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Fluorographane: a promising material for bipolar doping of MoS₂ †

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Received Xth January 2015, Accepted Xth XXXXXXXXXXXX 20XX

First published on the web Xth XXXXXXXXXXXX 200X

DOI: 10.1039/b000000x

Using first principles calculations we investigate the structural and electronic properties of interfaces between fluorographane and MoS₂. Unsymmetrical functionalization of graphene with H and F results in an intrinsic dipole moment perpendicular to the plane of the buckled graphene skeleton. Depending on the orientation of this dipole moment, the electronic properties of a physically absorbed MoS₂ monolayer can be switched from *n*- to *p*-type or vice versa. We show that one can realize vanishing *n*-type/*p*-type Schottky barrier heights when contacting MoS₂ to fluorographane. By applying a perpendicular electric field, the size of the Schottky barrier and the degree of doping can be tuned. Our calculations indicate that a fluorographane monolayer is a promising candidate for bipolar doping of MoS₂, which is vital in the design of novel technological applications based on two-dimensional materials.

1 Introduction

Transition metal dichalcogenides (TMDCs) are a large family of layered materials with diverse electronic, thermal, magnetic and mechanical properties^{1–11}. TMDCs are expected to be used in various technological applications including catalysis⁸, energy storage¹², sensing⁹, and electronic devices such as field-effect transistors (FETs)^{13–16} and logic circuits^{4,9}. The performance of the TMDC based device applications is strongly influenced by the interfaces between TMDCs and metal electrodes. Thus, a detailed knowledge of the electronic properties of the interface between TMDCs and metal electrodes is essential in order to implement TMDCs in efficient device applications. The interface between a two-dimensional (2D) material and a metal electrode plays a key role in the performance of a device constructed from low-dimensional materials because the injection, collection, concentration, and mobility of the charge carriers are mainly determined by the interfaces^{17–24}.

Previous experimental works and theoretical calculations have suggested that the metal-TMDC interface suffers from the presence of a Schottky barrier that results in a higher contact resistance which obstructs the carrier injection across the interface¹⁷. Such Schottky barriers have been considered to originate from the pinning of the Fermi level close to the conduction band of TMDCs. The origin of the Fermi level pinning has been attributed to the modification of the metal work function by the formation of an interface dipole due to the charge redistribution and the formation of gap states. In the case of MoS₂, the latter are mainly of Mo *d*-orbitals char-

acter²⁵. Different scenarios have been proposed to eliminate the Fermi level pinning at the interface. A remedy to mitigate the Fermi level pinning has been proposed in Refs. 26 and 27. They suggested the insertion of a thin oxide layer that acts as a tunnel barrier between the metal electrode and TMDCs, thereby lowering the Schottky barrier height at the metal-TMDCs interface. Alternatively, Çakır *et al.* proposed to use self-assembled monolayers (SAMs) instead of oxides in order to control the Schottky barrier height and the energy level alignment (i.e., position of the metal Fermi level relative to either valence or conduction band edge of the TMDC) at the interface²⁸. Depending on the type of SAMs, it is also possible to dope TMDCs either *n*-type or *p*-type.

Another difficulty with TMDC/metal interfaces is to achieve a *p*-type contact. For instance, while MoS₂ is a promising model system for applications in optoelectronic devices and FETs, its practical use has been restricted to *n*-type transistors^{13,29}. Even if high work function metals are used, MoS₂/metal interfaces mostly show a *n*-type Schottky contact. Such behavior hampers the use of TMDCs in logic circuits since it is necessary to fabricate both *n*- and *p*-type FETs. Interfacing TMDCs with other 2D materials could result in *p*-type doping of TMDCs. Similarly, to realize a vanishing Schottky barrier height at the interface one can insert another 2D material between the metal electrode and TMDC^{30,31}. Musso *et al.* demonstrated that a single layer of graphene-oxide forms a *p*-type contact with MoS₂ and the size of the Schottky barrier at the interface can be tuned by changing the concentration of oxygen³². Moreover, interfacial MoS₂ with graphene-oxide instead of traditional metals weakens the Fermi level pinning at the graphene-oxide/MoS₂ interface. Due to the lack of dangling bonds, individual 2D materials preserve their physical and chemical properties in

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heterostructures.

