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High Performance Full Color OLEDs based on a Class of Molecules with Dual Carrier Transport Channels and Small Singlet-Triplet Splitting

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Two deep blue emitting materials **PPI-PPITPA** and **PPI-PPIPCz** with dual carrier transport property and small singlet-triplet splitting feature are designed and synthesized. **PPI-PPITPA** and **PPI-PPIPCz** not only were used as non-doped emitting layer to fabricate highly efficient deep blue OLEDs, but also as hosts to construct high performance green, yellow and red phosphorescent OLEDs.

Organic light-emitting devices (OLEDs) are of current interest from both scientific and practical points of view due to their applications in the next generation full-color flat-panel displays and solid-state lighting sources.^[1] Generally, fine quality full-color displays or white color lighting sources are constructed by rational combination of high performance deep blue (B), green (G), yellow (Y) and red (R) emitters together with different host materials.^[2] Therefore, achieving high performance R, Y, G and B electroluminescence (EL) based on simple material system with an aim to minimize the manufacture cost of materials and simplify the fabrication process is an important issue for OLED applications. In principle, a deep blue emitting material with balanced carrier transport characteristic and relatively high triplet energy $(E_{\rm T})$ may be employed as an efficient host for G, Y and R phosphorescent emitters, simultaneously. The high $E_{\rm T}$ enables G, Y and R phosphors to harvest the triplet energy of blue emitter. However, the high



performance non-doped blue electroluminescent materials are usually not suitable hosts for phosphorescent OLEDs (PhOLEDs) due to their low $E_{\rm T}$ and poor carrier transport property.^[3] On the other hand the efficient hosts for **C V** and **R** phosphore often

other hand, the efficient hosts for **G**, **Y** and **R** phosphors often displayed low efficiency when they were used as emitters in blue color OLEDs.^[4] So far, it is still a challenge to achieve highly efficient full color OLEDs based on a deep blue emitting material.

To gain the multi-functional organic materials, which can be used not only as emitters to fabricate highly efficient deep blue OLEDs, but also as hosts to construct highly efficient **G**, **Y** and **R** OLEDs, we proposed the molecular design and synthesis strategy as following: (**i**) the molecules are supposed to have hole transport moiety (donor) and electron transport moiety (acceptor) with high photoluminescent (PL) quantum yield (Φ_{PL}) in solid state; (**ii**) the electron transport moieties and the hole transport moieties should be able to separately assemble together to form electron and hole transport channels, respectively, in solid state; (**iii**) to obtain deep blue emission, the donor-acceptor (D-A) molecule should have relatively weak charge transfer property since strong D-A structure

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can induce red shifted emission; (iv) the singlet-triplet splitting (ΔE_{ST}) should be quite small to ensure the triplet excited state energy is high enough to excite **G** phosphorescent dopant. In this communication, we report two organic molecules (**PPI-PPITPA** and **PPI-PPIPCz**) composed of two phenyl-phenanthroimidazol (**PPI**) and one triphenylamine (**TPA**) or phenyl-carbazole (**PCz**) groups (**Figure 1**). Using **PPI-PPITPA** and **PPI-PPIPCz** as non-doped deep blue emitters and hosts for **G**, **Y** and **R** phosphorescent dopants, highly efficient full color OLEDs are fabricated.

PPI-PPITPA and PPI-PPIPCz are synthesized through a simple procedure (Scheme S1) and fully characterized by NMR, mass spectra, and element analyses (see Supporting Information). The PPI-PPITPA molecule displays a V-type geometry feature (Figure S1) as demonstrated by single-crystal X-ray analysis. The theoretically calculated molecular orbital distribution of PPI-PPITPA is shown in Figures 2a and 2b. The HOMO orbital is totally localized at PPITPA group, while the LUMO orbital distributes on the PPI group and imidazole ring of PPITPA. The molecular orbital distribution of PPI-PPIPCz is similar to PPI-PPITPA (Figure S2). Obviously, the HOMO and LUMO of PPI-PPITPA and PPI-PPIPCz display adequate separation feature, which benefits the hole- and electron-transport (bipolar property) and reduction of the singlet-triplet splitting.^[5] The molecular packing (Figure 2c) of PPI-PPITPA reveals that the PPI and PPITPA moieties separately aggregate into stack columns indicating that the electron and hole transport channels are perfectly constructed, respectively. Though stacking model in the film is different from that in single crystal, the current–voltage characteristics (Figure S3) of single carrier devices also verify that both compounds have bipolar transport ability. The PPI (LUMO) and PPITPA (HOMO) moieties individually undertake the electron and hole transport

