

Promising new applications for Castanea sativa shell: nutritional composition, antioxidant activity, amino-acids and vitamin E profile

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1	Promising new applications for Castanea sativa shell: nutritional composition,			
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22 Abstract

23

24 The present study was aimed to assess the macronutrient composition and the amino acids 25 and vitamin E profile of *Castanea sativa* shell from different Portuguese producing regions 26 (Minho, Trás-os-Montes and Beira-Alta). The nutritional composition was similar for all 27 samples, with a high moisture content and low fat amounts. Arginine and leucine were the 28 predominant essential amino acids (EAA) accounting for 3.55 - 7.21% and 1.59 - 2.08%, 29 respectively, for samples of the different production zones. All the shells presented high 30 contents of vitamin E (481.5 mg/100 g sample, 962.8 mg/100 g sample and 567.5 mg/100 g 31 sample, respectively, for Minho, Trás-os-Montes and Beira-Alta). The predominant vitamer 32 was γ -tocopherol (670 mg/100 g sample for Trás-os-Montes). The antimicrobial and 33 antioxidant activity of C. sativa shell were also determined. Trás-os-Montes extracts 34 displayed the highest antioxidant activity (EC₅₀ = $31.8 \pm 1.3 \mu g/mL$ for DPPH; 8083.5 \pm 164.8 35 μ mol/mg db for FRAP). The total phenolic content (TPC) varied from 241.9 mg to 796.8 mg 36 gallic acid equivalents (GAE)/g db sample, being the highest TPC obtained for Trás-os-Montes. 37 The total flavonoid content (TFC) varied from 31.4 to 43.3 mg of catechin equivalents (CEQ)/g 38 db sample. No antimicrobial activity was observed. Results showed the potentialities of C. 39 sativa shell extracts. 40 41

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44 Key-words: Castanea sativa by-products; shell; sustainability; antioxidants; vitamin E;
45 amino acids

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47 48	1. Introduction
49	Castanea sativa Mill. is a species of the Fagaceae family, abundant in south Europe and Asia.
50	The fruits (chestnut) are an added value resource in producing countries, with an increased
51	economic value due not only to its nutritional properties, but also to the beneficial health
52	effects related with its consumption. In Portugal, chestnut trees are mainly used for nut
53	production, representing a total area of about 35.000 ha, with a production of about 19.000
54	tons per year 1 . These fruits are a highly appreciated seasonal nut (autumn) in the
55	Mediterranean countries ² , which can be consumed raw or cooked, although roasting or
56	boiling are still the most common cooking procedures ³ . Chestnut processing and
57	consumption generate high amounts of by-products, mainly shell, inner skin, outer skin and
58	bur. These may represent a new challenge for the industry: to design innovative applications,
59	answering simultaneously to sustainability concerns, while contributing to create new added
60	value products and economic resources. Recently, different authors have suggested some
61	potential applications of C. sativa shell as heavy metal adsorbent, as phenol substitutes in
62	adhesive formulation or even as chrome substitutes in leather tanning ⁴ . However, only few
63	studies have evaluated the antioxidant potential of this by-product ⁵ .

64 Due to nutritional characteristics of the nut and for all the potentialities that C. sativa shell has shown so far, we believe that this by-product could also be used by other industries, such 65 as pharmaceutical, food or cosmetic, as a natural source of bioactive compounds, generating 66 67 more profits, reducing pollution costs and improving social, economic and environmental 68 sustainability. Nevertheless, economic and environmental sustainability should be a priority for 69 industries, encompassing the type of procedures, chemicals and solvents used to extract 70 interesting compounds. For this reason, the extraction process should be designed as simple as possible, also considering the effective costs for industries ⁶. In that sense, following other 71

72 previous studies that evaluated the combination of different solvents and temperatures to 73 extract antioxidant compounds, the hydro-alcoholic solvent was chosen together with an extraction temperature of 50 °C^{4, 6-8}. Besides, it is well known that climatic conditions such as 74 75 temperature and humidity affect the antioxidant activity and the macronutrient composition 76 of plant products. According to Dinis et al. leaf is the most flexible part of the chestnut tree in 77 its response to environmental conditions⁹. The aim of this study was to evaluate the 78 macronutrients, vitamin E and amino acids contents of C. sativa shells from three different 79 producing regions in Portugal, namely, Minho, Trás-os-Montes and Beira-Alta. Also, the 80 antioxidant and antimicrobial activity of hydro-alcoholic extracts of these samples were 81 evaluated. To the best of our knowledge, no studies have been published regarding the 82 macronutrients, amino acids, vitamin E content or even antimicrobial activity of this agro-83 industrial waste.

84

85 2. Materials and methods

86 2.1.Chemicals and reagents

For the macronutrients analysis all analytical grade reagents were purchased from Panreac (Barcelona, Spain) and Merck (Darmstadt, Germany). Tocopherols (α , β , Υ and δ) and tocotrienols (α , β , Υ and δ) were purchased from Calbiochem (La Jolla, California, USA) and tocol was obtained from Matreya Inc. (Pennsylvania, USA). Butylated hydroxytoluene (BHT) was from Aldrich (Madrid, Spain).

