

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



Journal Name

Communication

Received 00th January 20xx,

A general method for *N*-glycosylation of nucleobases promoted by $(p\text{-Tol})_2\text{SO}/\text{Tf}_2\text{O}$ with thioglycoside as donor

Guang-jian Liu, Xiao-tai Zhang and Guo-wen Xing*

Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Based on a preactivation strategy with $(p\text{-Tol})_2\text{SO}/\text{Tf}_2\text{O}$ system, a series of nucleosides were synthesized by coupling various thioglycosides with pyrimidines and purines under mild conditions. High yields and excellent β -stereoselectivities were obtained with either armed or disarmed *N*-glycosylation donors by tuning the amount of $(p\text{-Tol})_2\text{SO}$ additive.

Nucleoside is arguably one of the most prominent compounds since the structure of DNA was elucidated by Watson and Crick¹. To date, the modified nucleoside has been used increasingly in expanding genetic code², developing new antiviral and antitumor drugs³, as well as creating biological probes⁴. Therefore, the study on nucleoside synthesis has continuously been a topical subject, and a large number of studies have been reported^{5,6}. Among various methodologies to construct nucleoside, the *N*-glycosylation reaction involving the direct coupling of a sugar moiety with nucleobase is the simplest synthetic strategy, which has been classified into three types: (1) acid-catalyzed fusion of peracylated sugars with nucleobases⁷; (2) the metal salt of heterocyclic systems reacts with protected sugar halides⁸; (3) the silyl-Hilbert-Johnson method developed by Vorbrüggen et al.⁹. Especially, the Vorbrüggen reaction has been used widely and developed into the most prevailing route for nucleoside synthesis¹⁰. Most recently, further more efficient methods were devised to enhance the product yield and the selectivity of the coupling reaction. For example, Yu et al. coupled the glycosyl ortho-alkynylbenzoates with nucleobases using a gold(I) complex as an activator in high regioselectivity¹¹ and stereoselectivity¹². Fraser-Ried¹³ and co-workers utilized the reverse synthesis strategy permitting prior extensive structural modifications to the ribose unit. In addition, Jamison's research group¹⁴ telescoped cumbersome nucleoside synthesis processes into a continuous one-flow multistep sequence using pyridinium triflates as catalyst. According to these advances, many novel nucleoside analogues have been successfully synthesized. However, it is still desired to develop new universal strategies for various armed/disarmed sugar donors and pyrimidine/purine acceptors to synthesize nucleosides under mild conditions.

Thioglycoside, one of the most useful donors for *O*-glycosylation, is convenient to prepare and stable under many reaction conditions^{15a}. However, fewer literatures¹⁶ using thioglycoside for nucleoside synthesis were reported. Moreover, the substrate scope was mainly limited to the pentose sugars and pyrimidine nucleobases. Especially, low

yields and poor selectivities were obtained with ambident and weakly nucleophilic purines as acceptors^{16c-f}. Over the years, the preactivation protocol with $(p\text{-Tol})_2\text{SO}/\text{Tf}_2\text{O}$ system, developed from the traditional $\text{Ph}_2\text{SO}/\text{Tf}_2\text{O}$ pair,^{15b} has been demonstrated to be a practical, straightforward, and high stereoselectivity method for *O*- and *C*-glycosylations using *N*-acetyl-5-*N*,4-*O*-oxazolidione protected thiosialoside as donor¹⁷. In continuation of our studies on this glycosylation protocol, we are intrigued whether this method is suitable for *N*-glycosylation, and finally establish one general method for synthesis of various *O*-, *C*-, and *N*-glycoconjugates.

Herein, we report a facile and general method, based on the preactivation strategy promoted by $(p\text{-Tol})_2\text{SO}/\text{Tf}_2\text{O}$, for the production of nucleoside analogues from armed/disarmed sugar donors and various nucleobases in high yields and stereoselectivities.

