

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



Journal Name

COMMUNICATION

Copper(II)-mediated regioselective *N*-arylation of pyrroles, indoles, pyrazoles and carbazole *via* dehydrogenative coupling

Received 00th January 20xx,
Accepted 00th January 20xx

Pradeep Sadhu and Tharmalingam Punniyamurthy*

DOI: 10.1039/x0xx00000x

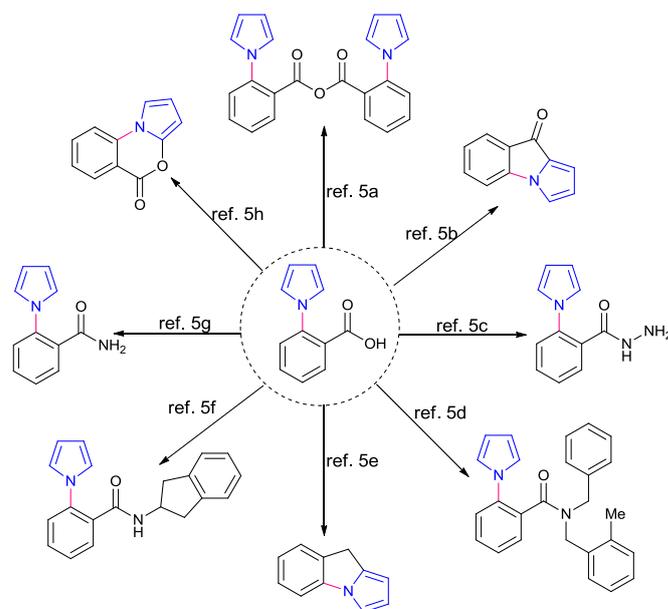
www.rsc.org/

A copper(II)-mediated regioselective *N*-arylation of azoles has been developed using 8-aminoquinoline amide as a directing group. This reaction shows broad substrate scope with different azoles such as pyrroles, indoles, pyrazoles and carbazole with good yields.

C-H Functionalization reactions using directing groups have emerged as more reliable synthetic tools for the regioselective construction of carbon-carbon and carbon-heteroatom bonds.^{1,2} Among them, the formation of C-N bonds³ has attracted considerable attention due to their presence in numerous compounds that are important in biological and pharmaceutical sciences. In particular, *N*-arylated azoles serve as versatile building blocks for the construction of nitrogen-containing structural scaffolds that exhibit interesting medicinal properties (Scheme 1).^{4,5} Commonly, they are prepared by the cross-coupling^{6,7} of azoles with aryl halides,⁸ aryl boronic acids,⁹ aryl bismuths¹⁰ and aryl lead reagents.¹¹ Recently, Patureau and co-workers reported a cross-dehydrogenative coupling approach for the *ortho*-selective amination of diaryl amines with carbazole using Ru as the catalyst and Cu(OAc)₂ as the oxidant at 150 °C (Scheme 2a).¹² In this context, the development of new synthetic routes using inexpensive metal source¹³ with broad range of nitrogen heterocycles is highly desirable. Herein, we report an efficient copper(II) acetate mediated *ortho*-selective dehydrogenative C-N cross-coupling of arenes with pyrroles, indoles, pyrazoles and carbazole using a removable amide as a directing group (Scheme 2b). This protocol is simple, general and effective at moderate temperature (70 °C) with good yields.

First, we commenced the optimization studies with *N*-(quinolin-8-yl)benzamide **1a** and pyrrole as the model substrates using different

bases, solvents and varied amounts of Cu(OAc)₂ (Table 1). Gratifyingly, the amination selectively occurred at the *ortho*-position to give the target product **2a** in 34% yield when the substrates were stirred with 1 equiv of Cu(OAc)₂ at 70 °C in the presence of K₂CO₃ in DMSO (entry 1). The use of Cs₂CO₃ as the base led to an increase in the yield to 44% (entry 3), while Na₂CO₃ afforded inferior results (entry 2). Subsequent screening of the solvents led to an increase in the yield to 51% using DMF, whereas NMP, *i*PrOH and 1,4-dioxane were not effective (entries 4-7). Increasing the amount of Cu(OAc)₂ (1.5 equiv) and pyrrole (3 equiv) led to the further enhancement in the yield to 79% (entries 8-11). Similar results were observed using oxygen atmosphere in place of air (entry 12-13). Recrystallization of **2a** in CH₃CN gave single crystal whose structure was determined by X-ray analysis (see Supporting Information). A control experiment confirmed that without Cu(OAc)₂ the formation of **2a** was not observed and the starting material was recovered intact (entry 14).

