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# COMMUNICATION

# All-In-One Azides: Empowered Click Reaction for *in vivo* Labeling and Imaging of Biomolecules

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Yaning Su,<sup>a, c,  $\ddagger$ </sup> Li Li,<sup>b, c,  $\ddagger$ </sup> Haibin Wang,<sup>c</sup> Xiaochen Wang<sup>c</sup> and Zhiyuan Zhang<sup>c,d</sup>  $\ast$ 

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We designed and synthesized all-in-one (AIO) reactive azide reagents for bioothogonal reactions with highly efficient Cu(I) ligand moieties, an azido group, and functional tags for imaging or purification. The AIO reagents displayed fast and efficient click ligation and can be applied in a wide range of *in vivo* systems.

The past two decades have witnessed the development and widespread application of bioorthogonal reactions for selectively labeling biomolecules without using genetic modifications.<sup>1</sup> mong all the bioorthogonal reactions, ligand assisted copper(I) catalyzed azide-alkyne cycloaddition (CuAAC) reactions are the most broadly applied one.<sup>2</sup> To date, CuAAC has been widely applied in the labeling and imaging of proteins,<sup>3</sup> glycans,<sup>4</sup> nucleic acids<sup>5</sup> and other macromolecules<sup>6</sup> both *in vitro* and *in vivo*. To guarantee the reaction speed and the overall efficiency of CuAAC reactions *in vivo*, excessive amounts of Cu(I) complex (copper sources, ligands, and sodium ascorbate) are typically required, which results in copper (I) mediated cellular toxicity that casts a shadow on their further application in living systems.<sup>1a,7</sup>

One strategy to improve the biocompatability of CuAAC reactions is to modify the structure and compatibility of the Cu(I) ligand to improve ligation kinetics and efficiency and decrease copper loading. Ligands with enhanced water solubility such as THPTA,<sup>7</sup> BTTAA<sup>8</sup>, and BTTES<sup>3</sup> (Scheme 1) have been developed and been shown to accelerate the CuAAC reaction significantly and to act as ROS scavengers to decrease Cu(I)-induced cell toxicity. A more effective strategy is developing highly reactive azides by combining Cu(I) chelating moieties and the azide component in one molecule to raise the effective copper concentration. This strategy was first exploited by Ting et al., who combined a copper chelating pyridine group into the substrate azide skeleton (Scheme 1,



Scheme 1 (a) Structure of classic triazole ligands and chelating azides. (b) Structure features of the All-In-One reagents.

 $CA_1$ ).<sup>9</sup> This chelating azide showed much faster kinetics than standard azides, and had efficient performance in bioorthogonal labeling with the assistance of THPTA as a ligand. Recently, the Taran group combined the triazole ring, a stronger Cu(I) chelating moiety into the azide reagent (Scheme 1,  $CA_2$ ).<sup>10</sup> Their system showed even faster kinetics than ligand catalyzed CuAAC reactions and was effective in labeling and imaging tubulin without additional ligands.

The structure of CA<sub>2</sub> was derived from triazole ligands for CuAAC, with two triazole rings linked to the nitrogen atom at the center of the skeleton. It is speculated that the fast kinetics is resulted from the intramolecular chelation pattern that enhances the electrophilicity of azido group and formation of the metallacycle intermediate.<sup>10</sup> With the third arm of the nitrogen atom left for an azido group acting as a war-head for further bioorthogonal coupling, tags for detection or purification for application in biogocical systems would have to be added to the triazole rings. However, since modifications at this position are necessary to improve the reactivity and solubility as shown in THPTA and BTTES, this tag linkage might undermine the properties of the tagged reactive azides.

In order to preserve the positions for compound property adjustment and to provide a more flexible scaffold for tag conjugation, herein, we report the design and synthesis of a

<sup>&</sup>lt;sup>a.</sup> College of Life Sciences, Beijing Normal University, Beijing 100875, China

<sup>&</sup>lt;sup>b.</sup> School of Life Sciences, Peking University, Beijing 100871, China
<sup>c</sup> National Institute of Biological Sciences (NIBS), Beijing 102206, China. E-mail:

 <sup>&</sup>lt;sup>d</sup> Collaborative Innovation Center for Cancer Medicine, Beijing 102206, China, E-India.

*<sup>‡</sup>* These authors contributed equally to this work.