With donor **1a** in hand, we initially performed the model *N*-glycosylation of **1a** with trimethylsilylated uracil **2a** by the preactivation protocol. A solution of **1a** and $(p\text{-Tol})_2\text{SO}$ (1.2 equiv) was preactivated by Tf_2O (1.2 equiv) in dichloromethane at -70°C for 0.5 h, followed by addition of **2a** (3.0 equiv) in CH_3CN . The resulted mixture was stirred for 2 h, and then warmed up to -50°C for another 2 h. The coupling product **3** (Table 1, entry 1) was obtained in 60% yield with exclusive β -stereoselectivity. Based on our previous findings¹⁷ that an excessive amount of $(p\text{-Tol})_2\text{SO}$ can raise the glycosylation yields by stabilizing the active intermediates derived from donors, we increased the amount of $(p\text{-Tol})_2\text{SO}$ from 1.2 equiv to 3.0 equiv, and found that the corresponding yields were not increased dramatically (Table 1, entries 2-3). These inferior results, which we attributed to the deactivating effect of the acetyl protecting group on donor **1a**, prompted us to optimize the reaction conditions by modulating both the temperature of preactivation and the amount of $(p\text{-Tol})_2\text{SO}$ at the same time. Surprisingly, when the preactivation temperature was elevated to -60°C , the isolated product yield jumped significantly from 66% to 99% with increasing the amount of $(p\text{-Tol})_2\text{SO}$ from 1.2 equiv to 3.0 equiv (Table 1, entries 4-6).

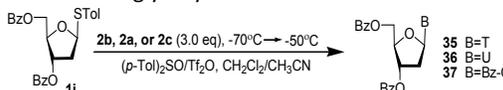
During above optimization studies, it was revealed that the efficiency of the *N*-glycosylation was significantly dependent on the preactivation temperature and $(p\text{-Tol})_2\text{SO}$ amount,

Department of Chemistry, Beijing Normal University, Beijing 100875, China. E-mail: gwxing@bnu.edu.cn

Electronic supplementary information (ESI) available: Detailed experimental procedures. See DOI: 10.1039/x0xx00000x

which are closely related to the donor reactivity and the stability of the glycosylation transition state respectively. Higher preactivation temperature could benefit the unreactive donors, and more (*p*-Tol)₂SO would be helpful to trap the less stable glycosylation intermediates.

Table 1 Effect of the (*p*-Tol)₂SO amount and preactivation temperature on *N*-glycosylation with **1a** as the donor



Entry	<i>T</i>	(<i>p</i> -Tol) ₂ SO	Yield	α/β
1		1.2 equiv	60%	β
2	-70°C	2.0 equiv	57%	β
3		3.0 equiv	79%	β
4		1.2 equiv	66%	β
5	-60°C	2.0 equiv	91%	β
6		3.0 equiv	99%	β

Next, a series of frequently used thioglycosides **1a-f** and nucleobase derivatives **2a-f** (Fig. 1) were prepared to explore the substrate scope of the *N*-glycosylation promoted by (*p*-Tol)₂SO/Tf₂O. We first investigated the reaction of pyrimidine nucleobases **2a-c** with donors **1a-f**, derived from D-galactose, D-glucose and D-ribose (Scheme 1). Under the optimized reaction conditions (-60°C → -40°C, (*p*-Tol)₂SO (3.0 equiv)), the coupling reactions of **1a** with TMS-thymine (**2b**) and TMS-N⁴-benzoylcytosine (**2c**) provided the nucleosides **4** and **5** in excellent yields (> 96%).

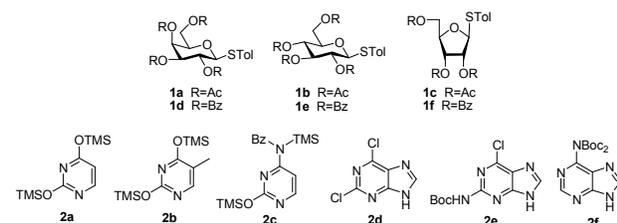
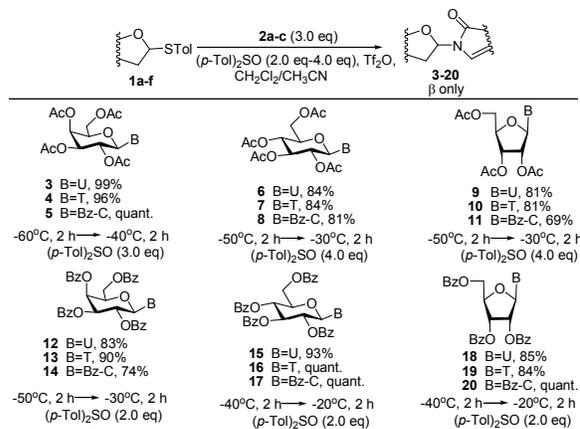