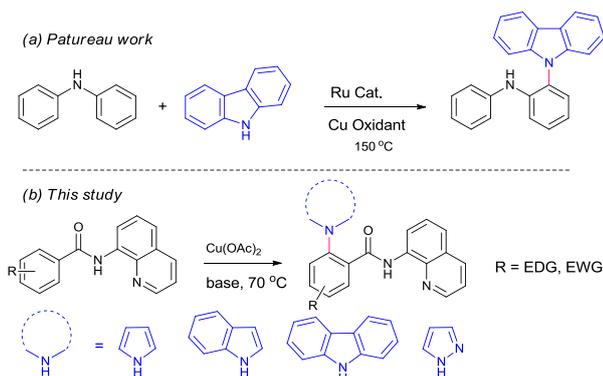


Scheme 1 For some examples of the utilities of 2-(1H-pyrrol-1-yl)benzoic acid

^a Department of Chemistry, Indian Institute of Technology Guwahati, Guwahati 781039, India. E-mail: tpunni@iitg.ernet.in

[†] Footnotes relating to the title and/or authors should appear here.

Electronic Supplementary Information (ESI) available: Supporting information for this article having experimental procedure, crystal structure and data of **2a**, characterization data and NMR spectra (¹H and ¹³C) of **2a-m**, **3a-j**, **4a-b**, **5** and **7** is given via a link at the end of the document. See DOI: 10.1039/x0xx00000x



Scheme 2 N-Arylation of azoles via dehydrogenative cross-coupling

Table 1 Optimization of the reaction conditions

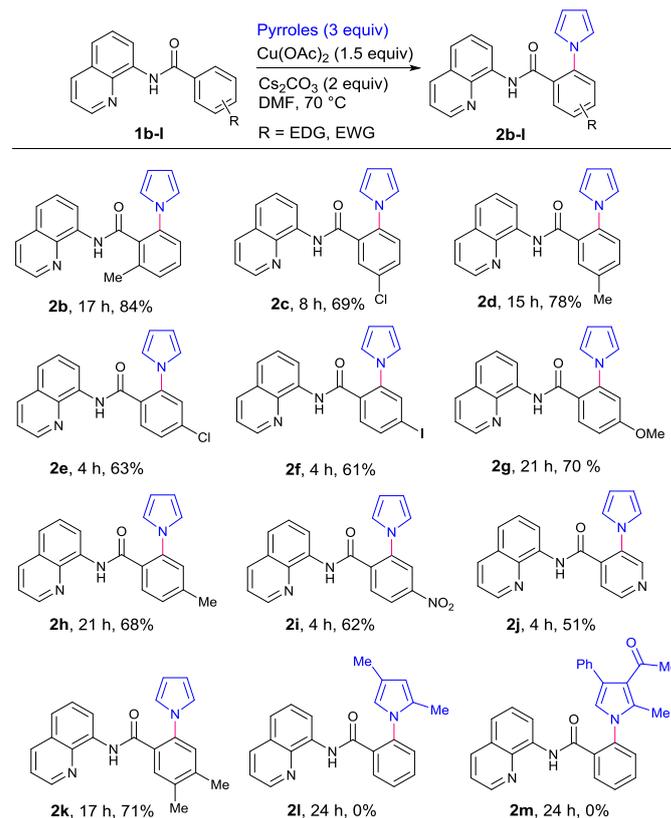
Entry	[Cu] (equiv)	Base	Solvent	Yield (%) ^{a,b}
1	Cu(OAc) ₂ (1)	K ₂ CO ₃	DMSO	34
2	Cu(OAc) ₂ (1)	Na ₂ CO ₃	DMSO	trace
3	Cu(OAc) ₂ (1)	CS ₂ CO ₃	DMSO	44
4	Cu(OAc) ₂ (1)	CS ₂ CO ₃	NMP	10
5	Cu(OAc) ₂ (1)	CS ₂ CO ₃	<i>i</i> PrOH	trace
6	Cu(OAc) ₂ (1)	CS ₂ CO ₃	1,4-dioxane	0
7	Cu(OAc) ₂ (1)	CS ₂ CO ₃	DMF	51
8	Cu(OAc) ₂ (1.5)	CS ₂ CO ₃	DMF	71
9	Cu(OAc) ₂ (2.0)	CS ₂ CO ₃	DMF	70
10 ^c	Cu(OAc) ₂ (1.5)	CS ₂ CO ₃	DMF	64
11 ^d	Cu(OAc) ₂ (1.5)	CS ₂ CO ₃	DMF	79
12 ^e	Cu(OAc) ₂ (1.0)	CS ₂ CO ₃	DMF	50
13 ^e	Cu(OAc) ₂ (0.2)	CS ₂ CO ₃	DMF	18
14	-	CS ₂ CO ₃	DMF	0