92 Ascorbic acid, 1,1-diphenyl-2-picrylhydrazyl (DPPH[•]) free radical, catechin, Folin–Ciocalteu's 93 reagent, gallic acid, iodine, 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid, glycerol, 94 a water-soluble derivative of vitamin E (Trolox) and α -tocopherol, were all purchased from 95 Sigma - Aldrich (Steinheim, Germany). Ethanol reagent grade, sodium acetate, sodium

96 carbonate decahydrate, sodium nitrite, aluminium chloride and sodium hydroxide were 97 purchased from Merck (Darmstadt, Germany). 98 HPLC-grade acetonitrile and 1,4-dioxane were from Fluka (Madrid, Spain). HPLC grade n-99 hexane was from Merck (Darmstad, Germany). Purified water was obtained from a Milli-Q 100 water purification system (Millipore, Bedford, MA, USA). 101 Vitek-2 identification cards were from Biomerieux (Crappone, France). Brain Heart Infusion, 102 Nutrient agar, Sabouraud dextrose agar and Mueller Hinton broth were purchased from Difco 103 Laboratories (USA). RPMI was from Biochrom (Berlin, Germany). 104 2.2. Samples 105 C. sativa fruits were collected during September 2013 in three different regions of Portugal, 106 namely, Minho, Trás-os-Montes and Beira-Alta, and dried at room temperature for 4 weeks. 107 Samples were prepared simulating the traditional processing of roasting. Chestnuts were 108 baked in the oven (Tecnogás, Portugal) at 240 °C for 35 minutes. Subsequently, shell was 109 separated from chestnut fruit, milled at particle size of approximately 0.1 mm using an A11 110 basic analysis mill (IKA Wearke, Staufen, Germany) and stored in plastic tubes at 4°C until the 111 extracts preparation.

112

113 **2.3. Preparation of hydro-alcoholic extracts**

Powdered samples (1 g) were submitted to solvent extraction by maceration with 20 mL of ethanol: water (1:1) at 40 °C for 30 minutes (Mirac, Thermolyne, USA). Extracts were filtered through Whatman No. 1 filter paper, concentrated under vacuum at 37 °C and kept under refrigeration (4 °C) prior to use.

118

119 **2.4. Macronutrients analysis**

Moisture content was instrumentally determined using an infrared moisture analyzer (SMO 01, Scaltec Instruments, Germany). The ash content was determined by incinerating the sample in a muffle furnace at 550 °C, according to 923.03 method ¹⁰. Protein content (factor 6.25) was determined using the Kjeldahl procedure ¹¹. Total fat was determined by Soxhlet ¹². Total carbohydrate content was determined by difference ¹³. Analyses were performed in triplicate and results are expressed as g/100 g.

126 **2.5.Total vitamin E and vitamers quantification**

127 Lipid fraction for vitamin E quantification was obtained by Soxhlet extraction with petroleum 128 ether (2.5 h). The chromatographic analysis was carried out in an HPLC integrated system 129 equipped with AS-2057 automated injector, a PU-2089 pump and a MD-2018 multi-130 wavelength diode array detector (DAD) coupled to a fluorescence detector FP-2020 (Jasco, 131 Japan), programmed for excitation at 290 and emission at 330 nm. The chromatographic separation of the compounds was achieved on a normal phase SupelcosilTM LC-SI (3 μ m; 75 X 132 3.0 mm; Supelco, Bellefonte, PA, USA) according to Alves *et al.* ¹⁴. Chromatographic data were 133 134 analyzed using JASCO-Chrom NAV Chromatography Software (Jasco, Japan). Results are 135 expressed in mg/100 g of sample. Analyses were performed in triplicate.

136

2.6. Amino acids quantification

Amino acids (Histidine (His), Arginine (Arg), Serine (Ser), Aspartic acid + Glutamic Acid (Asp + Glu), Threonine (Thr), Glycine (Gly), Alanine (Ala), Proline (Pro), Valine + Methionine (Val + Met), Phenylalanine (Phe), Isoleucine (Ile), Leucine (Leu), Ornithine (Orn), Lysine (Lys) and Tyrosine (Tyr)) were analyzed by reversed-phase HPLC with fluorescence detection after submitting samples to an acidic hydrolysis (HCl 6 mol/L, 110 °C, 24 h) ¹⁵⁻¹⁶ and derivatization with dansyl chloride ¹⁷⁻¹⁸. The protein fraction obtained from Kjeldahl procedure was used.

143 The chromatographic analysis was carried out in an HPLC integrated system equipped with an 144 AS-950 automated injector, a PU-980 pump, a CO-2060 Plus oven, and an FP-920 fluorescence 145 detector (Jasco, Japan) programmed for excitation at 335 nm and emission at 514 nm. The 146 compounds separation was achieved in a C18 RP Luna column (4.6 x 250 mm, 5 µm; Supelco, Bellefonte, PA, USA) from Phenomenex (Torrance, CA, USA), according to Pimentel et al.¹⁸. 147 148 Chromatographic data were analyzed using the JASCO-Chrom NAV Chromatography Software 149 (Jasco, Japan). The amino acids were identified by retention time comparison with authentic 150 standards. Quantification was carried out on the basis of the internal standard method. Results 151 are expressed in mg/ 100 g of sample. Analyses were performed in triplicate.