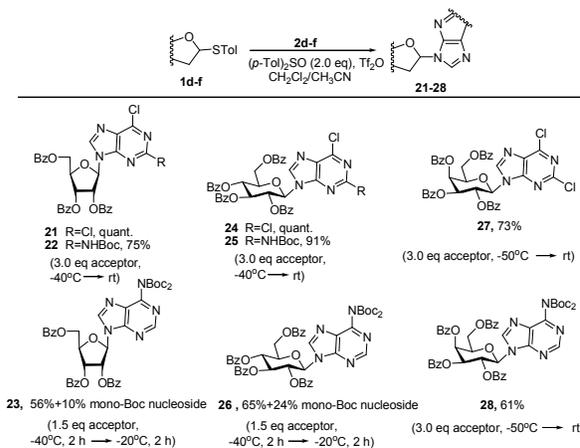
Fig. 1 Structures of glycosyl donors and acceptors for nucleosides syntheses

Under the same reaction conditions for **1a**, the *N*-glycosylation of **1b** with **2a** afforded product **6** in a higher yield (76%) than that observed by Yu¹¹ (67%) with glycosyl ortho-alkynylbenzoate as donor. Moreover, repetition of the experiment of **1b** and **2a** by warming up the preactivation temperature to -50°C, a much higher yield (84%, Scheme 1) was obtained, so were **2b** and **2c**. In addition, the peracylated ribofuranose **1c** reacted with TMS-pyrimidines (**2a-c**) to provide the desired β nucleosides **9-11** in good yields.

Taking into account that the benzoyl protected donors are less reactive than the corresponding acetyl protected donors¹⁸, we performed the coupling reactions of **1d-f** with nucleobases (**2a-c**) with a higher preactivation temperature. Much to our delight, among all these *N*-glycosylation reactions, satisfactory yields of the coupling products **12-20** were provided with eight out of nine examples ≅ 83% (Scheme 1).



Scheme 1 *N*-Glycosylation of glycosyl donors with pyrimidines



Scheme 2 *N*-Glycosylation of glycosyl donors with purines

Then, we focused on the real challenge of *N*-glycosylation with purines. Since the Boc-protected purine nucleobases have been prepared^{19a} and used in *N*-glycosylations and other reactions with good solubility and high N9/N7 regioselectivity^{10,11,19b-d}, we selected the purine analogues, such as 2,6-dichloropurine (**2d**), N²-*tert*-butoxycarbonyl-2-amino-6-chloropurine (**2e**) and N⁶-bis(*tert*-butoxycarbonyl)-adenine (**2f**), as acceptors to extend the *N*-glycosylations with thioglycosides as donors. As expected, donor **1f** was coupled with acceptor **2d** smoothly and furnished the exclusive product **21** quantitatively. Notably, the structure of the coupling product **21** was determined unequivocally on the basis of the precedent²⁰ as well as the combined analysis of 1D and 2D NMR (¹H, ¹³C, HSQC, HMBC) experiments. In particular, the strong HMBC correlation between H-1' and C4 rather than H-1' and C5 was telltale (see ESI for details) to confirm compound **21** is N9 product. Similarly, the N²-*tert*-butoxycarbonyl-2-amino-6-chloropurine (**2e**) also underwent *N*-glycosylation with **1f** to preponderantly give the desired N9-nucleoside **22** in good yield (75%). In the case of N⁶-bis(*tert*-butoxycarbonyl)adenine (**2f**), the coupling product **23** was isolated along with the corresponding mono-Boc-protected nucleoside as a side product (Scheme 2).

In a similar manner, the coupling reactions of the three purine nucleobases (**2d-f**) with **1d** and **1e** were also examined. The N9-nucleosides **24-28** were obtained in moderate to excellent yields (61-100%, Scheme 2), and the corresponding

Journal Name COMMUNICATION

N7-nucleosides were not detected. The structures of **22-28** were definitely corroborated from NMR experiments as described for the nucleoside **21**.

