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Transient transmembrane secretion of H₂O₂: A mechanism for citral-caused inhibition of aflatoxin production from *Aspergillus flavus*

Accepted 00th January 20xx DOI: 10.1039/x0xx00000x

Received 00th January 20xx,

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A polydopamine- Fe_3O_4 nanocomposite-based H_2O_2 electrochemical sensor is fabricated to real-time monitor the transmembrane release of reactive oxygen species from citraltreated *Aspergillus flavus*, revealing a mechanism involving transient transmembrane secretion of H_2O_2 for citral-caused inhibition of aflatoxin production from fungus for the first time.

Aflatoxins are a group of highly toxic and carcinogenic metabolites that are produced by Aspergillus flavus and A. parasiticus.¹ A low quantity (parts per billion) of aflatoxins can lead to significantly negative impacts on human and animal health.² Aflatoxin-producing fungi are widely spread in nature and can contaminate food and feed at each procedure of grain production.³ How to reduce the generation of aflatoxins from the fungi is one of the hottest research topics in microbiology. The use of natural essential oils for the protection of grains has been regarded as a facile, safe and effective way to restrain the fungi growth and aflatoxin production.⁴ As a typical natural essential oil presented in various plants, citral shows strong inhibitory effects on aflatoxin production and is proposed to be a very promising substance for the control of aflatoxins in grains.⁵ However, existing studies are limited to the dose-response of citral and aflatoxin production. The mechanism for citral-induced inhibition on aflatoxin biosynthesis in fungus is still obscure.

Reactive oxygen species (ROS) such as superoxide, hydrogen peroxide (H_2O_2) and hydroxyl radical are continuously generated during cellular metabolic processes. They act as signaling molecules in the maintenance of cellular normal physiological functions.⁶ The involvement of ROS in the generation of aflatoxins has also been

⁺ Footnotes relating to the title and/or authors should appear here. Electronic Supplementary Information (ESI) available: [Experimental details, materials characterization data and electrochemical results]. See DOI: 10.1039/x0xx00000x reported in previous research works.⁷ The enhancement of oxidative stress may result in the production of aflatoxins from toxigenic strain.⁸ Furthermore, the application of antioxidants could repress the biosynthesis of aflatoxins from *Aspergillus* species.⁹ Thus, ROS could play an important role in citral-caused inhibition on aflatoxin production.

 $\rm H_2O_2$ is one of the most stable ROS with a half-life of several hours, which could diffuse across membranes through water channels.¹⁰ In recent years, electrochemical sensors have been employed to real-time monitor $\rm H_2O_2$ in biological systems due to their simplicity, fast response, high sensitivity and low detection limit.¹¹ In the present work, we fabricated a polydopamine (PDA)-Fe₃O₄ nanocomposite based non-enzymatic H₂O₂ sensor to real-time monitor the H₂O₂ release from citral-treated *A. flavus* cells. The intracellular oxidative stress, morphological changes and the aflatoxin B₁ (AFB₁) production of *A. flavus* mycelia were also measured to reveal the role of H₂O₂ in citral-induced aflatoxin suppression.

Fe₃O₄ nanoparticles (NPs) were decorated on PDA microspheres via a facile in situ co-precipitation method and the as-prepared nanocomposites were characterized with transmission electron microscopy (TEM) and scanning electron microscopy (SEM) (Fig. 1). In comparison to the smooth surface of the PDA microspheres (Fig. S1 in ESI), a lot of small dots with the size less than 10 nm are discovered, suggesting the successful growth of Fe₃O₄ NPs in the nanocomposites. The average size of the PDA-Fe₃O₄ nanocomposite is ~500 nm. The surface attached NPs are of single-crystalline structures with lattice spacing distances of 0.25 and 0.30 nm, corresponding to the (311) and (220) planes of the face-centered cubic Fe₃O₄, respectively. The excellent crystalline structure (Fig. S2 in ESI) of the NPs could guarantee the electrocatalytic activity of the nanocomposites. The abundant surface chemical groups of PDA render the nanocomposites hydrophilic for the easy access of H₂O₂ molecules in the following sensing applications (Fig. S3 in ESI).

