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# Infrared spectra of small anionic water clusters from density functional theory and wavefunction theory calculations<sup>†</sup>

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We performed systematic theoretical studies on small anionic water/deuterated water clusters  $W/D_{n=2-6}^-$  at both density functional theory (B3LYP) and wavefunction theory (MP2) levels. The focus of the study is to examine the convergence of calculated infrared (IR) spectra with respect to the increasing of the number of diffuse functions. It is found that at the MP2 level for larger clusters ( $n = 4 - 6$ ), only one extra diffuse function is needed to obtain the converged relative IR intensities, while one or two more sets extra diffuse functions are needed for smaller clusters. Such behaviour is strongly associated with the convergence of the electronic structure of corresponding clusters at the MP2 level. It is striking to observe that at the B3LYP level, the calculated relative IR intensities for all the clusters under investigations are diverse and show no trend of convergence when increasing the numbers of the diffuse functions. Moreover, the increasing contribution from the extra diffuse functions to the dynamic IR dipole moment indicates that the B3LYP electronic structure also fails to converge. These results manifest that MP2 is a preferential theoretical method, as compared to the widely used B3LYP, for the IR intensity of dipole bounded electron systems.

## 1 Introduction

Anionic water clusters ( $W_n^-$ ) were first detected by mass spectrometry in the early 1980s,<sup>1,2</sup> and have attracted great attention in many areas<sup>3-7</sup> since then. There are two kinds of structure for  $W_n^-$ : interior structure and surface structure.<sup>8</sup> In the latter structure, which is preferential for small  $W_n^-$ ,<sup>8</sup> the excess electron is bounded outside the molecular kernel through the dipole-electron interaction.

To identify structures of  $W_n^-$ , diverse optical methodologies including photoelectron,<sup>9</sup> vibrational induced autodetachment<sup>10</sup> and fragmentation<sup>11</sup>, vibrational predissociation,<sup>12,13</sup> and time-resolved photoelectron<sup>14,15</sup> spectroscopies have been used. Among them, vibrational predissociation spectroscopy which avoids the so-called “Fano” resonance as well as continuum background<sup>16</sup> can provide high resolution of “fingerprint” vibrational information that is equivalent to

the corresponding infrared (IR) spectrum. The high resolution IR spectrum of  $W_6^-$  in the OH stretching region was first observed by vibrational predissociation spectroscopy in 1998.<sup>12</sup> In addition, to further circumvent the continuum background, anionic deuterated water clusters  $D_n^-$  were used to detect the IR spectra for  $D_3^-$  in the DOD bending region<sup>17</sup> and  $D_{4-6}^-$  in the OD stretching region<sup>13</sup>. Besides, the IR spectra of partially deuterated anionic water hexamers were also reported in the literature.<sup>18</sup>

Because of rapidly increasing number of isoenergetic conformers with the size of  $W_n^-$ ,<sup>19,20</sup> theoretical investigations are essential for identifying their structures. The theoretical fundamental of  $W_n^-$  in the surface state could be traced back to the quantum theory of an electron in the field of a finite dipole.<sup>21</sup> In such case, the value of the dipole moment should be larger than 1.625 Debye to bound an extra electron.<sup>22-26</sup> If further considering the electronic-rotational coupling, it has been proved that the critical value is dependent on the moments of inertia of the molecule.<sup>27</sup> Thus, for real molecules that are free to rotate, the critical dipole moment is in the range from 2.0 to 2.5 Debye.<sup>27,28</sup> Notice that, neutral water monomer has dipole moment around 1.85 Debye,<sup>29</sup> which is larger than the critical value (1.625 Debye) for bounding an extra electron if it is fixed but smaller than the critical value if it is freely rotating. Thus, water monomer even in its lowest rotational level would not bind an extra electron and there was no experimental evidence for it. Indeed, previous theoretical investigations showed that a single water molecule at or near its equilibrium geometry does not have the ability to capture

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**Table 1** Exponents of diffuse functions.

	$m, n$	$O_s$	$O_p$	$H_s$
<i>PmnD</i>	1	0.0105625000000	0.0105625000000	0.0045000000000
	2	0.0013203125000	0.0013203125000	0.0005625000000
	3	0.0001650390625	0.0001650390625	0.0000703125000
<i>AmnD</i>	1	0.0092200000000	0.0074675000000	0.0031575000000
	2	0.0011525000000	0.0009334375000	0.0003946875000
	3	0.0001440625000	0.0001166796875	0.0000493359375
<i>SmnD</i>	1	0.0112537500000	0.0079812500000	0.0040500000000
	2	0.0014067187500	0.0009976562500	0.0005062500000
	3	0.0001758398437	0.0001247070313	0.0000632812500

an extra electron through its dipole.<sup>30,31</sup>

Although the *ab initio* calculations for  $W_{1,2}^-$  were prior to the experimental observation,<sup>30,32</sup> the theoretical IR investigations were initiated in 1999,<sup>18</sup> shortly after the observation of the high resolution experimental spectra.<sup>12</sup> For theoretical IR spectra, density functional theory (DFT) and wavefunction theory (WFT) have been widely used although the former overestimates and the latter (at the Møller-Plesset second order perturbation (MP2)<sup>33</sup> level) underestimates the vertical electron-detachment energy (VDE) of  $W_n^-$ .<sup>34,35</sup> For DFT, the Becke's three-parameters exchange and Lee-Yang-Parr correlation hybrid functional (B3LYP)<sup>36</sup> was most popular<sup>13,18,20,34,37-44</sup> because of its reliable frequency shifts. Through the comparison of the theoretical IR spectra at the B3LYP level with experimental observations, the structures of  $W_{4-6}^-$  were found to be 4Rf, 5W3f, and 6Af, respectively.<sup>13,34</sup> Here we followed the terminology proposed by Kim and co-workers.<sup>34</sup> In these structures, a common motif of "free" water is highlighted for capturing the extra electron (see Fig. 1 for details). For WFT, at the MP2 level, the IR spectra of  $W_{3-5,7}^-$  were predicted.<sup>34,45</sup> In addition, the structure of  $D_3^-$  was identified as 3L by comparing the calculated (at the MP2 and coupled-cluster singles and doubles (CCSD)<sup>46</sup> levels) and measured IR spectra in the bending region.<sup>17</sup> Moreover, the calculated potential energy surface at the MP2 level reveals the quantum probabilistic structure of  $W_2^-$  is a 2CsI structure which is the transition state between two stable conformers.<sup>47</sup>

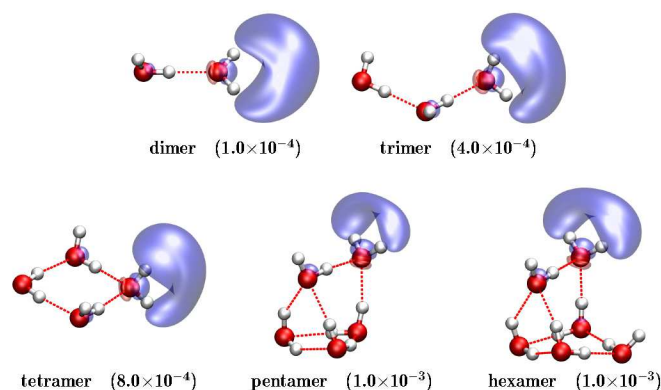
The systematic investigations for the calculated VDEs including their convergence with respect to the diffuse functions have been performed.<sup>48-50</sup> However, to the best of our knowledge, there was no convergence study for the IR spectra with respect to the diffuse functions although it is more important for the structures of small  $W_n^-$ . We should emphasize here, because the low-lying isoenergetic conformers of  $W_n^-$  and their spectral similarity, the systematic investigation for

the IR spectra is remarkably indispensable. For instance, the structure of  $W_6^-$  was first assigned as the "tweezerslike",<sup>37</sup> then the linear network,<sup>18</sup> and finally the 6Af<sup>13,34</sup>. During the evolution, the improvement of theoretical spectra played the crucial role. Therefore, in this work, the convergence of IR spectra for  $W/D_{2-6}^-$  with respect to the diffuse functions is performed. The convergence of dynamic IR dipole moment is also studied, which allows us to inspect the convergence of electronic structures.