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unique version of reactive azides, the All-In-One (AIO) copper chelating azide reagents with a tetrahedron carbon atom at the core of the structural skeleton(Scheme 2). By placing a tetrahedron carbon atom at the center, a total of four arms can be utilized to fulfill different aspirations. With two arms linked by triazole rings and the third arm as an azido group, the fourth arm remains free for various choices for tag fragments. This leaves positions  $R_1$  and  $R_2$  (Scheme 2) free for the convenient adjustment of reactivity and biocompatibility of AIO reagents.

We first evaluated the kinetics and overall performance of AIO reagent **1** in a pure chemistry system. (See ESI for full synthesis details). A model reaction, in which AIO reagent **1** reacts with a coumarin derivative to form a fluorescent product **5** (Fig. 1a), was selected for the kinetic study; the formation of product was measured via its fluorescence.<sup>15</sup> (Fig. S1, S2, S3) The model reaction was carried out in the presence of different concentrations of Cu(I) (0.25 to 4 equivalent). The fluorescence intensity of **5** was measured every 6 seconds during a total time course of 5 minutes. The yield and time course of the reaction was compared with TBTA, the typical CuAAC ligand, BTTES and THPTA, a water soluble ligand, and most





Fig.1 (a) Model reaction for kinetic measurements. (b) Time course under different CuSO<sub>4</sub> equivalents. For AIO reagent 1 and A19, reactions were performed with 10  $\mu$ M 1 or A19, 100  $\mu$ M 7-ethynylcoumarin, 100 mM sodium ascorbate and different equivalence of CuSO<sub>4</sub> accordingly. Control tractions with triazole ligands were performed with 50  $\mu$ M 2-azidoethenal, 50  $\mu$ M CuSO<sub>4</sub>, 500  $\mu$ M 7-ethynylcoumarin, 100 mM sodium ascorbate. All reactions are carried out in DMF: H<sub>2</sub>O 1:1 solution.

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importantly, with A19 (Scheme 1, CA<sub>2</sub>), the chelating reactive azide Taran reported. The kinetic study results indicate that AlO reagent **1** is extremely active, yielding product **5** at 60% within 90 seconds(Fig. 1b). The chelating azide A19 showed similar kinetics but a little lower yields (48%), and other control reactions catalyzed by ligands only had yields of no higher than 13%, at 5min (Fig. 1b).

The observed rate constant  $k_{obs}$  of **1** was calculated. When the CuSO<sub>4</sub> level was under 1 equivalent, the reaction rate depends on Cu(I) concentration, increasing from 28 M<sup>-1</sup> • s<sup>-1</sup> at 0.25 equivalent of Cu to 121 M<sup>-1</sup> • s<sup>-1</sup> at 1 equivalent of Cu (Table S1, ig. S4, Fig. S6). However, the reaction rate was highly similar when the CuSO<sub>4</sub> level was in the range of 1~4 equivalents. Under the same reaction condition (with 1 equivalent of Cu(I)),  $k_{obs}$  of reactions catalyzed by other ligands were also tested and compared. The  $k_{obs}$  (121 M<sup>-1</sup> • s<sup>-1</sup>) of AIO reagent **1** is much higher than that of TBTA (0.39 M<sup>-1</sup> • s<sup>-1</sup>), THPTA (0.99 M<sup>-1</sup> • s<sup>-1</sup>), BTTES (1.23 M<sup>-1</sup> • s<sup>-1</sup>), and slightly higher than **A19** (82 M<sup>-1</sup> • s<sup>-1</sup> Table S1, Fig.S4).

Since the differences in the amounts of  $CuSO_4$  were observed to affect the rate constant, we next investigated the order of Cu in this reaction. By plotting  $ln[k_{obs}]$  versus  $ln[CuSO_4]$ , we discovered that the order of Cu appeared to differ depending on its loading. The order was determined to be 0.99 under conditions of insufficient CuSO<sub>4</sub>, while with excessive amounts of CuSO<sub>4</sub>, the rate order was 0.08, which indicates that the reaction rate is independent on the CuSO<sub>4</sub> concentration when the Cu supply is more than one equivalent (Fig. S5).

It was proposed that copper chelating azides facilitates the formation of metallacycle intermediate in the ratedetermining step, leading to the fast kinectics.<sup>10,12</sup> To test if AIO reagents have similar reaction mechanism, we measured the association constant (Ka) of AIO reagent **1**, **A19** and their corresponding products (**5** and **25**) with Cu(I). Chelating azides and their products all have similar affinity with Cu(I), which are about 20-fold higher than that of non-chelating azide with premixed TBTA and Cu(I) complex (Fig. S8).<sup>§</sup> These results provided evidence that chelating azides coordinate to copper in a more efficient way than standard azides, which might in turn promote the metallacycle intermediate formation.

