

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

Target-selective photo-degradation of AFP-L3 and selective photo-cytotoxicity against HuH-7 hepatocarcinoma cells using an anthraquinone-PhoSL hybrid

Mai Okuyama,[†] Haruna Ueno,[†] Yuka Kobayashi,^{††} Hirokazu Kawagishi,^{†††} Daisuke Takahashi,[†] and Kazunobu Toshima^{†*}

Received (in XXX, XXX) Xth XXXXXXXXX 20XX, Accepted Xth XXXXXXXXX 20XX

DOI: 10.1039/b000000x

A purpose-designed anthraquinone-*Pholiota squarrosa* lectin (PhoSL) hybrid effectively degraded α -fetoprotein-L3 (AFP-L3) associated with liver cancer. Degradation was achieved using light irradiation in the absence of any additives and under neutral pH conditions. Moreover, the hybrid effectively exhibited selective photo-cytotoxicity against HuH-7 hepatocarcinoma cells upon photo-irradiation.

Hepatocellular carcinoma (HCC) is one of the most common causes of cancer-related mortality worldwide and represents more than 90% of primary liver cancers.¹ HCC is diagnosed in over half a million people worldwide each year, is the fifth most common cancer globally and the third most common cause of cancer-related death (after lung and gastric cancer), and thus represents a major worldwide health problem.

Levels of α 1-6 fucosylated α -fetoprotein (AFP-L3) are elevated in HCC patients. Core Fuc α 1-6 oligosaccharide, shown in Fig. 1, has several biological functions, including as a biomarker for HCC.² Therefore, an agent able to mediate the selective degradation of AFP-L3 by specific recognition of the core Fuc α 1-6 oligosaccharide would be a very attractive and one of the useful candidates as an HCC anti-cancer drug. In this context, Kobayashi and Kawagishi *et al.* recently isolated a novel Fuc α 1-6-specific lectin from the mushroom *Pholiota squarrosa*. The purified lectin, designated as PhoSL (*P. squarrosa* lectin), consists of 40 amino acids (NH₂-APVPVTKLVCDGDTYKCTAYLDFGDGRWVAQWDTNVFHTG-COOH).³ PhoSL is stable at a range of pH values and temperatures and is robust during purification and assay. We previously reported that certain anthraquinone derivatives degrade proteins by emitting reactive oxygen species under photo-irradiation conditions⁴ and we thus hypothesized that an anthraquinone hybrid incorporating PhoSL would effectively and selectively degrade AFP-L3 under

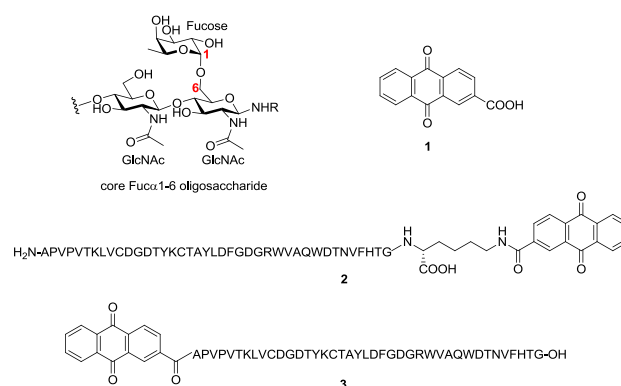


Figure 1. The chemical structures of core Fuc α 1-6 oligosaccharide, anthraquinone derivative **1**, and anthraquinone-PhoSL hybrids **2** and **3**.

photo-irradiation conditions, thus selectively exhibiting photo-cytotoxicity against HCC cells overexpressing AFP-L3. We here present the molecular design, chemical synthesis and evaluation of a novel light-activatable anthraquinone-lectin (PhoSL) hybrid that effectively degrades AFP-L3 and exhibits selective cytotoxicity against HCC cells under photo-irradiation.

