

## Bioassay Studies

Many of the pyroglutaminols 13, and their derivatives, showed significant activity in holeplate assay against $S$. aureus and E. coli (Table 4, ESI), in which assessment is made by measuring the diameter of the zone of inhibition after overnight incubation, using cephalosporin c as a positive control. ${ }^{55}$ Although not providing full Minimum Inhibitory Concentration data, this method provides a useful preliminary assessment of antibacterial activity. Both fluorenyl containing alcohols $\mathbf{1 3 c}$ and $\mathbf{1 3 1}$ and diacid 18a (entries 1,2 and 5 respectively) were very active against $S$. aureus, while mono acid 18b and mono-bromo 13e (entries 6 and 8 respectively) were very active against E. coli. Remarkably, the alternative mono-bromo diastereoisomer $\mathbf{1 3 f}$ (entry 7) was inactive, as were all of the phenyl analogues (entries 13-16). Of interest is the relative activity of the fluorenyl systems compared to the inactivity of the phenyl analogues (entries 13-16). A possible structural reason for these activities is the planar aromatic ring systems for the fluorenyl analogues, whereas the gemdiphenyl system is non-planar (as observed in the crystal structures, Figures 2 and 3).

## Reaction Mechanism

Mechanistically, the cyclopropanation reaction may proceed via carbene (singlet or triplet) insertion or cycloaddition processes (pyrazoline formation) followed by extrusion of nitrogen (Scheme 8, Paths A and B). The former seems unlikely since the majority of reactions are conducted with excess diazo compound in refluxing acetone, well below the decomposition temperature for diazofluorene and diphenylmethyldiazo compounds (decomposition is reported at $>120^{\circ} \mathrm{C}$, and some are stable even to $>165^{\circ} \mathrm{C}^{37}$ ), and for the more stable diazo compounds $\mathbf{3 c}$ and $\mathbf{3 e}$, the reaction could even be conducted at room temperature. For lactam $\mathbf{1 b}$, there is also the possibility of dimerisation (Path C). The diazo compound $\mathbf{2}$ and $\mathbf{3}$ are also prone to conversion to the azine (Path D), and Huisgen et al. ${ }^{56}$ gives the half life ( $\mathrm{t}_{1 / 2}$ ) of self-decomposition (Path E) of diphenyldiazomethane (2c) (one of the least stable diazo compounds used in solution) in a DMF solution of 72 days at $25^{\circ} \mathrm{C}$ and 10 days at $40{ }^{\circ} \mathrm{C}$. The rate of formation of carbenes derived from the more stable diazofluorenes (via Path E) at room temperature ( $20^{\circ} \mathrm{C}$ ) would therefore seem to be negligible, and in the cases of $\mathbf{3 a} \mathbf{a} \mathbf{d}$ and 3e-f, unreacted diazo compound could even be isolated pure from the reaction mixture after column chromatography. Therefore, diazofluorenes would most likely yield cyclopropanes by initial pyrazoline formation from 1,3-dipolar cycloaddition with enones $\mathbf{1 a}$ and $\mathbf{1 b}$ (Path A)
followed by extrusion of $\mathrm{N}_{2}$ (Path F). Furthermore, the rate of reaction of enone $\mathbf{1 b}$ is greater than 1a, consistent with a concerted pathway giving the pyrazoline products $\mathbf{3 0}$ and $\mathbf{3 1}$ which immediately collapse leading to the formation of the observed products 33 and 34 (Path F). In the case of $\mathbf{1 a}$, both pyrazolines are formed, only one of which immediately collapses to the products $\mathbf{3 3}$ and $\mathbf{3 4}$, and the other ( $\mathbf{3 0}$ ), which can easily be isolated, only does so at elevated temperature. The collapse of pyrazolines to cyclopropanes has been investigated in detail, ${ }^{57-59}$ and the formation of both pyrazoline and cyclopropyl products have been observed in the reactions of diazo compounds with unsaturated maleimide systems. ${ }^{60}$ The rates of such 1,3dipolar cycloadditions for a series of diaryldiazo compounds were found to be largely governed by the HOMO(diazomethane)-LUMO(dipolarophile) interaction. ${ }^{56,}{ }^{61}$ Diphenyldiazomethane is a nucleophilic 1,3 -dipole ${ }^{61-63}$ which reacts preferentially with electrophilic $\mathrm{C}=\mathrm{C}$ bonds, and the rate of reaction increases when the dipolarophile becomes more electron deficient. This is reflected in relative reaction rates, and for example, in the reaction of diazofluorene (3a), 2-bromodiazofluorene (3c) and 2,7-dibromodiazofluorene (3d) with both $\mathbf{1 a}$ and $\mathbf{1 b}$, yields and rates are typically: $\mathbf{8 a}$ ( $95 \%, 1-2$ days), $\mathbf{8 c}$ and $\mathbf{8 c}{ }^{\prime}(69 \%, 2$ days), $\mathbf{8 d}(64 \%, 3$ days), $7 \mathbf{a}(66 \%, 1$ week), and $7 \mathbf{b}(<5 \%, 3$ weeks). Increased steric hindrance in the transition state might also account for the observed selectivity of $45: 24$ (approx. $2: 1$ ) for $\mathbf{8 c}: \mathbf{8 c}$. The collapse of 1-pyrazolines to give cyclopropane and alkene products has been the subject of a number of investigations, and the product distributions are a combination of steric and electronic effects. ${ }^{57-59}$ Amongst the mechanisms for thermal collapse of pyrazoline adducts, the formation of a $(90,90)$ trimethylene intermediate ${ }^{64-67}$ has been proposed and is used to explain the general retention of stereochemistry of the initial pyrazoline. The stepwise or concerted formation of similar diradical species has also been investigated, ${ }^{68}$ and Nakano et al. ${ }^{64}$ found that the thermal stability of a series of pentasubstituted 1-pyrazolines increased with variation of the diazo substitution in the order fluorenyl < diphenyl < phenylmethyl < dimethyl. The abnormal stability of certain pyrazolines has been attributed to the inhibition of conformations which are favourable for decomposition. ${ }^{69}$ Thus, it seems that fluorenyl-spiro-cyclopropanes $\mathbf{8 a - 8 f}$ are likely to form via $(90,90)$ trimethylene intermediates, from rapid collapse of their respective pyrazoline adducts (Path F), which must all be unstable at or above room temperature.

## Conclusion

We have demonstrated that direct cyclopropanation of electronically activated unsaturated pyrrolidinones using diaryl diazo compounds allows stereoselective access to
conformationally well-defined tri- and tetracycloannulated cyclopropyl systems in high yield, some of which exhibit high levels of antibacterial activity. Given the renaissance of interest in the use of pyroglutamates as structurally well-defined building blocks and in drug discovery, ${ }^{70-72}$ this approach demonstrates that escape from flatland is readily achieved using modular chemistry for the generation of 3D templates and scaffolds ${ }^{73}$ for rapid library construction. ${ }^{74,75}$

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## Experimental

For general experimental procedures, see our earlier report. ${ }^{76}(2 R, 5 S)$-1-Aza-3-oxa-8-oxo-2-phenyl-bicyclo-[3.3.0]oct-6-ene 1a and ( $2 R, 5 S$ )-1-aza-7-ethyloxycarbonyl-3-oxa-8-oxo-2-phenyl-bicyclo-[3.3.0]oct-6-ene 1b were prepared using literature methods. ${ }^{27}$
(2R,5S,6S,7S)-1-Aza-2,9,9-triphenyl-8-oxo-3-oxa-tricyclo[4.3.0 $\left.{ }^{1,5} .0^{6,7}\right]$-nonane
(3aS,6R,8aR,8bS)-3,3,6-triphenyl-7-oxa-3,3a,6,7,8,8a-hexahydropyrazolo[3,4-a]
pyrrolizin-4(8bH)-one 5 and (2R,5S)-1-aza-6-benzhydryl-8-oxo-2-phenyl-3-oxa-

## bicyclo[3.3.0]oct-6-ene 6a

A mixture of $\mathbf{2 c}(282 \mathrm{mg}, 1.45 \mathrm{mmol})$ and $\mathbf{1 a}(288 \mathrm{mg}, 1.43 \mathrm{mmol})$ in acetone $(15 \mathrm{~mL})$ was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 3 d . The mixture was concentrated in vacuo to give a viscous red oil which was purified by column chromatography over silica eluting with $\mathrm{EtOAc} /$ Petrol ( $1: 7 \rightarrow 1: 5$ ) to give (in order of elution) 6a as a yellow oil (138 $\mathrm{mg}, 26 \%), \mathbf{4} \mathrm{a}$ as a white powder ( $120 \mathrm{mg}, 23 \%$ ) and $\mathbf{5}$ as a white solid ( $146 \mathrm{mg}, 26 \%$ ).

6a; $\mathrm{R}_{\mathrm{f}}=0.36(\mathrm{EtOAc} /$ Petrol, $1: 3) ;[\alpha]_{\mathrm{D}}{ }^{21}+69.6\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3017,1702$ $(\mathrm{C}=\mathrm{O}), 1495,1452,1334,1217,1030,756,702,667 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.28(1 \mathrm{H}, \mathrm{t}, J$ 8.3, C(4)HH), $3.71(1 \mathrm{H}, \mathrm{dd}, J 6.9,8.2, \mathrm{C}(4) \mathrm{H} H), 4.49-4.54(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(5) H), 5.01(1 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CHPh}_{2}\right), 5.73(1 \mathrm{H}, \mathrm{t}, J 1.3, \mathrm{C}(7) H), 6.16(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.18-7.25(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.27-7.40$ $(9 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.48-7.52(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 52.7\left(C \mathrm{CH}_{2}\right), 66.1(C(5))$,
$68.3(C(4)), 87.2(C(2)), 125.4(C(7)), 126.1(\mathrm{ArCH}), 127.6(p-\mathrm{ArCH}), 127.6\left(p-\mathrm{Ar}{ }^{\prime} C H\right)$, $128.4(\mathrm{ArCH}), 128.5(p-\mathrm{ArCH}), 128.6,128.7,128.9,129.0(\mathrm{all} \mathrm{ArCH}), 138.8,139.4,139.5$ (all ArC$), 166.8$ and $177.0(C(6)$ and $(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 426\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right)$;
$\mathbf{4 a ;} \mathrm{R}_{\mathrm{f}}=0.25(\mathrm{EtOAc} /$ Petrol, $1: 3)$; m.p. $180{ }^{\circ} \mathrm{C}($ dec. $) ;[\alpha]_{\mathrm{D}}{ }^{18}+78.3\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right)$; $v_{\max } / \mathrm{cm}^{-1}($ film $) 1704(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.80(1 \mathrm{H}, \mathrm{ddd}, J 1.1,2.4,6.1, \mathrm{C}(6) H)$, $2.86(1 \mathrm{H}, \mathrm{dd}, J 2.2,6.2, \mathrm{C}(7) H), 3.59(1 \mathrm{H}, \mathrm{ddd}, J 2.5,7.6,9.8, \mathrm{C}(4) H \mathrm{H}), 3.75-3.83(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(5) H), 4.14-4.19(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) \mathrm{H} H), 6.21(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.46(2 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H), 7.05-7.09$ (2H, m, ArH ), $7.12-7.45$ ( $9 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.51-7.53$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ); $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 31.9$ $(C(7)), 36.6(C(6)), 43.2\left(C\left(9^{\prime}\right)\right), 58.5(C(5)), 68.5(C(4)), 86.7(C(2)), 125.4,127.0,127.1$, 127.4, 128.0, 128.0, 128.7, 129.1, 129.9 (all ArCH), 136.9, 138.4, 143.0 (all ArC ), 176.3 $(C=O)$; HRMS $368.1643\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{NO}_{2}\right.$ requires 368.1645); Found C, 82.31; H, 5.37; $\mathrm{N}, 3.89 \%\left(\mathrm{C}_{25} \mathrm{H}_{21} \mathrm{NO}_{2}\right.$ requires $\left.\mathrm{C}, 81.72 ; \mathrm{H}, 5.76 ; \mathrm{N}, 3.81 \%\right)$;

5; $\mathrm{R}_{\mathrm{f}}=0.19(\mathrm{EtOAc} /$ Petrol, $1: 3)$; m.p. $180{ }^{\circ} \mathrm{C}($ dec. $) ;[\alpha]_{\mathrm{D}}{ }^{21}+565.5\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right)$; $v_{\max } / \mathrm{cm}^{-1}($ film $) 1716(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.66(1 \mathrm{H}, \mathrm{t}, J 8.7, \mathrm{C}(4) H \mathrm{H}), 3.87(1 \mathrm{H}, \mathrm{d}$, $J$ 8.8, C(7)H), $4.21(1 \mathrm{H}, \mathrm{ddd}, J 2.4,6.7,8.9, \mathrm{C}(5) H), 4.48(1 \mathrm{H}, \mathrm{dd}, J 6.7,8.0, \mathrm{C}(4) \mathrm{H} H), 5.53$ $(1 \mathrm{H}, \mathrm{dd}, J 8.8, \mathrm{C}(6) H), 6.22(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.96-7.00(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.25-7.31(6 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$, 7.36-7.41 (1H, m, ArH), 7.43-7.44 (2H, m, ArH), 7.70-7.74 (2H, m, ArH); $\delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\left.\mathrm{CDCl}_{3}\right) 52.0(C(7)), 59.2(C(5)), 69.7(C(4)), 86.8(C(2)), 90.9(C(6)), 105.1 C \mathrm{Ph}_{2}\right), 125.4$, 127.2, 127.5 (all ArCH$), 128.1$ ( $p-\mathrm{ArCH}$ ), 128.2, 128.2 (both ArCH$), 128.3$ ( $\left.p^{\prime}-\mathrm{ArCH}\right), 128.4$ ( $\left.p^{\prime \prime}-\mathrm{ArCH}\right), 128.9(\mathrm{ArCH}), 137.7,139.0,140.3$ (all ArC ), $174.9(C=\mathrm{O})$; HRMS 396.1707 $\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{2}\right.$ requires 396.1707).
(2R,5S,6S,7S)-Spiro[1-aza-3-oxa-8-oxo-2-phenyl-tricyclo[4.3.0 ${ }^{1,5} .0^{6,7}$ ]nonane-9,9’fluorene] 7a

A mixture of $\mathbf{3 a}(771 \mathrm{mg}, 4.01 \mathrm{mmol})$ and $\mathbf{1 a}(251 \mathrm{mg}, 1.25 \mathrm{mmol})$ in acetone $(25 \mathrm{~mL})$ was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 16 h . The mixture was concentrated in vacuo to give a red solid. The red solid obtained was passed through a plug of silica, eluting with DCM/Petrol $(2: 3)$ to remove the excess of diazo 3a and give a light brown solid which was purified by column chromatography over silica eluting with EtOAc/Petrol (1:5) to yield 7a (302 mg, 65\%) as a white solid; $\mathrm{R}_{\mathrm{f}}=0.20(\mathrm{EtOAc} /$ Petrol, $1: 3)$; m.p. $153{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{21}+257.3\left(\mathrm{c}=6.2, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}(\mathrm{film}) 3061,1709(\mathrm{C}=\mathrm{O}), 1449,1343,1221,1159$, $1027,774,749,732,716,699 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.98(1 \mathrm{H}, \mathrm{d}, J 6.1, \mathrm{C}(7) H), 3.03(1 \mathrm{H}$, d, $J 6.1, \mathrm{C}(6) H), 3.83(1 \mathrm{H}, \mathrm{t}, J 8.6, \mathrm{C}(4) H \mathrm{H}), 4.22(1 \mathrm{H}, \mathrm{dd}, J 9.0,6.4, \mathrm{C}(5) H), 4.41(1 \mathrm{H}, \mathrm{t}, J$ 7.0, C(4)HH), $6.52(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.91(1 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{Ar} H), 6.98-7.04(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.26-$ $7.51(7 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.57-7.61(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.78(1 \mathrm{H}, \mathrm{d}, J 7.5, \operatorname{Ar} H), 7.81-7.85(1 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 35.5(C(7)), 37.5(C(6)), 39.2(C(9)), 57.9(C(5), 69.7(C(4))$, $88.6(C(2)), 118.6,119.8,120.4,123.7,126.0,126.0,127.2,127.5,127.6,128.6,128.7$ (all ArCH ), 138.5, 138.8, 140.3, 142.2, 145.5 (all ArC ), $174.8(C=\mathrm{O}) ; \mathrm{m} / \mathrm{z}(\mathrm{ESI}+) 424$ $\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 34 \%\right)$; $\mathrm{HRMS} 388.1308\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{25} \mathrm{H}_{19} \mathrm{NNaO}_{2}\right.$ requires 388.1308).

## ( $2 R, 5 S, 6 S, 7 S$ )-2', ${ }^{\prime}$ '-Dibromospiro[1-aza-3-oxa-8-oxo-2-phenyl-tricyclo[4.3.0 $\left.{ }^{1,5} .0^{6,7}\right]$

## nonane-9,9'-fluorene] 7b

A mixture of diazo $\mathbf{3 b}(320 \mathrm{mg}, 0.91 \mathrm{mmol})$ and $\mathbf{1 a}(123 \mathrm{mg}, 0.61 \mathrm{mmol})$ in acetone $(10 \mathrm{~mL})$ was heated at reflux for 3 weeks (under an $\mathrm{N}_{2}$ atmosphere) before being concentrated in vacuo. Column chromatography over silica gel eluting with $\mathrm{EtOAc} / \operatorname{Petrol}(1: 3)$ yielded a residue containing cyclopropane $\mathbf{1 6 1 / 7 b}$ and other unidentified products ( 30 mg ) after concentration. This residue was dissolved in EtOAc and the solution was allowed to evaporate slowly leaving crystalline $\mathbf{1 6 1 / 7 b}(3 \mathrm{mg}, 1 \%) ; \mathrm{R}_{\mathrm{f}}=0.35(\mathrm{DCM}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 1709(\mathrm{C}=\mathrm{O})$, 1451,$812 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.04(2 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H$ and $\mathrm{C}(7) H), 3.84(1 \mathrm{H}, \mathrm{dd}, J 8.0,9.0$,
$\mathrm{C}(4) H \mathrm{H}), 4.24(1 \mathrm{H}, \mathrm{dd}, J 6.4,9.0, \mathrm{C}(5) H), 4.45(1 \mathrm{H}, \mathrm{dd}, J 6.4,8.0, \mathrm{C}(4) \mathrm{H} H), 6.51(1 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(2) H), 7.06\left(1 \mathrm{H}, \mathrm{d}, J 1.6, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.37-7.41(1 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.44-7.53(4 \mathrm{H}, \mathrm{m}, \operatorname{ArH})$, 7.56-7.59 (2H, m, $\operatorname{Ar} H), 7.60(1 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{Ar} H), 7.64(1 \mathrm{H}, \mathrm{d}, J 8.1, \operatorname{Ar} H), 7.74(1 \mathrm{H}, \mathrm{d}, J 1.6$, $\left.\operatorname{ArC}\left(8{ }^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3} ; \mathrm{Me}_{4} \mathrm{Si}\right) 36.0,37.6(C(6)$ and $C(7))$, $38.8\left(C\left(9^{\prime}\right)\right), 57.8$ $(C(5)), 69.5(C(4)), 89.1(C(2)), 121.2,121.6$ (both ArCH$), 121.6(\mathrm{ArCBr}), 121.8(\mathrm{ArCBr})$, $122.2(\mathrm{ArCH}), 125.9(p-\mathrm{PhCH}), 126.9,128.8,129.0,130.8,131.0($ all ArCH$), 136.6,138.3$, 140.1, 142.0, 147.2 (all ArC), 174.3 ( $C=\mathrm{O}$ ); $m / z$ (ESI-) 522 ([M-H] ${ }^{-}$, 78\%); HRMS 521.9536 ([M-H], $\mathrm{C}_{25} \mathrm{H}_{16} \mathrm{BrNO}_{2}$ requires 521.9534).

## (2R,5S,6S,7R)-1-Aza-9,9-di(p-iodophenyl)-7-ethoxycarbonyl-8-oxo-3-oxa-2-phenyl-

 tricyclo[4.1.0 ${ }^{1,5} .0^{6,7}$ ]nonane $4 c$A solution of $\mathbf{2 b}(370 \mathrm{mg}, 0.83 \mathrm{mmol})$ and freshly prepared $\mathbf{1 a}(223 \mathrm{mg}, 0.82 \mathrm{mmol})$ in acetone ( 10 mL ) was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 16 h . The mixture was concentrated in vacuo to give a red oil which was purified by column chromatography over silica eluting with EtOAc/Petrol (1:5) to yield $\mathbf{4 c}(106 \mathrm{mg}, 19 \%)$ as a white crystalline solid; $\mathrm{R}_{\mathrm{f}}=0.23(\mathrm{EtOAc} /$ Petrol, $1: 3)$; m.p. $200{ }^{\circ} \mathrm{C}(\mathrm{dec}) ;[\alpha]_{\mathrm{D}}{ }^{21}-30.2(\mathrm{c}=$ 6.3, $\mathrm{CHCl}_{3}$ ); $\mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}($ film $) 1741(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.04(3 \mathrm{H}, \mathrm{t}, J 7.2$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.44(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.56-3.66(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) H \mathrm{H}, \mathrm{C}(5) H), 4.04(2 \mathrm{H}, \mathrm{q}, J 7.2$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.21(1 \mathrm{H}, \mathrm{dd}, J 6.4,5.1, \mathrm{C}(4) \mathrm{H} H), 6.23(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.45-6.49(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar})$ ), 7.07-7.12 (2H, m, ArH), 7.16-7.27 (5H, m, ArH), 7.59-7.64 (2H, m, ArH), 7.72 (2H, d, $J$ 8.46, $\mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 32.6(C(6)), 46.2$, $48.9(C(4)$ and $\left.C\left(9^{\prime}\right)\right), 56.2(C(5)), 62.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 68.4(C(4)), 87.2(C(2)), 93.4(\mathrm{ArCI}), 93.8(\mathrm{ArCI})$, 125.3, 128.3, 128.4, 130.1, 130.9, 136.6, 137.8, 137.8, 137.9, 138.5, 164.4 and 171.2 (both $(C=\mathrm{O})) ; \quad m / z \quad(\mathrm{ESI}+) 750\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, \quad 100 \%\right) ; \quad \mathrm{HRMS} 713.9595\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $\mathrm{C}_{28} \mathrm{H}_{23} \mathrm{NNaO}_{4}$ requires 713.9609).
(2R,5S,6S,7R)-1-Aza-9,9-diphenyl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo [4.1.0 ${ }^{1,5} .0^{6,7}$ ]nonane 4d, and (2R,5S)-1-Aza-6-benzhydryl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-bicyclo-[3.3.0]oct-6-ene 6d

A solution of $\mathbf{2 c}(0.71 \mathrm{~g}, 3.7 \mathrm{mmol})$ and freshly prepared $\mathbf{1 a}(0.98 \mathrm{~g}, 3.6 \mathrm{mmol})$ in acetone ( 25 mL ) was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 3 d . The mixture was concentrated in vacuo to give a yellow oil which was purified by column chromatography over silica eluting with EtOAc/Petrol (1:5) to yield $\mathbf{4 d}(0.46 \mathrm{~g}, 29 \%)$ as a white powder and $\mathbf{6 d}(0.60 \mathrm{~g}, 46 \%)$ as a pale yellow, viscous oil. Also obtained from the reaction was the azine of diphenyl diazomethane.

4d; $\mathrm{R}_{\mathrm{f}}=0.16\left(\right.$ EtOAc/Petrol 1:3); m.p. $177-179{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{21}+235.6\left(\mathrm{c}=3.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-}$ ${ }^{1}$ (film) $1743(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6}\right) 0.76\left(3 \mathrm{H}, \mathrm{m}, J 7.07, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.11(1 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(6) H), 3.23-3.31(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) H \mathrm{H}, \mathrm{C}(5) H), 3.71(1 \mathrm{H}, \mathrm{dd}, J 6.19,4.93, \mathrm{C}(4) \mathrm{H} H), 3.85(2 \mathrm{H}, \mathrm{q}$, $\left.J 7.07, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 6.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.59-6.64(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.89-7.05(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, 7.23-7.27 (2H, m, ArH), $7.41(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{ArH})$; $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.7\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 33.0$ $(C(6)), 46.3$ and $50.2\left(\left(C(7)\right.\right.$ and $(C(9)), 56.2(C(5)), 61.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 68.4(C(4)), 87.0$ ( $C(2)$ ), 125.3, 127.3, 127.6, 127.9, 128.0, 128.2, 128.5, 128.9, 129.1 (all ArCH$), 137.4,138.1$, $138.7(\mathrm{ArC}), 164.7$ and 171.8 (both $C=\mathrm{O}) ; \mathrm{m} / \mathrm{z}(\mathrm{ESI}+) 901\left([2 \mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right), 499$ $\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 52 \%\right) ;$ HRMS $462.1667\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{25} \mathrm{NNaO}_{4}\right.$ requires 462.1676); 6d; $\mathrm{R}_{\mathrm{f}}=0.35(\mathrm{EtOAc} /$ Petrol, $1: 3) ;[\alpha]_{\mathrm{D}}{ }^{21}+195.3\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3020$, 1745 and $1718(\mathrm{C}=\mathrm{O}), 754,702,669 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6} ; \mathrm{Me}_{4} \mathrm{Si}\right) 1.00(3 \mathrm{H}, \mathrm{t}, J 7.1$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.62(1 \mathrm{H}, \mathrm{t}, J 8.5, \mathrm{C}(4) H \mathrm{H}), 2.88(1 \mathrm{H}, \mathrm{dd}, J 6.8,8.5, \mathrm{C}(4) \mathrm{H} H), 4.01(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.34(1 \mathrm{H}, \mathrm{dd}, J 6.8,8.5, \mathrm{C}(5) H), 6.15\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CHPh}_{2}\right), 6.42(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.90-$ $6.94(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.01-7.29(11 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.74(2 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6}\right)$ $13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 50.4\left(\mathrm{CHPh}_{2}\right), 61.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.1(C(5)), 68.4(C(4)), 87.6(C(2))$,
126.5, 127.3, 127.9, 128.2 (all ArCH), 128.5 ( $($ (7)), 128.6, 128.6, 128.7, 129.0, 129.6 (all $\operatorname{ArCH}), 139.7,139.9,140.4$ (all ArC), 162.2, 168.8, 172.3 ( $C(6)$ and $2 \times(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+)$ $498\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right)$; HRMS $440.1856\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{28} \mathrm{H}_{26} \mathrm{NO}_{4}\right.$ requires 440.1856).
( $2 R, 5 S, 6 S, 7 R$ )-1-Aza-9,9-di(p-methoxyphenyl)-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyltricyclo[4.1.0 ${ }^{1,5} .0^{6,7}$ nonane 4 e and (2R,5S)-1-Aza-6-( $p, p^{\prime}$ 'dimethoxy)benzhydryl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-bicyclo-[3.3.0]oct-6-ene 6e

A solution of diazo $\mathbf{2 d}(247 \mathrm{mg}, 0.97 \mathrm{mmol})$ and $\mathbf{1 b}(143 \mathrm{mg}, 0.52 \mathrm{mmol})$ in acetone $(5 \mathrm{~mL})$ was degassed and stirred in an atmosphere of $\mathrm{N}_{2}$ for 16 h at r.t. Then, EtOH ( 3 mL ) was added and the mixture was heated at reflux for 4 h until the solution had turned yellow. Concentration in vacuo resulted in a yellow solid containing cyclopropane $\mathbf{4 e}$ and olefin $\mathbf{6 e}$. Purification by column chromatography over silica gel eluting with EtOAc/Petrol (1:4 $\boldsymbol{\rightarrow}$ 1:1) yielded cyclopropane $\mathbf{4 e}(168 \mathrm{mg}, 64 \%)$ as a crystalline white solid and olefin $\mathbf{6 e}(41 \mathrm{mg}, 16$ $\%)$ as a yellow oil.