Table 2 *N*-glycosylation of perbenzyl protected donor **1g** with **2a-c** using different amount of (*p*-Tol)₂SO.

Entry	Product	(<i>p</i> -Tol) ₂ SO	Yield ^a	α/β ^b
1	29	1.2 equiv	76%	1:2.5
2		2.0 equiv	92%	1:2.7
3		3.0 equiv	99%	1:3.7
4		4.0 equiv	97%	1:7.9
5		6.0 equiv	quant.	>1:30
6	30	6.0 equiv	quant.	β
7		31	6.0 equiv	quant.

^a Isolated yields, ^b Determined by NMR.

Table 3 *N*-glycosylation of perbenzyl protected donor **1h** with **2a-c** under different preactivation conditions.

Entry	Product	<i>T</i>	(<i>p</i> -Tol) ₂ SO	Yield ^a	α/β ^b
1	32	-70°C	4.0 equiv	64%	1:4.3
2		6.0 equiv	56%	1:4.6	
3		-60°C	2.0 equiv	92%	1:5.6
4		4.0 equiv	quant.	1:8.7	
5		6.0 equiv	99%	1:17	
6	33	-60°C	2.0 equiv	93%	1:3.6
7		6.0 equiv	99%	1:25	
8		34	-60°C	6.0 equiv	80%

^a Isolated yields, ^b Determined by NMR.

Next, in an effort to ascertain the efficiency of the *N*-glycosylation based on (*p*-Tol)₂SO/Tf₂O preactivation protocol thoroughly, and especially to probe the influence of different added amount of (*p*-Tol)₂SO on the reaction, some complementary experiments with perbenzyl protected thioglycosides (**1g** and **1h**) as donors were performed (Tables 2 and 3).

The exploratory reactions of **1g** with **2a** were undertaken with different amount of (*p*-Tol)₂SO. As shown in Table 2, with increasing the amount of (*p*-Tol)₂SO from 1.2 equiv to 6.0 equiv, both the yield and stereoselectivity of the reaction were increased significantly (Table 2, entries 1-5). Remarkably, the similar excellent yields (~100%) and β-stereoselectivities were also obtained in the syntheses of **30** and **31** with 6.0 equiv. of (*p*-Tol)₂SO used for the *N*-glycosylations (Table 2, entries 6-7). To the best of our knowledge, this is the first time to obtain the perbenzyl protected nucleosides in so high stereoselectivity and yield simultaneously without the anchimeric assistance at the C2 position of glycosyl donors²¹.

Continuously, Table 3 illustrates the results of our efforts to scope the *N*-glycosylations between perbenzyl protected glucopyranoside (**1h**) and nucleobases. Interestingly, when the preactivation temperature was -70°C, the results of the reaction were not satisfactory, even 6.0 equiv of (*p*-Tol)₂SO was used (Table 3, entry 1, 2). The somewhat surprising fact was comparable with the previous observations (Table 1), which probably arised from the low reactivity of donor **1h** at

very low reaction temperature. Whereupon, analogous reactions of **1h** with TMS-pyrimidines (**2a-c**) were done using different amounts of (*p*-Tol)₂SO with the preactivation temperature elevated to -60°C. More efficient *N*-glycosylations were achieved (Table 3, entries 5, 7-8), indicating that the yield and stereoselectivity of *N*-glycosylations could be tunable to a certain degree by changing the amount of (*p*-Tol)₂SO and the preactivation temperature according to the characteristics of the used donors.