152 **2.7. Determination of total phenolic content**

153 Total phenolic content (TPC) was determined spectrophotometrically according to the Folin-Ciocalteu procedure ¹⁹, with minor modifications ²⁰. Briefly, 500 µL of extract was mixed with 154 155 2.5 mL of Folin–Ciocalteu reagent (10 x dilution) and allowed to react for 5 min. Then 2.5 mL of Na_2CO_3 7.5 % solution was added and allowed to stand for 15 min at 45 °C and 30 min at room 156 157 temperature, before the absorbance being determined at 765 nm using a Synergy HT 158 Microplate Reader (BioTek Instruments, Inc., Winooski, VT, USA). A calibration curve for the 159 standard gallic acid was used to obtain a correlation between sample absorbance and standard 160 concentration (linearity range = 5 - 100 μ g/mL, R^2 > 0.998). The total polyphenol content (TPC) 161 of the extracts was expressed as mg of gallic acid equivalents (GAE) per gram of plant material 162 on dry basis (db).

163 **2.8. Determination of total flavonoid content**

Total flavonoid content (TFC) was determined by a colorimetric assay based on the formation of flavonoid-aluminum compound according to Rodrigues *et al.* ⁶. Briefly, 1 mL of a diluted extract was mixed with 4 mL of ultrapure water and 300 μ L of 5 % (*w/v*) NaNO₂ solution. After

167 5 min, 300 μ L of 10% (*w/v*) g/100 mL AlCl₃ solution were spiked, and after 1 min, 2 mL of 1 168 mol/L NaOH and 2.4 mL of ultrapure water were also added. The absorbance was read at 510 169 nm using the Synergy HT Microplate Reader. Catechin was used as reference to plot the 170 standard curve (linearity range = 0 - 400 μ g/mL, R^2 > 0.999). Total flavonoid concentration 171 (TFC) was expressed as milligrams of catechin equivalents (CAE) per gram of plant material on 172 db.

173 **2.9.** In vitro antioxidant activity

The antioxidant activity of the samples was evaluated by DPPH[•] radical-scavenging activity and
ferric reducing antioxidant power (FRAP) as detailed in the following sections.

176 **2.9.1 DPPH free radical scavenging assay**

177 The reaction mixture was made directly on a 96 wells plate and consisted of a solution of different sample concentrations (30 µL) and methanol solution (270 µL) containing DPPH 178 radicals (6 x 10^{-5} mol/L), in each well. The mixture was left to stand for 30 min in the dark. The 179 reduction of the DPPH radical was determined by measuring the absorption at 517 nm²¹. The 180 181 radical scavenging activity (RSA) was calculated as a percentage of DPPH discolouration using 182 the equation: % RSA = [(ADPPH–AS)/ADPPH] × 100, where AS is the absorbance of the solution 183 when the sample extract has been added at a particular level, and ADPPH is the absorbance of the DPPH solution ²². The extract concentration providing 50% of radicals scavenging activity 184 (EC_{50}) was calculated from the graph of RSA percentage against extract concentration. BHA and 185 186 α -tocopherol were used as standards.

187

2.9.2 Ferric reducing antioxidant power (FRAP) assay

188 FRAP assay was carried out by the method of Benzie and Strain ²³, with minor modifications. 189 The method is based on the reduction of a ferric 2,4,6-tripyridyl-s-triazine complex ($Fe^{3+}-TPTZ$) 190 to the ferrous form ($Fe^{2+}-TPTZ$). An aliquot (90 µL) of an extract (with appropriate dilution, if

191 necessary) was added to 2.7 mL of FRAP reagent (10 parts of 300 mM sodium acetate buffer at 192 pH 3.6, 1 part of 10 mM TPTZ solution and 1 part of 20 mM FeCl₃ · 6H₂O solution) and the 193 reaction mixture was incubated at 37 °C. The increase in absorbance at 592 nm was measured 194 after 30 min. Solutions of known Fe (II) concentrations (FeSO₄-7H₂O) were used for calibration. 195 A calibration curve was prepared with ferrous sulphate (linearity range: 150 - 2000 μ M, R^2 > 196 0.996), and the results were expressed as ferrous sulphate equivalent per gram of plant 197 material on db.

