# Environmental Science Processes & Impacts

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](http://www.rsc.org/Publishing/Journals/guidelines/AuthorGuidelines/JournalPolicy/accepted_manuscripts.asp).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](http://www.rsc.org/help/termsconditions.asp) and the Ethical quidelines still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



rsc.li/process-impacts



Environmental Science: Processes

& Impacts Accepted Manuscript





- 
- 

# 

- 
- 
- 
- 

# **1. Introduction**

The use of nanoparticles (NPs) in commercial products and industrial applications has increased greatly in recent years since they have eminent magnetic, electrical and optical 86 properties with low cost and low energy needs. It is believed that products produced with NPs will have a significant role in future and improve the quality of life, so the amount of variety

Environmental Science: Processes

& Impacts Accepted Manuscript

and importance of these products are increasing continuously. As a natural consequence of this increase, NPs are released to the environment without any control. Their release can happen during the production, usage or disposal processes of the NP included products. Especially NPs usage in the hygiene and cosmetic industries increase their contact with water and cause an important concern for the wastewater treatment plants (WWTPs) with a biological treatment process. The contact of shower, laundry, cleaning and rain waters with the products containing NPs cause NPs to leave the product and follow the pathways of water to the WWTP. In the "*Nanotechnology Products Risk Assessment*" report [1] of European Union Scientific Committee, it is stated that NPs tend to aggregate and adsorb to organic materials and, therefore, accumulate in the solid part of wastewaters and reach to high concentrations. Similarly, there are studies presenting that NPs are removed from wastewaters 99 to a large extend and accumulate in sludge  $[2 - 6]$ . This accumulation causes NPs to emerge in waste activated sludge (WAS) processing operations.

WAS management is a global environmental problem. Besides, WAS is also a quite destructive pollution source itself since most of the pollutants are sorbed onto suspended solids and as a result they are found in sludge through sedimentation occurring in primary and secondary clarifiers. It is stated that 53% of the sludge in EU-27 is used in agriculture directly or after composting and more than 40% of produced biosolids are applied to land in USA and Canada [7]. It is necessary to stabilize the sludge to prevent the application of pollutants accumulated in WAS to the soil. Anaerobic digestion (AD) is one of the most technically mature and cost-effective process to convert sludge into methane-rich biogas and, therefore, stabilize the sludge [8]. During AD, several biochemical pollutants such as surfactants and medicine residues accumulated in sludge can be biodegraded and, therefore, not become a threat to the application of sludge to the soil after AD [7]. Furthermore, many materials can

have inhibitory effects on AD and inhibition value varies according to the material itself and process conditions.

The potential risk of releasing NPs into the natural environment threatens human health and ecosystems [9]. However, compared to NP synthesis and applications, limited study has been focused on environmental and health impacts [10] suggesting that the fate and transport of NPs should be investigated thoroughly to minimize their toxic effects [11, 12]. Higher organisms as targets were used in most of the work performed on the toxicology of NPs for defining ecotoxicological effects [13]. Toxicity of NPs on aerobic microorganisms and aquatic organisms has been investigated in detail [14 – 25] however limited number of research [6, 10, 13, 26 – 33] addressed the impacts of NPs on AD. Therefore, it is important to determine the effects of NPs on microorganisms participate in WAS AD process [28].

Since available information and detailed scientific contributions of NPs effects on WAS AD are insufficient, any effort related to inhibition risk of NPs should help in the regulation of production and use of NPs in the industry. In this work, short and long term impacts of three different NPs on WAS AD were examined with acute and chronic impact tests. Inhibition 127 amounts of NPs on AD and  $EC_{50}$  values were calculated from acute and chronic impact test results. Moreover, efforts have been made to explain the inhibition effects of NPs on AD process by the help of modeling of AD. Finally, FISH analysis was applied to determine the effect of NPs on methanogenetic diversity involved in WAS AD.

- 
- - **2. Materials and Methods**
	- **2.1. WAS characterization analyses**

WAS is taken from the sludge centrifuge of a municipal WWTP with a design capacity of 136  $31250 \text{ m}^3/\text{day}$ . The anaerobic seed sludge was withdrawn from the WAS AD reactors of

Environmental Science: Processes

& Impacts Accepted Manuscript

137 another WWTP with a design capacity of 210000  $\text{m}^3/\text{day}$  in Antalya. The analyses of dry matter (TS), organic matter (VS) and chemical oxygen demand (COD) were performed according to Standard Methods [34]. The contents of lignin, cellulose, hemicellulose and soluble matter were determined by Van Soest [35] using the Fiberbag system (Gerhardt). Total Kjeldahl nitrogen (TKN) was determined by TKN analyzer (Buchi Digest Automat K-438 and Buchi Auto Kjeldahl Unit K-370). Soluble carbohydrate concentration as glucose and soluble reducing sugar were determined by Anthrone [36] (Dreywood, 1946) and Dinitrosalicylic acid (DNS) methods [37] (Miller, 1959), respectively. Protein concentration and extractive matter including lipids were quantified by Lowry method [38] (Lowry et al., 1951) and soxhlet extraction [39] (Bridoux et al., 1994), respectively. Elemental analyses were performed by the CHNS analyzer (LECO, CHNS-932).