## 2 Computational Details

For anionic species, diffuse basis sets are important for the first principles calculations.<sup>48,51-53</sup> Here, we used three basis sets of triple zeta quality for  $W_n^-$ : the Pople's 6-311++G(d,p),<sup>54,55</sup> the augmented Dunning's correlation consistent basis set aug-cc-pVTZ with Davidson's linear transformation for contraction,<sup>56-58</sup> and the Sadlej's polarized basis set<sup>59</sup>. Based on these basis sets, up to three sets of diffuse function were added. They were labeled as *PmnD*, *AmnD*, and *SmnD* with  $m, n = 0, 1, 2, 3$ , respectively. Here  $m$  and  $n$  represent the number of  $s$  and  $p$  diffuse functions for oxygen and the number of  $s$  diffuse function for hydrogen, respectively. Meanwhile, the corresponding exponents used the scaled values of the outermost exponents by 0.125 consecutively.<sup>34</sup> Specifically, all used exponents were listed in Table 1. We should emphasize that the scheme of diffuse functions was widely used for  $W_n^-$ . For instance, P10D is the 6-311++G(d,p)+diff( $sp$ ) used in Ref. 13 for exploring the structures of  $W/D_{4-5}^-$  and A22D is the aug-cc-pVTZ+diff( $2s2p/2s$ ) used in Ref. 45 for calculating the VDEs of  $W_{3-5}^-$ .

The B3LYP functional<sup>36</sup> was chosen for DFT, while, MP2<sup>33</sup> for WFT. As mentioned before, there was no existence of  $W_1^-$ . Thus, here we did not consider this species.



**Fig. 1** Optimized anionic water clusters and their MP2 spin density at the MP2/S33D level. The values in parenthesis are the corresponding iso-values for the MP2 spin density.

The most probable conformers for  $W_{2-6}^-$  related to the experimental observation were considered. Followed by the terminology proposed by Kim and co-workers,<sup>34</sup> they are 2CsI, 3L, 4Rf, 5W3f, and 6Af, respectively. All degrees of freedom of the clusters were optimized at corresponding theoretical levels and all optimized structures at the MP2/S33D level were depicted in Fig. 1. Besides, the distributions of extra electron shown in Fig. 1 demonstrate all species are dipole bounded systems. The pruned “ultrafine” grids were used for numerical integrations in DFT. Our test calculate showed that the total energy converges to  $10^{-6}$  a.u. when more intense “superfine” grids were used. The calculated expectation values of the total spin angular momenta ( $\langle S^2 \rangle$ ) for all species were 0.75 indicating no spin contamination in the calculations. The analytical frequency calculations were performed to confirm that each geometry is a true local minimum except  $W_2^-$ . As mentioned before,  $W_2^-$  is a transition state which has only one imaginary frequency. Unfortunately, we could not locate the 2CsI structure at the B3LYP level as well as the MP2/*AmmD* level.<sup>34</sup> As a result, we used the optimized geometries with corresponding aug-cc-pVDZ+diff basis sets<sup>47</sup> instead of *AmmD* at the MP2 level and did not consider it for B3LYP. The calculated IR intensities were broadened by the Lorentzian function and the frequencies were scaled properly to facilitate comparison with the experimental counterparts. The specific full width at the half-maximum (fwhm) for Lorentzian function as well as scaling factors were given in the following individually. All electronic structure calculations were performed with Gaussian 09 suite of programs.<sup>60</sup>

In both DFT and WFT, the dipole moment could be calculated as

$$\mu = -\text{Tr}(PX) + \sum_A Z_A R_A, \quad (1)$$

where  $P$  is the generalized one-particle density matrix,  $X$  is the multipole matrix,  $Z_A$  is the nuclear charge for atom  $A$ , and

$R_A$  is the corresponding Cartesian coordinates. The dynamic IR dipole moment of the  $k$ -th normal mode ( $d\mu_k$ ) could be calculated from the direct differentiation of dipole moment, *i.e.*,

$$d\mu_k = \frac{\partial \mu}{\partial Q_k} = -\text{Tr} \left( \frac{\partial P}{\partial Q_k} X + P \frac{\partial X}{\partial Q_k} \right) + \sum_A Z_A \frac{\partial R_A}{\partial Q_k}. \quad (2)$$

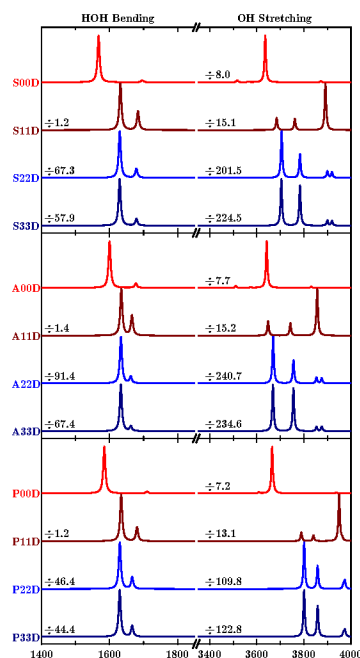
The second term in Eq. 2 represents the nuclear contribution to  $d\mu_k$ . Notice that the Tr would run over all basis set index, thus, the electronic contribution for  $d\mu_k$  in Eq. 2 (the first term) could be easily decomposed to each atomic orbitals. In other words,  $d\mu_k$  could be decomposed to different components, *i.e.*, nuclear, core, valence, polarization, and diffuse. The analytical calculations of  $\partial X/\partial Q_k$  and  $\partial R_A/\partial Q_k$  are straightforward.<sup>61–63</sup> On the other hand, for  $\partial P/\partial Q_k$ , it could be also calculated from the couple-perturbed method analytically.<sup>64,65</sup> However, for the MP2 method, we obtained it numerically by the central finite difference method<sup>66</sup> with  $\Delta Q_k = \pm 1.0 \times 10^{-3}$ . Finally, the projection of all components along the direction of total  $d\mu_k$  was calculated. It is noted that current scheme focuses on the basis set decomposition for  $d\mu_k$ , which is different from the methods based on atomic polar tensors.<sup>67–69</sup> In principle, this scheme is in the framework of Mulliken population analysis<sup>70</sup> and can be easily extended for the non-resonant Raman intensities.<sup>71</sup>

### 3 Results and Discussions

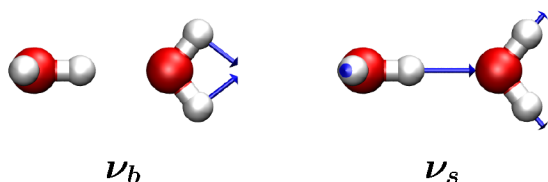
#### 3.1 Dimer

We start our study from  $W_2^-$  which is the simplest species of  $W_n^-$  observed by experiments.<sup>9</sup> Although there were no experimental IR spectra available for  $W_2^-$ , we could benefit from its theoretical IR spectra because they are much simpler than the other larger  $W_n^-$ .

