ChemComm

Accepted Manuscript

This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](http://www.rsc.org/Publishing/Journals/guidelines/AuthorGuidelines/JournalPolicy/accepted_manuscripts.asp).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](http://www.rsc.org/help/termsconditions.asp) and the Ethical quidelines still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

www.rsc.org/chemcomm

Journal Name

AL SOCI
C**hemis**

COMMUNICATION

Asymmetric Alkynylation/Hydrothiolation Cascade: Enantioselec tive Synthesis of Thiazolidine-2-imines from Imine, Acetylene and Isothiocyanate

Received 00th January 20xx, Accepted 00th January 20xx DOI: 10.1039/x0xx00000x

Alok Ranjan,^a Anupam Mandal,^a Swapnil G. Yerande,^b Dattatraya H. Dethe,*^a

www.rsc.org/

Multicomponent reaction amongst imine, terminal alkyne, and isothiocyanate in presence of catalytic chiral copper-pybox complex proceedes enantioselectively to give enantiopure thiazolidine-2-imine (60-99% *ee***) by highly regioselective intramolecular 5-***exo***-***dig* **hydrothiolation reaction.**

Azolines are five membered nitrogen atom containing heterocycles belonging to many natural products and pharmaceuticals.¹ In addition; many enantiopure azolidine derivatives have received attention due to their application as a synthetic intermediates, auxiliaries, ligands and catalysts for asymmetric synthesis. Various thiazolidines exhibit antiinflammatory, 2 antiviral, 3 anticonvulsant and cardiovascular properties.⁴ In 2013 Punniyamurthy and Sengoden⁵ reported the on water iron catalysed cycloaddition of aziridines with heterocummules for the synthesis of azolidines in racemic form. In 2007 Kwon and coworkers⁶ reported an elegant asymmetric synthesis of azoline derivatives using bisphosphine-catalyzed mixed double-Michael reactions. Despite these advances for synthesis of azoline derivatives, methods for enantioselective synthesis of thiazolidine derivatives, are scarce.

In 2011 Toste *et al.* reported a beautiful monophosphine gold (I) catalyzed enantioselective synthesis of oxazolidines by multicomponent one pot transformation (Scheme 1A).⁷ Singh *et. al.* reported an unprecedented Cu(I)–pybox-diPh-catalyzed highly enantioselective alkynylation/lactamization cascade (up to > 99% *ee*) (Scheme 1B) for the synthesis of diversely substituted isoindolinones and tetrahydroisoquinoline.⁸ Besides drawing inspiration from the work of Toste *et al.*⁷ and Singh *et. al.⁸* we also took a clue from our recently reported synthesis of imidazolidine-2-thione through base catalyzed intramolecular 5-*exo-dig* hydroamination of the

Scheme 1 Current state of the art and proposed asymmetric alkynylati n/hydrothiolation cascade.

propargylthiourea.⁹ It was envisioned that the enantiopure thiazolidine-2-imine can also be synthesised by an asymmetric alkynylation/hydrothiolation cascade from imine, acetylene and isothiocyanate (Scheme 1C). We hypothesized that in this multicomponent reaction first the alkyne will add on to the chiral copper-pybox catalyst activated iminium ion to general propargylamine *in situ,* the propargylamine thus formed will react with isothiocyanate in addition/intramolecular hydrothiolation sequence where isothiocyanate will play a du role as an electrophile and nucleophile to construct thiazolidine-2-imine.

We started our investigation by optimizing conditions for the one-pot preparation of racemic thiazolidine-2-imine 4. In typical experiment, solvent (3 mL), imine **2a** (0.50 mmol), phenyl acetylene 1**a** (0.65 mmol), benzyl isothiocyanate **3a** (0.55 mmol) and catalyst (10 mol %) were combined in a 20 \ldots vial at room temperature (30 °C) equipped with a magnetic still bar and a screw cap. The resultant mixture was stirred at room temperature for 12 h. When this reaction was carried out using 10 mol % triflic acid, the reaction proceeded smoothly to provide the corresponding hydrothiolation product 4a, albe t in 45% yield. Surprisingly hydroamination product **5a** was not observed (Table 1, entry 1). When the solvent was change $\mathfrak t$ from toluene to dichloromethane yield of **4a** improved up to **ChemCommunication**
 ChemCommunication
 ChemCommunication
 Chemcommunication
 Chemcommunication
 Chemcommunication
 Chemcommunication
 Chemcommunication
 Chemcommunication

CIF or other electronic format see DOI: 10.1039/x0xx00000x

a.Department of Chemistry, Indian Institute of Technology Kanpur-208016, Uttar Pradesh, India.

b.Acoris Research (A Division of Hikal Ltd). 3A, International Biotech Park, Hinjewadi, Pune 411 057, Maharashtra, India.