- 198 **2.10.** Antimicrobial assay
- 199 **2.10.1 Microorganisms strains**

200 Eight microorganisms (seven bacteria and one yeast) were included in the study. Both 201 American Type Culture Collection (ATCC) and clinical isolates were selected, corresponding to 202 Staphylococcus aureus (ATCC 6538 and a clinical isolate), S. epidermidis (clinical isolate), 203 Escherichia coli (ATCC 1576 and a clinical isolate), Klebsiella peumoniae (ATCC 4352), 204 Pseudomonas aeruginosa (ATCC 9027) and Candida albicans (ATCC 10231). Clinical isolates 205 were identified to species-level with Vitek-2 identification cards and all strains were kept 206 frozen in Brain Heart Infusion with 20% glycerol at -70° C until testing. For each experiment 207 microorganisms were subcultured twice in Nutrient agar for bacteria and Sabouraud dextrose 208 agar for *C. albicans*, in order to access culture viability.

209 2.10.2 MIC by broth dilution assay

The antibacterial activity was tested according to the Clinical and Laboratory Standards Institute (CLSI) M7-A6 micromethod and the antifungal activity was tested according to M27-A3 micromethod from the same protocol, after 48 h of incubation at 37 °C $^{24-25}$. Briefly, twofold serial dilutions of extracts were performed in Mueller Hinton broth for bacteria and in RPMI for yeast. Concentrations ranging from 10 µg/mL to 1000 µg/mL were tested.

Microorganism's growth was visually compared for each concentration with the growth control (without plant extract). Minimal inhibitory concentration (MIC) was defined as the lowest extract concentration able to completely inhibit microorganism growth, corresponding to 100% MIC value. All determinations were performed in duplicate for each assay and three independent experiments were run with concordant results.

220 2.11. Statistical analysis of data

221 Data were reported as mean ± standard deviation of at least triplicate experiments. Statistical 222 analysis of the results was performed with SPSS 22.0 (SPSS Inc., Chicago, IL, USA). One-way 223 ANOVA was used to investigate the differences between different samples for all assays. Post 224 hoc comparisons of the means were performed according to Tukey's HSD test. In all cases, P < 225 0.05 was accepted as denoting significance.

226 **3. Results and discussion**

227 3.1 Castanea sativa shell

228 3.1.1 Macronutrient composition

Through the assessment of several physical and chemical parameters it is possible to infer the state of maturation and conservation, as well as the nutritional and commercial values of fruits and their by-products ²⁶. To the best of our knowledge, this is the first study about the ash, moisture, fat, carbohydrates and protein content of *C. sativa* shell, showing promising results for the valorization of this by-product. Table 1 summarizes the macronutrient composition determined in this work.

235

236

Insert Table 1

237

238 Moisture is one of the most representative fractions in all samples, with statistical differences 239 between them. Shell, as a chestnut protection, is expected to have lower moisture content in 240 relation to the fruit. The difference between samples could result from the different climatic 241 conditions of each production region. For example, Minho region presents the highest 242 precipitation levels which might lead to the higher moisture value observed.

243

244 Regarding protein content, it ranged between 2.77 and 3.13 g/100 g, respectively, for Trás-os-245 Montes and Minho. Comparing with chestnut, the content was lower, as the fruit normally 246 presents a 6.0 to 8.6% of protein in their composition. The ash values found in the shell samples were also slightly different from contents described for chestnuts ²⁷⁻²⁸. Concerning 247 the fat content, the chestnut shell from Trás-os-Montes region presented the highest value, 248 statistically different from the other regions. The value found was also lower than the one 249 250 normally described in the fruit ²⁹. The lower protein, ash and fat content in relation to the 251 values described for chestnut fruit could be expected due to the higher presence of others 252 constituents such as dietary fiber, normally composed by cellulose, hemicellulose and lignin, 253 common in a higher proportion in fruit shells.

254

255 **3.1.2 Amino acids content**

The amino acids content evaluated in the chestnuts shell is shown in Table 2.

257

Insert Table 2

Arginine (Arg) and Leucine (Leu) were the predominant essential amino acids (EAA) in the chestnut shell samples accounting for 355 – 721/100 g sample and 159 – 20/ 100 g sample, respectively. Tyrosine (Tyr), phenylalanine (Phe) and threonine (Thr) were present, but in lower amounts, 14 - 20, 63 - 91 and 74 - 125, respectively. Histidine (His) was not detected in any of the analyzed samples. Glutamic and aspartic acid (Asp + Glu) where, in their turns, the predominant non-essential amino acids (NEA) detected. A balanced protein intake is important to maintain a healthy diet and general wellbeing. The development of new supplements rich in amino acids is of major interest for industry and consumers. According to the obtained results, that showed a balanced profile of amino acid in the chestnut shell, the use of this by-product as a possible and cheap source of amino acids should not be neglected.

269

3.1.3 Total Vitamin E and vitamers quantification

270 Vitamin E is the major lipid-soluble antioxidant in the cell antioxidant defence system and is 271 exclusively obtained from the diet. Vitamin E is an essential nutrient, occurring in eight main chemical forms of related structure, tocopherols (α -, β -, γ -, and δ -) and tocotrienols (α -, β -, γ -, 272 and δ -). α -Tocopherol is the most common form found in photosynthetic tissues ³⁰. Vitamin E 273 274 is known as one of the most powerful fat-soluble antioxidant, helping in the promotion of 275 cardiovascular health, enhancement of the immune system function, aid in skin repair and protecting cell membranes from damage caused by free radicals ³¹. Table 3 summarizes the 276 277 Vitamin E profile and total content of *C. sativa* shell.