We utilized anthraquinone derivative **1** as a protein photodegrading agent (Fig. 1). The structure of the sugar binding pocket of PhoSL remains unclear; therefore, **2** and **3** were tentatively designed as anthraquinone-PhoSL hybrids. Hybrid **2** comprises anthraquinone **1** linked at the C-terminal of PhoSL through a lysine spacer between **1** and PhoSL, while hybrid **3** consists of anthraquinone **1** attached to the N-terminal of PhoSL. Designed hybrids **2** and **3** were synthesized using standard solid-phase synthesis (see SI) and their binding ability and specificity for the core Fuc α 1-6 oligosaccharide were examined by capillary affinity electrophoresis using fluorescently (8-aminopyrene-1,3,6-trisulfonic acid, APTS) labelled oligosaccharides **4-8**.⁵ PhoSL was used as a positive control for the assay; as shown in Fig. 2-a, only the peak corresponding to the oligosaccharide **8**, which contains the core α 1-6 fucoside moiety, shifted to a longer retention time in a dose dependent manner, confirming that PhoSL bound to **8** specifically and exhibited no significant affinity for oligosaccharides **4-7** which lack the core α 1-6 fucoside structure. Furthermore, hybrid **2** also bound only to **8**,

[†] Department of Applied Chemistry, Faculty of Science and Technology, Keio University, 3-14-1 Hiyoshi, Kohoku-ku, Yokohama 223-8522, Japan. E-mail: toshima@aplc.keio.ac.jp; Fax: +81 45-566-1576.

^{††} J-Oil Mills, Inc. Yokohama 245-0064, Japan.

^{†††} Research Institute of Green Science and Technology, Shizuoka University, Shizuoka 422-8529, Japan.

^{††††} Electronic Supplementary Information (ESI) available: See DOI: 10.1039/b000000x/

similar to PhoSL (Fig. 2-b)), whereas hybrid **3** did not bind oligosaccharides **4-8** (see SI: Fig. S2). Interestingly, the binding ability of **2** to **8** was stronger than that of PhoSL.

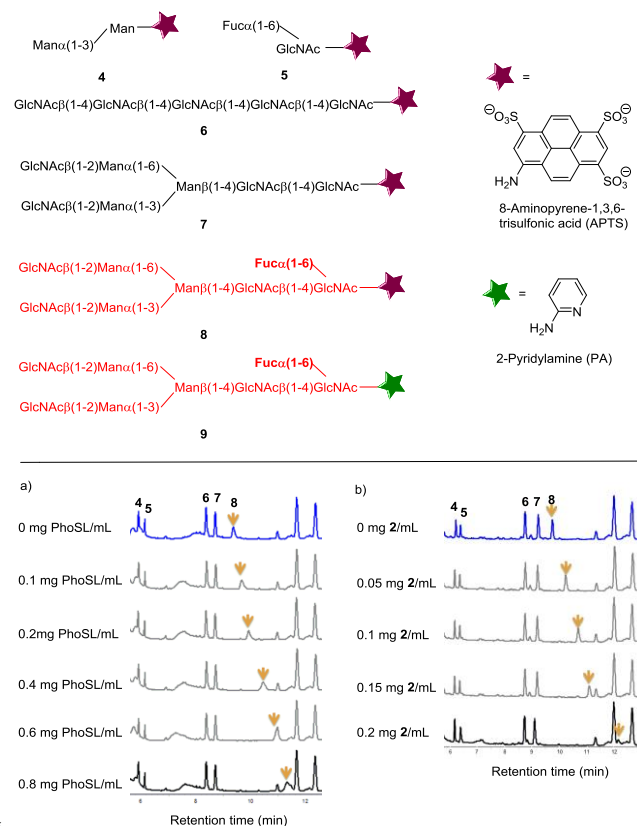


Figure 2. Assay of the interaction of PhoSL and hybrid **2** with the fluorescently (8-aminopyrene-1,3,6-trisulfonic acid, APTS)-labelled oligosaccharides **4-8** by capillary affinity electrophoresis. PhoSL and hybrid **2** were present at the concentrations indicated. a) Capillary electrophoresis profile using PhoSL. b) Capillary electrophoresis profile using hybrid **2**.

With these favorable results for hybrid **2** in hand, we next examined the photo-degradation ability of **2** against the target oligosaccharide **9** which possesses 2-pyridylamine (PA)⁶ at the reducing terminal as a label for fluorescence detection during HPLC analysis. The results of the photo-degradation experiment (365 nm, 100 W), monitored by HPLC, are shown in Fig. 3. No degradation of **9** occurred under these photo-irradiation conditions, even in the presence of a large excess of **2**. However, **2** degraded under photo-irradiation conditions, strongly suggesting that **2** produced reactive oxygen species which can photo-degrade oligosaccharides⁷, proteins,⁴ and itself, but not **9** (Fig. 4-a). This is probably due to the distance between the anthraquinone moiety in **2** and **9** in the sugar-binding pocket in **2** to support the degradation of **9**; however, if the anthraquinone moiety in **2** could be brought sufficiently close to AFP-L3 upon recognition of the core fuc α -1-6 oligosaccharide, then AFP-L3 would be photo-degraded by **2** (Fig. 4-b)).