In addition, above *N*-glycosylation protocol was also applied into the syntheses of 2-deoxyribonucleosides with 2-deoxy-β-thioriboside **1i** and TMS-thymine (**2b**) as glycosyl donor and acceptor, respectively (Table 4). More interestingly, when the amount of (*p*-Tol)₂SO additive increased from 0.8 equiv to 4.0 equiv, the product yield of 2-deoxyribonucleoside **35** was elevated from 53% to ~100%, whereas the β-stereoselectivity of the reaction was decreased from β/α = 7.8:1 to 2.8:1 (Table 4, entries 1-5). The results are similar to our previous research findings for another kind of deoxysugar, sialic acid, wherein the α-stereoselectivity was dropped down with increasing the Ph₂SO amount,^{17a,17d} indicating that the detailed glycosylation mechanism could be different between deoxysugar and ordinary sugar involved reactions. It is noted that much lower β-stereoselectivity (β/α = 1.6:1) and reaction yield (43%) were obtained in the same *N*-glycosylation of **1i** and **2b** with NIS/TfOH^{16e} as promoter. Considering the influences of both reaction yield and β-stereoselectivity on the coupling, 1.2 equiv of (*p*-Tol)₂SO was employed for the following *N*-glycosylation of TMS-uracil (**2a**) and TMS-N4-benzoylcytosine (**2c**). The expected 2-deoxyribonucleosides **36** and **37** were prepared in good β-stereoselectivities (β/α = 4.5:1 to 5.1:1, Table 4, entries 6-7) and 79-54% yields.

Table 4 *N*-glycosylation of perbenzoyl protected 2-deoxy-β-thioriboside **1i** with nucleobases **2a-c**.

Entry	Product	(<i>p</i> -Tol) ₂ SO	Yield ^a	α/β ^b
1	35	0.8 equiv	53%	1:7.8
2		1.2 equiv	75%	1:6.6
3		1.6 equiv	86%	1:5.1
4		2.0 equiv	99%	1:3.2
5		4.0 equiv	quant.	1:2.8
6	36	1.2 equiv	79%	1:4.5
7	37	1.2 equiv	54%	1:5.1

^a Isolated yields, ^b Determined by NMR.

In summary, a series of nucleosides were prepared efficiently by coupling the readily available thioglycosides with pyrimidine and purine nucleobases based on (*p*-Tol)₂SO/Tf₂O preactivation protocol under mild conditions. The *N*-glycosylation outcome is highly dependent on the preactivation temperature, and especially the added amount of (*p*-Tol)₂SO in the reaction. Generally, high yields and excellent β-stereoselectivities were obtained even perbenzyl protected sugars and 2-deoxy-β-thioriboside were used as donors by tuning the amount of (*p*-Tol)₂SO additive. Combined with our previous work¹⁷, the (*p*-Tol)₂SO/Tf₂O preactivation protocol discussed in this study has been demonstrated its

usefulness and generality for the efficient syntheses of various glycoconjugates including *N*-, *O*- and *C*-glycosides.

The project was financially supported by the Natural Science Foundation of China (21272027) and Beijing Municipal Commission of Education.

