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Mukherjee, Kakoli; AstraZeneca India Pvt Ltd Hosagrahara, Vinayak; AstraZeneca India Pvt Ltd Solapure, Suresh ; AstraZeneca India Pvt Ltd Hameed P, Shahul; AstraZeneca India Pvt Ltd

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## Scaffold Morphing Led to Evolution of 2,4-Diaminoquinolines and Aminopyrazolopyrimidines As Inhibitors of ATP Synthesis Pathway<sup> $\dagger$ </sup>

Subramanyam J. Tantry,<sup>a</sup> Vikas Shinde,<sup>a</sup> Gayathri Balakrishnan,<sup>a</sup> Shankar D. Markad,<sup>a</sup> Amit K. Gupta,<sup>a</sup> Jyothi Bhat,<sup>a</sup> Ashwini Narayan,<sup>a</sup> Anandkumar Raichurkar,<sup>a</sup> Lalit Kumar Jena,<sup>a</sup> Sreevalli Sharma,<sup>a</sup> Naveen Kumar,<sup>a</sup> Robert Nanduri,<sup>a</sup> Sowmya Bharath,<sup>a</sup> Jitendar Reddy,<sup>a</sup> Vijender Panduga,<sup>a</sup> Prabhakar K. R,<sup>a</sup> Karthikeyan Kandaswamy,<sup>a</sup> Parvinder Kaur,<sup>a</sup> Neela Dinesh,<sup>a</sup> Supreeth Guptha,<sup>a</sup> Ramanatha Saralaya,<sup>a</sup> Manoranjan Panda,<sup>a</sup> Suresh Rudrapatna,<sup>a</sup> Meenakshi Mallya,<sup>a</sup> Harvey Rubin,<sup>d</sup> Takahiro Yano,<sup>d</sup> Khisi Mdluli,<sup>b</sup> Christopher Cooper,<sup>b</sup> V Balasubramanian,<sup>a</sup> Vasan K. Sambandamurthy,<sup>a</sup> Vasanthi Ramachandran,<sup>a</sup> Radha Shandil,<sup>a</sup> Stefan Kavanagh,<sup>c</sup> Shridhar Narayanan,<sup>a</sup> Pravin Iyer,<sup>a</sup> Kakoli Mukherjee,<sup>a</sup> Vinayak P Hosagrahara,<sup>a</sup> Suresh Solapure,<sup>a</sup> Shahul Hameed P,<sup>a</sup>\* Sudha Ravishankar.<sup>a</sup>\*

The success of bedaquiline as an anti-tubercular agent for the treatment of multi-drug resistant tuberculosis has validated ATP synthesis pathway and in particular ATP synthase as an attractive target. However, limitations associated with its use in the clinic and the drug-drug interactions with rifampicin have prompted research efforts towards identifying alternate ATP synthesis inhibitors with differentiated mechanism of action. A biochemical assay was employed to screen AstraZeneca's corporate compound collection to identify inhibitors of mycobacterial ATP synthesis. The high throughput screen resulted in the identification of 2, 4-diaminoquinazolines as inhibitors of ATP synthesis pathway. A structure activity relationship for quinazolines was established and the knowledge was utilized to morph quinazoline core into quinoline and pyrazolopyrimidine to expand the scope of chemical diversity. The morphed scaffolds exhibited a 10-fold improvement in the enzyme potency and over 100fold improvement in selectivity against inhibition of mammalian mitochondrial ATP synthesis. These novel compounds were bactericidal and demonstrated growth retardation of Mycobacterium tuberculosis in the acute mouse model of tuberculosis infection.

<sup>c</sup>AstraZeneca, Alderley Park, Mereside, Cheshire, UK.

\* The authors declare no competing interests.

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<sup>&</sup>lt;sup>a</sup> Innovative Medicines, AstraZeneca India Pvt. Ltd., Bellary Road, Hebbal, Bangalore 560024, India.

<sup>&</sup>lt;sup>b</sup>Global Alliance for TB Drug Development, 40 Wall Street, 24th Floor. New York, NY 10005, USA.

<sup>&</sup>lt;sup>d</sup>University of Pennsylvania, 111 Clinical Research Building, 415 Curie Boulevard, Philadelphia PA 19104, USA

Electronic Supplementary Information (ESI) available: [Materials and Methods; Synthesis schemes 1A and 1B; Fig. S1: 3D flexible superimposition of compound **3h**, compound **8d** and compound **17a** with Bedaquiline, Figs. S2-S3: 2D binding site interaction of inhibitors - Bedaquiline, compound **3h**, 8d, 12 and 17a at the ATPase binding site, Figs. S4-S7: NMR spectra and HPLC profiles of compounds 8g and 17a], associated with this article can be found in the online version. See DOI: 10.1039/x0xx00000x

#### 1. Introduction

Tuberculosis (TB) caused by *Mycobacterium tuberculosis*, a highly infectious disease has plagued mankind for thousands of years. TB still remains a global pandemic causing around 1.4 million deaths and 9 million new incidences each year.<sup>1</sup> The financial burden of TB on developing countries is estimated to be between one and three trillion dollars over the next decade. Though the overall numbers of new cases have decreased by 40% since 1990, the global prevalence of multidrug resistant (MDR) and extensively drug resistant (XDR) TB has gone up substantially.<sup>2</sup>

TB can remain dormant for years after infection and can reactivate once the host become immunocompromised as in the case of HIV coinfected populations.<sup>3</sup> Considerable progress has been made in TB drug discovery in the recent past and a number of drug candidates are at different stages of clinical trials.<sup>4</sup> Diaryl quinoline (TMC207 or bedaquiline, **Fig. 1**), discovered by Johnson & Johnson pharmaceuticals and marketed as Sirturo<sup>TM</sup>, received expedited approval by the FDA for the treatment of MDR-TB.<sup>5</sup> Thus a new TB drug with a different mode of action got approved after a gap of 40 years.<sup>5,6</sup> Bedaquiline inhibits ATP synthase and causes depletion in intracellular ATP levels.<sup>7</sup> This target is essential for the survival of bacteria under both replicating and non-replicating states, with a six times lower intracellular ATP levels during the non-replicating state as compared to the levels in the actively replicating state. Maintaining this threshold level of ATP is absolutely essential to keep the bacteria viable. Any further depletion would lead to a detrimental effect on bacterial survival.<sup>8</sup> Although, ATP synthase is highly conserved and of fundamental importance to both prokaryotes and eukaryotes, bedaquiline is highly selective towards the mycobacterial enzyme. This critical information provides the rationale behind inhibiting a highly conserved target for clinical progression.<sup>9</sup> The unprecedented success and novelty of bedaquiline has triggered several groups to investigate this pathway.