In order to control or modify the physical and chemical properties of graphene^{33–35}, doping with foreign atoms appears as a promising approach. So far, synthesis of hydrogenated graphene, fluorographene, and chlorographene have been achieved. Theoretical calculations have suggested that interesting materials can be obtained via functionalization of graphene³⁶. Among them, fluorographene is obtained by adsorption of H and F atoms on opposite sides of graphene^{37–40}. Recently, the piezoelectric potential of fluorographene has been shown using first principles calculations^{41,42}.

In this study, we found that contacting MoS₂ to a fluorographene monolayer is a promising strategy for the design of interfaces with zero Schottky barrier height. In addition, bipolar doping of MoS₂ is achievable depending on which side fluorographene is interfaced with the MoS₂ monolayer. We also show that the Schottky barrier height and hence the degree of *n*- or *p*-type doping of MoS₂ can be tuned by using an out of plane electric field.

2 Computational Methodology

First principles calculations based on density functional theory are carried out by using the Vienna ab initio simulation package (VASP)^{43,44} code using the GGA-PBE⁴⁵ functional to treat the electronic exchange correlation interaction. The projected augmented-wave (PAW) method is used to describe the electron and ion interaction. Plane waves up to a kinetic energy cutoff of 400 eV are included in the calculations. For the structure optimizations, the Brillouin zone is sampled by a Γ centered $2 \times 2 \times 1$ *k*-point mesh within the Monkhorst-Pack scheme⁴⁶. To get accurate band structures, each high symmetry direction in the Brillouin zone is sampled by using 12 *k* points. We assume convergence when the difference of the total energies between two consecutive ionic steps is less than 10^{-5} eV and the maximum force allowed on each atom is 0.01 eV/Å. A vacuum region of at least 20 Å along the *z*-direction is used to separate the periodic images in order to avoid spurious interactions. To prevent interaction between the dipoles of repeated slabs, we include a dipole correction in the calculations. To obtain the electrostatic potential, and hence the potential drop at the interface, we calculate the plane averaged potential along the surface normal by using the following expression,

$$\bar{V}(z) = \frac{1}{A} \int \int_{cell} V(x, y, z) dx dy, \quad (1)$$

where *A* is the area of the surface unit cell. Since the semi-local functionals such as GGA fail to describe weakly interacting systems, we also take into account the van der Waals (vdW) interaction within the Grimme approach to provide a

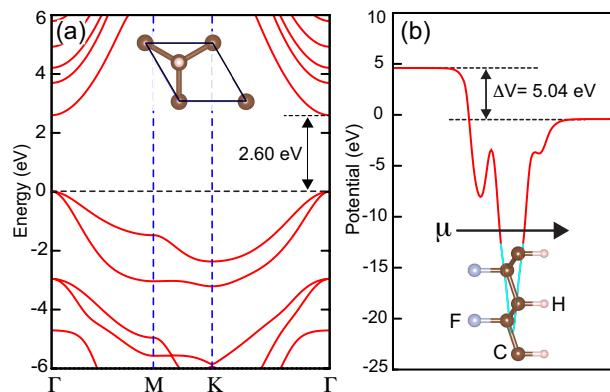


Fig. 1 (a) Calculated band structure of bare fluorographene and (b) plane averaged electrostatic potential energy across bare fluorographene. Black arrow shows the direction of the dipole moment.

better description of the interfaces between the weakly interacting structures⁴⁷. All the pair interactions up to a radius of 25 Å are included in the calculations.