Notes and references

- (a) A. J. A. Cobb, *Org. Biomol. Chem.*, 2007, **5**, 3260-3275; (b) D. M. Huryn and M. Okabe, *Chem. Rev.*, 1992, **92**, 1745-1768.
- I. Luyten and P. Herdewijn, *Eur. J. Med. Chem.*, 1998, **33**, 515-576;.
- (a) J. Stagg and M. J. Smyth, *Oncogene*, 2010, **29**, 5346-5358; (b) E. De Clercq, *Med. Res. Rev.*, 2010, **30**, 667-707; (c) A. Matsuda and T. Sasaki, *Cancer Sci.*, 2004, **95**, 105-111; (d) E. D. Clercq, *Nat. Rev. Drug Discovery*, 2007, **6**, 1001-1018; (e) A. Dumas and N. W. Luedtke, *J. Am. Chem. Soc.*, 2010, **132**, 18004-18007.
- (a) J. N. Wilson and E. T. Kool, *Org. Biomol. Chem.*, 2006, **4**, 4265-4274; (b) A. P. Silverman and E. T. Kool, *Chem. Rev.*, 2006, **106**, 3775-3789; (c) J. P. May, L. J. Brown, I. van Delft, N. Thelwell, K. Harley and T. Brown, *Org. Biomol. Chem.*, 2005, **3**, 2534-2542; (d) N. J. Greco and Y. Tor, *J. Am. Chem. Soc.*, 2005, **127**, 10784-10785; (e) S. G. Srivatsan and Y. Tor, *J. Am. Chem. Soc.*, 2007, **129**, 2044-2053; (f) R. W. Sinkeldam, N. J. Greco and Y. Tor, *Chem. Rev.*, 2010, **110**, 2579-2619.
- (a) L. J. Wilson, M. W. Hager, Y. A. El-Kattan and D. C. Liotta, *Synthesis*, 1995, **1995**, 1465-1479; (b) G. Romeo, U. Chiacchio, A. Corsaro and P. Merino, *Chem. Rev.*, 2010, **110**, 3337-3370; (c) H. Vorbrüggen and C. Ruh Pohlenz, *Synthesis of nucleosides*, Wiley Online Library, 2000; (d) J. W. Arico, A. K. Calhoun, K. J. Salandria and L. W. McLaughlin, *Org. Lett.*, 2010, **12**, 120-122; (e) M. Ferrero and V. Gotor, *Chem. Rev.*, 2000, **100**, 4319-4348; (f) N. Shimomura, T. Matsutani and T. Mukaiyama, *Bull. Chem. Soc. Jpn.*, 1994, **67**, 3100-3106.
- (a) S. Hanessian, G. Huang, C. Chenel, R. Machaalani and O. Loiseleur, *J. Org. Chem.*, 2005, **70**, 6721-6734; (b) S. Knapp, V. V. Thakur, M. R. Madduru, K. Malolanarasimhan, G. J. Morriello and G. A. Doss, *Org. Lett.*, 2006, **8**, 1335-1337; (c) J. Liao, J. Sun and B. Yu, *Carbohydr. Res.* 2009, **344**, 1034-1038; (d) J. Huchting and C. Meier, *Eur. J. Org. Chem.*, 2014, **2014**, 3423-3429; (e) C. S. Stauffer and A. Datta, *J. Org. Chem.*, 2008, **73**, 4166-4174.
- (a) J. Im, J. Kim, S. Kim, B. Hahn and F. Toda, *Tetrahedron Lett.* 1997, **38**, 451-452; (b) E. Diekmann, K. Friedrich and H. G. Fritz, *J. Prakt. Chem.*, 1993, **335**, 415-424.
- (a) T. Ukita, H. Hayatsu and Y. Tomita, *Chem. Pharm. Bull.*, 1963, **11**, 1068-1073; (b) Z. Kazimierczuk, H. B. Cottam, G. R. Revankar and R. K. Robins, *J. Am. Chem. Soc.*, 1984, **106**, 6379-6382; (c) H. Kawakami, H. Matsushita, Y. Naoi, K. Itoh and H. Yoshikoshi, *Chem. Lett.* 1989, **18**, 235-238.
- (a) H. Vorbrueggen, *Acc. Chem. Res.*, 1995, **28**, 509-520; (b) L. Birkofer, A. Ritter and H. P. Kuhlthau, *Angew. Chem.*, 1963, **75**, 209-210; (c) T. Nishimura, B. Shimizu and I. Iwai, *Chem. Pharm. Bull.*, 1963, **11**, 1470-1472; (d) E. Wittenburg, *Chem. Ber.*, 1968, **101**, 1095-1114; (e) U. Niedballa and H. Vorbrueggen, *J. Org. Chem.*, 1974, **39**, 3654-3660.