revealing that this molecule should have good electron and hole transport abilities. Both of the two compounds exhibit excellent thermal stability as evidenced by their high decomposition temperatures (corresponding to 5% weight loss) of 568 °C for **PPI-PPITPA** and 524 °C for **PPI-PPIPCz** (Figure S5). The glass-transition temperatures (T_g) are as high as 187 °C for **PPI-PPITPA** and 193 °C for **PPI-PPIPCz** (Figure S6). The distinguished thermal properties of the two compounds are favourable to form uniform thin films upon thermal evaporation, which are advantageous for application in OLEDs.

The photophysical properties of PPI-PPITPA and PPI-PPIPCz were analyzed by UV-vis and PL measurements (Figure 3 and Table S1). In dilute CH₂Cl₂ solution, the two compounds exhibit intense violet to deep blue emission with high Φ_{PL} of 0.90 for **PPI**-PPITPA and 0.73 for PPI-PPIPCz. And the emission maximum of PPI-PPIPCz (402 nm) is much blue-shifted with respect to that of PPI-PPITPA (446 nm) in solution, which should be ascribed to the one more benzene ring in PPI-PPITPA, inducing an elongated conjugation. Interestingly, in the thin film state, **PPI-PPIPCz** shows similar emission band with **PPI-PPITPA**, peaking at around 450 nm. The Φ_{PL} of the thin films were measured to be 0.53 for **PPI-PPITPA** and 0.89 for PPI-PPIPCz. Such high fluorescence efficiencies are essential for high-efficiency deep blue OLEDs. Based on lowtemperature PL measurements, the triplet energy levels are estimated to be 2.40 eV for PPI-PPITPA and 2.48 eV for PPI-PPIPCz, respectively, which are sufficiently high for the excitation of GYR phosphorescent dopants. The ΔE_{ST} values of PPI-PPITPA and PPI-PPIPCz were calculated to be 0.38 eV and 0.35 eV, respectively. The small ΔE_{ST} are advantageous for efficient energy transfer from the triplet excited state of hosts to GYR phosphorescent emitters.



Figure 2. Calculated spatial distributions of the HOMO (a) and LUMO (b) and molecular packing (c) of **PPI-PPITPA**.

functions. The calculated electron (t_e) and hole (t_h) transfer integrals for **PPI-PPITPA** are 0.025 and 0.061 eV, respectively, suggesting that **PPI-PPITPA** can act as a bipolar transport material. The **PPI-PPITPA** shows clear reduction and oxidation waves (Figure S4),



Figure 3. Room-temperature UV-vis absorption spectra, PL spectra of **PPI-PPIPA** and **PPI-PPIPCz** in CH_2Cl_2 solution and in neat films, as well as their phosphorescence spectra in thin film at 77 K.

We first examined the two compounds' EL property as blue emitters by fabricating non-doped OLEDs with a structure of [ITO/NPB (40 nm)/TCTA (5 nm)/**PPI-PPITPA** or **PPI-PPIPCz** (30 nm)/TPBI (50 nm)/LiF (1 nm)/Al] (device **B1** with emitting layer (EML) composed of **PPI-PPITPA** and device **B2** with EML composed of **PPI-PPIPCz**). The energy-level diagram and molecular structures of the used materials are shown in **Figure S7**. In devices **B1** and **B2**, we utilized NPB (4,4'-bis[N-(1-naphthyl)-Nphenyl amino] biphenyl) as the hole-transporting layer (HTL), TCTA (4,4',4''- tris(Ncarbazolyl)triphenylamine) as the excitonblocking layer, TPBI (1,3,5-tris(*N*-phenylbenzimidazol-2-yl)benzene) as the electron-transporting layer (ETL) and hole-blocking layer. LiF served as electron-injecting layer and Al served as cathode. The nonJournal Name

