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Access to side-chain carbon information in deuterated solids under fast MAS through non-rotor-synchronized mixing

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Natalia Kulminskaya(a)#, Suresh Kumar Vasa(a)#, Karin Giller(a), Stefan Becker(a), Ann Kwan(b), Margaret Sunde(b), Rasmus Linser(a)*

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We demonstrate the accessibility of aliphatic ¹³C side chain chemical shift sets for solid-state NMR despite perdeuteration and fast MAS using isotropic, non-rotor-synchronized ¹³C-¹³C mixing. Combined with amide proton detection, we unambiguously and sensitively detect whole side chain to backbone correlations for two proteins using around 1 mg of sample.

A fundamental step for analysis of structure and dynamics in proteins by NMR spectroscopy is resonance assignment for the different nuclei. Countless types of multidimensional NMR experiments have been developed for isotopically labeled proteins, typically with the focus on ¹³C and ¹⁵N nuclei in the solid-state. Backbone chemical shifts deliver important information about the secondary structure, molecular packing of the protein, mobility and many other parameters. Side chain chemical shifts, on the other hand, provide equally important information specific for the residue type and play an essential role in reporting amino acid interactions with neighboring amino acids, other proteins, small molecules, and lipids or water. Moreover, side-chain chemical shifts are also sensitive to conformational changes of the protein. In particular, sidechain dihedral angles χ_i can be defined from chemical shifts and can be used in structure calculation of the protein.¹⁻³ Exploitation of carbon side chain chemical shifts has proved fundamental in solid-state NMR.⁴⁻⁶ One of the crucial building blocks of such experiments is the mixing of magnetization among side chain nuclei, which is elicited mostly by protondriven spin diffusion PDSD⁷, dipolar-assisted rotational resonance mixing DARR⁸, or other schemes exploiting the strong proton-dipolar coupling network. Such mixing schemes are crucial for residue type identification in the course of protein assignment and for structural information from side chain carbon-carbon through-space correlations.

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Proton-detected solid-state NMR on the other hand has been rapidly expanding approach having grown to a worldwide standard recently.⁹⁻¹³ In conjunction with extensive ²H labeling (perdeuteration), proton-detected solid-state NMR now e... bles solution NMR-like correlations¹⁴⁻¹⁶. In presence of "ultr. fast" Magic-Angle Spinning (using 1.3 mm rotors or smaller) and 100% back-exchange of labile protons, proton-detecte solid-state NMR represents an eight-fold increase in sensitivi v per amount of protein compared with ¹³C detection.¹⁷ The sensitivity advantages are even increased in combination with paramagnetic relaxation enhancement, which can overcome slow ¹H and ¹³C longitudinal T_1 relaxation and enable fast rec cling of the experiments in the absence of high-power decoupling.^{18, 19} Thus, proton detection in combination with ultr fast MAS becomes especially useful for proteins which are difficult to produce in large amounts.²⁰⁻²²

Unfortunately, the price to pay for overcoming the hurdles of a strong proton-proton dipolar-coupling network by fast spir ning and particularly by deuteration is the inaccessibility "proton-dependent" conventional mixing schemes like DARR,⁸ PDSD,⁷ or CHHC.²³ Utilization of deuterons has been suggested. as a potential remedy^{24, 25} if an additional (fourth) deuteriu. channel is available. Radio-frequency-driven recoupling RFDR,²⁶ HORROR,²⁷ and symmetry-based sequences²⁸ can be an option for replacing PDSD and DARR experiments. Howeve, under fast MAS, such sequences can face limitations in tern of maximum radio frequency power that can be tolerated L_{i} the sample. This becomes increasingly problematic as spinnir, speeds are increased. Thus, the mentioned elements tend ι be limited to short mixing times, which significantly reduces the available options for getting the long-range correlations necessary for identifying side chain resonance sets. As demon strated by studies employing low-power variants of the esta' lished recoupling sequences,²⁹⁻³² there is a growing incentiv to find homonuclear mixing schemes amenable for the rapid' increasing MAS rates.

Here we demonstrate the effectiveness of an isotropic home nuclear mixing scheme from solution NMR, MOCCA,³³⁻³⁵ for amide proton-detected NMR approaches at sample spinning of

^{a.} Max-Planck Institute for Biophysical Chemistry, Department NMR-Based Structural Biology, Am Fassberg 11, 37077 Göttingen

^{b.} School of Medical Sciences and School of Molecular Bioscience, University of Sydney, Sydney, Australia

[#] Authors contributed equally to this work.