**2.2. Preparing NP suspensions and determination of particle sizes** 

150 Ag, CuO and CeO<sub>2</sub> NPs were purchased from Alfa Aesar. The stock NP suspension of 2  $g/L$ was prepared for BMP analyses [ 40, 41]. The stock NP suspension of 30 g/L was prepared 152 for anaerobic inhibition test and then sonicated (25°C, 30 W, 20 kHz, Sonics&Materials Vibra Cell) for 1 hour in order to break the NP aggregates in this suspension. Sodium dodecylbenzene sulfonate (SDBS) was added to the NP suspensions with a final concentration of 0.1 mM to prevent the aggregation of NPs for biochemical methane potential (BMP) and anaerobic inhibition tests. The final NP concentrations in BMP reactors were set to 5, 50, 150, 250 and 500 mgNP/gTS and final concentrations of 5, 50, 150, 250, 500, 750 and 1000 158 mgNP/gTS for each NP in the anaerobic inhibition test reactors were prepared from the stock suspensions. Lower concentrations of NPs were chosen to be environmentally relevant and to be able to compare to literature, however higher concentrations of NPs were chosen with consideration of a worst-case scenario such as an industrial spill and also to fill the lack of

investigation on the impacts of high concentrations to AD in literature as suggested by Nyberg et al [26] as well.

Particle size analysis for each stock NP suspension was performed by dynamic light scattering technique using a Malvern Zetasizer Nano ZS (Malvern Instruments, USA). The instrument 166 has a 4 mW He-Ne laser with a wavelength of 633 nm, and a measurement angle of 173 $\degree$ .

It is a well-known phenomenon that when high concentrations of NPs are applied in solutions with high ionic strengths, NPs tend to agglomerate and to form larger sizes. Therefore, to control the concentration of NPs in the solutions for a better stability, the use of dispersants, pH adjustment or various mixing methods need to be applied [42]. In this study, a cationic surfactant SDBS was used and ultrasound treatment was applied to stabilize the suspensions, and the pH of the suspensions were adjusted to pH 7.5±0.5. Commercially determined particle sizes and average particle sizes after being used in stock suspensions are given in Table 1. Particle size results showed that SDBS addition and ultrasound treatment could not stop the agglomeration of NPs completely which might be due to the high ionic strength, high NP concentration, neutral pH of the stock suspensions and the consequential formation of aggregates [43-45]. The particle size results are consistent with the literature [28, 46].

**Environmental Science: Processes & Impacts Accepted Manuscript**

Environmental Science: Processes

& Impacts Accepted Manuscript

**2.3. Biochemical methane potential (BMP) test for long term inhibition** 

The long term impacts of NPs on AD of WAS are investigated by the measurement of methane production. The methane production was studied by batch BMP tests following the procedures established by Carrere et al. [40] to investigate the long term (chronic) inhibition impacts of different NP concentrations on WAS AD.

#### **2.4. Anaerobic short term inhibition test**

Environmental Science: Processes

& Impacts Accepted Manuscript

The test was performed according to the "*ISO 13641-1 Water Quality: Determination of inhibition of gas production of anaerobic bacteria*" standard [47] to observe the short term (acute) inhibition impacts of NPs on AD of WAS. A 400 mL of working volume containing WAS, anaerobic seed sludge, NP suspension, and distilled water was used in the 500 mL reactors. Total gas pressures in the reactors were measured for 72 hours in every 12 hours by Digitron 2085p handheld digital manometer. Inhibition values caused by different concentrations of each NP were calculated as a percentage (%) using the equation given in the standard.

 $EC_{10}$ ,  $EC_{20}$  and  $EC_{50}$  concentrations were calculated by using EPA-TRAPv1.22 (Environmental Protection Agency - Toxicity Relationship Analysis Program Version 1.22) program. Model type and data analysis type were used as threshold Sigmoid and nonlinear regression, respectively.

#### **2.5. Mathematical Modeling**

The Gompertz equation [48], Logistic [49] and Transference functions [50] were used for the 202 determination of the biogas production potential  $(P_M)$ , the maximum methane production rate 203 (R<sub>M</sub>) and duration of lag phase ( $\lambda$ ) for digestion of WAS containing Ag, CeO<sub>2</sub> and CuO NPs. Furthermore, first-order reaction kinetic model [51] was used to determine hydrolysis rate 205 constant ( $k_H$ ) and overall reaction rate constant ( $k_R$ ). The models used for evaluating the AD of WAS are presented in Table 2. Model simulations and evaluations were performed using 207 the AQUASIM 2.0 [52]. The parameters  $P_M$ ,  $R_M$  and  $\lambda$  were estimated for each model by comparing calculated results with measured data, according to the best fit obtained between the experimental methane production and the model simulation. The parameters were estimated by the weighted least squares method. Model parameters were estimated by

minimizing the sum of the squares of the weighted deviations between measurements and predicted model results.

# **2.6. Fluorescence in situ hybridization (FISH) analysis**

FISH was used to determine the inhibition impact of NPs on Archaea population in the WAS AD. FISH analyses were carried according to the Amann et al. [53]. Oligonucleotide probes were EUB338 mix for all bacteria [54], ARC915 for all Archaea [55], MX825 for *Methanosaeta spp.* [56], MSMX860 for *Methanosarcinales spp.* [56] and MC1109 for *Methanococcus spp.* [56]. The details of the oligonucleotide probes are presented in Table 3. Olympus BX51 fluorescent microscope was used for imaging and images were evaluated by ImageJ program (J 1.47 software) [57].

### **3. Results and Discussion**

#### **3.1. WAS characterization**

The results of WAS characterization analyses are presented in Table 4. TS and VS were found as 182.60 gTS/kgSample (18.26%) and 124.86 gVS/kgSample (12.49%), respectively. Slightly lower results were reported for TS as 14.58% [58] and 13.8% [59] and for VS as 10.63% [58] and 11.8% [59]. Sludge centrifugation applied for dewatering in WWTP results in higher TS and VS values, compared to literature.