- (a) S. Knapp, *Chem. Rev.*, 1995, **95**, 1859-1876; (b) K. Chow and S. Danishefsky, *J. Org. Chem.*, 1990, **55**, 4211-4214; (c) U. Niedballa and H. Vorbrueggen, *J. Org. Chem.*, 1974, **39**, 3672-3674; (d) H. Vorbrüggen and C. Ruh-Pohlenz, *Handbook of nucleoside synthesis*, John Wiley & Sons, 2001.
- Q. Zhang, J. Sun, Y. Zhu, F. Zhang and B. Yu, *Angew. Chem. Int. Ed.*, 2011, **50**, 4933-4936.
- F. Yang, Y. Zhu and B. Yu, *Chem. Commun.*, 2012, **48**, 7097-7099.
- B. Fraser-Reid, P. Ganney, C. V. S. Ramamurty, A. M. Gomez and J. C. Lopez, *Chem. Commun.*, 2013, **49**, 3251-3253.
- A. Sniady, M. W. Bedore and T. F. Jamison, *Angew. Chem. Int. Ed.*, 2011, **50**, 2155-2158.
- (a) G. Lian, X. Zhang, B. Yu, *Carbohydr. Res.* 2015, **403**, 13-22; (b) J. D. C. Codee, R. E. J. N. Litjens, R. den Heeten, H. S. Overkleeft, J. H. van Boom, G. A. van der Marel, *Org. Lett.*, 2003, **5**, 1519-1522.
- (a) H. Sugimura, K. Sujino and K. Osumi, *Tetrahedron Lett.*, 1992, **33**, 2515-2516; (b) J. Nokami, M. Osafune, Y. Ito, F. Miyake, S. Sumida and S. Torii, *Chem. Lett.*, 1999, **28**, 1053-1054; (c) K. Mitsudo, W. Matsuda, S. Miyahara and H. Tanaka, *Tetrahedron Lett.*, 2006, **47**, 5147-5150; (d) K. Mitsudo, T. Kawaguchi, S. Miyahara, W. Matsuda, M. Kuroboshi and H. Tanaka, *Org. Lett.*, 2005, **7**, 4649-4652; (e) G. Mata and N. W. Luedtke, *J. Org. Chem.*, 2012, **77**, 9006-9017; (f) H. Sugimura, K. Osumi, Y. Kodaka and K. Sujino, *J. Org. Chem.*, 1994, **59**, 7653-7660; (g) S. Knapp and W. Shieh, *Tetrahedron Lett.*, 1991, **32**, 3627-3630; (h) H. Sugimura, I. Muramoto, T. Nakamura and K. Osumi, *Chem. Lett.*, 1993, **22**, 169-172.
- (a) Y. J. Wang, J. Jia, Z. Y. Gu, F. F. Liang, R. C. Li, M. H. Huang, C. S. Xu, J. X. Zhang, Y. Men and G. W. Xing, *Carbohydr. Res.*, 2011, **346**, 1271-1276; (b) X. T. Zhang, Z. Y. Gu and G. W. Xing, *Carbohydr. Res.*, 2014, **388**, 1-7; (c) Z. Y. Gu, X. T. Zhang, J. X. Zhang and G. W. Xing, *Org. Biomol. Chem.*, 2013, **11**, 5017-5022; (d) Z. Y. Gu, J. X. Zhang, G. W. Xing, *Chem. Asian J.*, 2012, **7**, 1524-1528; (e) X. T. Zhang, Z. Y. Gu, L. Liu, S. Wang, G. W. Xing, *Chem. Commun.*, 2015, **51**, 8606-8609.
- Z. Zhang, I. R. Ollmann, X. Ye, R. Wischnat, T. Baasov and C. Wong, *J. Am. Chem. Soc.*, 1999, **121**, 734-753.
- (a) S. Dey and P. Garner, *J. Org. Chem.*, 2000, **65**, 7697-7699; (b) S. Fletcher, V. M. Shahani, A. J. Lough and P. T. Gunning, *Tetrahedron*, 2010, **66**, 4621-4632; (c) B. Y. Michel and P. Strazewski, *Tetrahedron*, 2007, **63**, 9836-9841; (d) A. Porcheddu, G. Giacomelli, I. Piredda, M. Carta and G. Nieddu, *Eur. J. Org. Chem.*, 2008, **2008**, 5786-5797.
- (a) P. Ciuffreda, S. Casati and A. Manzocchi, *Magn. Reson. Chem.*, 2007, **45**, 781-784; (b) J. Kjellberg and N. G. Johansson, *Tetrahedron*, 1986, **42**, 6541-6544; (c) M. T. Chenon, R. J. Pugmire, D. M. Grant, R. P. Panzica and L. B. Townsend, *J. Am. Chem. Soc.*, 1975, **97**, 4636-4642.
- (a) S. Hanessian, J. J. Conde, H. K. Hoan and B. Lou, *Tetrahedron*, 1996, **52**, 10827-10834; (b) J. Liao, J. Sun and B. Yu, *Tetrahedron Lett.* 2008, **49**, 5036-5038.