As mentioned before, here we only considered  $W_2^-$  at the MP2 level. Notice that there is a blue shift in frequency in both the bending and stretching regions when adding extra diffuse functions. The blue shift is also observed for other  $W_n^-$  and also at the B3LYP level. By increasing the number of diffuse functions, the calculated IR spectra nearly converged with three sets of extra diffuse functions as shown in Fig. 2. To analyze  $d\mu_k$ , we select two vibrational modes involving the “free” water, namely:  $\nu_b$  in the bending region and  $\nu_s$  in the stretching region. In the bending region, the HOH bending mode ( $\nu_2$ ) of the “free” water is less mixed with those from other water molecules. Thus, here,  $\nu_b$  represents the  $\nu_2$  mode of the “free” water. On the other hand, in the stretching region, the mode mixing is significant for the symmetrical OH stretching modes ( $\nu_1$ ) of the “free” water. We define  $\nu_s$  as the symmetrical mixing of the  $\nu_1$  modes involving the “free” water. The other two important modes, labeled as  $\nu'_s$  and  $\nu_a$  hereafter, are



**Fig. 2** Calculated IR spectra for anionic water dimer at the MP2 level. All calculated spectra were broadened by a Lorentzian function with a fwhm of  $7.5 \text{ cm}^{-1}$ . The values are the corresponding scale factor for the spectra with respect to the maximum intensity in the bending region with S00D, A00D, and P00D, respectively.



**Fig. 3** Schematic vibrational modes for  $\nu_b$  and  $\nu_s$ .

the anti-symmetrical mixing of  $\nu_1$  involving the “free” water and the non-mixing anti-symmetrical OH stretching mode ( $\nu_3$ ) of the “free” water.  $\nu_b$  and  $\nu_s$  always appear at the lowest band in the corresponding region and their schematic drawing were depicted in Fig. 3. It should be noticed that, usually,  $\nu_s$  has less proportion of the “free” water, while,  $\nu_s'$  has more proportion. In the bending region of the calculated IR spectra of  $W_2^-$ , calculations with all three sets of extra diffuse functions show that  $\nu_b$  is the most intense band. In the stretching region,  $\nu_s$  and  $\nu_s'$  both have significant intensities, while the contribution of  $\nu_a$  could be negligible.

The detailed analysis of  $d\mu_k$  for  $W_2^-$  are listed in Table 2. The contribution from diffuse functions dominates  $d\mu_k$  when more than two sets are added. Both the outermost and the second outermost diffuse functions have significant contributions in  $d\mu_k$  with the most diffuse basis sets. On the other

**Table 2** Components of dynamic IR dipole moment for anionic water dimer. N: Nuclear; C: Core basis set; V: Valence basis set; P: Polarization basis set; D: Diffuse basis set; T: Total.

$m, n$	N	C	V	P	D	T
$\nu_b$						
<i>SmnD</i>						
00	-0.49	-0.04	-0.54	-0.26	2.11	0.78
11	-0.11	-0.01	-0.08	0.05	1.00	0.85
22	0.50	0.05	0.10	-0.15	5.94	6.44
33	0.51	0.05	0.10	-0.15	5.49	5.99
<i>AmnD</i>						
00	-0.50	-0.12	0.65	0.05	0.75	0.82
11	-0.01	0.00	0.03	0.00	0.95	0.96
22	0.50	0.12	-0.44	-0.02	7.71	7.88
33	0.51	0.13	-0.45	-0.02	6.55	6.72
<i>PmnD</i>						
00	-0.47	-0.06	0.42	0.09	0.78	0.76
11	-0.25	-0.03	0.04	0.05	1.02	0.83
22	0.55	0.07	-0.23	-0.08	4.86	5.18
33	0.55	0.08	-0.24	-0.08	4.75	5.07
$\nu_s$						
<i>SmnD</i>						
00	0.34	0.03	0.81	-0.29	-0.44	0.45
11	0.36	0.04	-0.26	0.01	1.41	1.57
22	0.45	0.05	-0.15	0.08	10.74	11.16
33	0.45	0.05	-0.14	0.08	11.21	11.64
<i>AmnD</i>						
00	0.39	0.10	0.33	-0.11	-0.26	0.44
11	0.35	0.09	0.03	-0.15	1.49	1.80
22	0.45	0.11	0.15	-0.22	12.29	12.78
33	0.46	0.11	0.15	-0.22	12.08	12.59
<i>PmnD</i>						
00	0.24	0.03	-0.46	0.12	0.39	0.31
11	0.32	0.04	-0.37	-0.04	1.23	1.19
22	0.45	0.06	-0.39	-0.08	7.93	7.98
33	0.46	0.06	-0.40	-0.08	8.37	8.42

hand, the contribution from diffuse function only has minor variation. In addition, the calculated VDE at the MP2/S33D level is 0.018 eV, which confirms that  $W_2^-$  is stable. These results are consistent with the experimental observation<sup>9</sup> of  $W_2^-$ . Consequently, the MP2 level with the most diffuse basis sets used here has the ability to describe the extra electron in  $W_2^-$  because the calculated IR spectra almost converged (only a small variation of  $\nu_s'$  shown in Fig. 2). It is nice to see that the calculated IR spectra with *SmnD* are similar to their counterparts with *AmnD* shown in Fig. 2 (even the two high frequency bands in the stretching region are similar). This result is also consistent with previous theoretical results.<sup>51,72,73</sup> Therefore, for larger  $W_{n \geq 3}^-$ , only *SmnD* and *PmnD* are used in order to

**Table 3** Components of dynamic IR dipole moment for anionic water trimer. N: Nuclear; C: Core basis set; V: Valence basis set; P: Polarization basis set; D: Diffuse basis set; T: Total.

$m, n$	MP2						B3LYP					
	N	C	V	P	D	T	N	C	V	P	D	T
$\nu_b$												
<i>SmnD</i>												
00	0.00	-0.06	-0.89	0.03	1.38	0.48	0.00	-0.06	-0.86	0.01	1.31	0.41
11	-0.01	0.08	0.81	-0.18	-0.22	0.48	0.00	-0.03	0.01	-0.02	0.18	0.14
22	-0.01	0.08	0.82	-0.18	0.04	0.75	0.00	0.00	0.17	-0.03	17.89	18.03
33	-0.01	0.08	0.82	-0.18	0.03	0.75	0.01	-0.07	-0.99	0.20	2.68	1.83
<i>PmnD</i>												
00	0.00	-0.09	0.19	0.07	0.30	0.47	0.00	-0.08	0.04	0.06	0.34	0.36
10	0.00	-0.07	0.04	0.06	0.23	0.27	0.01	-0.10	0.01	0.07	0.41	0.40
11	-0.01	0.11	0.00	-0.06	0.30	0.34	0.01	-0.10	0.04	0.07	0.10	0.12
22	-0.01	0.13	-0.02	-0.07	0.97	1.00	0.00	-0.06	0.00	0.07	1.84	1.85
33	-0.01	0.13	-0.02	-0.07	0.98	0.98	0.00	0.05	0.03	-0.03	2.64	2.68
$\nu_s$												
<i>SmnD</i>												
00	0.00	0.01	-0.08	-0.24	0.62	0.30	0.00	-0.03	-0.45	-0.14	1.01	0.40
11	-0.01	0.07	0.73	0.06	1.10	1.96	-0.01	0.07	0.74	0.08	0.64	1.53
22	-0.01	0.07	0.75	0.05	1.99	2.85	0.00	-0.04	-0.15	0.04	32.53	32.38
33	-0.01	0.07	0.75	0.05	1.97	2.85	-0.01	0.08	1.12	0.08	21.87	23.14
<i>PmnD</i>												
00	0.00	0.01	-0.13	0.01	0.38	0.26	0.00	-0.05	0.44	-0.09	0.12	0.42
10	-0.01	0.09	-0.27	0.04	0.90	0.76	0.00	0.08	-0.57	0.06	1.13	0.69
11	-0.01	0.10	-0.24	0.02	1.55	1.43	0.00	0.09	-0.64	0.06	1.76	1.26
22	0.00	0.11	-0.26	0.02	3.00	2.86	0.00	-0.09	0.32	0.01	5.42	5.66
33	-0.01	0.11	-0.26	0.02	2.98	2.84	0.00	-0.03	0.34	-0.06	5.40	5.66

save computational costs. Furthermore, only  $\nu_b$  and  $\nu_s$  are analyzed in detail for  $d\mu_k$  in the following.