**Environmental Science: Processes & Impacts Accepted Manuscript**

Environmental Science: Processes

& Impacts Accepted Manuscript

tCOD values were found as 1862.09 mgCOD/gVS and similar results were also reported by 231 Chang et al. [60] as  $1075 - 3399$  mg/L and by Ji et al. [61] as 2080 mgCOD/gVS. However, relatively higher sCOD values of 198.36 mgCOD/gVS were found than the results given by 233 Chang et al. [60] (as  $5 - 159$  mg/L) and Ji et al. [61] 28.1 mgCOD/gVS.

234 TKN was found as  $96.35$  mgTKN/gVS (1.3%) which is lower than the results (3.73 – 7.81%)

in the literature [62, 63]. The composition of soluble fraction, cellulose, hemicellulose and

Environmental Science: Processes

& Impacts Accepted Manuscript

lignin contents of the WAS were found as 93.64, 0.76, 3.12 and 2.48%, respectively. Even though the results are lower than those of Mottet et al. [64], the findings are in the comparable distribution with the reported values for five different WAS samples. Sugar, protein and extractable material containing lipid content of WAS were determined as 98.30 mgGlu/gVS, 231.25 mgPro/gVS and 0.12%, respectively, and higher results were shown by Mottet et al. 241 [64] as  $170 - 300 \text{ mgGlu/gVS}$ ,  $340 - 470 \text{ mgPro/gVS}$  and  $0 - 90 \text{ mg/gVS}$ , respectively in WAS. The elemental analysis resulted as 32.59% C, 5.29% H, 5.31% N and 0.72% S showing that the results were consistent with the reported compositions by Heo et al. [65].

#### **3.2. NPs long term impacts on AD of WAS**

To determine the long term effects of NPs on WAS AD, 5, 50, 150, 250 and 500 mgNP/gTS 247 (equal to 11, 110, 330, 550 and 1100 mgNP/L) concentrations of Ag, CuO and CeO<sub>2</sub> NPs 248 were used in BMP tests. Methane produced by anaerobic seed sludge  $(31.26 \text{ mLCH}_4/\text{gVS})$ was subtracted from the samples' methane amount to normalize the BMP results. To determine the SDBS effect on methane potential, concentrations of 0.09, 0.86, 2.61, 4.34 and 8.68 mgSDBS/gTS which are equal to the SDBS concentrations added to the BMP reactors with NP suspensions were prepared as controls and their BMP tests were performed under 253 same conditions. Low standard deviation (stdev)  $(2.78 \text{ mLCH}_4/\text{gVS})$  showed that SDBS had a negligible impact on methane production.

Methane productions of raw WAS, anaerobic seed sludge and WAS treated with 5 different 256 concentrations of Ag, CuO and CeO<sub>2</sub> NPs are presented in Fig. 1 (a), (b) and (c), respectively. 257 Methane production of the control sample (raw WAS) was measured as  $72.3 \text{ mLCH}_4/\text{gVS}$ . Concentrations of 150, 250 and 500 mgAg/gTS caused more than 5% inhibition on the methane production of WAS. The observed inhibitions are 6.5, 7.8 and 12.1%, respectively 260 and the decreasing trend of biogas production started after the  $17<sup>th</sup>$  day of the BMP test. No

significant reduction in methane production was observed at 5 and 50 mgAg/gTS concentrations.

There are limited studies in literature evaluating impacts of NPs on AD. Low concentrations of Ag NPs have been mostly investigated and no impact was reported [6, 13, 30]. Barrena et al. [13], investigated the inhibition effects of 10, 16 and 18 mgAg/L NP concentrations on cellulose degradation during 21 days. It was demonstrated that Ag NPs do not have a significant effect on bacterial consortium since the biogas production was not significantly 268 different  $(p<0.05)$  from the control test. Yang et al. [6] examined the inhibition effects of 10, 20 and 40 mgAg/L NP concentrations on WAS AD. Long term (28 days) WAS AD of control and Ag NP treated groups showed no significant difference of biogas production (*p=0.36*) in 271 their study. Garcia et al. [30] employed a higher NP concentration of 170 mgAg/L on cellulose degradation. They also observed no statistical differences between control and Ag NP treated sample. Gonzalez-Estrella et al. [10] exposed anaerobic granular sludge to 1500 mgAg/L for 10 days. They indicated that Ag NPs did not pose a serious inhibitory effect to acetoclastic or hydrogenotrophic methanogenic assays with the normalized specific methanogenic activity value above 70%. The results of this study are consistent with the literature for the lower concentrations. However, when higher concentrations of Ag NPs (150, 278 250 and 500 mgAg/gTS) were used to simulate a worst case scenario, an inhibition more than 10% was observed. In contrast to Gonzalez-Estrella et al's 10 day study, we observed inhibition on biogas production for the higher concentrations of Ag NPs. On the other hand, low inhibition effects were observed after 17 days in this study. Yang et al. [32], determined the impact of Ag NPs on AD of municipal solid waste more than 200 days continuous operation. It is found that 10 mgAgNPs/kgWaste or higher concentrations inhibited methanogenesis in Yang et al. study. They explained inhibition with the silver bioavailability because of slow and long term silver release from nanosilver dissolution under long term AD.