278

Insert Table 3

Regarding the total vitamin E content, there were statistical differences observed between regions. Trás-os-Montes presented the highest value and Minho the lowest one, with values ranging between 962.8 mg/100 g and 481.5 mg/100 g sample, respectively. Moreover, a considerable difference between the Vitamin E profiles in shells from different regions was found. This variation might be due to climatic differences as tocopherol acts essentially as antioxidant compound, being affected by different conditions such as temperature or humidity. Probably is a response of the plant to environmental stress.

From a qualitative point of view, shells present a vitamin E profile quite similar to chestnut $^{32-}$ 287 33 . Likewise to what is described for the chestnut fruit, Y-tocopherol was the prevailing

vitamer detected in shells, accounting for more than half of the total profile determined for all
studied samples. Interestingly, this vitamer is often present in trace amounts in other natural
products.

291 3.2 Castanea sativa hydro-alcoholic extracts

The extraction yield (expressed as w/w percentages) for chestnut shell from Trás-os-Montes and Minho are globally similar, varying from 12.57 ± 0.01 % to 13.67 ± 0.02 %, which can be accepted as satisfactory ^{27, 34}. In the shell samples collected in Beira-Alta region, the extraction yield was 4.10 ± 0.10 %, considerably lower than those obtained for other samples. Indeed, comparing the results with the ones described by Barreira *et al.* (2008), the extraction yield obtained was inferior, what could be explained by the higher polarity of water, the extractor solvent used by those authors ³⁸.

299

300 **3.2.1 Total phenolic, total flavonoid content and antioxidant activity**

Table 4 presents the total phenolic and flavonoid contents, the EC₅₀ values and the antioxidant
 activity based on FRAP assay obtained for all shell extracts.

303 Insert Table 4

304 Among all extracts analyzed, the chestnut shell from Trás-os-Montes presented the highest 305 value (796.80 ± 1.07 mg GAE/g db), followed by Minho and Beira-Alta. The significant 306 differences observed can be explained on the basis of environmental conditions, as all samples 307 were treated in the same away. Minho and Beira-Alta are rainfall regions, compared to Trás-308 os-Montes, which is more dry and arid. These factors can affect the environmental and 309 ecological quality of the soil, such as water regulation and biodiversity. The temperature rise, 310 associated with conditions of greater dryness, may also have a direct effect on organic matter 311 decline, due to a lower production of biomass and mineralization. The variation observed

between the extracts from the different regions may reflect that. Indeed, organic compounds and the development of defenses to combat stress caused by situations of unfavorable weather conditions, may enhance biological differences in composition between the three regions, justifying their antioxidant properties variations ³⁵.

316 The results obtained for the chestnut shell extracts from Trás-os-Montes were in accordance 317 with the ones presented by Barreira et al. (2010), who also evaluated the TPC of chestnut shell of different cultivars from the same region, using water as solvent extractor ³⁶. In turn, the TPC 318 values presented by Vázquez et al. (2008), that extracted chestnut shell with water and 319 different organic solvents, were lower ⁵. In this sense, it is possible to infer that the use of 320 321 ethanol and water as extraction solvents achieves a reasonable content of TPC in the extract, 322 and simultaneously comply with sustainability concept as it uses green chemicals. The values 323 obtained can also be compared to other C. sativa by-products, like chestnut bur, for which 324 considerably lower TPC values have been reported. This reinforces the potentialities of chestnut shell as a source of bioactive compounds³⁷. The obtained TPC content is considerable, 325 326 revealing the great potential of the assayed by-products extracts as a new source of antioxidant compounds 6-7. 327

Flavonoids are an important group of phenolic compounds, contributing to the antioxidant activity of plant products. In this assay, the highest TFC values were found in sample extracts collected in Trás-os-Montes. Barreira *et al.* (2008) found a TFC slightly higher for shell extracts from Trás-os-Montes that could be explained by natural seasonal variations and also by the use of different solvents ³⁸.

Regarding EC₅₀ values, Trás-os-Montes shell extracts also presented the best results (31.80 \pm 1.13 µg/ml), however no statistic differences (P > 0.05) were observed between the different regions. Once more, the values found were slight inferior to the values reported by Barreira *et*

336 al. (2010), whose EC₅₀ values for chestnut shell extracts ranged between 82.41 and 159.99 337 μ g/mL for the same sample ³⁶. Regarding the evaluation of the antioxidant capacity of extracts 338 based on FRAP assay, the obtained results showed a high antioxidant activity of all regions 339 tested, being Trás-os-Montes again the best (8083.50 \pm 164.80 μ mol of ferrous sulphate/ g db). The values were higher than those reported in other works ^{6-7, 36}, strengthening once more the 340 341 potential of the chestnut shell. The antioxidant capacity of plant products is usually related to 342 a high phenolic content, and also to a higher vitamin E contents. In this work, the highest 343 antioxidant activity was registered for the extracts that simultaneously presented the higher 344 phenolic and flavonoid contents (samples from Trás-os-Montes), which reinforces the 345 importance of these compounds to the antioxidant activity.