doped devices B1 and B2 exhibit deep-blue emission with CIE coordinates of (0.15, 0.08) and (0.15, 0.07), respectively, both of which are very close to the NTSC (National Television Standards Committee) blue standard (CIE: 0.14, 0.08) and remain almost unchanged over a wide range of driving voltage. As shown in Figure 4 and Figure S8, the maximum external quantum efficiencies (EQEs) of Device B1 and B2 are 7.7% and 8.1%, respectively. Device B1 using **PPI-PPITPA** as emitter shows a low turn-on voltage of 3.2 V and a maximum luminance of 13950 cd m⁻² at 10.5 V. The maximum current (CE) and power (PE) efficiencies of this device are 5.6 cd A⁻¹ and 5.5 lm W⁻¹, respectively. Device B2 shows a similar performance. It turns on at 3.4 V and achieves a maximum brightness of 13820 cd m⁻² at 10.5 V. The maximum CE and PE can be up to 4.8 cd A⁻¹ and 4.5 lm W⁻¹. To the best of our knowledge, these efficiencies are the highest values reported for non-doped blue OLEDs with CIEy below 0.10.^[6] Moreover, the EQE of these two devices maintains 5.4% and 6.0% respectively at a high brightness of 1000 cd m⁻², indicating relatively small efficiency roll-offs.



Figure 4. (a) Current density–voltage–brightness (*J*–*V*–*L*) characteristics; (b) Current efficiencies and power efficiencies versus current density curves; (c) External quantum efficiency versus luminance characteristics; (d) EL spectra and CIE1931 coordinates of devices **B2**, **G2**, **Y2** and **R2** based on **PPI-PPIPCz**.

The EQE of OLEDs can be expressed by the well-known equation of EQE = $\eta_{out} \times \eta_{rc} \times \eta_{\gamma} \times \Phi_{PL}$,^[7] where η_{out} is the light-out-coupling efficiency (~20%), $\eta_{\rm rc}$ is the product of the charge recombination efficiency that is ideally 100% if holes and electrons are fully balanced and completely recombined to form excitons, η_{γ} is efficiency of radiative exciton production (25% for conventional fluorescent OLEDs), Φ_{PL} is photoluminescence (PL) quantum yield of the emitter molecules (0.53 for PPI-PPITPA and 0.89 for PPI-PPIPCz). Thus, the maximum EQE values of PPI-PPITPA and PPI-PPIPCz devices can be calculated as 2.7% and 4.5%, respectively. Clearly, the obtained EQEs (7.7% for PPI-PPITPA, 8.1% for PPI-PPIPCz) overcome the theoretical EQE limit of 5% for the conventional fluorescent OLEDs. Since no delayed fluorescence was observed from transient PL (Figure S8), the high EQEs did not seem to be in accordance with TADF.^[8] At this stage we cannot give precise explanation for this phenomena and detail experiments should be performed to disclose the mechanism.

To explore the potential applications of these two compounds as host materials for G, Y and R phosphorescent dopants, we fabricated GYR PhOLEDs with configurations of [ITO/NPB (40 nm)/TCTA (5 nm)/EML (30 nm)/TPBI (50 nm)/LiF (1 nm)/Al (100 nm)] (EML: PPI-PPITPA:5 wt% Ir(ppy)₃ for device G1, PPI-PPITPA:5 wt% Ir(ppy)₂(dipig) for device **Y1**, **PPI-PPITPA**:8 wt% Ir(MDQ)₂(acac) for device R1 and PPI-PPIPCz:5 wt% Ir(ppy)₃ for device G2, PPI-PPIPCz:5 wt% Ir(ppy)₂(dipig) for device Y2, PPI-PPIPCz:8 wt% $Ir(MDQ)_2(acac)$ for device **R2**). The green phosphorescent complex Ir(ppy)₃ is *fac*-tris(2-phenylpyridine) iridium(III), yellow complex bis(2-phenylpyridinato)-N,N'-diisopropyl-Ir(ppy)₂(dipig) is diisopropylguanidinate iridium(III) and red complex Ir(MDQ)₂(acac) bis(2-methyldibenzo-[f,h]quinoxaline) is acetylacetonate iridium(III) (Figure S9).