4e; $\mathrm{R}_{\mathrm{f}}=0.06(\mathrm{EtOAc} /$ Petrol, $1: 4)$; m.p. $149{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}-69.7\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) 2957, 1738 (C=O), 1435, 1368, 1252, 1169, 1072, 1024, 880, 823, 781; $\delta_{\mathrm{H}}(500.3 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.44(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.60-3.66(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) \mathrm{HH}$, $\mathrm{C}(5) H), 3.74\left(3 \mathrm{H}, \mathrm{s}, \mathrm{C}^{\prime} H_{3} \mathrm{O}\right), 3.80\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}\right), 3.94-4.05\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.19(1 \mathrm{H}$, dd, $J 11.3,12.9, \mathrm{C}(4) \mathrm{H} H), 6.22(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.49(2 \mathrm{H}, \mathrm{d}, J 7.7, o-\mathrm{Ph} H), 6.76-6.79(2 \mathrm{H}, \mathrm{m}$, $\operatorname{ArH}\left(o-\right.$ to $\left.\left.\mathrm{OC}^{\prime} \mathrm{H}_{3}\right)\right), 6.89\left(2 \mathrm{H}, \mathrm{d}, J 8.9, \mathrm{Ar} H\left(o-\right.\right.$ to $\left.\left.\mathrm{OCH}_{3}\right)\right), 7.05-7.09(2 \mathrm{H}, \mathrm{m}, m-\mathrm{Ph} H), 7.14-$ $7.19(1 \mathrm{H}, \mathrm{m}, p-\mathrm{Ph} H), 7.26-7.29\left(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\left(m-\right.\right.$ to $\left.\left.\mathrm{OC}^{\prime} \mathrm{H}_{3}\right)\right), 7.39(2 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{ArH}$ ( $m-$ to $\left.\left.\mathrm{OCH}_{3}\right)\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 33.1(C(6)), 47.0$ and $49.1(C(7)$ and $\left.C\left(9^{\prime}\right)\right)$, 55.2 and $55.3\left(\mathrm{OC}^{\prime} \mathrm{H}_{3}\right.$ and $\left.\mathrm{OCH}_{3}\right)$, $56.4(\mathrm{C}(5)), 61.6\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 68.6(\mathrm{C}(4)), 87.0$ $(C(2)), 113.8\left(\mathrm{ArCH}\left(o-\right.\right.$ to $\left.\left.\mathrm{OC}^{\prime} \mathrm{H}_{3}\right)\right), 114.5\left(\mathrm{ArCH}\left(o-\right.\right.$ to $\left.\left.\mathrm{OCH}_{3}\right)\right), 125.5(o-\mathrm{PhCH}), 128.0(\mathrm{~m}-$ $\mathrm{PhCH}), 128.1(p-\mathrm{PhCH}), 129.3\left(\mathrm{ArCH}\left(m-\right.\right.$ to $\left.\left.\mathrm{OC}^{\prime} \mathrm{H}_{3}\right)\right), 129.9\left(\mathrm{ArC}\left(p-\right.\right.$ to $\left.\left.\mathrm{OC}^{\prime} \mathrm{H}_{3}\right)\right), 130.0$
$\left(\operatorname{ArCH}\left(m-\right.\right.$ to $\left.\left.\mathrm{OCH}_{3}\right)\right), 131.1\left(\mathrm{ArC}\left(p-\right.\right.$ to $\left.\left.\mathrm{OCH}_{3}\right)\right), 138.3(\mathrm{PhC}), 158.8$ and $159.1\left(\mathrm{Ar}^{\prime} \mathrm{COCH}_{3}\right.$ and $\mathrm{ArCOCH}_{3}$ ), 164.9, 171.8 (both $C=\mathrm{O}$ ); $m / z(\mathrm{ESI}+) 500\left([\mathrm{M}+\mathrm{H}]^{+}, 87 \%\right)$; HRMS 522.1883 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{30} \mathrm{H}_{29} \mathrm{NNaO}_{6}\right.$ requires 522.1887);

6e; $\mathrm{R}_{\mathrm{f}}=0.13(\mathrm{EtOAc} /$ Petrol, $1: 4) ;[\alpha]_{\mathrm{D}}{ }^{18}+31.7\left(\mathrm{c}=2.2, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 2936,2838$, 2251, 1742 (C=O), 1718 ( $\mathrm{C}=\mathrm{O}$ ), 1609, 1583, 1511, 1462, 1420, 1371, 1341, 1303, 1252, $1178,1113,1071,952,836,812,734,700 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.23(3 \mathrm{H}, \mathrm{t}, J 7.1$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.92(1 \mathrm{H}, \mathrm{t}, J 8.5, \mathrm{C}(4) H \mathrm{H}), 3.08(1 \mathrm{H}, \mathrm{dd}, J 6.8,8.5, \mathrm{C}(4) \mathrm{H} H), 3.79(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}^{\prime} H_{3} \mathrm{O}\right), 3.80\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}\right), 4.20\left(2 \mathrm{H}, \mathrm{q}, J 7.1 \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.51(1 \mathrm{H}, \mathrm{dd}, J 6.8,8.5, \mathrm{C}(5) H)$, $6.02\left(1 \mathrm{H}, \mathrm{s}, \mathrm{C} H\left(\mathrm{Ar}_{2}\right), 6.14(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.83-6.88\left(4 \mathrm{H}, \mathrm{m}, o-\right.\right.$ to $\mathrm{OC}^{\prime} \mathrm{H}_{3}$ and $\left.\mathrm{OCH}_{3}\right), 7.02-$ $7.08\left(4 \mathrm{H}, \mathrm{m}, m-\right.$ to $\mathrm{OC}^{\prime} \mathrm{H}_{3}$ and $\left.\mathrm{OCH}_{3}\right), 7.31-7.39(3 \mathrm{H}, \mathrm{m}, m-\mathrm{Ph} H, p-\mathrm{Ph} H), 7.60-7.64(2 \mathrm{H}, \mathrm{m}$, $o-\mathrm{Ph} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 48.6\left(\mathrm{CH}(\mathrm{Ar})_{2}\right), 55.2$ and $55.3\left(\mathrm{OC}^{\prime} \mathrm{H}_{3}\right.$ and $\left.\mathrm{OCH}_{3}\right), 61.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 64.8(C(5)), 68.5(C(4)), 87.1(C(2)), 114.1$ and $114.3(\mathrm{ArCH}$ (o- to $\mathrm{OC}^{\prime} \mathrm{H}_{3}$ and $\left.\mathrm{OCH}_{3}\right)$ ), $126.0(o-\mathrm{PhCH}), 128.4(m-\mathrm{PhCH}), 128.6(p-\mathrm{PhCH}), 128.8(C(7)), 129.3$, 129.3 (both $\mathrm{ArCH}\left(m-\right.$ to $\mathrm{OC}^{\prime} \mathrm{H}_{3}$ and $\left.\mathrm{OCH}_{3}\right)$ ), 131.7 and $132.2\left(\mathrm{ArC}\left(p-\right.\right.$ to $\mathrm{OC}^{\prime} \mathrm{H}_{3}$ and $\left.\left.\mathrm{OCH}_{3}\right)\right), 138.5(\mathrm{PhC}), 158.6$ and $159.2\left(\mathrm{Ar}^{\prime} \mathrm{COCH}_{3}\right.$ and $\left.\mathrm{ArCOCH}_{3}\right), 161.8,172.5,171.1(\mathrm{C}(6)$ and $2 \times(C=O)) ; m / z(E S I+) 499\left([\mathrm{M}-\mathrm{H}]^{-}, 100 \%\right) ;$ HRMS $522.1886\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{30} \mathrm{H}_{29} \mathrm{NNaO}_{6}\right.$ requires 522.1887).

## (2R,5S,6S,7R)-Spiro[1-aza-7-ethoxycarbonyl-8-oxo-3-oxa-2-phenyl-tricyclo

## [4.3.0 $0^{1,5} \cdot 0^{6,7}$ |nonane-9,9'-fluorene] 8a

A mixture of $\mathbf{3 a}(1.35 \mathrm{~g}, 7.00 \mathrm{mmol})$ and freshly prepared $\mathbf{1 b}(0.91 \mathrm{~g}, 3.34 \mathrm{mmol})$ in acetone $(50 \mathrm{~mL})$ was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 19 h . The mixture was concentrated in vacuo to give a red solid which was purified by column chromatography over silica eluting with $\mathrm{DCM} /$ Petrol $(1: 1)$ to remove the excess of diazo 3a and a close running impurity, then $\mathrm{EtOAc} /$ Petrol $(1: 4)$ to yield $\mathbf{8 a}$ as a white solid $(1.39 \mathrm{~g}, 95 \%) ; \mathrm{R}_{\mathrm{f}}=0.20$
(EtOAc/Petrol, $1: 3$ ); m.p. 193-195 ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}+145.5\left(\mathrm{c}=2.4, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 2988$, $1713(\mathrm{C}=\mathrm{O}), 1451,1372,1275,1159,1133,1024,948,749,701 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.03$ (3H, t, J 7.1, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.44(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.93(1 \mathrm{H}, \mathrm{dd} J 8.0,9.5, \mathrm{C}(4) H \mathrm{H}), 4.10-4.16$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.19(1 \mathrm{H}, \mathrm{dd}, J 6.1,9.5, \mathrm{C}(5) H), 4.49(1 \mathrm{H}, \mathrm{dd}, J 6.1,8.0, \mathrm{C}(4) \mathrm{H} H), 6.52$ $(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.94-6.99(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.01(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H), 7.23-7.28(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$, 7.31-7.52 (6H, m, ArH), 7.57-7.61 (2H, m, ArH), 7.77 (1H, d, J 7.6, ArH) 7.81 (1H, d, J 7.6, $\mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.7(C(6)), 43.3$ and $49.4\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $56.4(C(5)), 62.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.7(C(4)), 88.8(C(2)), 119.8,120.3,120.4,123.8,125.9$, 127.2, 127.4, 127.9, 128.1, 128.5, 128.8 (all ArCH), 138.4, 138.9, 139.2, 142.0, 142.4 (all $\operatorname{ArC}), 164.5$ and 170.8 (both $(C=O)$ ); $m / z(\mathrm{ESI}+) 496\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right), 460$ $\left([\mathrm{M}+\mathrm{Na}]^{+}, 12 \%\right), 438\left([\mathrm{M}+\mathrm{H}]^{+}, 4 \%\right), 898\left([2 \mathrm{M}+\mathrm{Na}]^{+}, 5 \%\right)$; HRMS $460.1529\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $\mathrm{C}_{28} \mathrm{H}_{23} \mathrm{NNaO}_{4}$ requires 460.1525 ); Found C, 76.92 ; H, 5.44; N, $3.28 \% ; \mathrm{C}_{28} \mathrm{H}_{23} \mathrm{NO}_{4}$ requires C, 76.87; H, 5.30; N, 3.20 \%.
( $2 R, 5 S, 6 S, 7 R, 9 S$ )-4'-Hydroxymethylspiro[1-aza-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyltricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right]$ nonane-9,9'-fluorene] $\quad 8 \mathrm{~b} \quad$ and $\quad(2 R, 5 S, 6 S, 7 R, 9 R)-5{ }^{\prime}-$

## Hydroxymethylspiro[1-aza-7-ethoxycarbonyl-8-oxo-3-oxa-2-phenyl-tricyclo

[4.3.0 ${ }^{1,5} .0^{6,7}$ ]nonane-9,9'-fluorene] $8{ }^{\text {b }}$ '
A mixture of diazo $\mathbf{3 b}(200 \mathrm{mg}, 0.89 \mathrm{mmol})$ and $\mathbf{1 b}(200 \mathrm{mg}, 0.73 \mathrm{mmol})$ in acetone $(10 \mathrm{~mL})$ was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 2 d . After concentration in vacuo the crude mixture was purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:3) to give $\mathbf{8 b}$ (122 g, 32\%) and $\mathbf{8 b}{ }^{\prime}(122 \mathrm{~g}, 32 \%)$ as clear gums.

8b; $\mathrm{R}_{\mathrm{f}}=0.25\left(\mathrm{EtOAc} /\right.$ Petrol, 2:3); $[\alpha]_{\mathrm{D}}{ }^{19}+131.0\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3487 \mathrm{br}$ (OH), 2983, $1714(\mathrm{C}=\mathrm{O}), 1440,1374,1275,1135,1023,733,700 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $1.04\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 1.98\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.45(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.93(1 \mathrm{H}, \mathrm{dd}, J$
8.0, 9.4, C(4)HH), 4.06-4.17 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$ ), $4.20(1 \mathrm{H}, \mathrm{dd}, J 6.0,9.4, \mathrm{C}(5) \mathrm{H}), 4.50(1 \mathrm{H}$, dd, J 6.0, 8.0, C(4)HH), $5.10\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{OH}\right), 6.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.94-7.01(2 \mathrm{H}, \mathrm{m}$, $\left.\operatorname{ArC}\left(1^{\prime}\right) H, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.23-7.28\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.35-7.49\left(5 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(3^{\prime}\right) H, \operatorname{ArC}\left(6^{\prime}\right) H\right.$, $m-\mathrm{Ph}, p-\mathrm{Ph}), 7.55-7.60\left(3 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(8^{\prime}\right) H, o-\mathrm{Ph}\right), 7.98\left(1 \mathrm{H}, \mathrm{d}, J 7.8, \operatorname{ArC}\left(5^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(100.6$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.2(C(6)), 43.0,49.9\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $56.4(C(5)), 62.2$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 63.7\left(\mathrm{CH}_{2} \mathrm{OH}\right), 69.8(C(4)), 88.9(\mathrm{C}(2))$, $119.7\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right)$, $123.4\left(\mathrm{ArC}\left(8^{\prime}\right) \mathrm{H}\right)$, 124.3 ( $\left.\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 125.9(o-\mathrm{PhCH}), 127.0\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 127.1\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 127.7,128.4,128.6$ and $128.8\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}, \operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}, o-\mathrm{PhCH}\right.$ and $\left.m-\mathrm{PhCH}\right), 135.4,136.9,138.3,139.4,142.1$, 143.1 (all ArC ), 164.6, 170.8 (both $(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+) 468\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS $490.1626\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{29} \mathrm{H}_{25} \mathrm{NNaO}_{5}\right.$ requires 490.1625);
$\mathbf{8 b}{ }^{\prime} ; \mathrm{R}_{\mathrm{f}}=0.28\left(\mathrm{EtOAc} /\right.$ Petrol, 2:3); $[\alpha]_{\mathrm{D}}{ }^{22}+133\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) 3484 (br, $\mathrm{OH}), 2983,1712(\mathrm{C}=\mathrm{O}), 1374,1275,1135,1020,742,700 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.02(3 \mathrm{H}$, t, $\left.J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.10\left(1 \mathrm{H}, \mathrm{t}, J 5.5, \mathrm{CH}_{2} \mathrm{OH}\right), 3.43(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.91(1 \mathrm{H}, \mathrm{dd}, J 8.0,9.4$, $\mathrm{C}(4) \mathrm{HH}), 4.06-4.17\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.17(1 \mathrm{H}, \mathrm{dd}, J 6.1,9.4, \mathrm{C}(5) H), 4.47(1 \mathrm{H}, \mathrm{dd}, J 6.1$, 8.0, C(4)HH), 5.01-5.11 ( $\left.2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 6.48(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.95(1 \mathrm{H}, \mathrm{t}, J 7.8, \mathrm{ArH}), 7.00$ $(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H), 7.26(1 \mathrm{H}, \mathrm{td}, J 0.6,7.6, \mathrm{Ar} H), 7.35-7.47(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.51(1 \mathrm{H}, \mathrm{d}, J 7.5$, $\mathrm{Ar} H), 7.54-7.59(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.95(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.2(C(6)), 43.1$ and 49.9) $\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $56.3(C(5)), 62.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 63.7$ $\left(\mathrm{CH}_{2} \mathrm{OH}\right), 69.7(C(4)), 88.9(C(2)), 120.0,122.8,123.9($ all ArCH$), 125.9(\mathrm{PhCH}), 126.9$, 127.1, 128.1, 128.2 (all ArCH$), 128.6(\mathrm{PhCH}), 128.8(\mathrm{ArCH}), 135.9,138.3,138.7,139.9$, 140.2, 142.5, (all ArC), 164.5, 170.8 (both ( $C=\mathrm{O}$ )); $m / z\left(\mathrm{ESI}+\right.$ ) 468 ( $[\mathrm{M}+\mathrm{H}]^{+}, 100 \%$ ); HRMS $490.1625\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{29} \mathrm{H}_{25} \mathrm{NNaO}_{5}\right.$ requires 490.1625).
(2R,5S,6S,7R,9S)-2'-Bromospiro[1-aza-7-ethoxycarbonyl-8-oxo-3-oxa-2-phenyl-tricyclo $\left[4.3 .0^{1,5} \cdot 0^{6,7}\right]$ nonane-9,9'-fluorene] 8 c and (2R,5S,6S,7R,9R)-7'-Bromospiro[1-aza-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right]$ nonane-9,9'-fluorene] $8 c^{\prime}$,

Diazo 3c ( $1.26 \mathrm{~g}, 4.61 \mathrm{mmol})$ and $\mathbf{1 b}(0.66 \mathrm{~g}, 2.42 \mathrm{mmol})$ were stirred in DCM $(20 \mathrm{~mL})$ at r.t. for 2 d in an atmosphere of $\mathrm{N}_{2}$. After concentration the crude mixture was purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:3) to afford 8c ( $0.56 \mathrm{~g}, 45 \%)$ and $\mathbf{8 c}{ }^{\prime}(0.30 \mathrm{~g}, 24 \%)$ as white solids;

8c; $\mathrm{R}_{\mathrm{f}}=0.52(\mathrm{EtOAc} /$ Petrol, $1: 3)$; m.p. $148-149{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+156.9\left(\mathrm{c}=0.84, \mathrm{CHCl}_{3}\right)$; $\mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2981, $1714(\mathrm{C}=\mathrm{O}), 1447,1373,1272,1133,1023,827,773,728,701$; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.10\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.44(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.91(1 \mathrm{H}, \mathrm{dd}, J$ 8.1, 9.4, C(4)HH), 4.09-4.25 (3H, m, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}, \mathrm{C}(5) H\right), 4.49(1 \mathrm{H}, \mathrm{dd}, J 6.1,7.9, \mathrm{C}(4) \mathrm{H} H)$, $6.50(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.97\left(1 \mathrm{H}, \mathrm{t}, J 7.7, \operatorname{ArC}\left(6^{\prime}\right) H\right), 7.15\left(1 \mathrm{H}, \mathrm{d}, J 1.5, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.34(1 \mathrm{H}, \mathrm{t}$, $\left.J 7.3, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.38-7.52(5 \mathrm{H}, \mathrm{m}, \operatorname{ArH}), 7.55-7.62(3 \mathrm{H}, \mathrm{m}, \operatorname{ArH}), 7.75(1 \mathrm{H}, \mathrm{d}, J 7.5$, $\left.\operatorname{ArC}\left(8{ }^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3} ; \mathrm{Me}_{4} \mathrm{Si}\right) 14.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.8(C(6)), 42.9$ and $49.4(C(7)$ and $\left.C\left(9^{\prime}\right)\right)$, $56.3(C(5)), 62.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.6(C(4)), 88.9(C(2)), 120.5(\mathrm{ArCH}), 120.9$ ( ArCBr ), 121.1, 123.8, 123.8 (all ArCH), $125.8(\mathrm{PhCH}), 127.8,128.3$ (both ArCH ), 128.6 $(\mathrm{Ph} C \mathrm{H}), 128.8,131.0(p-\mathrm{Ph} C \mathrm{H}), 138.2(\mathrm{Ph} C), 138.2,138.7,141.3,144.0$ (all ArC), 164.1 $(C=\mathrm{O}), 170.3(C=\mathrm{O})$; HRMS $538.0651\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{22} \mathrm{NNaO}_{4}\right.$ requires 538.0624);
$\mathbf{8 c} ; \mathrm{R}_{\mathrm{f}}=0.35($ EtOAc/Petrol, $1: 3) ;$ m.p. $194{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+42.2\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) 2982, 1715 (C=O), 1443, 1371, 1287, 1163, 1131, 1024, 829, 734, 700; $\delta_{\mathrm{H}}(400.2 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.42(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.90(1 \mathrm{H}, \mathrm{t}, J 8.7, \mathrm{C}(4) H \mathrm{H}), 4.06-$ $4.18\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}, \mathrm{C}(5) H\right), 4.44(1 \mathrm{H}, \mathrm{dd}, J 6.2,7.9, \mathrm{C}(4) \mathrm{H} H), 6.51(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.97$ $\left(1 \mathrm{H}, \mathrm{t}, J 7.7, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.26\left(1 \mathrm{H}, \mathrm{t}, J 7.6, \operatorname{ArC}\left(2^{`}\right) H\right), 7.35-7.51(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.54-7.61$ $(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.69(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{ArCH}), 7.74\left(1 \mathrm{H}, \mathrm{d}, J 1.6, \operatorname{ArC}\left(8^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 13.8\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.0(C(6)), 43.0$ and $49.4\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $56.3(C(5)), 62.2$
$\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.5(\mathrm{C}(4)), 88.2(\mathrm{C}(2)), 119.8,120.3(\mathrm{ArCH}), 121.3(\mathrm{ArCBr}), 121.4(\mathrm{ArCH})$, $125.7(\mathrm{PhCH}), 126.7,127.5,128.1,128.7$ (all ArCH), $128.9(\mathrm{PhCH}), 131.2(p-\mathrm{PhCH}), 138.0$ ( $\mathrm{Ph} C$ ), 138.1, 140.8, 141.2, 141.6 (all ArC), $164.1(C=\mathrm{O}), 171.0(C=\mathrm{O})$; HRMS 538.0636 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{22} \mathrm{NNaO}_{4}\right.$ requires 538.0624).

## (2R,5S,6S,7R)-2',7'-Dibromospiro[1-aza-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-

 tricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right.$ nonane-9,9'-fluorene] 8dA mixture of diazo 3d (302 mg, 0.86 mmol$)$ and $\mathbf{1 b}(180 \mathrm{mg}, 0.66 \mathrm{mmol})$ in acetone $(10 \mathrm{~mL})$ was purged with $\mathrm{N}_{2}$ before being heated to reflux and left for 24 h . After concentration in vacuo the crude material was purified by column chromatography over silica gel eluting with EtOAc/Petrol ( $1: 3$ ) to afford product $\mathbf{8 d}(251 \mathrm{mg}, 64 \%)$ as a white solid; $\mathrm{R}_{\mathrm{f}}=0.29$ (EtOAc/Petrol, $1: 3$ ); m.p. $185-187{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}+69.1\left(\mathrm{c}=0.75, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 1716$ $(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.10\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.44(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.91(1 \mathrm{H}$, dd, $J$ 8.1, 9.2, C(4)HH), 4.12-4.26 (3H, m, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}, \mathrm{C}(5) H\right), 4.48(1 \mathrm{H}, \mathrm{dd}, J 6.2,8.0$, $\mathrm{C}(4) \mathrm{H} H), 6.51(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.13\left(1 \mathrm{H}, \mathrm{d}, J 1.5, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.43-7.60(9 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.73$ $\left(1 \mathrm{H}, \mathrm{d}, J 1.6, \operatorname{ArC}\left(8^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.2(C(6)), 42.6$ and 49.5 $\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right), 56.3(C(5)), 62.6\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.5(C(4)), 89.4(C(2)), 121.2(\mathrm{ArCH})$, $121.3(\mathrm{ArC}), 121.5(\mathrm{ArCH}), 121.9(\mathrm{ArC}), 123.8(\mathrm{ArCH}), 125.8(\mathrm{PhCH}), 126.9(\mathrm{ArCH}), 128.9$ ( ArCH ), $129.0(\mathrm{PhCH}), 131.2,131.5$ (both ArCH$), 137.2,137.9,140.2,140.6,143.7(\mathrm{ArC})$, 163.8, 170.6 (both $(C=O)$ ); $m / z\left(\right.$ ESI + ) 654 ( $\left.\left[\mathrm{M}+\mathrm{MeCN}^{2}+\mathrm{NH}_{4}\right]^{+}, 48 \%\right)$; HRMS 617.9709 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{21} \mathrm{Br}_{2} \mathrm{NNaO}_{4}\right.$ requires 617.9711).
( $2 R, 5 S, 6 S, 7 R, 9 S$ )-1,4'-Diazaspiro[7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo $\left[4.3 .0^{1,5} \cdot 0^{6,7}\right]$ nonane-9,9'-fluorene] 8 e and $(2 R, 5 S, 6 S, 7 R, 9 R)-1,5$ '-Diazaspiro[7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right]$ nonane-9,9'-fluorene] 8e'

Diazo 3e ( $1.56 \mathrm{~g}, 8.07 \mathrm{mmol})$ and $\mathbf{1 b}(1.34 \mathrm{~g}, 4.90 \mathrm{mmol})$ were stirred in DCM $(25 \mathrm{~mL})$ at r.t. for 2 d in an atmosphere of $\mathrm{N}_{2}$. After concentration the crude mixture was purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:3 $\boldsymbol{\rightarrow} 1: 1)$ to afford $\mathbf{8 e}(0.83 \mathrm{~g}$, $39 \%)$ as a yellow foam and $\mathbf{8 e} \mathbf{e}^{\prime}(1.90 \mathrm{~g}, 58 \%)$ as a white solid.

