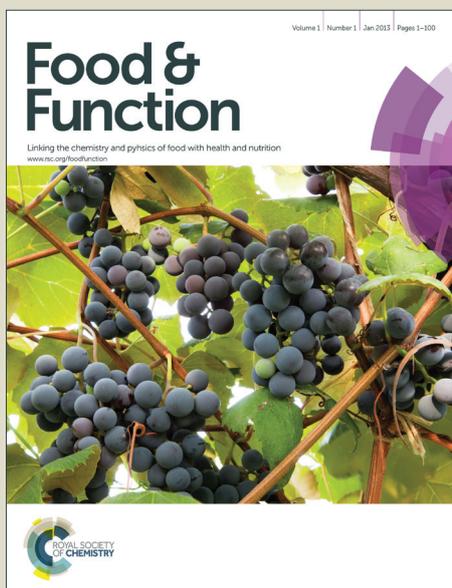


Food & Function

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1 ***In vitro* fermentation of chewed mango and banana: Particle size, starch and**
2 **vascular fibre effects**

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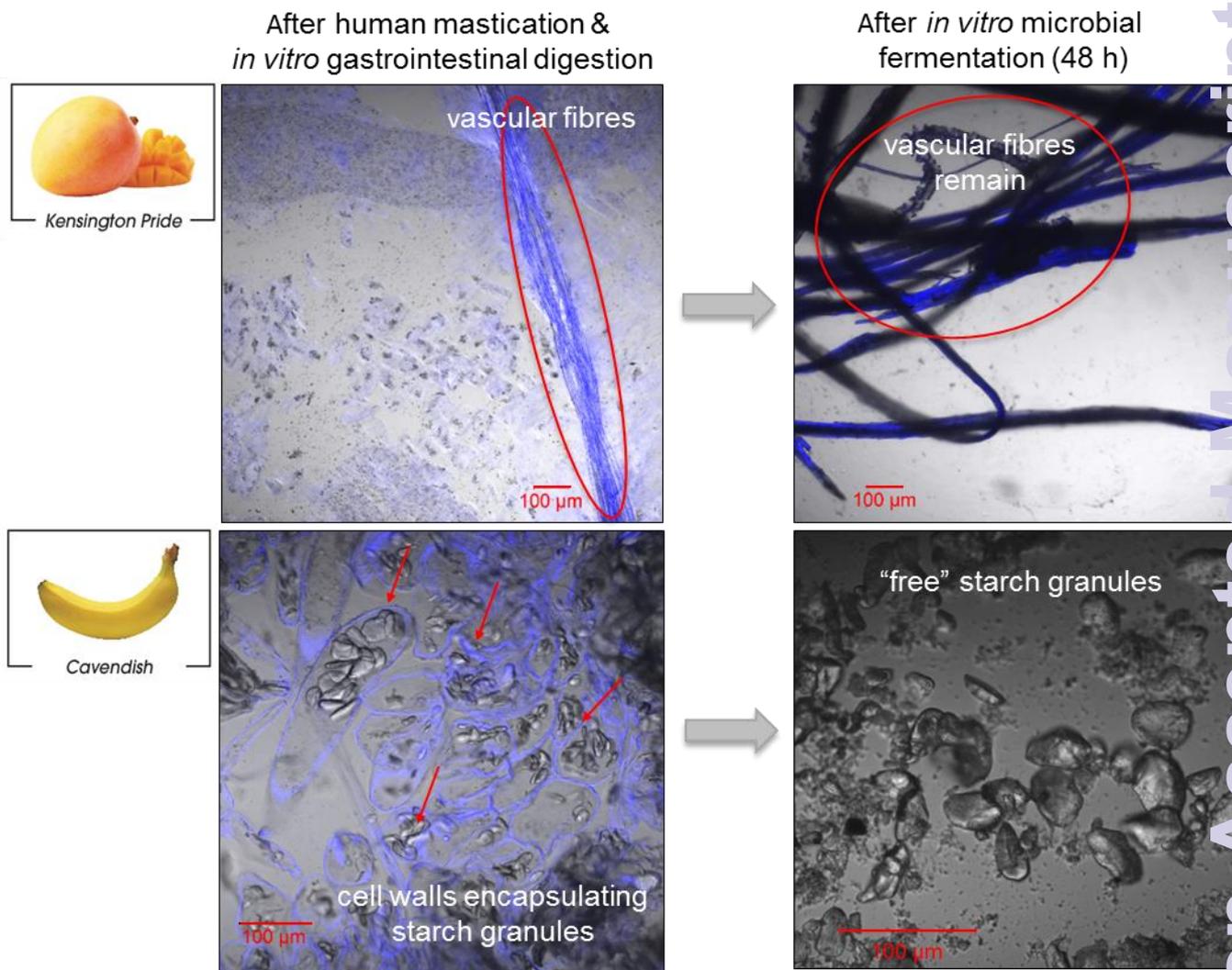
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15 **Table of contents entry**

16

17 Colour graphic:



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20

21 The presence of resistant starch in chewed banana and vascular fibres in chewed mango,
 22 have greater effects on microbial fermentation kinetics than particle size.

23 **Abstract**

24 Fruits (and vegetables) contain cellular structures that are not degraded by human
25 digestive enzymes. Therefore, the structure of the insoluble fraction of swallowed fruits is
26 mostly retained until intestinal microbial fermentation. *In vitro* fermentation of mango and
27 banana cell structures, which survived *in vivo* mastication and *in vitro* gastrointestinal
28 digestion, were incubated with porcine faecal inoculum and showed intensive metabolic
29 activity. This included degradation of cell walls, leading to the release of encapsulated cell
30 contents for further microbial metabolism. Production of cumulative gas, short chain fatty
31 acids and ammonia were greater for mango than for banana. Microscopic and
32 spectroscopic analyses showed this was due to a major fermentation-resistant starch
33 fraction present in banana, that was absent in mango. This study demonstrated distinctive
34 differences in the fermentability of banana and mango, reflecting a preferential degradation
35 of (parenchyma) fleshy cell walls over resistant starch in banana, and the thick cellulosic
36 vascular fibres in mango.

37

38 **Keywords:** microbial fermentation, gas production, mango, banana, resistant starch,
39 cellulose.

40 1. Introduction

41 *In vivo* investigations of dietary polysaccharide fermentation in the human colon can be
42 challenging due to inaccessibility, inconsistency and limitations of dietary control for
43 human volunteers ¹. There is continued interest in the development of relevant *in vitro*
44 models for the digestive process, but uncertainty exists as to how to best to represent the
45 unit processes involved. Non-invasive *in vitro* colonic fermentation models can be used to
46 monitor differences in substrate fermentability before and after gastrointestinal digestion ²,
47 ³, and to elucidate the potential role of microbiota in the metabolism of partially and/or non-
48 digestible components of the diet such as dietary fibre. *In vitro* colonic models involving
49 faecal or caecal microbiota of human, rat and pig have proved useful for investigating
50 metabolic processes mediated by intestinal microbiota ⁴. For comparative purposes, it is
51 important that the microbiota used as an inoculum is from a well-defined source.

52
53 Selected fruits and vegetables have been examined *in vitro* using colonic fermentation
54 techniques ⁵⁻⁹. The volume of gas produced as a result of fermentation acts as an indicator
55 of potential fermentability of the substrate in response to a controlled microbial population.
56 The predominant gas (CO₂) is derived from primary fermentation and the reaction of acidic
57 fermentation end-products with basic bicarbonate ions ^{10, 11}. Short chain fatty acids (SCFA)
58 such as acetic, propionic and butyric acids are major products of carbohydrate
59 fermentation, and are known to be beneficial in terms of energy contribution and health ¹²,
60 whereas ammonia (NH₃), one of the end-products of protein and peptide fermentation ¹³,
61 has potentially negative effects on the long-term health of colonic epithelial cells ¹⁴.

62
63 Previous authors have used a variety of pre-treatments for fruits and vegetables in *in vitro*
64 digestion studies including milling or grinding with hammer mills, blenders, mortars and
65 pestles ¹⁵⁻¹⁷, and homogenisation ¹⁸⁻²¹, often preceded by air-drying ²² or lyophilisation ²³,
66 ²⁴. In addition, wet liquid samples such as purees or juices have been prepared ^{20, 25}.
67 However, these high-shear techniques result in the disintegration and collapse of most of
68 the cellular structures, which does not necessarily simulate the human mastication
69 process, consequently leading to significant overestimates in phytonutrient bioaccessibility.
70 For example, the carotenoid bioaccessibility of pureed mango is significantly higher and
71 twice that of *in vivo* masticated mango fractions of varying particle sizes ²⁶.

72
73 The aim of this study was to investigate the fermentation kinetics and end-products of
74 fresh mango and banana flesh using a standardised batch fermentation model (with a

75 faecal inoculum obtained from pigs fed a well-defined diet) after *in vivo* human mastication
76 and *in vitro* digestion processing. In this context, the fruits were prepared in such a way
77 that the state and condition of these samples at the start of colonic fermentation were as
78 physiologically comparable as possible to the microstructures achieved at the beginning of
79 the colon during human consumption, *i.e.* using minimal artificial processing. Following
80 mastication, samples were size-fractionated to allow comparison of the effect of particle
81 size on subsequent fermentation behaviour.