PDA-Fe₃O₄ nanocomposites were modified on a glassy carbon electrode (GCE) to fabricate an electrochemical sensor towards H₂O₂ reduction. Cyclic voltammetry (CV) was employed to characterize the catalytic behavior of the sensor over a potential ranging from -0.9 to +0.2 V at a scan rate of 50 mV/s in pH 7.4

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phosphate buffer saline (PBS) with bare GCE, PDA microspheresimmobilized GCE (PDA/GCE) and Fe₃O₄ NPs-functionalized GCE (Fe₃O₄/GCE) for comparisons (Fig. 1D). The nanocomposite-base electrode shows a well-defined reduction peak with the highest current intensity in a 0.01 M PBS (pH 7.4) containing 1.0 mM H₂O₂, verifying the catalytic activity of Fe_3O_4 NPs to the reduction of H_2O_2 . Fig. 1E illustrates a typical amperometric response of the sensor on successive addition of H₂O₂. The catalytic currents reach the steadystate value (95% of the maximum) within 5 s after each $\rm H_2O_2$ injection. The fast response of the sensor can meet the requirements of cell-released H_2O_2 measurement. The currents show a linear response against the H₂O₂ concentration in the range of 0.5 to 6.0 μ M (R²=0.995) with a sensitivity of 51.06 μ A/mM and a detection limit of 4.9×10^{-8} M (S/N = 3). In comparison with the reported non-enzymatic H_2O_2 sensors, the PDA-Fe₃O₄ based one has a comparable sensitivity and a lower detection limit (Table S1 in ESI). Inset II of Fig. 1E illustrates the great selectivity of the sensor. The hydrophilicity of PDA and the well-dispersibility of Fe₃O₄ NPs in the nanocomposites may contribute to the excellent performance of the sensor. A micromolar level of steady-state extracellular H_2O_2 could be generated in a biological system with the cell density of $10^5/\textrm{mL}.^{12}$ The results in this assay confirm that the PDA-Fe $_3O_4$ nanocomposite-based H₂O₂ sensor can be applied to monitor the secretion of H₂O₂ in an A. *flavus* mycelia suspension system.



Fig. 1 (A) TEM image of the PDA-Fe₃O₄ nanocomposites; (B) SEM image of the PDA-Fe₃O₄ nanocomposites; (C) HRTEM image of the Fe₃O₄ nanoparticles on the PDA-Fe₃O₄ nanocomposites; (D) CV of bare GCE (a), PDA/GCE (b), Fe₃O₄/GCE (c) and PDA-Fe₃O₄-Nafion/GCE (d) in a 0.01 M PBS buffer (pH 7.4) containing 1.0 mM H₂O₂ with a scan rate of 50 mVs⁻¹; (E) Amperometric response of PDA-Fe₃O₄-Nafion/GCE on successive injection of H₂O₂ into a 0.01 M PBS buffer (pH 7.4). Applied potential: -0.3 V. Inset I: plot of H₂O₂ peak current versus H₂O₂ concentration. Inset II: Interfering effect of ascobic acid (AA), glucose (Glu), dopamine (DA) and uric acid (UA) on the performance of PDA-Fe₃O₄-Nafion/GCE.