346 **3.2.2 Antimicrobial activity**

The minimum inhibitory concentrations (MICs) for the active plant extracts were determined using modified methods of the CLSI against microorganisms selected to cover different Grampositive bacteria (*S. aureus* and *S. epidermidis*), Gram negative bacteria (*E. coli, K. pneum*oniae, *P. aeruginosa*) and yeasts (*C. albicans*), that normally inhabit or proliferate in the human body. The microorganism's strains were selected considering possible future application of this food by-product in different industries such as pharmaceutical or cosmetic. Results did not reveal inhibitory activity against any of the microorganisms tested.

354

355 **4. Conclusion**

The use of *C. sativa* shell as a source of functional compounds and their application for different purposes, such as pharmaceutical, food supplementation or even cosmetic, is a promising field which requires interdisciplinary research, taking into account sustainability concerns.

360 In this work, Castanea sativa shell from different Portuguese regions (Minho, Trás-os-361 Montes and Beira-Alta) and their hydro-alcoholic extracts were analyzed and compared. Shell 362 extracts from Trás-os-Montes presented the highest TPC, TFC, and antioxidant capacity 363 confirmed by their highest value in the FRAP assay and lower EC₅₀. Regarding the amino acid 364 content, all extracts are rich in essential amino acids. The vitamin E results were also very 365 promising. However, none of the extracts presented antimicrobial activity. The results of this 366 work, thus, suggested that C. sativa shell extracts, especially from Trás-os-Montes, could be a 367 new promising active ingredient with potential applications, such as in food supplements. 368 Further investigations are now needed to establish the citotoxicity of these extracts in 369 different cell lines, such as Caco-2 cell model or fibroblasts, in order to test their incorporation 370 in different food and/or cosmetic products.

371

372 Conflict of interest

The authors declare no conflict of interest. This article does not contain any studies withhuman or animal subjects.

375

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383

384 **References**

385 386	1.	Portuguese National Institute of Statistics IP. ISSN 0079-4139; ISBN 978-989-25-0198-7
387	2	B R Cruz A S Abraão A M Lemos and F M Nunes Carbobydr Polym 2013 94 594-
388	۷.	602.
389	3.	N. Braga, F. Rodrigues and M. B.P.P.Oliveira, Nat Prod Res, 2014, 1-18.
390	4.	G. Vázquez, J. González-Alvarez, J. Santos, M. S. Freire and G. Antorrena, Ind Crop Prod,
391		2009, 29 , 364-370.
392	5.	G. Vázquez, E. Fontenla, J. Santos, M. S. Freire, J. González-Álvarez and G. Antorrena,
393		Ind Crop Prod, 2008, 28 , 279-285.
394	6.	F. Rodrigues, A. Palmeira-de-Oliveira, J. das Neves, B. Sarmento, M. H. Amaral and M.
395		B. P.P.Oliveira, Ind Crop Prod, 2013, 49 , 634-644.
396	7.	F. Rodrigues, A. Palmeira-de-Oliveira, J. das Neves, B. Sarmento, M. H. Amaral and M.
397		B. P. P. Oliveira, <i>Pharm Biol</i> , 2014, 4 , 1-9.
398	8.	A. S. G. Costa, R. C. Alves, A. F. Vinha, S. V. P. Barreira, M. A. Nunes, L. M. Cunha and
399		M. B. P. P. Oliveira, Ind Crop Prod, 2014, 53 , 350-357.
400	9.	L. T. Dinis, F. Peixoto, T. Pinto, R. Costa, R. N. Bennett and J. Gomes-Laranjo, Environ
401		Exp Bot, 2011, 70 , 110-120.
402	10.	AOAC, in Official methods of analysis of AOAC International, editor W. Horwitz,
403		Gaithersburg. USA, 17th edition, 2000, chapter 32, p. 2.
404	11.	AOAC, Ash of flour (923.03) in Official methods of analysis of AOAC International,
405		editor W. Horwitz, Gaithersburg. USA, 17 th edition, 2000.
406	12.	AOAC, Fat (crude) (920.39) in Official methods of analysis of AOAC International, editor
407		W. Horwitz, Gaithersburg. USA, 17 th edition, 2000.
408	13.	K. Tontisirin, W. C. MacLean and P. Warwick, Food and Agriculture Organization of the
409		United Nation editor, Rome, Italy, 2003, pp. 1-66.
410	14.	R. C. Alves, S. Casal and M. B. P. P. Oliveira, Food Sci Technol Int, 2009, 15, 57-63.
411	15.	M. Fountoulakis and HW. Lahm, <i>J Chromatogr A</i> , 1998, 826 , 109-134.
412	16.	A. M. G. Paramás, J. A. G. Bárez, C. C. Marcos, R. J. García-Villanova and J. S. Sánchez,
413		Food Chem, 2006, 95 , 148-156.
414	17.	J. L. Navarro, M. Aristoy and L. Izquierdo, Rev Agroquim Tecnol, 1984, 24, 85-93.
415	18.	F. B. Pimentel, R. C. Alves, A. S. G. Costa, D. Torres, M. F. Almeida and M. B. P. P.
416		Oliveira, Food Chem, 2014, 149 , 144-150.
417	19.	V. L. Singleton and J. A. J. Rossi, <i>Am J Enol Vitic</i> , 1965, 16 , 144-158.
418	20.	R. Alves, A. Costa, M. Jerez, S. Casal, J. Sineiro, M. Núñez and M. B. P. P. Oliveira, J Agr
419		Food Chem, 2010, 58 , 12221-12229.
420	21.	R. Guimarães, L. Barros, J. C. M. Barreira, M. J. Sousa, A. M. Carvalho and I. C. F. R.
421		Ferreira, <i>Food Chem Toxicol</i> , 2010, 48 , 99-106.
422	22.	L. Barros, P. Baptista and I. C. F. R. Ferreira, Food Chem Toxicol, 2007, 45, 1731-1737.
423	23.	I. F. F. Benzie and J. J. Strain, in <i>Methods in Enzymology</i> , In: Lester P, editor. Methods
424		in Enzymology: Academic Press, 1999, 299, 15-27.
425	24.	C.L.S.I. Clinical Laboratory Standards Institute, NCCLS editor, Wayne, Pennsylvania,
426		2008.
427	25.	M. Wikler, D. Low, F. Cockerill, D. Sheehan, W. Craig, F. Tenover, M. Dudley, J.
428		Turnidge, G. Eliopoulos, M. Weinstein, D. Hecht, B. Zimmer, J. Hindler, M. Ferraro and
429		J. Swenson, NCCLS, Wayne, Pennsylvania., 2003, 26.
430	26.	A. F. Vinha, R. C. Alves, S. V. P. Barreira, A. Castro, A. S. G. Costa and M. B. P. P.
431		Oliveira, <i>LWT - Food Sci Technol</i> , 2014, 55 , 197-202.
432	27.	O. Borges, B. Gonçalves, J. L. S. de Carvalho, P. Correia and A. P. Silva, Food Chem,
433		2008, 106 , 976-984.