^{*} Corresponding author: rali@nmr.mpibpc.mpg.de

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55 kHz. We have previously demonstrated the feasibility of MOCCA mixing for traditional 2D carbon-carbon correlations for partially (30%) back-exchanged, deuterated proteins at moderate spinning speed.³⁶



Isotropic ¹³C-¹³C mixing in the context of proton-detected NMR at "ultra-fast" spinning (rotor diameters of 1.3 mm and smaller) enables us to obtain proton-detected side-chain carbon spectra on uniformly ¹⁵N, ¹³C, ²H-labeled proteins backexchanged in 100% H₂O:¹⁷ For methods development, we employed the SH3 domain of α -spectrin.⁴ We then applied the approach to obtain side chain assignments from functional amyloid rodlets composed of a fungal hydrophobin protein³⁷⁻ ³⁹. Hydrophobin rodlets are characterized by a structured core together with substantial sample heterogeneity.⁴⁰ In the absence of a dipolar-coupling network and high magnetic field at ultra-fast MAS, the solution NMR sequence proves capable of establishing connectivites between the protein backbone and the entire sets of side chain carbon resonances in 3D CCCANHtype experiments, without compromising sample integrity by sample heating. Although heterogeneous samples usually tend to require 4D and 5D experimental setup to overcome the resonance overlap, here near-complete aliphatic side-chain resonance assignments could be obtained using a single 3D CCCANH experiment.

First, we tailored MOCCA conditions and optimized the CO-PORADE approach⁴¹ for ultra-fast MAS. Secondly, we used MOCCA as a building block for 3D CCCANH experiments to obtain residue-specific side-chain chemical-shift information. Depending on the sample-dependent doping characteristics, see below, we could use either one mixing block (before frequency encoding, Fig. 1A) or two mixing blocks (before and after frequency encoding, Fig. 1B) to access intra-residual connectivites in the absence of side-chain protons. MOCCA mixing involves optimization of two parameters, ... 180° pulse length and the delay between the pulses (Fig. 1 and S1. First, we optimized the 180° pulse length using RFD ' conditions without phase cycling the RFDR pulse. (See the SI for details.) In the next step, we evaluated the performance of MOCCA at different mixing times and compared with RFDR an example of a well-established rotor-synchronized recoupling sequence as shown for exemplary residues in Fig. 2 and S3 (SI). It is evident from the figure that we could use longer mixing times amounting to similar amounts of energy dissipated into the probe with MOCCA in comparison to RFDR. Although the RFDR buildup behavior is slightly faster than MOCCA, the latter performs well with mixing times as long 7 ms. This enables relayed transfers providing many high intensity cross peaks with improved performance particular, for 2- or 3- bond transfers (Figs. 1A and S3). A comparison of 2D correlation spectra with extensive mixing times for bo 1 RFDR and MOCCA is shown in Fig. S4. Taking into account the feasibility of long mixing times using MOCCA with very low longitudinal magnetization loss (Fig. S5), the approach might be useful for other applications like long-range magnetiza..... transfer through ³J couplings or similar. In contrast, carbonyl and aromatic buildup is less efficient than with RFDR. might be due to the faster buildup for RFDR, which wou 1 mean sufficient mixing even for partial DD introduction in the presence of off-resonance effects. Similar observations a e made with moderate MAS up to 25 kHz with 30% proton backexchanged samples.³⁶



Figure 2. (A). 2D ¹³C-¹³C correlation experiments using MOCCA with the COPORADE approach⁴¹ (see Fig. S10), highlighting aliphatic side-chain carbons of the SH3 domain it 55.5 kHz MAS. The experiment was recorded with a MOCCA isotropic mixing time of σ ms and a total experimental time of 3 hours. (B) 2D experimental optimization of t' delay, Δ , between the 180° degree pulses of MOCCA for efficient multiple-bond transfer. The total experimental time for each 2D experiment was 3 hours. The optim al experimental delays are shown in dark and light green colors. (C) Experimental build- ρ curves of signal intensity for MOCCA (blue) and RFDR (red), adhering to similar limit tions for power dissipation: Examples of ¹³C-¹³C transfer efficiency for three-bond magnetization transfer. The radiofrequency pulse strength for RFDR and MOCCA w s set to 59.5 kHz. The 180-degree pulse was set 8.37 µs and the delay between the pulses, Δ , was set to 61.5 µs and 9.63 µs for MOCCA and RFDR, respectively.