c. Electronic supplementary information (ESI) available: Experimental procedures and characterization of all new compounds. For ESI and crystallographic data in

Table 1 Optimization of Reaction Conditions for the One Pot Formation of Thiazolidine-2-iminea

70% (Table 1, entry 3). Desired product **4a** was also formed in moderate yield when 10 mol % FeCl₃ was used as a catalyst (Table-1 entries 2, 4 and 5). Conversely, we detected the formation of the hydroamination product **5a** along with the hydrothiolation product $4a$ when 10 mol % of $Cu(OTf)_{2}$ and $Rh(OAc)_2$ (Table 1, entries 6-8) were used. Lewis acid $In(OTf)_3$ and Fe(OTf) $_3$ were found proficient in carrying out this transformation (Table 1 entries 9 and 10). Compound **4a** was formed in only 40% yield when copper iodide was used as a catalyst (Table 1, entry 11). After screening various solvent and catalyst combinations, it was found that the use of CuOTf toluene complex and toluene as a solvent leads to the exclusive formation of **4a**.

Scheme 2 Tentative Mechanism of Multicomponent Reaction

The 90% yield obtained of **4a** using CuOTf toluene is a marked improvement compared to yield obtained when CuI was use as a catalyst (Table 1, entry 12 vs. entry 11). The plausibl mechanism for this multicomponent reaction depicted in Scheme 2. Initial coordination of Cu with alkyne forms the πcomplex **6**, further complexation of **6** with imine **2** provides the complex $\overline{7}$, and deprotonation by imine produces copper acetylide with a coordinated iminium ion. An addition reaction leads to propargylamine 9 and regenerates the copper catalyst. Propargylamine **9** is then trapped with isothiocyanate to generate propargylthiourea **10**; the alkyne then coordinates with copper to form the second π-alkyne complex **11**. 5-*exo*dig cyclization by nucleophilic attack of the thiourea sulphur (hydrothiolation) forms the product and regenerates the copper catalyst.

Many points about this multicomponent atom econom. transformation are worthy of mention. First cyclization s generated only the five member thiazolidine product **4** in highly regioselective fashion through 5-exo-dig cyclization, we did not observe the 6-*endo-dig* cyclization product under the optimized condition. Prior to this work we have reported b^{max} catalyzed hydroamination reaction for the construction \sqrt{f} imidazolidine-2-thione **5** from propargylthiourea **10** by 5-*exo*dig cyclization through nucleophilic attack of the thioure nitrogen.⁹ In the present investigations using CuOTf as a catalyst thiazolidine-2-imine 4 was formed exclusively and formation of 5-exo-dig hydroamination product 5, and 6-endodig hydroamination product were not observed at all. We believe that under this optimized reaction condition, *in sit* formed propargylthiourea 10 exist in a thiol tautomeric form the resultant more nucleophilic sulphur attacks the activate. alkyne resulting in the exclusive formation of thiazolidine-2 imine **4. Chemcommand**
 ChemCommand
 ChemCommande
 Chemcommande
 Chemcommande
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcommandee
 Chemcom

Having achieved a proof of concept for our proposed hypothesis, we turned our attention to test the scope of the established protocol. To this end a small library of thiazolid. \sim 2-imine **4** were synthesized and the results obtained are summarized in Table 2. Gratifyingly, the reaction was proved to be very general under the optimized conditions, performin well in all of the cases examined. At first, the scope of the reaction with various alkynes was studied. The aromat alkynes reacted well with variety of imines and isothiocyanate under this condition to afford products 4a-4o, in good t excellent yield ranging from 80-95%. Aliphatic alkynes γ heptyne and 1-hexyne participated effortlessly in this multicomponent reaction and furnished product 4b, 4c and 4 **k.** Not only aliphatic isothiocyanate like allyl isothiocyanate and cyclopropyl isothiocyanate but also substituted phonyl isothiocyanate and substituted benzoyl isothiocyand engaged proficiently in this multicomponent reaction $t²$ furnish compounds **4a-o**.

The structure of thiazolidine-2-imine was unambiguously established by single crystal X-ray analysis of compound **4a.** These hydrothiolation reactions proceed, with a complet control over the stereochemistry. The exocyclic double bond formed was confirmed to be *Z*-isomer by single crystal X-ray analysis of 4a (see supporting information).¹⁰

2 | *J. Name*., 2012, **00**, 1-3 This journal is © The Royal Society of Chemistry 20xx

Journal Name COMMUNICATION

After successfully establishing the reaction conditions for the synthesis of thiazolidines **4**, we embarked on enantioselective variant of this multicomponent reaction. We had significant options with regards to ligand selection for carrying out this transformation in the enantioselective fashion. The mainstream of Cu-catalyzed asymmetric alkynylation reaction developed have utilized quinap¹¹, pinap¹², bis-imine¹³ and

