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# A new ratiometric fluorescent probe for rapid, sensitive and selective detection of endogenous hydrogen sulfide in mitochondria

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We have developed a new ratiometric fluorescent probe composed coumarin-merocyanine dyad based on FRET mechanism. The probe showed clear dual-emission signal changes in blue and red spectral windows upon H<sub>2</sub>S in a dose dependent manner under a single wavelength excitation. The probe targeted to mitochondria with high selectivity and sensitivity toward H<sub>2</sub>S.

Hydrogen sulfide (H<sub>2</sub>S) is an important endogenous gasotransmitter which is associated with diverse physiological processes, such as vasodilation,<sup>1</sup> angiogenesis,<sup>2</sup> apoptosis,<sup>3</sup> inflammation,<sup>4</sup> and neuromodulation.<sup>5</sup> It has been reported that the physiological concentration of H<sub>2</sub>S in mammalian serum is in the 30–100  $\mu$ M range, whereas those in the brain can be as high as 160  $\mu$ M.<sup>6</sup> However, abnormal H<sub>2</sub>S levels are implicated in many diseases such as Alzheimer's disease,<sup>7</sup> Down's syndrome,<sup>8</sup> diabetes,<sup>9</sup> and liver cirrhosis.<sup>10</sup>

Heightened research interest into the mechanism of the physiological and pathological functions of H<sub>2</sub>S has led to the development of fluorescent probes that can easily and reliably detect and quantify H<sub>2</sub>S in a living biological system.<sup>11</sup> Most of them were designed based on the strong nucleophilicity<sup>12</sup> and reduction capability of  $H_2S_1^{13}$  as well as  $H_2S$ -mediated metal sulfide precipitation from fluorophore-ligated metals.<sup>14</sup> Most of the probes based on the reduction of aromatic azide by H<sub>2</sub>S showed a delayed response. The probes based on nucleophilic substitution or addition reaction suffered the possible interference from biological thiols. Therefore, a challenging issue in developing fluorescent probes for H<sub>2</sub>S is how to secure selectivity over competing biothiols. Moreover, most of the probes based on fluorescent intensity may be influenced by many environmental effects.<sup>15</sup> However, ratiometric fluorescent probes, in contrast, can eliminate these

shortcomings by self-calibration of two emission bands, often corresponding to the unreacted probe and its subsequent reaction product.<sup>16a</sup>

A number of ratiometric fluorescent probes were developed based on Förster resonance energy transfer (FRET). FRET is the interaction between two fluorophores linked by a nonconjugated spacer in the same molecule. The fluorophore with shorter wavelength emission as the energy donor may collect radiation at excitation wavelength and transfer this energy to the longer wavelength-emitting fluorophore (the acceptor). FRET-based sensing system can rule out any influence of excitation backscattering effects on fluorescence detection because of the large shift between donor excitation and acceptor emission. In addition, two well-separated emission bands with comparable intensities can ensure accuracy in determining their intensities and ratios. However, developing FRET cassettes is still limited by the requirement that the donor emission must have more overlap with the acceptor absorption. Therefore, consideration must be given to both spectral overlap and the emission shifts of FRET systems.<sup>12b</sup> In addition, other factors should also be considered in the design of probes for biological systems, such as high photo-stability, high quantum yield, good solubility in aqueous and lipid environments, good cell permeability, and low interference from biological environments. Xanthene dyes are usually used as acceptor in the FRET dyad.<sup>16b</sup> A spectral overlap between the donor emission and the acceptor absorption can be controlled by ring-opening or ring-closing of xanthene.<sup>16c</sup> To date, a lot of xanthene-based FRET sensors have been reported for cations, anions and biomolecules.<sup>17</sup> Besides, the FRET process can be controlled by the cleavage of an appropriate linker.<sup>18</sup> Recently, the coumarin-naphthalimide dyad has been developed as a FRET platform because of the coumarin emission well overlaps with the naphthalimide absorption.<sup>19</sup> However, the less emission shifts of these FRET systems seem not so satisfactory. Yuan et al reported a new FRET system with two-photon excitation in which anthocyanidin-analogue dye was an acceptor. The probe showed distinguished improvements in ratiometric response

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### COMMUNICATION

### Journal Name

with a large emission shift and comparative fluorescence intensity.  $^{\rm 12b}$ 