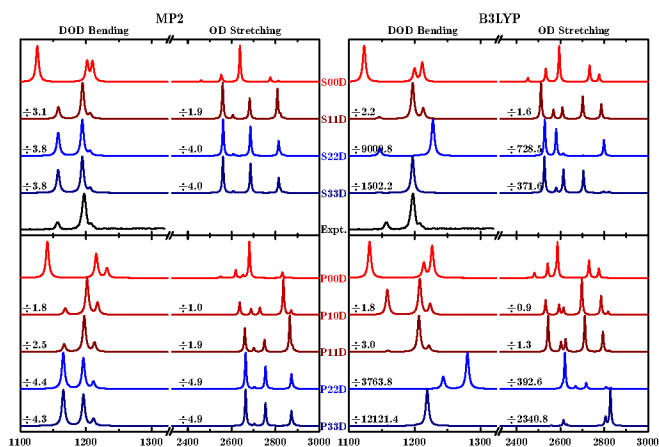
### 3.2 Trimer

$D_3^-$  is the smallest  $W/D_n^-$  for which the experimental IR spectrum has been reported.<sup>17</sup> Because of the low VDE, only the IR signal in the bending region was recorded. Nevertheless, it provides a good reference to validate the theoretical predictions.

All calculated IR spectra for  $D_3^-$  as well as the experimental results<sup>17</sup> were depicted in Fig. 4. At the MP2 level, convergence were reached for the relative IR spectra after two extra diffuse function were added in the calculations. The converged spectrum with *SmnD* agrees well with previous theoretical predictions at the CCSD/aug-cc-pVDZ(2s2p/2s) level in Ref. 17. Furthermore, it is also in good agreement with the experimental spectrum in the bending region, where the most intense band is contributed by  $\nu_2$  of the other two water molecules rather than that of the “free” water. These results indicate that the correlation is adequate for IR at the MP2

level, although MP2 underestimates the VDEs.<sup>34,35</sup> On the other hand, the converged IR spectrum with *PmnD* overestimates the intensity of  $\nu_b$  which becomes the most intense one in the bending region. This result reveals that *SmnD* are better than *PmnD* for IR simulations. In the stretching region, there are three significant bands. Specifically, they are  $\nu_s$ ,  $\nu'_s$  and  $\nu_a$  from low to high frequency. As compared with the spectra of  $W_2^-$ ,  $\nu_a$  is more clear in  $D_3^-$ .

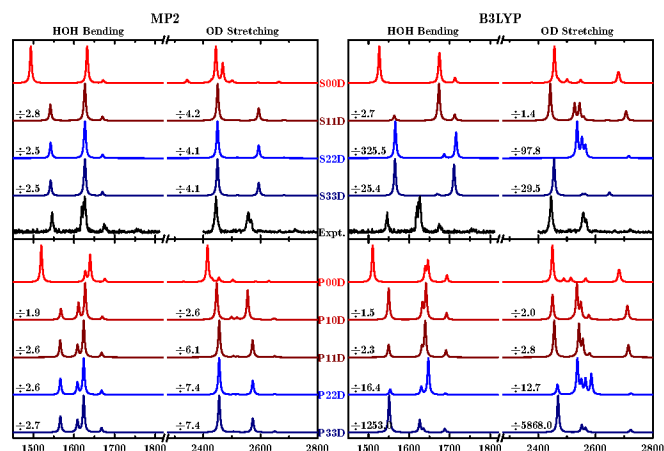
The components of  $d\mu_k$  for  $D_3^-$  listed in Table 3 are fully converged after two sets of extra diffuse functions were added, which is consistent with the convergence of the relative intensities shown in Fig. 4. The contribution of the diffuse functions dominates  $d\mu_k$  for the converged results except for the  $\nu_b$  mode with *SmnD*. Surprisingly, the dominant component for  $\nu_b$  with S33D is the valence part. This result should be attributed to the cancellation of the contribution from diffuse functions. In fact, the second outermost diffuse functions have the largest absolute value in  $d\mu_k$ . The full convergence of all components reveals that the electronic structure of the extra electron in  $D_3^-$  has been adequately described by the MP2 method with two sets of extra diffuse functions. Here the nu-



**Fig. 4** Calculated IR spectra for anionic water trimer at the MP2 and B3LYP levels. All calculated spectra were broadened by a Lorentzian function with a fwhm of  $5.0 \text{ cm}^{-1}$  in the bending region and  $7.5 \text{ cm}^{-1}$  in the stretching region. The frequency scaling factors are 0.97, 0.975, 0.985, and 1.00 in the bending region for the MP2/*Smm*D, MP2/*Pmm*D, B3LYP/*Smm*D, and B3LYP/*Pmm*D, respectively. The values are the corresponding scale factor for the spectra with respect to the maximum intensity of bending and stretching regions with S00D and P00D, respectively. The experimental IR spectrum in the bending region extracted from Ref. 17 was also included for comparison.

clear contributions are all around zero which could be attributed to the fact that the mass ratio between D and O are close to the nuclear charge ratio between them in  $D_n^-$ . Because of all vibrations are invariant for the center of mass, in this case, the magnitude of the nuclear contributions should be around zero.

The calculated relative intensities for  $D_3^-$  at the B3LYP level do not have any trend of convergence as shown in Fig. 4. It is interesting to see that, the result at the B3LYP/P10D level is similar to the converged prediction at the MP2/*Smm*D level in the bending region and seems to reproduce the experimental spectrum reasonably well. However, its large deviation in the stretching region when compared with the MP2/*Smm*D result as well as the unconverged behaviour indicate that such a resemblance is merely a coincidence. This case also highlights the importance of convergence investigation of the present work. Without convergence, comparison between theoretical and experimental results can lead to unreliable interpretations. The components of  $d\mu_k$  in Table 3 show that the diffuse functions make a dominant contribution at the B3LYP level. However, their contribution is rather unstable. Therefore, the variance of the calculated IR spectra could be attributed to the questionable description of the extra electron in  $D_3^-$  at the B3LYP level, which further confirms the coincidence of the results with P10D. It is noted that, in anionic systems, the extra electron has a tendency of partial ionization because of the self-interaction errors associated with the approx-



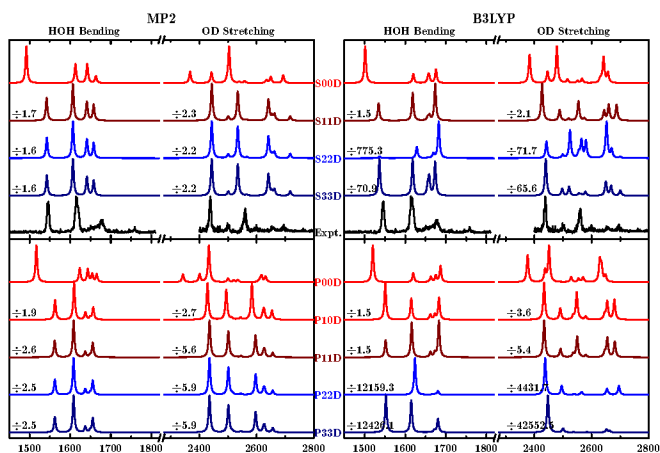
**Fig. 5** Calculated IR spectra for anionic water tetramer at the MP2 and B3LYP levels. All calculated spectra were broadened by a Lorentzian function with a fwhm of  $5.0 \text{ cm}^{-1}$  in the bending region and  $7.5 \text{ cm}^{-1}$  in the stretching region. The frequency scaling factors are 0.96, 0.96, 1.00, 0.99 in the bending region as well as 0.95, 0.92, 0.97, 0.97 in the stretching region for the MP2/*Smm*D, MP2/*Pmm*D, B3LYP/*Smm*D, and B3LYP/*Pmm*D, respectively. The values are the corresponding scale factor for the spectra with respect to the maximum intensity of bending and stretching regions with S00D and P00D, respectively. The experimental IR spectra extracted from Ref. 13 were also included for comparison.

