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### Asymmetric cyclopropanation of conjugated cyanosulfones using a novel cupreine organocatalyst : Rapid access to δ³-amino acids

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An organocatalytic asymmetric synthesis of a novel, highly functionalised cyclopropane system furnished with versatile substituents and containing a quaternary centre is described. The process utilises a new bifunctional catalyst based on the cinchona alkaloid framework and the products made using this catalyst were obtained as single diastereoisomers, with very high enantioselectivities (up to 96% ee). We have also demonstrated that these resulting cyclopropanes are very useful synthetic intermediates to interesting products, such as the difficult to access δ³-amino acids.

The cyclopropane moiety is an important and common motif in a number of biologically active compounds and natural products.[1] It is also of synthetic use, especially if functionalised with the correct arrangement of electron-donating and electron-withdrawing groups, as it can then undergo a variety of nucleophilic, electrophilic or pericyclic reactions.[2] If the cyclopropane is asymmetric in nature, then the products of these transformations also tend to be enantiopure. As a result, routes to enantiopure cyclopropanes have been extensively studied,[3] particularly with respect to the Simmons-Smith reaction,[4] and transition metal catalysed reactions using carbene intermediates.[5] Organocatalysis has also demonstrated its utility in their synthesis. Aggarwal,[6] Gaunt,[7] and MacMillan[8] have all utilized ylides to achieve this, whilst the Michael-initiated ring closure (MIRC) has also proven to be a powerful method of choice.[9] In spite of this, MIRCs are dominated by compounds containing the nitro-group functionality, thus limiting substrate scope.[10] Therefore, in order to increase the diversity of this class of reaction, we sought to use the versatile conjugated cyanosulfone **1** shown in Scheme 1 instead. Previously our group has demonstrated this substrate to be a useful Michael acceptor in an organocatalytic domino process,[11] and so we wished to examine its utility in the MIRC cyclopropanation with

Scheme 1 Concept – organocatalytic synthesis of a versatile cyclopropane
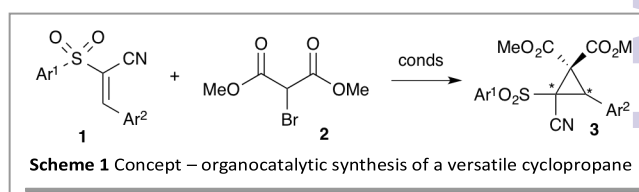
bromomethylmalonate **2**. We felt that the resulting novel cyclopropane product **3**, which is highly functionalised and contains a quaternary stereogenic centre, could undergo a variety of transformations to give useful products.

With this in mind, a test reaction was devised that utilised a variety of solvents and bifunctional catalysts (Table 1 for summary, Supporting Information for full investigation), mostly based around the framework of the cinchona alkaloids, which are well-known for their ability to activate electrophiles and pronucleophiles simultaneously.[12] Unfortunately, the most well-established bifunctional organocatalyst type - the thiourea-derived systems - failed to catalyse the reaction under our screening conditions. However, the quinine-derived organocatalyst cupreine **I** was able to impart some enantioselectivity, though both this and the yield were low (Entry 1). Encouragingly however, total diastereocontrol was observed in all cases. In an attempt to improve both enantioselectivity and yield, it was hypothesised that adding greater bulk to the secondary alcohol of the catalyst might improve the selectivity of the reaction. A number of cupreine-derived organocatalysts were therefore synthesised and screened (selected examples given, see supporting information for full study). In spite of the low yields obtained once again from this screen, we were encouraged to find that the acylated cupreine systems did indeed improve the stereoselectivity markedly (interestingly, the hydrocupreine derivatives performed better in terms of enantioselectivity than the cupreine derivatives), and that novel catalyst system **VII** performed best in this respect (Entry 11).