Environmental Science: Processes

& Impacts Accepted Manuscript

Our results are consistent with the results of Yang et al. [32] in the context of longer exposure time. It is thought that the release of Ag ions creates the antibacterial impact of Ag NPs by destroying cell membrane or causing enzyme inactivation or protein dephosphorylation [66]. As seen in the Figure 1 (b), CuO NPs inhibited the methane production beginning from the 150 mgCuO/gTS concentration. Inhibition on methane production occurred at the beginning of the experiment and 150, 250 and 500 mgCuO/gTS concentrations showed severe inhibition. The toxic impact of CuO NPs should be associated with its heavy metal characteristic. There are a very limited number of studies [10, 29, 33, 67] searching the impacts of CuO NPs on AD in literature. Gonzalez-Estrella et al. [10], tested CuO NPs at high concentrations (250, 500 and 1500 mgCuO/L) with methanogenic assays during 10 days. Significant inhibitions were observed and CuO NPs affected the methanogenic activity decreasing the specific methanogenic activity to 79.9, 71.3 and 54.6% for the assays supplied with 250, 500 and 1500 mgCuO/L concentrations, respectively. Obtained results from this study are consistent with the work of Gonzalez-Estrella et al. [10], for the CuO NPs which is thought that copper has a similar inhibition impact mechanism associated with its solubility [29].

302 As seen from the Figure 1 (c),  $CeO<sub>2</sub> NP$  has a positive impact on WAS AD.  $CeO<sub>2</sub>$ concentrations of 150, 250 and 500 mg/gTS increased the methane production to a maximum of 18.8, 25.5 and 9.2%, respectively. There are also very limited studies [10, 30, 68] 305 investigating the impacts of  $CeO<sub>2</sub>$  NPs to AD. Gonzalez-Estrella et al. [10], exposed 306 anaerobic sludge to 1500 mg  $CeO<sub>2</sub> NP/L$  concentration for 10 days, and no inhibition was 307 observed. Garcia et al. [30] found that  $CeO<sub>2</sub> NP$  concentrations <160 mg/L had no effect on methane production and inhibition impact reached to a maximum value when 640 mg/L concentration was applied. In the study of Garcia et al. [30], it is believed that the results were 310 due to the production process of  $CeO<sub>2</sub> NP. CeO<sub>2</sub> NPs$  takes oxygen to its crystal structure

during its synthesis process and has the property of holding the oxygen or releasing it according to the environmental conditions. This property can cause disruption of the respiration mechanism of microorganisms and cause inhibition. Our findings present a different result than Garcia et al. [30] and Gonzalez-Estrella et al. [10] works and there is no literature to compare the findings of this study.

- 
- **3.3. NPs short term impacts on AD of WAS**

The impacts of 5, 50, 150, 250, 500, 750 and 1000 mgNP/gTS concentrations of CuO, Ag and CeO<sub>2</sub> NPs were investigated on WAS AD. SDBS was also added to the stock suspensions of NPs to determine the SDBS effect on anaerobic inhibition. Low stdev (11.3 mL biogas) showed that SDBS had a negligible impact on inhibition of AD. This result is consistent with the results of Mu et al. [28] on the impact of SDBS on AD.

Methane productions of WAS treated with 5, 50, 150, 250, 500, 750 and 1000 mgAg/gTS (equal to 30, 299, 896, 1494, 2987, 4481, 5975 mgAg/L) concentration, control sample without Ag NPs and reference DCP sample are given in Figure 1 (d). In long term tests, Ag NP caused lower inhibition on biogas production especially at higher concentrations. However, similar results are not observed in short term tests. Samples including Ag NPs did not present any significant difference on the biogas production than that of the control sample 329 (Figure 1 (d)). Except WAS containing 1000 mg/gTS concentration of Ag NP, all of the Ag NP samples and control sample biogas productions were determined very close to each other. Stdev and average stdev are calculated as 12 and 9.1 mL, respectively. Therefore, it is concluded that 1000 mgAg/gTS sample has a very low inhibition impact on AD. Ag NPs lower inhibition impact was observed on biogas production as well as its lower positive impact. Inhibition calculations show that all of the impacts were below 10% except 1000 mgAg/gTS concentration. 5, 50, 250, and 500 mgAg/gTS concentrations increased biogas

oð

Environmental Science: Processes

**Impacts Accepted Manuscript** 

336 pressure 2.5%, 1.9%, 4.1%, and 5.8%, respectively. 150 and 1000 mgAg/gTS concentrations decrease biogas pressure 3.73% and 11.3%, respectively. It can be seen from these results that Ag NP do not have a significant impact on AD.

CuO NP is determined as the most inhibitor NP both in BMP and short term inhibition tests. Methane productions of WAS containing 5, 50, 150, 250, 500, 750 and 1000 mgCuO/gTS (equal to 31, 310, 930, 1550, 3100, 4650, 6200 mgCuO/L) concentration, control sample without CuO NP and reference DCP sample are given in Figure 1 (e). As can be seen in the Figure 1 (e), all of the CuO NP concentrations had a negative impact on methane production and caused a production rate below the control reactor. According to the ISO13641-1 345 calculation, the inhibition amounts of 5, 50, 150, 250, 500, 750 and 1000 mgCuO/gTS concentrations are 5.8 30.8, 67.3, 70.5, 74.4, 81.3 and 84%, respectively.

Biogas and methane production dose-response curves during exposure to CuO NP of short 348 and long term tests are presented in Figure 2 (a) and (b), respectively.  $EC_{50}$  values of short 349 and long term inhibition tests were calculated as  $224.2 \text{ mgCuO/gTS} \approx 1388.6 \text{ mgCuO/L}$  and 350 215.1 mgCuO/gTS ( $\approx$ 473 mgCuO/L), respectively. EC<sub>10</sub> and EC<sub>20</sub> values with the confidence 95% intervals are calculated as 134.18 and 161.30 mgCuO/gTS, respectively, for long term inhibition test.