Figure 4 and Figure S8 present the current density-voltagebrightness (J-V-L) characteristics, device performances and EL spectra of the devices. The EL spectra are identical to the PL spectra (Figure S10) of the doped thin films. Device G2 exhibits a low turn-on voltage of 2.8 V. The maximum EQE of 17.8% and PE of 65.7 lm W⁻¹ are among the best values for green phosphorescent OLEDs with blue emitters as hosts.^[9] However, device G1 shows a much lower EL efficiency than device G2. The reason can be attributed to the lower triplet energy level ($E_T = 2.40 \text{ eV}$) of **PPI**-PPITPA, which may cause back energy transfer from the green guest triplet states to the host, resulting in a loss in efficiency. Yellow PhOLEDs with excellent performance were also achieved by employing these two compounds as hosts. For instance, devices Y1 exhibits the maximum EQE of 19.1%, CE of 61.8 cd A⁻¹, and PE of 65.5 lm W^{-1} , and device Y2 shows the similar behavior with the maximum EQE of 18.1%, CE of 56.5 cd A⁻¹, and PE of 58.8 lm W⁻¹. Finally, deep red PhOLEDs were fabricated by using PPI-PPITPA and PPI-PPIPCz as host. Device R2 exhibits the maximum luminance of 24810 cd m⁻² at 13.5 V and excellent EL efficiencies (18.0%, 21.3 cd A⁻¹, and 20.9 lm W⁻¹), whereas device R1 exhibits the maximum luminance of 18020 cd m⁻² at 11.0 V and EL efficiencies (16.8%, 19.8 cd A⁻¹, and 21.5 lm W⁻¹) (Figure 4 and Figure S8). The PE values of devices R1 and R2 are comparable to recently reported highly efficient deep red PhOLEDs.^[10] Above experimental results demonstrates that PPI-PPITPA and PPI-PPIPCz are universal host materials for GYR phosphorescent emitters. The detail EL parameters of the RYGB OLEDs are summarized in Table 1.

In conclusion, we have designed and synthesized two deep blue emitting materials **PPI-PPITPA** and **PPI-PPIPCz** with dual carrier transport property and small singlet-triplet splitting feature. The single crystal study of **PPI-PPITPA** revealed that the **PPI** and **PPITPA** moieties assemble into electron and hole transport channels, respectively, in solid state. The strong deep blue emission and good ambipolar carrier transport characteristics of **PPITPA** and **PPI-PPIPCz** ensure that the OLEDs with the two compounds as nondopant EML display extremely high EQE of 7.7%, 8.1% with true deep blue CIE (0.15, 0.08) and (0.15, 0.07). Furthermore, the small singlet-triplet splitting and good carrier transport properties allow **PPITPA** and **PPI-PPIPCz** as hosts to fabricate **RYG** phosphorescent emitting layers for PhOLEDs. High performance **G**, **Y** and **R** PhOLEDs with EQE of 17.8%, 18.1% and 18.0%, respectively, have been successfully obtained based on the two compounds and phosphors. These results demonstrate an efficient **Table 1.** Electroluminescent performances of the devices.^{a)}

Device	Dopant	V_{on}/V	Lmax/cd m ⁻²	CE ^{b)} /cd A ⁻¹	$PE^{b}/lm W^{-1}$	EQE ^{b)} /%	CIE $(x,y)^{c)}$
B1	no	3.2	13950	5.6, 4.6, 3.9	5.5,3.2,2.0	7.7, 6.4, 5.4	0.15, 0.08
B2	no	3.4	13820	4.8, 4.1, 3.6	4.5, 2.8, 1.8	8.1, 6.8, 6.0	0.15, 0.07
G1	Ir(ppy) ₃	2.8	7022	26.6, 11.5, 3.7	29.9, 8.3, 1.5	7.9, 3.5, 1.0	0.32, 0.60
G2	Ir(ppy) ₃	2.8	55020	60.6, 51.5, 41.6	65.7, 41.9, 24.2	17.8, 15.3, 12.4	0.32, 0.61
¥1	Ir(ppy)2(dipig)	2.9	68990	61.8, 55.5, 50.7	65.5, 37.6, 24.7	19.1, 17.2, 15.8	0.46, 0.53
¥2	Ir(ppy)2(dipig)	3.0	75350	56.5, 49.7, 46.6	58.8, 32.2, 23.7	18.1, 15.8, 15.1	0.45, 0.54
R1	Ir(MDQ) ₂ (acac)	2.8	18020	19.8, 18.3, 16.3	21.5, 14.1, 10.1	16.8, 15.6, 13.8	0.64,0.36
R2	Ir(MDQ) ₂ (acac)	2.3	39138	21.3, 16.9, 14.8	20.9, 11.2, 6.7	18.0, 14.2, 12.4	0.64,0.36