imate DFT functionals.<sup>74,75</sup> This situation would be worse for  $W/D_n^-$  since the electron is just weakly bounded. We have tested seven more representative DFT functionals for trimer. They are PBE,<sup>76</sup> BPL,<sup>77,78</sup> LC-BLYP,<sup>79</sup> M06-2X,<sup>80</sup> CAM-B3LYP,<sup>81</sup>  $\omega$ B97XD,<sup>82</sup> and B2PLYP<sup>83</sup> functionals. These functionals contain self-interaction corrected correlation (PL in BPL),<sup>78</sup> different fraction of exact exchange (CAM-B3LYP, M06-2X,  $\omega$ B97XD, and B2PLYP),<sup>80-83</sup> asymptotic correction by the range-separated method (CAM-B3LYP, LC-BLYP, and  $\omega$ B97XD),<sup>79,81,82</sup> dispersion corrected functional ( $\omega$ B97XD),<sup>82</sup> and the nonlocal perturbative correlations (B2PLYP)<sup>83</sup>. Calculated results show that the electronic self-consistent field fails at the M06-2X/S22D level, and geometrical optimizations fail at the PBE/S22D, BPL/S22D, and B2PLYP/S22D levels. Three range-separated functionals can give successful results with all *Smm*D basis sets. However, the convergence of IR spectra is not reached with the CAM-B3LYP (see ESI† for details), although it has two extra parameters for the range-separated algorithm.<sup>81</sup> On the other hand, the calculated IR spectra converged with the LC-BLYP and  $\omega$ B97XD. We find that the converged IR spectra at the LC-BLYP level in the bending region are similar with that at the MP2 level and, in the stretching region, the results from  $\omega$ B97XD and MP2 are similar (see ESI† for details). Nevertheless, we also find that the results from LC-BLYP in the

**Table 4** Components of dynamic IR dipole moment for anionic water tetramer. N: Nuclear; C: Core basis set; V: Valence basis set; P: Polarization basis set; D: Diffuse basis set; T: Total.

$m, n$	MP2						B3LYP					
	N	C	V	P	D	T	N	C	V	P	D	T
$v_b$												
<i>S<sub>m</sub>nD</i>												
00	-0.42	-0.04	-4.80	-1.69	7.74	0.78	-0.46	-0.04	-3.80	-1.83	6.77	0.64
11	0.15	0.01	0.22	0.05	0.43	0.87	-0.29	-0.03	-3.95	-0.91	5.57	0.39
22	0.02	0.00	-1.22	-0.22	2.22	0.80	-0.57	-0.05	-5.95	-1.27	19.39	11.55
33	0.02	0.00	-1.22	-0.22	2.22	0.80	-0.49	-0.04	-5.50	-1.22	10.49	3.24
<i>P<sub>m</sub>nD</i>												
00	-0.42	-0.06	0.35	0.09	0.78	0.74	-0.44	-0.06	0.15	0.08	0.92	0.64
10	-0.20	-0.03	0.02	0.05	0.72	0.56	-0.44	-0.06	0.03	0.08	1.12	0.73
11	0.25	0.03	-0.19	-0.03	0.75	0.82	-0.36	-0.05	0.08	0.07	0.83	0.56
22	0.22	0.03	-0.18	-0.03	0.75	0.80	-0.50	-0.07	0.18	0.07	1.32	1.01
33	0.23	0.03	-0.18	-0.03	0.75	0.80	-0.39	-0.05	0.08	0.06	22.97	22.66
$v_s$												
<i>S<sub>m</sub>nD</i>												
00	-0.01	0.06	2.67	-2.09	-0.23	0.41	0.00	0.04	0.35	-0.66	0.44	0.17
11	-0.01	0.09	-0.92	-1.15	4.78	2.79	-0.01	0.07	-0.85	-1.15	3.53	1.59
22	-0.01	0.09	-0.85	-1.14	4.68	2.77	0.00	0.05	-0.40	-0.86	2.56	1.34
33	-0.01	0.09	-0.85	-1.14	4.68	2.77	0.00	-0.05	0.42	0.86	5.99	7.22
<i>P<sub>m</sub>nD</i>												
00	0.00	0.07	-0.95	0.22	0.85	0.18	0.00	-0.07	1.18	-0.23	-0.78	0.12
10	-0.01	0.12	-1.10	0.24	2.20	1.45	-0.01	0.10	-1.64	0.23	2.23	0.91
11	-0.01	0.12	-1.08	0.22	2.98	2.23	-0.01	0.10	-1.65	0.23	2.63	1.31
22	-0.01	0.12	-1.08	0.22	3.20	2.45	0.00	-0.04	0.90	-0.16	0.78	1.48
33	-0.01	0.12	-1.08	0.22	3.20	2.46	0.00	-0.07	1.16	-0.18	59.14	60.04





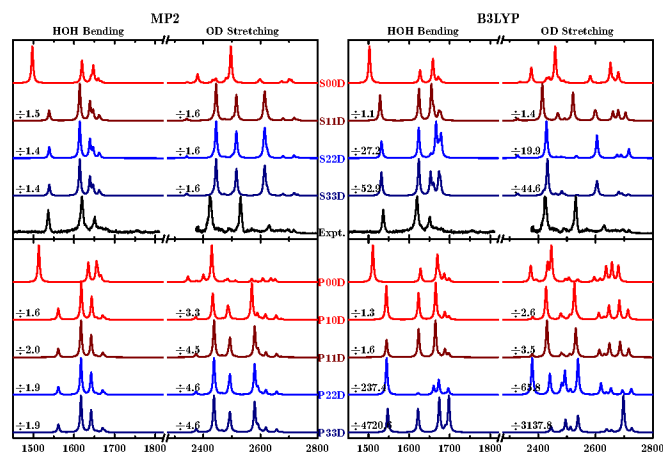
**Fig. 6** Calculated IR spectra for anionic water pentamer at the MP2 and B3LYP levels. All calculated spectra were broadened by a Lorentzian function with a fwhm of  $5.0 \text{ cm}^{-1}$  in the bending region and  $7.5 \text{ cm}^{-1}$  in the stretching region. The frequency scaling factors are 0.96, 0.96, 0.98, 0.99 in the bending region as well as 0.97, 0.93, 0.975, 0.97 in the stretching region for the MP2/*SmmD*, MP2/*PmmD*, B3LYP/*SmmD*, and B3LYP/*PmmD*, respectively. The values are the corresponding scale factor for the spectra with respect to the maximum intensity of bending and stretching regions with S00D and P00D, respectively. The experimental IR spectra extracted from Ref. 13 were also included for comparison.

stretching region and  $\omega\text{B97XD}$  in the bending region are different from the corresponding MP2 results (see ESI† for details). These results highlight the role of the self-interaction errors for  $\text{W}/\text{D}_n^-$  and encourage us to find the suitable functionals for current systems, which would be our future work.

### 3.3 Tetramer, Pentamer, and Hexamer

There were high experimental resolution IR spectra for  $\text{W}_{4-6}^-$  in the bending region and  $\text{D}_{4-6}^-$  in the stretching region.<sup>13</sup> Here we present our calculation results.