3 Pristine fluorographene

We first investigate the structural and electronic properties of bare fluorographene. Our GGA-PBE calculations predict that fluorographene is a direct band gap semiconductor at the Γ -point with a band gap of 2.6 eV, see Fig. 1(a). Absorption of H and F atoms on graphene results in a buckled structure (with a buckling distance of 0.48 Å) and enlarges the lattice parameter of graphene from 2.54 to 2.57 Å, (see inset of Fig. 1(b)). The C-C, C-H and C-F bond lengths are 1.56, 1.11 and 1.38 Å, respectively. Alternating decoration of pristine graphene with H and F atoms on alternating sides of graphene (see inset of Fig. 1(a)) converts the *sp*² hybridization to *sp*³. Because the H and F atoms are located on opposite sites of the buckled graphene plane, we observe a very large potential step of 5.04 eV along the direction perpendicular to fluorographene, see Fig. 1(b) for the plane averaged electrostatic potential energy across pristine fluorographene. Such a potential drop develops a dipole moment, pointing from the F-side to the H-side.

4 Fluorographene-MoS₂ bilayer heterostructure

Next, we study the structural and electronic properties of the heterostructure made of fluorographene and a monolayer of MoS₂. In order to construct an appropriate supercell with a minimal lattice mismatch, we consider an interface between a 5×5 fluorographene and a 4×4 layer of MoS₂. In this particular geometry, the calculated lattice mismatch between flu-

orographane and MoS₂ is about 2 %. The lattice parameter of fluorographane is forced to be commensurable to the lattice of MoS₂. In the construction of the heterostructure between fluorographane and MoS₂, we consider the electronegativity of the atoms. Since the F and S atoms carry a negative net charge, they avoid each other. In order to get a realistic stacking, we minimize the direct F-S interactions at the interface. In other words, we maximize the interatomic distances between the F and S atoms. In contrast, H carries a positive net charge such that it prefers to stay close to the S atoms. It is possible that the interface structure between fluorographane and MoS₂ is not the lowest energy structure due to the existence of a large number of possible stacking configurations. For this purpose, we performed calculations for very different stacking configurations and found that there is no significant change in the calculated electronic properties. In addition, in a recent work, an ultra smooth potential energy surface was predicted at the fluorographene-MoS₂ interface in the context of a study on friction, meaning that different stacking structures result in similar energetic.⁴⁸

Since fluorographane has two different terminations, two different kinds of heterostructures are possible. In the first (second) heterostructure, MoS₂ is placed on the hydrogen (fluorine) terminated side of fluorographane, as depicted in Fig. 2(a) and Fig. 2(b). In the GGA + vdW optimized structures, the distance between the two planes formed by Mo atoms of the MoS₂ and the midway of buckled C layers is around 6 Å. Apart from structural differences between the two heterostructures, the orientation of the intrinsic dipole moments is opposite with respect to MoS₂, which is expected to result in different electronic properties. For both types of heterostructures, we compute the band structure and obtain the *n*- and *p*-type Schottky barrier heights for MoS₂, see Fig. 2(c) and Fig. 2(d). The *n*-type Schottky barrier height (Φ_n) is defined as the energy difference between the Fermi level and the minimum of the conduction band of MoS₂ contacted to the hydrogenated side of fluorographane. For the system in which MoS₂ is contacted to the fluorinated side of fluorographane, the Fermi level of the combined system appears between the valence band of MoS₂ and the conduction band of fluorographane. For such a system, we define the *p*-type Schottky barrier height (Φ_p) as the energy difference between the minimum of the conduction band of fluorographane and the valence band of MoS₂. The first clear observation is that the position of the Fermi level with respect to the band edges of MoS₂ is totally different. Since MoS₂ is physisorbed on fluorographane, the band structure of MoS₂ is nearly unperturbed. Placing MoS₂ on the hydrogen terminated side of fluorographane yields a *n*-type zero Schottky barrier height. However, as one rotates the dipole of fluorographane by 180° (i.e., putting MoS₂ on the fluorine terminated side) the Fermi level appears very close to the top of the valence band, indicat-