After determining the optimized conditions for MOCCA at fact MAS, we incorporated this mixing sequence into a 3D CCCAN (or (H)CX(CA)NH) experiment in order to assign side-chain resonances in conjunction with proton detection for increase resolution and sensitivity. Here, we used the COPORADE ar proach, which enables both effective backbone and sidechain carbon polarization in deuterated samples, with paramagnet

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doping for fast relaxation and shortened recycle delays. Corresponding 3D sequences are shown in Figs. 1A and B using one or two mixing blocks, respectively. For a crystalline sample like SH3, paramagnetic chelates can travel throughout the protein lattice, and it has been shown that doping very effectively reduces the spin-lattice relaxation times of ¹H and ¹³C.¹⁹ The resulting relaxation enhancement for both nuclei ensures high sensitivity with fast repetition rates. For this kind of samples, abundant side-chain ¹³C polarization even for those carbons that are too far away from H^N to be polarized by H-C CP has been shown to be available in the indirect dimensions with one mixing block.⁴¹ For a 3D (H)CX(CA)NH spectrum recorded on the SH3 domain, representative strips are depicted in ¹³C correlations dispersed by their amide ¹H and ¹⁵N chemical shifts.



Figure 3. Illustrative strip plot from a ¹H-detected 3D (H)CX(CA)NH experiment, yielding side chain to backbone correlations for uniformly-labelled ¹⁵N, ¹³C, ²H SH3 domain (100% proton back exchanged). A 2D HN projection is shown on the bottom. Mixing time applied in the experiment is 32.5 ms.

For many microcrystalline proteins like SH3, good $^{1}\text{H}^{-15}\text{N}^{-1}$ resolution spectra are obtained, which enables an effective dispersion of side chain ¹³C strips. By contrast, as expected it is more challenging to study amyloids, where a low amide resolution is found due to sample heterogeneity (see Fig. S10). Here we applied the experimental approach to hydrophobin rodlets. This protein generates a hydrophobic coating on fungal spores, which improves their wettability characteristics.^{38,} The overall behavior of the rodlet samples is dominated by severe heterogeneity of the sample for both proton- and carbon-detected experiments as shown by Morris et al.⁴⁰ and has so far remained a challenge for solid-state NMR characterization. This can be derived from their amide spectral dispersion and sensitivity (see Fig. S10). In addition, a lower overall accessibility of the protein to the paramagnetic agent due to the packing in the fibrillar, amyloid state¹⁸ than in the crystalline state with solvent channels, give rise to large carbon T_1 values despite identical dopant concentrations. Whereas C^{α} can still be sufficiently polarized by H-C CP, the ¹³C Boltzmann polarization required for side chain carbons distant from H^N at the chosen recycle delay is low. However, we could partially alleviate this problem by inserting an additional MOCCA mixing block before chemical shift encoding (Fig. 1B) to mix protopolarized C^{α} magnetization into the side chain before the finct chemical shift evolution. Even though this way we still main tain higher C^{α} peaks, we observe an effective increase of ic chain polarization by initial CA-to-CX mixing.

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Apart from access to single side chain chemical shifts, the value of the access to aliphatic carbon resonance sets by MOCCA mixing is apparent from the spectra obtained on the heterogeneous hydrophobin sample. Fig. 4 shows strips from the 3D (H)CX(CA)NH NMR experiment, effectively overcoming the intrinsic sample heterogeneity in conjunction with the low amount of sample required. Overall, using the side chain the backbone correlation approach, we managed to identify an assign individual contributions of many assigned residues the HN plane by their side-chain carbon chemical shifts without going to higher spectral dimensions. The side chain st i effectively help to deconvolve and unambiguously assign the protein despite a lack of HN resolution.

Side-chain carbon shift accessibility is important for determining hydrophobic contacts within protein structures and with other molecules. Side-chain carbon shifts are also sensitive many factors including tertiary structural contacts, intermolecular contacts, ring currents, as well as rotamer confc mations. Apart from the possibility of resolving overlapping correlations for assignment purposes as shown for this hydrophobin rodlet sample, we thus expect manifold applications arising from the accessibility of side chain carbon chemic... shifts in conjunction with the sensitivity provided from protor detected solid-state NMR studies.