The calculated IR spectra of  $\text{W}_4^-$  in the bending region and  $\text{D}_4^-$  in the stretching region were depicted in Fig. 5. At the MP2 level, the relative IR intensities show convergence when only one extra diffuse function was added. The band assignment for *SmmD* and *PmmD* is the same. Specifically, in the bending region of the converged IR spectra, the first band is  $\nu_b$  as discussed before. The second band, *i.e.* the most intense band, belongs to  $\nu_2$  of the two water molecules connected with the “free” water (see Fig. 1 for the details). There are two almost degenerate vibrational modes and the one with slightly higher frequency is much intense than the lower one because of the concerted  $\nu_2$  motion of these water molecules. The third band belongs to  $\nu_2$  of the water molecule that is the farthest from the extra electron. In the stretching region, there



**Fig. 7** Calculated IR spectra for anionic water hexamer at the MP2 and B3LYP levels. All calculated spectra were broadened by a Lorentzian function with a fwhm of  $5.0 \text{ cm}^{-1}$  in the bending region and  $7.5 \text{ cm}^{-1}$  in the stretching region. The frequency scaling factors are 0.96, 0.96, 0.98, 0.99 in the bending region as well as 0.97, 0.93, 0.97, 0.97 in the stretching region for the MP2/*SmmD*, MP2/*PmmD*, B3LYP/*SmmD*, and B3LYP/*PmmD*, respectively. The values are the corresponding scale factor for the spectra with respect to the maximum intensity of bending and stretching regions with S00D and P00D, respectively. The experimental IR spectra extracted from Ref. 13 were also included for comparison.

are two bands. The lower band is  $\nu_s$  and the higher band is  $\nu_a$ . Here, the intensity of  $\nu_s'$  is suppressed because of the cancellation from the  $\nu_1$  motion. The present assignment is consistent with that in Ref. 13. Although the calculated IR spectra with *PmmD* show splittings in the main band in the bending region, there is no splitting of  $\nu_a$  in the stretching region. Thus, the two splitting bands observed in experiments would probably be attributed to some other isoenergetic conformers of  $\text{W}/\text{D}_4^-$ . This has also been addressed in the previous investigations for VDEs.<sup>43,84</sup>

The detailed components of  $d\mu_k$  for  $\text{W}/\text{D}_4^-$  were listed in Table 4. Although the relative intensities are converged when only one extra diffuse function was added, the absolute value of  $d\mu_k$  converges apparently after two sets of extra diffuse function were added. The largest absolute contribution comes from the second outermost *s* diffuse functions of H and O, which is consistent with the convergence of components of  $d\mu_k$ . Except diffuse functions, other terms also have significant contribution to  $d\mu_k$ . For instance, the valence functions have negative contribution to the  $\nu_b$  with *SmmD* and  $\nu_s$  with *PmmD*, while, both valence and polarization functions contribute to the  $\nu_s$  with *SmmD*. Here we also noticed that the nuclear component for  $\nu_b$  with *SmmD* is almost perpendicular to the total  $d\mu_k$ , while with *PmmD*, it is not.

At the B3LYP level, we find that, with S11D, P10D, and

**Table 5** Components of dynamic IR dipole moment for anionic water pentamer. N: Nuclear; C: Core basis set; V: Valence basis set; P: Polarization basis set; D: Diffuse basis set; T: Total.

$m, n$	MP2						B3LYP					
	N	C	V	P	D	T	N	C	V	P	D	T
$\nu_b$												
<i>SmnD</i>												
00	-0.46	-0.04	-3.74	-2.15	7.08	0.69	-0.51	-0.04	-2.63	-2.10	5.84	0.56
11	0.12	0.01	-0.21	-0.19	0.96	0.71	-0.33	-0.03	-4.61	-0.93	6.36	0.46
22	0.02	0.00	-1.48	-0.39	2.50	0.65	-0.53	-0.05	-5.72	-1.03	10.83	3.51
33	0.02	0.00	-1.49	-0.39	2.51	0.64	-0.39	-0.03	-3.46	-0.63	9.18	4.67
<i>PmnD</i>												
00	-0.49	-0.07	0.18	0.11	0.91	0.65	-0.52	-0.07	-0.11	0.10	1.14	0.53
10	-0.24	-0.03	0.14	0.05	0.73	0.64	-0.49	-0.07	0.04	0.09	1.07	0.65
11	0.21	0.03	0.01	-0.04	0.50	0.70	-0.39	-0.05	0.09	0.07	0.73	0.44
22	0.15	0.02	0.03	-0.03	0.48	0.66	0.56	0.08	-0.05	-0.10	6.24	6.73
33	0.15	0.02	0.03	-0.03	0.49	0.66	-0.26	-0.04	0.11	0.05	59.35	59.21
$\nu_s$												
<i>SmnD</i>												
00	0.00	0.00	1.35	-0.47	-0.16	0.73	0.00	-0.02	-0.01	0.02	0.99	0.98
11	0.00	0.04	1.39	-0.10	0.70	2.01	0.00	0.03	0.85	0.02	0.71	1.61
22	0.00	0.04	1.42	-0.11	0.62	1.97	0.00	-0.05	-0.29	0.25	6.48	6.40
33	0.00	0.04	1.42	-0.11	0.62	1.97	0.00	-0.04	-1.17	0.04	10.22	9.05
<i>PmnD</i>												
00	0.00	0.00	-0.74	0.12	1.18	0.56	0.00	-0.05	0.04	-0.01	0.93	0.91
10	0.00	0.04	-0.70	0.18	1.77	1.29	0.00	0.02	-0.89	0.17	1.94	1.24
11	0.00	0.06	-0.90	0.19	2.52	1.86	0.00	0.05	-1.23	0.20	2.50	1.52
22	0.00	0.06	-0.90	0.19	2.57	1.92	0.01	-0.10	1.38	-0.21	42.27	43.36
33	0.00	0.06	-0.90	0.19	2.57	1.92	0.00	-0.08	0.49	-0.03	133.98	134.36

P11D, the calculated spectra resemble some features of the experimental observation. Nonetheless, when more diffuse functions were added, the calculated results are totally different with the observations. The components in Table 4 show that the diffuse functions dominate the contribution in  $d\mu_k$  when even only one set of diffuse functions was added. Thus, the discrepancy between the calculated and experimental spectra should be attributed to the wrong description of the extra electron at the B3LYP level when more diffuse functions were used. Here, we should emphasize that the previous assignment for the structure of  $W/D_4^-$ , even though correct, was based on the unconverged calculation results.<sup>13,34</sup> The experimental and theoretical IR spectra are in agreement with each other merely by coincidence.

The calculated IR spectra for  $W/D_{5,6}^-$  were depicted in Figs. 6 and 7, respectively. At the MP2 level, the relative IR intensities again converge with one extra diffuse function. The assignment for converged spectra in the bending region is the same as that for  $W_4^-$ . The calculated spectra in this region are in agreement with the experimental observations. On the other hand, there are three bands in the stretching region. The same

as in  $D_3^-$ , these bands are assigned to  $\nu_s$ ,  $\nu'_s$  and  $\nu_a$  from low to high frequency, respectively. The calculated relative intensities of  $\nu_s$  and  $\nu'_s$  are comparable to the experimental results. However, the intensity of  $\nu_a$  is significantly overestimated by the MP2 method.

The components of  $d\mu_k$  for  $W/D_{5,6}^-$  in Tables 5 and 6 show that they are certainly converged when two sets of diffuse functions were added. With S33D and P33D, the largest absolute contribution of individual basis set belongs to the second outermost diffuse functions, which is consistent with the convergence of the total  $d\mu_k$ . For  $\nu_b$ , with *SmnD*, the diffuse functions have the largest contribution and the valence components have considerably negative contribution, while with *PmnD*, the diffuse functions dominant. For  $\nu_s$ , with *SmnD*, the valence components are the major part due to the cancellation of individual contributions from diffuse functions. However, with *PmnD*, the contributions from diffuse functions are dominant and the valence has the most negative contribution.

At the B3LYP level with S11D, P10D, and P11D, the calculated IR spectra in the stretching region are in agreement with experimental spectra (here the relative intensity of  $\nu_a$  is lower

**Table 6** Components of dynamic IR dipole moment for anionic water hexamer. N: Nuclear; C: Core basis set; V: Valence basis set; P: Polarization basis set; D: Diffuse basis set; T: Total.