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Table 1 Optimisation screen for the asymmetric cyclopropanation (abridged)
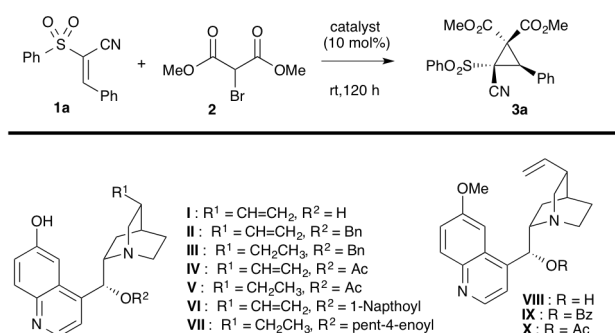
Entry	Cat	Solvent	Yield, %	dr[b]	er[c]
1	I	CH$_2$Cl$_2$	10	>19:1	68:32
2	I	THF	18	>19:1	59:41
3	I	CHCl$_3$	7	>19:1	62:38
4	I	PhMe	Trace	nd	nd
5	I	^{i}PrOH	Trace	nd	nd
6	II	CH$_2$Cl$_2$	12	>19:1	50:50
7	III	CH$_2$Cl$_2$	21	>19:1	54.5:45.5
8	IV	CH$_2$Cl$_2$	17	>19:1	79.5:20.5
9	V	CH$_2$Cl$_2$	17	>19:1	85:15
10	VI	CH$_2$Cl$_2$	11	>19:1	83.5:16.5
11	VII	CH$_2$Cl$_2$	12	>19:1	86:14
12	VIII	CH$_2$Cl$_2$	31	>19:1	50:50
13	IX	CH$_2$Cl$_2$	8	>19:1	50:50
14	X	CH$_2$Cl$_2$	12	>19:1	50:50

[a] Reaction conditions: room temperature, 120h. [b] Determined by ^{1}H-NMR. [c] Determined by chiral HPLC using a Chiralpak AD-H stationary phase.

**Table 2** Further optimisation study for the asymmetric cyclopropanation (abridged)[a]

Entry	Base[b]	Eq	Yield, %	Er
1	Na$_2$CO$_3$	1	92	85:15
2	Cs$_2$CO$_3$	1	86	53.5:46.5
3	K$_2$CO$_3$	0.5	88	86:14
4	K$_2$CO$_3$	1	90	85:15
5	K$_2$CO$_3$	2	92	84.5:15.5
6	K$_2$CO$_3$	4	90	83.5:16.5
7[c]	K$_2$CO$_3$	1	92	88:12

[a] In all cases only one diastereoisomer was observed by ^{1}H NMR [b] All bases presented in this table were dried in a vacuum oven before use. [c] Reaction performed at -10°C for 96h.

As with other studies using cupreine, it appears that the quinoline alcohol is critical to the enantioselectivity of the reaction, as if this is methylated, no selectivity occurs – although there is still

comparable turnover (Entries 12-14).[13] Clearly, however, the yield of the reaction required improvement, and so we endeavoured to achieve this by addition of base which is purported to enhance the reaction by neutralising the liberated HBr.[14] Owing to its marginally better selectivity profile, we screened a number of bases against organocatalyst **VII**, and pleasingly found that the yield improved significantly (Table 2), as did the rate of reaction. Impressively, we also found that the enantioselectivity was generally much better and that oven-dried potassium carbonate gave the best results in terms of this. These dry conditions are in contrast with a related study where water was shown to enhance the reaction.[14] Our final optimisation looked at the effect of catalyst loading and temperature, where ultimately, a loading of 10 mol% at the reduced temperature of -10°C for 96 h in dichloromethane was deemed to be the best conditions for this cyclopropanation reaction (Entry 7 - see supporting information for full study).

With an effective protocol for the enantioselective cyclopropanation in hand, we examined the generality and scope of the methodology. A variety of conjugated cyanosulfones were studied which varied in the electronics of the aryl group (Table 3). In most cases the initial enantioselectivity was good. More pleasing, however was the fact that the majority of these could be significantly improved through a *single* recrystallization. In all cases the diastereoselectivity was excellent and the yields were also excellent. The absolute crystal structure of **3c** was determined by X-ray crystallography (Figure 1) and the other products assigned by analogy.
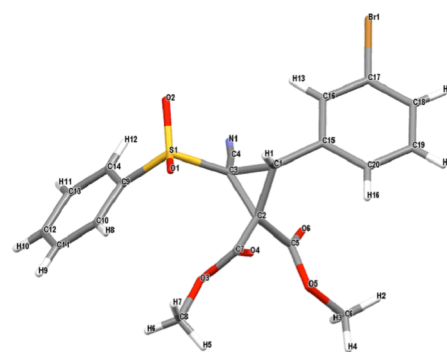
Figure 1 X-ray crystal structure of **3c**, showing the absolute configuration. CCDC no: 1403683, http://www.ccdc.cam.ac.uk.