To our knowledge, there are only a few works discussing short term impacts of CuO NPs on AD in the literature [29, 33]. Luna-delRisco et al. [29], applied ISO 13641 test for 7.5 – 480 mgCuO/L (0.8 – 53.3 mgCuO/gTS) concentrations. 15 mgCuO/L (1.7 mgCuO/gTS) concentration caused 30% of inhibition on biogas production compared to the control group without CuO NPs. The results of Luna-delRisco et al. [29] suggest lower inhibition concentrations than the findings of this study. It is thought that the differences between the results are due to the used concentration range for the assays. Gonzalez-Estrella et al. [10], 360 calculated  $EC_{50}$  values for the BMP test lasted 10 days.  $EC_{50}$  values were found as 223

#### **Page 15 of 37 Environmental Science: Processes & Impacts**

mgCuO/L and >1500 mg/L for the acetoclastic methanogens and hydrogenotrophic methanogens, respectively. The determined short term toxic impacts of CuO NP in this study are similar to the work of Gonzalez-Estrella et al. [10].

 CeO<sub>2</sub> NP did not have a significant effect on the AD according to the ISO13641-1 test results. 365 Methane productions of WAS containing 5, 50, 150, 250, 500, 750 and 1000 mgCeO $\gamma$ gTS (equal to 29, 284, 853, 1421, 2843, 4264, 5686 mgNP/L) concentration, control sample 367 without  $CeO<sub>2</sub>$  NPs and reference DCP sample are presented in Figure 1 (f). Only a 5.5% of 368 low inhibition impact was determined at the concentration of 1000 CeO<sub>2</sub>/gTS. CeO<sub>2</sub> NP did not cause a significant decrease in gas production in neither short nor long term inhibition tests.

#### **3.4. Modeling results**

373 Methane production of raw WAS and WAS containing all concentrations of Ag and  $CeO<sub>2</sub>$ 374 NPs were simulated with the regression coefficient  $(R^2)$  of > 0.93 using same kinetic 375 parameters presented in Table 5. According to results, Ag and  $CeO<sub>2</sub>$  NPs did not have an impact on WAS AD. Since, Ag and CeO<sub>2</sub> NPs did not lead to any inhibition, hydrolyses were 377 approximately completed in 3 days.  $k_H$  value was determined as 0.027745 for each 378 concentration of Ag and CeO<sub>2</sub> NPs as well as the raw WAS with the  $R^2$  of > 0.999.

Similarly, methane production from raw WAS and WAS containing 5 and 50 mgCuO/gTS were also modeled with relatively same kinetic parameters. As seen from Table 6, with the 381 increasing concentration of CuO NPs, starting from 150 mgCuO/gTS,  $P_M$  and  $R_M$  values were 382 decreased and  $\lambda$  values are increased. This means that CuO NPs inhibited the methanogenic consortium which resulted with low biogas production and spesific methane production rate. 384 Furthermore, increased  $\lambda$  indicates inhibition and that anaerobic consortium needs more adaptation time for the production of biogas. According to results, WAS containing 250 and

386 500 mgCuO/gTS were simulated with the higher  $R^2$  by Gompertz equation (Table 6), although first order modeling yielded best fit for the raw WAS and WAS containing 5, 50 and 150 mgCuO/gTS (Figure 3).

First order kinetic model was used to evaluate hydrolysis period of raw WAS and WAS treated with 5, 50, 150, 250 and 500 mgCuO/gTS concentrations (Table 7). Hydrolysis period 391 was completed in 3 days for raw WAS and WAS containing 5 mgCuO/gTS.  $k_H$  of WAS started to decrease with the addition of 50 mgCuO/gTS and  $k_H$  reached to 0.001610 d<sup>-1</sup> when WAS dosed with 500 mgCuO/gTS. On the other hand, hydrolysis period observed as 3 days with WAS containing 5, 50 and 150 mgCuO/gTS similar to raw WAS. Furthermore, hydrolysis period extended to 12 and 17 days when WAS contained 250 and 500 396 mgCuO/gTS, respectively, with the significant decrease in  $k<sub>H</sub>$ . There is only one modeling 397 work in literature performed by Doolette et al. [31]. Calculated  $k_H$  values of raw WAS and Ag 398 NP dosed WAS were  $0.13\pm0.020$  and  $0.12\pm0.014$ , respectively. Approximately similar k<sub>H</sub> and methane potential values for raw WAS and WAS dosed with Ag NP were found. Modeling result supported that Ag NP have no inhibition effect on WAS AD. Since there is no reference to modeling of CuO NP inhibition on WAS AD in literature, modeling result from this study is the first confirmation of CuO NP inhibition of the anaerobic sludge degradability.

#### **3.5. FISH results**

The abundance of Archaea was observed in raw WAS (Figure 4a). Observed abundance of Archaea was similar in short and long term tests carried out with WAS containing Ag and CeO<sub>2</sub> NPs (Figure 4b and c). On the other hand, a small decrease in the amount of the total Archaea was observed after the short term test carried out with WAS containing 1000 mgCuO/gTS compared to raw WAS (Figure 4d). Additionally, a remarkable decrease occurred in the amount of Archaea after the long term test of WAS dosed with 500

#### **Page 17 of 37 Environmental Science: Processes & Impacts**

mgCuO/gTS (Figure 4e) parallel to the observed inhibition of methane production in long term test of WAS containing CuO NP. This result suggests that Archaea population is not notably inhibited in short term with the CuO NP, but long term of CuO NP causes inhibition on Achaea.