8e; $\mathrm{R}_{\mathrm{f}}=0.42\left(\mathrm{EtOAc} /\right.$ Petrol, 1:1); m.p. $174-176{ }^{\circ} \mathrm{C}(\mathrm{EtOAc} /$ Petrol $) ;[\alpha]_{\mathrm{D}}{ }^{20}+96.4(\mathrm{c}=1$, $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2983, $1715(\mathrm{C}=\mathrm{O}), 1449,1420,1373,1258,1178,1023,738,700$; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.06\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.92(1 \mathrm{H}, \mathrm{dd}, J$ 8.0, 9.5, C(4)HH), 4.08-4.18 (2H, m, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.22(1 \mathrm{H}, \mathrm{dd}, J 6.1,9.5, \mathrm{C}(5) H), 4.52(1 \mathrm{H}$, dd, $J 6.1,8.0, \mathrm{C}(4) \mathrm{H} H), 6.51(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.07\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(7^{`}\right) H\right), 7.14(1 \mathrm{H}, \mathrm{dd}, J 4.9,7.9$, $\left.\operatorname{ArC}\left(2^{\prime}\right) H\right), 7.35\left(1 \mathrm{H}, \mathrm{dd}, J 1.4,7.9, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.39-7.49\left(4 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(6^{\prime}\right) H, m-\mathrm{Ph} H, p-\mathrm{Ph} H\right)$, $7.53\left(1 \mathrm{H}, \mathrm{d}, J 8.0, \operatorname{ArC}\left(8^{\prime}\right) H\right), 7.56-7.59(2 \mathrm{H}, \mathrm{m}, o-\mathrm{Ph} H), 8.14\left(1 \mathrm{H}, \mathrm{d}, J 7.2 \operatorname{ArC}\left(5^{\prime}\right) H\right), 8.60$ $\left(1 \mathrm{H}, \mathrm{dd}, J 1.4,4.9 \mathrm{ArC}\left(3^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.2(C(6)), 41.7$ and $48.9\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right), 56.2(C(5)), 62.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.7(C(4)), 88.9(C(2)), 121.2$ $\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 120.2\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 121.5\left(\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 125.8(o-\mathrm{PhCH}), 128.1\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 128.6$, 128.7 and $128.9\left(p-\mathrm{PhCH}, \operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right.$ and $\left.m-\mathrm{PhCH}\right), 129.5\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 135.9,138.2,139.5$, 141.7 (all $\operatorname{ArC}$ ), $149.3\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 158.2(\operatorname{ArC}), 164.3,170.1$ (both $\left.(C=\mathrm{O})\right) ; m / z(\mathrm{ESI}+) 439$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 877\left([\mathrm{M}+\mathrm{H}]^{+}, 91 \%\right), 461\left([\mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right) ;$ HRMS $461.1472[\mathrm{M}+\mathrm{Na}]^{+}$, $\mathrm{C}_{27} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{NaO}_{4}$ requires 461.1476);
$\mathbf{8} \mathbf{e}^{\prime} \mathrm{R}_{\mathrm{f}}=0.67$ (EtOAc/Petrol, 1:1); m.p. $85{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+111\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2980, $1716(\mathrm{C}=\mathrm{O}), 1280,737 ; \delta_{\mathrm{H}}\left(700.1 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.52$ $(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.92(1 \mathrm{H}, \mathrm{t}, J 8.7, \mathrm{C}(4) H \mathrm{H}), 4.09-4.18\left(3 \mathrm{H}, \mathrm{m}, \mathrm{C}(5) H, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.53(1 \mathrm{H}, \mathrm{t}$, $J 6.9, \mathrm{C}(4) \mathrm{H} H), 6.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.83\left(1 \mathrm{H}, \mathrm{dd}, J 4.9,7.9, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.07(1 \mathrm{H}, \mathrm{d}, J 7.8$, $\left.\operatorname{ArC}\left(1^{\prime}\right) H\right), 7.37\left(1 \mathrm{H}, \mathrm{t}, J 7.5, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.40-7.50\left(4 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(3^{\prime}\right) H, m-\mathrm{Ph} H, p-\mathrm{Ph} H\right), 7.56$ $(2 \mathrm{H}, \mathrm{d}, J 7.4, o-\mathrm{Ph} H), 7.67\left(1 \mathrm{H}, \mathrm{d}, J 7.9, \operatorname{ArC}\left(8^{\prime}\right) H\right), 8.09\left(1 \mathrm{H}, \mathrm{d}, J 7.5, \operatorname{ArC}\left(4^{\prime}\right) H\right), 8.53(1 \mathrm{H}$, d, $\left.J 4.7, \operatorname{ArC}\left(6^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.4(C(6)), 41.6$ and 48.4
$\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right), 56.6(C(5)), 62.4\left(\mathrm{CH}_{3} C H_{2} \mathrm{O}\right), 69.7(C(4)), 88.9(C(2)), 120.3\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right)$, $121.1\left(\operatorname{ArC}\left(4^{\prime}\right) \mathrm{H}\right), 121.3\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 125.8(o-\mathrm{PhCH}), 128.6(m-\mathrm{PhCH}), 128.5,129.0$ and $129.2\left(\mathrm{~m}-\mathrm{PhCH}, \operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 131.2\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 133.4,138.2,138.8,142.5$ (all $\operatorname{ArC}), 149.0\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 160.8(\mathrm{ArC}), 164.1,170.6($ both $(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 877\left([2 \mathrm{M}+\mathrm{H}]^{+}\right.$, $100 \%) ; 439\left([\mathrm{M}+\mathrm{H}]^{+}, 58 \%\right)$; HRMS $439.1649\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{27} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}_{4}\right.$ requires 439.1652).
( $2 R, 5 S, 6 S, 7 R$ )-1,4, ${ }^{\prime}$ '-Triazaspiro[7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo [4.3.0 ${ }^{1,5} .0^{6,7}$ ]nonane-9,9'-fluorene] $8 f$

A mixture of diazo $\mathbf{3 f}(340 \mathrm{mg}, 1.75 \mathrm{mmol})$ and $\mathbf{1 b}(170 \mathrm{mg}, 0.63 \mathrm{mmol})$ in acetone $(10 \mathrm{~mL})$ was heated to $40{ }^{\circ} \mathrm{C}$ for 3 d (under an $\mathrm{N}_{2}$ atmosphere) before being concentrated in vacuo. Column chromatography over silica gel eluting with $\mathrm{Et}_{2} \mathrm{O} / \mathrm{DCM}(1: 4)$ yielded $\mathbf{8 f}$ as a grey solid (114 mg, 41\%); $\mathrm{R}_{\mathrm{f}}=0.17\left(\mathrm{Et}_{2} \mathrm{O} / \mathrm{DCM}, 1: 4\right)$; m.p. dec. $113{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}+77.9(\mathrm{c}=1.65$, $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2984, $1716(\mathrm{C}=\mathrm{O}), 1565,1402,1259,1175,1021,745,701$; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.04\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.55(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.91(1 \mathrm{H}, \mathrm{dd}, J$ 8.2, 9.1, C(4)HH), 4.05-4.21 (3H, m, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}, \mathrm{C}(5) H\right), 4.55(1 \mathrm{H}, \mathrm{dd}, J 6.1,8.2, \mathrm{C}(4) \mathrm{H} H)$, $6.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.92\left(1 \mathrm{H}, \mathrm{dd}, J 4.8,8.1 \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.22-7.27\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.39-$ $7.49\left(4 \mathrm{H}, \mathrm{m}, m-\mathrm{Ph} H, p-\mathrm{Ph} H, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.53-7.57(2 \mathrm{H}, \mathrm{m}, o-\mathrm{Ph} H), 7.70-7.74(1 \mathrm{H}, \mathrm{m}$, $\left.\operatorname{ArC}\left(8^{\prime}\right) H\right), 8.65\left(1 \mathrm{H}, \mathrm{d}, J 4.7, \operatorname{ArC}\left(6^{\prime}\right) H\right), 8.74\left(1 \mathrm{H}, \mathrm{d}, J 4.7, \operatorname{ArC}\left(3{ }^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 36.8(C(6)), 48.2(C(7)), 56.3(C(5)), 62.6\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.6(C(4))$, $77.2\left(C\left(9^{\prime}\right)\right), 88.9(C(2)), 122.7,122.8\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 125.7(o-\mathrm{PhCH}), 128.5$ $\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 128.5(m-\mathrm{PhCH}), 129.0(p-\mathrm{PhCH}), 131.4\left(\mathrm{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 134.3,136.6$ (both ArC$)$, $138.0(\mathrm{PhC}), 150.1,150.4\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 156.8,159.2$ (both $\operatorname{ArC}$ ), 163.8, 169.9 (both $(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+) 440\left([\mathrm{M}+\mathrm{H}]^{+}, 77 \%\right), 879\left([2 \mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS 462.1422 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{NaO}_{4}\right.$ requires 462.1424).

## Dimethyldiazomalonate $\mathbf{1 1 b}^{77}$

A mixture of dimethyl malonate ( $1.73 \mathrm{~mL}, 15.1 \mathrm{mmol}$ ), p-ABSA ( $3.71 \mathrm{~g}, 15.4 \mathrm{mmol}^{2}$ ), $\mathrm{Et}_{3} \mathrm{~N}$ ( $6.33 \mathrm{~mL}, 45.4 \mathrm{mmol}$ ) in $\mathrm{MeCN}(50 \mathrm{~mL}$ ) was stirred for 14 h . The mix was washed with $\mathrm{Et}_{2} \mathrm{O} /$ Petrol $(1: 1)$ and the solid precipitate allowed to settle before the liquid was decanted away. This process was repeated several times before the combined organic washes were concentrated in vacuo to leave a yellow oil which was purified by column chromatography over silica eluting with $\mathrm{Et}_{2} \mathrm{O} / \operatorname{Petrol}(1: 4)$ to give 11b as a clear yellow oil ( $2.02 \mathrm{~g}, 84 \%$ ); $v_{\max } / \mathrm{cm}^{-1}$ (neat) 2959, $2139\left(\mathrm{CN}_{2}\right), 1763(\mathrm{C}=\mathrm{O}), 1696,1439,1334,1276,1192,1100,761$; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3} ; \mathrm{Me}_{4} \mathrm{Si}\right) 3.75\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3} ; \mathrm{Me}_{4} \mathrm{Si}\right) 52.2\left(\mathrm{CH}_{3}\right)$, $161.1(C=O)$.

## 6',7’-Diaza-5a,8-diethyloxycarbonyl-2-oxa-3-phenyl-1,2,3,5a,6,8b-hexahydrocyclo penta[a]pyrrolizin-5(8aH)-one 12

A solution of ethyldiazoacetate $\mathbf{1 1 a}(0.32 \mathrm{~mL}, 3 \mathrm{mmol})$ and $\mathbf{1 b}(0.27 \mathrm{~g}, 1 \mathrm{mmol})$ in acetone $(15 \mathrm{~mL})$ was heated at reflux for 16 h . The mixture was concentrated and the resulting oil was purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:2) to give 12 as yellow-orange oil (74\%); $\mathrm{R}_{\mathrm{f}}=0.46\left(\right.$ EtOAc/Petrol, 2:1); $[\alpha]_{\mathrm{D}}{ }^{18}-52.2\left(\mathrm{c}=4.63, \mathrm{CHCl}_{3}\right)$; $v_{\max } / \mathrm{cm}^{-1}($ film $) 3331(\mathrm{NH}), 2984,1723(\mathrm{C}=\mathrm{O}), 1565,1380,1221,1133,748,701 ; \delta_{\mathrm{H}}(400.2$ $\left.\mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6}\right) 0.84$ and $0.97\left(2 \times 3 \mathrm{H}, \mathrm{t}, J 7.1,2 \times \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.55(1 \mathrm{H}, \mathrm{dd}, J 8.0,8.9$, $\mathrm{C}(4) H \mathrm{H}), 3.83-4.05\left(5 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}, \mathrm{C}(5) H\right), 4.06(1 \mathrm{H}, \mathrm{d}, J 2.3, \mathrm{C}(6) H), 4.11(1 \mathrm{H}, \mathrm{dd}$, $J 6.2,8.0, \mathrm{C}(4) \mathrm{H} H), 6.43(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.99-7.09(3 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H), 7.41-7.45(2 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H)$, $7.95(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.0$ and $14.2\left(2 \times \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 51.3(C(6), 61.8$, $61.8\left(C(5)\right.$ and $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right)$, $63.5\left(\mathrm{CH}_{3} C^{\prime} \mathrm{H}_{2} \mathrm{O}\right)$, $70.3(C(4)), 80.3(C(7)), 87.8(C(2), 125.8$, $128.5(o-\mathrm{PhCH}$ and $m-\mathrm{PhCH}), 128.9(p-\mathrm{PhCH}), 137.6$ and $140.7\left(\mathrm{EtO}_{2} \mathrm{CC}=\mathrm{N}\right.$ and $\left.\mathrm{Ph} C\right)$,
161.3, 167.5, 171.7 (3 x $(C=O)$ ); HRMS $410.1324\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{NaO}_{6}\right.$ requires 410.1323).

## General Procedure for TFA Mediated Deprotection of Cyclopropane Adducts To a

 solution of cyclopropane $(1 \mathrm{mmol})$ in $\mathrm{DCM}(10-20 \mathrm{~mL})$ at r.t. was added a solution of TFA $(0.77 \mathrm{~mL}, 10 \mathrm{mmol})$ in water $(0.77 \mathrm{~mL})$. The mixture was stirred for 5 h before being concentrated in vacuo. Toluene ( 8 mL ) was added followed by 1 drop of water and the mixture was re-concentrated in vacuo. The addition of toluene and water followed by reconcentration was repeated (x2) to give a brown viscous oil which was purified by column chromatography over silica gel yielding deprotected alcohol.
## General Procedure for Hydrolysis of Esters

To a solution of ester $(0.1 \mathrm{mmol})$ in $\mathrm{MeOH}(2.25 \mathrm{~mL})$ and water $(0.75 \mathrm{~mL})$ was added lithium hydroxide ( $21 \mathrm{mg}, 0.5 \mathrm{mmol}$ (per mol of ester)). After stirring at r.t. for 2 d , MeOH was removed in vacuo. EtOAc ( 5 mL ) and water ( 3 mL ) were added and the organic layer was separated and concentrated to give the unwanted organic extracts. EtOAc ( 5 mL ) was added and conc. HCl was added dropwise until pH 1 was achieved. The organic layer was separated and the aqueous was extracted with $\operatorname{EtOAc}(4 \times 5 \mathrm{~mL})$ before the combined organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give product acid which required no further purification.
(2S,3R,4R)-1-Aza-6,6-diphenyl -4-ethoxycarbonyl-2-hydroxymethyl-5-oxo -bicyclo [3.1.0]hexane 13a
Following general procedure, after column chromatography eluting with EtOAc/Petrol(60/80) (2:1), cyclopropane $\mathbf{4 d}(304 \mathrm{mg}, 0.69 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 a}$ as a white solid ( 110 mg , $45 \%) ; \mathrm{R}_{\mathrm{f}}=0.15(\mathrm{EtOAc} /$ Petrol, $2: 1) ;$ m.p. $109{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{21}-152.7\left(\mathrm{c}=6.6, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3400 (br, NH, OH), 2981, 1733 (C=O), 1495, 1448, 1311, 1112, 1043, 750, 708;
$\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6}\right) 0.79\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.38(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.51(1 \mathrm{H}, \mathrm{t}, J 4.6$, $\mathrm{C}(2) H), 3.81-3.93\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.92\left(1 \mathrm{H}\right.$, br s, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 6.71(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$, 6.98-7.03 (1H, m, $p-\mathrm{Ph} H), 7.05-7.14\left(3 \mathrm{H}, \mathrm{m}, m-\mathrm{Ph} H, p-\mathrm{Ph}{ }^{\prime} H\right), 7.20\left(2 \mathrm{H}, \mathrm{t}, J 7.8, m-\mathrm{Ph}{ }^{\prime} H\right)$, $7.44(2 \mathrm{H}, \mathrm{d}, J 7.3, o-\mathrm{Ph} H), 7.59\left(2 \mathrm{H}, \mathrm{d}, J 7.2, o-\mathrm{Ph}{ }^{\prime} H\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{D}_{6}\right) 13.8$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 33.0(C(3)), 44.9$ and $47.1(C(6)$ and $C(4)), 54.0(C(2)), 61.1$ and $65.4\left(\mathrm{CH}_{2} \mathrm{OH}\right.$ and $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 127.3,127.5,128.3,128.5,129.1,129.4($ all PhCH$), 138.1$ and $140.5(\mathrm{PhC})$ and $\left(\mathrm{Ph}^{\prime} \mathrm{C}\right), 166.0$ and $171.3(2 \times(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 410\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right)$; HRMS $374.1359\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NNaO}_{4}\right.$ requires 374.1363).

## (2S,3R,4R)-1-Aza-di(p-methoxyphenyl)-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-6,6-

## bicyclo[3.1.0]hexane 13b

Following the general procedure, after column chromatography eluting with EtOAc, cyclopropane $\mathbf{4 e}(113 \mathrm{mg}, 0.23 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 b}$ as a white solid ( $75 \mathrm{mg}, 80 \%$ ); $\mathrm{R}_{\mathrm{f}}$ $=0.20(\mathrm{EtOAc}) ;[\alpha]_{\mathrm{D}}{ }^{17}-147\left(\mathrm{c}=1.3, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3346,2957,2838,1732(\mathrm{C}=\mathrm{O})$, $1696(\mathrm{C}=\mathrm{O}), 1608,1580,1512,1418,1404,1295,1248,1178,1113,1033,830,735,687$; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.90\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.10(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.42(1 \mathrm{H}, \mathrm{t}, J 5.0$, $\mathrm{C}(2) \mathrm{H}), 3.60-3.80\left(8 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3}{ }_{3} \mathrm{O}, \mathrm{CH}_{3} \mathrm{O}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.82-3.96\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.98(1 \mathrm{H}$, s, NH), 6.72 (2H, d, $J 8.8, \operatorname{Ar}^{\prime} H(o-$ to OMe) ), 7.72 (2H, d, $J 8.8, \operatorname{ArH}$ (o- to OMe)), 7.21 (2H, d, $J 8.8, \operatorname{Ar}^{\prime} H(m-$ to OMe$)$ ), $7.32(2 \mathrm{H}, \mathrm{d}, J 8.8, \mathrm{Ar} H(m-$ to OMe$)) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $13.7\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 32.6(C(3)), 44.2$ and $46.1(C(4)$ and $C(6))$, $53.2(C(2))$, 55.2, $55.2\left(\mathrm{CH}_{3} \mathrm{O}\right.$ and $\left.C^{\prime} \mathrm{H}_{3} \mathrm{O}\right), 61.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.3\left(\mathrm{CH}_{2} \mathrm{OH}\right), 113.7$ and $114.3(\mathrm{ArC}$ ' $H$ (o- to OMe), -$\operatorname{ArCH}(o-$ to OMe$)), 129.4\left(\mathrm{Ar}^{\prime} \mathrm{CH}(m-\right.$ to OMe$)$ ), 129.5 ( $\mathrm{Ar}^{\prime} C(p-$ to OMe$)$ ), 129.6 ( $\mathrm{ArCH}\left(m-\right.$ to OMe )), 131.8 ( $\mathrm{ArC}(p-$ to OMe$)$ ), 158.6 and 158.7 ( $\mathrm{ArC}{ }^{\prime} \mathrm{OMe}$ and ArCOMe ), 165.9 and 171.0 (both $C=\mathrm{O}$ ); $m / z\left(\mathrm{ESI}+\right.$ ) $412\left([\mathrm{M}+\mathrm{H}]^{+}, 87 \%\right), 434\left([\mathrm{M}+\mathrm{Na}]^{+}, 73 \%\right), 845$ $\left([\mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right)$; HRMS $434.1580\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{23} \mathrm{H}_{25} \mathrm{NNaO}_{6}\right.$ requires 434.1574).
(2S,3S,4R)-1-Aza-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro[bicyclo[3.1.0] hexane-6,9'-fluorene] 13c

Following the general procedure, after column chromatography eluting with EtOAc/Petrol(60/80) (2: 1), cyclopropane 8a(177 mg, 0.40 mmol$)$ yielded alcohol $\mathbf{1 3 c}$ as a white solid ( $130 \mathrm{mg}, 92 \%$ ); $\mathrm{R}_{\mathrm{f}}=0.30$ (EtOAc); m.p. $176-177{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{21}+74.0(\mathrm{c}=3.1$, $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) 3318, 1697, 1450, 1262, 1199, 1109, 731; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.23(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.14(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.88-3.95(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 4.06(1 \mathrm{H}, \mathrm{t}, J 5.4, \mathrm{C}(2) \mathrm{H}), 4.07-4.23\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 6.56(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.02$ $(1 \mathrm{H}, \mathrm{d}, J 7.8, \operatorname{Ar} H), 7.23-7.27(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.38(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar} H), 7.43(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar} H)$, $7.51(1 \mathrm{H}, \mathrm{d}, J 7.9 \mathrm{Ar} H), 7.78(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 7.85(1 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{Ar} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 36.9(C(3)), 40.4$ and $46.7(C(4)$ and $C(6))$, $53.4(C(2)), 62.2$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.4\left(\mathrm{CH}_{2} \mathrm{OH}\right), 119.8,120.5,120.6,122.7,127.2,127.7$, 127.7, 128.1, (all ArCH), 138.8, 139.3, 142.2, 142.5, (all ArC ), 164.9, 169.0 (both ( $C=\mathrm{O}$ )); $m / z$ (ESI+) 408 $\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right)$; HRMS $372.1201\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{21} \mathrm{H}_{19} \mathrm{NNaO}_{4}\right.$ requires 372.1206).

## (2S,3S,4R,6R)-1-Aza-2,5'-dihydroxymethyl-4-ethoxycarbonyl-5-oxo-spiro[bicyclo

## [3.1.0]hexane-6,9'-fluorene] 13d

Following the general procedure, after column chromatography eluting with EtOAc, cyclopropane $\mathbf{8 b}{ }^{\prime}(251 \mathrm{mg}, 0.54 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 d}$ as a white solid ( $65 \mathrm{mg}, 32 \%$ ); $\mathrm{R}_{\mathrm{f}}=0.24$ (EtOAc); m.p. dec. $170{ }^{\circ} \mathrm{C}(\mathrm{EtOAc}) ;[\alpha]_{\mathrm{D}}{ }^{18}+54.7(\mathrm{c}=1.65, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3357 (br, OH), $1694(\mathrm{C}=\mathrm{O}), 1429 ; \delta_{\mathrm{H}}(400.2 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 1.01(3 \mathrm{H}, \mathrm{t}, J 7.1$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.34(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.72-3.83\left(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 3.95(1 \mathrm{H}, \mathrm{t}, J 5.7, \mathrm{C}(2) H)$, 3.98-4.12 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$ ), $5.03\left(2 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{2} \mathrm{OH}\right), 7.03(1 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{ArH}), 7.24-7.31$ $(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.40(1 \mathrm{H}, \mathrm{td}, J 1.0,7.6, \mathrm{Ar} H), 7.46(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar} H), 7.51(1 \mathrm{H}, \mathrm{d}, J 7.5$,
$\mathrm{Ar} H), 7.95(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}(100.6 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 14.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.9(C(3))$, 41.2 and $49.2\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 55.2\left((C(2)), 62.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 63.7\left(\mathrm{ArCH}_{2} \mathrm{OH}\right), 65.5\right.$ $\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 121.4,122.6,125.1,127.8,127.8,128.2,128.8(\mathrm{ArCH}), 138.1,140.2,141.1$, 141.1, $144.4(\mathrm{ArC}), 166.5$ and $171.3(C=\mathrm{O}) ; m / z(\mathrm{ESI}+) 380\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 402\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, 97\%); HRMS $402.1305\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{22} \mathrm{H}_{21} \mathrm{NNaO}_{5}\right.$ requires 402.1312).

## (2S,3S,4R,6S)-1-Aza-2'-bromo-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro

 [bicyclo[3.1.0]hexane-6,9'-fluorene] 13e
## By Deprotection of Dibromo 8d

Following the general procedure, after column chromatography eluting with EtOAc/Petrol (2 : 1), dibromocyclopropane $\mathbf{8 d}(268 \mathrm{mg}, 0.45 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 e}$ as a white solid (134 $\mathrm{mg}, 59$ \%).