We speculate that as with all bifunctional organocatalytic systems, the reaction proceeds *via* simultaneous activation of the pronucleophile (bromomalonate) and the electrophilic cyanosulfone through co-ordination to the cupreine derived catalyst. We tentatively propose that the transition state shown in Figure 2 could account for this activation mode and the resulting stereochemistry. We also believe that the exceptionally high diastereoselectivity is due to the intermediate anion probably being long-lived enough to equilibrate to the thermodynamically more stable conformation, whereby the sulfone and the aryl group adopt positions that avoid unfavourable gauche interaction.
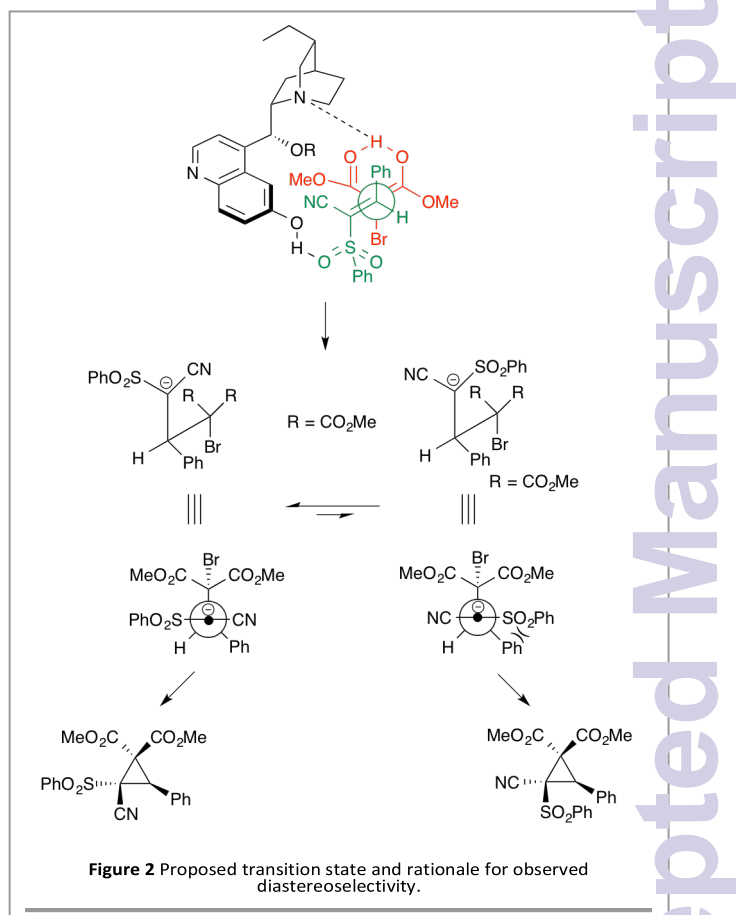
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Table 1 Optimisation screen for the asymmetric cyclopropanation