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To verify the role of ROS in citral-induced suppression of aflatoxin production, the secretion of H2O2 from A. flavus mycelia with/without citral treatment was real-time measured using the asprepared electrochemical sensor. In our investigation, it is found that the amount of aflatoxins in the conidia-inoculated culture medium explosively grows in the incubation time range of 60 to 108 h(Table S2 in ESI). Thus, mycelia after 60 h of culture were utilized in the following experiments. As being exhibited in curve (a) (Fig. 2), the addition of 300 ppm citral in the PBS without A. flavus mycelia does not cause any electrochemical response. 30% glycerin, the solvent of citral, also has no significant impact on the current in a mycelia suspension (curve b of Fig. 2). The data rule out the interference of stimulator in the electrochemical detection at the potential of -0.3 V versus Ag/AgCl. A remarkable enhancement of the reduction current is obtained in a PBS (0.01 M, pH 7.4) containing mycelia after the injection of 300 ppm citral (curve c in Fig. 2). Moreover, the introduction of catalase (100 U mL⁻¹), a selective scavenger of H_2O_2 , can induce the recovery of the current back to the background level, indicating that the increase of current is ascribed to citral-induced secretion of H_2O_2 from the mycelia. According to the calibration curve in Fig. 1E, the amount of the released H₂O₂ is calculated as 304.4 nM (the dry weight of mycelia is 0.2744 g). The data clearly show that citral-treatment could trigger the release of H_2O_2 from A. flavus cells. To the best of our knowledge, it is the first time to real-time monitor the section of H₂O₂ from fungi cells. It has been reported that citral can cause significant damage to cell membrane and cell wall of microbes.¹³ Thus, the citral treatment may also destroy the cell membrane and cell wall structure of A. flavus mycelia, consequently resulting in the H_2O_2 secretion.



Fig. 2 (a) Amperometric response of the sensor in a PBS solution without *A. flavus* cells under citral injection; (b) Amperometric response of the sensor in a PBS solution containing *A. flavus* cells under injection of glycerine, the solvent for citral; (c) Amperometric response of the sensor in a PBS solution containing *A. flavus* cells treated with citral (300 ppm), followed by a catalase injection (100 U mL⁻¹).

The secretion of H_2O_2 may cause the decrease of intracellular ROS level, further leading to the alleviation of oxidative stress in the mycelia. Since the formation of oxidative stress is a pre-requisite for the biosynthesis of aflatoxins,⁸ it is of great importance to evaluate the intracellular oxidative stress after the citral-treatment. Dichlorodihydro-fluorescein diacetate (DCFH-DA) assay was employed to

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show the intracellular oxidative stress in the present study (Fig. 3A and 3B). The glycerine-treated mycelia possess high intensity of green fluorescence, but the citral-treated ones demonstrate less fluorescence emission. The results indicate that the ROS level in the former samples is much higher than that in the latter ones. Therefore, the citral-induced section of H_2O_2 may reduce the amount of ROS in the mycelia, subsequently resulting in the alleviation of oxidative stress. Interestingly, it is observed that the average diameter of the citral-treated mycelia is smaller than that of glycerine-exposed ones (Fig. 3C and 3D). A lot of wrinkles can also be found on the surface of citral-treated mycelia. The findings prove the occurrence of the cellular secretion process and may suggest the co-release of other intracellular substances like H₂O molecules together with H₂O₂. Vacuoles in A. flavus cells are correlated to the biosynthesis of aflatoxins,¹⁴ thus the citral-caused secretion may also possibly inhibit the aflatoxin production via the change of vacuole volume.



Fig. 3 Effects of citral treatment on the intracellular ROS level and surface morphology of mycelia. Fluorescent micrographs of the mycelia treated with glycerine (A) and citral (B), respectively; SEM images of the mycelia treated with glycerine (C) and citral (D), respectively.