^aAll the reactions were performed with imine (0.50 mmol), acetylene (0.65 mmol) and isothiocyanate (0.55 mmol), 10 mol % CuOTf, solvent (3 mL) for 12 h at room temperature (30 °C).

pybim¹⁴ as ligands. Quest to find out new ligand with aim to increase the substrate scope of the asymmetric alkynylation reaction has led to discovery of pybox ligand.¹⁵

The problem, then, this multicomponent one-pot enantioselective synthesis of thiazolidine-2-imine presents is twofold: first acetylene has to undergo enantioselective addition on imine to form chiral propargylamine, and secondly the propargylthiourea intermediate formed by addition of propargylamine to isothiocyanate has to undergo regioselective (5 *exo dig vs* 6 *endo dig*) and chemoselective (hydrothiolation *vs* hydroamination) cyclization to form thiazolidine-2-imine. The difficulty of the task in our hand could be judged from the earlier reports that the relatively unhindered BINAP derived ligand was incapable to catalyze 5 exo-dig cyclization of propargylurea.⁹ With the aim of finding the right catalyst and ligand combination for the multicomponent enantioselective synthesis of thiazolidine-2 imine, we used C₂ symmetric bisoxazoline ligand **L1, L2** and L3 (Scheme 3). However, our initial attempts toward this end were severely jolted as we failed to achieve good enantioselectivity using ligand **L1** (Table 3 entries 1, 4 and 7).

Scheme 3 Chiral bis(oxazolinyl) ligands

Table 3 Effect of Conditions on the Enantioselectivity of the Multico mponent Reaction⁶

^aAll the reactions were performed with substituted imine (0.50 mmol), phenylacetylene (0.65 mmol) and benzoyl isothiocyanate (0.55 mmol), solvent (3 mL) for 3 days at 30 °C; ^bisolated yield after colum chromatography; ^cEnantiomeric excess was determined by chiralcel OD-. I column using Isopropanol:hexane (1:19).

Singh *et.al* were able to get a high enantiomeric excess when they used i-Pr-pybox-diPh ligand having diphenyl group at th C-5 position of the oxazoline rings. 8

Even though increase in enantioselectivity was observed whe ligand L2 was employed in the reaction (Table 3 entries 2, 5, 8 and 10), the enantioselectivity could be improved up to *ee* 90% (Table 3 entry 8). Lastly we turned our attention to ligand L3, which has been successfully used by Kang *et.al* for an asymmetric room temperature (30 °C) desymmeterization of symmetric $2,2$ disubstituted 1,3-diols to install all-carbon quaterna $\sqrt{ }$ stereocenter.¹⁶ Use of ligand L3 in combination with different counter ions and solvents was competent but not very selective (Table 3 entries 6 and 11-15). To our relief, high enantioselectivi' \prime (99% *ee*) was achieved when Cu(I)OTf/**L3** was employed as a catalyst system, (Table 3 entry 16). Temperature studies indicate that the product was obtained in good yield with excellent

COMMUNICATION Journal Name

enantioselectivity when the reaction was conducted at 30 °C. Lowering of the catalyst loading to 5 mol % led to slower reaction rates. When the reaction was performed in 10 volume of toluene, decrease in chemo selectivity was detected and hydroamination product was formed along with hydrothiolation product.

Table 4 Scope of the Multicomponent Reaction using Cu(OTf) Toluene pybox Complex*^a*

^aReaction conditions: All the reactions were performed with substituted imine (0.50 mmol), phenyl acetylene (0.65 mmol) and benzoyl isothiocyanate (0.55 mmol), 10 mol % CuOTf, 10 mol % pybox/L3, solvent (3 mL) for 3 days at 30 °C.

Having identified the optimum catalyst and reaction conditions, we set out to demonstrate the generality of our enantioselective protocol. The reaction was further extended to diverse imines, and the majority of them gave high enantioselectivity (Table 4). We obtained 99% *ee* when imine formed from 4- methoxybenzaldehyde (Table 4 entry **16o**) and 4-chlorobenzaldehyde was used, we obtained 99% enantiomeric excess (Table 4 entries **16a, 16f** and **16g**). Imine prepared from 3,4 dimethoxy benzaldehyde and aniline gave only 60% enantiomeric excess. (Table 4 entry **16s**) The yield obtained in this reaction was excellent in most of the cases. This three component reaction can be extended to *para* substituted phenylacetylene and *para* substituted benzoyl isothiocyanate. It is significant that the enantioselectivity was excellent in all types of substituted aromatic alkynes (Table 4 entries **16a** and **16c**) and isothiocyanate used (Table 4 entries **16b, 16g** and **16r**). The absolute configuration of the products was unambiguously assigned as "*S*" on the basis flack parameter¹⁷ of the X-ray crystal structure of compound 16l.