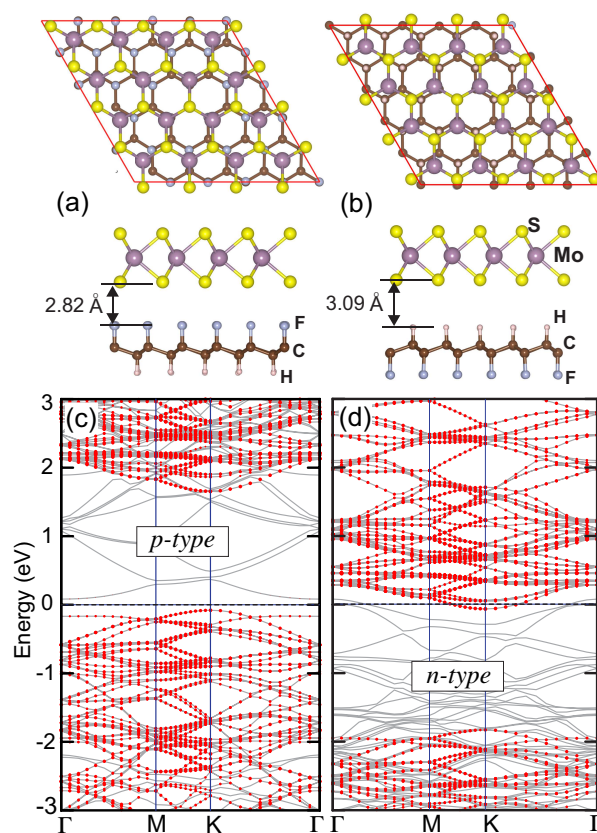


Fig. 2 Top and side views of the fluorographane-MoS₂ heterostructures for F-side (a) and H-side (b) towards MoS₂ and their corresponding band structures in (c) and (d). Red filled circles are used to show the band structure of MoS₂. Fermi level is at 0 eV.

ing a very strong *p*-type doping of the MoS₂ monolayer. The *p*-type Schottky barrier height is found to be 0.16 eV. This means that thermionic emission may obscure such a small Schottky barrier height at finite temperatures⁴⁹. As is clear that by changing the orientation of the intrinsic dipole moment of fluorographane with respect to MoS₂, the type of doping of MoS₂ can be switched from *n*- to *p*-type or vice versa. Without needing an external electric field, we can achieve zero or small Schottky barriers at the interfaces. This controllable doping of MoS₂ allows for the fabrication of MoS₂ based *p-n* junctions and transistors.

To explain Fermi level pinning, different mechanisms have been proposed. Among them, the role of surface states in the pinning of the Fermi level in the band gap of semiconductors was emphasized by Bardeen. The metal-induced gap states mechanism⁵⁰ can be directly correlated to the interfacial gap states that are basically metallic wave functions decaying into the semiconductor⁵¹. In addition, the presence of defect/disorder may result in a high density of interface states at the metal-semiconductor interfaces⁵². In contrast to the

standard metal-TMDC interfaces (for instance Sc-TMDC²⁴), due to the presence of the weak van der Waals interaction, the fluorographane-MoS₂ interface is free of interface states within the band gap of MoS₂, which are considered as one of the main origins of Fermi level pinning. Consequently, the fluorographane-MoS₂ interface does not suffer from Fermi level pinning. It is evident that exfoliated or chemically grown MoS₂ is not absolutely free of defects that may result in Fermi level pinning at the fluorographane-MoS₂ interface. The presence of defects (for instance S vacancy on MoS₂) may lead to localized states within the band gap of MoS₂, at which the Fermi level may be pinned⁵³⁻⁵⁶. Thus, the success of proposed fluorographane-MoS₂ heterostructures is largely dependent on the quality of the MoS₂ and fluorographane monolayers.