$m, n$	MP2						B3LYP					
	N	C	V	P	D	T	N	C	V	P	D	T
$\nu_b$												
<i>S<sub>mn</sub>D</i>												
00	-0.48	-0.04	-3.90	-1.44	6.56	0.70	-0.52	-0.05	-2.31	-1.50	4.97	0.60
11	-0.19	-0.02	-3.61	-0.66	4.92	0.45	-0.51	-0.04	-5.03	-1.17	7.27	0.51
22	-0.27	-0.02	-4.35	-0.78	5.88	0.46	-0.08	-0.01	-1.39	-0.34	3.93	2.11
33	-0.27	-0.02	-4.35	-0.78	5.88	0.46	-0.43	-0.04	-3.47	-0.85	8.27	3.48
<i>P<sub>mn</sub>D</i>												
00	-0.50	-0.07	0.06	0.12	1.06	0.67	-0.51	-0.07	-0.22	0.10	1.27	0.58
10	-0.36	-0.05	0.11	0.07	0.70	0.47	-0.55	-0.08	-0.12	0.11	1.26	0.62
11	0.06	0.01	0.04	-0.02	0.37	0.45	-0.54	-0.07	-0.08	0.10	1.08	0.49
22	-0.03	0.00	0.06	0.00	0.40	0.43	-0.04	-0.01	-0.06	0.01	8.95	8.85
33	-0.03	0.00	0.07	0.00	0.40	0.43	-0.58	-0.08	0.00	0.11	32.11	31.55
$\nu_s$												
<i>S<sub>mn</sub>D</i>												
00	0.00	0.02	2.01	-0.55	-0.78	0.70	0.00	-0.01	0.44	-0.07	0.39	0.75
11	0.00	0.04	1.89	-0.22	0.12	1.82	0.00	0.04	1.34	-0.10	0.10	1.37
22	0.00	0.04	1.93	-0.22	0.04	1.79	0.00	0.00	-0.92	-0.16	6.25	5.18
33	0.00	0.04	1.93	-0.22	0.04	1.79	0.00	-0.05	-0.73	0.35	8.17	7.74
<i>P<sub>mn</sub>D</i>												
00	0.00	0.02	-1.04	0.17	1.34	0.49	0.00	-0.04	-0.12	0.01	0.81	0.67
10	0.00	0.06	-1.02	0.24	1.97	1.26	0.00	0.04	-1.35	0.24	2.14	1.07
11	0.00	0.07	-1.17	0.25	2.60	1.74	0.00	0.06	-1.61	0.27	2.60	1.31
22	0.00	0.07	-1.17	0.25	2.60	1.75	0.00	-0.03	0.83	-0.15	3.64	4.29
33	0.00	0.07	-1.17	0.25	2.60	1.75	0.01	-0.11	1.76	-0.26	14.51	15.91

than that at the MP2 level). However, in the bending region, the calculated results are quite different from the experimental counterparts. Furthermore, when more diffuse functions were added, the IR spectra are diverse in both regions. Thus, we would attribute the similarity in the stretching region to coincidence again. The detailed components in Tables 5 and 6 indicate that the diffuse functions dominate in  $d\mu_k$  and become larger and larger. This result reveals that B3LYP has a questionable description of the extra electron in  $W/D_{5,6}^-$ .

## 4 Conclusions

In this work, we have examined the convergence of calculated IR spectra for  $W/D_{2-6}^-$  with respect to the number of diffuse functions at both MP2 and B3LYP levels. For the MP2 method, the convergence can be achieved with respect to the used basis sets. The converged IR spectra are in good agreement with experimental observations. It can be concluded that the MP2 method has the ability to correctly describe the extra electron in the dipole bounded systems. In contrast, the convergence has never been obtained for all clusters under in-

vestigations at the B3LYP level. The contribution of diffuse functions to the properties becomes larger with the increased number of diffuse functions. It certainly implies that it is questionable to use B3LYP to study the systems bounded with an extra electron, although such excises can be frequently found in the literature. The basis set dependence could be a useful measure to examine the applicability of some modern DFT functionals (also high level post Hartree-Fock methods), in particular, the functionals with correct asymptotic decaying as well as the doubly hybrid functionals, for such anionic water clusters.