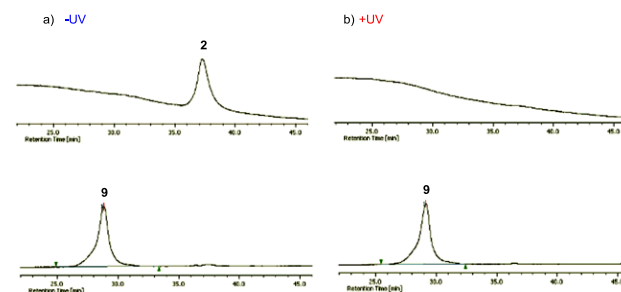


Figure 3. Photo-degradation of PA-labelled oligosaccharide **9** by hybrid **2**. **9** (500 nM) was incubated with **2** (60 μ M) in Tris-acetate buffer (100 mM, pH 7.0)/MeCN (7:3) at 25 $^{\circ}$ C for 2 h during irradiation with a UV lamp (365 nm, 100 W) placed 10 cm from the sample and the reaction mixture was analyzed by HPLC. a) HPLC profiles of **2** and **9** without photo-irradiation. b) HPLC profiles of **2** and **9** with photo-irradiation.

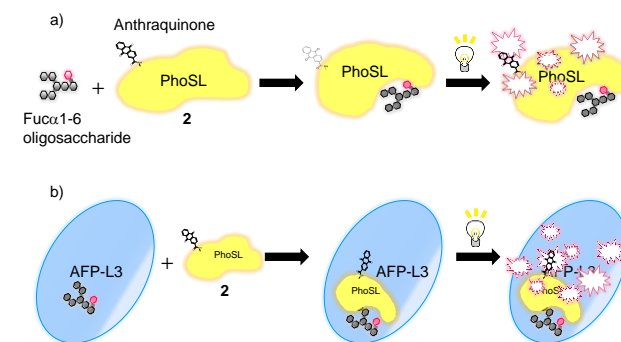


Figure 4. a) Self photo-degradation of **2** without photo-degradation of Fuc α -1-6 oligosaccharide. b) Photo-degradation of AFP-L3 by **2** following recognition of Fuc α -1-6 oligosaccharide by **2**.

We therefore conducted electron paramagnetic resonance (EPR) spin trapping experiments using 5,5-dimethyl-1-pyrroline-*N*-oxide (DMPO) to confirm the generation of reactive oxygen species from the reaction of photo-excited **2** with O₂. It was found that photo-irradiation of **2** in the presence of DMPO gave a product with the EPR spectrum characteristics of the DMPO-hydroxyl radical spin adduct DMPO- \cdot OH (Fig. 5-b), which results from the reaction of DMPO with \cdot OH and/or the decay of the DMPO-superoxide anion spin adduct DMPO- \cdot OOH.^{8,9} In addition, it was confirmed that no peaks corresponding to DMPO- \cdot OOH or DMPO- \cdot OH were detected either by treatment of DMPO with **2** without photo-irradiation or by photo-irradiation of DMPO in the absence of **2** (Fig. 5-a). These results indicate that a reactive oxygen species was certainly produced by the reaction of photo-excited **2** with O₂.

We next examined the degradation ability and selectivity of **2** against AFP-L3 upon photo-irradiation (365 nm, 100 W). AFP-L1 has no surface target oligosaccharide and was used as a negative control glycoprotein. The reaction was monitored by SDS-PAGE and the results are shown in Fig. 6. Comparison of lanes 3 and 4 with lane 2 in Fig. 6-a shows that neither photo-irradiation of AFP-L3 in the absence of **2** or treatment of AFP-L3 with **2** in the absence of photo-irradiation resulted in a change in the SDS-PAGE profile. In contrast, lanes 5-7 show the fading or disappearance of the bands corresponding to AFP-L3, indicating that hybrid **2** caused significant degradation of AFP-L3 upon

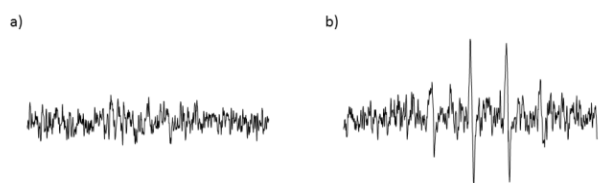


Figure 5. EPR spectrum obtained during photo-irradiation of **2** in the presence of DMPO. **2** (21 μM) and DMPO (500 mM) were incubated in 2% DMSO/PBS (pH 7.4, 137 mM NaCl, 2.7 mM KCl, 6.5 mM Na_2HPO_4 , 1.5 mM KH_2PO_4) containing 1 mM DETAPAC under irradiation with a UV lamp (365 nm, 100 W) placed 40 cm from a flat cell. a) Before irradiation. b) After 10 min irradiation. DETAPAC= diethylenetriaminepentaacetic acid.