It is worth mentioning that the band gap of fluorographane is quite sensitive to the type of stacking. In a previous study, it was shown that while the monolayer of fluorographane is a semiconductor with a band gap of 2.83 eV, bilayer fluorographane exhibits metallic behavior⁵⁷. This is because the uncompensated depolarization field, originating from the parallel dipole moments of the individual layers, in the lowest energy structure of bilayer fluorographane closes the band gap and metallizes bilayer fluorographane. However, if one constructs a non-polar bilayer of fluorographane, the band gap increases to 3 eV. In this stacking type, the same terminated side of individual fluorographane layers face to each other and the net dipole moment becomes zero. Similarly, we observe that when MoS₂ is put on the hydrogenated side of fluorographane, the separation of the valence and conduction band of absorbed fluorographane increases from 2.3 eV to 3.3 eV due to the reduction of the intrinsic dipole moment of fluorographane as a result of charge transfer from fluorographane to MoS₂. This charge transfer creates a dipole moment pointing from MoS₂ to fluorographane, which is antiparallel to the intrinsic dipole moment of fluorographane. Here, we have a gap of 2.3 eV for pristine fluorographane instead of 2.6 eV because of the applied strain to the fluorographane lattice to make it commensurable with the MoS₂ lattice. Applying a 2 % strain decreases the band gap of pristine fluorographane from 2.6 eV to 2.3 eV. Depending on the local structure at the interface, the C-C bond length of bare fluorographane varies within the range of 2.49-2.60 Å.

We also investigate the effect of a perpendicular electric field on the electronic properties of the fluorographane-MoS₂ heterostructure. The concentration of charge carriers on the MoS₂ monolayer in TMDC based FETs can be tuned by an external electric field. Such electric field can be produced by a gate electrode in a real device. Figure 3 shows the *n*-type and *p*-type Schottky barrier heights at the interface of fluorographane and MoS₂ subjected to an out of plane electric field. Here, we calculate the *p*-type (*n*-type) Schottky barrier, Φ_p (Φ_n),

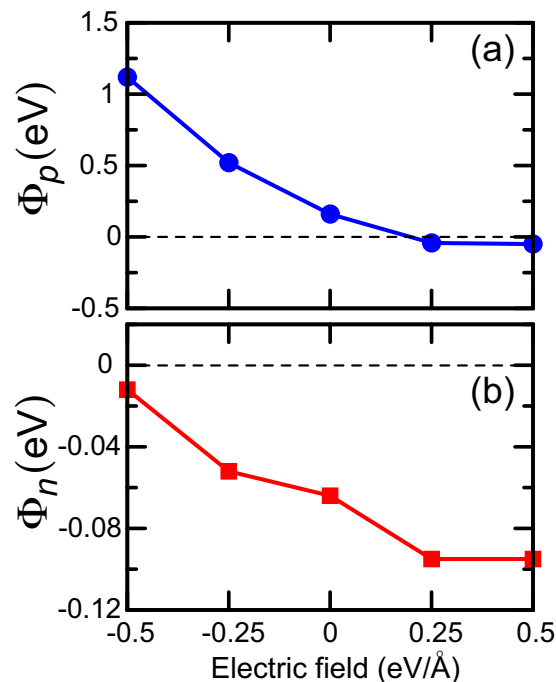


Fig. 3 The *p*- and *n*-type Schottky barriers as a function of external electric field for MoS₂ absorbed on (a) fluorine and (b) hydrogen terminated side of fluorographane, respectively. Φ_p and Φ_n denote the *p*- and *n*-type Schottky barriers, respectively.