## By Deprotection of Mono-Bromo 8c

Following the general procedure, after column chromatography eluting with EtOAc/Petrol (2 : 1), mono-bromocyclopropane $\mathbf{8 c}(250 \mathrm{mg}, 0.48 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 e}$ as a white solid $(149 \mathrm{mg}, 72 \%) ; \mathrm{R}_{\mathrm{f}}=0.38(\mathrm{EtOAc}) ;$ m.p. dec. $>195{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+101(\mathrm{c}=1, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3356 (br OH), $1717(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.11\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.14$ $(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.88-3.96\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 4.05(1 \mathrm{H}, \mathrm{t}, J 5.5, \mathrm{C}(2) H), 4.10-4.18(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{3} \mathrm{CHHO}\right), ~ 4.23-4.32\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CHHO}\right), 6.61(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.18\left(1 \mathrm{H}, \mathrm{d}, J 1.7, \mathrm{ArC}\left(1^{\prime}\right) H\right)$, 7.25-7.29 (1H, m, $\left.\operatorname{ArC}\left(6^{\prime}\right) H\right), 7.44\left(1 \mathrm{H}, \mathrm{t}, J 7.5, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.49\left(1 \mathrm{H}, \mathrm{d}, J 8.4, \operatorname{ArC}\left(5^{\prime}\right) H\right)$, $7.51\left(1 \mathrm{H}, \mathrm{dd}, J\right.$ 1.7, 8.1, $\left.\operatorname{ArC}\left(3^{\prime}\right) H\right), 7.64\left(1 \mathrm{H}, \mathrm{d}, J 8.1 \operatorname{ArC}\left(4^{\prime}\right) H\right), 7.82(1 \mathrm{H}, \mathrm{d}, J 7.5$, $\left.\operatorname{ArC}\left(8^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.1\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.1(C(3)), 40.1$ and $46.8,(C(4)$ and $\left.C\left(9^{\prime}\right)\right)$, $53.4(C(2)), 62.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.3\left(\mathrm{CH}_{2} \mathrm{OH}\right), 120.7\left(\mathrm{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 120.9\left(\mathrm{ArC}\left(2^{\prime}\right)\right)$, $121.1 \quad\left(\operatorname{ArC}\left(4^{\prime}\right) \mathrm{H}\right), \quad 122.8 \quad\left(\mathrm{ArC}\left(5^{\prime}\right) \mathrm{H}\right), \quad 124.0 \quad\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right), \quad 128.1 \quad\left(\mathrm{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 128.4$ $\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 138.8\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 138.3,138.6,141.5,144.3$ (all ArC ), 164.5, 168.7 (both
$(C=O)) ; m / z(E S I-) 428$ ([M-H] $\left.{ }^{-}, 100 \%\right), 426$ ([M-H] $\left.{ }^{-}, 97 \%\right)$; HRMS $450.0314\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{BrNNaO}_{4}$ requires 450.0311).
(2S,3S,4R,6R)-1-Aza-7'-bromo-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro [bicyclo[3.1.0]hexane-6,9'-fluorene] 13f

Following the general procedure, after column chromatography eluting with EtOAc, cyclopropane $\mathbf{8 c}$ ' ( $88 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) yielded alcohol $\mathbf{1 3 f}$ as a white solid ( $42 \mathrm{mg}, 58 \%$ ); $\mathrm{R}_{\mathrm{f}}$ $=0.61$ (EtOAc); m.p. dec. $204-206{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}-1.2(\mathrm{c}=0.52, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 1745$ $(\mathrm{C}=\mathrm{O}), 1684 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.12(1 \mathrm{H}, \mathrm{t}, J 5.3$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 3.15(1 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{C}(3) H), 3.89-3.98\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 4.03-4.07(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(2) H)$, 4.07-4.21 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$ ), $6.21(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.01\left(1 \mathrm{H}, \mathrm{d}, J 7.8, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.27(1 \mathrm{H}, \mathrm{td}, J$ 0.9, 7.8, $\left.\operatorname{ArC}\left(2^{`}\right) H\right), 7.39\left(1 \mathrm{H}, \operatorname{td}, J 0.9,7.7, \operatorname{ArC}\left(3^{`}\right) H\right), 7.56\left(1 \mathrm{H}, \mathrm{dd}, J 1.7,8.1, \operatorname{ArC}\left(6^{\prime}\right) H\right)$, $7.62\left(1 \mathrm{H}, \mathrm{d}, J 1.7, \operatorname{ArC}\left(8^{\prime}\right) H\right), 7.70\left(1 \mathrm{H}, \mathrm{d}, J 8.1, \operatorname{ArC}\left(5{ }^{\prime}\right) H\right), 7.70\left(1 \mathrm{H}, \mathrm{d}, J 8.1, \operatorname{ArC}\left(4^{\prime}\right) H\right)$; $\delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.1(C(3)), 40.1$ and $46.8\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 53.4$ $(C(2)), 62.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.4\left(\mathrm{CH}_{2} \mathrm{OH}\right), 119.9\left(\mathrm{ArC}\left(4^{\prime}\right) \mathrm{H}\right), 120.5\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 121.5$ $\left(\operatorname{ArC}\left(7^{\prime}\right)\right), 121.7\left(\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 125.9\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 127.6\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 128.0\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 131.3$ $\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 138.2,140.9,141.6,142.0,($ all ArC$), 164.5$ and 169.2, (both $(C=\mathrm{O})$ ); $m / z($ ESI-) 426 ([M-H] $\left.{ }^{-}, 99 \%\right), 428$ ([M-H], $100 \%$ ); HRMS 426.0346 ([M-H] ${ }^{-}, \mathrm{C}_{21} \mathrm{H}_{17} \mathrm{BrNO}_{4}$ requires 426.0346).
(2S,3S,4R,6S)-1,4'-Diaza-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro [bicyclo[3.1.0]hexane-6,9'-fluorene] 13h

To a solution of cyclopropane $\mathbf{8 e}(156 \mathrm{mg}, 0.36 \mathrm{mmol})$ in $\mathrm{DCM}(4 \mathrm{~mL})$ at r.t. was added a mixture containing TFA $(0.55 \mathrm{~mL}, 7.12 \mathrm{mmol})$ and water $(0.55 \mathrm{~mL})$. Purification by column chromatography over silica gel eluting with (EtOAc) yielded deprotected alcohol 13h $\mathbf{( 9 5} \mathbf{m g}$,
$76 \%)$ as a white solid; $\mathrm{R}_{\mathrm{f}}=0.06$ (EtOAc); m.p. dec. $>176{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{24}+29.9(\mathrm{c}=1.25$, $\mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) $3252(\mathrm{br}, \mathrm{OH}), 1700(\mathrm{C}=\mathrm{O}), 1591,1421,1284,1253,1180,1103$, $1045,755,737 ; \delta_{\mathrm{H}}(400.2 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 1.07\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.51(1 \mathrm{H}, \mathrm{s}$, C(3)H), $3.78(1 \mathrm{H}, \mathrm{dd}, J 5.9,11.2, \mathrm{CHHOH}), 3.83(1 \mathrm{H}, \mathrm{dd}, J 5.3,11.2, \mathrm{CH} H O H), 3.97(1 \mathrm{H}, \mathrm{t}$, $J 5.5, \mathrm{C}(2) H), 4.04-4.18\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 7.32\left(1 \mathrm{H}, \mathrm{dd}, J 5.1,7.9, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.44-7.49$ $\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(6^{\prime}\right) H\right), 7.52-7.57\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(1^{\prime}\right) H, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.61\left(1 \mathrm{H}, \mathrm{d}, J 7.9, \operatorname{ArC}\left(5{ }^{\prime}\right) H\right)$, $8.16\left(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{ArC}\left(8^{\prime}\right) H\right), 8.53\left(1 \mathrm{H}, \mathrm{dd}, J 1.2,5.1, \operatorname{ArC}\left(3{ }^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz} ; d 4-\mathrm{MeOD})$ $14.4\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.4(\mathrm{C}(3))$, 39.8 and $54.9\left(\mathrm{C}(4)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $55.3(\mathrm{C}(2)), 63.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right)$, $65.4\left(\mathrm{CH}_{2} \mathrm{OH}\right), 122.7\left(\mathrm{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 123.0\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 124.1\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 129.8\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right)$, 131.2, $133.4\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 138.7,141.5,141.8(\operatorname{all} \operatorname{ArC}), 148.7\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right)$, $158.3(\mathrm{ArC}), 166.0,170.6$ (both $(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+) 351\left([\mathrm{M}+\mathrm{H}]^{+}, 89 \%\right), 373\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $35 \%), 701\left([2 \mathrm{M}+\mathrm{H}]^{+}, 94 \%\right), 723\left([2 \mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right)$; HRMS $351.1332\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{4}\right.$ requires 351.1339 ).

## (2S,3S,4R,6R)-1,5'-Diaza-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro

 [bicyclo[3.1.0]hexane-6,9'-fluorene] 13iTo a solution of cyclopropane $\mathbf{8 e} \mathbf{e}^{\prime}(128 \mathrm{mg}, 0.29 \mathrm{mmol})$ in $\mathrm{DCM}(4 \mathrm{~mL})$ at r.t. was added a mixture containing TFA ( $0.45 \mathrm{~mL}, 5.84 \mathrm{mmol}$ ) and water $(0.45 \mathrm{~mL})$. Purification by column chromatography over silica gel eluting with (EtOAc) yielded deprotected alcohol $\mathbf{1 3 i}$ ( 68 mg , $67 \%$ ) as a white solid. $\mathrm{R}_{\mathrm{f}}=0.07$ (EtOAc); m.p. dec. $150-151{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+51.7(\mathrm{c}=0.92$, $\mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}(\mathrm{film}) 3265(\mathrm{br}, \mathrm{OH}), 1701(\mathrm{C}=\mathrm{O}), 1585,1457,1401,1256,1187,1108$; $\delta_{\mathrm{H}}(400.2 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 1.03\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.78(1 \mathrm{H}, \mathrm{dd}, J$ 5.9, 11.2, CHHOH), 3.83 (1H, dd, J 5.2, 11.2, CHHOH), 3.95 ( $1 \mathrm{H}, \mathrm{t}, J 5.6, \mathrm{C}(2) H), 4.02-4.16$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 7.14\left(1 \mathrm{H}, \mathrm{d}, J 7.3, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.37\left(1 \mathrm{H}, \mathrm{dd}, J 5.1,8.0, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.43-$ $7.53\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(3^{\prime}\right) H, \operatorname{ArC}\left(2^{`}\right) H\right), 7.96\left(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.0, \operatorname{ArC}\left(8{ }^{\prime}\right) H\right), 8.07-8.11(1 \mathrm{H}, \mathrm{m}$,
$\left.\operatorname{ArC}\left(4^{\prime}\right) H\right), 8.55\left(1 \mathrm{H}, \mathrm{dd}, 1.2,5.1, \operatorname{ArC}\left(6^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 14.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right)$, $38.6(C(3)), 39.6,49.9\left(C(4)\right.$, and $\left.C\left(9^{\prime}\right)\right), 55.6(C(2)), 63.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 65.4\left(\mathrm{CH}_{2} \mathrm{OH}\right), 122.2$ $\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 122.3\left(\mathrm{ArC}\left(4^{\prime}\right) \mathrm{H}\right), 123.4\left(\mathrm{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 129.4\left(\mathrm{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 131.3\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right)$, $133.6\left(\operatorname{Ar} C\left(8^{\prime}\right) \mathrm{H}\right), 136.2,137.9,145.0($ all ArC$), 147.9\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 160.7(\operatorname{ArC}), 165.7$, 170.5 (both $(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+) 351\left([\mathrm{M}+\mathrm{H}]^{+}, 74 \%\right), 701\left([2 \mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 723\left([2 \mathrm{M}+\mathrm{Na}]^{+}\right.$, $43 \%)$; HRMS $351.1339\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{4}\right.$ requires 351.1339).
(2S,3S,4R)-1,4',5'-Triaza-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-spiro[bicyclo

## [3.1.0]hexane-6,9'-fluorene] 13j

To a solution of cyclopropane $\mathbf{8 f}(59 \mathrm{mg}, 0.13 \mathrm{mmol})$ in DCM $(4 \mathrm{~mL})$ at r.t. was added a mixture containing TFA $(0.26 \mathrm{~mL}, 3.36 \mathrm{mmol})$ and water $(0.10 \mathrm{~mL})$. The mixture was stirred for 5 h before being concentrated in vacuo. Toluene ( 2 mL ) was added followed by 1 drop of water and the mixture was re-concentrated in vacuo. The addition of toluene and water followed by re-concentration was repeated (x 2) to give a brown viscous oil which was dissolved in the minimum amount of MeOH and heated to $60^{\circ} \mathrm{C}$. EtOAc was added dropwise to the solution until a white solid began to precipitate. After cooling to r.t. the brown liquid was removed before the white precipitate was washed with EtOAc ( $2 \times 1 \mathrm{~mL}$ ) and dried in vacuo to leave alcohol $\mathbf{1 3 j}$ as a white solid ( $45 \mathrm{mg}, 96 \%$ ); m.p. dec. $>110^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{25}+44(\mathrm{c}=$ $0.18, \mathrm{MeOH}$ ); $\mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3217 (br, OH ), 1733 (br, $\mathrm{C}=\mathrm{O}$ ), 1358, 1257, 1183, 1122; $\delta_{\mathrm{H}}(400.2 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 1.10\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.66(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.77-3.87(2 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 4.00(1 \mathrm{H}, \mathrm{t}, J 5.4, \mathrm{C}(2) H), 4.07-4.21\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 7.53(2 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\left.\operatorname{ArC}\left(2^{`}\right) H, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.74\left(1 \mathrm{H}, \mathrm{d}, J 7.7, \operatorname{ArC}\left(8^{`}\right) H\right), 8.07\left(1 \mathrm{H}, \mathrm{d}, J 7.9, \operatorname{ArC}\left(1^{`}\right) H\right), 8.74(2 \mathrm{H}$, br s, $\left.\operatorname{ArC}\left(3^{\prime}\right) H, \operatorname{ArC}\left(6^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 14.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.1(C(4))$, 38.6 $(C(3)), 48.3\left(C\left(9^{\prime}\right)\right), 55.6(C(2)), 63.4\left(\mathrm{CH}_{2} \mathrm{OH}\right), 65.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 125.1,125.3\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{Ar} C\left(7^{\prime}\right) \mathrm{H}\right), 132.2,133.4\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 137.1,140.1$ (both $\operatorname{ArC}$ ), 149.7,
$149.9\left(\operatorname{ArC} C\left(3^{\prime}\right) \mathrm{H}\right.$ and $\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}$ ), 155.7, 158.5 (both $\operatorname{ArC}$ ), 165.6, 170.2 (both $C=\mathrm{O}$ ); m/z (ESI+) $352\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS $374.1106\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{19} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{NaO}_{4}\right.$ requires 374.1111).

## (2S,3R,4S)-1-Aza-6,6-diphenyl-2-hydroxymethyl-5-oxo-bicyclo[3.1.0]hexane 13k

Following the general procedure, after column chromatography eluting with EtOAc, cyclopropane $\mathbf{4 a}(110 \mathrm{mg}, 0.30 \mathrm{mmol})$ yielded alcohol $\mathbf{1 3 k}$ as a white solid ( $62 \mathrm{mg}, 74 \%$ ); $\mathrm{R}_{\mathrm{f}}$ $=0.31(\mathrm{MeOH} / \mathrm{EtOAc}, 1: 9) ;$ m.p. $159-160{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{22}-33.2\left(\mathrm{c}=1.1, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) $3406,3350(\mathrm{NH}), 3055,1687,1496,1447,1266,737,706 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.49(1 \mathrm{H}$, d, $J 6.2, \mathrm{C}(4) H), 2.54(1 \mathrm{H}, \mathrm{d}, J 6.2, \mathrm{C}(3) H), 2.67\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.53(1 \mathrm{H}, \mathrm{t}, J 4.5$, $\mathrm{C}(2) H), 3.66(1 \mathrm{H}, \mathrm{dd}, J 6.3,11.2, \mathrm{C} H \mathrm{HOH}), 3.74(1 \mathrm{H}, \mathrm{dd}, J 11.2, \mathrm{CH} H \mathrm{OH}), 5.58(1 \mathrm{H}, \mathrm{s}$, $\mathrm{N} H), 7.11-7.32(8 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H), 7.42(2 \mathrm{H}, \mathrm{d}, J 7.92, \mathrm{Ph} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 30.9(C(4))$, $35.1(C(3)), 39.5(C(6)), 55.4(C(2)), 65.6\left(\mathrm{CH}_{2} \mathrm{OH}\right), 126.7(p-\mathrm{PhCH}), 127.0(\mathrm{PhCH}), 127.5$ ( $p$-Ph' $C H$ ), 128.5, 128.9, 129.8 (all PhCH$)$, 136.5, $143.4\left(\mathrm{Ph} C\right.$ and $\left.\mathrm{Ph} C^{\prime}\right), 175.5(C=\mathrm{O}) ; m / z$ (ESI-) 278 ([M-H] $\left.{ }^{-}, 100 \%\right)$; HRMS $302.1151\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NaNO}_{2}\right.$ requires 302.1151).

## (2S,3S,4S)-1-Aza-2-hydroxymethyl-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene] 131

 Following the general procedure, after column chromatography eluting with EtOAc, cyclopropane $7 \mathrm{a}(194 \mathrm{mg}, 0.53 \mathrm{mmol})$ yielded alcohol 131 as a white solid ( $136 \mathrm{mg}, 93 \%$ ); $\mathrm{R}_{\mathrm{f}}$ $=0.16(\mathrm{EtOAc}) ;$ m.p. $176-178{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{21}+235.5\left(\mathrm{c}=2.6, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}(\mathrm{film}) 3286$, $1682,1448,1204,1138,731 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.42(1 \mathrm{H}, \mathrm{d}, J 5.1,(\mathrm{C}(4) H)), 2.75(1 \mathrm{H}$, d, $J 5.1,(\mathrm{C}(3) H)), 3.37-3.61\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.76(1 \mathrm{H}$, br s, $(\mathrm{C}(2) H)), 4.57(1 \mathrm{H}$, br s, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 6.69(1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{ArCH}), 7.03(1 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{ArCH}), 7.09-7.22(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH})$, 7.22-7.32 (2H, m, $\operatorname{ArCH}), 7.63(2 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{ArCH}), 7.80(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 34.2(C(4)), 35.7\left(C\left(9^{\prime}\right)\right), 35.8(C(3)), 55.7(C(2)), 64.7\left(\mathrm{CH}_{2} \mathrm{OH}\right), 118.9,119.5$, 120.3, 122.3, 127.0, 127.4, 127.4, 127.4 (all ArCH ), 138.2, 140.1, 142.0, 145.5 (all ArC ),$174.7(C=\mathrm{O}) ; m / z(\mathrm{ESI}+) 336\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right) ;$ HRMS $300.0993\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NNaO}_{2}$ requires 300.0995).
(2S,3S,4R)-1-Aza-4-ethoxycarbonyl-2-methoxycarbonyl-5-oxo-spiro[bicyclo[3.1.0] hexane-6,9'-fluorene] 14a

A solution of $\mathbf{1 3 c}(149 \mathrm{mg}, 0.43 \mathrm{mmol})$ and PDC $(825 \mathrm{mg}, 2.34 \mathrm{mmol})$ in DMF $(10 \mathrm{~mL})$ was stirred under $\mathrm{N}_{2}$ for 2 d at $40^{\circ} \mathrm{C}$. After cooling to $0{ }^{\circ} \mathrm{C}$, a solution of diazomethane in $\mathrm{Et}_{2} \mathrm{O}$ was distilled into the reaction mixture. After standing for 30 mins , the mixture was purged with $\mathrm{N}_{2}$ to remove excess diazomethane. Sat. aq. $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and EtOAc ( 20 mL ) were added and the layers were separated. The aqueous layer was extracted with EtOAc ( $3 \times 20$ mL ) before the combined organic extracts were dried over $\mathrm{MgSO}_{4}$ and the solvent removed in vacuo to leave a brown oil which was purified by column chromatography over silica eluting with EtOAc/Petrol ( $1: 2$ ) to give 14a ( $59 \mathrm{mg}, 37 \%$ ) as a white wax; $\mathrm{R}_{\mathrm{f}}=0.35(\mathrm{EtOAc} /$ Petrol, $2: 1) ;[\alpha]_{\mathrm{D}}{ }^{18}-17.8\left(\mathrm{c}=5.1, \mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1}($ film $) 3286(\mathrm{br}, \mathrm{NH}), 1723,1450,1266,1110$, 1018,$744 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.06\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.62(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.88$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.08-4.28\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.52(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.09(1 \mathrm{H}, \mathrm{d}, J 7.7$, $\mathrm{Ar} H)$, 7.15-7.21 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ), 7.25-7.30 (1H, m, $\operatorname{Ar} H), 7.38-7.49(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.66(1 \mathrm{H}, \mathrm{s}$, $\mathrm{N} H), 7.80(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 7.87(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.0$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 36.9(C(3)), 41.0$ and $46.1(C(4)$ and $C(6))$, $53.2\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right)$, $54.0(C(2)), 62.2$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 119.9,120.6,120.7,122.3,127.2,127.9,128.0,128.3$ (all ArCH$), 138.1,139.3$, 141.7, 142.5 (all ArC$), 164.0,169.1,169.9($ all $(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 436\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}\right.$, $100 \%), 813\left(\left[2 \mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 30 \%\right), 777\left([2 \mathrm{M}+\mathrm{Na}]^{+}, 13 \%\right), 1154\left([3 \mathrm{M}+\mathrm{Na}]^{+}, 5 \%\right), 400$ $\left([\mathrm{M}+\mathrm{Na}]^{+}, 3 \%\right) ;$ HRMS $400.1153\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{22} \mathrm{H}_{19} \mathrm{NNaO}_{5}\right.$ requires 400.1155).

A solution of $131(61 \mathrm{mg}, 0.22 \mathrm{mmol})$ and PDC $(385 \mathrm{mg}, 1.11 \mathrm{mmol})$ in DMF $(4.5 \mathrm{~mL})$ was stirred under $\mathrm{N}_{2}$ for 2 d at $40^{\circ} \mathrm{C}$. After cooling to $0{ }^{\circ} \mathrm{C}$, a solution of diazomethane in $\mathrm{Et}_{2} \mathrm{O}$ was distilled directly into the reaction mixture. After standing for 30 min , the mixture was purged with $\mathrm{N}_{2}$ to remove excess diazomethane. The mixture was concentrated in vacuo to give a dark brown solid. Sat. aq. $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ and EtOAc $(10 \mathrm{~mL})$ were added and the layers were separated. The aqueous layer was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ) before the combined organic extracts were washed with brine, dried over $\mathrm{MgSO}_{4}$ and the solvent removed in vacuo to leave a pale green oil which was purified by column chromatography over silica eluting with EtOAc/Petrol (1:1) to give 14b (19 mg, 29\%) as a clear gum; $\mathrm{R}_{\mathrm{f}}=$ $0.24(\mathrm{EtOAc} /$ Petrol, $2: 1) ;[\alpha]_{\mathrm{D}}{ }^{21}+112.0\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3225,3063,2954$, 1749 and $1704(2 \times \mathrm{C}=\mathrm{O}), 1449,1216,773 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) ; 2.99(1 \mathrm{H}, \mathrm{d}, J 6.3$, $\mathrm{C}(3) H), 3.17(1 \mathrm{H}, \mathrm{d}, J 6.25, \mathrm{C}(4) H), 3.85\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.56(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.96(1 \mathrm{H}, \mathrm{d}$, $J 7.6, \mathrm{Ar} H), 7.01(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.19(1 \mathrm{H}, \mathrm{dt}, J 1.1,7.6, \mathrm{Ar} H), 7.33(1 \mathrm{H}, \mathrm{dt}, J 1.1,7.6, \mathrm{Ar} H)$, 7.38-7.46 (3H, m, ArH), $7.81(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 7.89(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 34.5,34.5(C(3)$ and $C(4)), 36.6(C(6)), 53.1\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 55.4(C(2)), 118.9,119.8$, 120.7, 122.1, 127.4, 127.6, 127.7, 127.8 (all ArCH), 138.5, 139.6, 142.4, 145.4 (ArC), 170.7, 172.7 (both $(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 364\left(\left[\mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}, 100 \%\right), 669\left(\left[2 \mathrm{M}+\mathrm{MeCN}+\mathrm{NH}_{4}\right]^{+}\right.$, $23 \%)$; HRMS $328.0944\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{19} \mathrm{H}_{15} \mathrm{NNaO}_{3}\right.$ requires 328.0944).

## Attempted synthesis of (2S,3R,4R)-1-aza-6,6-diphenyl-4-ethoxycarbonyl-2-methoxy carbonyl-5-oxo-bicyclo[3.1.0]-hexane 14a

A solution of $\mathbf{1 3 a}(114 \mathrm{mg}, 0.32 \mathrm{mmol})$ and $\operatorname{PDC}(565 \mathrm{mg}, 1.62 \mathrm{mmol})$ in DMF $(6 \mathrm{~mL})$ was stirred under $\mathrm{N}_{2}$ for 2 d at $40^{\circ} \mathrm{C}$. After cooling to $0{ }^{\circ} \mathrm{C}$, a solution of diazomethane in $\mathrm{Et}_{2} \mathrm{O}$ was distilled directly into the reaction mixture. After standing for $1 \mathrm{~h}, \mathrm{AcOH}$ was added to quench any excess diazomethane. The mixture was concentrated in vacuo to give a dark
brown solid. Sat. aq. $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ and $\mathrm{EtOAc}(10 \mathrm{~mL})$ were added and the layers were separated. The aqueous layer was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ) before the combined organic extracts were washed with brine, dried over $\mathrm{MgSO}_{4}$ and the solvent removed in vacuo to leave a pale green oil which was purified by column chromatography over silica eluting with $\mathrm{EtOAc} /$ Petrol $(1: 3)$ to give $\mathbf{1 4 a}(35 \mathrm{mg}, 29 \%)$ as a white wax whose spectroscopic properties were identical to those reported above.