²⁷ Abbreviation: V_{on} : Turn-on voltage. L_{max} : Maximum luminance. CE: The maximum current efficiency. PE: The maximum power efficiency. EQE: The maximum external quantum efficiency. ^{b)} in the order of maximum, then values at 100 and 1000 cd m⁻². ^{c)} Measure at 100 cd m⁻².

strategy to achieve full color OLEDs by developing the deep blue organic luminescent materials with ideal **D-A** molecular structure and molecular packing in solid state as well as small singlet-triplet splitting.

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Notes and references

- (a) M. A. Baldo, M. E. Thompson, S. R. Forrest, *Nature*, 2000, 403, 750; (b) S. Reineke, F. Lindner, G. Schwartz, N. Seidler, K. Walzer, B. Lüssem, K. Leo, *Nature*, 2009, 459, 234; (c) A. C. Grimsdale, K. L. Chan, R. E. Martin, P. G. Jokisz, A. B. Holmes, *Chem, Rev.*, 2009, 109, 897; (d) W. C. H. Choy, W. K. Chan, Y. Yuan, *Adv. Mater.*, 2014, 26, 5368; (e) T. Fleetham, G. Li, L. Wen, J. Li, *Adv. Mater.*, 2014, 26, 7116.
- (a) S. Chen, G. Tan, W-Y. Wong, H-S. Kwok, Adv. Funct. Mater., 2011, 21, 3785; (b) G. M. Farinola, R. Ragni, Chem. Soc. Rev., 2011, 40, 3467; (c) B. P. Yan, C. C. C. Cheung, S. C. F. Kui, H. F. Xiang, V. A. L. Roy, S. J. Xu, C. M. Che, Adv. Mater., 2007, 19, 3599; (d) H. Sasabe, J. I. Takamatsu, T. Motoyama, S. Watanabe, G. Wagenblast, N. Langer, O. Molt, E. Fuchs, C. Lennartz, J. Kido, Adv. Mater., 2010, 22, 5003; (e) R. J. Wang, D. Liu, H. C. Ren, T. Zhang, H. M. Yin, G. Y. Liu, J. Y. Li, Adv. Mater., 2011, 23, 2823; (f) K. T. Kamtekar, A. P. Monkman, M. R. Bryce, Adv. Mater., 2010, 22, 572; (g) M. C. Gather, A. Köhnen, K. Meerholz, Adv. Mater., 2011, 23, 233; (h) C. Han, Z. Zhang, H. Xu, J. Li, G. Xie, R. Chen, Y. Zhao, W. Huang, Angew. Chem. Int. Ed., 2012, 51, 10104; (i) C. Fan, L. Zhu, T. Liu, B. Jiang, D. Ma, J. Qin, C. Yang, Angew. Chem. Int. Ed., 2014, 53, 2147.
- 3 (a) W. Y. Lai, Q. Y. He, R. Zhu, Q. Q. Chen, W. Huang, Adv. Funct. Mater., 2008, 18, 265; (b) L. Wang, Y. Jiang, J. Luo, Y. Zhou, J. H. Zhou, J. Wang, J. Pei, Y. Cao, Adv. Mater., 2009, 21, 4854; (c) L. Xiao, Z. Chen, B. Qu, J. Luo, S. Kong, Q. Gong, J. Kido, Adv. Mater., 2011, 23, 926; (d) C.-G. Zhen, Y.-F. Dai, W.-J. Zeng, Z. Ma, Z.-K. Chen, J. Kieffer, Adv. Funct. Mater., 2011, 21, 699.
- 4 (a) K. Wang, F. C. Zhao, C. G. Wang, S. Y. Chen, D. Chen, H. Y. Zhang, Y. Liu, D. G. Ma, Y. Wang, *Adv.Funct. Mate.*, 2013, 23, 2672;

(b) K. Wang, S. P. Wang, J. B. Wei, S. Y. Chen, D. Liu, Y. Liu, Y. Wang, J. Mater. Chem. C., 2014, 2, 6817.