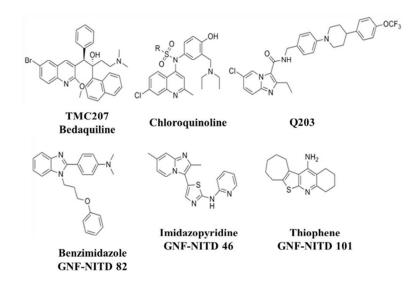
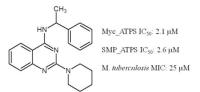


Fig. 1. Structures of ATP synthesis pathway inhibitors

**Fig. 1** represents the structures of a few leading mycobacterial ATP synthase inhibitors reported in the literature to date.<sup>10-12</sup> In order to identify novel scaffolds targeting mycobacterial ATP-synthesis,

we screened AstraZeneca's corporate compound collection by employing a high throughput mycobacterial ATP synthesis assay (Myc\_ATPS) which used *Mycobacterium smegmatis* membrane Journal Name

preparation (inverted membrane vesicles - IMVs)<sup>13</sup> to measure ATP synthesis via the oxidative phosphorylation (Ox-Phos) process. Compound 3a (2, 4-diaminoquinazoline, Fig. 2) emerged as one of the promising hits from this high throughput screening (HTS) effort. However, this initial hit (3a) was not selective as it inhibited the mammalian mitochondrial ATP synthesis assay (SMP ATPS) with equal potency. This report describes the results of medicinal chemistry structure activity relationship (SAR) efforts leading to the evolution of potent mycobacterial ATP synthesis pathway inhibitors, aminopyrazolopyrimidines and 2, 4diaminoquinolines with about 20-fold and 120-fold selectivity window respectively.



#### Fig. 2: Structure of HTS hit (compound 3a)

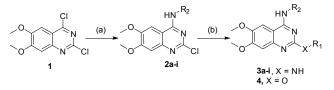
#### 2. Results and Discussion

#### 2.1. Chemistry

The syntheses of compounds derived from quinazoline, quinoline and pyrazolopyrimidine series are described in **schemes 1-4.** 

#### 2.1.1. Quinazoline derivatives (3a-i)

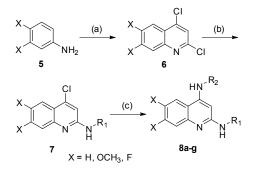
The required R2 groups were incorporated by the displacement of 4-chloro functionality of 2,4dichloro-6,7-dimethoxy-quinazoline with corresponding R2-amines to give intermediates **2a-i** in good yields. Finally, 2-chloro of intermediates **2a-i** was displaced with R1-amines under microwave conditions to provide compounds **3a-i** (**Scheme 1**). While most of the raw materials used as substitutions at R1 and R2 positions were procured commercially, one of the key intermediate 6-[2-(dimethylamino)ethoxy]-5-fluoropyridin-3amine utilized for the synthesis of many of the potent derivatives was obtained following a 4-step synthetic protocol starting from corresponding 3-fluoropyridin-2-ol (*Scheme 1A, supporting information 1*). Similarly, compound 4 with ether linkage at C2 position was synthesized by modification of the intermediate **2h** (*Scheme 1B, supporting information 1*)



Scheme 1. Synthesis of 2, 4-diaminoquinzolines. Reagents and conditions: (a)  $R_2$ -NH<sub>2</sub>, NaH, DMF, 80 °C, 2 h, yield: 45-55% and (b)  $R_1$ -NH<sub>2</sub>, Na<sub>2</sub>CO<sub>3</sub>, *n*-BuOH, MW, 170 °C, 1.5 h, yield: 19-45%.

#### 2.1.2. Quinoline derivatives (8a-g)

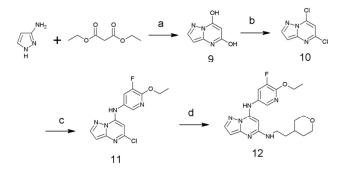
Condensation of malonic acid with aniline/ substituted anilines **5** followed by treatment with POCl<sub>3</sub> gave substituted/unsubstituted 2,4dichloroquinolines **6**. C<sub>2</sub>-substitution with R<sub>1</sub>amines followed by C4-substitution with R<sub>2</sub>-amines under Buchwald-Hartwig reaction conditions gave compounds **8a-g** (**Scheme 2**).



Scheme 2. Synthesis of 2, 4-diaminoquinolines. Reagents and conditions: (a) Malonic acid, POCl<sub>3</sub>, 120 °C, 6 h, yield: 60-65%; (b)  $Pd_2(dba)_3$ , XantPhos, NaOtBu, PhCH<sub>3</sub>, 120 °C, MW 1 h, yield: 35-40%; (c)  $Pd_2(dba)_3$ , XantPhos, NaOtBu, PhCH<sub>3</sub>, 120 °C, MW 1 h, yield: 7-72%.

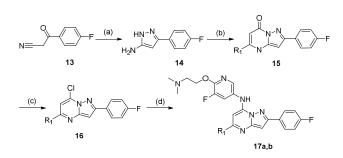
#### 2.1.3. Pyrazolopyrimidine derivatives (12, 17a,b)

Synthesis of compound 12 is described in scheme 3a. The condensation of diethyl malonate and 1H-pyrazol-3-amine with metallic sodium in ethanol resulted in cyclized 2,4-dihydroxy pyrazolopyrimidine intermediate 9. Treatment of intermediate 9 with POCl<sub>3</sub> provided chloro intermediate 10. Nucleophilic substitution of chloro intermediate 10 with 6-ethoxy-5-fluoropyridin-3-amine followed by a second nucleophilic substitution at C2 position resulted in compound 12.