^a Reaction conditions: *N*-(quinolin-8-yl)benzamide **1a** (0.2 mmol), Cu(OAc)₂, pyrrole (0.4 mmol), base (0.4 mmol), solvent (1 mL), 70 °C, 21 h. ^b Isolated yield. ^c Pyrrole (0.3 mmol) was used. ^d Pyrrole (0.6 mmol) was used. ^e Using O₂ balloon.

Having the optimal conditions in hand, we sought to further explore the reaction scope with a broad range of *N*-(quinolin-8-yl)benzamide derivatives (Scheme 3). The substrates **1b** having 2-Me substituent underwent the reaction to produce the target product **2b** in 84% yield. The substrates **1c-d** bearing 3-Cl and 3-Me functionalities readily reacted to afford the desired products **2c** and **2d** in 69 and 78% yields, respectively, while **1e-i** bearing electron donating and electron-withdrawing groups in the *para* position proceeded reaction smoothly to provide the corresponding *ortho*-aminated products **2e-i** in 61-70% yields. Heterocyclic compound, isonicotinamide **1j** underwent reaction to give the aminated product **2j** in 51% yield, whereas the reaction of 3,4-diMe substituted substrate **1k** produced the target product **2k** in 71% yield. However, the substrate **1a** with substituted pyrroles, 2,4-dimethylpyrrole and 1-(2-methyl-4-phenyl-1*H*-pyrrol-3-yl)ethan-1-one, failed to produce the desired **2l** and **2m**, which might be due to

the steric hindrance of the substituents present in the pyrroles. These results suggest that the protocol is compatible with the substrates bearing substitution at *ortho*, *meta* as well as *para* positions. Furthermore, the substrates with electron withdrawing group **2i** exhibit greater reactivity compared to that having electron donating group **2g**. In addition, the substrates having halides such as chlorine **2c**, **2e** and iodine **2i** are tolerated under these reaction conditions.