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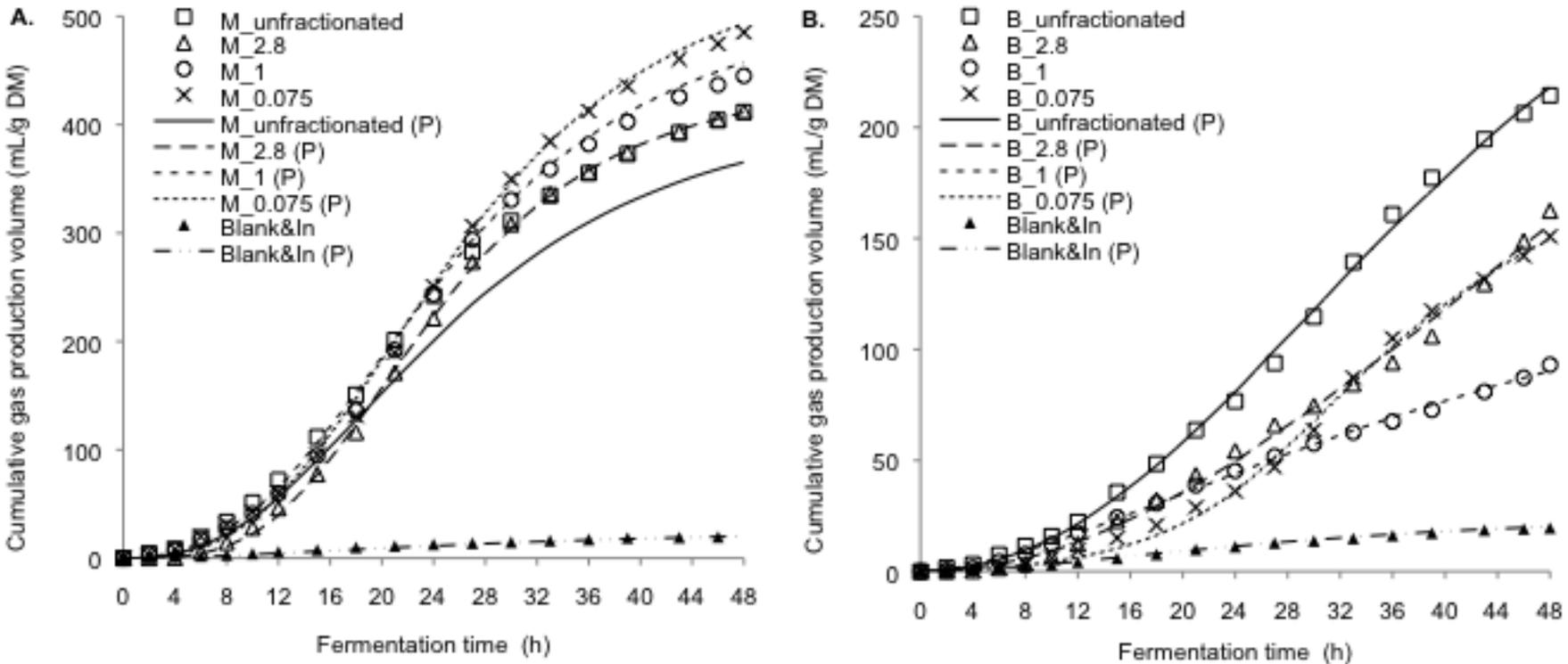
83 2. Results

84 2.1. Fermentation kinetics of mango and banana

85 The cumulative gas production profiles (DMCV) for mango and banana are shown in Fig 1.
86 Measured data points fitted well to mathematical predictions of Groot's model ²⁷ for all
87 substrate types except for the unfractionated mango where the measured gas values were
88 higher than the predicted curve. A comparison of replicates (n=4 per particle size per fruit)
89 indicated that the sample bottles 1, 2, 7, 11, 15 and 17 behaved as outliers (Fig S1,
90 supplementary data); therefore, these data were excluded from Proc GLM analysis to
91 avoid false means because the high variation in raw data contributed to exaggerated
92 estimates for $T_{1/2}$, TR_{max} and R_{max} values based on the curve fitting results. Each substrate
93 fraction consisted of a range of particle sizes, for example, the 2.8 mm fraction contains
94 masticated mango or banana particles ranging from 2.8-5.6 mm while the 1 mm fraction
95 consists of particles from 1-2.8 mm and the 0.075 mm fraction consists of particles from
96 0.075-1 mm. Therefore, there might be heterogeneity in each fraction as a result of the
97 biological mastication process and/or during sub-sampling of these heterogeneous
98 samples into individual bottles as replicates, leading to accumulated variation in individual
99 bottles. Updated DMCV₄₈ means are shown in Fig 1A and 1B. Interestingly, there was an
100 apparent opposite trend for mango and banana as a function of particle size, although the
101 absolute differences in gas production volume between particle sizes were small. An
102 inverse relationship between particle size and gas production was observed for mango, in
103 contrast to banana where unfractionated and larger banana particles showed faster and
104 more extensive gas production. The higher distribution of larger bolus particles relative to
105 smaller particles in the unfractionated banana (mixture of particle sizes) is likely to have
106 contributed to the higher gas production volume. All substrates started with an initial lag
107 phase of 2-6 h, suggesting an adaption time is required for physical adhesion of cellulolytic
108 microbial species to the fibrous plant cell wall components ^{28, 29}.

109

110 Mango was more readily fermentable and to a greater extent, as the total gas production
111 for mango (440 mL/g DM) was significantly greater than for banana (113 mL/g DM)
112 ($P<0.0001$) (Table S1, supplementary data). Mango reached its maximum rate of gas
113 production ($R_{\max}=17$ mL/h) at 19 h while banana only reached its maximum rate ($R_{\max}=3$
114 mL/h) after 31 h, reflecting a significant fruit effect ($P<0.0001$). In addition, the half-time to
115 reaching asymptotic gas production differed significantly between mango and banana (25
116 h and 54 h respectively, $P<0.0001$). Neither the end-point gas production (48 h) nor R_{\max}
117 were significantly different between particle sizes ($P>0.05$), but $T_{1/2}$ and TR_{\max} occurred
118 significantly later for the 0.075 mm and 2.8 mm particles ($P=0.02$ and $P=0.0002$
119 respectively). The effects of fruit type and particle size, and any interactions of the
120 fermentation kinetic parameters ($DMCV_{48}$, $T_{1/2}$, TR_{\max} , R_{\max}), are shown in Table S1. In
121 banana, the $T_{1/2}$ and R_{\max} of 1 mm particles were not significantly different ($P>0.05$) from
122 the other particle sizes. However, the time at which the 1 mm particles reached the
123 maximum rate of gas production ($TR_{\max} = 19$ h) was significantly shorter ($P<0.0001$) than
124 the other sizes ($TR_{\max} = 29-41$ h) (Table S2), suggesting substrate heterogeneity from the
125 mastication process and/or from mixing complications before sub-sampling. As
126 fermentation of the substrates apparently would have extended beyond 48 h, gas
127 production asymptotes were extrapolated rather than observed (Fig 1), particularly for
128 banana.



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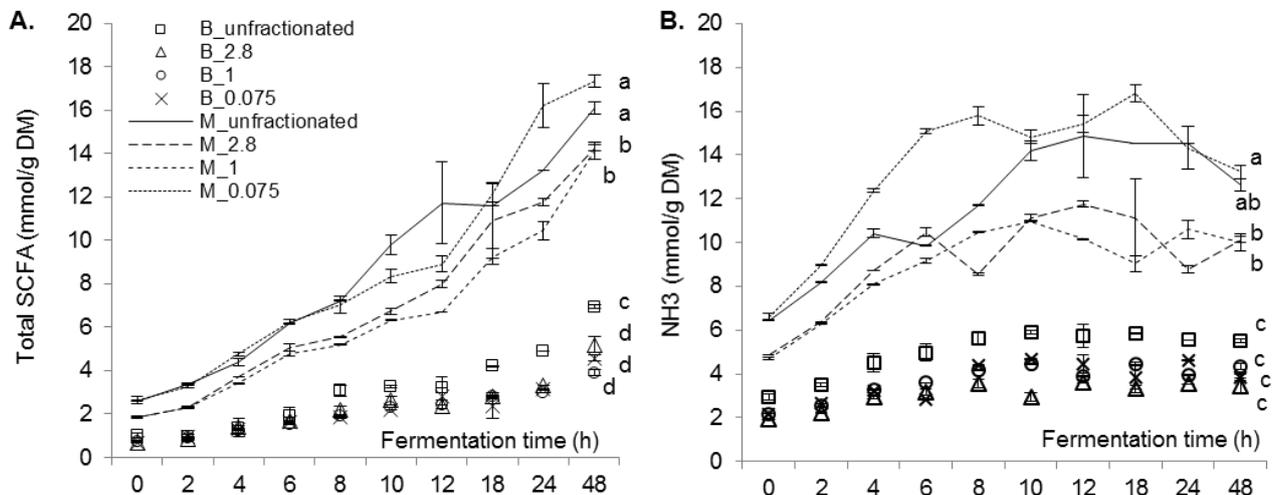
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Figure 1. (Measured) cumulative gas production volume (DMCV) profiles of masticated (A) mango and (B) banana data points (□, Δ, O, x) and experimental blank (▲), and fitted monophasic curve (P) predictions (—, ---, - - -, ---, - · -) according to Groot’s mathematical model (n=4 for each particle size- unfractionated, 2.8 mm, 1 mm, 0.075 mm). DMCV values have been corrected for dry matter content per substrate fermentation bottle during 48 h microbial fermentation *in vitro*. Data is expressed as means±standard deviation. Note the different y-axis scales for the two profiles. M, mango; B, banana; Blank&In, blank containing only inoculum and medium.

135 **2.2. pH, SCFA, BCR and NH₃ in mango and banana**

136 At the end of fermentation, the pH values for mango and banana ranged from 6.15-6.55
 137 (Table S1). There was no significant difference ($P>0.05$) in pH either due to fruit or particle
 138 size, indicating^{30, 31} that the buffering capacity of the medium was sufficient for the
 139 fermentations taking place.