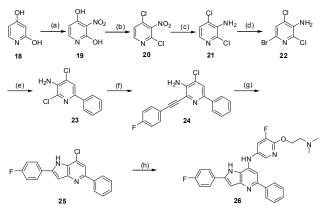
Scheme 3a. Synthesis of aminopyrazolopyrimidine 12. Reagents and conditions: (a) Na, EtOH, 85  $^{\circ}$ C, 18 h, yield: 62%; (b) POCl<sub>3</sub>, 120  $^{\circ}$ C, 3 h, yield: 30%; (c) 6-ethoxy-5-fluoropyridin-3-amine, NaH, DMF, 75  $^{\circ}$ C, 2 h, yield: 94%; (d) 2-(tetrahydro-2*H*-pyran-4-yl) ethanamine, Na<sub>2</sub>CO<sub>3</sub>, DIPEA, n-BuOH, MW, 170  $^{\circ}$ C, 1.25 h, yield: 34%.

Synthesis of compounds 17a and 17b is described in scheme 3b. Condensation of 3-(4fluorophenyl)-3-oxopropanenitrile 13 with hydrazine hydrate under heating conditions gave 5aminopyrazole derivative 14. The pyrazolopyrimidinone ring 15 was constructed by the reaction of 5-aminopyrazole derivative 14 with various R<sub>1</sub>-substituted β-ketoesters. Treatment of compound 15 with POCl<sub>3</sub> provided chloro intermediate 16. Nucleophilic substitution of chloro intermediate 16 with 6-[2-(dimethylamino)ethoxy]-5-fluoropyridin-3-amine (Scheme 1A, supporting information 1) afforded compounds 17a and 17b (Scheme 3b).



Scheme 3b. Synthesis of aminopyrazolopyrimidines. Reagents and conditions: (a)  $NH_2NH_2.H_2O$ , EtOH, 80 °C, 4 h, yield: 33%; (b)  $R_1$ -CO-CH<sub>2</sub>- COOEt, AcOH, 120 °C, 12 h, yield: 48-55%; (c) POCl<sub>3</sub>, 135 °C, 3 h, yield: 60-68%; (d) 1d, K<sub>2</sub>CO<sub>3</sub>, DMF, 75 °C, 2 h, yield: 22-33%.

The synthesis of compound **26** with azaindole core is provided in **scheme 4**.



Scheme 4. Synthesis of azaindole 26. Reagents and conditions: (a) HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, 0 °C, 0.5 h, yield: 85%; (b) POCl<sub>3</sub>, 135 °C, 12 h, yield: 72%; (c) Fe, NH<sub>4</sub>Cl, EtOH-H<sub>2</sub>O, (4:1) 80 °C, 5 h, yield: 82%; (d) NBS, DMF, 0-25 °C, 1 h, yield: 47%; (e) PhB(OH)<sub>2</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>, K<sub>2</sub>CO<sub>3</sub>, 1,4-dioxane, H<sub>2</sub>O, 85 °C, 12 h, yield: 81%; (f) 1-ethynyl-4fluorobenzene, Pd(Ph)<sub>2</sub>Cl<sub>2</sub>, CuI, TEA, 80 °C, 5 h, yield: 45%; (g) KO<sup>t</sup>Bu, THF, MW, 100 °C, 1 h, yield: 78%; (h) 1d, <sup>t</sup>BuXPhos, THF, NaO<sup>t</sup>Bu, MW, 100 °C, 2 h, yield: 63%.

#### 2.2. Structure Activity Relationship

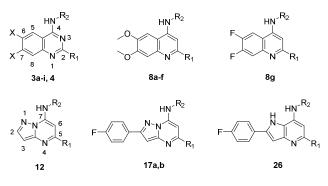
The initial hit obtained from HTS campaign, compound 3a (Fig. 2), was less potent in both enzyme assay (IC<sub>50</sub> of 2.1 µM) and antimycobacterial potency (M. tuberculosis minimum inhibitory concentration (MIC) of 25 µM) as compared to some of the reported inhibitors of this pathway. Moreover, it had poor selectivity as it inhibited mammalian mitochondrial ATP synthesis (SMP ATPS) with an IC<sub>50</sub> of 2.6  $\mu$ M. The later assay was similar to Myc ATPS assay in principle but used submitochondrial particles (SMP) from bovine heart mitochondria as the source of mitochondrial Ox-Phos components. Since ATP synthesis is a ubiquitous pathway and highly conserved across species, our initial efforts were focused on understanding the structure activity relationship (SAR) requirements of quinazoline series and improving the selectivity index (ratio of SMP ATPS  $IC_{50}$  to Myc ATPS  $IC_{50}$ ). The molecular structure of compounds synthesized as part of this SAR effort along with their enzyme inhibitory and anti-mycobacterial potency have been listed in Table 1. Simple bioisosteric replacement of piperidine by pyrrolidine ring at C2 position tolerated for potency but the selectivity index got worsened for compound 3b (Compound 3a vs 3b, Table 1). In the next round of design, we introduced polarity in the molecules at C2 and C4 position to improve selectivity and potency. Polarity modification hypothesis around compound **3a** resulted in achieving a 15-fold improvement in Myc ATPS IC<sub>50</sub>. Improved enzyme potency for compounds 3d, 3e, 3g also translated into a 6 to 8fold improvement in MIC but the selectivity index remained a concern for guinazoline series (Table 1) except for compounds 3d, 3g and 3i.