(Φ_n), for the MoS₂ monolayer placed on the fluorine (hydrogen) terminated side of fluorographane. As seen from Fig. 3, the application of a negative and positive electric field results in different shifts in the bands of MoS₂. For instance a negative electric field applied to the MoS₂ monolayer absorbed on the F terminated side of fluorographane makes the conduction band of fluorographane closer to the Fermi level and the valence band of MoS₂. This means that we can decrease the size of the *p*-type Schottky barrier height, and hence it is possible to attain a barrierless interface by applying a perpendicular electric field. This easy tunability of the electronic properties allows one to change the *p*-type Schottky barrier height by 1 eV. In contrast, the change in Φ_n for the MoS₂ monolayer interfaced with hydrogen terminated side of fluorographane is only 0.1 eV. As seen from Fig. 3(b), Φ_n is always negative, implying a barrierless interface. When the applied electric field switches from negative to positive, the Fermi level moves upwards within the conduction band of MoS₂, thereby increasing the carrier concentration within this band.

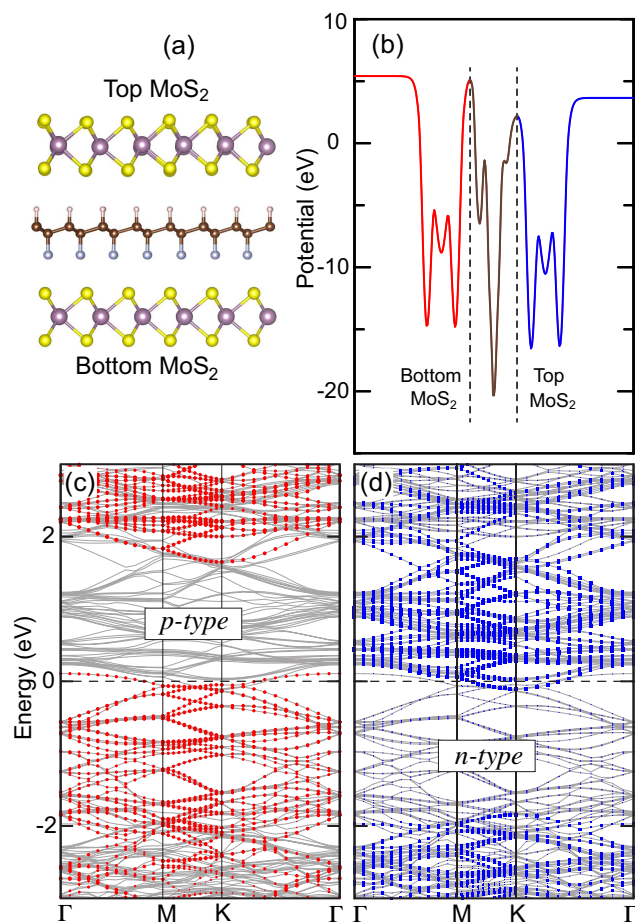


Fig. 4 (a) Optimized structure of bilayer MoS₂ separated by fluorographane. (b) Plane averaged electrostatic potential energy across the MoS₂-fluorographane-MoS₂ trilayer. Calculated band structure projected on the bottom layer in (c) and the top layer (d) MoS₂ states. Red (blue) color is used for bottom (top) MoS₂. Fermi level is at 0 eV.

5 MoS₂-Fluorographane-MoS₂ trilayer heterostructure

We investigate a bilayer of MoS₂ intercalated with a monolayer of fluorographane, see Fig. 4(a). The presence of a potential difference across the fluorographane monolayer breaks the interface symmetry and results in a *p*-type (*n*-type) doping of the bottom (top) MoS₂ monolayer. Such potential difference and the resultant dipole moment triggers a charge transfer from the bottom MoS₂ layer to the top MoS₂ layer. As a result of this charge transfer, while the conduction band of the top MoS₂ monolayer crosses the Fermi level, the valence band of the bottom MoS₂ monolayer appears at the Fermi level. Thus, to obtain such bipolar doping, we do not need to apply an external electric field. In this particular stacking, the

energy gap between the valence and conduction band of fluorographane increases from 2.3 eV to 4.4 eV. We also calculate the plane-averaged electrostatic potential along the normal of the MoS₂-fluorographane-MoS₂ trilayer, see Fig. 4(b). A significant electrostatic potential difference of 1.75 eV is found between the bottom and top MoS₂ monolayers. Besides the intrinsic dipole moment of fluorographane, an additional dipole moment, pointing from the top MoS₂ monolayer to the bottom MoS₂ monolayer, is formed due to the charge transfer. This newly formed dipole moment is antiparallel to the dipole moment of fluorographane and weakens the strength of the latter. Such a device structure can be utilized to construct efficient *p*-*n* junctions, field effect transistors or logic devices that require to have both *n*-type and *p*-type materials together on the nanoscale.