- 434 28. S. Pereira-Lorenzo, A. M. Ramos-Cabrer, M. B. Díaz-Hernández, M. Ciordia-Ara and D.
 435 Ríos-Mesa, *Sci Hort*, 2006, **107**, 306-314.
- 436 29. B. Gonçalves, O. Borges, H. S. Costa, R. Bennett, M. Santos and A. P. Silva, *Food Chem*,
 437 2010, **122**, 154-160.
- 438 30. M. Hasanuzzaman, K. Nahar and M. Fujita, in *Emerging Technologies and Management*439 *of Crop Stress Tolerance*, P. Ahmad and S. Rasool editors, Academic Press, San Diego,
 440 2014, 267-289.
- 441 31. O. Guralp, *Maturitas*, 2014, **79**, 476-480.
- 442 32. M. D. Zlatanov, G. A. Antova, M. J. Angelova-Romova and O. T. Teneva, J Sci Food
 443 Agric, 2013, 93, 661-666.
- 33. J. C. M. Barreira, R. C. Alves, S. Casal, I. C. F. R. Ferreira, M. B. P. P. Oliveira and J. A.
 Pereira, *J Agric Food Chem*, 2009, **57**, 5524-5528.
- 446 34. J. De La Montaña Míguelez, M. Míguez Bernárdez and J. M. García Queijeiro, *Food* 447 *Chem*, 2004, **84**, 401-404.
- 448 35. Strategies for adaptation of agriculture and forestry to climate change. In: Ministry of
 449 Agriculture S, Environment and Territory, editor. Lisbon: Portuguese Government,
 450 2013, 1-88.
- 451 36. J.C.M. Barreira, I. C. F. R. Ferreira, M. B. P. P. Oliveira and J. A. Pereira, *Food Sci Technol*452 *Int*, 2010, **16**, 209-216.
- 453 37. E. M. Balboa, M. L. Soto, D. R. Nogueira, N. González-López, E. Conde, A. Moure, M. P.
 454 Vinardell, M. Mitjans and H. Domínguez, *Ind Crop Prod*, 2014, **58**, 104-110.
- 455 38. J. C. M. Barreira, I. C. F. R. Ferreira, M. B.P. P. Oliveira and J. A. Pereira, *Food Chem*,
 456 2008, **107**, 1106-1113.

Table Captions

Table 1 – Nutritional composition of the *C. sativa* shell from different production regions (mean \pm standard deviation). Different letters (a, b, c) in the same line indicate significant differences between mean values (P < 0.05).

Table 2 - Amino acids content (mg/100 g sample) of the shell of chestnut samples from different production regions (mean \pm standard deviation). Different letters (a, b, c) in the same line indicate significant differences between mean values (P < 0.05). nd – not detectable.