## Attempted synthesis of (2S,3R,4R)-4-ethoxycarbonyl-5-oxo-6,6-diphenyl-1-aza

## bicyclo[3.1.0]hexane-2-aldehyde ${ }^{78}$

To a solution of oxalyl chloride ( $20 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ) in dry $\mathrm{DCM}(1 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was added DMSO $(50 \mu \mathrm{~L})$. After 15 mins a solution of $\mathbf{1 3 a}(70 \mathrm{mg}, 0.19 \mathrm{mmol})$ in $\mathrm{DCM}(0.33 \mathrm{~mL})$ was added dropwise. The mixture was then left to stir at $-78{ }^{\circ} \mathrm{C}$ for 35 min before triethylamine $(0.32 \mathrm{~mL})$ was added (all in one go) and the left to stir for 45 min before taking to $-42{ }^{\circ} \mathrm{C}$. After 10 min , water ( 1 mL ) was added (all in one go) and the mixture was allowed to warm to r.t. before DCM ( 5 mL ) and water ( 5 mL ) were added. The organic layer was removed and the aqueous layer washed with $\operatorname{DCM}(2 \times 10 \mathrm{~mL})$. The combined organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo. The crude was purified by column chromatography over silica eluting with EtOAc/Petrol (1:5) to give $\mathbf{1 5}(46 \mathrm{mg}, 66 \%)$ as a white solid; m.p. $176-177{ }^{\circ} \mathrm{C}, \mathrm{Rf}=0.73(\mathrm{EtOAc}) ;[\alpha]_{\mathrm{D}}{ }^{18}-84.8\left(\mathrm{c}=1.4, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3428(\mathrm{OH})$, 3060, 2981, 1732 (C=O), 1496, 1409, 1264, 1138, 1085, 1022; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.89$ ( $6 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$ ), 2.30-3.20 (2H, br s, OH), $3.33(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{C}(2) H), 3.37(2 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(3) H), 3.90\left(4 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 5.16(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{C}(H)(\mathrm{N})(\mathrm{OH})), 7.14-7.26(6 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph} H)$, 7.26-7.40 ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H$ ), $7.48(4 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ph} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.6$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 31.7(C(3)), 44.0$ and $48.9(C(4)$ and $C(6)), 52.9(C(2)), 61.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 74.5$ $(\mathrm{RC}(\mathrm{H})(\mathrm{N})(\mathrm{OH})), 127.6,127.8,128.2,128.6,128.7,129.0($ all PhCH$), 136.0,129.0$ (both

PhC ), 164.5 and 168.4 (both $(C=\mathrm{O})$ ); $m / z(\mathrm{ESI}+) 757\left(\left[\mathrm{M}+\mathrm{NH}_{4}+\mathrm{MeCN}\right]^{+}, 100 \%\right), 721$ (([M+Na] $\left.]^{+}, 25 \%\right)$.

## (2S,3R,4R)-1-Aza-6,6-di(p-methoxyphenyl)-2-hydroxymethyl-5-oxo-bicyclo[3.1.0]

## hexane-4-carboxylic acid 16

Following general procedure, a solution of $\mathbf{1 3 b}(12 \mathrm{mg}, 29 \mu \mathrm{~mol}), \mathrm{MeOH}(0.9 \mathrm{~mL})$, water ( 0.3 mL ) and lithium hydroxide ( $6 \mathrm{mg}, 143 \mu \mathrm{~mol}$ ) was stirred at r.t. for 2 d . The work-up procedure yielded 16 as a white solid ( $11 \mathrm{mg}, 100 \%$ ) which required no further purification; m.p. dec. $>134{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{22}-105(\mathrm{c}=0.55, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3315(\mathrm{brOH}), 2936,2838$, $1731(\mathrm{C}=\mathrm{O}), 1671(\mathrm{C}=\mathrm{O}), 1609,1581,1512,1441,1292,1248,1178,1114,1032,841,735$; $\delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 3.25(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.45(1 \mathrm{H}, \mathrm{t}, J 5.1, \mathrm{C}(2) H), 3.55-3.69(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 3.71\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.73\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OC}^{\prime} H_{3}\right), 6.77\left(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{Ar}{ }^{\prime} H(o-\right.$ to OMe$)$ ), $6.85\left(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{ArH}\right.$ (o- to OMe)), 7.31 (2H, d, $J 8.4, \mathrm{Ar}^{\prime} H(m-$ to OMe$)$ ), 7.39 (2H, d, $J$ 8.4, $\mathrm{ArH}(m-\mathrm{to} \mathrm{OMe})) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 35.0(C(3)), 45.5\left(C\left(9^{\prime}\right)\right), 54.9(C(4))$, $55.0(C(2)), 55.7\left(\mathrm{CH}_{3} \mathrm{O}\right), 55.7\left(\mathrm{C}^{\prime} \mathrm{H}_{3} \mathrm{O}\right), 65.3\left(\mathrm{CH}_{2} \mathrm{OH}\right), 114.8\left(\mathrm{Ar}{ }^{\prime} \mathrm{CH}(o-\mathrm{to} \mathrm{OMe})\right), 115.3$ $(\mathrm{ArCH}(o-$ to OMe$)), 130.8\left(\mathrm{Ar}^{\prime} \mathrm{CH}(m-\right.$ to OMe$\left.)\right), 130.9(\mathrm{ArCH}(m-$ to OMe$)), 131.5$ and $133.7\left(\operatorname{Ar}^{\prime} C(p\right.$ - to OMe$)$ and $\operatorname{ArC}(p$ - to OMe$)$ ), 160.2 and 160.4, $\left(\mathrm{Ar}^{\prime} \mathrm{COCH}_{3}\right.$ and $\mathrm{ArCOCH}_{3}$ ), 161.4 and 174.3 (both ( $\mathrm{C}=\mathrm{O}$ )); $\mathrm{m} / \mathrm{z}$ (ESI-) 382 ([M-H] ${ }^{-}, 100 \%$ ); HRMS 406.1257 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{21} \mathrm{H}_{21} \mathrm{NNaO}_{6}\right.$ requires 406.1261).

## (2S,3S,4R)-1-Aza-2-hydroxymethyl-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-4carboxylic acid 17a

Following the general procedure, a solution of $\mathbf{1 3 c}(90 \mathrm{mg}, 0.26 \mathrm{mmol}), \mathrm{MeOH}(6 \mathrm{~mL})$, water $(2 \mathrm{~mL})$ and lithium hydroxide $(57 \mathrm{mg}, 1.36 \mathrm{mmol})$ was stirred at r.t. for 2 d . The work-up procedure yielded $\mathbf{1 7 a}$ as a white powder ( $80 \mathrm{mg}, 97 \%$ ) which required no further
purification; m.p. $196^{\circ} \mathrm{C}($ dec. $) ;[\alpha]_{\mathrm{D}}{ }^{20}+70.5(\mathrm{c}=1.65, \mathrm{EtOH}) ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{MeOD}, \mathrm{D}_{2} \mathrm{O}\right)$ $3.30(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.77-3.87\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.97(1 \mathrm{H}, \mathrm{t}, J 5.4, \mathrm{C}(2) H), 7.17(1 \mathrm{H}, \mathrm{d}, J 7.7$, $\mathrm{Ar} H), 7.28-7.36(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.41(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar} H), 7.44-7.52(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.84(1 \mathrm{H}, \mathrm{d}$, $J 7.58, \mathrm{Ar} H), 7.91(1 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{MeOD}, \mathrm{D}_{2} \mathrm{O}\right) 37.5(C(3)), 40.7\left(C\left(9^{\prime}\right)\right)$, $54.3(C(2)), 64.3\left(\mathrm{CH}_{2} \mathrm{OH}\right), 120.0(\mathrm{ArCH}), 120.7,121.2,122.9,127.6,127.8,128.1,128.5$, $139.2(\mathrm{ArC}), 139.5(\mathrm{ArC}), 142.6(\mathrm{ArC}), 142.7(\mathrm{ArC}), 167.3$ and 170.9 (both $(C=\mathrm{O})$ ); $\mathrm{m} / \mathrm{z}$ (ESI-) 962 ([3M-H] $\left.{ }^{-}, 100 \%\right), 641$ ([2M-H], $94 \%$ ); HRMS 320.0923 ([M-H] $]^{-}, \mathrm{C}_{19} \mathrm{H}_{14} \mathrm{NO}_{4}$ requires 320.0917 ).

## (2S,3S,4R,6R)-1,5'-Diaza-2-hydroxymethyl-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-

## fluorene]-4-carboxylic acid 17b

To a solution of $\mathbf{1 3 i}(11 \mathrm{mg}, 33 \mu \mathrm{~mol})$ in $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(3: 1)(1.2 \mathrm{~mL})$ was added $\mathrm{LiOH}(7 \mathrm{mg}$, $167 \mu \mathrm{~mol})$. The mix was stirred vigorously for 2 d and concentrated in vacuo. $\mathrm{D}_{2} \mathrm{O}(1 \mathrm{~mL})$ was added and the solution was concentrated, to leave $\mathbf{1 7 b}$ as a mixture containing excess LiOD; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 2.97(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.65-3.74\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(2) H\right), 6.95$ $\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(4^{\prime}\right) H\right), 7.09\left(1 \mathrm{H}, \mathrm{dd}, J 5.0,8.2, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.23-7.29\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H\right.$, $\left.\operatorname{ArC}\left(3^{\prime}\right) H\right), 7.52\left(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.2, \operatorname{ArC}\left(8^{\prime}\right) H\right), 7.67-7.71\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(1^{\prime}\right) H\right), 8.22(1 \mathrm{H}, \mathrm{dd}, J$ 1.3, 5.0, $\left.\operatorname{ArC}\left(6^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 37.6(C(4)), 38.0(C(3)), 51.0\left(C\left(9^{\prime}\right)\right), 55.0(C(2))$, $64.1\left(\mathrm{CH}_{2} \mathrm{OH}\right), 120.2\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 120.7\left(\mathrm{ArC}\left(4^{\prime}\right) \mathrm{H}\right), 121.9\left(\mathrm{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 127.5$ and 129.5 $\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 130.8\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 134.5,136.7,144.1$ (all ArC ), 147.0 $\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 159.2(\mathrm{ArC}), 170.0$ and 173.1 (both $\left.(C=\mathrm{O})\right) ; m / z(\mathrm{ESI}-) 321$ ([M-H] ${ }^{-}, 100 \%$ ); HRMS 321.0874 ([M-H], $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires 321.0881 ).
(2S,3S,4R)-1,4',5’-Triaza-2-hydroxymethyl-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9’-fluorene]-4-carboxylic acid 17c

To a solution of $\mathbf{1 3 j}(16 \mathrm{mg}, 46 \mu \mathrm{~mol})$ in $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(3: 1)(1.8 \mathrm{~mL})$ was added $\mathrm{LiOH}(10$ $\mathrm{mg}, 228 \mu \mathrm{~mol})$. The mix was stirred vigorously for 2 d and concentrated in vacuo. $\mathrm{D}_{2} \mathrm{O}(1$ mL ) was added and the solution was concentrated, to leave $\mathbf{1 7 c}$ as a mixture containing excess LiOD; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right)$ 3.16-3.20 ( 1 H , br s, $\mathrm{C}(3) H$ ), 3.68-3.78 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}$ ), 3.81$3.87(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(2) H), 7.23-7.31\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.44-7.49$ and 7.73-7.78(2H, m, $\left.\operatorname{ArC}\left(1^{`}\right) H, \operatorname{ArC}\left(8^{\prime}\right) H\right), 8.42-8.48\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(3{ }^{\prime}\right) H, \operatorname{ArC}\left(6^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 35.9$ $(C(4)), 38.6(C(3)), 51.2\left(C\left(9^{\prime}\right)\right), 56.1(C(2)), 64.3\left(\mathrm{CH}_{2} \mathrm{OH}\right), 123.7$ and $123.7\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 129.9$ and $131.3\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 136.1,139.5$ (both $\left.\operatorname{ArC}\right), 148.3$ and $148.4\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 154.3,157.3$ (both $\operatorname{ArC}$ ), 168.3, 173.0 (both $(C=\mathrm{O})$ ); $m / z$ (ESI-) 322 ([M-H] ${ }^{-}, 100 \%$ ); HRMS 328.0918 ([M-H] ${ }^{-}, \mathrm{C}_{17} \mathrm{H}_{11} \mathrm{LiN}_{3} \mathrm{O}_{4}$ requires 328.0915).
(2S,3S,4R)-1-Aza-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-biscarboxylic acid

## 18a

Following the general procedure, a solution of $\mathbf{1 4 a}(25 \mathrm{mg}, 66 \mu \mathrm{~mol})$, $\mathrm{MeOH}(2.1 \mathrm{~mL})$, water $(0.7 \mathrm{~mL})$ and lithium hydroxide $(40 \mathrm{mg}, 953 \mu \mathrm{~mol})$ was stirred at r.t. for 3 d . The work-up procedure yielded 18a as a white solid ( $22 \mathrm{mg}, 100 \%$ ) which required no further purification; m.p. dec. $>171{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}+20.3(\mathrm{c}=0.6, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3584,3386(\mathrm{br} \mathrm{COO}-\mathrm{H})$, $1680(\mathrm{C}=\mathrm{O}), 1447,1214 ; \delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 3.54(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 4.51(1 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(2) H), 7.21(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H), 7.27-7.34(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.42(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar} H), 7.47(1 \mathrm{H}$, t, $J 7.5, \operatorname{Ar} H), 7.53(1 \mathrm{H}, \mathrm{d}, J 7.9, \operatorname{Ar} H), 7.87(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{Ar} H), 7.94(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{Ar} H)$; $\delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 38.7(C(3)), 42.1\left(C\left(9^{\prime}\right)\right), 54.9(C(2)), 55.3(C(4)), 120.9,121.7$, 122.2, 123.6, 128.5, 128.6, 129.0, 129.4 (all ArCH), 139.9, 140.6, 143.6, 144.0 (all ArC ), 167.2, 171.2, 173.2 (all ( $C=O$ ) ); $m / z$ (ESI-) 334 ([M-H] ${ }^{-}, 100 \%$ ); HRMS 334.0739 ([M-H] ${ }^{-}$, $\mathrm{C}_{19} \mathrm{H}_{13} \mathrm{NO}_{5}$ requires 334.0721).

Following the general procedure, a solution of $\mathbf{1 4 b}(10 \mathrm{mg}, 33 \mu \mathrm{~mol})$, $\mathrm{MeOH}(1.5 \mathrm{~mL})$, water $(0.5 \mathrm{~mL})$ and lithium hydroxide ( $20 \mathrm{mg}, 477 \mu \mathrm{~mol}$ ) was stirred at r.t. for 2 d . The work-up procedure yielded 18b as a glassy solid ( $9 \mathrm{mg}, 100 \%$ ) which required no further purification; m.p. dec. $>171{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{18}+101.7(\mathrm{c}=0.35, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3240(\mathrm{br} \mathrm{COOH}), 3070$, $1750(\operatorname{sh} \mathrm{C}=\mathrm{O}), 1700(\mathrm{C}=\mathrm{O}), 1449,1218,733 ; \delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 2.98(1 \mathrm{H}, \mathrm{d}, J 6.3$, $\mathrm{C}(3) H), 3.18(1 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{C}(4) H), 4.55(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.09(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 7.30-7.49$ (5H, m, ArH), $7.85(1 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H), 7.95(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD})$ $35.6(C(4)), 36.2(C(3)), 37.5\left(C\left(9^{\prime}\right)\right), 57.0(C(2)), 120.3,120.7,121.7,123.2,128.4,128.4$, 128.9, $128.9(\operatorname{all} \mathrm{Ar} C \mathrm{H}), 139.6,141.0,143.8,146.9($ all ArC$), 173.6$ and 175.3 (both $(C=\mathrm{O})$ ); $m / z$ (ESI-) 290 ([M-H], 94\%); HRMS $314.0787\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{18} \mathrm{H}_{13} \mathrm{NNaO}_{3}\right.$ requires 314.0788).
( $2 R, 5 S, 6 S, 7 R, 9 S$ )-4'-Methoxycarbonylspiro[1-aza-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-tricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right]$ nonane-9,9'-fluorene] 19a

To a mixture of $\mathbf{8} \mathbf{b}(213 \mathrm{mg}, 0.46 \mathrm{mmol})$, TEMPO ( $11 \mathrm{mg}, 0.07 \mathrm{mmol}$ ), $\mathrm{NaH}_{2} \mathrm{PO}_{4}(3.1 \mathrm{~mL}$, $0.67 \mathrm{M})$ and $\mathrm{MeCN}(3.6 \mathrm{~mL})$ at $37{ }^{\circ} \mathrm{C}$ was added $\mathrm{NaClO}_{2}(120 \mathrm{mg}, 1.32 \mathrm{mmol})$ followed by 5 drops of $8 \%$ aq. NaOCl . The mixture was stirred vigorously for 5 h before allowing to cool to r.t. and then poured over an ice-cold solution of sodium sulfite ( 1.0 g in 1.0 mL water and 2.0 g of ice). EtOAc ( 5 mL ) was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 5 \mathrm{~mL}$ ), before the combined organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to yield the crude acid a white solid which was used without further purification; HRMS $482.1598\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{29} \mathrm{H}_{24} \mathrm{NO}_{6}\right.$ requires 482.1599); The residue was dissolved in EtOAc ( 10 mL ) and diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo to leave 19a as a white foam ( $210 \mathrm{mg}, 93 \%$ ) which required
no further purification. $[\alpha]_{\mathrm{D}}^{25}+114\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 2983,1716(\mathrm{C}=\mathrm{O})$, $1435,1272,1135,1027,731 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.46(1 \mathrm{H}$, $\mathrm{s}, \mathrm{C}(6) H), 3.93(1 \mathrm{H}, \mathrm{dd}, J 8.0,9.4, \mathrm{C}(4) H \mathrm{H}), 4.03\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.04-4.16(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.19(1 \mathrm{H}, \mathrm{dd}, J 6.1,9.4, \mathrm{C}(5) H), 4.50(1 \mathrm{H}, \mathrm{dd}, J 6.1,7.8, \mathrm{C}(4) \mathrm{H} H), 6.50(1 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(2) H), 6.98-7.03(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.11(1 \mathrm{H}, \mathrm{dd}, J 0.9,7.8, \operatorname{Ar} H), 7.27(1 \mathrm{H}, \mathrm{t}, J 7.8, \operatorname{Ar} H), 7.32-$ $7.60(7 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.72(1 \mathrm{H}, \mathrm{dd}, J 1.0,7.7, \mathrm{Ar} H), 8.26(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 13.8\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.8(C(6))$, 42.4 and $49.7\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $52.4\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right)$, 56.1 $(C(5)), 62.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.6(C(4)), 88.8(C(2)), 123.0,123.0,124.8,125.8,126.2(\mathrm{ArCH})$, $126.7\left(\mathrm{ArCCO}_{2} \mathrm{CH}_{3}\right), 127.9,128.0,128.5,128.8,129.2(\mathrm{ArCH}), 137.6(\mathrm{PhC}), 138.2,139.6$, 140.6, 143.6 (ArC), 164.4, 168.5, $170.4(C=O)$; HRMS $496.1754\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{30} \mathrm{H}_{26} \mathrm{NO}_{6}\right.$ requires 496.1755).

## ( $2 R, 5 S, 6 S, 7 R, 9 R$ )-5'-Methoxycarbonylspiro[1-aza-7-ethoxycarbonyl-3-oxa-8-oxo-2-

 phenyl-tricyclo $\left[4.3 .0^{1,5} .0^{6,7}\right]$ nonane-9,9'-fluorene] 19bA (1: 1) mixture of diastereoisomers $\mathbf{8 b}$ and $\mathbf{8} \mathbf{b}^{\prime}(291 \mathrm{mg})$ was oxidised by an identical procedure to that described for the preparation of 19a above, with quantitative conversion. Column chromatography over silica gel eluting with EtOAc/Petrol (1:3) furnished 19b. $[\alpha]_{\mathrm{D}}{ }^{25}+84\left(\mathrm{c}=1.2, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 2983,1721(\mathrm{C}=\mathrm{O}), 1433,1372,1276,1134$, 1049, 734; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.02\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.47(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(6) H), 3.92$ $(1 \mathrm{H}, \mathrm{dd}, J 8.0,9.4, \mathrm{C}(4) \mathrm{HH}), 4.01\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.06-4.15\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.18(1 \mathrm{H}$, dd, $J 6.1,9.4, \mathrm{C}(5) H), 4.51(1 \mathrm{H}, \mathrm{dd}, J 6.1,8.0, \mathrm{C}(4) \mathrm{H} H), 6.49(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 6.95-6.71(2 \mathrm{H}$, $\mathrm{m}, \mathrm{Ar} H), 7.28(1 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{Ar} H), 7.34-7.48(4 \mathrm{H}, \mathrm{m}, o-\mathrm{Ph} H, p-\mathrm{Ph} H, \mathrm{Ar} H), 7.55-7.57(2 \mathrm{H}, \mathrm{m}$, $m-\mathrm{Ph} H), 7.65-7.70(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 8.16(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(50.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.8$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 38.4(\mathrm{C}(6))$, 42.6 and $50.0\left(C(7)\right.$ and $\left.C\left(9^{\prime}\right)\right), 52.5\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 56.3(C(5)), 62.3$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 69.8(\mathrm{C}(4)), 88.9(C(2)), 119.9,124.4$ (both ArCH$), 125.7(p-\mathrm{PhCH})$, 126.2,
126.6 (both ArCH$), 127.4\left(\mathrm{ArCCO}_{2} \mathrm{Me}\right), 127.9(p-\mathrm{PhCH}), 127.9(\mathrm{ArCH}), 128.6(m-\mathrm{PhCH})$, 128.9, 129.4 (both ArCH ), 137.4, 138.3, 140.7 (all ArC ), 140.9 ( PhC ), 142.8 ( ArC ), 164.4, 168.7, 171.1 (all ( $\mathrm{C}=\mathrm{O}$ ) ); HRMS $518.1575\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{30} \mathrm{H}_{25} \mathrm{NNaO}_{6}\right.$ requires 518.1574).
(2S,3S,4R,6S)-1-Aza-4-ethoxycarbonyl-2-hydroxymethyl-4'-methoxycarbonyl-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene] 20a

To a solution of cyclopropane 19a ( $100 \mathrm{mg}, 0.20 \mathrm{mmol}$ ) in $\mathrm{DCM}(4 \mathrm{~mL})$ at r.t. was added a mixture containing TFA $(0.16 \mathrm{~mL}, 2.0 \mathrm{mmol})$ and water $(0.16 \mathrm{~mL})$. After removal of solvents the resultant oil was dissolved in EtOAc ( 1 mL ). Petrol was added dropwise until a white precipitate had formed and the filtrate was revoved via pipette leaving 20a as a white solid ( $74 \mathrm{mg}, 90 \%$ ) which required no further purification; m.p. dec. $>140{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}+79.9(\mathrm{c}=$ $\left.0.76, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3333 \mathrm{br}(\mathrm{OH}), 2383,2358,1723(\mathrm{C}=\mathrm{O}), 1442,1279 ; \delta_{\mathrm{H}}(500.3$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.98\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 3.27(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.36(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.69-$ $3.76(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{HOH}), 3.80-3.86(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} \mathrm{OH}) 3.90-3.95(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(2) H), 3.96-4.14$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.03\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCO}_{2} \mathrm{CH}_{3}\right), 7.13\left(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{ArC}\left(1^{\prime}\right) H\right), 7.20(1 \mathrm{H}, \mathrm{t}, J$ 7.6, $\left.\operatorname{ArC}\left(6^{\prime}\right) H\right), 7.26\left(1 \mathrm{H}, \mathrm{t}, J 7.7, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.39\left(1 \mathrm{H}, \mathrm{t}, J 7.6, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.50(1 \mathrm{H}, \mathrm{d}, J$ 7.9, $\left.\operatorname{ArC}\left(5^{\prime}\right) H\right), 7.57(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.71\left(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{ArC}\left(3^{\prime}\right) H\right), 8.29(1 \mathrm{H}, \mathrm{d}, J 7.9$, $\left.\operatorname{ArC}\left(8^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.4(C(3))$, 39.7 and $47.6(C(4)$ and $\left.C\left(9^{\prime}\right)\right), 52.5\left(\mathrm{ArCO}_{2} \mathrm{CH}_{3}\right), 53.7(\mathrm{C}(2)), 62.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 64.5\left(\mathrm{CH}_{2} \mathrm{OH}\right), 122.2\left(\mathrm{ArC}\left(5^{\prime}\right)\right)$, $123.3\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 125.0\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 126.3\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 126.8\left(\mathrm{ArC}^{\prime}\left(4^{\prime}\right) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 128.1$ $\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 128.7\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 129.2\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 137.8,139.6,140.6,144.0(\mathrm{all} \mathrm{ArC}), 165.2$, 168.8, $169.7(\operatorname{all}(C=O)) ;$ HRMS $430.1259\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{23} \mathrm{H}_{21} \mathrm{NNaO}_{6}\right.$ requires 430.1261).