photo-irradiation. In addition, it is clear that hybrid **2** exhibited only weak degradation activity towards AFP-L1 when the mixture of AFP-L1 and **2** was irradiated with a long-wavelength UV lamp (Fig. 6-b)). These results clearly demonstrated that interaction between **2**, which possesses the PhoSL moiety, and the target oligosaccharide on AFP-L3, which possesses the core α -1-6 fucoside structure, is very important for the effective and selective degradation of AFP-L3 under photo-irradiation. No SDS-PAGE bands corresponding to degraded peptide fragments or aggregated product were observed after photo-irradiation of **2** and AFP-L3, suggesting that the degradation reaction occurred non-site-specifically and that AFP-L3 was degraded into small peptide fragments too small to observe by SDS-PAGE.^{4,10,11} In addition, because no degradation of AFP-L3 by hybrid **2** occurred in the absence of light, UV light functioned as a trigger to initiate degradation by the hybrid.

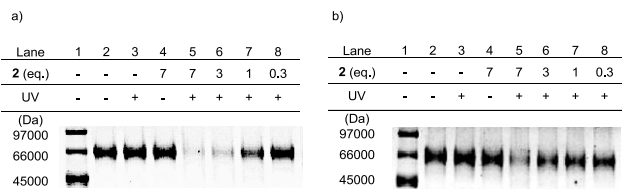


Figure 6. Photo-degradation of a) AFP-L3 and b) AFP-L1. AFP-L3 or AFP-L1 (3.0 μM) was incubated with hybrid **2** in 2% DMSO/PBS (pH 7.4, 137 mM NaCl, 2.7 mM KCl, 6.5 mM Na_2HPO_4 , 1.5 mM KH_2PO_4) at 25 $^\circ\text{C}$ for 2 h during irradiation with a UV lamp (365 nm, 100 W) placed 10 cm from the sample and the reaction mixtures were analyzed by tricine-SDS-PAGE (12% polyacrylamide gel): a) lane 1, size markers; lane 2, AFP-L3 alone; lane 3, AFP-L3 with UV; lane 4, AFP-L3 + **2** (21 μM) without UV; lanes 5-8, the concentrations of **2** were 21, 9, 3.0, and 0.90 μM with UV, respectively. b) lane 1, size markers; lane 2, AFP-L1 alone; lane 3, AFP-L1 with UV; lane 4, AFP-L1 + **2** (21 μM) without UV; lanes 5-8, the concentrations of **2** were 21, 9, 3.0, and 0.90 μM with UV, respectively.

These photo-degradation results were supported by MALDI-TOF MS analysis, as shown in Fig. 7: incubation of AFP-L3 with **2** without photo-irradiation gave the MS peaks corresponding to AFP-L3 and the adduct of AFP-L3 with **2** while incubation of AFP-L3 with **2** under photo-irradiation resulted in disappearance of those MS peaks corresponding to AFP-L3 and the adduct. No MS peaks corresponding to degradation fragments could be detected due to the randomness of the degradation, leading to many fragments present in very small amounts.^{4,10,11}