140



141

142 Figure 2. (A) Total short chain fatty acids (SCFA) and (B) ammonia (NH₃) production
 143 profiles of mango (□, Δ, O, x) and banana (—, — —, - - -, ---) particles (n=4 for each
 144 particle size- unfractionated, 2.8 mm, 1 mm, 0.075 mm) during 48 h microbial fermentation
 145 *in vitro*. SCFA and NH₃ concentrations are reported as mmol/g dry matter (DM). Data are
 146 expressed as means±standard deviation. ^{a,b,c,d} Different letters within substrates denote
 147 significance differences for end-point values (48 h) at $P<0.05$. M, mango; B, banana.

148

149 Changes in total SCFA and NH₃ concentrations with time are shown in Fig 2. All mango
 150 fractions consistently produced significantly larger amounts of total SCFA in comparison to
 151 banana ($P<0.0001$), which is in agreement with the retarded fermentability of banana as
 152 evidenced by the lower volume and rate of gas production. The total SCFA concentration
 153 initially started from 0.7 mmol/g and showed a gradual increase over 48 h for both fruit
 154 types. Small but significant particle size effects were observed for the total SCFA
 155 ($P<0.0001$). The unfractionated and smallest particles (0.075 mm) produced significantly
 156 higher total SCFA (12.4 and 11.9 mmol/g DM) than the 2.8 mm and 1 mm particles (10.4
 157 and 9.6 mmol/g DM). This trend was also observed for the major individual SCFA- acetic,
 158 butyric and propionic acids (Table S1).

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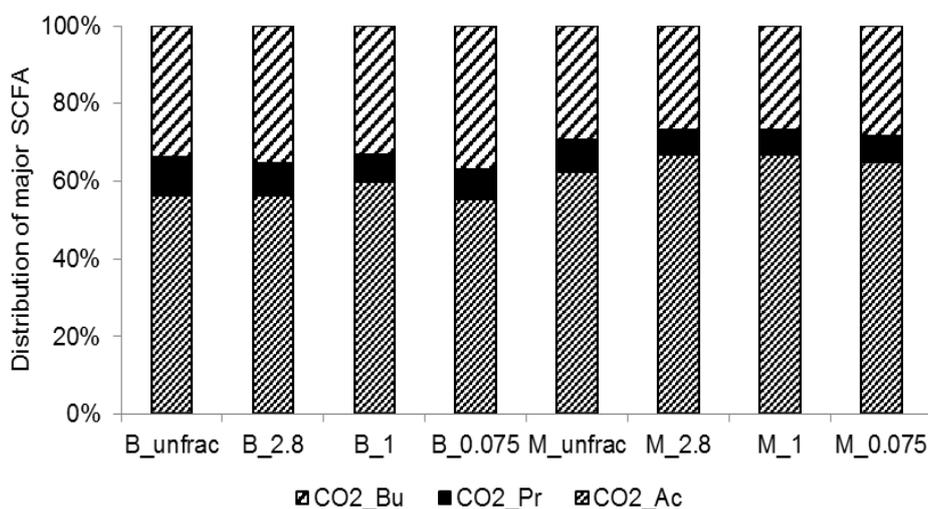
160 NH₃ concentrations were at least two-fold higher for all mango particle sizes compared to
 161 banana ($P=0.0026$). Similarly to SCFA production, the unfractionated and 0.075 mm
 162 mango particles had significantly higher NH₃ concentrations than those for banana.

163 However, there was no significant difference between particle sizes for banana ($P>0.05$).
 164 The concentration of NH_3 production peaked between 8 and 18 h, and then declined. From
 165 0 h, NH_3 concentrations of >2 mmol/g DM measured for both fruits suggests that some
 166 bacterial species within the porcine faecal microbiota have started actively fermenting
 167 some peptide/amino acid source present in the inoculum and/or medium.

168

169 Following the usual pattern for gut fermentation, acetic acid was the major SCFA produced
 170 (54-66%) in both fruits, followed by propionic (13-19%) and butyric acids (9-17%) (Fig 3),
 171 whereas valeric, isovaleric and isobutyric acids were minor SCFA, collectively accounting
 172 for $<10\%$. These SCFA concentrations (mmol/g DM) were subsequently converted into
 173 acetic acid equivalents (AAE) (Table S1) using their respective molar mass to obtain a
 174 branched-chain ratio (BCR). The BCR gives an indication of the proportion of SCFA likely
 175 to be related to protein fermentation⁴. Banana fermentation was associated with a higher
 176 proportion of branched-chain SCFA (isobutyric, isovaleric and valeric acids) to straight
 177 chain acids (acetic, propionic and butyric acids), leading to a significantly higher BCR
 178 ($P<0.0001$) for banana than for mango.

179



180

181 Figure 3. % Distribution of individual major short chain fatty acids (SCFA): acetic acid (Ac),
 182 propionic acid (Pr) and butyric acid (Bu) in mango and banana particles ($n=4$ for each
 183 particle size: unfractionated, 2.8 mm, 1 mm, 0.075 mm). M, mango; b, banana; CO_2 ,
 184 carbon dioxide.

185

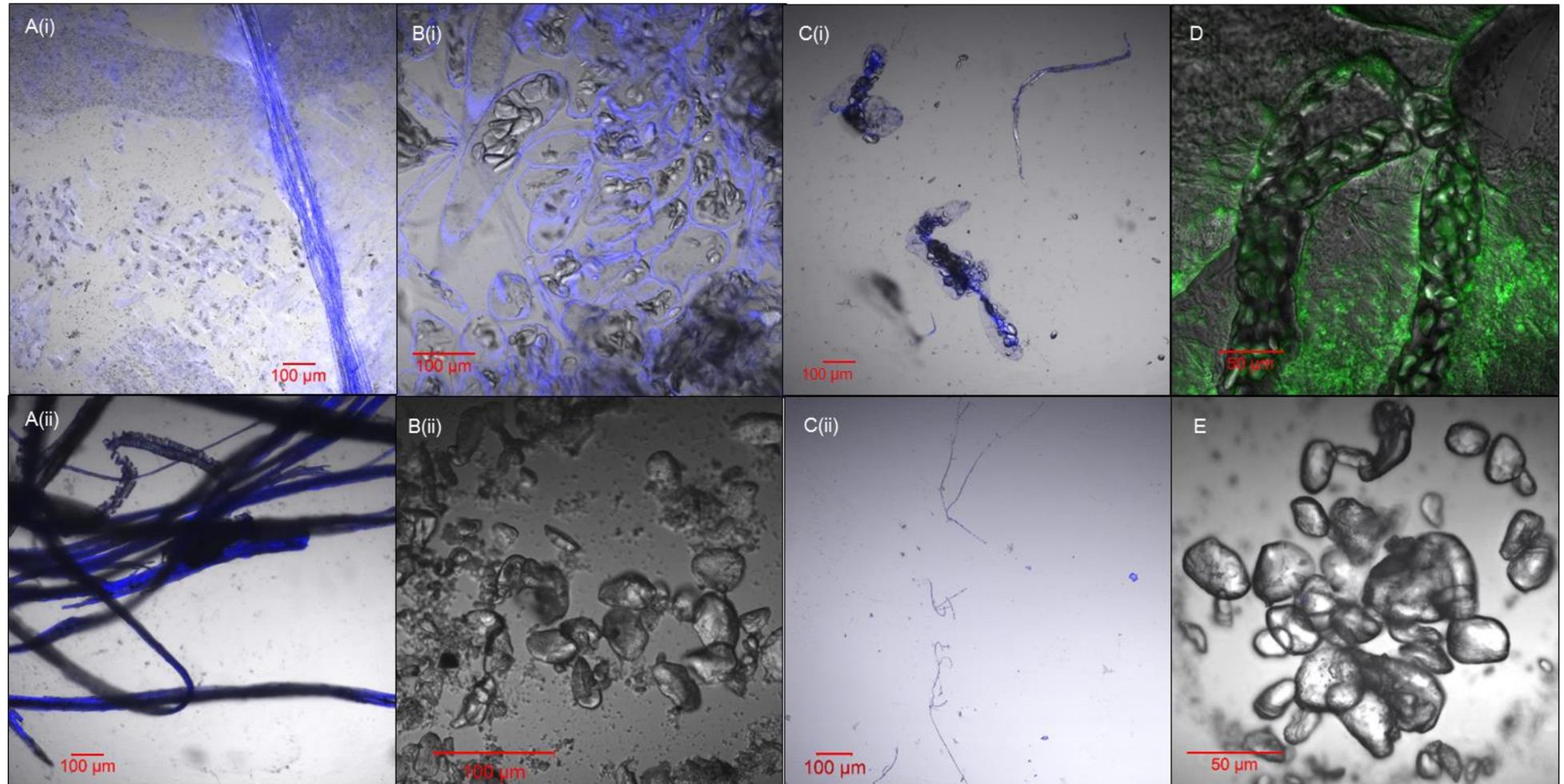
186 2.3. Physical structures and major polysaccharide composition affecting 187 fermentation

188 From confocal microscopy, it could be seen that masticated mango particles subjected to
 189 *in vitro* gastrointestinal digestion contained soft parenchyma (fleshy) tissue, which
 190 disappeared after microbial fermentation, leaving mostly cellulosic vascular fibres (Fig 4A).