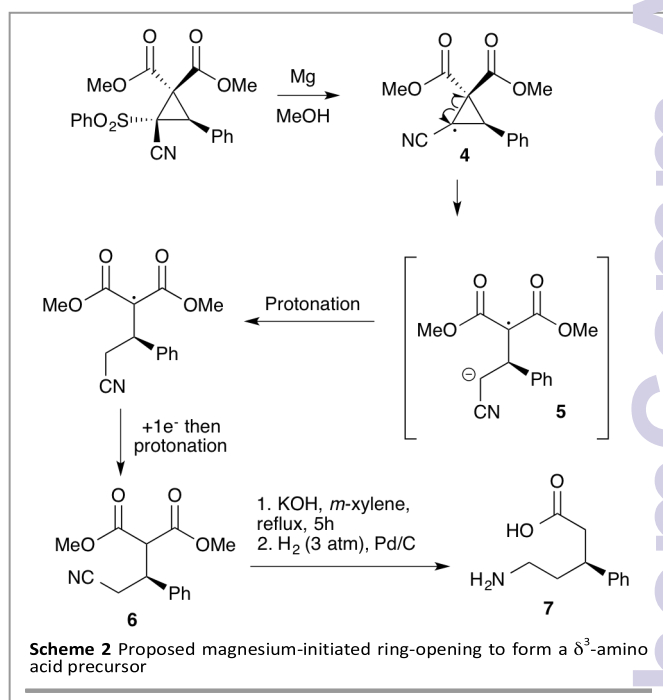| Entry	Product[a]	Yield%,[b] er,[c] (er after single recrystallisation)
1	3a	92%, 88:12 (90:10)
2	3b	>99%, 91:9 (96:4)
3	3c	>99%, 91:9 (98:2)
4	3d	94%, 89:11 (94.5:5.5)
5	3e	96%, 90:10 (91.5:8.5)
6	3f	73% 80:20 (80:20)
7	3g	91%, 80.5:19.5 (95:5)
8	3h	88%, 79:21 (88:12)
9	3i[d]	94%, 80:20

[a] Only one diastereoisomer observed in 1H NMR. [b] Isolated yield. [c] Determined by HPLC using a Chiralpak AD-H or Chiralcel OD stationary phase. [d] Did not recrystallize.

Figure 2 Proposed transition state and rationale for observed diastereoselectivity.

system (Scheme 3). This fascinating ring system appears in a wide variety of natural products and medicinally useful compounds, such as the antibiotic indolizomycin **9**, and cycloclavine **10**, an ergot alkaloid.
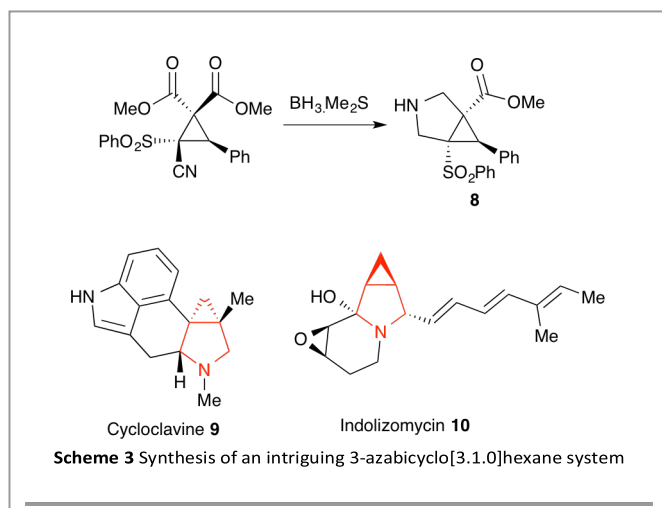
Finally, in order to demonstrate the utility of these compounds, we subjected them to a variety of transformative conditions. To begin with, we attempted a straightforward desulfonylation using refluxing magnesium in methanol[15] to obtain the δ^3-amino acid precursor **6** with no depreciation of enantiomeric ratio. We propose that this occurs *via* a radical desulfonylation, generating cyclopropane radical **4** which rapidly ring opens to the radical anion **5**. Protonation, followed by addition of a further electron to the malonate radical and protonation gives the δ^3-amino acid precursor **6**.[16] Hydrolysis/decarboxylation and reduction of the nitrile then gives the corresponding δ^3-amino acid (Scheme 2).