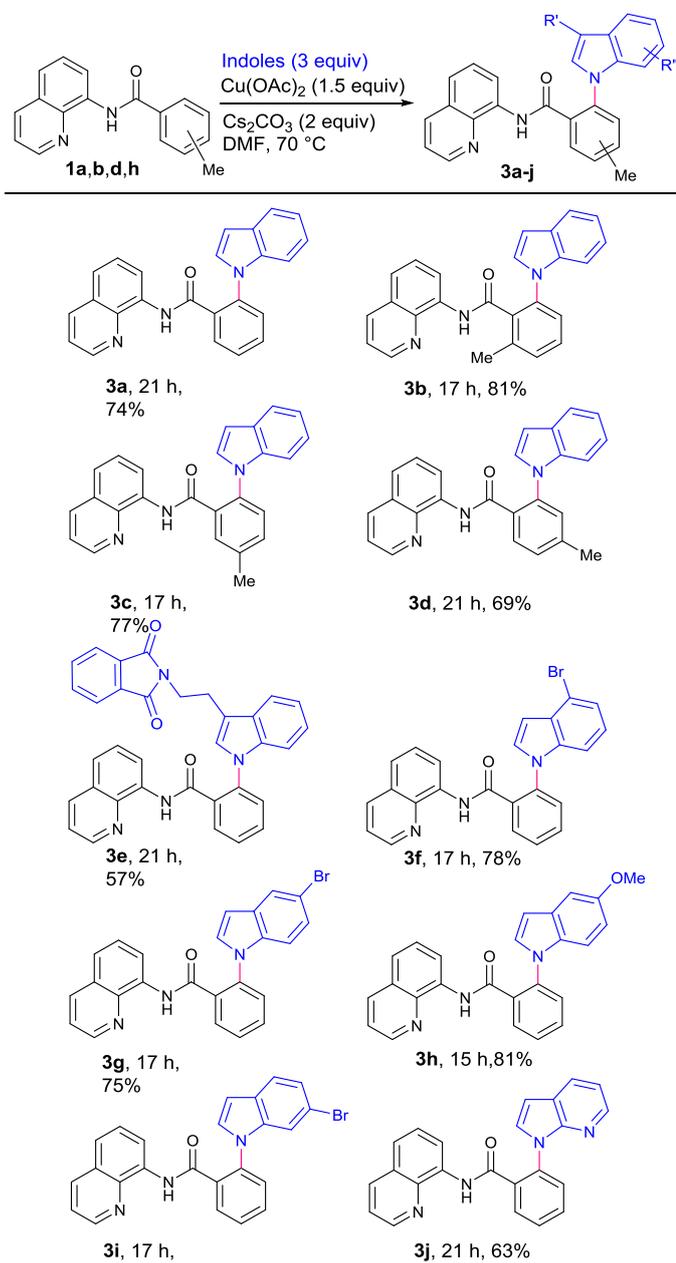
Next, the scope of the reaction was extended to the *N*-arylation of indoles with 8-aminoquinoline amides (Scheme 4). The reactions proceeded efficiently in high yields. For examples, *N*-(quinolin-8-yl)benzamide **1a** underwent reaction to furnish the target *ortho*-aminated product **3a** in 74% yield. In addition, the reaction of the substrates **1b**, **1d** and **1h** with methyl substituent at *ortho*, *meta* and *para*-positions produced the corresponding aminated compounds **3b-d** with 69-81% yields. Furthermore, the substrate **1a** with the substituted indole, 2-(2-(1*H*-Indol-3-yl)ethyl)isoindoline-1,3-dione, furnished **3e** in 57% yield. Similarly, the reactions of **1a** with the indoles bearing 4-bromo, 5-bromo, 5-methoxy and 6-bromo substituents afforded the corresponding cross-coupled products **3f-i** in 75-81% yields, while azaindole underwent reaction to produce **3j** in 63% yield.



Scheme 3 Substrate scope of *N*-(quinolin-8-yl)benzamides with pyrrole.^{a,b} Reaction conditions: *N*-(quinolin-8-yl)benzamide **1a** (0.2 mmol), Cu(OAc)₂ (0.3 mmol), pyrrole (0.6 mmol), CS₂CO₃ (0.4 mmol), DMF (1 mL), 70 °C. ^b Isolated yield.

Furthermore, we explored the scope of the procedure for the reactions of different azoles with *N*-(quinolin-8-yl)benzamide **1a** as

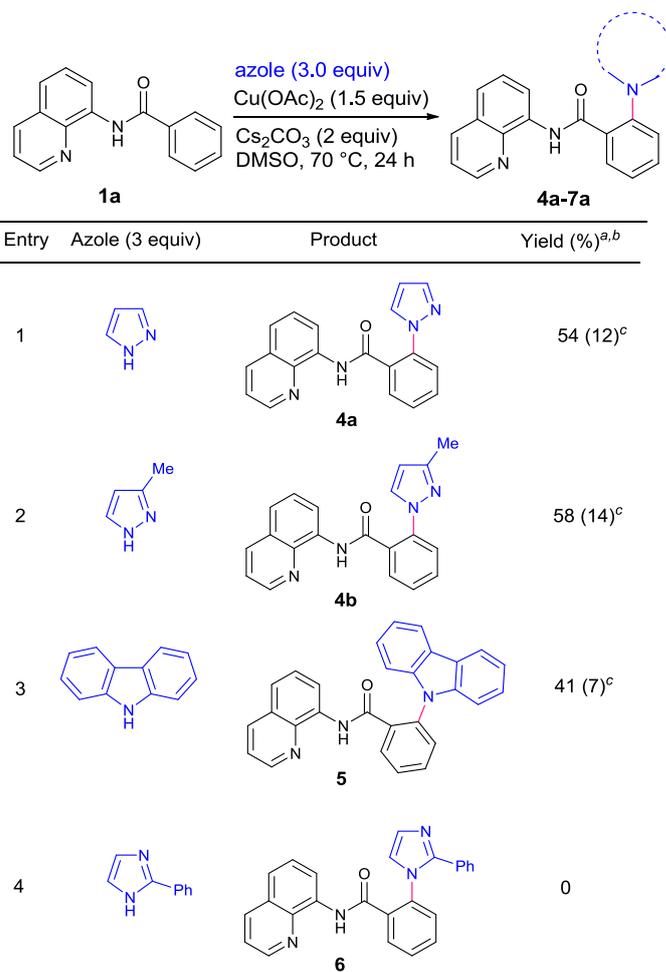
the representative example (Scheme 5). These reactions exhibited superior results employing DMSO as the solvent compared to that of DMF. For examples, pyrazole underwent reaction to produce the target product **4a** in 54% yield. Similar result was observed with 3-methylpyrazole leading to the formation of **4b** in 58% yield. In addition, the reaction of carbazole furnished the desired cross-coupled product **5** in 41% yield. However, 2-phenylimidazole failed react to yield **6** and the starting materials were recovered.