Though numerous examples of fluorescence-sensing strategies in the detection of H<sub>2</sub>S are reported, FRET system remains underused for sensing H<sub>2</sub>S.<sup>12b,19b,c</sup> Therefore, developing advanced probes with high sensitivity, selectivity and fastresponse for real time visualization of low concentration H<sub>2</sub>S in living biological samples is still needed. Especially, developing a new strategy for ratiometric fluorescent probe remains challenging. In this work, we have developed a new strategy to construct a ratiometric fluorescent probe based on the FRET process. The ratio emission signal of the coumarinmerocyanine dyad is controlled through the nucleophilic addition of H<sub>2</sub>S to the acceptor (Scheme 1). The probe (CPC) exhibited a large emission shift compared with most reported probes for  $H_2S$ . Prior to reaction with  $H_2S$ , the probe has a significant overlap of the coumarin emission with the merocyanine absorption and the FRET process occurs, and the dyad displays merocyanine fluorescence. However, interaction of merocyanine moiety with H<sub>2</sub>S affords nucleophilic addition product CPC-S, which is not an acceptor fluorophore due to the destruction of conjugate system, and the FRET process does not exist. The dyad displays coumarin emission. Thus, the dyad CPC might detect H<sub>2</sub>S as a ratiometric fluorescent probe based on the new FRET strategy. Probe CPC was synthesized by convenient coupling reactions (Scheme S1, ESI) and fully characterized (see ESI).

We first examined the emission spectra of probe CPC in PBS buffer (0.01 M, pH 7.4). As designed, upon excitation at 410 nm, the free probe exhibits weak coumarin fluorescence at 474 nm and strong merocyanine fluorescence at 587 nm. The distinct gap between the two bands is over 113 nm, which makes this probe favorable for the dual emission ratiometric imaging. These results are consistent with our design that the coumarin fluorescence in CPC is quenched through the FRET process. However, upon addition of Na<sub>2</sub>S (a standard source for hydrogen sulfide), a significant merocyanine acceptor emission peak at around 587 nm disappeared and simultaneously the emission peak of the coumarin donor obviously enhanced (Fig. S1, ESI), suggesting that conjugate system of acceptor in CPC was destroyed to afford an addition product CPC-S in the presence of Na2S and FRET from the coumarin unit to the merocyanine was effectively blocked.

To clarify the new FRET mechanism, we synthesized the **donor** and the **acceptor**, respectively. Upon addition of Na<sub>2</sub>S, the fluorescence intensity of the **Donor** in the same buffer solution did not change, however, that of the **Acceptor** decreased dramatically (Fig. S2 S3, ESI). It suggested that the merocyanine moiety of probe **CPC** interacted with Na<sub>2</sub>S while

the coumarin moiety did not. Notably, the coumarin emission with the merocyanine absorption has a significant overlap (Fig. S4, ESI). Thus the FRET process occurs effectively. When CPC was excited at maximum absorption wavelength of the coumarin, the typical coumarin emission at about 474 nm was nearly completely quenched compared with that of the reference Donor. Meantime, the fluorescence intensity of the merocyanine moiety at around 587 nm appeared relative to that of the reference Acceptor at the same concentration (Fig. S5, ESI). This suggested that the excitation energy of the coumarin donor was efficiently transferred to the merocyanine acceptor. The intramolecular energy transfer efficiency from the donor to the acceptor in CPC was calculated to be 98.35%. The efficient intramolecular energy transfer was collaborated by the excellent correspondence between the absorption and excitation spectra of the probe (Fig. S6, ESI).

The specific nature of probe **CPC** for  $H_2S$  was also evaluated by UV-vis absorption in the buffer solution. Probe **CPC** exhibited two major absorption band centered at 418 and 518 nm, which were assigned to the donor and the acceptor, respectively. Upon addition of different representative interfering species, no clear change in the absorption maximum was observed. However, upon addition of  $H_2S$ , the absorption at around 518 nm disappeared while the absorption at around 418 nm remains almost no change (Fig. S7, ESI). The changes in the absorption spectra were in good agreement with the variations in the emission profile. These results indicated that probe **CPC** had clear selectivity in absorption toward  $H_2S$  than other tested analytes.