To a solution of cyclopropane 19b ( $94 \mathrm{mg}, 0.19 \mathrm{mmol}$ ) in DCM ( 4 mL ) at r.t. was added a mixture containing TFA $(0.15 \mathrm{~mL}, 2.0 \mathrm{mmol})$ and water $(0.15 \mathrm{~mL})$. After removal of solvents the resultant oil was dissolved in EtOAc ( 1 mL ). Petrol was added dropwise until a white precipitate had formed and the filtrate was revoved via pipette leaving 20b as a white solid ( $57 \mathrm{mg}, 74 \%$ ) which required no further purification; $[\alpha]_{\mathrm{D}}{ }^{25}+33.7\left(\mathrm{c}=0.51, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-}$ ${ }^{1}$ (film) $3345(\mathrm{br}, \mathrm{OH}), 1724(\mathrm{C}=\mathrm{O}), 1704(\mathrm{C}=\mathrm{O}), 1434,1138,1117 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $1.00\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 2.71(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.21(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.79-3.90(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 3.96(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{C}(2) \mathrm{H}), 4.01-4.18\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 4.03\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCO}_{2} \mathrm{CH}_{3}\right)$, $7.00\left(1 \mathrm{H}, \mathrm{d}, J 7.6, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.18\left(1 \mathrm{H}, \mathrm{t}, J 7.7, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.27-7.31\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H\right)$, $7.37\left(1 \mathrm{H}, \mathrm{t}, J 7.3, \operatorname{ArC}\left(3^{\prime}\right) H\right), 7.60(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.63\left(1 \mathrm{H}, \mathrm{d}, J 7.9 \operatorname{ArC}\left(6^{\prime}\right.\right.$ or $\left.\left.8^{\prime}\right) H\right), 7.67$ $\left(1 \mathrm{H}, \mathrm{d}, J 7.9 \mathrm{ArC}\left(6^{\prime}\right.\right.$ or $\left.\left.8^{\prime}\right) H\right), 8.18\left(1 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{ArC}\left(4^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 37.9(C(3)), 39.7$ and $39.8\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 52.5\left(\mathrm{ArCO}_{2} \mathrm{CH}_{3}\right), 53.9(C(2)), 62.3$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}\right), 68.5\left(\mathrm{CH}_{2} \mathrm{OH}\right), 120.0\left(\mathrm{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 124.3\left(\mathrm{ArC}\left(4^{\prime}\right) \mathrm{H}\right)$, $125.4\left(\mathrm{ArC}\left(6^{\prime}\right.\right.$ or $\left.\left.8^{\prime}\right) \mathrm{H}\right)$, $127.0\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 127.4\left(\operatorname{ArC}\left(5^{\prime}\right) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 127.7\left(\mathrm{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 127.8\left(\mathrm{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 129.2$ ( $\operatorname{ArC}\left(6^{\prime}\right.$ or $\left.\left.8^{\prime}\right) \mathrm{H}\right), 137.4,140.6,140.8,143.1$ (all $\mathrm{Ar} C$ ), 165.0, 168.8, 168.9 (all ( $\left.C=\mathrm{O}\right)$ ); HRMS $430.1258\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{23} \mathrm{H}_{21} \mathrm{NNaO}_{6}\right.$ requires 430.1261).
(2S,3S,4R)-1-Aza-1-benzyl-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-bismethanol $21 a^{79}$

To a suspension of $\mathrm{LiAlH}_{4}(100 \mathrm{mg}, 2.64 \mathrm{mmol})$ in THF ( 3 mL ) at r.t. under an atmosphere of $\mathrm{N}_{2}$ was added a solution of cyclopropane 8a ( $250 \mathrm{mg}, 0.57 \mathrm{mmol}$ ) in THF ( 6 mL ). After stirring for 1 h , the mixture was heated to reflux for 14 h before being allowed to cool to r.t. Sat. aq. $\mathrm{NaHCO}_{3}$ was added dropwise to the mixture until gas evolution ceased and the reaction mixture was poured into sat. aq. $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$ before $\mathrm{EtOAc}(25 \mathrm{~mL})$ was added. The mixture was shaken vigorously and allowed to settle before the organic layer was removed via pipette. The aqueous was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ) (via pipette) and the
combined organics were dried over $\mathrm{MgSO}_{4}$ before being concentrated in vacuo. Column chromatography over silica gel eluting with EtOAc/Petrol $(1: 1 \rightarrow 3.5: 1)$ yielded diol 21a as a viscous oil (160 mg, 73\%); $\mathrm{R}_{\mathrm{f}}=0.20(\mathrm{EtOAc} /$ Petrol, $1: 1) ;[\alpha]_{\mathrm{D}}{ }^{30}-5.95\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right)$; $v_{\max } / \mathrm{cm}^{-1}($ film $) 3356(\mathrm{OH}), 3061,2872,1449,1176,1028,920,796,740,701 ; \delta_{\mathrm{H}}(400.2$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.78(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.00-3.50(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.46(1 \mathrm{H}, \mathrm{d}, J$ 10.1, $\mathrm{C}(4) \mathrm{CHHOH}), 3.70-3.76(3 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) \mathrm{CHHOH}, \mathrm{C}(2) H, \mathrm{C}(2) \mathrm{CHHOH}), 3.92-4.00(2 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(2) \mathrm{HHOH}, \mathrm{C}(5) \mathrm{HH}), 4.07(1 \mathrm{H}, \mathrm{d}, J 12.0, \mathrm{C}(5) \mathrm{H} H), 4.11\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2} \mathrm{~N}\right), 7.21(1 \mathrm{H}, \mathrm{d}, J$ 7.5, ArH), 7.29-7.50 (9H, m, ArH), 7.77 (1H, d, J 7.7, ArH), 7.85 (1H, d, J 7.5, ArH), 7.91 $(1 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 40.7(C(3)), 44.3$ and $44.7\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 54.2$ $\left(\mathrm{PhCH}_{2} \mathrm{~N}\right), 57.7\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 61.3(C(5)), 63.3(C(2)), 63.3\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 119.8$, 120.1, 122.3, 125.5, 126.2, 126.3, 126.3, 126.4, 127.2, 127.2 (all ArCH and $p-\mathrm{PhCH}$ ), 128.5 and $128.6(o-\mathrm{PhCH}$ and $m-\mathrm{PhCH}), 139.0,139.5,141.7,142.4,144.6(\mathrm{PhC}$ and 4 x ArC$) ; m / z$ (ESI+) $384\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 406\left([\mathrm{M}+\mathrm{Na}]^{+}, 39 \%\right) ;$ HRMS $384.1960\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{26} \mathrm{H}_{26} \mathrm{NO}_{2}\right.$ requires 384.1958 ).

## (2S,3S,4R)-1-Aza-1-benzyl-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-bisaldehyde $21 b^{34}$

To a solution of oxalyl chloride ( $0.05 \mathrm{~mL}, 0.55 \mathrm{mmol}$ ) in DCM $(2.7 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was added dry DMSO ( $0.08 \mathrm{~mL}, 1.13 \mathrm{mmol}$ ), dropwise. After 15 min a solution of diol 21a ( 103 mg , $0.27 \mathrm{mmol})$ in $\mathrm{DCM}(1 \mathrm{~mL})$ was added dropwise. Then, $\mathrm{Et}_{3} \mathrm{~N}(0.45 \mathrm{~mL}, 3.23 \mathrm{mmol})$ was added and the mixture was allowed to warm to $-30{ }^{\circ} \mathrm{C}$ over 40 min . Water ( 5 mL ) was added to quench the reaction mixture. The organic layer was separated and the aqueous was extracted with DCM ( $3 \times 5 \mathrm{~mL}$ ), before the combined organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to leave a brown oil which was purified over a plug of silica gel eluting with $\mathrm{DCM} /$ Petrol $(1: 1)$ to give bis-aldehyde 21b as a viscous oil $(90 \mathrm{mg}, 88 \%) ; \delta_{\mathrm{H}}(400.2$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.28(1 \mathrm{H}, \mathrm{d}, J 10.9, \mathrm{C}(5) H \mathrm{H}), 3.54(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 4.06(1 \mathrm{H}, \mathrm{d}, J 3.0, \mathrm{C}(2) H)$,
$4.17\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2} \mathrm{~N}\right), 4.20(1 \mathrm{H}, \mathrm{d}, J 10.9, \mathrm{C}(5) \mathrm{H} H), 7.26-7.50(11 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.84(1 \mathrm{H}, \mathrm{d}, J$ 7.5, $\mathrm{Ar} H), 7.90(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 9.70(1 \mathrm{H}, \mathrm{d}, J 3.0, \mathrm{C}(2) \mathrm{CHO}), 9.80(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(4) \mathrm{CHO})$; $\delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 40.2(C(3)), 48.2\left(C\left(9^{\prime}\right)\right), 50.6(C(5)), 51.2(C(4)), 54.7\left(\mathrm{PhCH}_{2} \mathrm{~N}\right)$, $69.4(C(2)), 120.4,120.6,122.9,125.4,126.6,126.8,127.6,127.7,127.8$ (all ArCH and $p$ $\mathrm{PhCH})$, 128.7, $129.0(o-\mathrm{Ph} C H$ and $m-\mathrm{PhCH}), 137.7,139.2,140.0,141.0,142.0(\mathrm{PhC}$ and 4 x $\mathrm{ArC}), 196.1$ and $200.4(b o t h(C=\mathrm{O})) ; m / z(\mathrm{ESI}+) 379\left([\mathrm{M}+\mathrm{H}]^{+}, 71 \%\right)$.

## (3S,4R)-1-Aza-1-benzyl-4-methoxycarbonyl-2-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-

fluorene] $\mathbf{2 3}{ }^{\mathbf{8 0 , 8}} \mathbf{8 1}$
To a solution of freshly prepared bisaldehyde 21b ( $100 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) and 2-methyl-2butene ( 5 mL ) in $\mathrm{MeCN}(5 \mathrm{~mL})$ and 2-methyl-2-propanol $(8.5 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was slowly added a solution of $\mathrm{NaClO}_{2}(1.33 \mathrm{~g}, 14.71 \mathrm{mmol})$ and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(1.33 \mathrm{~g}, 11.1 \mathrm{mmol})$ in water $(2.8$ mL ). The mixture was stirred at $\mathrm{r} . \mathrm{t}$. for 1.5 h before being quenched by dropwise addition of sat. aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(6 \mathrm{~mL})$. Then, EtOAc $(10 \mathrm{~mL})$ and water $(10 \mathrm{~mL})$ were added. The organic layer was separated and the aqueous layer was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to leave a yellow oil. The resultant oil was dissolved in EtOAc ( 3 mL ) and diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo. The resultant oil was purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:3) to furnish pyrrolidinone 23 as a viscous oil ( $12 \mathrm{mg}, 12 \%$ over two steps); $\mathrm{R}_{\mathrm{f}}=0.19$ (EtOAc/Petrol, 1:3); $[\alpha]_{\mathrm{D}}{ }^{17}-176\left(\mathrm{c}=0.82, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\text {max }} / \mathrm{cm}^{-1}$ (film) $1736(\mathrm{C}=\mathrm{O}), 1698$ $(\mathrm{C}=\mathrm{O}), 1480,1438,1250,1122,911,734,703 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.53\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}\right)$, $3.60(1 \mathrm{H}, \mathrm{d}, J 1.6, \mathrm{C}(3) H), 3.78(1 \mathrm{H}, \mathrm{dd}, J 1.6,12.4, \mathrm{NC} H \mathrm{HPh}), 3.88(1 \mathrm{H}, \mathrm{d}, J 12.4$, $\mathrm{NCH} H \mathrm{Ph}), 4.18(1 \mathrm{H}, \mathrm{d}, J 14.4, \mathrm{C}(5) H \mathrm{H}), 5.02(1 \mathrm{H}, \mathrm{d}, J 14.4, \mathrm{C}(5) \mathrm{H} H), 6.73(1 \mathrm{H}, \mathrm{d}, J 7.6$,
$\mathrm{Ar} H), 7.12(1 \mathrm{H}, \mathrm{td}, J 1.2,7.6, \mathrm{ArH}), 7.23-7.28(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.34-7.46(7 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 7.75$ $(1 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{Ar} H), 7.84(1 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 38.6(C(3)), 40.7$ and $42.1(C(4)$ and $C(9)), 46.7\left(\mathrm{NCH}_{2} \mathrm{Ph}\right), 48.5(C(5)), 52.5\left(\mathrm{CH}_{3} \mathrm{O}\right), 119.7,120.6,122.1,122.6$, 127.1, 127.3, 127.7, 128.0, $128.2(\mathrm{ArCH}$ and $p-\mathrm{PhCH}), 129.0$ and $129.1(\mathrm{o}-\mathrm{PhCH}$ and $p-$ $\mathrm{PhCH}), 134.8(\mathrm{Ph} C), 139.0,139.6,141.2,142.3($ all ArC$), 167.4$ and 168.3 (both $(C=\mathrm{O})$ ); $m / z$ $(\mathrm{ESI}+) 396\left([\mathrm{M}+\mathrm{H}]^{+}, 36 \%\right)$; HRMS $418.1417\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{26} \mathrm{H}_{21} \mathrm{NNaO}_{3}\right.$ requires 418.1414).
(2S,3R,4R)-1-Aza-1-tert-butyloxycarbonyl-6,6-diphenyl-bicyclo[3.1.0]hexane-2,4-

## bismethanol 24a

$\mathrm{BH}_{3}(1 \mathrm{M}$ in THF, 2.5 mL ) was added to a solution of lactam 13a $(87 \mathrm{mg}, 0.25 \mathrm{mmol})$ in THF $(1 \mathrm{~mL})$ under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 4 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before the THF was removed in vacuo. EtOAc ( 5 mL ) was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 10 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give a white residue. The residue was dissolved in DCM ( 3 mL ) and di-tert-butyl dicarbonate ( 87 $\mathrm{mg}, 0.40 \mathrm{mmol}$ ) was added. The solution was stirred for 12 h before being concentrated in vacuo to give a white residue which was purified by column chromatography over silica gel eluting with EtOAc/Petrol (2:1) to furnish 24a as a white foam ( $63 \mathrm{mg}, 64 \%$ ); $\mathrm{R}_{\mathrm{f}}=0.22$ (EtOAc/Petrol, 2: 1); $[\alpha]_{\mathrm{D}}{ }^{25}-146.9\left(\mathrm{c}=0.5, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3397(\mathrm{OH}), 2976,1669$ $(\mathrm{C}=\mathrm{O}), 1418,1171,1125,710 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $(\mathrm{A}: \mathrm{B})=$ (6:4), $1.13\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right), 1.14\left(9 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right), 2.01-2.06\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{m}\right.$, $\left.\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{O} H^{\mathrm{A}}\right), 2.13\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right), 2.16-2.22\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{m}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{O} H^{\mathrm{B}}\right), 2.23\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}\right.$, $\left.\mathrm{C}(3) H^{\mathrm{B}}\right), 2.79-2.83\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{m}, \mathrm{C}(2) \mathrm{CH}_{2} \mathrm{O} H^{\mathrm{A}}\right), 3.40\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{dd}, J 3.2,11.5, \mathrm{C}(4) \mathrm{C} H^{\mathrm{B}} \mathrm{HOH}\right), 3.51$ $\left(1 \mathrm{H}^{\mathrm{A}}, \operatorname{dd}, J 3.6,11.5, \mathrm{C}(4) \mathrm{C} H^{\mathrm{A}} \mathrm{HOH}\right), 3.68-4.03\left(5 \mathrm{H}^{\mathrm{A}}\right.$ and $6 \mathrm{H}^{\mathrm{B}}, \mathrm{m}, \mathrm{C}(2) H^{\mathrm{B}}, \mathrm{C}(2) \mathrm{CH}_{2}{ }^{\mathrm{A}+\mathrm{B}} \mathrm{OH}$,
$\left.\mathrm{C}(5) \mathrm{CH}_{2}{ }^{\mathrm{A}+\mathrm{B}}, \mathrm{C}(4) \mathrm{H} H^{\mathrm{A}+\mathrm{B}} \mathrm{OH}\right), 4.20\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{dd}, J 3.1,5.2, \mathrm{C}(2) H^{\mathrm{A}}\right), 7.10-7.16(2 \mathrm{H}, \mathrm{m}, p-\mathrm{Ph} H, p-$ $\left.\mathrm{Ph} H^{\prime}\right), 7.20-7.39(8 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.1$ and $28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 32.3$ and $32.9(C(3)), 39.2$ and $39.8,42.7$ and $42.8\left(C(4)\right.$ and $\left(C\left(9^{\prime}\right)\right), 48.6$ and $49.0(C(5)), 59.3(C(2))$, 63.4 and $63.6\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 65.9$ and $66.2\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 79.3$ and $79.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 126.6$ and $126.7(p-\mathrm{PhCH}), 126.8\left(p-\mathrm{Ph}{ }^{\prime} \mathrm{CH}\right), 128.0$ and 128.1, 128.6, 128.7 and 128.8), 129.0 and $129.1\left(o-\mathrm{PhCH}, o-\mathrm{Ph} \cdot C H, m-\mathrm{Ph} C H\right.$ and $\left.m-\mathrm{Ph} h^{\prime} C H\right), 138.1$ and 138.3 and $141.9(\mathrm{Ph} C$ and $\left.\mathrm{Ph}^{\prime} C\right), 152.4$ and $153.8(C=\mathrm{O}) ; m / z(\mathrm{ESI}+) 813\left([\mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right), 396\left([\mathrm{M}+\mathrm{H}]^{+}, 43 \%\right)$; HRMS $418.1984\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{24} \mathrm{H}_{29} \mathrm{NNaO}_{4}\right.$ requires 418.1989).

## (2S,3R,4R)-1-Aza-1-tert-butyloxycarbonyl-6,6-di(p-methoxyphenyl)-bicyclo

## [3.1.0]hexane-2,4-bismethanol 24b

$\mathrm{BH}_{3}(1 \mathrm{M}$ in THF, 5.0 mL ) was added dropwise to a solution of $\mathbf{1 3 b}(285 \mathrm{mg}, 0.69 \mathrm{mmol})$ in THF ( 7 mL ) under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 15 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before the THF was removed in vacuo. EtOAc ( 15 mL ) was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 10 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give a white residue. The residue was dissolved in DCM ( 5 mL ) and di-tert-butyl dicarbonate (227 $\mathrm{mg}, 1.04 \mathrm{mmol}$ ) was added. The solution was stirred for 12 h before being concentrated in vacuo to give an oil which was purified by column chromatography over silica gel eluting with $\mathrm{EtOAc} /$ Petrol ( $2: 1$ ) to furnish diol $\mathbf{2 4 b}$ as a colourless viscous oil ( $123 \mathrm{mg}, 39 \%$ ); $\mathrm{R}_{\mathrm{f}}=$ $0.11(\mathrm{EtOAc} /$ Petrol, $2: 1) ;[\alpha]_{\mathrm{D}}{ }^{20}-101.2\left(\mathrm{c}=0.75, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3406(\mathrm{OH}), 2935$, 1669 (C=O), 1608, 1511, 1413, 1367, 1289, 1246, 1176, 1124, 1036, 911, 835, 733; $\delta_{\mathrm{H}}(500.3$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $(\mathrm{A}: \mathrm{B})=(3: 2) ; 1.14\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CH}_{3}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right), 1.15\left(9 \mathrm{H}^{\mathrm{A}}\right.$, $\left.\mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right), 2.06\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right), 2.15\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{B}}\right), 2.37\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{br} \mathrm{s}, \mathrm{OH}^{, \mathrm{A}}\right), 2.71$
$\left(1 \mathrm{H}^{\mathrm{B}}\right.$, br $\left.\mathrm{s}, \mathrm{OH}^{\mathrm{B}}\right), 2.85\left(1 \mathrm{H}^{\mathrm{A}}, \operatorname{br} \mathrm{s}, \mathrm{OH}^{\mathrm{A}}\right), 3.17\left(1 \mathrm{H}^{\mathrm{B}}\right.$, br s, $\left.\mathrm{OH}^{\mathrm{B}}\right), 3.38\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 11.5\right.$, $\left.\mathrm{C}(2) \mathrm{C} H^{B} \mathrm{HOH}\right), 3.50\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 11.4, \mathrm{C}(2) \mathrm{C} H^{4} \mathrm{HOH}\right), 3.65\left(11 \mathrm{H}^{\mathrm{A}}+12 \mathrm{H}^{\mathrm{B}}, \mathrm{m}, \mathrm{C}(2) H^{\mathrm{B}}\right.$, $\left.\mathrm{C}(2) \mathrm{H} H^{(\mathrm{A}+\mathrm{B})} \mathrm{OH}, \mathrm{CH}_{3}{ }_{3} \mathrm{O}, \mathrm{CH}_{3} \mathrm{O}, \mathrm{C}(5) H_{2}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 4.17\left(1 \mathrm{H}^{\mathrm{A}}\right.$, dd, $\left.J 3.4,4.8, \mathrm{C}(2) H^{\mathrm{A}}\right)$, 6.73-6.83 ( $4 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H^{(\mathrm{A}+\mathrm{B})}(o-$ to OMe$)$ ), 7.17-7.27 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H^{(\mathrm{A}+\mathrm{B})}$ ( $m$ - to OMe $)$ ); $\delta_{\mathrm{C}}(125.8$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.0$ and $28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 32.5 and $33.0(C(3))$, 39.4 and $39.9(C(4)), 41.2$ and $41.2\left(C\left(9^{\prime}\right)\right)$, 48.7 and $49.0\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 55.0\left(\mathrm{C}^{\prime} \mathrm{H}_{3} \mathrm{O}\right), 55.2\left(\mathrm{CH}_{3} \mathrm{O}\right)$, 59.3 and $59.5(\mathrm{C}(2))$, 63.2 and $63.5\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 65.7$ and $66.0(\mathrm{C}(5)), 79.3$ and $79.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 114.1$ ( $\mathrm{Ar}^{\prime} \mathrm{CH}(o-$ to OMe$)$ ), 114.1 ( $\mathrm{ArCH}(o-$ to OMe$)$ ), 128.8 and 128.9 ( $\mathrm{Ar}{ }^{\prime} \mathrm{CH}(m-$ to OMe$)$ ), 129.8 and $129.9\left(\mathrm{ArCH}(m-\right.$ to OMe$)$ ), 130.7 and $130.9\left(\mathrm{Ar}^{\prime} C(p-\right.$ to OMe$)$ ), 134.4 ( $\mathrm{ArC}(p$ - to $\mathrm{OMe})$ ), 152.6 and $153.8(C=\mathrm{O}), 158.0$ and 158.1 ( $\mathrm{ArC} C^{\prime} \mathrm{OMe}$ ), 158.1 and 158.2 ( ArCOMe ); $m / z(\mathrm{ESI}+) 456\left([\mathrm{M}+\mathrm{H}]^{+}, 92 \%\right), 912\left([2 \mathrm{M}+\mathrm{H}]^{+}, 100 \%\right) ;$ HRMS $472.2188\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, $\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NNaO}_{6}$ requires 472.2200).
(2S,3S,4R)-1-Aza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4bismethanol 24 c and (2S,3S,4R)-1-Aza-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-

## 2,4-bismethanol 25c

Method 1: A mixture of lactam 13c ( $211 \mathrm{mg}, 0.60 \mathrm{mmol}$ ) and $\mathrm{LiAlH}_{4}(110 \mathrm{mg}, 2.90 \mathrm{mmol})$ in THF ( 10 mL ) was heated at reflux under an atmosphere of $\mathrm{N}_{2}$ for 16 h . After cooling to r.t. the mixture was added dropwise to water ( 50 mL ) under a constant flow of $\mathrm{N}_{2}$ and then concentrated to remove THF. EtOAc ( 20 mL ) was added and the precipitate was filtered and washed with EtOAc ( $2 \times 15 \mathrm{~mL}$ ). The organic layer was separated and the aqueous extracted with EtOAc ( 3 x 15 mL ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo leaving an off white solid. The residual solid was dissolved in EtOAc ( 5 mL ) and di-tert-butyl dicarbonate ( $127 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was added. This mixture was stirred at r.t. for 14 h before the solvent was removed in vacuo. Purification
by column chromatography over silica gel eluting with EtOAc/Petrol (2:1 $\rightarrow 1: 1$ ) gave 24c as a white foam ( $32 \mathrm{mg}, 13 \%$ ) and $\mathbf{2 5 c}$ as a crystalline solid ( $100 \mathrm{mg}, 54 \%$ ).