The stereochemistry of the exocyclic double bond was found to be Z (see supporting information).¹⁰

In summary, we have developed a system in which a copper catalysed both the alkynylation and the subsequent 5-*exo-dig* cyclization of the corresponding propargylthiourea to affor cyclic five-membered thiazolidine-2-imines. This cascade reaction forms a C-C, N-C and C-S bond in a one-pot. The importance of the chemistry described here lies in the established new synthetic method to prepare wide variety \overline{f} thiazolidine-2-imine using mild reaction conditions. The chiral Cu(I)OTf and pybox/**L3** complex was established to be new and an efficient catalyst for the enantioselective three-component synthesis of thiazolidine-2-imines from imines, alkynes ard. isothiocyanates. The method is very simple, furnished a wid variety of thiazolidines in good to excellent yields (up to 95%) with excellent enantioselectivities (up to 99% *ee*). **Chemcommunical Chemcommunical Chemcommuni**

A. R. and A. M. thank the Council of Scientific and Industrial Research (CSIR), New Delhi, for the award of a research fellowship. Financial support from the Department of Science and Technology (DST), Government of India, New Delhi (project no. SB/S1/OC- 01/2014) is gratefully acknowledg- d^2 We thank Pragati Pandey and Subhrajit Rout for HPLC analysis

Notes and references

- 1 (a) M. L. Barreca, E. De Clercq, *J. Med. Chem.* 2002, **45**, 5410; (b) B. Goel, A. Kumar, *Eur. J. Med. Chem.* 1999, **34**, 265. (c) A. Verma, S. K. Saraf, *Eur. J. Med. Chem.* 2008, **43**, 897.
- 2 S. Cuzzocrea, B. Zingarelli, E. H. Gilard, A. L. Salzman, U. Szabo, *Free Radical Biol. Med.* 1998, **24**, 450.
- 3 M. L. Barreca, *Bioorg. Med. Chem. Lett.* 2001, **11**, 1793-1796.
- 4 S. Nagar, H. H. Singh, J. N. Sinha, S. S. Parmar, *J. Med. Chem*. 1973, **16**, 178.
- 5 M. Sengoden, T. Punniyamurthy, *Angew Chem. Int. Ed.* 2013, **52**, 572.
- 6 V. Sriramurthy, G. A. Barcan, O. Kwon, *J. Am. Chem. Soc.* 2007, **129**, 12928.
- 7 M. J. Campbell, F. D. Toste, *Chem. Sci.* 2011, **2**, 1369.
- 8 V. Bisai, A. Suneja, V. K. Singh *Angew Chem. Int. Ed.* 2014, **26**, 10737.
- 9 A. Ranjan, R. Yerande, P. B. Wakchaure, S. G. Yerande, D. H. Dethe, *Org. Lett.* 2014, **16**, 5788.
- 10 CCDC 1403793 (**4a**) and CCDC 1403803 (**16l**) contains the supplemen tary crystallographic data for this paper.
- 11 (*a*) N. Gommermann, C. Koradin, K. Polborn, P. Knochel, *Angew. Chem. Int. Ed.* 2003, 42, 5763; (b) N. Gommerman P. Knochel *Tetrahedron* 2005, **61**, 11418; (c) Gommermann, P. Knochel, *Chem. Eur. J.* 2006*,* **12***,* 4380*.*
- 12 T. F. Knöpfel, P. Aschwanden, T. Ichikawa, T Watanabe, E. M. Carreira, *Angew. Chem. Int. Ed*. 2004, **43**, 5971.
- 13 (a) F. Colombo, M. Benaglia, S. Orlandi, F. Usuelli, G. Celentano, *J. Org. Chem.* 2006, 71, 2064; (b) F. Colombo, '1. Benaglia, S. Orlandi, F. Usuelli, J. Mol. Catal. A: Chem. 2006, **260**, 128.
- 14 S. Nakamura, M. Ohara, Y. Nakamura, N. Shibata; T. Toru *Chem. Eur. J.* 2010, **16**, 2360.
- 15 (a) A. Bisai, V. K. Singh, *Org. Lett*. 2006, **8**, 2405; (b) A. Bisai, Singh, K. *Tetrahedron* 2012**, 68**, 3480; (c) Z. Li, Jiang, W. Weike Su, *Green Chem.* 2015, **17**, 2330; (d) C .Wei, C. -J. Li, *J. Am. Chem. Soc.* 2002, **124**, 5638.
- 16 J. Y. Lee, Y. S. You, S. H. Kang, *J. Am. Chem. Soc.* 2011, **133**, 1772.
- 17 H. D. Flack *Acta Cryst.* 1983, **A39**, 876.

4 | *J. Name*., 2012, **00**, 1-3 This journal is © The Royal Society of Chemistry 20xx