To examine the selectivity, probe CPC (5  $\mu$ M) was treated with various biologically relevant species including the representative anions, reactive oxygen species, reducing agents, small molecule thiols, and Na<sub>2</sub>S in the aqueous buffer. Addition of these representative interfering species induced no marked ratio increase. However, just addition of 50 µM Na<sub>2</sub>S can elicit remarkable increase in the ratio value  $(I_{474}/I_{587})$  (Fig. S8, ESI). The reaction was completed within 7 min and the ratio value peaked. Thus, probe CPC could fast sense H<sub>2</sub>S in a real-time in vivo imaging (Fig. S9, ESI). In addition, the CPC ratiometric sensing ability for H<sub>2</sub>S in the presence of other biological thiols was also studied (Fig. S10, ESI). These results suggested that probe CPC has a high selectivity toward H<sub>2</sub>S over other biological species and is able to detect H<sub>2</sub>S without any distinct interference from these biological thiols. Moreover, an increase in fluorescence signal ratios was observed at physiological pH (Fig. S11, ESI).

To obtain further insight into the sensitivity of probe **CPC** for H<sub>2</sub>S, the fluorescence intensity change was closely monitored upon the addition of H<sub>2</sub>S (0–11 equiv.) into the probe (Fig. 1). Fluorescence titration demonstrated that its emission band at 587 nm decreased distinctly, meanwhile the band at 474 nm increased obviously. The intensity ratio of the two emission bands (I<sub>474</sub>/I<sub>587</sub>) increased from 0.73 to 40.93, and the final enhancement factor is about 56-fold. The fluorescence intensity remained changeless after the concentration of H<sub>2</sub>S was up to 55  $\mu$ M (Fig. S12, ESI). The intensity ratio was linearly related to the concentration of H<sub>2</sub>S and the detection limit was

### Journal Name

calculated to be as low as 40 nM (Fig. S13, ESI), which is superior to some reported probes suffering from the limitation of poor detection limit.<sup>12*h*-*k*, 13*b*, 13*e*, 18*b*, <sup>20</sup> These results suggest that probe **CPC** is highly sensitive to H<sub>2</sub>S and is potentially useful for quantitative determination of H<sub>2</sub>S. Moreover, upon addition of Na<sub>2</sub>S, the absorption peak at 518 nm decreased gradually and another peak located at 418 nm changed little (Fig. S14, ESI). The colour changed from red to light yellow (Insert of Fig. S14, ESI).</sup>

Although our attempt to separate HS<sup>-</sup> nucleophilic addition product **CPC-S** was failed, the Thermo LCQ Fleet mass spectrum of **CPC** solution in the presence of excess Na<sub>2</sub>S showed one major signal with m/z of 623.15 and another minor peak with m/z of 645.98, which can be assigned as [CPC-S + H]<sup>+</sup> and [CPC-S + Na]<sup>+</sup>, respectively (Fig. S15, ESI).

The desirable fluorescence properties of probe CPC for  $H_2S$  prompted us to utilize it for the detection of intracellular  $H_2S$ . HeLa cells stained by probe CPC (3  $\mu$ M) alone showed very dim fluorescence in the blue channel and a strong fluorescence in the red channel. Incubation with  $Na_2S$  (0.5, 1 and 5 mM) for 1 h led to a distinct increase in blue channel, accompanied by the dramatic drop of red channel fluorescence, and a drastic enhancement of emission ratio (blue/red) in a dose dependent manner can be observed (Fig. 2).

Encouraged by these promising results, we evaluated the feasibility of probe CPC for detecting endogenous H<sub>2</sub>S. It is known that cysteine could be catalyzed by cystathionine  $\beta$ synthase (CBS) and cystathionineg  $\gamma$ -lyase (CSE) in living cells for  $H_2S$  production.<sup>19b</sup> As shown in Fig. 3, without addition of cysteine, strong fluorescence in the red channel and faint fluorescence in the blue channel were observed. However, after the cells were incubated with cysteine, a fluorescent emission in the red channel decreased and a significant emission in the blue channel increased. The ratios of emission intensity (blue/red) markedly increased in the presence of cysteine. The inhibition assay was also carried out by inhibiting CBS and CSE.<sup>12e,21</sup> When the cells were incubated in a sequence with propargylglycine (PAG) or aminooxyacetic acid (AOAA) or PAG/AOAA and Cys, followed by incubation with probe CPC, almost no fluorescence change was observed compared to the control group. Therefore, probe CPC is capable of detecting extraneous and endogenous H<sub>2</sub>S in living cells.



Fig. 1 Fluorescence spectra of the probe (5  $\mu$ M) with the addition of Na<sub>2</sub>S (0, 0.5, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 equiv.) inPBS buffer (pH 7.4, 0.01 M) : DMF (6: 4, v/v). ( $\lambda_{ex}$  = 410 nm , slit: 12 nm/10 nm). Insert: a visual fluorescence change photograph under illumination with a 365 nm UV lamp.