191 This was confirmed by ^{13}C CP/MAS NMR spectra of fermented 2.8 mm mango particles (Fig
192 5A) where the cellulose C-1 peak (dominant signal at 105 ppm ^{32, 33}) remained after
193 fermentation, indicating that it was not well fermented. This is consistent with the
194 micrographs (Fig 4A) where these vascular fibres are structured but loosely attached to
195 the rest of the sample material. Vascular fibres were also present in banana after
196 fermentation, but these were less pronounced than in mango, evidently thinner (Fig 4Cii),
197 and were not resolved from more major peaks by NMR.

198

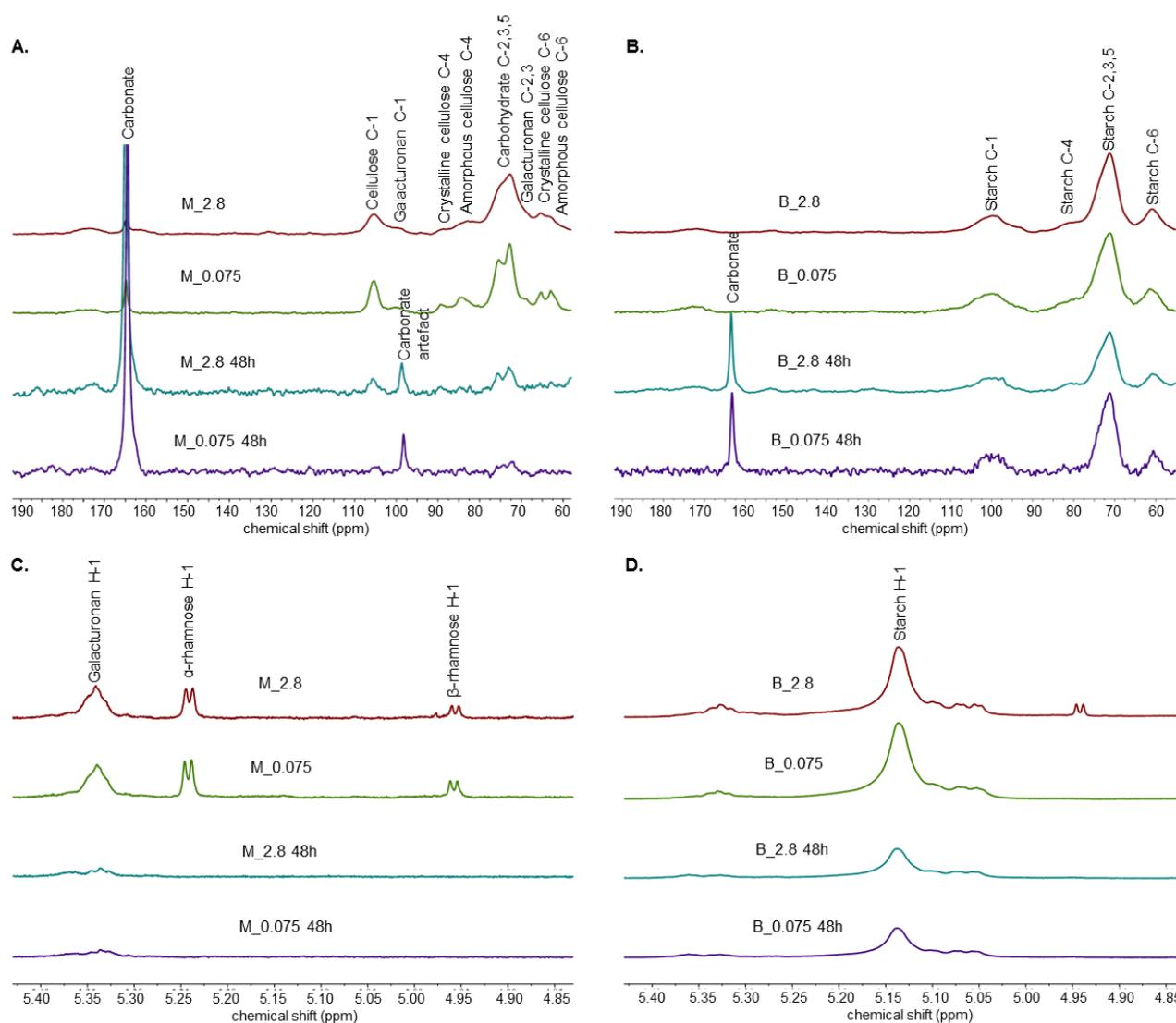
199 Banana particles after chewing and *in vitro* gastric and small intestinal digestion comprised
200 mostly starch both before and after fermentation (Fig 5B). Before fermentation, starch
201 granules were generally observed to be encapsulated by intact cell walls (Fig 4B). After
202 fermentation, cell walls surrounding the starch were no longer detected, but liberated
203 starch granules were still clearly visible (Fig 4Bii). This was supported by the solid state
204 ^{13}C NMR spectra of samples both before and after fermentation showing a characteristic
205 starch spectrum ³⁴ including a C-1 peak, which overlapped with lower intensity cellulose
206 signals (95-105 ppm). Evidence of partial fermentation of banana starch was observed
207 using confocal microscopy, where the apparently roughened or scratched surfaces of the
208 granule morphology is typical of the early stages of starch breakdown by amylolytic
209 enzymes, as previously observed for both raw ³⁵ and ripe bananas ³⁶ and potato starch ³⁷.



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Figure 4. Differential interference contrast images of A(i) mango (2.8 mm), B(i) banana (2.8 mm) and C(i) banana (0.075 mm) cellular structures in blue fluorescence before microbial fermentation, and (D) banana starch granules (10-30 μm in length) in green fluorescence before fermentation. Thick cellulose vascular fibres remained in A(ii) mango after fermentation, whereas fermented banana comprised mostly of B(ii) starch and C(ii) some vascular fibres. Image (E) shows the rough and/or scratched surfaces of released banana starch granules after 48 h fermentation (63x magnification).

216 The ^1H NMR spectra (Fig 5C, 5D) indicated that pectic galacturonan was present in both
 217 mango and banana but this was not obvious in the ^{13}C CP/MAS NMR spectra because of the
 218 overlap of the spinning side band (*ca* 99 ppm) from the carbonate peak (164 ppm) with the
 219 galacturonan C-1, which would appear from 98-101 ppm³³. Rhamnose was present in
 220 mango as the major monosaccharide, which has been similarly reported in *Ataulfo* and
 221 *Tommy Atkins* cultivars³⁸, but was not observed in banana. Rhamnose, along with
 222 galacturonan residues were apparently utilised by the faecal microbiota, as they were not
 223 observed for either fruit after fermentation. Bacterial species capable of degrading pectin
 224 and/or cellulose in porcine faecal microbiota have been reported in numerous studies³⁹⁻⁴¹.
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Figure 5. ^{13}C CP/MAS and ^1H NMR spectra of masticated (A, C) mango and (B, D) banana particles of 2.8 mm and 0.075 mm respectively, before and after microbial fermentation (48 h). In (A) mango samples after fermentation, the chemical shift at 99 ppm is an artefactual spinning side band from the intense carbonate residue (164 ppm). The peaks at 105, 101, 89, 85, 69, 65 and 63 ppm are identified as cellulose C-1, galacturonan C-1, crystalline

234 cellulose C-4, amorphous cellulose C-4, crystalline C-6 and amorphous cellulose C-6
235 respectively⁴². The peaks at 72-77 ppm correspond to the C-2, 3, 5 of carbohydrates. In
236 (B), the peaks at 105-95, 85-80 and 62 ppm are identified as C-1, C-4 and C-6 of starch,
237 with C-2, 3, 5 of starch at 77-67 ppm³⁴. In (C), the peaks before fermentation are assigned
238 to anomeric protons of pectic galacturonan (5.4 ppm)⁴², and monomeric rhamnose (5.24
239 and 4.96 ppm)⁴³. The broad peak from 5.05-5.25 ppm in (D) banana is from anomeric
240 protons of starch residues. M, mango; B, banana; 2.8 mm, 2.8 mm chewed fraction; 0.075,
241 0.075 mm chewed fraction. Results are consistent with monosaccharide analysis⁴⁴ after
242 hydrolysis in either 1M H₂SO₄ (hydrolyses all polysaccharides except crystalline cellulose)
243 and 12M H₂SO₄ (hydrolyses all polysaccharides), which showed that cellulose and 1M
244 H₂SO₄ solubilised glucan (e.g. starch) were the main components present after
245 fermentation of mango and banana respectively, with additional sugars characteristic of
246 pectin and other cell wall components present before but to a lesser extent after
247 fermentation.
248

249 3. Discussion

250 3.1. Effects of fruit and particle size on fermentation kinetics

251 Gas kinetics profiles showed significant differences between chewed mango and chewed
252 banana both in terms of kinetics and end-points. Disintegration of the plant cell wall
253 network and cell structures during *in vivo* mastication led to particles of varying sizes. The
254 largest chewed fraction (2.8 mm) consisted of more fermentation-resistant cellulosic
255 vascular tissues, whereas the 1 mm and 0.075 mm fractions contained mostly single cells
256 and ruptured cell fragments, and little or no vascular fibres²⁶. There was no significant
257 particle size effect ($P=0.43$) on the cumulative gas production, however, there was a trend
258 where the smaller mango particles of 0.075 mm were fermented more rapidly and
259 extensively, and produced more gas (485 mL/g DM) than the larger (>1 mm) and
260 unfractionated particles (411-445 mL/g DM). While decreasing particle size confers an
261 expansion of surface area available for microbial accessibility and/or attachment^{45, 46}, the
262 relative amount of vascular fibres is also a potential factor influencing this particle size
263 effect. Fig 2 shows that the significant difference in surface area due to particle size was
264 associated with kinetic rate (*i.e.* active fermentation) rather than lag (*i.e.* colonisation).
265

266 In contrast, the larger banana cell-cluster particles (2.8 mm) produced more gas (136 mL/g
267 DM) than the smallest particles (0.075 mm) (93 mL/g DM). Larger banana particles may
268 have contained a higher proportion of more fermentable cell wall structures. Similarly, in a
269 previous study, multi-cellular carrot particles (137-298 μ m) were fermented faster (23
270 mL/h) compared to 50-75 μ m single carrot cells and fragments (8 mL/h)⁶. It appears that
271 the plant cellular composition or architecture has a more significant impact than particle
272 size or available surface area (exposed to the faecal microbiota). Fruit and vegetable
273 matrices of varying physical and structural characteristics *i.e.* taproot or fruit, appeared to

274 have a strong influence on substrate fermentability, as did the cell contents. In this study,
275 differences in substrate fermentability were due to a fruit effect, rather than a particle size
276 effect, presumably due to the soft tissue structure of mango and banana.