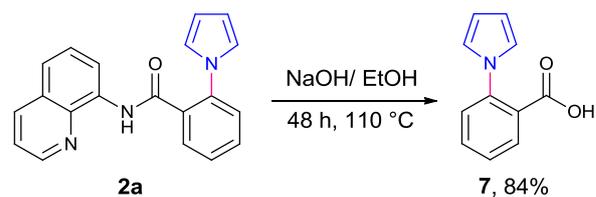
Scheme 4 Substrate scope of *N*-(quinolin-8-yl)benzamide with indoles.^{a, b} ^a Reaction conditions: *N*-(quinolin-8-yl)benzamide **1a** (0.2 mmol), $\text{Cu}(\text{OAc})_2$ (0.3 mmol), indole (0.6 mmol), Cs_2CO_3 (0.4 mmol), DMF (1 mL), 70 °C. ^b Isolated yield.

Finally, the removal of the directing group was studied using compound **2a** as the representative example (Scheme 6).^{13e} The compound **2a** underwent readily hydrolysis with NaOH in ethanol at

110 °C to afford 2-(1*H*-pyrrol-1-yl) benzoic acid **7** in 84% yield, which is a key intermediate for the construction of diverse nitrogen containing structural frameworks as described in Scheme 1.⁵

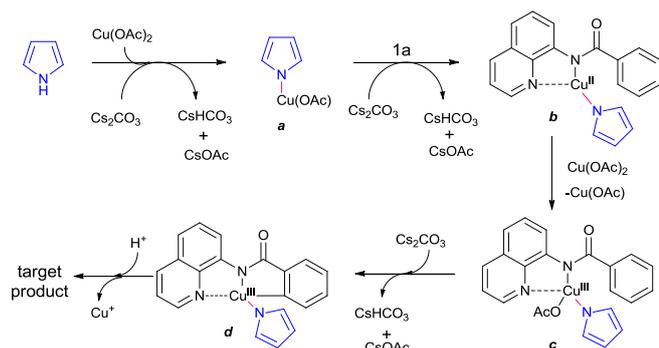


Scheme 5 Substrate scope of *N*-(quinolin-8-yl)benzamide **1a** with different azoles.^{a, c} ^a Reaction conditions: *N*-(quinolin-8-yl)benzamide **1a** (0.2 mmol), $\text{Cu}(\text{OAc})_2$ (0.3 mmol), azole (0.6 mmol), Cs_2CO_3 (0.4 mmol), DMSO (1 mL), 70 °C, 24 h. ^b Isolated yield. ^c Using DMF.



Scheme 6. Directing group removal

The proposed reaction pathway is shown in Scheme 7. In the presence of base, azole may undergo reaction with $\text{Cu}(\text{OAc})_2$ to produce copper(II) species **a**,¹⁴ which may react with the substrate **1a** via ligand exchange to give copper(II) intermediate **b**. $\text{Cu}(\text{OAc})_2$ -mediated oxidation of **b** may lead to the formation of copper(III) species **c**. Intramolecular C-H cupration of the aryl ring may give the intermediate **d**, which can provide the target product by reductive elimination followed by protonation. This proposed reaction pathway also explains the necessity of excess $\text{Cu}(\text{OAc})_2$ and base to realize the products in good yields.



Scheme 7. Proposed reaction pathway

In summary, we have developed an efficient copper-mediated 8-aminoquinoline amide directed *N*-arylation of pyrroles, indoles, pyrazoles and carbazole *via* intermolecular cross-dehydrogenative coupling reaction. This protocol exhibits broad substrate scope and high functional group tolerance, which may open a new avenue for further development of dehydrogenative coupling protocols for the coupling azoles with hydrocarbons.