This is an intriguing result, as reports on the synthesis of enantiopure δ-amino acids appear to be extremely rare in the literature. Our methodology represents a rapid new way of accessing this particular class, being just *five* linear steps.[17]

Finally, we also achieved the synthesis of the intriguing 3-azabicyclo[3.1.0]hexane system **8** through the borane mediated reduction of the nitrile, leading to concomitant cyclisation of the resulting amine onto the *syn*-ester followed by *in situ* reduction of the γ-lactam to the fused pyrrolidine

Scheme 2 Proposed magnesium-initiated ring-opening to form a δ^3-amino acid precursor

Scheme 3 Synthesis of an intriguing 3-azabicyclo[3.1.0]hexane system

Conclusions

We have developed the enantioselective synthesis of a highly functionalised cyclopropane which can be transformed into several useful substrates. In particular, magnesium initiated radical desulfonylation-ring opening leads to the corresponding δ^3-amino acid, and thus this route represents an extremely expedient way of accessing this class of molecule. In addition we have utilised the cyclopropane constructs to access a novel 3-azabicyclo[3.1.0]hexane system.

Notes and references

1   For a review on the occurrences of cyclopropanes in nature and recent syntheses see D. Y.-K. Chen, R. H. Pouwer and J.-A. Richard, *Chem Soc. Rev.*, 2012, **41**, 4631.

2   M. A. Cavitt, L. H. Phun and S. France, *Chem. Soc. Rev.*, 2014, **43**, 804.

3   (a) H. Pellissier *Tetrahedron* 2008, **64**, 7041; (b) G. Bartoli, G. Bencivenni and R. Dalpozzo, *Synthesis* 2014, **46**, 979; (c) H. Lebel, J.-F. Marcoux, C. Molinaro and A. B. Charette, *Chem. Rev.* 2003, **103**, 977; (d) A. B. Charette and J.-F. Marcoux, *Synlett* 1995, 1197.

4   (a) H. E. Simmons and R. D. Smith, *J. Am. Chem. Soc.*, 1958, **80**, 5323; (b) H. E. Simmons and R. D. Smith, *J. Am. Chem. Soc.*, 1959, **81**, 4256; For catalytic asymmetric variations see (c) H. Takahashi, M. Yoshioka, M. Ohno and S. Kobayashi, *Tetrahedron Lett.* 1992, **33**, 2575; (d) H. Shitama and T. Katsuki, *Angew. Chem. Int. Ed.*, 2008, **47**, 2450; (e) H. Du, J. Long and Y. Shi, *Org. Lett.* 2006, **8**, 2827; (f) J. Long, Y. Yuan and Y. Shi, *J. Am. Chem. Soc.* 2003, **125**, 13632; (g) J. Long, H. Du, K. Li and Y. Shi, *Tetrahedron Lett.* 2005, **46**, 2737; (h) H. Y. Kim, A. E. Lurain, P. García- García, P. J. Carroll and P. J. Walsh, *J. Am. Chem. Soc.* 2005, **127**, 13138; (i) R. G. Cornwall, O. A. Wong, H. Du, T. A. Ramirez and Y. Shi, *Org. Biomol. Chem.*, 2012, **10**, 5498.

5   For recent discussion on this extensive area of catalysis, see (a) A.F. Trindade, J. A. S. Coelho, C. A. M. Afonso, L. F. Veiros

and P. M. P. Gois, *ACS Catal.*, 2012, **2**, 370; (b) M. P. Doyle, Duffy, M. Ratnikov and L. Zhou, *Chem. Rev.*, 2010, **110**, 704; (c) H. M. L. Davies and J. R. Denton, *Chem. Soc. Rev.*, 2009, **38**, 3061 and references therein.

6   V. K. Aggarwal, E. Alonso, G. Fang, M. Ferrara, G. Hynd and M. Porcelloni, *Angew. Chem., Int. Ed.* 2001, **40**, 1433.

7   (a) C. D. Papageorgiou, M. A. Cubillo de Dios, S. V. Ley, M. J. Gaunt, *Angew. Chem., Int. Ed.* 2004, **43**, 4641; (b) C. C. C. Johansson, N. Bremeyer, S. V. Ley, D. R. Owen, S. C. Smith and M. J. Gaunt, *Angew. Chem., Int. Ed.* 2006, **45**, 6024.