277

278 **3.2. Degradation of cell walls is more extensive than that of resistant starch**

279 Chewed pieces of mango fruit were readily fermented as evidenced by the 48 h DMCV
280 and SCFA values, leaving mostly long cellulosic vascular fibres after fermentation, which
281 would be expected to be difficult to degrade by intestinal microbiota⁴⁷⁻⁴⁹ depending on the
282 chemical structure, microbial species and residence time in the gut⁵⁰. The strands of
283 cellulose in mango appeared not strongly connected to the parenchyma (fleshy) tissue,
284 and were sometimes observed as separate strands before microbial fermentation, but
285 clearly separated after fermentation. There appears to be a hierarchy in substrate
286 utilisation as evidenced by the preferential degradation of (thinner) primary parenchyma
287 cell walls over cellulosic vascular fibres.

288

289 Banana was far less efficiently fermented than mango, likely due to differences in their
290 polysaccharide compositions. Cellulose was the major polysaccharide component of
291 mango before and after fermentation (Figs 4A, 5A), while starch was the major component
292 of banana before and after fermentation (Figs 4B, 5B). Starch granules encapsulated
293 within intact banana cell walls survived mastication, as well as *in vitro* gastrointestinal
294 digestion. The thin banana cell walls present before fermentation were apparently all
295 fermented, releasing the starch granules that the cell walls had previously encapsulated.
296 Most starch granules appeared smooth after mastication and 'digestion', but exhibited a
297 parallel-striated surface after fermentation (Fig 4E). The fact that numerous and relatively
298 intact starch granules were observed at the end of the fermentation shows that they were
299 not rapidly fermented as soon as their encapsulating cell wall had been degraded. Indeed,
300 banana starch in the granular form is relatively resistant to digestion by pancreatic
301 enzymes⁵¹, similar to other B-type starches such as potato. When treated with pancreatic
302 amylases and amyloglucosidases *in vitro*, potato starch granules showed the same type of
303 'scratching' or exo-corrosion³⁷ as was found in the present study for banana starch
304 granules after microbial fermentation. The smooth dense surface of (released) banana
305 starch granules could also partially account for the intrinsic resistance of such granules to
306 enzyme-catalysed hydrolysis by faecal microbiota. Additionally, banana starch has been
307 previously reported to be highly resistant to *in vivo* human small intestinal digestion⁵²⁻⁵⁴.
308 The thick external layer (several μm) of larger blockets⁵⁵, which surrounds the banana

309 granules, is composed of a hard and well organised material⁵¹, and has been proposed to
310 impede enzyme action and thus reduce hydrolysis rate. This same external layer can
311 survive digestion of the granule interior giving rise to starch 'ghosts'⁵⁶, and it is possible
312 that some residual starch is also in this form. Colonic bacteria reportedly utilise a Starch
313 Utilisation System to reach these starch structures to extract glucose for energy⁵⁷, but the
314 evidence from this study suggests colonic microbiota may not be any more effective than
315 pancreatic amylases in overcoming the hard surface layer of banana starch granules.

316

317 Striations on the starch granule surface indicate the presence of microbial amylolytic
318 activity in the fermentation medium, leading to erosion, but with limited hydrolytic effect.
319 Some areas of the starch granule are more likely to be difficult to hydrolyse than other
320 areas (crystalline regions appearing after partial hydrolysis)³⁵ and banana starch was
321 described as having B-type crystallinity, which is typically associated with slow amylase
322 digestion⁵⁸. Additionally, during weighing of the masticated fractions into the fermentation
323 bottles, it was noted that banana fractions had a more physically compact structure, which
324 made it relatively difficult to obtain seemingly homogenous sub-samples. Accessibility, as
325 influenced by the entrapping matrix of banana cells and/or cell clusters appeared to restrict
326 access of the microbiota and/or their enzymes into the substrates.

327

328 **3.3. Higher SCFA and NH₃ in mango but higher BCR in banana**

329 Differences in SCFA and NH₃ between fruits were more pronounced than differences in
330 particle sizes, where 68% and 64% significantly greater concentrations of SCFA and NH₃
331 respectively, were observed for mango as compared with banana ($P<0.0001$). This
332 correlates well with the 74% greater DMCV₄₈ value for mango and is consistent with the
333 expectation that more rapid and extensive fermentation is generally associated with higher
334 SCFA production *in vitro*⁵⁹. Particle size played a small role in SCFA production and had
335 no significant effect on NH₃ production ($P>0.05$). Another fermentation study of wheat bran
336 also found that finer wheat particles (50 μm) produced higher SCFA concentrations than
337 larger clusters (758 μm)⁶⁰.

338

339 A lower total SCFA production typically corresponds to a proportionate increase in NH₃
340 levels. SCFA production in banana showed a lower total production, but there was no
341 concomitant increase in NH₃, which was expected. Banana fermentation was associated
342 with a higher BCR ($P<0.0001$), further validating the differences between these two fruits.
343 Branched-chain SCFA are usually formed as a result of bacteria metabolising undigested

344 and endogenous proteins, peptides and amino acids, particularly when carbohydrates are
345 in short supply as an energy source¹³ or are difficult to utilise as found in this study. Here,
346 NH₃ production was reduced for banana, which reflected the difference in availability of
347 fermentable carbohydrate^{39, 61}, thus increasing the BCR.

348

349 The composition of dietary polysaccharides available for fermentation also influenced the
350 proportions of SCFA produced. Resistant starch in banana appeared to favour an
351 increased proportion of butyric acid (Fig 4), agreeing with previous reports that *in vitro*
352 colonic fermentation of resistant starch is associated with elevated butyrate levels⁶²⁻⁶⁴.
353 Conversely, the higher acetic acid % observed in mango can be ascribed to the presence
354 of higher levels of cellulose and pectin, which is consistent with reported studies^{65, 66}
355 showing that acetate production predominates over propionate and butyrate in diets
356 containing higher levels of non-starch polysaccharides.

357

358 4. Experimental

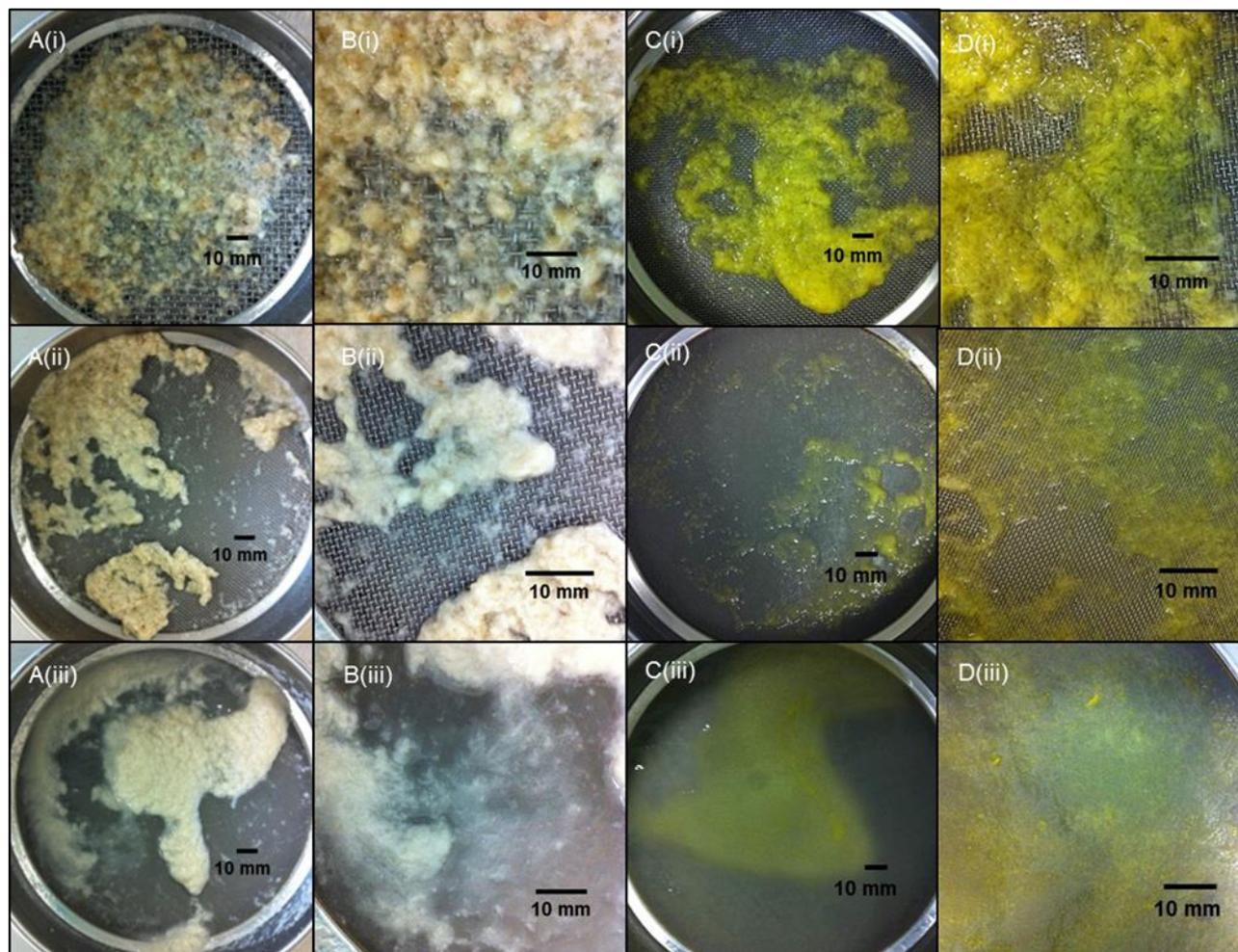
359 4.1. Preparation of fruit substrates