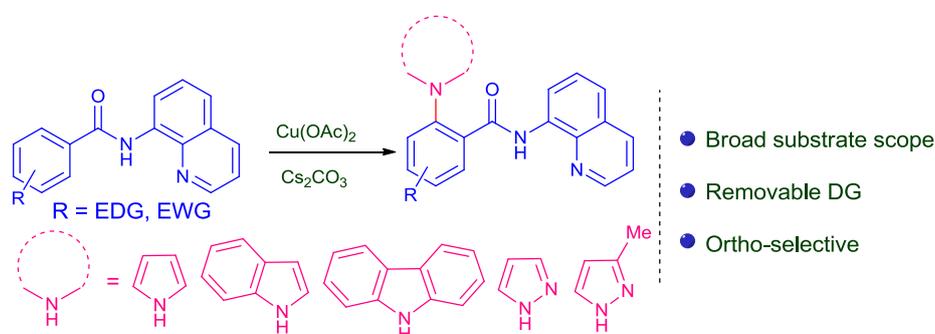
We gratefully acknowledge Department of Science and Technology (SR/S1/OC-55/2011) and Council of Scientific and Industrial Research (02(0088)/12/EMR-II) for financial support. P. S. thanks UGC for SRF Fellowship. We also thank Central Instrumental Facility, IIT Guwahati for NMR facilities.