Method 2: $\mathrm{BH}_{3}$ ( 1 M in THF, 17 mL ) was added to a solution of lactam $\mathbf{1 3 c}(882 \mathrm{mg}, 2.52$ mmol ) in THF ( 5 mL ) under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 25 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before THF was removed in vacuo. EtOAc ( 20 mL ) was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 15 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give a white residue. The residue was dissolved in DCM ( 20 mL ) and di-tert-butyl dicarbonate ( $881 \mathrm{mg}, 4.03 \mathrm{mmol}$ ) was added. The solution was stirred for 12 h before being concentrated in vacuo to give a white residue which was purified by column chromatography over silica gel eluting with EtOAc/Petrol (2:1 $\rightarrow 1: 1$ ) to furnish $\mathbf{2 4 c}$ as a white foam (472 $\mathrm{mg}, 48 \%$ ) and $\mathbf{2 5 c}$ as a crystalline solid ( $398 \mathrm{mg}, 51 \%$ );

24c; $\mathrm{R}_{\mathrm{f}}=0.44(\mathrm{EtOAc} /$ Petrol, $2: 1) ;[\alpha]_{\mathrm{D}}^{22}-21.4\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}(\mathrm{film}) 3385(\mathrm{OH})$, 2361, $1669(\mathrm{C}=\mathrm{O}), 1393,1170,1125,739 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $(\mathrm{A}: \mathrm{B})=(6: 4), 1.56\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right), 1.57\left(9 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right), 2.70\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right)$, $2.72\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{B}}\right), 2.98\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{br} \mathrm{s}, \mathrm{O} H^{\mathrm{A}}\right), 3.49-4.40\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{m}, \mathrm{O} H^{\mathrm{B}}\right.$ and $9 \mathrm{H}, \mathrm{m}, \mathrm{O} H$, $\left.\mathrm{C}(2) \mathrm{H}, \mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(5) \mathrm{HH}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 6.98(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar} H), 7.12-7.44(5 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H), 7.83(1 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H), 7.89(1 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.5$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 38.2$ and $39.2(C(3)), 42.9,43.2$ and 43.2, $43.3\left(C(4)\right.$ and $\left(C\left(9^{\prime}\right)\right), 52.4$ and 52.5 $\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 60.3$ and $60.5(C(2)), 60.6$ and $60.8\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 65.0(C(5)), 80.6$ and 80.7 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 119.8$ and 119.9, 120.4 and 120.4, 122.1 and $122.2,123.3$ and $123.4(4 \times \mathrm{ArCH})$, 126.5, 126.6, 126.6, 126.7, 126.7, 126.8, 127.0, 127.2 (4 x ArCH), 139.5 and 139.6, (141.0 and $141.1,141.8$ and $141.9,143.8$ and $144.1(4 \times \mathrm{ArC}), 154.2(C=\mathrm{O}) ; m / z(\mathrm{ESI}+) 416$
$\left([\mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right), 394\left([\mathrm{M}+\mathrm{H}]^{+}, 57 \%\right)$; HRMS $416.1831\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{24} \mathrm{H}_{27} \mathrm{NNaO}_{4}\right.$ requires 416.1832);

25c; $\mathrm{R}_{\mathrm{f}}=0.20(\mathrm{EtOAc}) ;$ m.p. dec. $>250{ }^{\circ} \mathrm{C}(\mathrm{EtOAc}) ;[\alpha]_{\mathrm{D}}{ }^{19}+134(\mathrm{c}=0.69, \mathrm{MeOH}) ; \delta_{\mathrm{H}}(500.3$ $\mathrm{MHz} ; d 4-\mathrm{MeOD}) 3.13(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.77(1 \mathrm{H}, \mathrm{dd}, J 6.2,11.1, \mathrm{C}(2) \mathrm{C} H \mathrm{HOH}), 3.82(1 \mathrm{H}, \mathrm{dd}$, $J 5.8,11.1, \mathrm{C}(2) \mathrm{CH} H \mathrm{OH}), 3.89(1 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{C}(4) \mathrm{CHHOH}), 4.00(1 \mathrm{H}, \mathrm{t}, J 5.8, \mathrm{C}(2) H), 4.27$ $(1 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{C}(4) \mathrm{CH} H O H), 7.25(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 7.29(1 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{Ar} H), 7.37(1 \mathrm{H}, \mathrm{t}, J$ 7.4, ArH), 7.40-7.45 (2H, m, ArH), 7.49 (1H, d, J 7.9, ArH), 7.87 (1H, d, J 7.5, ArH), 7.90 $(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 38.0(C(3)), 41.7$ and $47.0\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, 55.4 $(C(2)), 57.3\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 65.7\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 121.0,121.4,123.5,123.7,128.0,128.4$, 128.6, 128.7 (all ArCH), 140.7, 141.5, 143.1, 143.8 (all ArC), 176.7 ( $C=0$ ); $m / z$ (ESI-) 306 ([M-H] $\left.{ }^{-}, 100 \%\right)$; HRMS $330.1099\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{19} \mathrm{H}_{17} \mathrm{NNaO}_{3}\right.$ requires 330.1101).

## (2S,3S,4R,6S)-1,4'-Diaza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-

## fluorene]-2,4-bismethanol 24d

$\mathrm{BH}_{3}(1 \mathrm{M}$ in THF, 6.6 mL$)$ was added dropwise to a solution of $\mathbf{1 3 h}(330 \mathrm{mg}, 0.94 \mathrm{mmol})$ in THF ( 4 mL ) under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 15 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before the THF was removed in vacuo. EtOAc $(15 \mathrm{~mL})$ was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 10 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give a white residue. The residue was dissolved in EtOAc ( 5 mL ) and di-tert-butyl dicarbonate (308 $\mathrm{mg}, 1.41 \mathrm{mmol}$ ) was added. The solution was stirred for 12 h before being concentrated in vacuo to give an oil which was purified by column chromatography over silica gel eluting with $\mathrm{EtOAc} /$ Petrol $(2: 1)$ to furnish diol $\mathbf{2 4 d}$ as a white solid $(152 \mathrm{mg}, 41 \%) ; \mathrm{R}_{\mathrm{f}}=0.32$ (EtOAc); m.p. softens at $100{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}-69.1(\mathrm{c}=1.0, \mathrm{MeOH}) ; \delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD})$
$1.58\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.60\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 2.92(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 2.94(0.5 \mathrm{H}, \mathrm{s}$, $\mathrm{C}(3) \mathrm{H}), 3.70-4.35\left(7 \mathrm{H}, \mathrm{m}, \mathrm{C}(2) \mathrm{H}, \mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(5) \mathrm{H}_{2}\right), 7.01(1 \mathrm{H}, \mathrm{dd}, J 9.6$, 8.3, $\left.\operatorname{ArC}\left(5^{\prime}\right) H\right), 7.33\left(1 \mathrm{H}, \mathrm{dd}, J 7.8,5.1, \operatorname{ArC}\left(2^{\prime}\right) H\right), 7.36-7.40\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(6^{\prime}\right) H\right), 7.49(1 \mathrm{H}$, d, $\left.J 7.3, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.73\left(1 \mathrm{H}, \mathrm{d}, J 7.7, \operatorname{ArC}\left(1^{\prime}\right) H\right), 8.18\left(1 \mathrm{H}, \mathrm{d}, J 7.8, \operatorname{ArC}\left(8^{\prime}\right) H\right), 8.48(1 \mathrm{H}, \mathrm{d}$, $\left.\left.J 5.1, \operatorname{ArC}\left(3^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 29.0\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 39.7$ and $40.7(C(3)), 42.4$ and $42.5\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 53.6$ and $53.9\left(\mathrm{C}(4) C \mathrm{H}_{2} \mathrm{OH}\right), 59.1$ and $60.4\left(\mathrm{C}(2) C \mathrm{H}_{2} \mathrm{OH}\right), 62.2$ and $62.3(C(2)), 63.5$ and $64.3(C(5)), 82.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 122.3\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), 122.7\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right)$, $124.4\left(\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 128.4\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 130.3$ and 130. $\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 132.0\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 140.6$ and 140.7, 142.0, 143.9 and $144.0(\operatorname{all} \mathrm{ArC}), 148.0\left(\mathrm{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 155.3$ and $155.4(\mathrm{ArC}), 159.2$ and $159.3(C=O) ; m / z(E S I+) 395\left([M+H]^{+}, 100 \%\right) ;$ HRMS $395.1961\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{23} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{4}\right.$ requires 395.1965 ).

## (2S,3S,4R,6R)-1,5'-Diaza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-

 fluorene]-2,4-bismethanol 24 e and $\quad(2 S, 3 S, 4 R, 6 R)-1,5$ '-Diaza-5-oxo-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-bismethanol 25e$\mathrm{BH}_{3}(1 \mathrm{M}$ in THF, 6 mL$)$ was added dropwise to a solution of $\mathbf{1 3 i}(308 \mathrm{mg}, 0.88 \mathrm{mmol})$ in THF ( 5 mL ) under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 15 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before the THF was removed in vacuo. EtOAc ( 15 mL ) was added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 10 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give a white residue. The residue was dissolved in a small quantity of $\mathrm{MeOH}(0.5 \mathrm{~mL})$ and EtOAc (3 mL ) was added. The MeOH was removed slowly in vacuo at $40^{\circ} \mathrm{C}$, leaving a precipitate in the EtOAc solution. The precipitate was allowed to settle and the EtOAc was removed via pipette, before being washed with more EtOAc ( $2 \times 3 \mathrm{~mL}$ ) and dried in vacuo to leave 25e ( 60
$\mathrm{mg}, 22 \%$ ) as a white powder. The combined EtOAc washes were concentrated in vacuo, leaving an oil, to which, EtOAc ( 4 mL ) and di-tert-butyl dicarbonate ( $164 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) were added. The solution was stirred for 12 h before being concentrated in vacuo to give an oil which was purified by column chromatography over silica gel eluting with EtOAc/Petrol ( $2: 1$ ) to furnish diol $\mathbf{2 4 e}$ as a wax ( $77 \mathrm{mg}, 22 \%$ );

24e; $\mathrm{R}_{\mathrm{f}}=0.51(\mathrm{EtOAc}) ;[\alpha]_{\mathrm{D}}{ }^{25}-12.1\left(\mathrm{c}=0.65, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3375(\mathrm{brOH}), 1691$ $(\mathrm{C}=\mathrm{O}), 1393,1163,1122,746 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.54\left(3 / 7 \mathrm{x} \mathrm{9H}, \mathrm{~s},\left(\mathrm{CH}_{3}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right)$, $1.57\left(4 / 7 \times 9 H, s,\left(\mathrm{CH}_{3}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right), 2.73\left(4 / 7 \times 1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right), 2.76\left(3 / 7 \times 1 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{B}}\right), 3.87-$ 4.37 (7H, m, C(2)H, C(2) $\left.\mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(5) \mathrm{H}_{2}\right), 7.07-7.12\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 7.19-$ $7.26\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}, \operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 7.39-7.45\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}, \operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 8.07-8.11(1 \mathrm{H}$, $\left.\left.\mathrm{m}, \operatorname{ArC}\left(4^{\prime}\right) \mathrm{H}\right), 8.50-8.54\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 28.5\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 38.3 and $39.3(C(3)), 41.3$ and 41.4 and $43.7\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, 52.7, and $52.8\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right)$, 60.4, $60.6\left(C(2)\right.$ and $\left.\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 64.8$ and $\left.64.9(C(5)), 80.9\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 121.0,121.2\right)$ $\left(\operatorname{ArC}\left(4^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 122.0$ and $122.1\left(\operatorname{ArC}\left(8^{\prime}\right.\right.$ or $\left.\left.2^{\prime}\right) \mathrm{H}\right), 127.2$ and 127.3 and 128.6 $\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right.$ and $\left.\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 130.7$ and $130.9\left(\mathrm{ArC}\left(8^{\prime}\right.\right.$ or $\left.\left.2^{\prime}\right) \mathrm{H}\right), 135.4$ and $135.5,138.8$ and 138.9, 144.5, 144.7 (All $\operatorname{ArC}), 147.4\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right),(153.4,154.0)(\mathrm{ArC}),(160.2,160.2)(C=\mathrm{O})$; $m / z(\mathrm{ESI}+) 395\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right) ;$ HRMS $395.1962\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{23} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{4}\right.$ requires 395.1965); 25e; m.p. dec. $>20{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{25}+74(\mathrm{c}=1, \mathrm{MeOH}) ; \delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 3.22(1 \mathrm{H}, \mathrm{s}$, C(3)H), $3.75(1 \mathrm{H}, \mathrm{dd}, J 6.2,10.9, \mathrm{C}(2) \mathrm{HHOH}), 3.80(1 \mathrm{H}, \mathrm{dd}, J 5.7,10.9, \mathrm{C}(2) \mathrm{HHOH}), 3.88$ $(1 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{C}(4) H \mathrm{HOH}), 3.95(1 \mathrm{H}, \mathrm{t}, J 5.9, \mathrm{C}(2) H), 4.30(1 \mathrm{H}, \mathrm{d}, J 12.6 \mathrm{C}(4) \mathrm{HHOH}), 7.30$ (1H, dd, $J$ 5.0, 8.0, $\left.\operatorname{ArC}\left(2^{\prime}\right) H\right), 7.35-7.39\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(8^{\prime}\right) H\right), 7.49-7.54\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(6^{\prime}\right) H\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) H\right), 7.91\left(1 \mathrm{H}, \mathrm{dd}, J 1.3,8.0, \operatorname{ArC}\left(1^{\prime}\right) H\right), 8.10-8.14\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(5^{\prime}\right) H\right), 8.52(1 \mathrm{H}, \mathrm{dd}, J$ 1.3, 5.0, $\left.\operatorname{ArC}\left(3^{\prime}\right) H\right) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 38.3(C(3)), 40.0$ and $47.1\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right)$, $55.8(C(2)), 57.1\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 65.8\left(\mathrm{C}(2) \mathrm{CH}_{2} \mathrm{OH}\right), 122.1\left(\mathrm{ArC}\left(5^{\prime}\right) H\right), 122.9\left(\mathrm{ArC}\left(2^{\prime}\right) H\right)$, $124.0\left(\operatorname{ArC}\left(8^{\prime}\right) H\right), 128.8$ and $130.3\left(\operatorname{ArC}\left(6^{\prime}\right) H\right.$ and $\left.\operatorname{ArC}\left(7^{\prime}\right) H\right), 132.2\left(\operatorname{ArC}\left(1^{\prime}\right) H\right), 136.9$,
139.4, 145.1 (all $\operatorname{ArC}$ ), $148.7\left(\operatorname{ArC}\left(3^{\prime}\right) H\right), 161.1$ ( $\operatorname{ArC),176.0(C=O);~m/z~(ESI-)~} 307$ ([M-H] ${ }^{-}$ , 100\%); HRMS $309.1234\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}_{3}\right.$ requires 309.1234).

## (2S,3R,4R)-1-Aza-1-tert-butyloxycarbonyl-6,6-diphenyl-bicyclo[3.1.0]hexane-2-methanol

 24 g$\mathrm{BH}_{3}(1 \mathrm{M}$ in THF, 2 mL ) was added dropwise to a solution of $\mathbf{1 3 k}(79 \mathrm{mg}, 0.28 \mathrm{mmol})$ in THF ( 2 mL ) under an atmosphere of $\mathrm{N}_{2}$. The mixture was then heated at reflux for 2 d before cooling to r.t. $\mathrm{NaOH}(1 \mathrm{M}, 10 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 10 min before the THF was removed in vacuo. EtOAc $(10 \mathrm{~mL})$ and water $(5 \mathrm{~mL})$ were added and the organic layer was separated. The aqueous layer was extracted with EtOAc ( $4 \times 12 \mathrm{~mL}$ ) before the combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo to give crude secondary amine as a white residue; $m / z(\mathrm{ESI}+) 267\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS $266.1539\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{18} \mathrm{H}_{20} \mathrm{NO}\right.$ requires 266.1539); The residue was dissolved in DCM ( 3 mL ) and di-tert-butyl dicarbonate ( $93 \mathrm{mg}, 0.43 \mathrm{mmol}$ ) was added. The solution was stirred for 12 h before being concentrated in vacuo to give an oil which was purified by column chromatography over silica gel eluting with $\operatorname{EtOAc} /$ Petrol $(1: 3)$ to furnish $\mathbf{2 4 g}$ as a colourless gum ( $85 \mathrm{mg}, 82 \%$ ); $\mathrm{R}_{\mathrm{f}}=0.15(\mathrm{EtOAc} /$ Petrol, $1: 3) ;[\alpha]_{\mathrm{D}}{ }^{25}-25\left(\mathrm{c}=0.5, \mathrm{CHCl}_{3}\right)$; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.14\left(5.4 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.15\left(3.6 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.59-2.36(3 \mathrm{H}$, m, C(3)H, C(4)H, OH), 3.52-3.58 (1H, m, C(5)HH), 3.69-3.77 (1.4H, m, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), ~ 3.69-3.77$ $(0.6 \mathrm{H}, \mathrm{m}, \mathrm{C}(5) \mathrm{H} H), 3.79-3.86\left(0.6 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.90(0.4 \mathrm{H}, \mathrm{d}, J 11.7, \mathrm{C}(5) \mathrm{H} H), 3.99(0.4 \mathrm{H}$, t, $J 5.2, \mathrm{C}(2) H), 4.24(0.6 \mathrm{H}, \mathrm{dd}, J 4.0,7.7, \mathrm{C}(2) H), 7.05-7.14(3 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H), 7.17-7.26(3 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph} H), 7.31-7.39(4 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H) ; \delta_{C}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.0$ and 28.1$)\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 30.3$ and $30.4(C(4)), 33.2$ and $33.7(C(3)), 35.7$ and $35.8(C(6)), 46.1$ and $46.2(C(5)), 59.6$ and 59.7 $(C(2)), 65.1$ and $66.2\left(\mathrm{CH}_{2} \mathrm{OH}\right), 79.5$ and $79.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 125.9,126.7,127.0,128.3,128.6$, 129.1, 129.2 (all PhCH$), 136.0$ and $136.3\left(\mathrm{Ph}^{\prime} C\right), 145.2$ and $145.4(\mathrm{PhC}), 152.6$ and 154.7
$\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR})\right) ; m / z(\mathrm{ESI}+) 388\left([\mathrm{M}+\mathrm{Na}]^{+}, 82 \%\right), 366\left([\mathrm{M}+\mathrm{H}]^{+}, 71 \%\right) ;$ HRMS 388.1877 $\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NNaO}_{3}\right.$ requires 388.1883).
(2S,3S,4R)-1-Aza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4biscarboxylic acid 26a ${ }^{53}$

To a solution of ester $\mathbf{2 6 b}(40 \mathrm{mg}, 0.09 \mathrm{mmol})$ in $\mathrm{MeOH}(3 \mathrm{~mL})$ and water $(1 \mathrm{~mL})$ was added lithium hydroxide ( $37 \mathrm{mg}, 0.89 \mathrm{mmol}$ ). After stirring at r.t. for 2 d , MeOH was removed in vacuo. EtOAc ( 5 mL ) was added and the organic layer was separated and concentrated to give unwanted organic material. Then, EtOAc ( 5 mL ) and $\mathrm{HCl}(0.25 \mathrm{M}, 10 \mathrm{~mL})$ were added sequentially to the aqueous mixture. The organic layer was separated and the aqueous was extracted with EtOAc ( $3 \times 5 \mathrm{~mL}$ ) before the combined organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give diacid 26a as a white foam ( $36 \mathrm{mg}, 96 \%$ ) which required no further purification; $[\alpha]_{\mathrm{D}}{ }^{19}+489\left(\mathrm{c}=2, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 2980(\mathrm{COO}-\mathrm{H}), 1708(\mathrm{C}=\mathrm{O})$, $1396,1259,1180,1146,738 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.48\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.53(4.5 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 3.06(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.28(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 4.01(0.5 \mathrm{H}, \mathrm{d}, J 12.8, \mathrm{C}(5) H \mathrm{H}), 4.17$ $(0.5 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{C}(5) H \mathrm{H}), 4.25(0.5 \mathrm{H}, \mathrm{d}, J 12.8, \mathrm{C}(5) \mathrm{H} H), 4.28(0.5 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{C}(5) \mathrm{H} H)$, $4.51(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 4.69(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.02\left(0.5 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{Ar} H^{\prime}\right), 7.10(0.5 \mathrm{H}, \mathrm{d}, J 7.7$, $\left.\mathrm{Ar} H^{\prime}\right), 7.12-7.44(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.72$ ( $0.5 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{ArH}^{\prime}$ '), 7.75 ( $0.5 \mathrm{H}, \mathrm{d}, J 7.2, \operatorname{Ar} H^{\prime}$ '), 7.84 $\left(1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{Ar} H^{\prime \prime}\right)$, $9.0\left(2 \mathrm{H}, \mathrm{br}, 2 \times \mathrm{CO}_{2} H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.2$ and 28.4 $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 37.9$ and $39.1(C(3))$, 43.4, 44.0, 44.2, $44.3\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 49.8$ and 49.9 $(C(5)), 60.5$ and $60.9(C(2)), 82.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 119.4$ and $119.6(\mathrm{ArCH} '), 120.7\left(\mathrm{ArCH}{ }^{\prime \prime}\right)$, 121.9 and $122.2(\mathrm{ArCH}), 123.5$ and $123.8\left(\mathrm{ArCH}^{\prime}\right), 127.2$ and $127.3,127.5$ and $127.7,127.6$ $(4 \times \operatorname{ArCH}), 138.5$ and $138.6,139.2$ and 139.3, 142.5 and $142.8(4 \times \mathrm{ArC}), 153.3$ and 154.0 $\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR})\right), 171.6$ and $171.8\left(\mathrm{CO}_{2} \mathrm{H}\right), 174.0$ and $174.8\left(C^{\prime} \mathrm{O}_{2} \mathrm{H}\right) ; m / z(\mathrm{ESI}-) 420([\mathrm{M}-$ $\mathrm{H}]^{-}, 100 \%$ ); HRMS 420.1455 ([M-H] $]^{-}, \mathrm{C}_{24} \mathrm{H}_{22} \mathrm{NO}_{6}$ requires 420.1453).
(2S,3S,4R)-1-Aza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-

## bismethyl-ester 26b

A mixture of alcohol $\mathbf{2 4 c}(300 \mathrm{mg}, 0.76 \mathrm{mmol})$, TEMPO ( $36 \mathrm{mg}, 0.23 \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{CN}(7.6$ mL ), and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(7.6 \mathrm{~mL}, 0.67 \mathrm{M})$ was warmed to $45^{\circ} \mathrm{C}$ and allowed to cool to $37{ }^{\circ} \mathrm{C}$. Then, $\mathrm{NaClO}_{2}$ ( $414 \mathrm{mg}, 4.57 \mathrm{mmol}$ ) was added followed by 3 drops of $10-13 \% \mathrm{aq} . \mathrm{NaOCl}$. The solution was maintained at $37^{\circ} \mathrm{C}$ for 20 h and then poured over an ice-cold saturated solution of sodium thiosulfate $(10 \mathrm{~mL})$. The resulting mixture was extracted with EtOAc (4 x 10 mL ) and the combined layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to yield the crude diacid 26a. A portion of the crude diacid 26a ( $39 \mathrm{mg}, 93 \mu \mathrm{~mol}$ ) was dissolved in EtOAc ( 3 mL ) and diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo. The resultant oil was purified by column chromatography over silica gel eluting with DCM/Petrol (1:1) to furnish ester 26b as a white wax ( $41 \mathrm{mg}, 34 \%$ over two steps); $\mathrm{R}_{\mathrm{f}}=$ $0.14(\mathrm{DCM} /$ Petrol, $1: 1) ;[\alpha]_{\mathrm{D}}{ }^{21}-68.4\left(\mathrm{c}=0.8, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 1740(2 \times(\mathrm{C}=\mathrm{O}))$, $1704(\mathrm{C}=\mathrm{O})$; 1449, 1386, 1268, 1205, 1121, 737; $\delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $\left.(\mathrm{A}: \mathrm{B})=(11: 10) ; 1.50\left(9 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}{ }_{3}\right)_{3} \mathrm{CO}\right), 1.57\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{B}}{ }_{3}\right)_{3} \mathrm{CO}\right),\right)^{3.15}$ $\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right), 3.18\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{B}}\right), 3.58\left(3 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(4) \mathrm{CO}_{2} \mathrm{C} H_{3}^{\mathrm{B}}\right), 3.58\left(3 \mathrm{H}^{\mathrm{A}}, \mathrm{s}\right.$, $\left.\mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{A}}\right), 3.86\left(3 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{A}}\right), 3.87\left(3 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{B}}\right), 4.04\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J\right.$ 12.1, $\left.\mathrm{C}(5) H^{\mathrm{B}} \mathrm{H}\right), 4.11\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 12.2, \mathrm{C}(5) H^{\mathrm{A}} \mathrm{H}\right), 4.25\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 12.1, \mathrm{C}(5) \mathrm{H}^{\mathrm{B}} H\right), 4.30\left(1 \mathrm{H}^{\mathrm{A}}\right.$, $\left.\mathrm{d}, J 12.2, \mathrm{C}(5) \mathrm{H}^{\mathrm{A}} H\right), 4.58\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{A}}\right), 4.70\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{B}}\right), 7.11\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 7.9\right.$, $\left.\operatorname{Ar} H^{\mathrm{B}}\right), 7.14\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 7.9, \operatorname{Ar} H^{\mathrm{A}}\right), 7.23-7.48(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.79(1 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar} H), 7.90$ $(1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.2$ and $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 37.9$ and $38.8(C(3))$, 43.4, 43.5, 43.7, $44.5\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 49.5$ and $49.6(C(5)), 52.4,52.5,52.6,52.8$ $\left(\mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}, \mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 60.2$ and $60.6(\mathrm{C}(2)), 81.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 119.6$ and 119.6, 120.7,
121.7 and $121.8,123.7$ and 123.7, 127.1 and 127.1, 127.3 and 127.5, 127.4, $127.7(\mathrm{all} \mathrm{ArCH})$, 138.3 and $138.4,139.4,142.2$ and $142.3,142.6$ and 142.6 (all ArC), 152.6 and 153.1 $\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR}), 167.5\right.$ and $167.5\left(\mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 171.0$ and $171.3\left(\mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}$ (ESI + ) $450\left([\mathrm{M}+\mathrm{H}]^{+}, 77 \%\right)$; HRMS $472.1722\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{26} \mathrm{H}_{27} \mathrm{NNaO}_{6}\right.$ requires 472.1731).

## (2S,3S,4R)-1-Aza-spiro[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-biscarboxylic acid 27

A solution of diacid 26a in TFA ( 2 mL ) was stirred at room temperature for 3 h before being concentrated in vacuo to give a viscous brown oil. The oil was washed with EtOAc ( 2 mL ) and allowed to stand until a white precipitate had settled. The organic solution was removed via pipette to give amino acid analogue 27 as a white solid ( $9 \mathrm{mg}, 59 \%$ ); m.p. dec. $>235^{\circ} \mathrm{C}$; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3} / d-\mathrm{TFA}(6: 1)\right) 3.82(1 \mathrm{H}, \mathrm{d}, J 2.5, \mathrm{C}(3) H), 4.40(1 \mathrm{H}, \mathrm{d}, J 13.9$, $\mathrm{C}(5) H \mathrm{H}), 4.68(1 \mathrm{H}, \mathrm{d}, J 13.9, \mathrm{C}(5) \mathrm{H} H), 5.28(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.12(1 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar} H), 7.34-$ 7.38 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ), 7.39-7.43 (2H, m, ArH), 7.48-7.52 (1H, m, $\mathrm{Ar} H), 7.60-7.64$ (1H, m, $\mathrm{Ar} H)$, $7.86(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H), 8.01(1 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3} / d-\mathrm{TFA}(6: 1)\right) 38.7$ $(C(3)), 44.6$ and $48.5\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 49.9(C(5)), 60.8(C(2)), 120.5,122.2,122.9,123.1$, 127.4, 127.9, 129.3, 129.8 (All ArCH), 136.4, 138.8, 140.0, 143.3 (All ArC), 171.9 ( $C=0$ ) (one ( $C=O$ ) missing - obscured by $d$-TFA); $m / z$ (ESI-) 320 ([M-H], 100\%); HRMS 320.0930 ([M-H], $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{NO}_{4}$ requires 320.0928).