360 Fully ripe *Kensington Pride* mangoes and *Cavendish* bananas were procured from local
361 stores in Brisbane, Queensland, Australia, two to three days before each of the three
362 chewing sessions. The fruits were selected based on typical eating maturity, at stage 6 of
363 mango ripeness (peel is yellow with a pink-red blush and flesh is slightly firm)⁶⁷ and stage
364 6 of banana ripeness (peel is completely yellow)⁶⁸. Mangoes were stored at 4-6°C while
365 bananas were stored at ambient temperature prior to the chewing sessions. The chewing
366 selection and process, and bolus collection have been previously described²⁶. From the
367 twenty participants recruited, their expectorated boluses were collected and size
368 fractionated, with each sieve fractions being weighed to obtain a % distribution of particle
369 sizes ranging from >5.6 mm to 0.075 mm. This generated an individual mastication profile,
370 allowing the participants to be categorised into various type of chewers. This profile was
371 then used for the selection of a coarse chewer (with higher proportion of larger particle
372 size fractions), in addition to another criterion: their consistency in producing a similar
373 particle size distribution in each chewing experiment. In this study, a participant
374 representing a coarse chewer was selected from the twenty participants and re-invited for
375 three subsequent chewing sessions for bolus collection.

376

377 The fruits were first subjected to *in vivo* human mastication and wet-sieve fractionation (Fig
378 6)²⁶, followed by *in vitro* gastrointestinal digestion⁶ and centrifugation at 3000 g for 10 min

379 (Avant®JE centrifuge, JA14 rotor). The pellets were then washed three times with water
 380 (1:3) to remove salivary components such as enzymes, soluble sugars and amino acids.
 381 Samples were stored at 4°C prior to *in vitro* microbial fermentation. The chewing
 382 experiment was approved by the Medical Research Ethics Committee at The University of
 383 Queensland (Ethical clearance no. 2012000683) and was performed in compliance with
 384 relevant laws and institutional guidelines. Participants gave informed consent to the
 385 experiment.
 386



387
 388 Figure 6. Images of masticated and fractionated (A & B) banana and (C & D) mango bolus
 389 particles captured on sieves of size (i) 2.8 mm, (ii) 1 mm and (iii) 0.075 mm. B(i-iii) and D(i-
 390 iii) show magnified views of each fraction respectively.
 391

392 4.2. Preparation of faecal inoculum

393 Faecal inoculum was prepared as described previously⁴. Faeces were collected directly
 394 from five pigs (~35 kg) under ethics approval of the University of Queensland Animal
 395 Ethics Committee (SAFS/111/13/ARC). Prior to faecal collection, the pigs were fed a semi-
 396 purified diet based on highly digestible maize starch and fishmeal for ten days to avoid
 397 adaption of the gut microbiota to any of the substrates being used⁶⁹. Faeces were kept in

398 pre-warmed CO₂-filled vacuum flasks during transport to the laboratory. To avoid (as far as
399 possible) any effect of genetic variation of the pigs, the faeces from all five pigs were
400 combined to make an inoculum representative of pigs as a whole. The faeces were then
401 mixed (1:5) with pre-warmed saline (9 g/L NaCl), homogenised for 1 min under CO₂ and
402 strained through four layers of muslin cloth within 2 h of faecal collection.

403

404 **4.3. Cumulative gas production**

405 The cumulative gas production technique was carried out as previously described ⁴. Fresh
406 unfractionated and fractionated mango (4.8±0.4 g) and banana (3.1±1.4 g) particles (each
407 particle size, n=4), were weighed into 120 mL serum bottles containing 76 mL basal
408 solution, 1 mL vitamin/phosphate buffer solution, 4 mL bicarbonate buffer and 1 mL
409 reducing agent ⁷⁰. Unfractionated mango and banana refers to expectorated mango and
410 banana boluses that were not sieved, containing mixed size particles. Faecal inoculum (4
411 mL) was added to each serum bottle and incubated at 39°C. Experimental blanks
412 containing only inoculum and medium were also included. A steady stream of O₂-free CO₂
413 flowed into the fermentation bottles at all times prior to sealing with butyl rubber stoppers
414 and aluminium crimp seals. The medium contained resazurin, known as an oxygen-
415 reduction indicator, which did not turn pink, confirming there was no oxygen contamination
416 in the bottles. Cumulative gas readings were measured using a pressure transducer (Type
417 453A, Bailey and Mackey Ltd., Birmingham, UK) and a LED digital readout voltmeter
418 (Tracker 200) after insertion of a hypodermic syringe needle through the butyl rubber
419 stopper above the fermentation solution. The head-space pressure and volume of gas
420 were measured in each fermentation bottle (178 bottles) at 0, 2, 4, 6, 8, 10, 12, 15, 18, 21,
421 24, 27, 30, 33, 36, 39, 43, 46 and 48 h of the fermentation period according to the method
422 of Williams et al. (2005) ⁴. Then, the pressure and volume of gas recorded for each bottle
423 was regressed to provide a corrected volume at each time per bottle. After cumulative gas
424 readings were carried out for the bottles at their respective time intervals, they were placed
425 immediately in iced-water to inhibit further microbial activity prior to sampling for post-
426 fermentation analyses.

427

428 **4.4. Post-fermentation analyses**

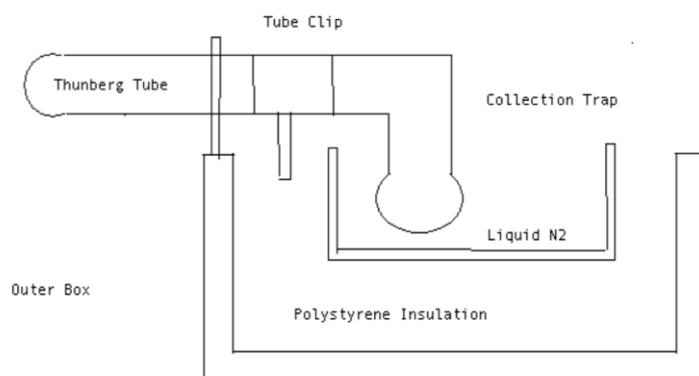
429 pH of the fermentation solutions was recorded and aliquots were taken from all the
430 fermentation time bottles at their respective time intervals for SCFA and NH₃ analyses.
431 The remaining bottle contents were centrifuged at 4000 g for 10 min at 4°C and washed
432 twice with water. Dry matter (DM) of the fermented samples (and substrates before

433 fermentation) was determined by drying to a constant weight at 103°C (ISO 6496, 1999)
434 and then ashing by combustion at 550°C (ISO 5984, 1978).

435

436 SCFA in the fermented samples was extracted ⁷² with modifications to the microvacuum
437 distillation apparatus (Fig 7), which has been expanded to distill 12 samples at one time.
438 Sample aliquots (0.9 mL) and 1 M sulphuric acid (0.1 mL) containing 500 mM formic acid
439 were added to the Thunberg tubes, frozen with liquid N₂ and vacuumed distilled. SCFA
440 concentrations of the extracted aliquots were then analysed by gas chromatography using
441 an Agilent GC-FID (HP6890) (Agilent Technologies, Mulgrave, VIC, Australia) and DB-
442 FFAP capillary column (30 m x 0.5 mm) at a flow rate of 5.3 mL/min with helium as the
443 carrier gas. Injector and detector temperatures were 250°C and oven temperature was
444 programmed from 90°C (1 min) to 190°C (1 min) at 10°C/min). The injection volume was
445 0.5 µL. *Iso*-caproic acid was used as an internal standard. The SCFA mixed reference
446 comprised of acetic acid (52.51 mM), propionic acid (13.4 mM), iso-butyric acid (1.07 mM),
447 n-butyric acid (5.45 mM), iso-valeric acid (0.91 mM), n-valeric acid (0.92 mM), n-caproic
448 acid (0.16 mM) and heptanoic acid (0.15 mM) (Sigma-Aldrich, Castle Hill, NSW, Australia).
449 The BCR was calculated as the ratio of mainly branched chain acids (isobutyric, isovaleric
450 and valeric acids- end-products of protein fermentation) concentration, to straight chain
451 acids (acetic, propionic and butyric acids) that had all been corrected to mg of AAE using
452 their respective molar masses.

453



454

455 Figure 7. Simplified schematic diagram of vacuum distillation apparatus. An insulated box
456 is constructed with a water-tight tray such that the Thunberg tube is held at a slight angle
457 horizontally above the box by clips, which prevents liquid from running into the collection
458 trap if the sample should melt. The bulb of the collection trap is in the open tray, which is
459 filled with liquid N₂ so that the bulb is below the surface.