Notes and references

- For recent reviews on C-H Functionalization, see: (a) D. A. Colby, R. G. Bergman and J. A. Ellman, *Chem. Rev.* 2010, **110**, 624; (b) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147; (c) A. E. Wendlandt, A. M. Suess and S. S. Stahl, *Angew. Chem., Int. Ed.*, 2011, **50**, 11062; (d) L. Ackermann, *Chem. Rev.*, 2011, **111**, 1315; (e) J. Wencel-Delord, T. Dröge, F. Liu and F. Glorius, *Chem. Soc. Rev.*, 2011, **40**, 4740; (f) P. B. Arockiam, C. Bruneau and P. H. Dixneuf, *Chem. Rev.*, 2012, **112**, 5879; (g) G. Rouquet and N. Chatani, *Angew. Chem., Int. Ed.*, 2013, **52**, 11726; (h) F. Zhang and D. R. Spring, *Chem. Soc. Rev.*, 2014, **43**, 6906; (i) M.-L. Louillat and F. W. Patureau, *Chem. Soc. Rev.*, 2014, **43**, 901; (j) O. Daugulis, J. Roane and L. D. Tran, *Acc. Chem. Res.*, 2015, **48**, 1053; (k) Z. Chen, B. Wang, J. Zhang, W. Yu, Z. Liu and Y. Zhang, *Org. Chem. Front.*, 2015, **2**, 1107.
- For some recent examples on (a) N. Schröder, J. Wencel-Delord and F. Glorius, *J. Am. Chem. Soc.*, 2012, **134**, 8298; (b) K. Parthasarathy and C. Bolm, *Chem. -Eur. J.*, 2014, **20**, 4896; (c) K. Shibata and N. Chatani, *Org. Lett.*, 2014, **16**, 5148; (d) X. Qin, X. Li, Q. Huang, H. Liu, D. Wu, Q. Guo, J. Lan, R. Wang and J. You, *Angew. Chem., Int. Ed.*, 2015, **54**, 7167; (e) M. Corbet and F. D. Campo, *Angew. Chem., Int. Ed.*, 2013, **52**, 9896; (f) G. Rouquet and N. Chatani, *Chem. Sci.*, 2013, **4**, 2201; (g) J. Fernández-Salas, S. Manzini, L. Piola, A. M. Z. Slawin and S. P. Nolan, *Chem. Commun.*, 2014, **50**, 6782; (h) V. S. Thirunavukkarasu, S. I. Kozhushkov and L. Ackermann, *Chem. Commun.*, 2014, **50**, 29; (i) P. Becker, D. L. Priebbenow, R. Pirwerdjan and C. Bolm, *Angew. Chem., Int. Ed.* 2014, **53**, 269; (j) P. Lennartz, G. Raabe, C. Bolm, *Adv. Synth. Catal.*, 2012, **354**, 3237; (k) C.-G. Feng, M. Ye, K.-J. Xiao, S. Li and J.-Q. Yu, *J. Am. Chem. Soc.*, 2013, **135**, 9322; (l) P. Sadhu, S. K. Alla and T. Punniyamurthy, *J. Org. Chem.*, 2013, **78**, 6104; (m) N. Dastbaravardeh, T. Toba, M. E. Farmer and J.-Q. Yu, *J. Am. Chem. Soc.*, 2015, **137**, 9877.
- For some examples on directed *C-N* bond formation, see: (a) Q. Shuai, G. Deng, Z. Chua, S. Bohle and C.-J. Li, *Adv. Synth. Catal.*, 2010, **352**, 632; (b) A. John and K. M. Nicholas, *J. Org. Chem.*, 2011, **76**, 4158; (c) M. Shang, S.-H. Zeng, S.-Z. Sun, H.-X. Dai and J.-Q. Yu, *Org. Lett.*, 2013, **15**, 5286; (d) L. D. Tran, J. Roane and O. Daugulis, *Angew. Chem., Int. Ed.*, 2013, **52**, 6043; (e) L. Wang, D. L. Priebbenow, W. Dong and C. Bolm, *Org. Lett.*, 2014, **16**, 2661; (f) A. M. Martinez, N. Rodriguez, R. G. Arrayás and J. C. Carretero, *Chem. Commun.*, 2014, **50**, 2801; (g) Q. Li, S.-Y. Zhang, G. He, Z. Ai, W. A. Nack and G. Chen, *Org. Lett.*, 2014, **16**, 1764; (h) T. Matsubara, S. Asako, L. Ilies and E. Nakamura, *J. Am. Chem. Soc.*, 2014, **136**, 646; (i) M. Shang, S.-Z. Sun, H.-X. Dai and J.-Q. Yu, *J. Am. Chem. Soc.*, 2014, **136**, 3354; (j) D. Mahesh, P. Sadhu and T. Punniyamurthy, *J. Org. Chem.* 2015, **80**, 1644; (k) Q. Yan, Z. Chen, W. Yu, H. Yin, Z. Liu and Y. Zhang, *Org. Lett.*, 2015, **17**, 2482.
- For some examples, see: (a) G. Greco, E. Novellino, I. Fiorini, V. Nacci, G. Campiani, S. M. Ciani, A. Garofalo, P. Bernasconi and T. Mennini, *J. Med. Chem.*, 1994, **37**, 4100; (b) C. Fischer and B. Koenig, *Beilstein J. Org. Chem.*, 2011, **7**, 59; (c) T. Balle, J. Perregaard, M. T. Ramirez, A. K. Larsen, K. W. Sjøby, T. Liljefors and K. Andersen, *J. Med. Chem.*, 2003, **46**, 265; (d) Y. Wu, Y. Li, S. Gardner and B. S. Ong, *J. Am. Chem. Soc.*, 2005, **127**, 614; (e) M. M. M. Gee, S. Gemma, S. Butini, A. Ramunno, D. M. Zisterer, C. Fattorusso, B. Catalanotti, G. Kukreja, I. Fiorini, C. Pisano, C. Cucco, E. Novellino, V. Nacci, D. C. Williams and G. Campiani, *J. Med. Chem.*, 2005, **48**, 4367; (f) Y. Liu, M. Nishiura, Y. Wang and Z. Hou, *J. Am. Chem. Soc.*, 2006, **128**, 5592.
- For some examples, see: (a) S. Rault, M. C. D. Sévricourt, A. M. Godard and M. Robba, *Tetrahedron Lett.*, 1985, **26**, 2305; (b) H. Strobel, P. Wohlfart, A. Safarova, A. Walsler, T. Suzuki and R. M. Dharanipragada, *PCT. Int. Appl.*, WO 2002064545 A1, 2002; (c) K. Burri, J. Hoffner, K. Islam and S. Mukhija, *PCT. Int. Appl.*, WO 2002070464 A3, 2002; (d) G. Yang, H. Alan, P. John, P. Wallace, T. Andrew, Y. Taeyoung and Z. He, *PCT. Int. Appl.*, WO 2003082826 A1, 2003; (e) S. I. Druzhinin, S. A. Kovalenko, T. A. Senyushkina, A. Demeter, R. Januskevicius, P. Mayer, D. Stalke, R. Machinek and K. A. Zachariasse, *J. Phys. Chem. A*, 2009, **113**, 9304; (f) F. Aiello, A. Garofalo and F. Grande, *Tetrahedron Lett.*, 2010, **51**, 6635; (g) W. K. De, L. J. R. M. Maes, L. Meerpoel, *PCT. Int. Appl.*, WO 2012084804 A1, 2012; (h) F. Grande, A. Brizzi, A. Garofalo and F. Aiello, *Tetrahedron*, 2013, **69**, 9951.
- For selected review, see: F. Monnier and M. Taillefer, *Angew. Chem., Int. Ed.*, 2009, **48**, 6954.
- For examples see: (a) J. Hassan, M. Sevignon, C. Gozzi, E. Schulz and M. Lemaire, *Chem. Rev.*, 2002, **102**, 1359; (b) S. V. Ley and A. W. Thomas, *Angew. Chem., Int. Ed.*, 2003, **42**, 5400.
- For examples using aryl iodide, see: (a) D. W. Old, M. C. Harris and S. L. Buchwald, *Org. Lett.*, 2000, **2**, 1403; (b) J. C. Antilla, J. M. Baskin, T. E. Barder and S. L. Buchwald, *J. Org. Chem.*, 2004, **69**, 5578; (c) L. Rout, S. Jammi and T. Punniyamurthy, *Org. Lett.*, 2007, **9**, 3397; (d) D. T. Ziegler, J. Choi, J. M. Muñoz-Molina, A. C. Bissember, J. C. Peters and G. C. Fu, *J. Am. Chem. Soc.*, 2013, **135**, 13107.
- For some examples using aryl boronic acid, see: (a) P. Y. S. Lam, C. G. Clark, S. Saubern, J. Adams, M. P. Winters, D. M. T. Chan and A. Combs, *Tetrahedron Lett.*, 1998, **39**, 2941; (b) S. Yu, J. Saenz and K. Srirangam, *J. Org. Chem.*, 2002, **67**, 1699.