## 5 Acknowledgments

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## References

- M. Armbruster, H. Haberland and H.-G. Schindler, *Phys. Rev. Lett.*, 1981, **47**, 323–326.
- H. Haberland, H.-G. Schindler and D. R. Worsnop, *Ber. Bunsenges. Phys. Chem.*, 1984, **88**, 270–272.
- E. Hart and M. Anbar, *The Hydrated Electron*, Wiley-Interscience, New York, 1970.
- P. J. Campagnola, D. J. Lavrich, M. J. DeLuca and M. A. Johnson, *J. Chem. Phys.*, 1991, **94**, 5240–5242.
- S. T. Arnold, R. A. Morris, A. A. Viggiano and M. A. Johnson, *J. Phys. Chem. A*, 1996, **100**, 2900–2906.
- M. K. Beyer, B. S. Fox, B. M. Reinhard and V. E. Bondybej, *J. Chem. Phys.*, 2001, **115**, 9288–9297.
- C. Desfrancois, S. Carles and J. P. Schermann, *Chem. Rev.*, 2000, **100**, 3943–3962.
- R. N. Barnett, U. Landman, C. L. Cleveland and J. Jortner, *J. Chem. Phys.*, 1988, **88**, 4429–4447.
- J. V. Coe, G. H. Lee, J. G. Eaton, S. T. Arnold, H. W. Sarkas, K. H. Bowen, C. Ludewigt, H. Haberland and D. R. Worsnop, *J. Chem. Phys.*, 1990, **92**, 3980–3982.
- A. Bar-on and R. Naaman, *J. Chem. Phys.*, 1989, **90**, 5198–5199.
- L. A. Posey and M. A. Johnson, *J. Chem. Phys.*, 1988, **89**, 4807–4814.
- P. Ayotte, C. G. Bailey, J. Kim and M. A. Johnson, *J. Chem. Phys.*, 1998, **108**, 444–449.
- N. I. Hammer, J.-W. Shin, J. M. Headrick, E. G. Diken, J. R. Roscioli, G. H. Weddle and M. A. Johnson, *Science*, 2004, **306**, 675–679.
- A. E. Bragg, J. R. R. Verlet, A. Kammrath, O. Cheshnovsky and D. M. Neumark, *Science*, 2004, **306**, 669–671.
- D. H. Paik, I.-R. Lee, D.-S. Yang, J. S. Baskin and A. H. Zewail, *Science*, 2004, **306**, 672–675.
- C. G. Bailey, J. Kim and M. A. Johnson, *J. Phys. Chem.*, 1996, **100**, 16782–16785.
- N. I. Hammer, J. R. Roscioli, M. A. Johnson, E. M. Myshakin and K. D. Jordan, *J. Phys. Chem. A*, 2005, **109**, 11526–11530.
- P. Ayotte, G. H. Weddle, C. G. Bailey, M. A. Johnson, F. Vila and K. D. Jordan, *J. Chem. Phys.*, 1999, **110**, 6268–6277.
- C. J. Tsai and K. Jordan, *Chem. Phys. Lett.*, 1993, **213**, 181–188.
- S. B. Suh, H. M. Lee, J. Kim, J. Y. Lee and K. S. Kim, *J. Chem. Phys.*, 2000, **113**, 5273–5277.
- R. F. Wallis, R. Herman and H. W. Milnes, *J. Mol. Spectrosc.*, 1960, **4**, 51–74.
- K. Fox and J. E. Turner, *J. Chem. Phys.*, 1966, **45**, 1142–1144.
- M. Mittleman and V. Myerscough, *Phys. Lett.*, 1966, **23**, 545–546.
- J.-M. Lévy-Leblond, *Phys. Rev.*, 1967, **153**, 1–4.
- C. A. Coulson and M. Walmsley, *Proc. Phys. Soc.*, 1967, **91**, 31–32.
- W. B. Brown and R. E. Roberts, *J. Chem. Phys.*, 1967, **46**, 2006–2007.
- W. Garrett, *Chem. Phys. Lett.*, 1970, **5**, 393–397.
- O. H. Crawford, *Mol. Phys.*, 1971, **20**, 585–591.
- S. A. Clough, Y. Beers, G. P. Klein and L. S. Rothman, *J. Chem. Phys.*, 1973, **59**, 2254–2259.
- K. D. Jordan and J. J. Wendoloski, *Chem. Phys.*, 1977, **21**, 145–154.
- D. M. Chipman, *J. Chem. Phys.*, 1978, **82**, 1080–1083.
- D. M. Chipman, *J. Phys. Chem.*, 1979, **83**, 1657–1662.
- C. Møller and M. Plesset, *Phys. Rev.*, 1934, **46**, 618–622.
- H. M. Lee, S. Lee and K. S. Kim, *J. Chem. Phys.*, 2003, **119**, 187–194.
- K. D. Jordan and F. Wang, *Annu. Rev. Phys. Chem.*, 2003, **54**, 367–396.
- P. J. Stephens, F. J. Devlin, C. F. Chabalowski and M. J. Frisch, *J. Phys. Chem.*, 1994, **98**, 11623–11627.
- K. S. Kim, S. Lee, J. Kim and J. Y. Lee, *J. Am. Chem. Soc.*, 1997, **119**, 9329–9330.
- H. M. Lee and K. S. Kim, *J. Chem. Phys.*, 2002, **117**, 706–708.
- H. M. Lee, S. Bum Suh and K. S. Kim, *J. Chem. Phys.*, 2003, **118**, 9981–9986.
- H. M. Lee, S. B. Suh and K. S. Kim, *J. Chem. Phys.*, 2003, **119**, 7685–7692.
- H. M. Lee, S. B. Suh, P. Tarakeshwar and K. S. Kim, *J. Chem. Phys.*, 2005, **122**, 044309.
- J. M. Herbert and M. Head-Gordon, *J. Am. Chem. Soc.*, 2006, **128**, 13932–13939.
- J. M. Herbert and M. Head-Gordon, *Proc. Natl. Acad. Sci. U. S. A.*, 2006, **103**, 14282–14287.
- J. R. Roscioli and M. A. Johnson, *J. Chem. Phys.*, 2007, **126**, 024307.
- J. Kim, S. B. Suh and K. S. Kim, *J. Chem. Phys.*, 1999, **111**, 10077–10087.
- G. D. Purvis III and R. J. Bartlett, *J. Chem. Phys.*, 1982, **76**, 1910–1918.
- J. Kim, J. Y. Lee, K. S. Oh, J. M. Park, S. Lee and K. S. Kim, *Phys. Rev. A*, 1999, **59**, R930–R933.
- J. M. Herbert and M. Head-Gordon, *J. Phys. Chem. A*, 2005, **109**, 5217–5229.
- K. Yagi, Y. Okano, T. Sato, Y. Kawashima, T. Tsuneda and K. Hirao, *J. Phys. Chem. A*, 2008, **112**, 9845–9853.
- V. P. Vysotskiy, L. S. Cederbaum, T. Sommerfeld, V. K. Vooora and K. D. Jordan, *J. Chem. Theory Comput.*, 2012, **8**, 893–900.
- W. Koch and M. C. Holthausen, *A Chemist's Guide to Density Functional Theory*, Wiley-VCH Verlag GmbH, Weinheim New York, 2nd edn, 2001.
- P. Skurski, M. Gutowski and J. Simons, *Int. J. Quantum Chem.*, 2000, **80**, 1024–1038.
- T. Sommerfeld, *J. Chem. Phys.*, 2007, **126**, 027101.
- R. Krishnan, J. S. Binkley, R. Seeger and J. A. Pople, *J. Chem. Phys.*, 1980, **72**, 650–654.
- T. Clark, J. Chandrasekhar, G. W. Spitznagel and P. V. R. Schleyer, *J. Comp. Chem.*, 1983, **4**, 294–301.
- T. H. Dunning Jr., *J. Chem. Phys.*, 1989, **90**, 1007–1023.
- R. A. Kendall, T. H. Dunning Jr. and R. J. Harrison, *J. Chem. Phys.*, 1992, **96**, 6796–6806.
- E. R. Davidson, *Chem. Phys. Lett.*, 1996, **260**, 514–518.
- A. J. Sadlej, *Collect. Czech. Chem. Commun.*, 1988, **53**, 1995–2016.
- M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, *Gaussian 09 Revision B. 01*, Gaussian Inc. Wallingford CT 2009.
- S. Obara and A. Saika, *J. Chem. Phys.*, 1986, **84**, 3963–3974.
- M. Head-Gordon and J. A. Pople, *J. Chem. Phys.*, 1988, **89**, 5777–5786.
- T. P. Hamilton and H. F. Schaefer III, *Chem. Phys.*, 1991, **150**, 163–171.
- J. Gerratt and I. M. Mills, *J. Chem. Phys.*, 1968, **49**, 1719–1729.
- J. A. Pople, R. Krishnan, H. B. Schlegel and J. S. Binkley, *Int. J. Quantum Chem.*, 1979, **16**, 225–241.

- 66 B. Fornberg, *Math. Comp.*, 1988, **51**, 699–760.
- 67 W. T. King, G. B. Mast and P. P. Blanchette, *J. Chem. Phys.*, 1972, **56**, 4440–4446.
- 68 J. Cioslowski, *J. Am. Chem. Soc.*, 1989, **111**, 8333–8336.
- 69 A. Milani and C. Castiglioni, *J. Phys. Chem. A*, 2010, **114**, 624–632.
- 70 R. S. Mulliken, *J. Chem. Phys.*, 1955, **23**, 1833–1840.
- 71 Further differentiation of Eq. 2 with respect to external electric field  $F$  reads  
$$\frac{\partial \alpha}{\partial Q_k} = -\text{Tr} \left( \frac{\partial^2 P}{\partial F \partial Q_k} X + \frac{\partial P}{\partial F} \frac{\partial X}{\partial Q_k} \right).$$
  
Thus, the dynamic Raman polarizability could be easily decomposed into different atomic basis set.
- 72 M. D. Halls and H. B. Schlegel, *J. Chem. Phys.*, 1998, **109**, 10587–10593.
- 73 M. D. Halls and H. B. Schlegel, *J. Chem. Phys.*, 1999, **111**, 8819–8824.
- 74 F. Jensen, *J. Chem. Theory Comput.*, 2010, **6**, 2726–2735.
- 75 M.-C. Kim, E. Sim and K. Burke, *J. Chem. Phys.*, 2011, **134**, 171103.
- 76 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865–3868.
- 77 A. D. Becke, *Phys. Rev. A*, 1988, **38**, 3098–3100.
- 78 J. P. Perdew and A. Zunger, *Phys. Rev. B*, 1981, **23**, 5048–5079.
- 79 H. Iikura, T. Tsuneda, T. Yanai and K. Hirao, *J. Chem. Phys.*, 2001, **115**, 3540–3544.
- 80 Y. Zhao and D. Truhlar, *Theor. Chem. Acc.*, 2008, **120**, 215–241.
- 81 T. Yanai, D. P. Tew and N. C. Handy, *Chem. Phys. Lett.*, 2004, **393**, 51–57.
- 82 J.-D. Chai and M. Head-Gordon, *Phys. Chem. Chem. Phys.*, 2008, **10**, 6615–6620.
- 83 S. Grimme, *J. Chem. Phys.*, 2006, **124**, 034108.
- 84 T. Sommerfeld, S. D. Gardner, A. DeFusco and K. D. Jordan, *J. Chem. Phys.*, 2006, **125**, 174301.