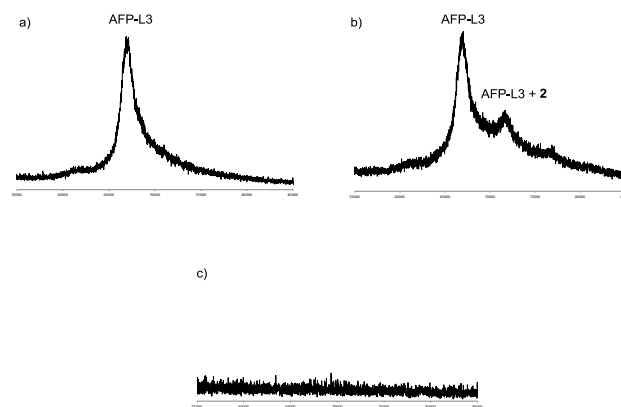


Figure 7. MALDI-TOF MS profile of AFP-L3 obtained following incubation with hybrid **2**. AFP-L3 (1.5 eq.) was incubated without a) or with b) and c) **2** (21 μM) in 2% DMSO/PBS (pH 7.4, 137 mM NaCl, 2.7 mM KCl, 6.5 mM Na_2HPO_4 , 1.5 mM KH_2PO_4) at 25 $^\circ\text{C}$ for 2 h without a) and b) or with c) photo-irradiation with a UV lamp (365 nm, 100 W) placed 10 cm from the mixture. The resulting products were analysed by MALDI-TOF MS (matrix: 3,5-dimethoxy-4-hydroxycinnamic acid). Peaks were observed at m/z 67000 in a), indicating AFP-L3; at m/z 67000 and 72000 in b), indicating AFP-L3 and AFP-L3+2; those peaks were not observed in c).

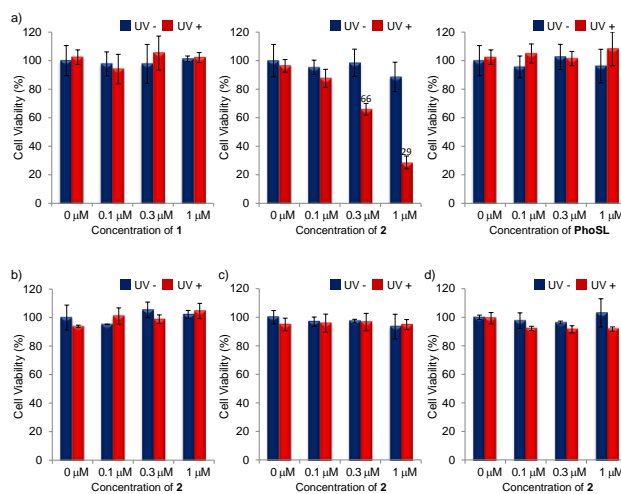


Figure 8. Effects of **1**, **2**, and PhoSL on a) HuH-7 cell, b) A549 cell, c) HeLa cell, or d) HLE cell proliferation. Cells were seeded into 96-well plates (2.00×10^3 cells per well). After 24 h, compounds were added at the indicated concentrations and the cells were incubated for 24 h at 37 $^\circ\text{C}$ in 5% CO_2 in air. Cells were irradiated for 30 min with a UV lamp (365 nm, 30 W) placed 20 cm from the plate. Cells were then kept for 72 h at 37 $^\circ\text{C}$ in 5% CO_2 in air, then MTT reagent was added to each well and the cells were incubated for up to 3 additional hours. The absorbance at 540 nm was read using a plate reader.

Finally, we examined the inhibitory activity of **2** on the cell growth of HuH-7, A549, HeLa, and HLE cells using the MTT assay (Figure 8).^[12] HLE cells are a hepatocellular carcinoma cell line derived from liver cancer and do not display overexpressed AFP-L3 on the surface, whereas HuH-7 cells are a hepatocellular carcinoma cell line derived from liver cancer which overexpress AFP-L3. A549 and HeLa cells are a human adenocarcinoma cell line derived from lung cancer and a human epithelial carcinoma cell line derived from cervical cancer, respectively, both of which do not display overexpressed AFP-L3 on the surfaces. As

expected, hybrid **2** significantly inhibited the growth of HuH-7 cells upon light irradiation, whereas no inhibition was observed when the anthraquinone derivative **1** without PhoSL, or PhoSL alone, was used instead of **2** (Figure 8-a)). In addition, **2** did not show any cytotoxicity against HuH-7 cells without photo-irradiation. Furthermore, **2** exhibited no inhibitory activity towards the growth of HLE, A549, and HeLa cells, even with photo-irradiation (Figure 8-b-d)). These results clearly indicate that the anthraquinone-PhoSL hybrid **2** exhibits effective and selective photo-cytotoxicity against HuH-7 cells which display overexpressed AFP-L3 containing the core fuc α 1-6 oligosaccharide. In addition, although we have not yet obtained direct evidence of photo-degradation of the target glycoprotein, AFP-L3, on HuH-7 cells, these results strongly suggest that selective photo-degradation of the target glycoprotein on the cancer cells by hybrid **2** took place and induced significant photo-cytotoxicity

In conclusion, anthraquinone-PhoSL hybrid **2** selectively and effectively degraded AFP-L3 upon irradiation with long-wavelength UV light in the absence of any additives and under neutral pH conditions. Furthermore, **2** showed significant and selective inhibitory activity toward HuH-7 cell growth upon photo-irradiation. These results showed that selective photo-degradation of a target protein can be realized by specific recognition of a surface oligosaccharide and a light activatable molecule, anthraquinone, and that the hybrid molecule can inhibit select cancer cell growth. The results presented here will contribute to the molecular design of novel glycoprotein photo-degrading agents by the selective recognition of specific oligosaccharide structures, thus providing agents for controlling the functions of proteins involved in diseases.