- 10 For example using organobismuth reagents, see: P. Petiot, J. Dansereau and A. Ganon, *RSC Adv.*, 2014, **4**, 22255.
- 11 For example using aryl lead, see: P. López-Alvarado, C. Avendaño and J. C. Menéndez, *J. Org. Chem.*, 1995, **60**, 5678.
- 12 M.-L. Louillat, A. Biafora, F. Legros and F. W. Patureau, *Angew. Chem., Int. Ed.*, 2014, **53**, 3505.
- 13 For some examples on the use of first row transition metals, see: (a) X. Chen, X.-S. Hao, C. E. Goodhue and J.-Q. Yu, *J. Am. Chem. Soc.*, 2006, **128**, 6790; (b) L. D. Tran, I. Popov and O. Daugulis, *J. Am. Chem. Soc.*, 2012, **134**, 18237; (c) A. M. Suess, M. Z. Ertem, C. J. Cramer and S. S. Stahl, *J. Am. Chem. Soc.*, 2013, **135**, 9797; (d) M. Corbet and F. D. Campo, *Angew. Chem., Int. Ed.*, 2013, **52**, 9896; (e) T. Truong, K. Klimovica and O. Daugulis, *J. Am. Chem. Soc.*, 2013, **135**, 9342; (f) M. Shang, H.-L. Wang, S.-Z. Sun, H.-X. Dai and J.-Q. Yu, *J. Am. Chem. Soc.*, 2014, **136**, 11590; (g) S. Wang, R. Guo, G. Wang, S.-Y. Chen and X.-Q. Yu, *Chem. Commun.*, 2014, **50**, 12718; (h) J. Liu, L. Yu, S. Zhuang, Q. Gui, X. Chen, W. Wang and Z. Tan, *Chem. Commun.*, 2015, **51**, 6418; (i) E. R. Fruchey, B. M. Monks and S. P. Cook, *J. Am. Chem. Soc.*, 2014, **136**, 13130; (j) J. Li and L. Ackermann, *Angew. Chem., Int. Ed.*, 2015, **54**, 8551.
- 14 For some examples, see: (a) W. Zhu, D. Zhang, N. Yang and H. Liu, *Chem. Commun.*, 2014, **50**, 10634; (b) P.-F. Larsson, C.-J. Wallentin and P.-O. Norrby, *ChemCatChem*, 2014, **6**, 1277.

Graphical Abstract



Copper-mediated regioselective *N*-arylation of pyrroles, indoles, pyrazoles and carbazole is described using 8-aminoquinoline amide as a directing group via dehydrogenative coupling. The protocol has broad substrate scope with good yields at moderate temperature.