## (2S,3S,4R,6S)-1,4'-Diaza-1-tert-butyloxycarbonyl-spiro[bicyclo[3.1.0]hexane-6,9'-

fluorene]-2,4-bismethyl-ester 28a and (2S,3S,4R,6S)-1,4'-diaza-1-tert-butyloxycarbonyl-

## 4-hydroxymethyl-spiro[bicyclo [3.1.0]hexane-6,9'-fluorene]-2-methyl-ester 28b

A mixture of alcohol $\mathbf{2 4 a}(62 \mathrm{mg}, 0.16 \mathrm{mmol})$, TEMPO ( $13 \mathrm{mg}, 0.08 \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{CN}(3.1$ $\mathrm{mL})$, and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(2.7 \mathrm{~mL}, 0.67 \mathrm{M})$ was warmed to $45{ }^{\circ} \mathrm{C}$ and allowed to cool to $37{ }^{\circ} \mathrm{C}$. Then, $\mathrm{NaClO}_{2}(110 \mathrm{mg}, 1.22 \mathrm{mmol})$ was added followed by 6 drops of $8 \%$ aq. NaOCl . The solution was maintained at $37{ }^{\circ} \mathrm{C}$ for 16 h and then poured over an ice-cold solution of
$\mathrm{Na}_{2} \mathrm{SO}_{3}(1.6 \mathrm{~g})$ in water $(1.6 \mathrm{~mL})$ and ice $(3.4 \mathrm{~g})$. The resulting mixture was extracted with EtOAc (4 x 5 mL ) and the combined layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo, before being dissolved in EtOAc ( 3 mL ). Diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo. THF ( 3 mL ) was added to the aqueous layer which was then stirred vigorously before diazomethane (approx. 3 mmol ) was distilled directly into the mixture. After addition was complete, the THF was removed in vacuo before the aqueous was extracted with EtOAc ( $3 \times 5 \mathrm{~mL}$ ). The combined organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo. Both fractions were combined and purified by column chromatography over silica gel eluting with EtOAc/Petrol (1:1) to give diester 28a as a white wax ( $3 \mathrm{mg}, 4 \%$ ), and 28b as a clear viscous oil ( $20 \mathrm{mg}, 30 \%$ );

28a; $\mathrm{R}_{\mathrm{f}}=0.64\left(\mathrm{EtOAc} /\right.$ Petrol, 2:1); $\mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2962, 2360, 2341, $1739(\mathrm{C}=\mathrm{O}), 1705$ $(\mathrm{C}=\mathrm{O}), 1438,1387,1260,1209,1175,1019,740 ; \delta_{\mathrm{H}}\left(400.2 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.49(4.5 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}^{\mathrm{A}}{ }_{3}\right)_{3} \mathrm{CO}\right), 1.56\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{C}_{3}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right)$, , $3.21\left(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{A}}\right), 3.24\left(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H^{\mathrm{B}}\right)$, $3.59\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{A}}\right), 3.60\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{B}}\right), 3.86\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{B}}\right)$, $3.87\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}{ }^{\mathrm{A}}\right), 4.01\left(0.5 \mathrm{H}, \mathrm{d}, J 12.3, \mathrm{C}(5) H^{\mathrm{A}} \mathrm{H}\right), 4.09(0.5 \mathrm{H}, \mathrm{d}, J 12.4$, $\left.\mathrm{C}(5) H^{\mathrm{B}} \mathrm{H}\right), 4.28\left(0.5 \mathrm{H}, \mathrm{d}, J 12.3, \mathrm{C}(5) \mathrm{H} H^{\mathrm{A}}\right), 4.34\left(0.5 \mathrm{H}, \mathrm{d}, J 12.4, \mathrm{C}(5) \mathrm{H} H^{\mathrm{B}}\right), 4.59(0.5 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}(2) H^{\mathrm{A}}\right), 4.71\left(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{B}}\right), 7.15-7.25\left(2 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(2^{\prime}\right) H, \operatorname{ArC}\left(5^{\prime}\right) H\right), 7.39-7.49(1 \mathrm{H}, \mathrm{m}$, $\left.\operatorname{ArC}\left(6^{\prime}\right) H\right), 7.54-7.60\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.75\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \operatorname{ArC}\left(1^{\prime}\right) H\right), 8.39(1 \mathrm{H}, \mathrm{br}$ s, $\left.\operatorname{ArC}\left(8^{\prime}\right) H\right), 8.58\left(1 \mathrm{H}, \mathrm{d}, J 4.9, \operatorname{ArC}\left(3^{`}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.2$ and $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 37.9 and $38.8(C(3)), 42.2,42.4,43.9$ and $44.5\left(C(4)\right.$ and $\left.\left.C\left(9^{\prime}\right)\right), 49.1(C(5)), 52.8,52.9\right)$ $\left(\mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}\right.$ and $\left.\mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 60.0$ and $60.4(C(2)), 81.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 121.4$ and 123.8, $123.9\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right.$ and $\left.\left.\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 128.7\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right)\right), 152.5$ and $153.0\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR}), 167.3\right.$ and $167.4\left(\mathrm{C}(4) \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 170.7$ and $171.0\left(\mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}(\mathrm{ESI}+) 451\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS $451.1879\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{25} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{6}\right.$ requires 451.1864).

28b; $\mathrm{R}_{\mathrm{f}}=0.12\left(\mathrm{EtOAc} /\right.$ Petrol, 2:1); $[\alpha]_{\mathrm{D}}{ }^{25}-33\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}($ film $) 3200(\mathrm{OH})$, $3000,1750,1699,1406,1175,1126,739 ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.49\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $1.59\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 2.71(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 2.72(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.64(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $3.83\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.83\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.07-4.21\left(3 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}, \mathrm{C}(5) \mathrm{HH}\right)$, $4.34(0.5 \mathrm{H}, \mathrm{d}, J 12.1 \mathrm{C}(5) \mathrm{H} H), 4.37(0.5 \mathrm{H}, \mathrm{d}, J 12.1 \mathrm{C}(5) \mathrm{H} H), 4.59(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 4.69$ ( $0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.05\left(0.5 \mathrm{H}, \mathrm{d}, J 4.9, \operatorname{ArC}\left(2^{`}\right) H\right), 7.07\left(0.5 \mathrm{H}, \mathrm{d}, J 4.9, \operatorname{ArC}\left(2^{`}\right) H\right), 7.07(1 \mathrm{H}$, m, $\operatorname{Ar}\left(5^{\prime}\right) H$ ), 7.24-7.31 (2H, m, $\left.\operatorname{ArC}\left(6^{\prime}\right) H, \operatorname{ArC}\left(7^{\prime}\right) H\right), 7.35-7.38\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(1^{\prime}\right) H\right), 7.91-$ $7.95\left(1 \mathrm{H}, \mathrm{m}, \operatorname{ArC}\left(8^{\prime}\right) H\right), 8.28\left(1 \mathrm{H}, \mathrm{dd}, J 1.1,4.9, \operatorname{ArC}\left(3^{\prime}\right) H\right) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.3$ and $\left.28.5\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 37.9$ and $38.3(C(3)), 41.0,43.8,44.5\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 60.7$ and 61.1 $(C(2)), 61.1$ and $61.2(C(5)), 77.2\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 81.1$ and $81.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 120.2$ and 120.3 $\left(\operatorname{ArC}\left(2^{\prime}\right) \mathrm{H}\right), \quad 121.6 \quad\left(\operatorname{ArC}\left(8^{\prime}\right) \mathrm{H}\right), 123.1 \quad\left(\operatorname{ArC}\left(5^{\prime}\right) \mathrm{H}\right), 127.5 \quad\left(\operatorname{ArC}\left(7^{\prime}\right) \mathrm{H}\right), 129.2,129.3$ $\left(\operatorname{ArC}\left(1^{\prime}\right) \mathrm{H}\right), 129.3129 .5\left(\operatorname{ArC}\left(6^{\prime}\right) \mathrm{H}\right), 137.6,140.6,140.7$ and $140.8(\operatorname{all} \mathrm{ArC}), 147.3$ $\left(\operatorname{ArC}\left(3^{\prime}\right) \mathrm{H}\right), 152.7$ and $153.4\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR})\right), 158.3(\mathrm{ArC}), 172.1$ and $172.2\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right) ; m / z$ (ESI + ) $423\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right)$; HRMS $423.1909\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{24} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{5}\right.$ requires 423.1914).

## Attempted Synthesis of ( $\mathbf{2 S , 3 S}, \mathbf{4 R , 6 R}$ )-1,5'-diaza-1-tert-butyloxycarbonyl-spiro

[bicyclo[3.1.0]hexane-6,9'-fluorene]-2,4-bismethyl-ester 28d and (2S,3S,4R,6R)-1,5'-diaza-1-tert-butyloxycarbonyl-4-hydroxymethyl-spiro[bicyclo [3.1.0]hexane-6,9'-fluorene]-2-methyl-ester 28 .

A mixture of alcohol $\mathbf{2 4 b}(53 \mathrm{mg}, 0.13 \mathrm{mmol})$, TEMPO ( $12 \mathrm{mg}, 0.08 \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{CN}(2.6$ mL ), and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(2.3 \mathrm{~mL}, 0.67 \mathrm{M})$ was warmed to $45^{\circ} \mathrm{C}$ and allowed to cool to $37{ }^{\circ} \mathrm{C}$. $\mathrm{NaClO}_{2}(97 \mathrm{mg}, 1.07 \mathrm{mmol})$ was added followed by 9 drops of $8 \%$ aq. NaOCl . The solution was maintained at $37^{\circ} \mathrm{C}$ for 16 h and then poured over an ice-cold solution of $\mathrm{Na}_{2} \mathrm{SO}_{3}(1.4 \mathrm{~g})$ in water $(1.4 \mathrm{~mL})$ and ice $(2.9 \mathrm{~g})$. The resulting mixture was extracted with EtOAc ( $4 \times 5 \mathrm{~mL}$ ) and the combined layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo, before being
dissolved in EtOAc ( 3 mL ). Diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo. THF ( 3 mL ) was added to the aqueous layer which was then stirred vigorously before diazomethane (approx. 3 mmol ) was distilled direcly into the mixture. After addition was complete, the THF was removed in vacuo before the aqueous was extracted with EtOAc ( $3 \times 5 \mathrm{~mL}$ ). The combined organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated in vacuo. Both fractions were combined and purified by column chromatography over silica gel eluting with $\mathrm{EtOAc} /$ Petrol $(1: 2)$ to give ester 28e as a clear viscous oil $(4 \mathrm{mg}, 7 \%) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3385 (br OH), 2977, 1750 (C=O), 1700 (C=O), 1393, 1207, 1369, 800; $\delta_{\mathrm{H}}(400.2 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 1.49\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.57\left(4.5 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 2.80(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 2.82$ $(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(3) H), 3.85\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.85\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.96(0.5 \mathrm{H}, \mathrm{d}, J$ 12.0, $\mathrm{C}(5) H \mathrm{H}), 4.03(0.5 \mathrm{H}, \mathrm{d}, J 12.2, \mathrm{C}(5) H \mathrm{H}), 4.08-4.23\left(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 4.30(0.5 \mathrm{H}, \mathrm{d}, J$ 12.0, C(5)HH), $4.34(0.5 \mathrm{H}, \mathrm{d}, J 12.2, \mathrm{C}(5) \mathrm{H} H), 4.56(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 4.67(0.5 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H)$, 7.12-7.20 (2H, m, ArH), 7.40-7.47 (3H, m, ArH), 8.08-8.18 (1H, m, ArH), 8.55 ( $0.5 \mathrm{H}, \mathrm{dd}, J$ 1.6, 5.3, $\left.\operatorname{ArC}\left(8^{`}\right) H\right), 8.57\left(0.5 \mathrm{H}, \mathrm{dd}, J 1.6,5.3, \operatorname{ArC}\left(6{ }^{\prime}\right) H\right) ; \delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.3$ and $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 38.8 and $39.6(C(3)), 41.1$ and 41.2, 43.5 and $44.2\left(C(4)\right.$ and $\left.C\left(9^{\prime}\right)\right), 52.3$ and $52.4(C(5))$, 52.7 and $52.8\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 60.4$ and $60.6\left(\mathrm{C}(4) \mathrm{CH}_{2} \mathrm{OH}\right), 61.0$ and $61.3(\mathrm{C}(2))$, $81.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 121.3,121.5,121.7,121.9,121.9,127,7,129.0($ all ArCH$), 131.5,144.0$, 147.0, 159.8 (all ArC ), 152.9 and $153.4\left(\mathrm{C}(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR})\right), 171.9$ and $172.0\left(\mathrm{C}(2) \mathrm{CO}_{2} \mathrm{CH}_{3}\right)$; $m / z(\mathrm{ESI}+) 867\left([2 \mathrm{M}+\mathrm{Na}]^{+}, 100 \%\right), 423\left([\mathrm{M}+\mathrm{H}]^{+}, 63 \%\right) ;$ HRMS $445.1740\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, $\mathrm{C}_{24} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{NaO}_{5}$ requires 445.1734).
(2S,3R,4R)-1-Aza-1-tert-butyloxycarbonyl-6,6-diphenyl-bicyclo[3.1.0]hexane-2-methylester 29a ${ }^{54}$

A mixture of alcohol $\mathbf{2 4 g}(85 \mathrm{mg}, 0.23 \mathrm{mmol})$, TEMPO ( $6 \mathrm{mg}, 0.04 \mathrm{mmol}), \mathrm{CH}_{3} \mathrm{CN}(1.8 \mathrm{~mL})$, and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(1.6 \mathrm{~mL}, 0.67 \mathrm{M})$ was warmed to $37{ }^{\circ} \mathrm{C}$. Then, $\mathrm{NaClO}_{2}(63 \mathrm{mg}, 0.70 \mathrm{mmol})$ was added followed by 3 drops of $8 \%$ aq. NaOCl . The solution was maintained at $37^{\circ} \mathrm{C}$ for 5 h and then poured over an ice-cold solution of $\mathrm{Na}_{2} \mathrm{SO}_{3}(0.25 \mathrm{~g}$ in 0.5 mL water and 1 g of ice). The resulting mixture was extracted with EtOAc ( $4 \times 5 \mathrm{~mL}$ ) and the combined layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was dissolved in EtOAc ( 10 mL ) and diazomethane (approx. 3 mmol ) was distilled directly into the solution. After 0.5 h , the excess diazomethane was quenched with AcOH and the solvents were removed in vacuo. The resultant oil was purified by column chromatography over silica gel eluting with DCM to furnish ester 29a as a white wax ( $62 \mathrm{mg}, 68 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.14(\mathrm{DCM}) ;[\alpha]_{\mathrm{D}}{ }^{20}-8.39(\mathrm{c}=1.55$, $\mathrm{CHCl}_{3}$ ); $v_{\max } / \mathrm{cm}^{-1}$ (film) 2976, 1752, 1702, 1405, 1174, 1113, 760, 709; $\delta_{\mathrm{H}}(400.2 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $(\mathrm{A}: \mathrm{B})=(5: 4), 1.14\left(9 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right), 1.15\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s}\right.$, $\left.\left(\mathrm{CH}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right), 2.20-2.29(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(4) H), 2.31\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 7.3, \mathrm{C}(3) H^{\mathrm{B}}\right), 2.37\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 7.3\right.$, $\left.\mathrm{C}(3) H^{\mathrm{A}}\right), 3.66-3.74\left(1 \mathrm{H}^{\mathrm{A}}\right.$ and $\left.2 \mathrm{H}^{\mathrm{B}}, \mathrm{m}, \mathrm{C}(5) H \mathrm{H}, \mathrm{C}(5) \mathrm{H} H^{\mathrm{B}}\right), 3.76\left(3 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CO}_{2} \mathrm{C} H^{\mathrm{A}}{ }_{3}\right)\right), 3.77$ $\left(3 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CO}_{2} \mathrm{CH}^{\mathrm{B}}\right)\right), 3.87\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 11.3, \mathrm{C}(5) \mathrm{H} H^{\mathrm{A}}\right), 4.39\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{A}}\right), 4.58\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}\right.$, $\left.\mathrm{C}(2) H^{\mathrm{B}}\right), 7.07-7.15(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.18-7.25(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.30-7.37(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(100.6$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 30.1$ and $30.3(C(4)), 34.0$ and $34.2(C(3)), 36.4(C(6)), 46.1$ and $46.4(C(5)), 52.3\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 59.5$ and $60.0(C(2)), 79.4$ and $79.5\left(\mathrm{O}\left(C\left(\mathrm{CH}_{3}\right)_{3}\right), 126.2\right.$, 126.9, 127.1, 128.4, 128.8) (all PhCH ), 135.7 and 135.8 ( PhArC ), 144.7 and 144.8 ( $\mathrm{Ph}{ }^{\prime} \mathrm{ArC}$ ), 152.2 and $152.6\left((\mathrm{CO})\left(\mathrm{NR}_{2}\right)(\mathrm{OR})\right), 173.6$ and $176.6\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right) ; \mathrm{m} / z(\mathrm{ESI}+) 394.2\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, 48\%); HRMS $416.1837\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{24} \mathrm{H}_{27} \mathrm{NNaO}_{4}\right.$ requires 416.1832).
(2S,3R,4R)-1-Aza-1-tert-butyloxycarbonyl-6,6-diphenyl-bicyclo[3.1.0]hexane-2carboxylic acid 29b ${ }^{53}$

To a solution of $\mathbf{2 9} \mathbf{a}(60 \mathrm{mg}, 0.15 \mathrm{mmol})$ in $\mathrm{MeOH}(5.1 \mathrm{~mL})$ and water $(1.7 \mathrm{~mL})$ was added lithium hydroxide ( $32 \mathrm{mg}, 0.76 \mathrm{mmol}$ ). After stirring at r.t. for 2 d , MeOH was removed in vacuo. $\operatorname{EtOAc}(5 \mathrm{~mL})$ was added and the organic layer was separated and concentrated to give unwanted organic material. Then, EtOAc ( 5 mL ) and $\mathrm{HCl}(0.25 \mathrm{M}, 10 \mathrm{~mL})$ were added sequentially to the aqueous mixture. The organic layer was separated and the aqueous was extracted with EtOAc ( $4 \times 5 \mathrm{~mL}$ ) before the combined organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give acid 29b as a white solid ( $56 \mathrm{mg}, 97 \%$ ) which required no further purification; m.p. dec. $190^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}-40\left(\mathrm{c}=1.25, \mathrm{CHCl}_{3}\right) ;[\alpha]_{\mathrm{D}}{ }^{22}-51.2(\mathrm{c}=0.5$, $\mathrm{CHCl}_{3}$ ); $\mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 2978, $1700(\mathrm{C}=\mathrm{O}), 1495,1426,1368,1172,1139,760,709$; $\delta_{\mathrm{H}}\left(500.3 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ Approximate rotamer ratio $(\mathrm{A}: \mathrm{B})=(2: 1), 1.15\left(9 \mathrm{H}^{\mathrm{A}}, \mathrm{s},\left(\mathrm{CH}^{\mathrm{A}}\right)_{3} \mathrm{CO}\right)$, $1.18\left(9 \mathrm{H}^{\mathrm{B}}, \mathrm{s},\left(\mathrm{CH}_{3}^{\mathrm{B}}\right)_{3} \mathrm{CO}\right), 2.26\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{dd}, J 4.5,7.2, \mathrm{C}(4) H^{\mathrm{B}}\right), 2.30\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{dd}, J 4.5,7.2\right.$, $\left.\mathrm{C}(4) H^{\mathrm{A}}\right), 2.47\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 7.2, \mathrm{C}(3) H^{\mathrm{B}}\right), 2.66\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 7.4, \mathrm{C}(3) H^{\mathrm{A}}\right), 3.62\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{dd}, J 4.7\right.$, $\left.11.5, \mathrm{C}(5) H^{\mathrm{A}} \mathrm{H}\right), 3.70\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{dd}, J 4.6,11.4, \mathrm{C}(5) H^{\mathrm{B}} \mathrm{H}\right), 3.76\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{d}, J 11.5, \mathrm{C}(5) \mathrm{H} H^{\mathrm{A}}\right), 3.90$ $\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{d}, J 11.4, \mathrm{C}(5) \mathrm{H} H^{\mathrm{B}}\right), 4.42\left(1 \mathrm{H}^{\mathrm{B}}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{B}}\right), 4.53\left(1 \mathrm{H}^{\mathrm{A}}, \mathrm{s}, \mathrm{C}(2) H^{\mathrm{A}}\right), 7.08-7.16(3 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H), 7.19-7.27(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.30-7.40(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 28.0$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 30.1$ and $30.6(C(4)), 32.3$ and $34.0(C(3)), 36.1$ and $36.5(C(6)), 46.2$ and 46.6 $(C(5)), 59.8$ and $59.9(C(2)), 79.9$ and $81.3\left(\mathrm{O}\left(C\left(\mathrm{CH}_{3}\right)_{3}\right)\right), 126.3,126.8,126.9,127.2,127.3$, 128.4, 128.4, 128.8, 128.9, 128.9, 129.1 ( 6 x PhCH), 135.6 and 135.7 ( PhC ), 144.4 and 144.6 (Ph'C), 152.2 and $154.9\left(C(\mathrm{O})\left(\mathrm{NR}_{2}\right)(\mathrm{OR}), 173.6\right.$ and $176.6\left(\mathrm{CO}_{2} \mathrm{H}\right) ; m / z(\mathrm{ESI}-) 378\left([\mathrm{M}-\mathrm{H}]^{\top}\right.$, 100\%); HRMS $402.1676\left([\mathrm{M}+\mathrm{Na}]^{+}, \mathrm{C}_{23} \mathrm{H}_{25} \mathrm{NNaO}_{4}\right.$ requires 402.1676).

## (2S,3R,4R)-1-Aza-6,6-diphenyl-bicyclo[3.1.0]hexane-2-carboxylic acid 29c

A solution of acid $\mathbf{2 9 b}(20 \mathrm{mg}, 53 \mu \mathrm{~mol})$ in $\mathrm{DCM}(1 \mathrm{~mL})$ and dry $\mathrm{HCl}\left(2 \mathrm{M}\right.$ in $\left.\mathrm{Et}_{2} \mathrm{O}, 3 \mathrm{~mL}\right)$ was stirred vigorously for 3 h . The mixture was concentrated before EtOAc ( 1 mL ) was added followed by dropwise addition of petrol. After the precipitate had settled removal of the
solvent (via pipette) and concentration in vacuo left 29c as a white foam ( $15 \mathrm{mg}, 100 \%$ ); $[\alpha]_{\mathrm{D}}{ }^{19}-24.8(\mathrm{c}=0.85, \mathrm{MeOH}) ; \mathrm{v}_{\max } / \mathrm{cm}^{-1}$ (film) 3406 (br), 3059, 3027, 1730 (sh), 1677 (C=O), 1495, 1447, 1198, 1136, 711, 698; $\delta_{\mathrm{H}}(500.3 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 2.71(1 \mathrm{H}, \mathrm{t}, J 5.9$, $\mathrm{C}(4) H), 2.86(1 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{C}(3) H), 3.58(1 \mathrm{H}, \mathrm{d}, J 12.5, \mathrm{C}(5) H \mathrm{H}), 3.88(1 \mathrm{H}, \mathrm{dd}, J 5.3,12.3$, $\mathrm{C}(5) \mathrm{H} H), 4.45(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(2) H), 7.13-7.18(1 \mathrm{H}, \mathrm{m}, p-\mathrm{Ph} ' H), 7.21-7.29\left(4 \mathrm{H}, \mathrm{m}, o-\mathrm{and} m-\mathrm{Ph}{ }^{\prime} H\right)$, 7.36-7.42 ( $1 \mathrm{H}, \mathrm{m}, p-\mathrm{Ph} H)$, 7.47-7.54 (4H, $\mathrm{m}, o-$ and $m-\mathrm{Ph} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz} ; d 4-\mathrm{MeOD}) 32.0$ $(C(4)), 36.1(C(3)), 41.2(C(6)), 48.0(C(5)), 62.7(C(2)), 128.0\left(o-\mathrm{Ph}{ }^{\prime} C H\right), 128.4\left(\mathrm{Ph}{ }^{\prime} \mathrm{CH}\right)$, $129.6(o-\mathrm{PhCH}), 129.8\left(\mathrm{Ph}{ }^{\prime} \mathrm{CH}\right), 131.3,131.7($ both PhCH$), 136.2$ and $145.6\left(\mathrm{PhC}\right.$ and $\left.\mathrm{Ph}{ }^{\prime} C\right)$, $170.9\left(\mathrm{CO}_{2} \mathrm{H}\right) ; m / z(\mathrm{ESI}+) 280\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right) ;$ HRMS $280.1330\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{C}_{18} \mathrm{H}_{18} \mathrm{NO}_{2}\right.$ requires 280.1332 ).

## References

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## Uncatalysed diaryldiazo cyclopropanations on bicyclic lactams: access to conformationally constrained amino acids

L. Harris, M. Gilpin, A.L. Thompson, A.R. Cowley and M.G. Moloney

$\mathrm{R}=\mathrm{H}, \mathrm{CO}_{2} \mathrm{Et}$


Acetone, $\mathrm{N}_{2}$, Reflux




Scheme 1



Figure 1


8c



Ph

Figure 3



17a $\mathrm{A}=\mathrm{B}=\mathrm{CH}$ (Fluorenyl)
17b $A=C H, B=N$ (Azafluorenyl) 17c $A=N, B=N$ (Diazafluorenyl)

Scheme 2




20a $\mathrm{R}^{1}=\mathrm{CO}_{2} \mathrm{Me}, \mathrm{R}^{2}=\mathrm{H}$ 20b $\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{CO}_{2} \mathrm{Me}$


Scheme 4



Scheme 5


24d $X=N, Y=C H$
24e $X=C H, Y=N$


$4 a$


4 c


4 e


8d


7a


4d


8a


8e'


13a


16


25c
Figure 4