The amount of *Methanosaeta spp.* population was reduced by exposure of CuO NP, although no change was observed in the amount of *Methanosaeta spp.* in the presence of other NPs at any concentration. With the addition of NPs, the small decrease was observed in the amount of *Methanosarcinales spp.* compared to raw WAS. This observation pointed out that *Methanosarcinales spp.* is not heavily affected by CuO NPs inhibition of WAS AD. On the other hand, *Methanococcus spp.* was detected in raw WAS and survived in WAS containing CuO NP while these species were not observed in the WAS dosed with Ag NP. The absence of *Methanococcus spp.* in WAS including Ag NP indicated the negative effect of Ag NP on this species.

#### **4. Conclusions**

CuO NPs were found to be the most toxic on biogas production both in long and short term inhibition tests. Ag NPs did not cause an inhibition impact in short term test, eventhough in long term tests, Ag NPs caused slight inhibition on biogas production, especially at higher concentrations which was interpreted that exposure time is as important as dosage. As 431 opposed to Ag and CuO NPs, CeO<sub>2</sub> NP caused an increase in methane production in long term BMP test.

433 Ag and  $CeO<sub>2</sub>$  NPs and raw WAS  $k_H$  values were determined as same which revealed no 434 inhibition impact on AD, caused from Ag and  $CeO<sub>2</sub>$  NPs. On the other hand,  $k_H$  values of 435 CuO NPs decreased as the concentrations increase for the First Order model. Estimated  $P_M$ 

Environmental Science: Processes

& Impacts Accepted Manuscript

 

436 and R<sub>M</sub> decreased while the  $\lambda$  increased with the increase of CuO NPs concentration in Gompertz, Logistic and Transference models, which supports the experimental findings from long and short term inhibition tests.

FISH imaging revealed that the abundance of Archaea observed in raw WAS were similar in

440 ISO and BMP tests carried out with WAS containing Ag and  $CeO<sub>2</sub>$  NPs. On the other hand,

CuO NP caused decrease of Archaea.

The results of this study are an important contribution to published data related to inhibition of NPs on WAS AD. Additionally, the presented results emphasize the need for a deeper understanding of the interaction of NPs with the environment considering their rapid increase in industrial use and release to the environment.

#### **5. Acknowledgements**

This study has been financially supported (Grant no: 2012.02.0121.027) by Akdeniz University.

### **6. References**

- [1] Scientific Comitee on Emerging and Newly Identified Health Risks (SCENIHR), Risk assessment of products of nanotecnologies, 2009.
- [2] T.M. Benn, P. Westerhoff, Nanoparticle silver released into water from commercially available sock fabrics, Environ. Sci. Technol. 42 (2008): 4133-4139.
- [3] L.K. Limbach, R. Bereiter, E. Müller, R. Krebs, W.J. Stark, Removal of oxide nanoparticles in a model wastewater treatment plant: Influence of agglomeration and surfactants on clearing efficiency, Environ. Sci. Technol. 42 (2008): 5828-5833.
- [4] M.A. Kiser, H. Ryu, H. Jang, K. Hristoski, P. Westerhoff, Biosorption of nanoparticles to heterotrophic wastewater biomass, Water Res. 44 (2010): 4105-4114.
- [5] Y. Wang, P. Westerhoff, K.D. Hristovski, Fate and biological effects of silver, titanium 462 dioxide and  $C_{60}$  (Fullerene) nanomaterials during simulated wastewater treatment processes, J. Hazard. Mater. 201-202 (2012): 16-22.

[6] Y. Yang, Q. Chen, J.D. Wall, Z. Hu, Potential nanosilver impact on anaerobic digestion at moderate silver concentrations, Water Res. 46 (2012): 1176-1184. [7] A.S. Stasinakis, Review on the fate of emerging contaminants during sludge anaerobic digestion, Bioresour. Technol. 121 (2012): 432-440. [8] Y. Cao, A. Pawlowski, Sewage sludge-to-energy approaches based on anaerobic digestion and pyrolysis: Brief overview and energy efficiency assesment, Renewable and Sustainable Energy Reviews, 16 (2012): 1657-1665. [9] S.K. Brar, M. Verma, R.D. Tyagi, R.Y. Surampalli, Engineered nanoparticles in wastewater and wastewater sludge – Evidence and impacts, Waste Manage. 30 (2010): 504-520. [10] J. Gonzalez-Estrella, R. Sierra-Alvarez, J.A. Field, Toxicity assessment of inorganic nanoparticles to acetoclastic and hydrogenotrophic methanogenic activity in anaerobic granular sludge, J. Hazard. Mater. 260 (2013): 278-285. [11] ENHRES, Engineered nanoparticles: review of health and environmental safety 2010, http://ihcp.jrc.ec.europa.eu/whats-new/enhres-final-report (15.10.2014) [12] A. Kahru, A. Ivask, Mapping the dawn of nanoecotoxicological research, Acc. Chem. Res. 46 (2012): 823-833 [13] R. Barrena, E. Casals, J. Colon, X. Font, A. Sanchez, V. Puntes, Evaluation of the ecotoxicity of model nanoparticles, Chemosphere, 75 (2009): 850-857. [14] R.D. Handy, F. Von Der Kammer, J.R. Lead, M. Hassellöv, R. Owen, M. Crane, The ecotoxicology and chemistry of manufactured nanoparticles, Ecotoxicology, 17 (2008): 287-314. [15] S.J. Klaine, P.J.J. Alvarez, G.E. Batley, T.F. Fernandes, R.D. Handy, D.Y. Lyon, S. Mahendra, M.J. Mclaughlin, J.R. Lead, Nanomaterials in the environment: Behavior, fate, bioavailability and effects, Environ. Toxicol. Chem. 27 (2008): 1825-1851. [16] D.M.A. Alrousan, P.S.M. Dunlop, T.A. McMurray, J.A. Byrne, Photocatalytic 490 inactivation of E. coli in surface water using immobilised nanoparticle  $TiO<sub>2</sub>$  films, Water Res. 43 (2009): 47-54. [17] A. Kahru, H. C. Dubourguier, From ecotoxicology to nanotoxicology, Toxicology, 269 (2010): 105-119. [18] K, Tae Kim, S.J. Klaine, J. Cho, S.H. Kim, S.D. Kim, Oxidative stres responses of 495 Daphnia Magna exposed to  $TiO<sub>2</sub>$  nanoparticles according to size fraction, Sci. Total Environ. 408 (2010): 2268-2272.