- 5 (a) L. Duan, J. Qiao, Y. D. Sun, Y. Qiu, *Adv. Mater.*, 2011, 23, 1137;
 (b) C.-J. Zheng, J. Wang, J. Ye, M. -F. Lo, X.-K. Liu, M.-K. Fung, X.-H. Zhang, C.-S. Lee, *Adv. Mater.*, 2013, 25, 2205; (c) D. D. Zhang, L. Duan, Y. L. Li, H. Y. Li, Z. Y. Bin, D. Q. Zhang, J. Qiao, G. D. Dong, L. D. Wang, Y. Qiu, *Adv. Funct. Mater.*, 2014, 24, 3551.
- 6 (a) W. J. Li, Y. Y. Pan, R. Xiao, Q. M. Peng, S. T. Zhang, D. G. Ma,
 F. Li, F. Z. Shen, Y. H. Wang, B. Yang, Y. G. Ma, *Adv. Funct. Mater.*, 2013, 24, 1609; (b) R. Kim, S. Lee, K. H. Kim, Y. J. Lee, S.
 K. Kwon, J. J. Kim and Y. H. Kim, *Chem. Commun.*, 2013, 49, 4664;
 (c) J. N. Moorthy, P. Venkatakrishnan, D. F. Huang, T. J. Chow, *Chem. Commun.*, 2008, 2146-2148.
- 7 V. Jankus, C. -J. Chiang, F. Dias, A. P. Monkman, *Adv. Mater.*, 2013, 25, 1455.
- 8 (a) H. Uoyama, K. Goushi, K. Shizu, H. Nomura, C. Adachi, *Nature*, 2012, 492, 234; (b) Q. Zhang, J. Li, K. Shizu, S. Huang, S. Hirata, H. Miyazaki, C. Adachi, *J. Am. Chem. Soc.*, 2012, 134, 14706; (c) S. Y. Lee, T. Yasuda, Y. S. Yang, Q. Zhang, C. Adachi, *Angew. Chem. Int. Ed.*, 2014, 53, 6402; (d) J. W. Sun, J. -H. Lee, C. -K. Moon, K. -H. Kim, H. Shin, J. -J. Kim, *Adv. Mater.*, 2014, 26, 5684; (e) D. Zhang, L. Duan, C. Li, Y. Li, H. Li, D. Zhang, Y. Qiu, *Adv. Mater.*, 2014, 26, 5050; (f) X. K. Liu, Z. Chen, C. J. Zheng, M. Chen, W. Liu, X. H. Zhang, C. S. Lee, *Adv. Mater.*, 2015, DOI: 10.1002/adma.201500013.
- ⁹ (a) H. H. Chou, C. H. Cheng, Adv. Mater., 2010, 22, 2468; (b) S.-J. Su,
 C. Cai and J. Kido, Chem. Mater., 2011, 23, 274; (c) S. L. Gong, Y.
 H. Chen, J. J. Luo, C. L. Yang, C. Zhong, J. G. Qin, D. G. Ma, Adv.
 Funct. Mater., 2011, 21, 1168; (d) X. K. Liu, C. J. Zheng, M. F. Lo, J.
 Xiao, C. S. Lee, M. K. Fungc, X. H. Zhang, Chem. Commun., 2014,
 50, 2027; (e) P.-I. Shih, C.-H. Chien, F.-I. Wu, C.-F. Shu, Adv. Funct.
 Mater., 2007, 17, 3514.
- (a) Y. Tao, Q. Wang, Y. Shang, C. L. Yang, L. Ao, J. Qin, D. G. Ma, Z. Shuai, *Chem. Commun.*, 2009, 77; (b) H. Sasabe , T. Chiba , S.-J. Su , Y. J. Pu , K. I. Nakayama . J. Kido, *Chem. Commun.*, 2008, 5821; (c) F. M. Hsu, C. H. Chien, Y. J. Hsieh, C. H. Wu, C. F. Shu, S. W. Liu, C. T. Chen, *J. Mater. Chem.*, 2009, **19**, 8002; (d) C. H. Chen, L. C. Hsu, P. Rajamalli, Y. W. Chang, F. L. Wu, C. Y. Liao, M. J. Chiu, P. Y. Chou, M. J. Huang, L. K. Chu, C. H. Cheng, *J. Mater. Chem.* C., 2014, **2**, 6183; (e) C. H. Fan, P. P. Sun, T. H. Su, C. H. Cheng, *Adv. Mater.*, 2011, **23**, 2981.

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