8   (a) R. K. Kunz and D. W. C. MacMillan, *J. Am. Chem. Soc.* 2006, **127**, 3240; For similar work see (b) A. Hartikka and P. I. Arvidsson, *J. Org. Chem.* 2007, **72**, 5874; (c) A. Hartikka, A. T. Slosarczyka and P. I. Arvidsson, *Tetrahedron: Asymmetry* 2007, **18**, 1403.

9   See reference 3 and the appropriate section of C. M. R. Volla, I. Atodiresei and M. Rueping, *Chem. Rev.* 2013, **114**, 2390 for detailed review.

10 (a) A. J. A. Cobb, in *Enantioselective Organocatalyzed Reactions II: Asymmetric C-C Bond Formation Processes*; R. Mahrwald, Ed.; Springer: Dordrecht, 2011. (b) L. S. Aitken, R. Arezki, A. Dell'Isola and A. J. A. Cobb, *Synthesis* 2013, **45** 2627.

11 (a) S. Rajkumar, K. Shankland, J. M. Goodman and A. J. A. Cobb, *Org. Lett.* 2013, **15**, 1386. For a related process, see (b) S. Meninno, G. Groce and A. Lattanzi, *Org. Lett.* 2013, **15**, 3436.

12 For minireviews, see: (a) M. Kotke and P. R. Schreiner, *Hydrogen Bonding in Organic Synthesis*; P. Pihko, Ed.; Wiley VCH, 2009, pp. 141 – 351; (b) H. Miyabe and Y. Takemoto, *Bull. Chem. Soc. Jpn.* 2008, **81**, 785-795; (c) S. J. Connon, *Chem. Commun.* 2008, **22**, 2499; (d) T. Marcelli, J. H. van Maarseveen and H. Hiemstra, *Angew. Chem. Int. Ed.* 2006, **45**, 7496; Other work by us in this area includes : (e) W. J. Nodes, D. R. Nutt, A. M. Chippindale and A. J. A. Cobb, *J. Am. Chem. Soc.* 2009, **131**, 16016; (f) S. Rajkumar, K. Shankland, G. D. Brown and A. J. A. Cobb, *Chem. Sci.* 2012, **3**, 584.

13 The C6-OH of cupreine has been shown to be very important in other reactions for obtaining high selectivities. See: (a) H. Li, Y. Wang, L. Tang and L. Deng, *J. Am. Chem. Soc.* 2004, **126**, 9906; (b) H. Li, Y. Wang, L. Tang, F. Wu, X. Liu, C. Guo, B. M. Foxman and L. Deng, *Angew. Chem., Int. Ed.* 2005, **44**, 105; (c) X. Liu, H. Li and L. Deng, *Org. Lett.* 2005, **7**, 167; (d) H. Li, Song, X. Liu and L. Deng, *J. Am. Chem. Soc.* 2005, **127**, 8948; (d) R. Dodda, J. J. Goldman, T. Mandal, C.-G. Zhao, G. A. Broker and E. R. T. Tiekink, *Adv. Synth. Catal.*, 2008, **350**, 537; (e) P. Chauhan and S. S. Chimni, *Tetrahedron Lett.* 2013, **54**, 4613; (f) P. Chauhan and S. S. Chimni, *RSC Advances*, 2012, **2**, 737.

14 U. Das, Y.-L. Tsai and W. Lin, *Org. Biomol. Chem.* 2013, **11**, 44.

15 G. H. Lee, I. K. Youn, E. B. Choi, H. K. Lee, G. H. Yon, H. C. Yang and C. S. Pak, *Curr. Org. Chem.* 2004, **8**, 1263.

16 The only previous report of this compound is in European Pat., EP1512678 A1, 2005, which obtained it in a 60% ee.

17 For other (longer) syntheses, see (a) R. Karla, B. Ebert, C. Thorkildsen, C. Herdeis, T. N. Johansen, B. Nielsen and P. Krogsgaard-Larsen, *J. Med. Chem.* 1999, **42**, 2053. (b) Z.-T He, Y.-B. Wei, H.-J. Yu, C.-Y. Sun, C.-G. Feng, P. Tian and G.-Q. Lin. (c) M. I. Attia, C. Herdeis and H. Bräuner-Osborne, *Molecules* 2013, **18**, 10266 (non-chiral).