- [19] J. Peralta-Videa, L. Zhao, M.L. Lopez-Moreno, G. De La Rosa, J. Hong, J.L. Gardea-Torresdey, Nanomaterials and the environment: A review for the biennium 2008-2010, J. Hazard. Mater. 186 (2011): 1-15.
	- [20] I. Rodea-Palomares, S. Gonzalo, J. Santiago-Morales, F. Leganes, E. Garcia-Calvo, R. Rosal, F. Fernandez-Pinas, An insight into the mechanisms of nanoceria toxicity in aquatic photosynthetic organisms, Aquat. Toxicol. 122-123 (2012): 133-143.
	- [21] X. Zhao, R. Liu, Recent progress and perspectives on the toxicity of carbon nanotubes at organism, organ, cell and biomacromolecule levels, Environ. Int. 40 (2012): 244-256.
- [22] F. Mirzajani, H. Askarı, S. Hamzelou, M. Farzaneh, A. Ghassempour, Effect of silver nanoparticles on *Oryza sativa* L. and its rhizosphere bacteria, Ecotoxicol. Environ. Saf. 88 (2013): 48-54.
- [23] F. Seitz, M. Bundschuh, R.R. Rosenfeldt, R. Schultz, Nanoparticle toxicity in *Daphnia magna* reproduction studies: The importance of test design, Aquat. Toxicol. 126 (2013): 163-168.
- [24] J. Chen, Y.Q. Tang, Y. Li, Y. Nie, L. Hou, X.Q. Li, X.L. Wu, Impacts of different nanoparticles on functional bacterial community in activated sludge, Chemosphere, 104 (2014): 141-148.
- [25] M. Faria, J.M. Navas, A.M.V.M. Soares, C. Barata, Oxidative stress effects of titanium dioxide nanoparticles aggregates in zebrafish embryos, Sci. Total Environ. 470–471 (2014): 379-389.
- [26] L. Nyberg, R.F. Turco, L. Nies, Assessing the impact of nanomaterials on anaerobic microbial communities, Environ. Sci. Technol. 42 (2008): 1938-1943.
- [27] H. Mu, Y. Chen, Long-term effect of ZnO nanoparticles on waste activated sludge anaerobic digestion, Water Res. 45 (2011): 5612-5620.
- 521 [28] H. Mu, Y. Chen, N. Xiao, Effects of metal oxide nanoparticles  $(TiO_2, A_2O_3, SiO_2)$  and ZnO) on waste activated sludge anaerobic digestion, Bioresour. Technol. 102 (2011): 10305-10311.
- [29] M. Luna-delRisco, K. Orupold, H.C. Dubourguier, Particle-size effect of CuO and ZnO on biogas and methane production during anaerobic digestion, J. Hazard. Mater. 189 (2011): 603-608.
- [30] A. Garcia, L. Delgado, J.A. Tora, E. Casals, E. Gonzalez, V. Puntes, X. Font, Carrera, J. A. Sanchez, Effect of cerium dioxide, titanium dioxide, silver and gold nanoparticles on the activity of microbial communities intended in wastewater treatment, J. Hazard. Mater. 64-72 (2012): 199-200.
- 



- [44] E. Bae, H.-J. Park, J. Lee, Y. Kim, J. Yoon, K. Park, K. Choi, J. Yi, Bacterial cytotoxicity of the silver nanoparticle related to physicochemical metrics and agglomeration properties, Environ. Toxicol. Chem. 29 (2010): 2154–2160.
	- [45] H.H. Liu, S. Surawanvijit, R. Rallo, G. Orkoulas, Y. Cohen, Analysis of nanoparticle agglomeration in aqueous suspensions via constant-number Monte Carlo simulation, Environ. Sci. Technol. 45 (2011): 9284–9292.
		- [46] R. Karthik, R. Harish Nagarajan, B. Raja, P. Damodharan, Thermal conductivity of CuO–DI water nanofluids using 3-*ω* measurement technique in a suspended micro-wire, Exp. Therm. Fluid Sci. 40 (2012): 1-9.
	- [47] ISO 13641-1:2003, Water Quality Determination of Inhibition of Gas Production of Anaerobic Bacteria, Part 1: General Test, 2003.
	- [48] I. Buendia, F. Fernandez, J. Villasenor, L. Rodriguez, Feasibility of anaerobic co-digestion as a treatment option of meat industry wastes, Bioresour. Technol. 100 (2009): 1903–1909.
	- [49] A.J. Mawson, R.L. Earle, V.F. Larsen, Degradation of acetic and propionic acids in the methane fermentation, Water Res. 25 (1991): 1549–1554.
	- [50] G. Redzwan, C. Banks, The use of a specific function to estimate maximum methane production in a batch-fed anaerobic reactor. J. Chem. Technol. Biotechnol. 79 (2004): 1174–1178.
	- [51] P. Llabres-Luengo, J. Mata-Alvarez, Kinetic study of the anaerobic digestion of straw-pig manure mixtures. Biomass 14 (1987): 129–142.
	- [52] P. Reichert, J. Ruchti, W. Simon, Aquasim 2.0. Swiss Federal Institute for Environmental Science and Technology (EAWAG), CH-8600 Duebendorf, Switzerland, 1998.
	- [53] R.I. Amann, In situ identification of microorganisms by whole cell hybridization with rRNA-targeted nucleic acid probes. In: A Akkermans, J van Elsas, F de Bruijn (eds) Molecular rmicrobial ecology manual, Kluwer, London, pp MMEM–3.3.6/1–MMEM-3.3.6/15, 1995.
	- [54] R.I. Amann, B.J. Binder, R.J. Olson, S.W. Chisholm, R. Devereux, D.A. Stahl, Combination of 16S rRNA-targeted oligonucleotide probes with flow cytometry for analyzing mixed microbial populations. Appl. Environ. Microbiol. 56 (1990): 1919- 1925.
- 