This research was supported in part by grants for Scientific Research (B) (No. 23310153 and 26282212) and Scientific Research on Innovative Areas “Chemical Biology of Natural Products” (No. 24102528 and 26102738) from the Ministry of Education, Culture, Sports, Science and Technology of Japan (MEXT).

Notes and references

- 1 a) H. B. El-Serag, *Gastroenterology* 2012, **142**, 1264; b) R. Lencioni, *Hepatology* 2010, **52**, 762; c) H. Nishikawa, Y. Osaki, *Int. J. Oncol.* 2013, **43**, 1333; d) J. M. Llovet, *J. Gastroenterology* 2005, **40**, 225; e) J. M. Llovet, S. Ricci, V. Mazzaferro, et al., *N. Engl. J. Med.* 2008, **359**, 378; f) H. B. El-Serag, *N. Engl. J. Med.* 2011, **365**, 1118.
- 2 H. Tateno, S. Nakamura-Tsuruta, J. Hirabayashi, *Glycobiology* 2009, **19**, 527.
- 3 Y. Kobayashi, H. Tateno, H. Dohra, K. Moriwaki, E. Miyoshi, J. Hirabayashi, H. Kawagishi, *J. Biol. Chem.* 2012, **287**, 33973.
- 4 a) Y. Aoki, S. Tanimoto, D. Takahashi, K. Toshima, *Chem. Commun.* 2013, **49**, 1169; b) A. Suzuki, M. Hasegawa, M. Ishii, S. Matsumura, K. Toshima, *Bioorg. Med. Chem. Lett.* 2005, **15**, 4624; c) K. Toshima, *Mol. Biosyst.* 2013, **9**, 834; d) S. Tanimoto, D. Takahashi, K. Toshima, *Chem. Commun.* 2012, **48**, 7659.
- 5 K. Nakajima, M. Kinoshita, Y. Oda, T. Masuko, H. Kaku, N. Shibuya, K. Kakehi, *Glycobiology* 2004, **14**, 793.
- 6 H. Hase, T. Ikenaka, Y. Matsushima, *Biochem. Biophys. Res. Commun.* 1978, **85**, 257.
- 7 a) D. Takahashi, T. Miura, K. Toshima, *Chem. Commun.* 2012, **48**, 7595; b) Y. Imai, S. Hirono, H. Matsuba, T. Suzuki, Y. Kobayashi, H. Kawagishi, D. Takahashi, K. Toshima, *Chem. Asian J.* 2012, **7**, 97; c) D. Takahashi, S. Hirono, C. Hayashi, M. Igarashi, Y.

- Nishimura, K. Toshima, *Angew. Chem. Int. Ed.* 2010, **49**, 10096; d) M. Ishii, S. Matsumura, K. Toshima, *Angew. Chem. Int. Ed.* 2007, **46**, 8396; e) D. Takahashi, K. Toshima, *Chem. Commun.* 2012, **48**, 4397.
- 8 E. Wertz, B. Bolton in *Electron Spin Resonance*, McGraw-Hill, New York, 1972.
- 9 H. M. Swartz, J. R. Bolton, D. C. Borg in *Biological Application of Electron Spin Resonance*, Wiley-Interscience, New York, 1972.
- 10 a) G. B. Jones, J. M. Wright, G. Hynd, J. K. Wyatt, M. Yancisin, M. A. Brown, *Org. Lett.* 2000, **2**, 1863; b) T. Furuta, M. Sakai, H. Hayashi, T. Asakawa, F. Kataoka, S. Fujii, T. Suzuki, Y. Suzuki, K. Tanaka, N. Fishkin, K. Nskanishi, *Chem. Commun.* 2005, **41**, 4575.
- 11 a) R. T. Dean, S. Fu, R. Stocker, M. J. Davies, *Biochem. J.* 1997, **324**, 1; b) M. J. Davies, *Biochem. Biophys. Res. Commun.* 2003, **305**, 761; c) M. Gracanin, C. L. Hawkins, D. I. Pattison, M. J. Davies, *Free Rad. Bio. Med.* 2009, **47**, 92.
- 12 T. Mosmann, *J. Immunol. Methods* 1983, **65**, 55.