In addition to the MoS₂-fluorographane-MoS₂ trilayer heterostructure as discussed above, we also study trilayers in which both MoS₂ monolayers are put either on the hydrogen or fluorine terminated side of fluorographane. These trilayer structures show similar electronic properties as their corresponding MoS₂-fluorographane bilayer heterostructures as is apparent from Fig. 2. Due to the presence of fluorographane, the bands of the individual MoS₂ layers shift with respect to each other by an amount of 0.15 eV. We demonstrate that the order of the monolayers in the stacking determines the electronic properties. Depending on the type of application, the electronic properties of TMDCs can be tuned by changing the type and sequence of individual monolayers in a stacking, which gives us ample opportunities for the design of new and efficient devices.

It is well known that semi-local exchange correlation functionals (such as GGA) give incorrect descriptions of properties like band gaps, band structures, and energy level alignment between the weakly interacting systems. In order to improve predictions on these properties, more accurate non-local exchange correlation functionals (e.g. hybrid exchange functionals or GW methods) were proposed. In order to check the reliability of our GGA calculations, we performed HSE06 calculations for the selected systems^{58–60}. Although HSE06 has a remarkable effect on the band gap of the MoS₂ and fluorographane monolayers, we found that the Fermi level is predicted to be at nearly the same position relative to the band edges of the individual layers. This means that MoS₂ monolayer exhibits a strong *n* (*p*-type) behavior when being placed on fluorine (hydrogen) terminated side of fluorographane.

6 Conclusion

In summary, we investigated the electronic and structural properties of the fluorographane-MoS₂ heterostructure by means of first-principles calculations based on density functional theory. First of all, as a substrate, the fluorographane

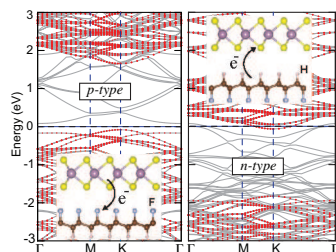


Fig. 5 TOC

monolayer is a suitable choice for MoS₂. We found that the electronic properties of MoS₂ are determined by the type of termination of fluorographane. Contacting MoS₂ to the hydrogen (fluorine) terminated side of fluorographane results in a very strong *n*- (*p*-) type doping of MoS₂. This promising property, originating from the presence of an intrinsic dipole moment, of fluorographane allows us to control the type of doping of MoS₂. Due to the weak interaction between fluorographane and MoS₂, the resulting heterostructure is free of Fermi level pinning. The calculated *n*-type and *p*-type Schottky barrier heights are 0 and 0.16 eV, respectively. Applying an external electric field helps to diminish the *p*-type Schottky barrier height and to control the doping level of the MoS₂ monolayer. Finally, we show that the presence of fluorographane intercalated between two MoS₂ layers initiates a charge transfer between MoS₂ layers in a MoS₂-fluorographane-MoS₂ trilayer. This charge transfer makes it possible to realize both *n*- and *p*-type doped materials together, which is a vital requirement for several device applications.

7 Acknowledgments

This work was supported by the Flemish Science Foundation (FWO-VI) and the Methusalem foundation of the Flemish government. Computational resources were provided by TUBITAK ULAKBIM, High Performance and Grid Computing Center (TR-Grid e-Infrastructure), and HPC infrastructure of the University of Antwerp (CalcUA) a division of the Flemish Supercomputer Center (VSC), which is funded by the Hercules foundation.

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