Fig. 2 (a) Fluorescence and bright field images and (b) the ratio of fluorescence intensity (blue/red) of HeLa cells incubated with probe CPC (3  $\mu$ M) for 1 h and then with Na<sub>2</sub>S (0.5, 1 and 5 mM) for 1 h. The ratio images were all obtained as F<sub>blue</sub>/F<sub>red</sub>. Excitation: 405 nm; emission channels: 405-560 nm (blue) and 560-700 nm (red). \*\* *P* < 0.01; (n = 3). Scale bar = 20  $\mu$ m.



Fig. 3 (a) First line: cells incubated with CPC (1  $\mu$ M) for 1 h; second line: cells incubated with Cys (200  $\mu$ M) for 1 h followed by CPC (1  $\mu$ M) for 1 h; third to fifth line: cells incubated with PAG (1 mM), AOAA (1 mM) and both PAG and AOAA for 1 h, respectively, followed by incubation with Cys (200  $\mu$ M) for 1 h, then incubated with CPC (1  $\mu$ M) for 1h. (b) From left to right: the relative ratio of fluorescence intensity of line 1, 2, 3, 4 and 5 in (a). The ratio images were all obtained as  $F_{blue}/F_{red}$ . Excitation: 405 nm; emission channels: 405-560 nm (blue) and 560-700 nm (red). \*\* P < 0.01; (n = 3). Scale bar = 20  $\mu$ m.



Fig. 4 Colocalization of the probe in HeLa cells. (a-e) probe CPC (1  $\mu$ M, 1 h) and Mito Tracker Deep Red (0.3  $\mu$ M, 0.5 h); (f-j) cells incubated with Na\_{2}S (1 mM) for 1 h, followed by incubation with CPC (1  $\mu$ M) for 1 h, then incubated with Tracker Deep Red (0.3  $\mu$ M) for 0.5 h. (a, f) The cell image of CPC,  $\lambda$ ex = 405 nm,  $\lambda$ em = 405-560 nm. (b, g) The cell image of Mito Tracker Deep Red,  $\lambda$ ex = 635 nm,  $\lambda$ em = 635-700 nm. (c, h) Bright field image. (d, i) Merged image. (e, j) Quantitation of co-localization coefficients (Pearson's coefficient): 0.924 and 0.891.

Furthermore, a colocalization assay with Mito Tracker Deep Red and probe **CPC** displayed that fluorescence of probe **CPC** was colocalized with that of Mito Tracker Deep Red, implying a preferential distribution of probe **CPC** in mitochondria. In addition, the probe could rest on mitochondria after sensing H<sub>2</sub>S (Fig. 4). Moreover, photostability of the probe was evaluated by the fluorescence images of HeLa cells exposed for different time. The ratio of blue/red fluorescence intensity indicated that the probe possesses good photostability during 720 S (Fig. S16, ESI). The SRB assay showed that probe **CPC** had little cytotoxicity in the 1-5  $\mu$ M range for living cells (Fig. S17, ESI). This preliminary imaging study suggested that probe **CPC** can be used for ratiometric tracking of H<sub>2</sub>S in mitochondria, which is of great importance to clarify the physiological roles of H<sub>2</sub>S in the organelle.

In summary, a new ratiometric fluorescent probe composed coumarin–merocyanine dyad based on FRET mechanism was rationally designed and synthesized. The probe showed clear dual-emission signal changes in blue and red spectral windows upon H<sub>2</sub>S in a dose dependent manner. The probe revealed high selectivity and sensitivity toward H<sub>2</sub>S. The probe exhibited the acceptable biocompatibility and targeted to mitochondria. Therefore, probe **CPC** can be used for ratiometric tracking of extraneous and endogenous H<sub>2</sub>S in mitochondria and might be a promising tool for the biological research of H<sub>2</sub>S.