Table 3 – Vitamin E total content and vitamers (mg / 100 g sample) of chestnut shell from different regions (mean \pm standard deviation). Different letters (a, b, c) in the same line indicate significant differences between mean values (P < 0.05). nd – not detectable.

Table 4 – Total polyphenol content (TPC), Total flavonoid content (TFC), Radical scavenging activity (RSA) EC_{50} values and antioxidant activities based on their abilities to reduce ferric iron (Fe³⁺) to ferrous iron (Fe²⁺) in the three hydro-alcoholic shell extracts. Values are expressed as means ± standard deviation (n = 6). GAE, gallic acid equivalents. CEQ, cathechin equivalents. Different letters (a, b, c) in the same column indicate significant differences between mean values (P < 0.05).

	Region			
	Minho	Trás-os-Montes	Beira-Alta	
Moisture (g/100g db)	38.61 ±1.78ª	21.29±1.56 ^c	26.21±0.40 ^b	
Ash (g/100g db)	1.60 ± 0.06	1.36 ± 0.04	1.08 ± 0.04	
Proteins (g/100g db)	3.13 ± 0.10	2.77 ± 0.00	2.89 ± 0.17	
Fat (g/100g db)	0.15 ± 0.00^{b}	0.52 ± 0.05^{a}	0.23 ± 0.02^{b}	
Carbohydrates (g/100 g db)	56.51	74.06	70.41	

		Zone			
Amino acids	Minho	Trás-os- Montes	Beira-Alta		
Essential amino acids (EAA)					
His	nd	nd	nd		
lle	132 ± 10^{a}	143 ± 4^{a}	99 ± 10^{b}		
Leu	208 ± 16^{b}	246 ± 10^{a}	159 ± 15 [°]		
Lys	132 ± 11 ^b	156 ± 4^{a}	109 ± 1^{c}		
Phe	79 ± 7 ^b	91 ± 4^{a}	63 ± 6 ^b		
Tyr	20 ± 2^{a}	19 ± 1^{a}	14 ± 7 ^b		
Thr	101 ± 5^{b}	125 ± 9^{a}	74 ± 6^{c}		
Arg	721 ± 71 ^ª	715 ± 78 ^ª	355 ± 19 ^b		
Val + Met	51 ± 4^{a}	58 ± 1ª	38 ± 4 ^b		
Non-essential am	nino acids (NEAA)			
Ala	84 ± 6^{b}	95 ± 4^{a}	51 ± 4^{c}		
Ser	121 ± 12 ^b	160 ± 13^{a}	106 ± 4^{b}		
Asp + Glu	403 ± 38 ^b	484 ± 32^{a}	268 ± 12 ^c		
Gly	71 ± 4 ^b	79 ± 4^{a}	51 ± 1 ^c		
Pro	182 ± 15 ^b	241 ± 9 ^a	185 ± 22 ^b		
Orn	9 ± 1^{a}	1 ± 1^{b}	2 ± 1 ^b		

	Region			
Compound analyzed	Minho	Trás-os-Montes	Beira-Alta	
α-tocopherol	65.94 ± 2.83 ^c	190.8 ± 0.00^{a}	162.1 ± 0.90 ^b	
β-tocopherol	12.00 ± 0.10^{b}	14.2 ± 0.30^{a}	$6.7 \pm 0.10^{\circ}$	
δ-tocopherol	29.86 ± 0.83^{a}	69.3 ± 0.70^{b}	33.6 ± 0.40^{b}	
Y-tocopherol	228.58 ± 0.85 ^c	670.4 ± 0.20^{a}	356.5 ± 1.70^{b}	
α-tocotrienol	19.37 ± 0.28 ^a	8.6 ± 0.10^{b}	$4.3 \pm 0.00^{\circ}$	
β-tocotrienol	34.56 ± 2.55	nd	nd	
Ύ-tocotrienol	67.79 ± 0.90	nd	nd	
δ -tocotrienol	23.35 ± 1.18 ^ª	9.6 ± 0.10^{b}	$4.3 \pm 0.10^{\circ}$	
Total	481.5 ± 1.19	962.8 ± 0.23	567.5 ± 0.53	

Region	TPC (mg GAE/ g db)	TFC (mg CEQ / g db)	EC₅₀ (μg/mL)	FRAP (μmol of ferrous sulphate/g db)
Minho	241.90 ± 0.26^{b}	$31.38 \pm 2.88^{\circ}$	37.61 ± 5.08 ^ª	6891.20 ± 355.00 ^b
Trás-os-Montes	$796.80 \pm 1.07^{\circ}$	43.33 ± 1.00 ^a	31.80 ± 1.13^{a}	8083.50± 164.80 ^a
Beira-Alta	$143.00 \pm 0.57^{\circ}$	37.95 ± 1.52 ^b	36.81 ± 4.55^{a}	6008.70± 138.60 ^c

Sustainability / Re-Use



Castanea sativa shell



Macronutrients analysis

Vitamin E and vitamers

Amino acids

Antioxidant activity

Antimicrobial activity