460

461 Analysis of NH₃ of the fermented samples involved a modified procedure ⁷³. Here, sample
462 aliquots were mixed with 0.2 N HCl (1:1) with the concentrations of ammonium and

463 nitrogen being determined using the reduction of ammonium ions by sodium salicylate and
464 nitroprusside in a weakly alkaline buffer (free chlorine). The resulting coloured complex
465 was measured using a UV-Vis spectrophotometer (OlympusAU400, Tokyo, Japan) at 650
466 nm.

467

468 **4.5. Nuclear magnetic resonance (NMR) spectroscopy**

469 **Solid-state ^{13}C CP/MAS NMR**

470 Mango and banana particles, before and after fermentation (0 h and 48 h respectively),
471 were freeze-dried and analysed by solid-state ^{13}C CP/MAS NMR spectroscopy using a
472 Bruker MSL-300 spectrometer (Bruker, Karlsruhe, Germany) at a frequency of 75.46 MHz.
473 Samples were lightly ground and stirred to ensure homogeneity, from which 200 mg was
474 packed into a 4 mm diameter, cylindrical, PSZ (partially stabilized zirconium oxide) rotor
475 with a Kelf end cap. The rotor was spun at 5-6 kHz at the magic angle (54.7°). The 90°
476 pulse width used was 5 μs , while a contact time of 1 ms and a recycle delay of 3 s was
477 used for all samples. The spectral width was 38 kHz, acquisition time 50 ms, time domain
478 points 2 k, transform size 4 k, and line broadening 50 Hz. At least 2400 scans were
479 accumulated for each spectrum.

480

481 **Solution state ^1H NMR**

482 Similarly, freeze-dried mango and banana samples, before and after fermentation (0 h and
483 48 h respectively) (5 mg) were dissolved at 80°C overnight in 650 μL of d_6 -DMSO
484 containing 0.5 wt % LiBr. After the samples were cooled to room temperature, sodium 3-
485 (trimethylsilyl)propionate-2,2,3,3- d_4 (TSP) in D_2O was added as an internal standard. The
486 addition of 50 μL of deuterated trifluoroacetic acid (d_1 -TFA) directly before each
487 measurement moved the HOD peak away from the diagnostic carbohydrate anomeric
488 signals⁷⁴. ^1H NMR spectra were measured on a Bruker Avance 500MHz spectrometer
489 operating at 298K equipped with a 5 mm PABBO probe using 12 μs 90° pulse, 3.91 s
490 acquisition time, 1 s relaxation delay and 64 scans.

491

492 **4.6. Confocal laser scanning microscopy (CLSM)**

493 Microscopy of mango and banana particles, before and after fermentation (0 h and 48 h
494 respectively) was carried out using CLSM (LSM700, Carl Zeiss, Germany) under 10x and
495 40x objective lenses, differential interference contrast (DIC) and Zen (Black) 2011
496 software. Fluorescence of cell walls was observed at an excitation λ of 355 nm, emission λ
497 from 400-440 nm, and laser power intensity of 2% after staining with Calcofluor. Starch

498 granules were stained with 3-aminopropyl-trimethoxysilane (APTS) followed by washing
 499 with 70% ethanol, incubating in APTS solution (10 mM APTS in 15% acetic acid) at 30°C
 500 overnight, washing five times with Milli-Q water and finally centrifuging at 3000 g for 10
 501 min. Fluorescence of starch granules was observed at an excitation λ of 488 nm.

502

503 4.7. Curve fitting and statistical analysis

504 Cumulative gas production measured as a function of time was corrected to the volume
 505 (mL) of gas produced per g of substrate DM (DMCV₄₈) and was fitted to the monophasic
 506 Michaelis-Menten model²⁷ shown in Eq. (1):

$$507 \text{ DMCV}_{48} = A/(1 + (C/t)^B) \quad (1)$$

508

509 where A is the asymptotic gas production (mL), B is the switching characteristic of the
 510 curve, C is the time at which half of the asymptotic value is reached ($T_{1/2}$) and t is the
 511 fermentation time (h). The maximal rate of gas production, R_{\max} (mL/h) and the time at
 512 which it occurs, TR_{\max} (h) were calculated from Eq. (2) and (3):

$$513 R_{\max} = (A(CB)B(TR_{\max}^{-(B-1)}))/(1+(C^B)TR_{\max}^{(-B)})^2 \quad (2)$$

514

$$515 TR_{\max} = C(((B-1)/(B+1))^{1/B}) \quad (3)$$

516

517 All parameters were tested for significant differences (effects of fruit, particle size and the
 518 interaction between fruit and particle size) using the Tukey-Kramer multiple comparison
 519 procedure as defined in Eq. (4):

$$520 Y = \mu + F_i + P_i + (F_i \times P_i) + \varepsilon_i \quad (4)$$

521

522 where Y is the dependent variable, μ is the mean, F_i is the effect of fruit, P_i is the effect of
 523 particle size, $(F_i \times P_i)$ is the interaction between fruit and particle size, and ε_i is the error
 524 term. Statistical analyses were performed using SAS (9.3) NLIN (curve fitting) and GLM
 525 (significant difference) procedures⁷⁵.

526

527 5. Conclusions

528 The investigation of the fermentation characteristics of masticated particles of mango and
 529 banana has demonstrated distinct differences between the two fruits in terms of cellular
 530 architecture and starch content, which seemed to outweigh any effects of particle size on
 531 colonic-microbial fermentability. A decrease in particle size and a concomitant increase in
 532 available surface area would have been expected to increase the total gas production by

533 enhancing microbial accessibility. However, colonic fermentation differences between
534 larger particle clusters (2.8-1 mm) and single cells or cell fragments (0.075 mm) were not
535 as significant in the soft tissues of mango and banana studied here, as compared to a
536 previous study on carrots with a more robust cellular structure ⁶. The fruit (parenchyma)
537 fleshy cells were fully or mostly fermented during fermentation, preferentially over resistant
538 starch in banana, and over the thick cellulosic vascular fibres in mango. The slow
539 fermentability of banana starch conferred by its intrinsic resistance and cell-wall
540 encapsulation may have implications on calorific availability, satiety, glucose metabolism,
541 and transit rates along the colon, and therefore deserves further study. The higher
542 absolute levels of butyrate production from mango could be important in terms of
543 contributing to anti-inflammatory and anti-carcinogenic properties ⁷⁶⁻⁷⁸ rather than the
544 higher % ratio of butyrate to acetate/propionate in banana. Further studies investigating
545 the extended fermentation of both fruits over 72 h and longer, and the microscopic
546 degradation of banana cell walls with time, preferably 3-hourly should also be explored in
547 future work.

548

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References

1. C. A. Edwards, G. Gibson, M. Champ, B. B. Jensen, J. C. Mathers, F. Nagengast, C. Rumney and A. Quehl, *J. Sci. Food Agric.*, 1996, 71, 209-217.
2. E. Bauer, B. A. Williams, C. Voigt, R. Mosenthin and M. W. A. Verstegen, *Arch. Anim. Nutr.*, 2010, 64, 394-411.
3. E. Bauer, B. A. Williams, C. Voigt, R. Mosenthin and M. W. A. Verstegen, *J. Sci. Food Agric.*, 2003, 83, 207-214.
4. B. A. Williams, M. W. Bosch, H. Boer, M. W. A. Verstegen and S. Tamminga, *Anim. Feed Sci. Technol.*, 2005, 123, 445-462.
5. S. Bazzocco, I. Mattila, S. Guyot, C. Renard and A. M. Aura, *Eur. J. Nutr.*, 2008, 47, 442-452.
6. L. Day, J. Gomez, S. K. Oiseth, M. J. Gidley and B. A. Williams, *J. Agric. Food Chem.*, 2012, 60, 3282-3290.
7. A. L. Molan, M. A. Lila, J. Mawson and S. De, *World J. Microbiol. Biotechnol.*, 2009, 25, 1243-1249.
8. M. H. Vong and M. L. Stewart, *Benef. Microbes*, 2013, 4, 291-295.
9. O. Piquer, C. Casado, S. Biglia, C. Fernandez, E. Blas and J. J. Pascual, *J. Anim. Feed Sci.*, 2009, 18, 743-757.
10. Z. S. Davies, D. Mason, A. E. Brooks, G. W. Griffith, R. J. Merry and M. K. Theodorou, *Anim. Feed Sci. Technol.*, 2000, 83, 205-221.
11. J. C. J. Groot, B. A. Williams, A. J. Oostdam, H. Boer and S. Tamminga, *British Journal of Nutrition*, 1998, 79, 519-525.
12. Y. T. Chiu and M. Stewart, *Journal of Medicinal Food*, 2012, 15, 120-125.
13. W. H. Hendriks, J. van Baal and G. Bosch, *Br. J. Nutr.*, 2012, 108, S247-S257.
14. R. Mosenthin, *Asian-Australasian Journal of Animal Science* 1998, 11, 608-619.
15. Y. S. Kim, J. K. Brecht and S. T. Talcott, *Food Chem.*, 2007, 105, 1327-1334.
16. M. W. Davey, J. Keulemans and R. Swennen, *J. Chromatogr. A*, 2006, 1136, 176-184.
17. H. Kim, J. Y. Moon, D. S. Lee, M. Cho, H. K. Choi, Y. S. Kim, A. Mosaddik and S. K. Cho, *Food Chem.*, 2010, 121, 429-436.
18. A. Z. Mercadante, D. B. Rodriguez-Amaya and G. Britton, *J. Agric. Food Chem.*, 1997, 45, 120-123.
19. C. M. Ajila, L. J. Rao and U. Rao, *Food Chem. Toxicol.*, 2010, 48, 3406-3411.
20. J. A. Manthey and P. Perkins-Veazie, *J. Agric. Food Chem.*, 2009, 57, 10825-10830.
21. J. D. Ornelas-Paz, E. M. Yahia and A. A. Gardea, *J. Agric. Food Chem.*, 2007, 55, 6628-6635.
22. R. M. Robles-Sanchez, H. Astiazaran-Garcia, O. Martin-Belloso, S. Gorinstein, E. Alvarez-Parrilla, L. A. d. I. Rosa, G. Yepiz-Plascencia and G. A. Gonzalez-Aguilar, *Food Research International*, 2011, 44, 1386-1391.
23. N. M. Shofian, A. A. Hamid, A. Osman, N. Saari, F. Anwar, M. S. P. Dek and R. Hairuddin, *Int. J. Mol. Sci.*, 2011, 12, 4678-4692.
24. J. Bouayed, L. Hoffmann and T. Bohn, *Food Chemistry: 128 (1) 14-21*, 2011, 128, 14-21.
25. M. Alothman, R. Bhat and A. A. Karim, *Food Chem.*, 2009, 115, 785-788.
26. D. Y. Low, B. D'Arcy and M. J. Gidley, *Food Research International*, 2015, 67, 238-246.
27. J. C. J. Groot, J. W. Cone, B. A. Williams, F. M. A. Debersaques and E. A. Lantinga, *Anim. Feed Sci. Technol.*, 1996, 64, 77-89.
28. P. J. Van Soest, *Nutritional ecology of the ruminant*, Cornell University Press, United States of America, 2nd edn., 1994.