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## Notes and references

- 1 G. Yang, L. Wu, B. Jiang, W. Yang, J. Qi, K. Cao, Q. Meng, A.K. Mustafa, W. Mu, S. Zhang, S.H. Snyder and R. Wang, *Science*, 2008, **322**, 587.
- A. Papapetropoulos, A. Pyriochou, Z. Altaany, G. Yang, A. Marazioti, Z. Zhou, M.G. Jeschke, L.K. Branski, D.N. Herndon, R. Wang and C. Szabó, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, 106, 21972.
- 3 G. Yang, L. Wu and R. Wang, FASEB J., 2006, 20, 553.
- 4 R.C. Zanardo, V. Brancaleone, E. Distrutti, S. Fiorucci, G. Cirino and J.L. Wallace, *FASEB J.*, 2006, **20**, 2118.
- 5 K. Abe and H. Kimura, *Neurosci.*, 1996, **16**, 1066.
- 6 J.L. Wallace, Trends Pharmacol. Sci., 2007, 28, 501.
- 7 K. Eto, T. Asada, K. Arima, T. Makifuchi and H. Kimura, Biochem. Biophys. Res. Commun., 2002, **293**, 1485.
- 8 P. Kamoun, M.C. Belardinelli, A. Chabli, K. Lallouchi and B. Chadefaux-Vekemans, *Am. J. Med. Genet.*, 2003, **116A**, 310.
- 9 W. Yang, G. Yang, X. Jia, L. Wu and R. Wang, J. Physiol., 2005, 569, 519.
- S. Fiorucci, E. Antonelli, A. Mencarelli, S. Orlandi, B. Renga, G. Rizzo, E. Distrutti, V. Shah and A. Morelli, *Hepatology*, 2005, 42, 539.
- (a) F. Yu, X. Han and L. Chen, *Chem. Commun.*, 2014, **50**, 12234; (b) A.R. Lippert, *J. Inorg. Biochem.*, 2014, **133**, 136; (c) K. Shimamoto and K. Hanaoka, *Nitric Oxide*, 2015, **46**, 72; (d) V.S. Lin, W. Chen, M. Xian and C.J. Chang, *Chem. Soc. Rev.*, 2015, **44**, 4596.
- (a) Z. Dai, L. Tian, B. Song, Z. Ye, X. Liu and J. Yuan, Anal. Chem., 2014, 86, 11883; (b) L. Yuan, F. Jin, Z. Zeng, C. Liu, S. Luo and J. Wu, Chem. Sci., 2015, 6, 2360; (c) C. Liu, W. Chen, W. Shi, B. Peng, Y. Zhao, H. Ma and M. Xian, J. Am. Chem. Soc., 2014, 136, 7257; (d) B. Peng, C. Zhang, E. Marutani, A. Pacheco, W. Chen, F. Ichinose and M. Xian, Org. Lett., 2015,

Page 4 of 4

17, 1541; (e) S. Singha, D. Kim, H. Moon, T. Wang, K.H. Kim,
Y.H. Shin, J. Jung, E. Seo, S.J. Lee and K.H. Ahn, Anal. Chem.,
2015, 87, 1188; (f) T. Liu, Z. Xu, D.R. Spring and J. Cui, Org.
Lett., 2013, 15, 2310; (g) K. Zheng, W. Lin, L. Tan and D.
Cheng, Anal. Chim. Acta, 2015, 853, 548; (h) Y. Chen, C. Zhu,
Z. Yang, J. Chen, Y. He, Y. Jiao, W. He, L. Qiu, J. Cen and Z.
Guo, Angew. Chem., Int. Ed., 2013, 52, 1688; (i) Z. Huang, S.
Ding, D. Yu, F. Huang and G. Feng, Chem. Commun., 2014, 50,
9185; (j) X. Li, S. Zhang, J. Cao, N. Xie, T. Liu, B. Yang, Q. He
and Y. Hu, Chem. Commun., 2013, 49, 8656; (k) P. Sathyadevi,
Y.J. Chen, S.C. Wu, Y.H. Chen and Y.M. Wang, Biosens.
Bioelectron., 2015, 68, 681;