Recently, halide inhibiton of CuAAC reaction has been reported.<sup>7,13</sup> We set out to evaluate the effect of halides on AIO reagents assisted CuAAC reaction. The model reaction was carried out using 1 equivalent of Cu(I), accompanied by the addition of 1 equivalent or 100 equivalent of NaX (X=Cl, Br, I). The real-time detection of the reaction course (Fig. S10, S11, Table S2) indicated that only minor inhibition was seen under the condition of 100 equivalent of sodium iodide, while no obvious inhibitions were seen on both the yield and reaction rate under the other halide conditions. The good halide tolerance property of AIO reagents provided a guarantee for applications in biological systems.

It has been reported that triazole ligands such as THPTA,<sup>7</sup> BTTES<sup>3</sup> and BTTAA<sup>8</sup> all have protective effect towards Cu(I) induced toxicity. Due to the structure similarity, we investigated whether AIO reagent **1** has protective effects for cell viability when they were co-incubated with Cu(I). Cells

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incubated with Cu(I) alone for 4 hours had an extremely high incidence of cell death (survival rate is less than 5%), even when the Cu(I) concentration was as low as 20  $\mu$ M. The presence of AlO reagent **1**, however, showed an obvious protective effect, increasing the rate of cell survival to more than 90% at even 100  $\mu$ M Cu(I)(Fig. S12). Complexes composed of Cu(I) and A19/THPTA/TBTA/BTTES also showed protective effects, all raising the cell survival rate to more than 40% at 100  $\mu$ M Cu(I) treatment.

Encouraged by the results of the evaluation of the AIO reagent properties, we next evaluated the utility of the AIO click reaction in live cells. For this purpose, we synthesized FITC or biotin tagged reagents **2** and **3** (Scheme 2, See ESI for full synthesis details). Jurkat cells were cultured in the presence of peracetylated N-(4-pentynoyl)mannosamine (Ac<sub>4</sub>ManNAI), a modified biosynthesis precursor of alkynyl sialic acid, so that the bioorthogonal alkyne group can be assembled on cell surface glycoproteins through a biosynthesis procedure (Fig. S13).<sup>3,4b</sup>

The reactivity of FITC-tagged AIO reagent 2 was investigated initially. The alkyne labeled cells were treated with a mixture of AIO 2 (100  $\mu$ M) and CuSO<sub>4</sub> (100  $\mu$ M), together with 10 mM sodium L-ascorbate for 1 hour. A clear and strong green fluorescence in the FITC channel was observed in the Ac<sub>4</sub>ManNAl treated cells, and this merged well with the red fluorescence from cell membrane stained with Dil (Fig. 2a, S14, S15). In the control sample lacking  $Ac_4ManNAI$  treatment, the green fluorescence was hardly visible (Fig. 2a, S14, S15), indicating that AIO reagent 2 can efficiently label membrane proteins on the surface of living cells. Next, the performance of AIO regent 3 (Scheme 2) was evaluated in the biotin pulldown process. Alkyne labeled Jurkat cells were collected and lysed. This was followed with the AIO click reaction, biotin affinity pull down experiments, and Western blotting. As indicated, the Western blots had clear bands in the Ac₄ManNAl treated sample but not in the control sample (Fig. 2b).

In addition to the mono-tag AIO reagents **2** and **3**, we also synthesized a bi-tag AIO reagent **4** (Scheme 2), which has both biotin and FITC moieties. This bi-functional AIO reagent can be

used for imaging and affinity purification at the same time. We tested the application of **4** using the aforementioned alkyne modified Jurkat cells. After affinity purification with the biotin tag, the isolated portion was examined for both the biotin and FITC tags; apparent and sufficient glycoprotein imaging and biotin labeling is shown(Fig. 2c). This bi-functional AIO reagent can be further applied in activity-based protein profilings to isolate labeled protein via fluorescent signals from SDS gel after biotin-strepavidin enrichment, which could simplify the whole procedure and at the same time lower the background signals.