Replacement of piperidine at the C2 position and  $\alpha$ -methyl benzylamine at C4 position in compound **3b** with 2-(tetrahydro-2H-pyran-4yl)ethanamine and 5-tert-butyl-1-methyl-1Hpyrazol-3-amine respectively yielded compound **3c** with no loss in potency suggesting that there is enough space for such hydrophobic groups. Introduction of additional polarity via dimethoxy

group at the C6, C7 positions of quinazoline ring resulted in a 10-fold improvement in enzyme potency and selectivity index for compound 3d (Compound 3c vs 3d, Table 1). Later, we confined to having dimethoxy group substitution at C6, C7 during the majority of SAR modification due to potency gains observed and for reasons of ease of synthesis. Further replacement of pyrazole at C4 position by suitably substituted thiazole and pyridine ring systems tolerated for potency but the selectivity index was compromised by 5 fold (compound 3d vs 3f, Table 1). Substituting the pyran at C2 by pyridine ring system retained the enzyme potency with improved selectivity (compound 3f vs 3g, Table 1). However, the compound **3g** lost the MIC completely which may be due to bacterial permeability. Compound 4 with an ether linkage at C2 position was found be inactive in both Myc ATPS assay and M. tuberculosis MIC assay (Table 1). This indicated the requirement of an amino group at C2 position for mycobacterial ATP synthesis inhibition. The nitrogen here could be either secondary or tertiary and indicated that it is probably making a critical interaction with the enzyme. One of the explanations for this observed phenomenon could be the possible quaternization of nitrogen at physiological pH which is not possible with ether linkage. Elongation of ethoxy substitution of C4 pyridine side chain with terminal basic group showed moderate potency with enhanced selectivity index and an absolute IC<sub>50</sub> of >25  $\mu$ M in SMP ATPS assay (Compound 3h vs 3i, Table 1). This result suggested that an additional base centre in the molecule may provide handles for selectivity improvement without compromising on the potency.

None of the other attempts made to further resolve the issue around selectivity was successful for compounds with 2, 4-diaminoquinazoline core. Hence, we decided to expand the chemical scope of central core ring into 2, 4-diaminoquinoline and 5phenylpyrazolo[1,5-a]pyrimidine. Majority of the SAR exploration on the new core was carried out

using 6-(2-(dimethylamino)ethoxy)-5fluoropyridin-3-amine at R<sub>1</sub> position as this particular group on quinazoline, compound **3i**, had earlier resulted in an absolute IC<sub>50</sub> value of >25 µM against mammalian mitochondrial ATP synthesis.



**Fig. 3.** Generic structures of synthesized compounds

Converting quinazoline core ring into quinoline by substituting N3 nitrogen with carbon weakened the enzyme potency marginally (Match pairs: 3h vs 8a and 3i vs 8b in Table 1). Encouraged with expanded chemical diversity of quinoline core with moderate enzyme potency, we then explored SAR of quinoline core with various R<sub>1</sub> substitutions to improve potency. Replacement of tetrahydropyranyl ethylamine substitution with 2-(2-chlorophenyl)ethanamine resulted in 60-fold improvement in enzyme potency and antimycobacterial activity (Match pair: Compound 8b vs 8c, Table 1). Compound 8c also showed selectivity improved index (Table 1).

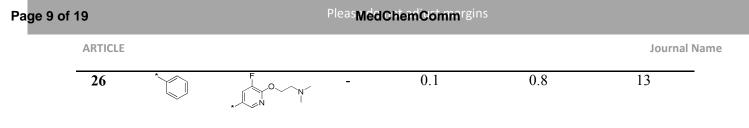
Table 1: SAR around 2,4-diaminoquinazoline (3a-3i, 4), 2,4-diaminoquinoline (8a-8f, 8g) and aminopyrazolopyrimidine (12, 17a-17b, 26) cores.

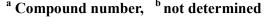
CN. <sup>a</sup>	R <sub>1</sub>	R <sub>2</sub>	X	Myc_ATPS IC <sub>50</sub> (µM)	M. tuberculosis MIC (μM)	Selectivity Index
<b>3</b> a	* N	, Ĉ	Н	2.1	25	2.6
3b	* N_>		Н	2.5	50	<0.04
3c	. <sup>II</sup>	N-N *	Н	1.7	6.2	4.5
3d	*- <sup>H</sup>	N-N *	OCH <sub>3</sub>	0.1	6.2	52
3e		s	OCH <sub>3</sub>	0.1	6.2	ND <sup>b</sup>
3f	*.H	* N	OCH <sub>3</sub>	0.4	3.1	10

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Journal	Name					AF	RTICLE
3g	*-N	↓ O ↓	OCH <sub>3</sub>	0.1	>200	96	
3h	+ <sup>H</sup> N	F V N	OCH <sub>3</sub>	0.6	1.9	2.3	
<b>3</b> i	*_NO	×	OCH <sub>3</sub>	0.9	25	>28	
4	*_00	*N	OCH <sub>3</sub>	25	>200	$ND^{b}$	
8a	*- <sup>H</sup>	F ↓ O ↓ N	-	0.8	6.2	3.2	
8b	* <sup>N</sup> ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	F O N I	-	3.7	200	1	
8c	*H	F O N	-	0.06	3.1	79	
8d	× CI	F ↓ O ↓ N	-	0.04	1.6	128	
8e	× C	F → O N	-	0.06	6.2	44	
8f	× N C	F H N N	-	0.1	1.6	51	
8g	, CI	F O N	-	0.1	3.1	19	
12		F N	-	21	>50	>1.2	
17a	*		-	0.5	6.2	20	
17b	* F F	F N N	-	0.5	12.5	19	





The potency improvement observed with compound 8c suggested that  $R_1$  pocket may be hydrophobic in nature and the aromatic ring system would be a better alternative to an alicyclic ring in order to improve the potency. Further, conformational constraining of phenyl ethylamine group with 3-(2-chlorophenyl)pyrrolidine at  $R_2$ position retained the potency and selectivity (Compound 8d vs 8c, Table 1). In a similar way, the conformational constraining of dimethylaminoethoxy group of R<sub>1</sub> substitution with N-methylpyrrolidine ring retained the enzyme potency but the selectivity index was compromised by 3-fold (Compound 8d vs 8e, Table 1). Replacement of oxygen linker of pyridine side chain with -NH linker in compound 8d retained enzyme potency but anti-mycobacterial potency and selectivity were compromised by 2 to 3-fold (compound 8e vs 8d, Table 1). Modification of dimethoxy substitution of quinoline ring with difluoro group again retained potency but selectivity index was compromised by 6 to 7-fold. Overall, quinoline core demonstrated improved potency and selectivity compared to quinazoline core. Compound 8d from quinoline series displayed the best IC<sub>50</sub>, MIC, and selectivity index of 40 nM, 1.6 µM and 128-fold respectively (Table 1).