- (a) S. Yang, Y. Qi, C. Liu, Y. Wang, Y. Zhao, L. Wang, J. Li, W. Tan and R. Yang, Anal. Chem., 2014, **86**, 7508; (b) B. Wang, P. Li, F. Yu, J. Chen, Z. Qu and K. Han, Chem. Commun., 2013, **49**, 5790; (c) Y. Cai, L. Li, Z. Wang, J.Z. Sun, A. Qin and B.Z. Tang, Chem. Commun., 2014, **50**, 8892; (d) J. Zhang and W. Guo, Chem. Commun., 2014, **50**, 4214; (e) L. Wei, Z. Zhu, Y. Li, L. Yi and Z. Xi, Chem. Commun., 2015, **51**, 10463; (f) N. Gupta, S.I. Reja, V. Bhalla, M. Gupta, G. Kaur and M. Kumar, Chem. Commun., 2015, **51**, 10875; (g) F. Yu, P. Li, P. Song, B. Wang, J. Zhao and K. Han, Chem. Commun., 2013, **49**, 2013; (h) B. Wang, P. Li, F. Yu, P. Song, X. Sun, S. Yang, Z. Lou and K. Han, Chem. Commun., 2013, **49**, 2445; (j) Z. Lou, P. Li and K. Han, Acc. Chem. Res., 2015, 48, 1358.
- 14 (a) X. Qu, C. Li, H. Chen, J. Mack, Z. Guo and Z. Shen, *Chem. Commun.*, 2013, **49**, 7510; (b) Z. Dong, X. Le, P. Zhou, C. Dong and J. Ma, *New J. Chem.*, 2014, **38**, 1802; (c) F. Ma, M. Sun, K. Zhang, H. Yu, Z. Wang and S. Wang, *Anal. Chim. Acta*, 2015, **879**, 104.
- (a) K. Komatsu, Y. Urano, H. Kojima and T. Nagano, J. Am. Chem. Soc., 2007, **129**, 13447; (b) D. Srikun, E. W. Miller, D. W. Domaille and C.J. Chang, J. Am. Chem. Soc., 2008, **130**, 4596.
- 16 (a) J. Fan, M. Hu, P. Zhan and X. Peng, *Chem. Soc. Rev.*, 2013,
  42, 29; (b) L. Yuan, W. Lin, K. Zheng and S. Zhu, *Acc. Chem. Res.*, 2013, 46, 1462; (c) Y. R. Zhang, X. P. Chen, J. Shao. J. Y. Zhang, Q. Yuan, J. Y. Miao and B. X. Zhao, *Chem. Commun.*, 2014, 50, 14241.
- (a) M.H. Lee, J.S. Kim and J.L Sessler, *Chem. Soc. Rev.*, 2015, 44, 4185; (b) H, Zhu, J. Fan, B. Wang and X. Peng, *Chem. Soc. Rev.*, 2015, 44, 4337; (c) G.R. Hamilton, S.K. Sahoo, S. Kamila, N. Singh, N. Kaur, B.W. Hyland and J.F. Callan, *Chem. Soc. Rev.*, 2015, 44, 4415; (d) Y. Chen, Y. Bai, Z. Han, W. He and Z. Guo, *Chem. Soc. Rev.*, 2015, 44, 4415; (d) Y. Chen, Y. Bai, Z. Han, W. He and Z. Guo, *Chem. Soc. Rev.*, 2015, 44, 4517; (e) J. Yin, Y. Hu and J. Yoon, *Chem. Soc. Rev.*, 2015, 44, 4619; (f) Z. Yang, J. Cao, Y. He, J.H. Yang, T. Kim, X. Peng and J.S. Kim, *Chem. Soc. Rev.*, 2014, 43, 4563; (g) X. Li, X. Gao, W. Shi and H. Ma, *Chem. Rev.*, 2014, 114, 590.
- 18 (a) L. Long, W. Lin, B. Chen, W. Gao and L. Yuan, Chem. Commun., 2011, 47, 893. (b) C. Zhang, L. Wei, C. Wei, J. Zhang, R. Wang, Z. Xi and L. Yi, Chem. Commun., 2015, 51, 7505.
- (a) X. Zhou, F. Su, H. Lu, P. Senechal-Willis, Y. Tian, R.H. Johnson and D.R. Meldrum, *Biomaterials*, 2011, **33**, 171. (b)
  L. He, W. Lin, Q. Xu and H. Wei, *Chem. Commun.*, 2015, **51**, 1510; (c) Z. Guo, G. Chen, G. Zeng, Z. Li, A. Chen, J. Wang and L. Jiang, *Analyst*, 2015, **140**, 1772.
- (a) X.J. Zou, Y.C. Ma, L.E. Guo, W.X. Liu, M.J. Liu, C.G. Zou, Y. Zhou and J.F. Zhang, *Chem. Commun.*, 2014, **50**, 13833; (b) X. Li, J. Cheng, Y. Gong, B. Yang and Y. Hu, *Biosens. Bioelectron.*, 2015, **65**, 302.
- 21 M. Martinez-Cutillas, V. Gil, N. Mañé, P. Clavé, D. Gallego, M.T. Martin and M. Jimenez, *Pharmacol. Res.* 2015, 93, 52.

4 | J. Name., 2012, 00, 1-3

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