To explore the further potential of this AIO labeling, we tested its application in chemical genetics, a recently developed research field that includes active small molecule-based target identification starting from phenotype screening.<sup>14</sup> Human epidermal growth factor receptor 2 (HER2) and its covalent binding inhibitor (compound 6, Canertinib analogue with EC<sub>50</sub>=20 nM for Ba/F3-HER2) (Fig. 3a) were chosen for our investigation of activity based targeting of specific protein (Fig. S16).<sup>15</sup> Treatment of HER2/H878Y-transformed BA/F3 cells with compound 6 localized the bioorthogonal alkyne group on HER2 through covalent binding of the probe with the intracellular kinase domain of HER2.<sup>15b</sup> Cells were then then incubated with the fluorescence AIO reagent 2/CuSO4/sodium ascorbate complex. In order to confirm the localization of the FITC labeling, cells were fixed and immunostained with HER2 antibody. The fluorescence of FITC was mainly concentrated at the cell membrane region, and the FITC fluorescence coincided well with the red fluorescence of the HER2 immunostaining(Fig. 3b, S17), indicating that AIO reagent 2 successfully penetrated into cells and labeled the protein-bound probe with the imaging tag.

The affinity pulldown assay showed a similar performance of probe **3** in targeting applications in complex systems. Cell lysates obtained from compound **6** preincubated cells were treated with AIO reagent **3** for the labeling of the biotin tag. Following strepavidin affinity pull-down, the HER2 signal was clearly present in the Western blot (Fig. 3c) compared with the control group, which had not been treated with compound **6**.



Fig.2 (a) Confocal microscopy images of probe 2 labeling on the surface of live Junket cells. (blue channel: Hoechest; green channel: FITC; red channel: Dil. Scale bar:  $5 \mu m$ ) (b) Western blot analysis for biotin tag after probe 3 labeling on the Junket cell proteome. (c) In-gel fluorescence and biotin Western blotting for the detection of labeling efficiency of bifunctinal probe 4. (+): cells cultured with 100  $\mu$ M Ac<sub>4</sub>ManNAI; (-): cells cultured without Ac<sub>4</sub>ManNAI. For all imaging and labelling, 100  $\mu$ M AlO reagents/Cu(I) complex was used.



Fig. 3 (a) Structures of canertinib and its alkynyl analogue 6. (b) Confocal microscopy images of probe 2 labeled HER2/H878Y-transformed BA/F3 cells. (blue channel: Hoechest; green channel: FITC; red channel: anti-HER2 immunostaining . Scale bar: 5 µm) (c) Werstern blot of HER2 attransformed BA/F3 proteome. (+): cells treated with 10 µM compound 6 for 3hr; (-): cells treated with DMSO. For all imaging and labelling, 100 µM AlO reagents/Cu(i) complex was used after compound 6 was washed away.

These results indicated that AIO labeling system can be used in activity based protein profiling (ABPP) applications for target identification, target localization and functional study with active small molecules.

After we successfully labeled cell surface glycoproteins and HER2 kinase in vivo, we focused our efforts on labeling and imaging in a living organism, C. elegans. C. elegans is an excellent model organism for studies of glycan dynamics in development, disease, and normal physiological processes.<sup>16</sup> We cultured C. elegans larvae on nematode growth medium (NGM)-agar plates without (untreated, "-") or with 5 mM Ac<sub>4</sub>GalNAI(treated, "+") for at least two generations. The mixstaged Ac<sub>4</sub>GalNAI worms were soaked in 100 µM AIO reagent 2/Cu(I) for 40 min, washed softly, and recovered for 4 h on NGM plates prior to imaging. In young adult hermaphrodites, GalNAI-specific labeling was clearly observed in pharynx and tail regions (Fig. 4b,e,f), in accordance with previous reports,<sup>26</sup> while no specific staining was observed in control worms (Fig. 4a,c,d). Similar staining patterns were observed in worms that were treated with much lower concentrations of AIO reagent or Ac<sub>4</sub>GalNAI (250 μM) (Fig. S18), highlighting the high labeling efficiency of AIO reagent 2.

To conclude, AIO reagents displayed efficient ligation and had excellent catalytic kinetics and obvious cellular protective effect against Cu(I)-induced toxicity. We successfully



Fig. 4 (a-b) Confocal imaging of live C. elegans young adult hermaphrodites after probe 2 lableing. Scale bar: 50  $\mu m$  (c-f) Higher magnification images of the pharynx (c, e) and tail region (d, f). Scale bar: 10  $\mu m$ . (+: Ac\_aGalNAl-cultured worms; -: normally cultured worms.) For labelling and imaging, 100  $\mu M$  AlO reagents/Cu(l) complex was used.

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