In an attempt to further expand chemical scope of the core ring system, we also morphed the central quinoline core to 5. 6-fused pyrazolopyrimdine ring system. The initial compound 12 was found to be inactive in both the biochemical and the anti-mycobacterial assay. In the absence of information on binding of these compounds at molecular level, we relied on chemical intuition to progress the series. Introduction of additional hydrophobic groups like fluorophenyl ring at the C2 position and a phenyl ring at the C5 position of pyrazolopyrimidine core showed some promise of enzyme potency and M. tuberculosis MIC. Compound 17a from pyrazolopyrimdine core demonstrated IC<sub>50</sub> and MIC of 0.5  $\mu$ M and 6.2  $\mu$ M respectively. Replacement of phenyl ring with suitable hydrophobic group like CF3 at the C5 position of pyrazolopyrimidine (17b) core was found to be equipotent as that of 17a. These results indicated that steric and/or hydrophobic factors could be mainly contributing to the potency rather than the planarity of phenyl ring (17b vs 17a, Table 1). Further scaffold morphing was achieved by removing the bridge-head nitrogen of pyrazolopyrimidine and modifying central core into 4-azaindole ring system (Matched pair: 17a vs 26, 
 Table 1). The resulting compound 26 showed best
MIC in the series with M. tuberculosis MIC of 0.8  $\mu$ M and Myc ATPS IC<sub>50</sub> of 0.1  $\mu$ M. Nevertheless, the selectivity index was compromised for compounds with pyrazolopyrimidine and azaindole cores compared to those with quinoline core. Among the different central core ring systems explored, compounds with quinoline core showed improved selectivity with compound 8d demonstrating about 128-fold selectivity index, best among all the compounds synthesized. Fig. 3 provides a snapshot of the generic structures of the compounds synthesized in this study towards understanding the SAR.

#### 2.3. Diaminoquinolines & aminopyrazolo pyrimidines do not damage mycobacterial membrane

During the SAR exploration, compounds were tested for potency in the Myc\_ATPS assay which used the inverted membrane vesicles prepared from *M. smegmatis* as source of Ox-Phos components. To ensure that the inhibition of ATP synthesis and the consequent *M. tuberculosis* MIC observed by these compounds were not due to non-specific perturbation of the mycobacterial membrane, an assay was performed to test their potential to damage membrane and the results are given in Table 2. M. bovis Bacillus Calmette-Guerin (BCG) cells were used as surrogate in this membrane damage assay for reasons like its closeness to M. tuberculosis both at the genome and at the cellular architecture level, translation of *M. tuberculosis* MIC's in BCG and the ease of handling. This assay, an adaptation from the one reported for S. aureus, used a fluorophore, 3, 3'- Diethyloxacarbocyanine iodide (DiOC<sub>2</sub>). A red shift in the emission wavelength is exhibited when DiOC<sub>2</sub> gets stacked in the bacterial cells.<sup>14</sup>

Table 2: Assessment of membrane damage potential

Compound	Myc_ATPS IC <sub>50</sub> (µM)	BCG Membrane damage IC <sub>50</sub> (µM)
3d	0.13	>100
8a	0.85	>100
8d	0.04	>100
8g	0.13	>100
17a	0.45	>100
17b	0.51	>100
26	0.13	>100
Bedaquiline	0.005	>100
СССР	4.8	3

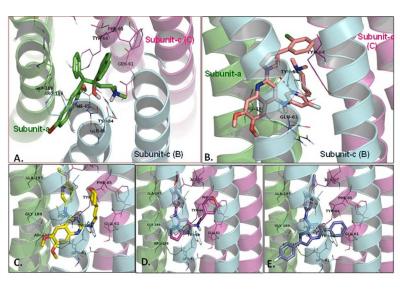
All compounds tested in this assay had an IC<sub>50</sub> of over 100 µM which is about hundred to thousand fold greater than the  $IC_{50}$  in the Myc ATPS assay suggesting that these compounds specifically inhibited mycobacterial ATP synthesis unlike the Page 10 of 19

positive control used in the assay, carbonyl cyanide m-chloro phenyl hydrazine (CCCP) had an IC<sub>50</sub> of 3  $\mu$ M in the membrane damage assay while its IC<sub>50</sub> in the Myc ATPS assay was also in the same range (Table 2).

#### 2.4. Modeling studies with compounds 3h, 12 & 17a

Encouraged by the observation that the inhibition of ATP synthesis is not due to nonspecific damage to the membrane, we utilized modeling studies to understand the possible interaction of these compounds with the ATP synthesis pathway component. The 2-dimensional (2D) tanimoto distance similarity (>50 %) as well as the 3-dimensional (3D) shape overlay similarity of select compounds **3h** (diaminoquinazoline core), 8d (diaminoquinoline core) and 17a (aminopyrazolopyrimidine core) with the crystal structure of bedaquiline led us to hypothesize that these inhibitors may occupy the binding cavity at the interface of the two c-subunits (chain B and chain C) and the a-subunit of ATP synthase similar that reported for bedaquiline with M. to tuberculosis ATP synthase (Fig. S1, supporting information 2).<sup>15</sup> In order to predict the possible mode of binding and to understand the rationale observed from the structure activity relationship data in Table 1, we performed docking studies of these molecules at the active site using the earlier reported homology model of *M. tuberculosis* ATP synthase.<sup>16</sup> Compounds were docked onto the bedaquiline binding site using Glide 6.1 extra precision(XP) method.<sup>17</sup> The binding site interaction of bedaquiline in the M. tuberculosis ATP synthase model is shown in Fig. 4, where it bound the enzyme diagonally at the interface of the two c-subunits (chain B and chain C) and the asubunit of ATP synthase. These interactions would plug the rotation of c-subunits resulting in the inhibition of ATP synthase and hence ATP synthesis (Fig. 4A).

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**Fig. 4.** Predicted binding mode of inhibitors. **A.** Bedaquiline (green colored carbon); **B.** Compound **8d** (amaranth pink colored carbon); **C.** Compound **3h** (yellow colored carbon) **D.** Compound **12** (pink colored carbon); **E.** Compound **17a** (blue colored carbon). Helices in cyan and pink color are chains A and B of c-subunit respectively whereas helix in green color is a-subunit of Fo component of ATP synthase.