Figure 4. Strip plots from a ¹H-detected 3D (H)CX(CA)NH experiment on 100% L exchanged EAS_{A15} hydrophobin rodlets. All experimental conditions are similar to the SH3 protein, however relying on out-and-back mixing of C^{α} magnetization, hence t. e low cross-peak-to-diagonal peak ratio. Mixing time is 60.7 ms. The experimental till e was around 48 hours. The strip plots of 39Q, 52T, 63A are shown with slightly high contour levels.

We have shown that by using a non-rotor-synchronized carbon-carbon liquid-state mixing sequence based on J-couplines together with low power requirements, we are able to obtain

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amide-resolved proton-detected sidechain-to-backbone correlations despite fast MAS and in the absence of side-chain protons. We demonstrate the sensitive access to side-chain resonances from only 1 mg of protein material for a microcrystalline protein as well as for a functional amyloid known to pose significant hurdles to solid-state NMR due to its inhomogeneity. As a *J*-coupling-based sequence with extremely low duty, MOCCA does not require rotor synchronization, allows long mixing times and multiple-bond transfers without major power dissipation at fast MAS. MOCCA provides access to the sets of side chain carbon shifts from a heterogeneous protein sample in spite of spectral overlap in the HN plane even in three dimensions, strongly facilitating unambiguous resonance assignment and reporting manifold local biophysical parameters.

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References

- 1. Y. Shen and A. Bax, J Biomol NMR, 2013, 56, 227-241.
- 2. R. E. London, B. D. Wingad and G. A. Mueller, *J Am Chem Soc*, 2008, **130**, 11097-11105.
- 3. M. Hong, T. V. Mishanina and S. D. Cady, *J Am Chem Soc*, 2009, **131**, 7806-7816.
- 4. F. Castellani, B.-J. van Rossum, A. Diehl, M. Schubert, K. Rehbein and H. Oschkinat, *Nature*, 2002, **420**, 98–102.
- 5. A. Loquet, N. G. Sgourakis, R. Gupta, K. Giller, D. Riedel, C. Goosmann, C. Griesinger, M. Kolbe, D. Baker, S. Becker and A. Lange, *Nature*, 2012, **486**, 276-279.
- 6. S. H. Park, B. B. Das, F. Casagrande, Y. Tian, H. J. Nothnagel, M. Chu, H. Kiefer, K. Maier, A. A. De Angelis, F. M. Marassi and S. J. Opella, *Nature*, 2012, **491**, 779-783.
- 7. N. Bloembergen, Physica, 1949, 15, 386-426.
- 8. K. Takegoshi, S. Nakamura and T. Terao, *Chem. Phys. Lett.*, 2001, **344**, 631-637.
- 9. M. J. Knight, A. J. Pell, I. Bertini, I. C. Felli, L. Gonnelli, R. Pierattelli, T. Herrmann, L. Emsley and G. Pintacuda, *Proc. Natl. Acad. Sci. U.S.A.*, 2012, **109**, 11095-11100.
- 10. V. Agarwal, S. Penzel, K. Szekely, R. Cadalbert, E. Testori, A. Oss, J. Past, A. Samoson, M. Ernst, A. Böckmann and B. H. Meier, *Angew. Chem. Int. Ed.*, 2014, **53**, 12253-12256.
- 11. P. Ma, J. D. Haller, J. Zajakala, P. Macek, A. C. Sivertsen, D. Willbold, J. Boisbouvier and P. Schanda, *Angew. Chem. Int. Ed.*, 2014, **53**, 4312-4317.
- 12. S. Asami, J. R. Porter, O. F. Lange and B. Reif, *J. Am. Chem. Soc.*, 2015, **137**, 1094-1100.
- 13. D. H. Zhou, J. J. Shea, A. J. Nieuwkoop, W. T. Franks, B. J. Wylie, C. Mullen, D. Sandoz and C. M. Rienstra, *Angew. Chem., Int. Ed.*, 2007, **46**, 8380-8383.