29. H. J. Flint, K. P. Scott, S. H. Duncan, P. Louis and E. Forano, *Gut Microbes*, 2014, 3, 289-306.
30. J. M. Campbell, G. C. Fahey and B. W. Wolf, *J. Nutr.*, 1997, 127, 130-136.
31. S. Yanahira, M. Morita, S. Aoe, T. Suguri, Y. Takada, S. Miura and I. Nakajima, *J. Nutr. Sci. Vitaminol.*, 1997, 43, 123-132.
32. J. K. T. Ng, Z. D. Zujovic, B. G. Smith, J. W. Johnston, R. Schroder and L. D. Melton, *Carbohydr. Res.*, 2014, 386, 1-6.
33. M. Dick-Perez, Y. A. Zhang, J. Hayes, A. Salazar, O. A. Zabolina and M. Hong, *Biochemistry*, 2011, 50, 989-1000.
34. I. Tan, B. M. Flanagan, P. J. Halley, A. K. Whittaker and M. J. Gidley, *Biomacromolecules*, 2007, 8, 885-891.
35. P. Y. Zhang, R. L. Whistler, J. N. BeMiller and B. R. Hamaker, *Carbohydr. Polym.*, 2005, 59, 443-458.
36. K. Kayisu, L. F. Hood and P. J. Van soest, *J. Food Sci.*, 1981, 46, 1885-1890.
37. S. Dhital, A. K. Shrestha and M. J. Gidley, *Carbohydr. Polym.*, 2010, 82, 480-488.
38. M. D. Garcia-Magana, H. S. Garcia, L. A. Bello-Perez, S. G. Sayago-Ayerdi and M. M. M. de Oca, *Plant Food Hum. Nutr.*, 2013, 68, 254-258.
39. M. A. Sappok, W. F. Pellikaan, M. W. Versteegen, G. Bosch, A. Sundrum and W. H. Hendriks, *Journal of Science and Food Agriculture*, 2012, 93, 987-994.
40. B. A. Williams, M. W. Bosch, A. Awati, S. R. Konstantinov, H. Smidt, A. D. L. Akkermans, M. W. A. Versteegen and S. Tamminga, *Anim. Res.*, 2005, 54, 191-201.
41. F. Rink, E. Bauer, M. Eklund and R. Mosenthin, *Arch. Anim. Nutr.*, 2011, 65, 445-459.
42. C. M. G. C. Renard and M. C. Jarvis, *Plant Physiology*, 1999, 119, 1315-1322.
43. A. M. A. de Bruyn, R. de Gussem and G. G. S. Dutton, *Carbohydr. Res.*, 1976, 47, 158-163.
44. P. Comino, K. Shelat, H. Collins, J. Lahnstein and M. J. Gidley, *J. Agric. Food Chem.*, 2013, 61, 12111-12122.
45. J. Parada and J. M. Aguilera, *J. Food Sci.*, 2007, 72, 21-32.
46. F. Guillon, A. Auffret, J. A. Robertson, J. F. Thibault and J. L. Barry, *Carbohydr. Polym.*, 1998, 37, 185-197.
47. H. Yu, R. G. Liu, D. W. Shen, Y. Jiang and Y. Huang, *Polymer*, 2005, 46, 5689-5694.
48. F. Ismailbeigi, J. G. Reinhold, B. Faraji and P. Abadi, *J. Nutr.*, 1977, 107, 510-518.
49. S. Otles and S. Ozgoz, *Acta scientiarum polonorum. Technologia alimentaria*, 2014, 13, 191-202.
50. J. W. Anderson and W. J. L. Chen, *Am. J. Clin. Nutr.*, 1979, 32, 346-363.
51. C. A. Soares, F. H. G. Peroni-Okita, M. B. Cardoso, R. Shitakubo, F. M. Lajolo and B. R. Cordenunsi, *J. Agric. Food Chem.*, 2011, 59, 6672-6681.
52. N. Faisant, A. Buleon, P. Colonna, C. Molis, S. Lartigue, J. P. Galmiche and M. Champ, *Br. J. Nutr.*, 1995, 73, 111-123.
53. J. H. Cummings, E. R. Beatty, S. M. Kingman, S. A. Bingham and H. N. Englyst, *Br. J. Nutr.*, 1996, 75, 733-747.
54. H. N. Englyst and J. H. Cummings, *Am. J. Clin. Nutr.*, 1986, 44, 42-50.
55. N. Faisant, D. J. Gallant, B. Bouchet and M. Champ, *Eur. J. Clin. Nutr.*, 1995, 49, 98-104.
56. B. Zhang, S. Dhital, B. M. Flanagan and M. J. Gidley, *J. Agric. Food Chem.*, 2014, 62, 760-771.
57. E. C. Martens, N. M. Koropatkin, T. J. Smith and J. I. Gordon, *J. Biol. Chem.*, 2009, 284, 24673-24677.

58. B. R. Hamaker and Y. C. Tuncil, *Journal of Molecular Biology*, 2014, 426, 3838-3850.
59. D. L. Topping and P. M. Clifton, *Physiol. Rev.*, 2001, 81, 1031-1064.
60. D. J. A. Jenkins, C. W. C. Kendall, V. Vuksan, L. S. A. Augustin, Y. M. Li, B. Lee, C. C. Mehling, T. Parker, D. Faulkner, H. Seyler, E. Vidgen and V. Fulgoni, *J. Am. Coll. Nutr.*, 1999, 18, 339-345.
61. J. W. Cone, A. H. van Gelder and F. Driehuis, *Anim. Feed Sci. Technol.*, 1997, 66, 31-45.
62. D. J. Rose, A. Keshavarzian, J. A. Patterson, M. Venkatachalam, P. Gillevet and B. R. Hamaker, *Mol. Nutr. Food Res.*, 2009, 53, S121-S130.
63. J. L. J. Casterline, C. J. Oles and Y. Ku, *J. Agric. Food Chem.*, 1997, 45, 2463-2467.
64. U. Lehmann, G. Jacobasch and D. Schmiedl, *J. Agric. Food Chem.*, 2002, 50, 5236-5240.
65. G. Anison and D. L. Topping, *Annual Nutrition of review*, 1994, 14, 297-320.
66. A. G. Low, in *Recent developments in pig nutrition 2*, eds. D. J. A. Cole, W. Haresign and P. C. Garnsworthy, Nottingham University Press, Loughborough, Leicestershire, UK, 1993, ch. 11, pp. 137-162.
67. Primary Industries & Fisheries, Queensland Government, Queensland, Australia, 2012.
68. Isopan, Banana ripening, <http://isopaninsulation.com/technologies/banana-ripening>, Accessed August 3, 2011.
69. D. J. Wang, B. A. Williams, M. G. Ferruzzi and B. R. D'Arcy, *J. Sci. Food Agric.*, 2013, 93, 276-283.
70. S. E. Lowe, M. K. Theodorou, A. P. J. Trinci and R. B. Hespell, *J. Gen. Microbiol.*, 1985, 131, 2225-2229.
71. M. K. Theodorou, B. A. Williams, M. S. Dhanoa, A. B. McAllan and J. France, *Anim. Feed Sci. Technol.*, 1994, 48, 185-197.
72. H. J. Vreman, J. A. Dowling, R. A. Raubach and M. W. Weiner, *Anal. Chem.*, 1978, 50, 1138-1141.
73. W. E. Baethgen and M. M. Alley, *Commun. Soil Sci. Plant Anal.*, 1989, 20, 961-969.
74. M. J. Tizzotti, M. C. Sweedman, D. Tang, C. Schaefer and R. G. Gilbert, *J. Agric. Food Chem.*, 2011, 59, 6913-6919.
75. SAS, SAS Institute Inc., Cary, NC, USA, 9.3 edn., 2011.
76. M. A. R. Vinolo, H. G. Rodrigues, R. T. Nachbar and R. Curi, *Nutrients*, 2011, 3, 858-876.
77. S. Tedelind, F. Westberg, M. Kjerrulf and A. Vidal, *World Journal of Gastroenterology*, 2007, 13, 2826-2832.
78. E. A. Williams, J. M. Coxhead and J. C. Mathers, *Proc. Nutr. Soc.*, 2003, 62, 107-115.