The binding pose analysis of compound **8d** at the active site of *M. tuberculosis* ATP synthase revealed that dimethoxyquinoline group of **8d** interacts at the interface of Atp-a and Atp-c (chain-B) subunits through  $\pi$ - $\pi$  interaction with Tyr-64 and H-bond interaction with Glu-61 and directs the fluoropyridin dimethoxyquinolin-4-amine group towards the interface of Atp-a and Atp-c (chain-C) subunits. Further, chlorophenyl group of the molecule nicely fitted into the hydrophobic cavity between the two Atp-c subunits (**Fig. 4B**). The occupancy of **8d** at the interface of Atp-a and Atp-c subunits caused the blockage of Atp-c subunit rotation and thereby its enzyme activity.

Similarly, the binding pose study of compound **3h** (**Fig. 4C**) revealed that the dimethoxy quinazoline group of **3h** occupied the cavity between Atp-a and Atp-c (chain-B) subunits. The amine group attached with pyridyl ring was found to interact with Arg-186 of Atp-a subunit, whereas the amine attached with pyran ring made H-bond contact with Glu-61 of Atp-c subunit (chain-B) providing strong anchorage to the molecule and place the pyran ring towards the hydrophobic space between the two Atp-c subunits. The predicted binding mode of compound 12 (Fig. 4D) revealed the H-bond contact of pyrimidine nitrogen with Tyr-64 of Atp-c subunit (chain-B). The loss of Myc\_ATPS inhibitory activity for compound 12 (IC<sub>50</sub> = 21  $\mu$ M) could be due to the lack of functional group to explore the hydrophobic pocket (constituting residues from Atp-a and Atp-c (chain-B) subunits) unlike bedaquiline where bromoquinoline exploits this region optimally (Fig. 4C). However, the pyran and pyridyl ring of the molecule showed similar binding pattern as noted for compound 3h.

The binding pose comparison study of compound **17a** (**Fig. 4E**) showed that the addition of para fluoro phenyl group at pyrazole ring in compound **17a** allowed the compound to regain its interface blocking capacity between Atp-a and Atp-c (chain-B) subunits and hence regain the inhibitory potency. Further H-bond contact of pyrimidine nitrogen with Tyr-64 of Atp-c subunit (chain-B) anchored the ring and directed the attached phenyl ring to occupy the hydrophobic space between the two Atp-c subunits resulting in ATP synthase inhibitory activity of compound **17a** (IC<sub>50</sub> = 0.5  $\mu$ M) which is comparable to compound **3h**. Thus our docking study not only suggested the plausible binding mode of these derivatives but also explained the activity variation among them.

# 2.5. Cross resistance profile of diaminoquinolines and aminopyrazolo pyrimidines.

studies led Computational test us to diaminoquinoline and aminopyrazolopyrimidine derivatives against bedaquiline resistant mutants to check if they would exhibit the same mechanism of action as bedaquiline. A select set of compounds from these subseries were profiled against a panel of drug sensitive and drug resistant M. tuberculosis strains which included two of the bedaquiline resistant strains raised and characterized in-house. MICs against all the five drug sensitive and seven single drug resistant strains were the same as that of wild-type *M. tuberculosis* strain with no cross resistance observed in any of the mutants tested (Table 3). This data not only indicated that

diaminoquinoline and aminopyrazolopyrimidine compounds are equipotent against a range of drug sensitive *M. tuberculosis* strains but they are also equally active against several drug resistant strains satisfying one of the key requirements of a new TB drug candidate. However, the absence of cross resistance against bedaquiline resistant strains suggested that diaminoquinolines and aminopyrazolopyrimidines may not interact with ATP synthase in the same manner as bedaquiline despite binding at the same site. Thus, the key interactions to plug the rotation of the two csubunits for diaminoquinoline and aminopyrazolopyrimidine derivatives could be different from the residues involved with the bedaquiline interaction. This needs to be further validated through characterization of mutants of M. tuberculosis resistant to diaminoquinoline and aminopyrazolopyrimidine derivatives.

					MIC (µN	()			
Inhibitor	Ref	erence		Single	Reference lab				
	Sensiti	ive strains						sti	rains
	<i>M.tb</i> ATCC- 27294	<i>M.tb</i> strains*	<i>M.tb</i> 35820 STR <sup>R</sup>	M.tb 35822 INH <sup>R</sup>	<i>M.tb</i> 19000 RIF <sup>R</sup>	<i>M.tb</i> 17003 EMB <sup>R</sup>	<i>M.tb</i> 12119 OFX <sup>R</sup>	<i>M.tb</i> bedaqui line <sup>R</sup> (I66M)	<i>M.tb</i> bedaqui line <sup>R</sup> (A63P)
8d	6.3	3.1 - 12.5	12.5	6.3	3.1	6.3	12.5	6.3	3.1
8g	3.1	3.1 - 12.5	25.0	3.1	3.1	6.3	25.0	3.1	3.1
17a	6.3	6.3 - 12.5	25	12.5	6.3	6.3	12.5	6.3	12.5
26	3.1	3.1 - 6.3	25	3.1	3.1	6.3	25	3.1	3.1
Bedaqu iline	0.2	<2	<2	<2	<2	<2	<2	10	>10
Streptom ycin (STR)	0.3	0.3	>11	0.3	0.3	0.3	0.3	0.3	0.3

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I ghia 💔 Crace	resistance r	arotile o	t diami	naamnalinee	and aminopyraz	alanyrimidines
1 abic 5. Cross	i constante p	<b>JI UIIIC U</b>	n ulaiiii	ivyumumus	a and a minuply $a$	VIUDVIIIIIUIIIUS

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Isoniazid (INH)	0.2	0.2	0.2	>15	0.2	0.2	0.2	0.2	0.2	
Rifampi cin (RIF)	0.04	0.04	0.04	0.04	>1.2	0.04	0.04	0.04	0.04	
Ethambu tol (EMB)	9.8	9.8	9.8	9.8	9.8	>20	4.9	9.8	9.8	
Ofloxaci n (OFX)	0.7	0.7	0.7	0.7	0.7	0.3	>5.6	0.7	0.7	

M.tb: M. tuberculosis

\*MIC tested against four of *M. tuberculosis* drug sensitive strains: *M. tuberculosis* 25618, *M. tuberculosis*-Erdman, *M. tuberculosis*-Beijing (E-47/94), *M. tuberculosis*-Harlingen.