After citral- or glycerine-treatment the A. flavus mycelia were harvested and cultured in a freshly prepared medium subsequently. The amount of AFB₁ in the culture medium was measured after 0, 24 and 48 h of incubation at 30 °C to evaluate the effects of citral on the production of aflatoxins (Table 1). Initially, the concentration of AFB₁ in both glycerine- and citral-treated groups is around 200 ng mL⁻¹, which may be attributed to desorption of mycelium surfaceadsorbed aflatoxin molecules. After 24 h of incubation only 8.2 \pm 3.5 ng mL⁻¹ of AFB₁ increment is detected in the citral group. However, the enhancement of AFB₁ concentration in the glycerine samples is 32.8 ± 4.9 ng mL⁻¹. From 24 to 48 h a sharp increase on the concentration of AFB₁ occurs in the glycerine-exposed group. The accumulated AFB₁ concentration reaches to 1689.4 \pm 2.4 ng mL^{-1} with an increment of 1439.9 ± 4.9 ng mL^{-1} . The high efficient biosynthesis of aflatoxin after 24 h of culture may lead to the tremendous growth of AFB₁ concentration. In contrast, the citraltreated samples show an enhancement of $145.4 \pm 1.0 \text{ ng mL}^{-1}$ on the AFB1 level, which is around 10 times lesser than the one observed in the glycerine group. The results strongly support that a

short-term exposure of *A. flavus* mycelia to citral could effectively diminish the generation of aflatoxins. The amount of AFB_1 released from the citral-treated mycelia may continuously increase after 48 h. However, its growth should be much slower than that of glycerine-exposed ones. The influence of long-term citral treatment on the production of aflatoxin has been well documented in the literatures.¹⁵ It is the first time to report the effective control of aflatoxins via a short-term citral exposure, which undoubtedly should be a safer and more economic approach. However, it should be noted that the treatment timing is very critical for the outcome of this strategy. Based on our data, the citral should be applied at least 24 h before the explosive production of aflatoxins.

Table 1 Generation of AFB_1 from the glycerine- and citral-treated mycelia

Time	Glycerine		Citral	
(h)	AFB_1	ΔAFB_1	AFB_1	ΔAFB_1
	(ng/mL)	(ng/mL)	(ng/mL)	(ng/mL)
0	216.6±2.3	-	196.2±2.0	-
24	249.5±2.5	32.8±4.9	204.4±1.5**	8.2±3.5**
48	1689.4±2.4	1439.9±4.9	349.8±2.4**	145.4±1.0**

**: p<0.01, compared with the glycerine-treated mycelia.

The short-term citral treatment caused inhibition of AFB_1 production in *A. flavus* mycelia could be explained as the follows. Normally, after certain duration of culture ROS gradually accumulate in the premature mycelia to trigger a series of biochemical reactions for the aflatoxin biosynthesis. However, upon the addition of citral large amount of ROS is secreted from the mycelia, further resulting in the decrease of intracellular ROS level. The failure formation of oxidative stress interrupts the signaling pathway to the aflatoxin biosynthesis. Therefore, the production of aflatoxins from citral-incubated *A. flavus* mycelia is inhibited significantly. More works are still carrying out in the authors' laboratory to elucidate the detailed pathway.



Fig. 4 Mechanism for citral-caused inhibition of aflatoxin production in *A. flavus*.

In summary, a PDA-Fe₃O₄ nanocomposite-based highly sensitive electrochemical sensor was fabricated for real-time monitoring the secretion of H_2O_2 molecules from *A. flavus* mycelia after short-time treatment with citral. The real-time measurement results associated with the intracellular ROS level and morphological changes of the mycelia prove the transit transmembrane release of

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 H_2O_2 in the citral exposure process. The remarkable inhibition effect of citral on AFB₁ production was also demonstrated by culturing the treated mycelia for another 48 h. A mechanism involving the transient secretion of H_2O_2 from the mycelia is proposed to explain citral-caused inhibition of aflatoxin production from A. flavus.

This work is financially supported by National Program on Key Basic Research Project of China (973 Program) under contract No.2013CB127800 and National Natural Science Foundation of China (No. 21205097 and No. 31200700). Z. S. Lu would like to thank the supports by Fundamental Research Funds for the Central Universities (XDJK2015B016) and Young Core Teacher Program of the Municipal Higher Educational Institution of Chongqing.

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