[55] D.A. Stahl, R. Amann, Development and application of nucleic acid probes. 205-248. In E. Stackebrandt and M. Goodfellow ed., Nucleic acid techniques in bacterial systematics. John Wiley & Sons Ltd., Chichester, England, 1991. [56] L. Raskin, J.M. Stromley, B.E. Rittmann, D.A. Stahl, Group-specific 16SrRNA hybridization probes to describe natural communities of methanogens, Appl. Env. Microbiol. 60 (1994): 1232-1240. [57] W. Rasband, ImageJ: Image Processing and Analysis in Java. 2004. http://rsb.info.nih.gov/ij/ (15.10.2014) [58] W. Qiao, X. Yon, J. Ye, Y. Sun, W. Wang, Z. Zhang, Evaluation of biogas production from different biomass wastes with/without hydrothermal pretreatment, Renew. Energy 36 (2011): 3313-3318. [59] V. Dubrovskis, I. Plume, V. Kotelenecs, E. Zabarovskıs, Anaerobic digestion of sewage sludge. 9th International Scientific Conference Engineering for rural development Proceedings, Volume (9) 2010: May 27-28. [60] C.N. Chang, Y.S. Ma, C.W. Lo, Application of oxidation-reduction potential as a controlling parameter in waste activated sludge hydrolysis, Chem. Eng. J. 90 (2002): 273-281. [61] Z. Ji, G. Chen, Y. Chen, Effects of waste activated sludge and surfactant addition on primary sludge hydrolysis and short-chain fatty aciids accumulation, Bioresour. Technol. 101 (2010): 3457-3462. [62] M.C. Rizk, R. Bergamasco, C.R.G. Tavares, Anaerobic codigestion of fruit and vegetable waste and sewage sludge, Int. J. Chem. React. Eng., 5 (2007): 1–10. [63] S. Babel, J. Sae-Tang, A. Pecharaply, Anaerobic codigestion of sewage and brewery sludge for biogas production and land application, Int. J. Environ. Sci. Tech. 6 (2009): 131-140. [64] A. Mottet, E. Francois, E. Latrille, J.P. Steyer, S. Deleris, F. Vedrenne, H. Carrere, Estimating anaerobic biodegradability indicators for waste activated sludge, Chem. Eng. J. 160 (2010): 488-496. [65] N.H. Heo, S.C. Park, J.S. Lee H. Kang, Solubilization of waste activated sludge by alkaline pretreatment and biochemical methane potential (BMP) tests for anaerobic co-digestion of municipal organic waste, Water Sci. Technol. 48 (2003): 211–219. [66] S. Prabhu, E.K. Poulose, Silver nanoparticles: mechanism of antimicrobial action, synthesis, medical applications and toxictiy effects, Int. Nano Lett. 2 (2012): 32.

Environmental Science: Processes

& Impacts Accepted Manuscript





- **Table 4**. WAS characterization results
	- 664 Table 5. Modeling results of raw WAS and WAS containing Ag and CeO<sub>2</sub> NPs
	- **Table 6.** Modeling results of raw WAS and WAS containing CuO NP
	- **Table 7.** First order kinetic modeling results of raw WAS and WAS containing CuO NP
- 

# **Supplementary File**

- **Figure 1.** FISH images for Raw WAS (a, c, e) and WAS dosed with 1000 mgCuO NP/gTS
- from ISO test (b, d, f). DAPI (blue), all bacteria (green) and (a, b) *Methanosaeta spp.* (red), (c,
- d) *Methanosarcinales spp.* (red), (e, f) *Methanococcus spp.* (red)
- 

**Figure 1.** BMP test results for the concentrations of 5, 50, 150, 250, 500 mgAg/gTS, raw WAS and anaerobic seed sludge (ASS) (a), 5, 50, 150, 250, 500 mgCuO/gTS, raw WAS and ASS (b) and 5, 50, 150, 250, 500 mgCeO $_2$ /gTS, raw WAS and ASS (c), Short term inhibition test results for the concentrations of 5, 50, 150, 250, 500, 750 and 1000 mgAg/gTS, control sample and reference DCP sample (d), 5, 50, 150, 250, 500, 750 and 1000 mgCuO/gTS, control sample and reference DCP sample (e) and 5, 50, 150, 250, 500, 750 and 1000 mgCe $O_2$ /gTS, control sample and reference DCP sample (f)



Environmental Science: Processes

& Impacts Accepted Manuscript





**Figure 2.** Dose-response curves during exposure to CuO NP of short term (a) and long term (b) AD

























# Environmental Science: Processes & Impacts **Page 34 of 37**



![](_page_35_Picture_332.jpeg)

![](_page_36_Picture_315.jpeg)

![](_page_37_Figure_2.jpeg)