### 2.6. Diaminoquinolines & aminopyrazolo pyrimidines deplete intracellular ATP pool.

Inhibitors of ATP synthase enzyme or the other components of ATP synthesis pathway are expected to deplete the cells off their intracellular ATP pool. In order to test the ability of diaminoquinoline and aminopyrazolopyrimidine derivatives in reducing the intracellular ATP pool, we tested a representative set of compounds from subseries these in the intracellular ATP measurement assay. Cell lysates of BCG cultures which had been exposed to 5X MIC concentrations of these compounds for 20 hours were used to measure the intracellular ATP concentration. While isoniazid which inhibits FasII system showed no depletion in ATP levels, bedaquiline used as positive control exhibited 76% depletion in ATP levels. Compounds belonging to diaminoquinoline and aminopyrazolopyrimidines were not as potent as bedaquiline except for compound 8d in reducing the cellular ATP levels suggesting a need to improve their potency further (Table 4).

Table 4: ATP pool depletion by diaminoquinolines & aminopyrazolopyrimidines.

Inhibitor	% ATP
	depletion
INH	-6
Bedaquiline	76
8f	22
8g	25
8a	5
8d	68
17a	25

3h	31
26	23
None	0

#### 2.7. Diaminoquinolines & aminopyrazolo pyrimidines are bactericidal but exhibit different kill rates

During the course of infection, M. tuberculosis is known to exist in various physiological states such as actively replicating and quiescent nonreplicating forms. Inhibitors of the energy production pathway are known to make the pathogen highly vulnerable resulting in bactericidal effect.<sup>7,18</sup> A select set of compounds from the subseries were tested for their growth inhibitory properties under aerobic and hypoxic conditions (used as a model for non-replicating bacilli). All the compounds tested were bactericidal under both aerobic and hypoxic conditions with a minimum bactericidal concentration (MBC) to MIC ratio reaching a maximum of 15 except for compound 8d (Table 5). However, the two scaffolds differed in kill rates as determined by the in vitro kill kinetics under replicating conditions. Compound 8g (diaminoquinoline) showed a faster kill rate, dependent on both concentration and time whereas 17a (aminopyrazolopyrimidine) seemed to display a dependence on time for cidality (Figs. 5A and **5B**).

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Hypoxic

M.tb

MBC

3.125

CC

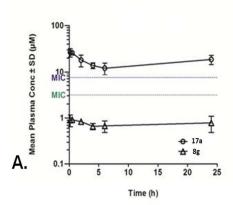
14

Compound

Fig. 5. In vitro survival kinetics of 8g and 17a: *M. tuberculosis* cells were exposed to  $0-200 \mu$ M of compounds 8g (A) and 17a (B) over a period of 7 days. Appropriate dilutions of culture samples were plated for measuring colony forming units on 7H11 plates on 0 (start of exposure), 3 and 7 days post compound exposure.

#### 2.8. Diaminoquinolines & amino pyrazolo pyrimidines retard growth of *M. tuberculosis* in a mouse model of TB infection

Based on the potency and selectivity for inhibition of mycobacterial ATP synthesis, 17a and 8g were tested for efficacy in an acute model of TB infection. Plasma concentration-time profiles in infected mice have been shown in Fig. 6A. In the infected mice, total plasma concentration of 17a was above MIC for 24 h whereas the same for 8g was below MIC during the entire dosing interval. Both the compounds were able to retard the growth of *M. tuberculosis* during the course of infection in mouse lungs unlike the reference drug isoniazid which exhibited a bactericidal effect by reducing the bacterial load by more than two logs in the two week treatment period (Fig. 6B).



(µM) (µM) (µM) 3d 6.2 25 50 3f 25 3.1 13 3h 2.2 19 3 50 6.2 25 3e 2.2 54 200 8d 3.1 38 50 8g 50 **8**e 6.2 50 28 100 17a 7.4 0.2 4 **Bedaquiline** 1.6 12 10 8 og10CFU/ml 100 50 25 12.5 6.25 3.125 CC 2 LOQ Α. 0 2 4 6 0 8 Days 12 10 200 100 8 log 10 CFU/ml 50 25 12.5 6 6.25

Table 5: Bactericidal activity of diamino quinolines & aminopyrazolopyrimidines

Aerobic

M.tb

MBC

4

2 LOQ

0

2

4

Days

6

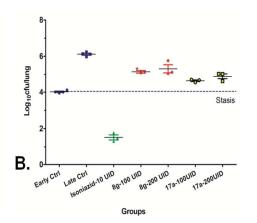
8

Β. 0

M.tb

MIC

Journal Name



**Fig. 6. (A) Pharmacokinetic profiles - 8g** and **17a** were orally administered as a suspension in 0.5% hydroxypropyl methylcellulose (HPMC) & Tween 80 at a 10-ml/kg dose volume and **(B)** Log<sub>10</sub> lung CFU in various groups: untreated early control (blue circles, filled) after 3 days of infection; untreated late control (blue squares, filled) after 2 weeks of infection; after treatment with isoniazid at 10 mg/kg (green symbols) or with **8g** (red symbols) or **17a** (yellow symbols) at 100 or 200 mg/kg once daily (UID), 6 days per week for 2 weeks.

Growth retardation effect by compound **17a** was more pronounced than that of **8g** probably due to its higher plasma concentration. However, the extent of growth inhibition observed with both compounds seemed to be maximal as no dose response effect was seen with the two doses (100 and 200 mg/kg) tested in the study. The growth inhibition observed with **8g** despite its lower plasma concentration could be due to the accumulation of the compounds in the mouse lungs.