14. R. Linser, U. Fink and B. Reif, J Magn Reson, 2008, 193, 89-93. 15. V. Chevelkov, K. Faelber, A. Diehl, U. Heinemann, H. Oschkinat and B. Reif, J. Biomol. NMR, 2005, 31, 295-310. 16. P. Schanda, M. Huber, R. Verel, M. Ernst and B. H. Meier, Angew. Chem., Int. Ed., 2009, 48, 9322-9325. 17. J. R. Lewandowski, J.-N. Dumez, U. Akbey, S. Lange, L. Emsley and H. Oschkinat, J. Phys. Chem. Lett., 2011, 2, 2205-2211. 18. N. P. Wickramasinghe, S. Parthasarathy, C. R. Jones, C. Bhardwaj, F. Long, M. Kotecha, S. Mehboob, L. W. M. Fung, J. Past, A. Samoson and Y. Ishii, Nature methods, 2009, 6, 215-218. 19. R. Linser, V. Chevelkov, A. Diehl and B. Reif, J. Magn. Reson., 2007, 189, 209-216. 20. J. M. Lamley, D. luga, C. Öster, H.-J. Sass, M. Rogowski, A. Oss, Past, A. Reinhold, S. Grzesiek, A. Samoson and J. R. Lewandowski, J Am. Chem. Soc., 2014, 136, 16800-16806. 21. R. Linser, B. Bardiaux, S. G. Hyberts, A. H. Kwan, V. K. Morris, M. Sunde and G. Wagner, J. Am. Chem. Soc., 2014, 136, 11002-11010 22. S. Wang, S. Parthasarathy, Y. Xiao, Y. Nishiyama, F. Long, I. Matsuda, Y. Endo, T. Nemoto, K. Yamauchi, T. Asakura, M. Takec T. Terauchi, M. Kainosho and Y. Ishii, ChemComm, 2015, DOI: 10.1039/c5cc04618a. 23. A. Lange, K. Seidel, L. Verdier, S. Luca and M. Baldus, J. Am. Chem. Soc., 2003, 125, 12640-12648. 24. U. Akbey, H. Oschkinat and B. J. van Rossum, J. Am. Chem. Soc. 2009, 131, 17054-+. 25. Ü. Akbey, F. Camponeschi, B.-J. van Rossum and H. Oschkinat, ChemPhysChem, 2011, 12, 2092-2096. 26. A. E. Bennett, J. H. Ok, R. G. Griffin and S. Vega, J. Chem. Phys. 1992, **96**, 8624-8627. 27. N. C. Nielsen, F. Creuzet, R. G. Griffin and M. H. Levitt, J. Chem. Phys., 1992, 96, 5668-5677. 28. M. H. Levitt, in Encyclopedia of Nuclear Magnetic Resonance, eds. D. M. Grant and R. K. Harris, 2002, vol. 9, pp. 165-196. 29. G. Teymoori, B. Pahari, E. Viswanathan and M. Edén, Journal o Magnetic Resonance, 2013, 236, 31-40. 30. A. B. Nielsen, S. K. Jain and N. C. Nielsen, Chem. Phys. Lett., 2011, 503, 310-315. 31. C. Herbst, J. Herbst, J. Leppert, O. Ohlenschläger, M. Görlach and R. Ramachandran, J Biomol NMR, 2011, 50, 277-284. 32. P. Bellstedt, C. Herbst, S. Häfner, J. Leppert, M. Görlach and R. R., J. Biomol. NMR, 2012, 54, 325-335. 33. Y. Yoshimura, N. V. Kulminskaya and F. A. Mulder, J Biomol NMR, 2015, 61, 109-121. 34. F. Kramer, W. Peti, C. Griesinger and S. J. Glaser, J Magn Reson 2001, 149, 58-66. 35. I. C. Felli, R. Pierattelli, S. J. Glaser and B. Luy, J Biomol NMR, 2009, 43, 187-196. 36. N. Kulminskaya, S. K. Vasa, K. Giller, S. Becker and R. Linser, J. Biomol. NMR, 2015, DOI: 10.1007/s10858-015-9980-1 37. H. A. B. Wösten, O. M. H. Devries and J. G. H. Wessels, Plant *Cell*, 1993, **5**, 1567–1574. 38. M. Sunde, A. H. Kwan, M. D. Templeton, R. E. Beever and J. P Mackay, Micron, 2008, 7, 773-784. 39. A. H. Y. Kwan, R. D. Winefield, M. Sunde, J. M. Matthews, R. G. Haverkamp, M. D. Templeton and J. P. Mackay, Proc. Natl Acad. Sc U.S.A., 2006, 103, 3621-3626.

40. V. K. Morris, R. Linser, K. L. Wilde, A. P. Duff, M. Sunde and A. H. Kwan, *Angew. Chem., Int. Ed.*, 2012, **51**, 12621-12625.
41. R. Linser, *J. Biomol. NMR*, 2011, **51**, 221-226.

4 | *J. Name.*, 2012, **00**, 1-3

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