#### Conclusion

In conclusion, we report here the discovery of diaminoquinoline, aminopyrazolopyrimdines and azaindoles as novel mycobacterial ATP synthesis inhibitors with potent *M. tuberculosis* MIC. Like bedaquiline, these compounds were found to

#### inhibit *M. tuberculosis* growth by reducing the intracellular ATP pool although not as effectively as bedaquiline. Docking studies implicated the mode of action to be the inhibition of ATP synthase enzyme, however, the exact molecular mechanism of how these compounds deplete the cells off ATP still needs to be elucidated as no cross resistance was observed with bedaquiline resistant mutants. Scaffold morphing efforts resulted in the identification of multiple cores which enhanced the scope of SAR expansion for the series. Compounds belonging to diaminoquinoline subseries exhibited best enzyme inhibition and anti-mycobacterial potency with a fairly good selectivity index among all the compounds analyzed. Representative compounds from the all subseries exhibited bactericidal activity against M. tuberculosis and were able to demonstrate growth retardation of the bacilli in the lungs of infected mice as compared to the uninhibited control in a murine model of tuberculosis infection.

Additional medicinal chemistry efforts are needed to further optimize the potency, selectivity and pharmacokinetic profiles of these lead series to identify a clinical candidate. Being a novel chemotype with activity against bedaquiline resistant mutant, we believe that this class of compounds has the potential to be developed as an anti-TB drug candidate.

#### Abbreviations

M.tb: M. tuberculosis

Ox-Phos: Oxidative phosphorylation

DiOC2: Diethyloxacarbocyanine iodide CCCP: Carbonyl cyanide m-chloro phenyl hydrazine

Myc\_ATPS: Mycobacterial ATP synthesis assay SMP\_ATPS: Mitochondrial ATP synthesis assay MIC: Minimum Inhibitory Concentration MBC: Minimum Bactericidal Concentration Page 16 of 19

#### **Author Contributions**

S.J.T., S.H.P, S.D.M., A.R., C.C., P.I. and M.P. are responsible for medicinal chemistry design and analyses. S.J.T., V.S., S.D.M. and L.K.J. are responsible for synthetic chemistry. S.R. and M.M. are responsible for compound purity. A.K.G and M.P. designed and performed the modelling. S.H.P., K.M., M.P., S.K. M.M. and V.B. performed the hit triage for the high through screen output. K.M., H.R., T.Y. and K.M. designed IC<sub>50</sub> determination experiments. G.B. and J.B. performed IC<sub>50</sub> determination experiments and analyzed the data. V.R., V.K.S., K.M., V.B. and S.R. designed the microbiology experiments. A.N., S.S., P.K., N.D., S.G., performed microbiology experiments. V.P.H., S.S., S.N. and R.S. Designed and Analyzed PK-PD experiments. N.K., R.N., S.B., J.R., V.P., P.K.R., K.K. and R.S. performed PK-PD experiments. S.R., G.B. and J.B designed, performed and analyzed ATP pool measurement assay. S.J.T., S.D.M., A.K.G., S.R. and S.H.P wrote the manuscript with contributions from all coauthors.

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#### **Funding Sources**

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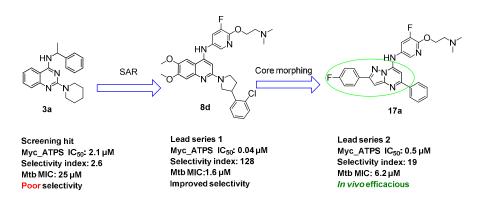
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#### Scaffold Morphing Led to Evolution of 2, 4-Diaminoquinolines and Aminopyrazolopyrimidines as Inhibitors of ATP Synthesis Pathway

Subramanyam J. Tantry,<sup>a</sup> Vikas Shinde,<sup>a</sup> Gayathri Balakrishnan,<sup>a</sup> Shankar D. Markad,<sup>a</sup> Amit K Gupta,<sup>a</sup> Jyothi Bhat,<sup>a</sup> Ashwini Narayan,<sup>a</sup> Anandkumar Raichurkar,<sup>a</sup> Lalit Kumar Jena,<sup>a</sup> Sreevalli Sharma,<sup>a</sup> Naveen Kumar,<sup>a</sup> Robert Nanduri,<sup>a</sup> Sowmya Bharath,<sup>a</sup> Jitendar Reddy,<sup>a</sup> Vijender Panduga,<sup>a</sup> Prabhakar K. R,<sup>a</sup> Karthikeyan Kandaswamy,<sup>a</sup> Parvinder Kaur,<sup>a</sup> Neela Dinesh,<sup>a</sup> Supreeth Guptha,<sup>a</sup> Ramanatha Saralaya,<sup>a</sup> Manoranjan Panda,<sup>a</sup> Suresh Rudrapatna,<sup>a</sup> Meenakshi Mallya,<sup>a</sup> Harvey Rubin,<sup>d</sup> Takahiro Yano,<sup>d</sup> Khisi Mdluli,<sup>b</sup> Christopher Cooper,<sup>b</sup> Balasubramanian V,<sup>a</sup> Vasan K. Sambandamurthy,<sup>a</sup> Vasanthi Ramachandran,<sup>a</sup> Radha Shandil,<sup>a</sup> Stefan Kavanagh,<sup>c</sup> Shridhar Narayanan,<sup>a</sup> Pravin Iyer,<sup>a</sup> Kakoli Mukherjee,<sup>a</sup> Vinayak P Hosagrahara,<sup>a</sup> Suresh Solapure,<sup>a</sup> Shahul Hameed P,<sup>a</sup>\* Sudha Ravishankar.<sup>a</sup>\*

<sup>a</sup> Innovative Medicines, AstraZeneca India Pvt. Ltd., Bellary Road, Hebbal, Bangalore 560024, India.



2, 4-diaminoquinazolines, 2, 4-diaminoquinolines and aminopyrazolopyrimidines, inhibitors of mycobacterial ATP synthesis, are novel lead molecules towards discovery and development of new anti-tubercular agents.

<sup>&</sup>lt;sup>b</sup> Global Alliance for TB Drug Development, 40 Wall Street, 24th Floor. New York, NY 10005, USA.

<sup>&</sup>lt;sup>c</sup>AstraZeneca, Alderley Park, Mereside, Cheshire, UK.

<sup>&</sup>lt;sup>d</sup>University of Pennsylvania, 111 Clinical Research Building, 415 Curie Boulevard, Philadelphia PA 19104, USA